

Supporting Information

**Mitigating Volume Expansion and Enhancing Cycling Stability of Ferrous
Fluosilicate-Modified Silicon-Based Composite Anodes for Lithium-Ion Batteries**

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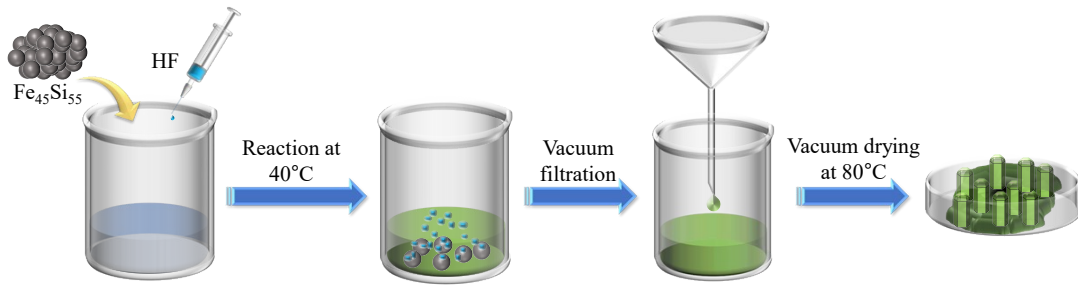


Fig. S1. Schematic illustration for the preparation of FeSiF₆.

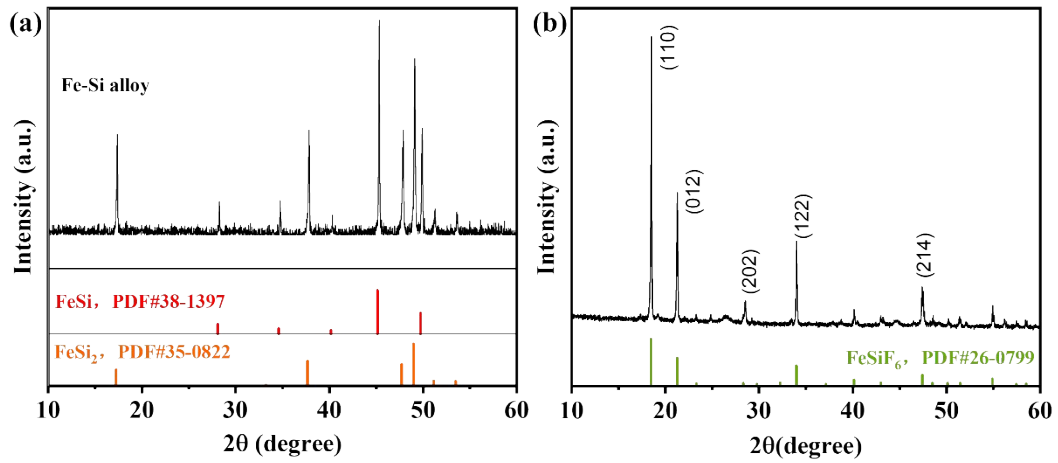


Fig. S2. XRD results of (a) Fe₄₅Si₅₅ and (b) as-synthesized FeSiF₆ composites.

