

Supporting Information

Designing Strategies for High-Redox-Potential Conjugated Carbonyl Organic Cathodes in Lithium- and Sodium-ion Batteries

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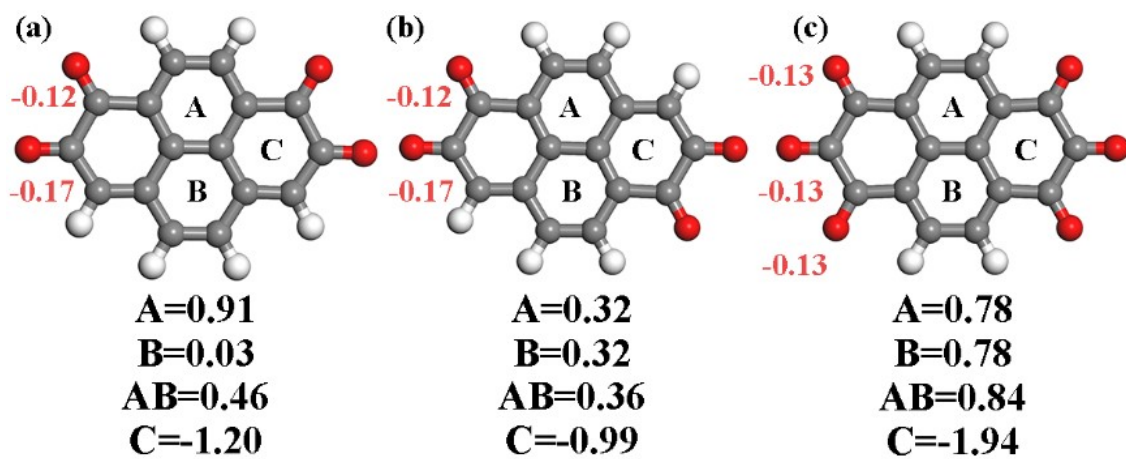


Figure S1. HOMA of various rings and BPI of various carbonyls in (a) AST, (b) BST and (c) PST, respectively. (The font color of HOMA was described in black, BPI was described in red.)

