Supplementary Information

Mo/PANI co-modified ultra-high nickel NCM9622 cathode composite with excellent cycle stability and high-rate performance for power batteries

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GITT test methods

Firstly, the newly assembled half-cells are subjected to three charge-discharge cycles at 0.1 C for initial activation, and then the cells are tested for charge-discharge at a current density of 0.1 C, a pulse time of 10 min, a relaxation time of 30 min, and a voltage window of 2.7-4.4 V. The Li⁺ diffusion coefficients (D_{Li^+}) can be calculated by the following equations:

$$
D_{Li} = \frac{4}{\pi \tau} \left[\left(\frac{m_B V_M}{M_B S} \right) \left(\frac{\Delta E_s}{\Delta E_\tau} \right) \right]^2 \left(\tau \ll L^2 / D \right)
$$

Where m_B is the mass (g) of the active substance, V_m and M_B are the molar volume $(\text{cm}^3 \text{ mol}^{-1})$ and molecular weight (g mol⁻¹) respectively, *S* is the electrode area (cm²), τ is the duration (s) of a single current pulse, ΔE_s is the steady-state voltage difference (V) after relaxation for 30 min, and ΔE_{τ} is the transient voltage difference (V) after the current pulse. The parameters ΔE_s , ΔE_t and τ in the formula can be obtained from the single pulse GITT curves of Figs. 6(e-g). Based on the data of discharge voltage (*V*) and corresponding pulse time (*τ*) in Fig. 6(e-g), the relationship between the discharge voltage (V) and the square root of pulse time $(\tau^{1/2})$ is obtained, as shown in Fig. $6(i)$

Fig. S1† Enlarged view corresponding to the (003) and (104) diffraction peaks in the XRD pattern.

Fig. S2† TGA curves of NCM96-Mo and NCM96-Mo@PANI materials.

Fig. S3† Mapping of O, Mn, Co and Ni elements of NCM96-Mo@PANI in STEM.

Fig. S4† (a) N 1s XPS spectra of NCM96-Mo@PANI material; (b) XPS full spectra of NCM9622 and NCM96-Mo@PANI materials.

Fig. S5† The first three cycles charge-discharge curves of NCM9622 (a), NCM96-Mo (b) and NCM96-Mo@PANI (c) materials at 2.7-4.4 V and 0.1 C.

Fig. S6† (a) Rate performances of NCM9622 and NCM9622@PANI materials, and (b) the first three cycles charge-discharge curves of NCM9622@PANI material at 2.7-4.4 V and 0.1 C.

Fig. S7† Cyclic performances of NCM9622, NCM96-Mo and NCM96-Mo@PANI samples at 2.7-4.4 V and 0.5 C.

Fig. S8† EIS diagrams of (a) NCM9622, (b) NCM96-Mo and (c) NCM96-Mo@PANI materials before and after 100 cycles at 1 C.

Fig. S9† CV curves of (a) NCM9622, (b) NCM96-Mo, (c) NCM96-Mo@PANI materials at the first three cycles.

Fig. S10† Integral results of the spin-up total density of states (TDOS) of NCM9622 and NCM96-Mo materials between 0 and 3 eV

Fig. S11† TEM image of NCM96-Mo@PANI material after 100 cycles at 1 C

Fig. S12† SEM image of $\text{Ni}_{0.96}\text{Co}_{0.02}\text{Mn}_{0.02}(\text{OH})_2$ precursor

Fig. S13† Cycling performances of NCM96-Mo material with 0.5%, 1% and 1.5%Mo doping.

Samples	$a=b(A)$	c(A)	c/a	$V(A^3)$	Li/Ni exchange $(\%)$	$R_{\rm wp}$ (%)
NCM9622	2.87228	14.18206	4.937562	101.327	2.39%	6.3
NCM96-Mo	2.87478	14.19823 4.938893		101.619	3.45%	6.4
NCM96-Mo@PANI	2.87426	14.20372 4.941696		101.621	3%	6.3

Table S1† Rietveld refinement data of materials

Samples	$100th$ capacity retention/rate	The range of operating voltage/V	References
NCM96-Mo@PANI	90.3%/0.5 C	$2.7 - 4.4$	This work
Ti-NM90	88.9%/0.5 C	$2.7 - 4.4$	
$Al-NM90$	88.2%/0.5 C	$2.7 - 4.4$	
Sn-NCM90	$87.9\%/0.5 \text{ C}$	$2.7 - 4.4$	2
$Co-NM90$	$86.3\%/0.5$ C	$2.7 - 4.4$	
LWO-NCM90	$84.6\%/0.5 \text{ C} (80\%)$	$3.0 - 4.3$	3
Ga-NCA94	84.4%/0.5 C	$2.7 - 4.3$	4
Zr-NCA94	79.2%/0.5 C	$2.7 - 4.3$	4

Table S2† Comparison of cycling stability of NCM96-Mo@PANI cathode with the Nirich cathodes reported before.

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Table S3† Impedance fitting data of NCM9622, NCM96-Mo and NCM96-

Samples	Before 100 cycles			After 100 cycles	
	$R_e(\Omega)$	$R_{\text{ct}}(\Omega)$	$R_e(\Omega)$	$R_{\rm ct}(\Omega)$	
NCM9622	4.356	113.2	7.252	228.1	
NCM96-Mo	8.305	63.51	8.346	117.1	
NCM96-Mo@PANI	2.888	26.78	3.628	35.46	

Mo@PANI materials before and after 100 cycles at 1 C