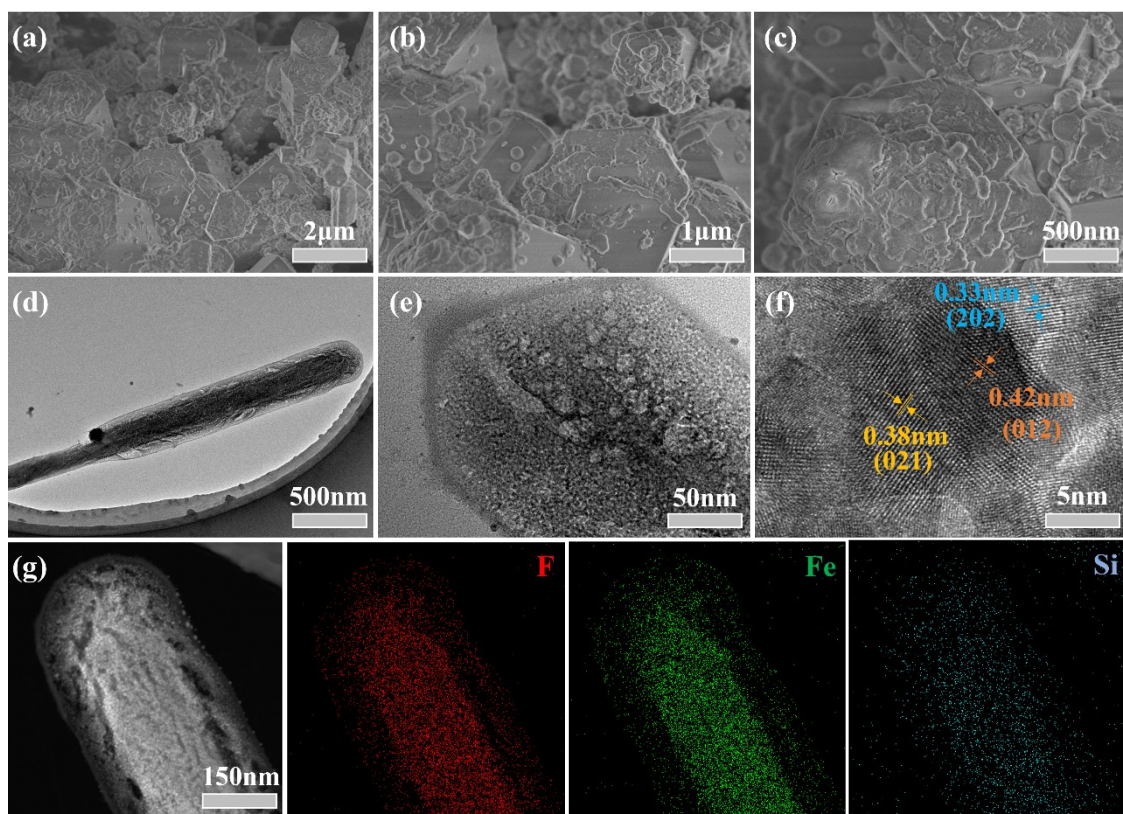


Fig. S3. (a)-(c) SEM images of pristine FeSiF_6 crystals. (d)-(e) TEM images of pristine FeSiF_6 crystals. (f) Inverse FFT image of which shows (202), (012) and (021) lattice planes of FeSiF_6 . (e) HAADF-STEM image of a single FeSiF_6 wire, (f-h) The corresponding elemental mapping results of F, Fe and Si.

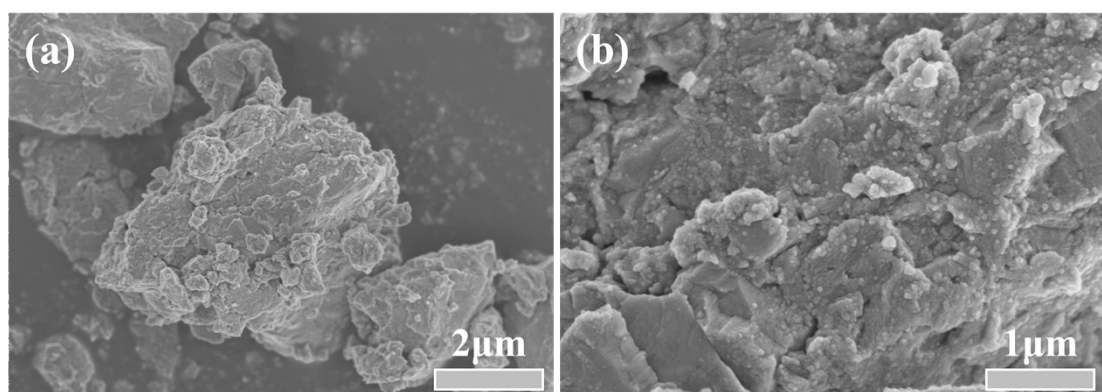


Fig. S4. (a)-(b) SEM images of pristine $\text{Fe}_{45}\text{Si}_{55}$ crystals.