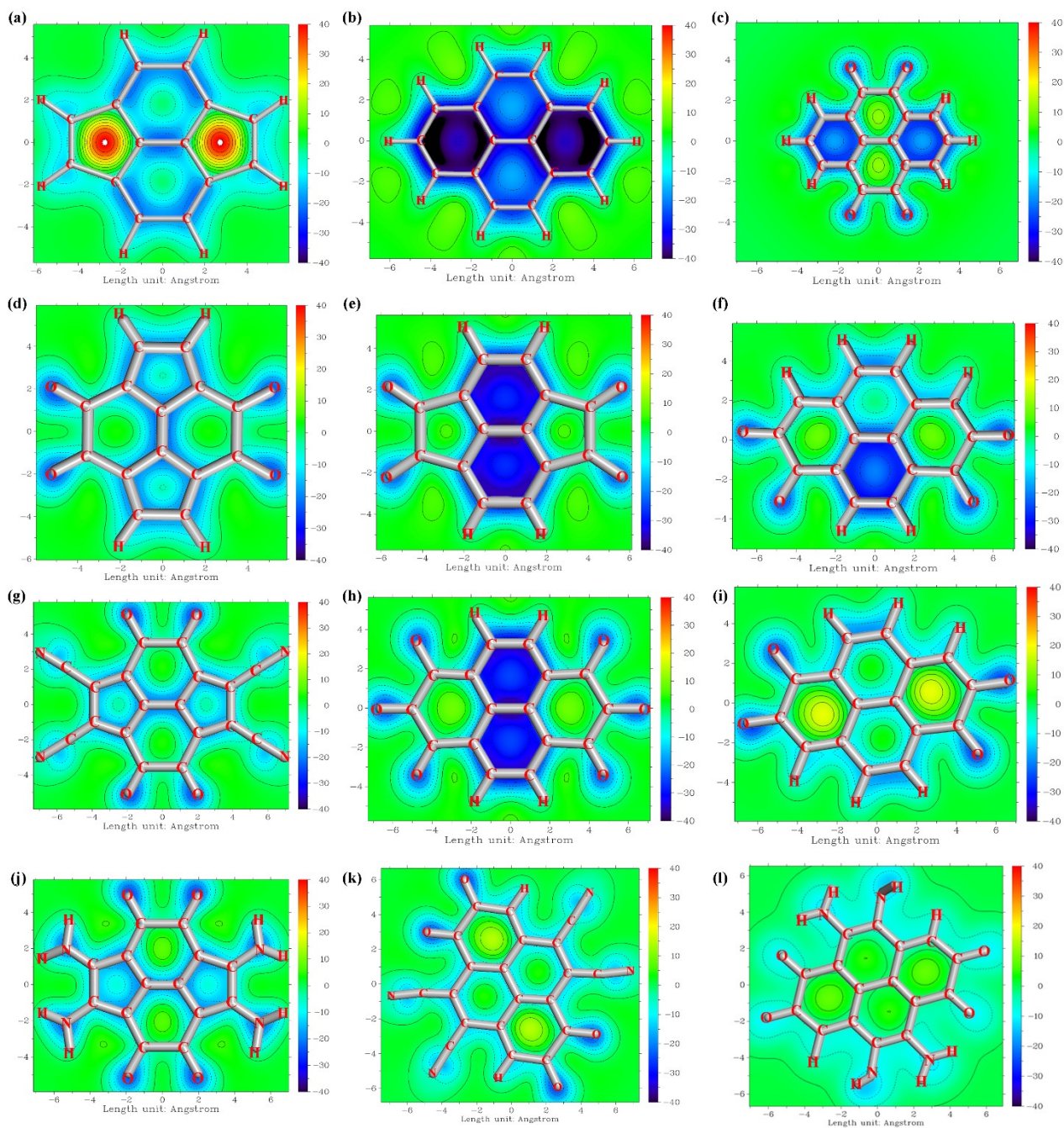


Figure S2. NICS-2D scan plane map (The ZZ component of magnetic shielding tensor above 1 Å of the plane of a ring center.) of (a) Pyracylene, (b) Pyrene, (c) PTO, (d) CAT (e) CAP, (f) AST, (g) CAT-CN, (h) PST, (i) BST, (j) CAT-NH₂, (k) BST-CN and (l) BST-NH, respectively. (The dark blue region indicates a high level of aromaticity, while the red region indicates a high level of anti-aromaticity.)