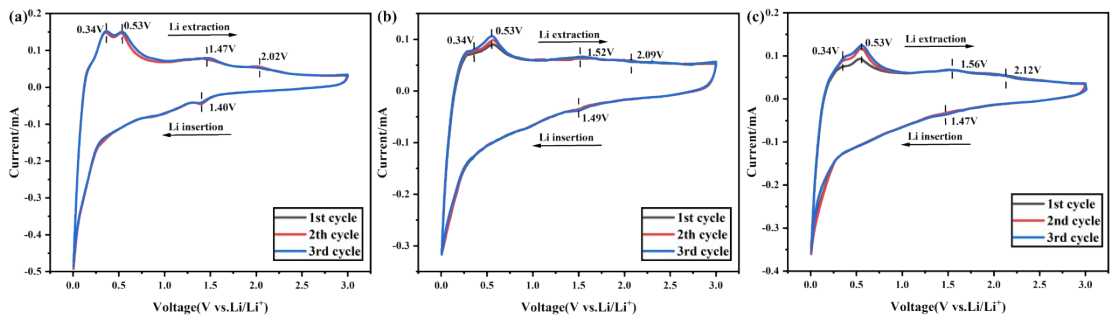


Fig. S5. Cyclic voltammetry profiles of (a) Fe-Si@C-T, (b) Fe-Si-T@C and (c) Fe-Si@F@C electrodes at a sweep rate of 0.5 mV s^{-1} .

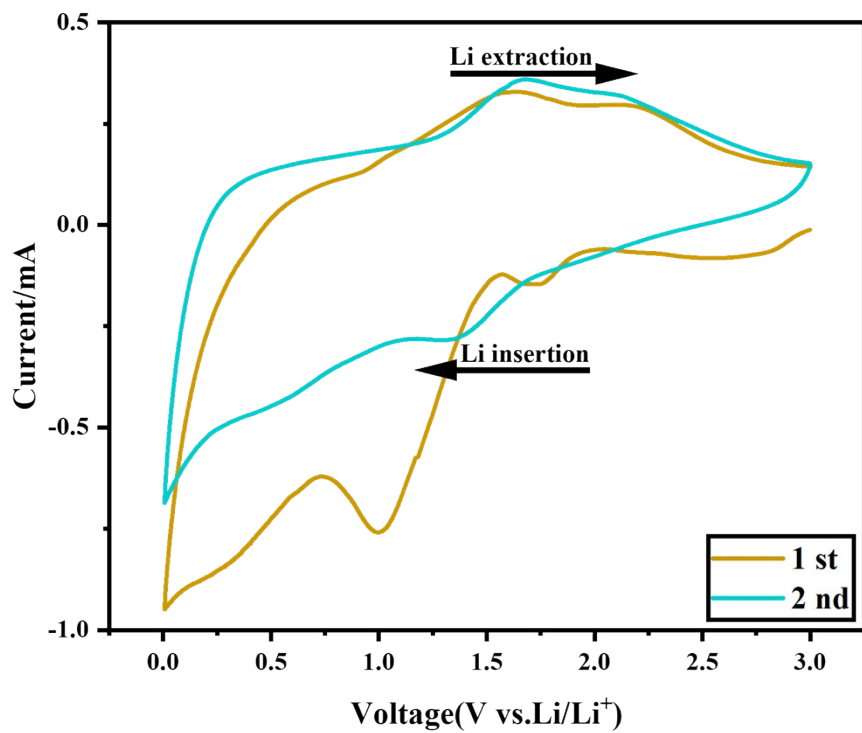


Fig. S6. Cyclic voltammetry profiles of FeSiF₆ composite anode in the initial five cycles at a sweep rate of 0.1 mV/s .