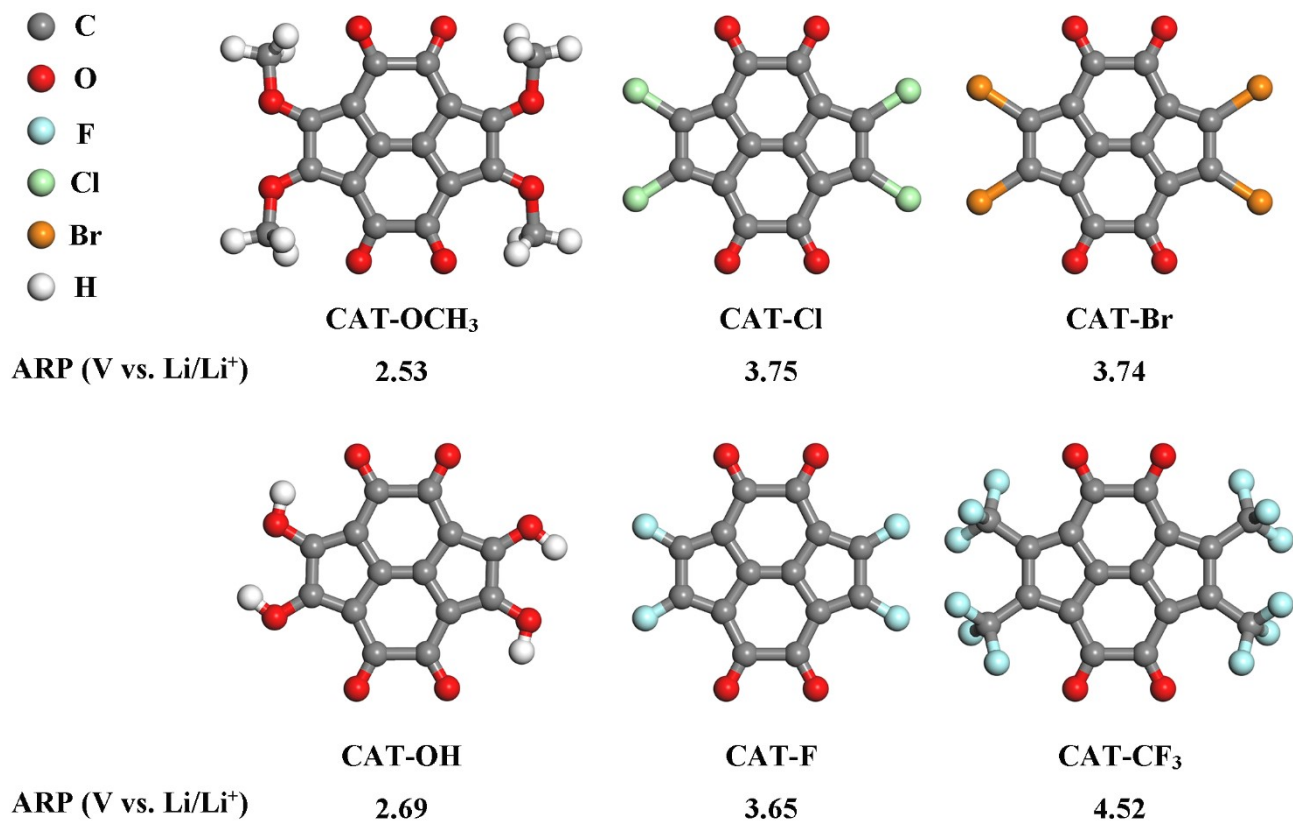


Figure S3. The structure and adiabatic redox potentials (ARP, V vs Li/Li⁺) of CAT-OCH₃, CAP-OH, CAT-Cl, CAT-F, CAT-Br and CAT-CF₃.