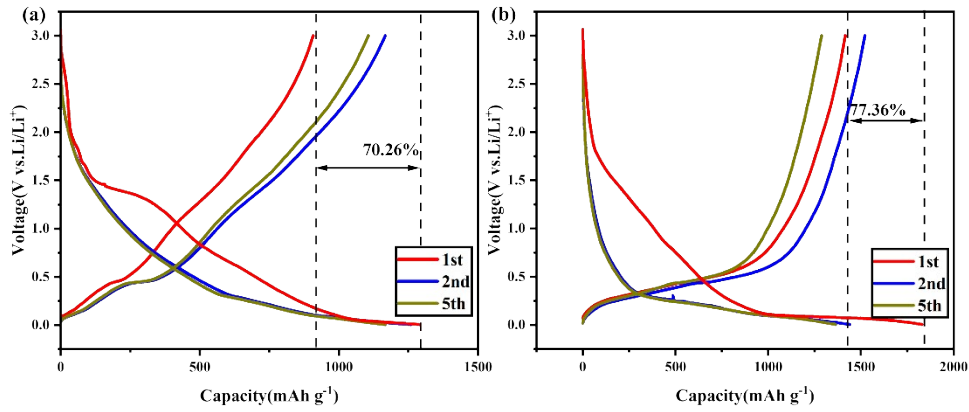


Fig. S7. Galvanostatic charge-discharge curves of (a) Fe-Si@C-T and (b) Fe-Si-T@C.

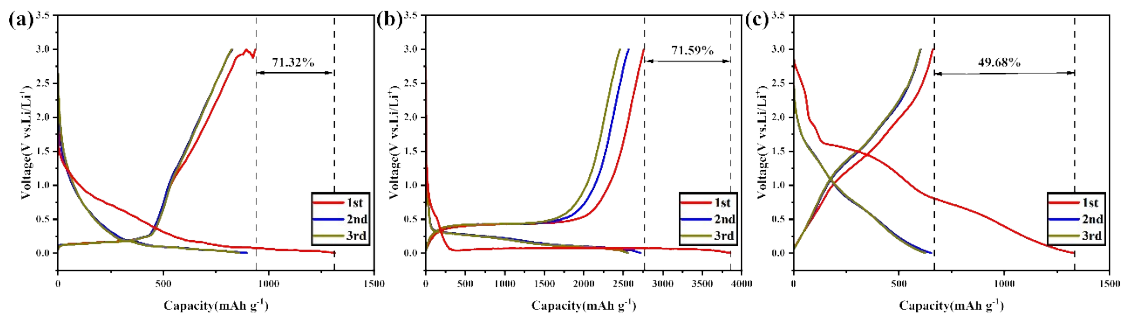


Fig. S8. GCD curves of (a) graphite (b) Si and (c) FeSiF₆ in the 1st, 2nd and 3rd cycles measured at 0.1A g⁻¹.

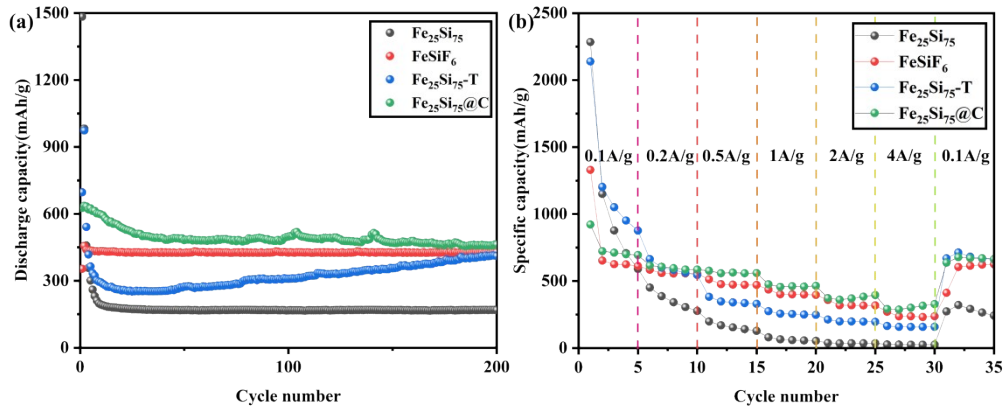


Fig. S9. (a) Cycling performance of Fe₂₅Si₇₅, FeSiF₆, Fe₂₅Si₇₅-T, and Fe₂₅Si₇₅@C at a current density of 1 A/g and corresponding fit curves. (b) Rate performance of Fe₂₅Si₇₅, FeSiF₆, Fe₂₅Si₇₅-T, and Fe₂₅Si₇₅@C at different current densities.