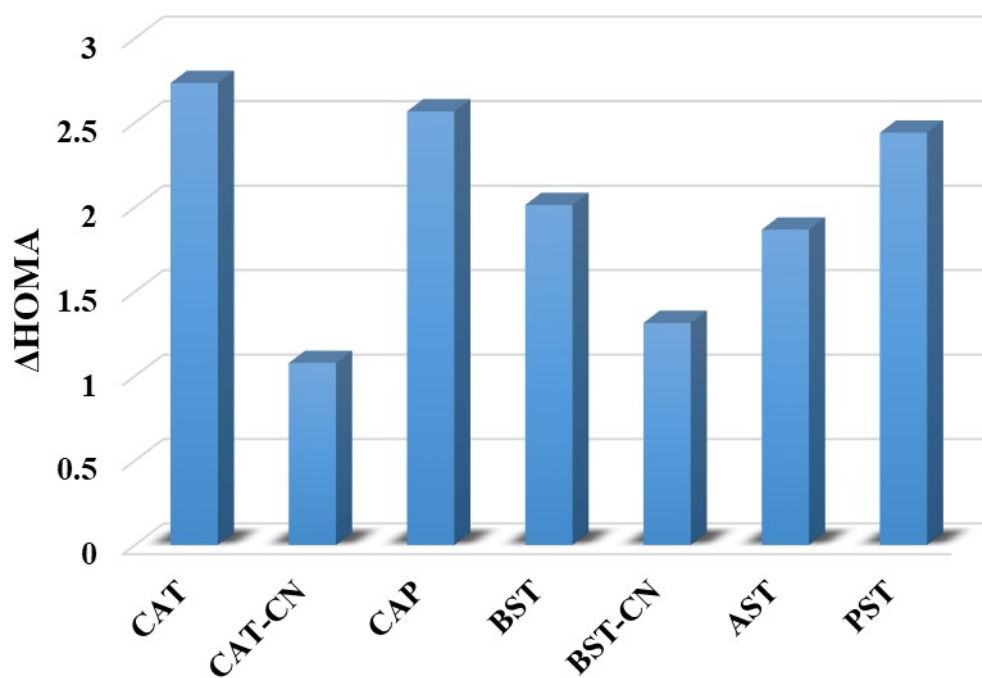


Figure S4. The difference in the Harmonic oscillator measure of aromaticity (Δ HOMA) between the fully charged states and the fully discharged states of CAT, CAT-CN, CAP, BST, BST-CN, AST and PST.

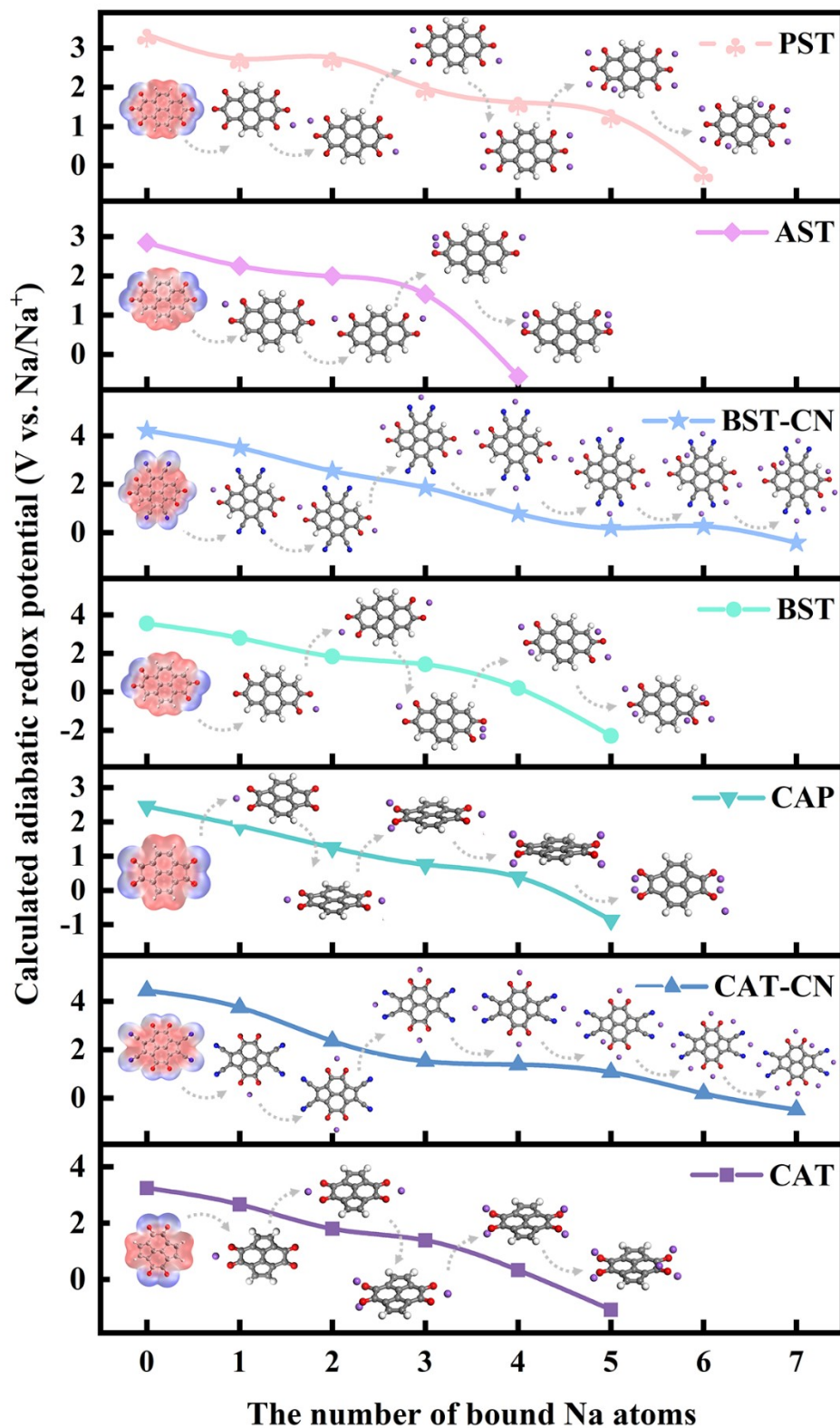


Figure S5. In the discharging process, the Na-ion storage sites and the structural evolution for seven pyrenetetrone derivatives (CAT, CAT-CN, CAP, BST, BST-CN, AST and PST) and their variety of ARP as a function of the number of bound Na atoms. (The atoms with grey, white, red, blue and purple depict carbon, hydrogen, oxygen, nitrogen and sodium, respectively.)

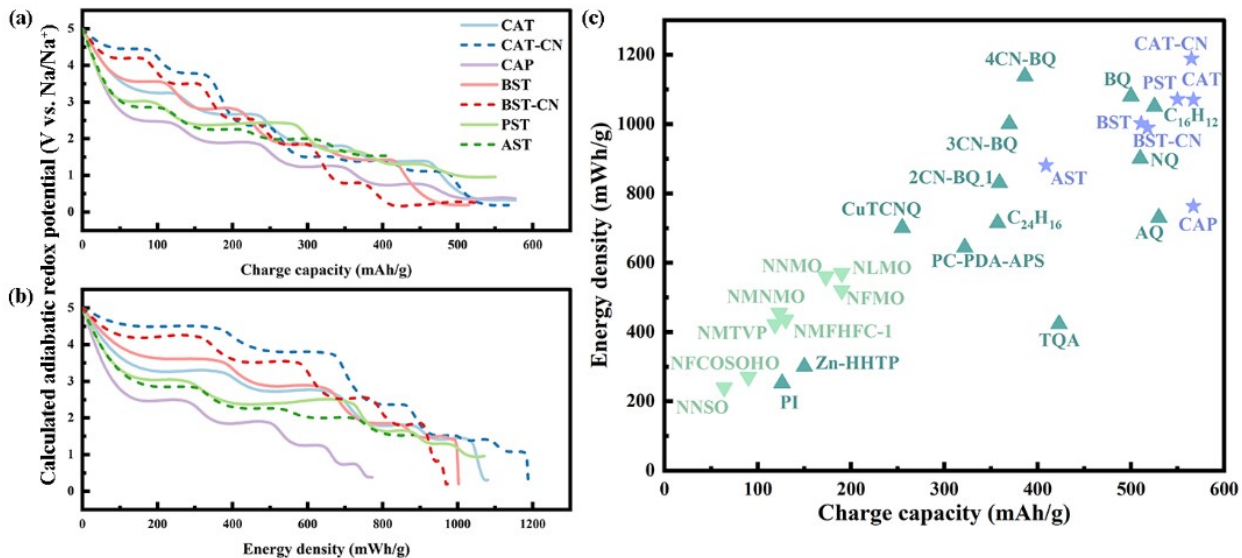


Figure S6. (a) Resultant profiles of the gravimetric charge capacities, (b) energy densities as a function of the calculated redox potential and (c) The Ragone plot shows the performance parameters of the pyrenetetrone derivatives reported by this work, the conventional inorganic compounds¹⁻⁸ and the outstanding organic compounds⁹⁻¹⁶ referred from previous studies.