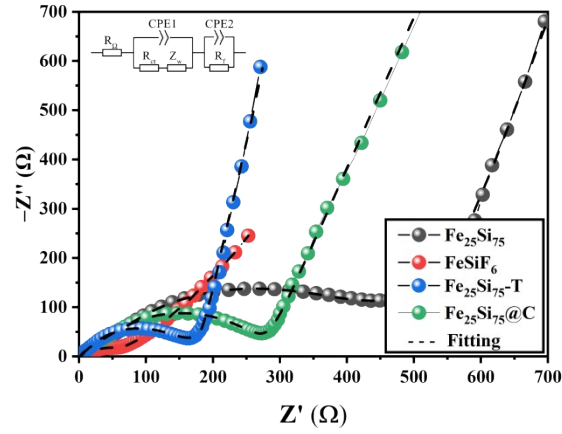


Fig. S10. EIS spectra of $\text{Fe}_{25}\text{Si}_{75}$, FeSiF_6 , $\text{Fe}_{25}\text{Si}_{75}\text{-T}$ and $\text{Fe}_{25}\text{Si}_{75}\text{@C}$ with a scan rate of 0.1 mV/s. The inset shows the equivalent circuit model used for EIS curve fitting.

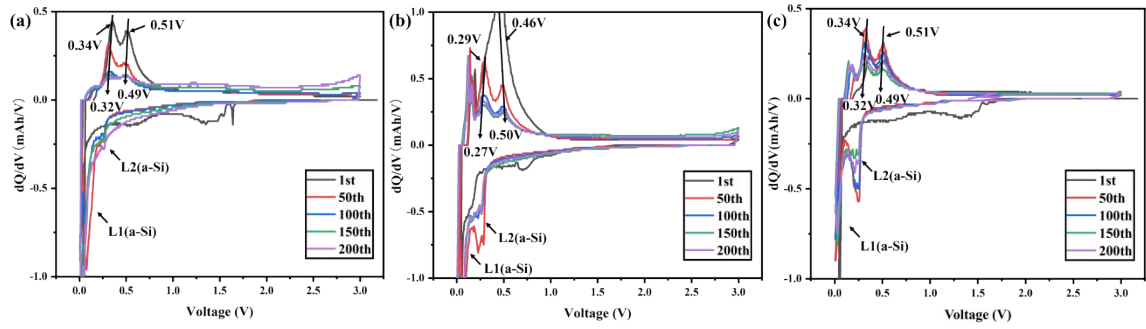


Fig. S11. The differential capacity curves of (a) Fe-Si@C-T , (b) Fe-Si-T@C , and (c) Fe-Si@F@C after different cycles at 1A/g.

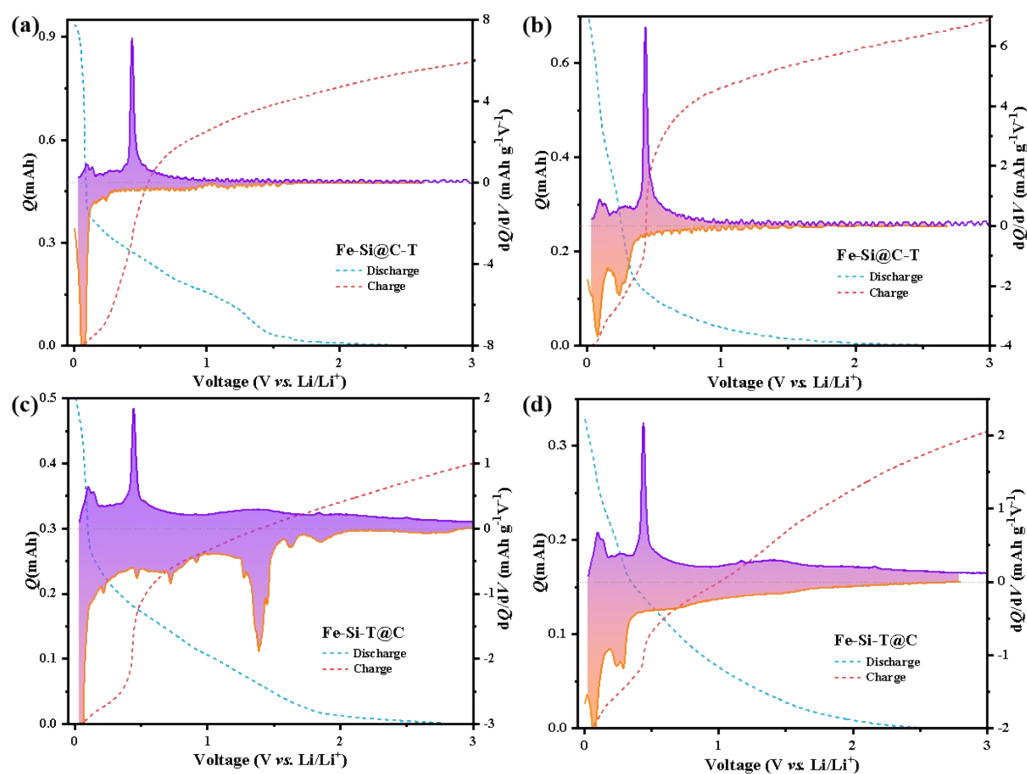


Fig. S12. Differential capacity curves for cycles of Fe-Si@C-T at (a) 1st cycle and (b) 5th cycle at 0.1A/g. (c) Differential capacity curves of Fe-Si-T@C in (a) 1st cycle and (b) 5th cycle at 0.1A/g.

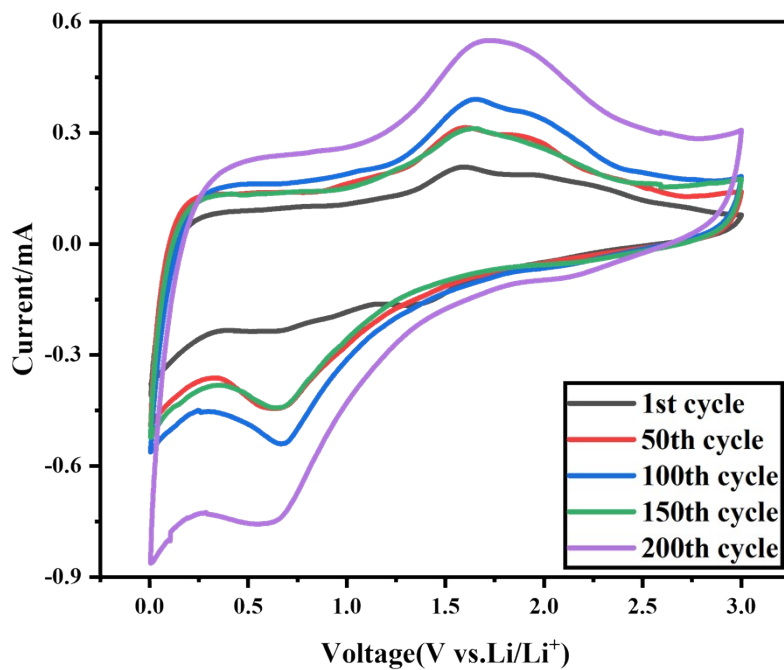


Fig. S13. Cyclic voltammety profiles of FeSiF₆ anode in the initial five cycles at a sweep rate of 0.5 mV s⁻¹.

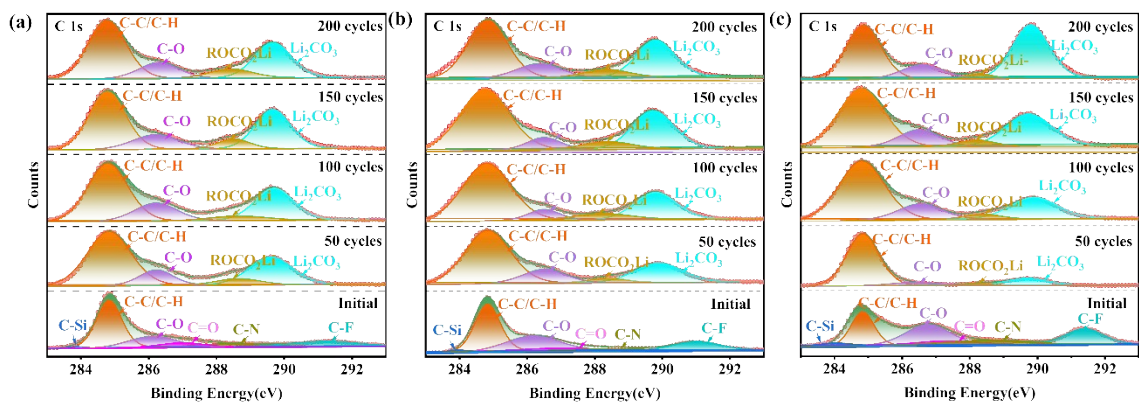


Fig. S14. High-resolution XPS spectra of C 1s of (a) Fe₂₅Si₇₅, (b) FeSiF₆, and (c) Fe-Si@F@C under different cycle times.

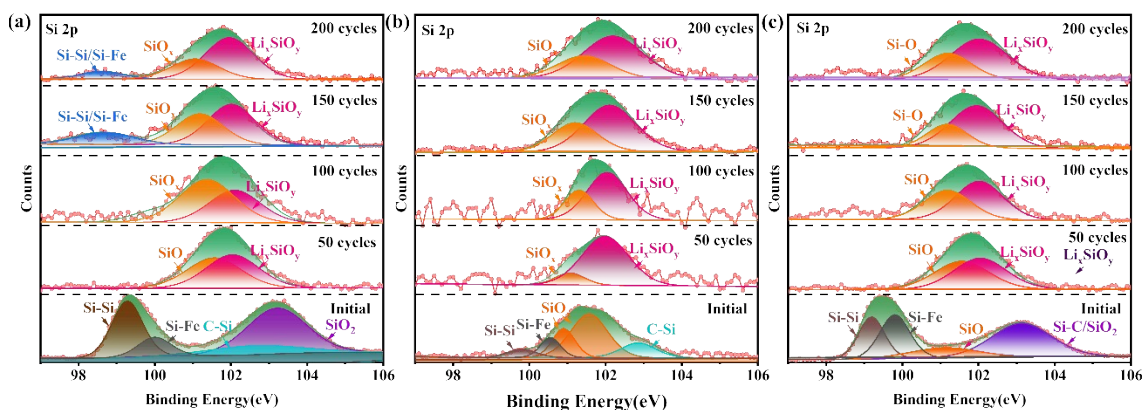


Fig. S15. High-resolution XPS spectra of Si 2p of (a) Fe₂₅Si₇₅, (b) FeSiF₆, and (c) Fe-Si@F@C under different cycle times.

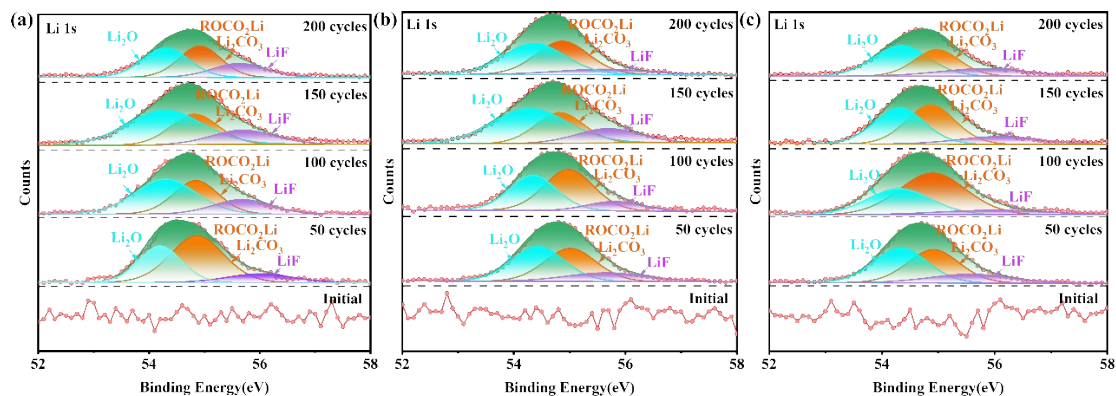


Fig. S16. High-resolution XPS spectra of Li 1s of (a) Fe₂₅Si₇₅, (b) FeSiF₆, and (c) Fe-Si@F@C under different cycle times.

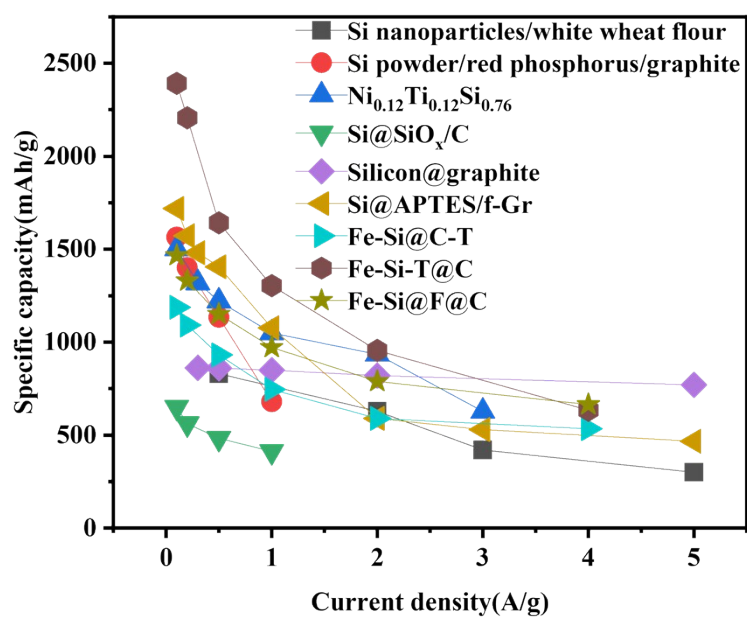


Fig. S17. Comparison of specific capacities of Fe-Si@C-T, Fe-Si-T@C, Fe-Si@F@C (this study), and other silicon-based composite anodes in literature.

Table S1. Performance comparison of silicon-based composite anodes.

Materials	Synthetic method	Synthetic method	Initial CE (%)	Anode performance	Rate	Reference
Si nanoparticles/white wheat flour	Ball milling and annealing	Core shell	71	880 mAh g ⁻¹ 92% retention after 90 cycles	210 mA g ⁻¹	1
Si powder/red phosphorus/graphite	Ball milling	Core shell	64.7	883.4 mAh g ⁻¹ after 200 cycles	200 mA g ⁻¹	2
Micronized Si/citric acid	Ball milling and carbonization	Core shell	82	850 mAh g ⁻¹ after 100 cycles	210 mA g ⁻¹	3
Amorphous-Si@SiO _x /C	Ball milling and carbonization of the citric acid	Core shell	82	1230 mAh g ⁻¹ after 100 cycles	500 mA g ⁻¹	4
N-doped carbon embedding Si nanoparticles	ball milling and pyrolysis carbonization	Core shell	80.8	794.7 mAh g ⁻¹ after 1000 cycles	1 A g ⁻¹	5
Ni _{0.12} Ti _{0.12} Si _{0.76}	One-step ball-milling method	The ternary compound	81.8	500 mAh g ⁻¹ after 50 cycles	840 mA g ⁻¹	6
Si@SiO _x /C	Two-step ball-milling	Core shell	82	726 mAh g ⁻¹ after 500 cycles	100 mA g ⁻¹	7
Silicon@graphite	a facile and scalable ball milling	Core shell	77	646.5 mAh g ⁻¹ after 100 cycles	200 mA g ⁻¹	8

References

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