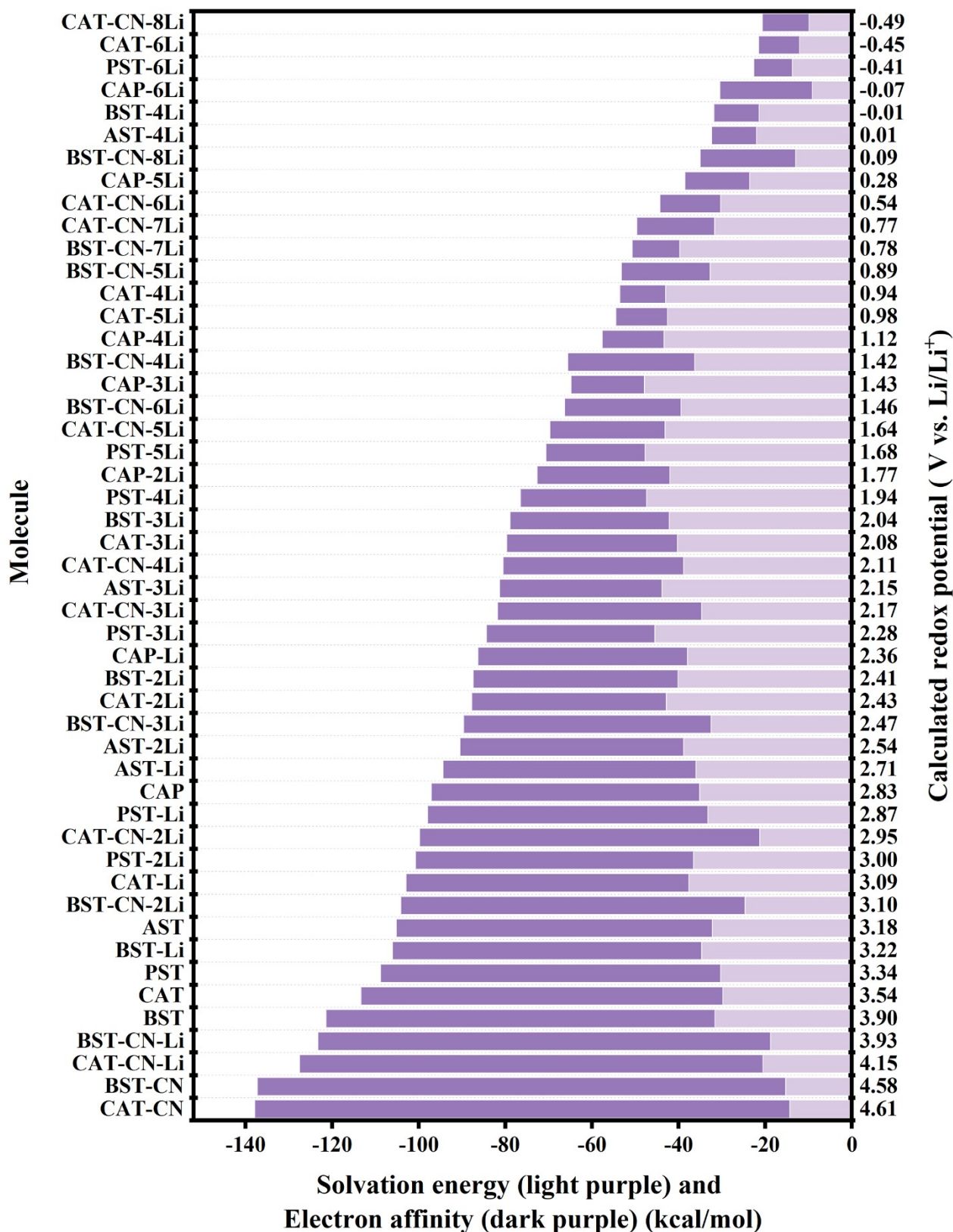


Figure S7. Contributions of solvation energy and electron affinity on redox properties. Solvation energy and electron affinity values for CAT, CAT-CN, CAP, PST, BST, BST-CN and AST with or without binding lithium atoms.

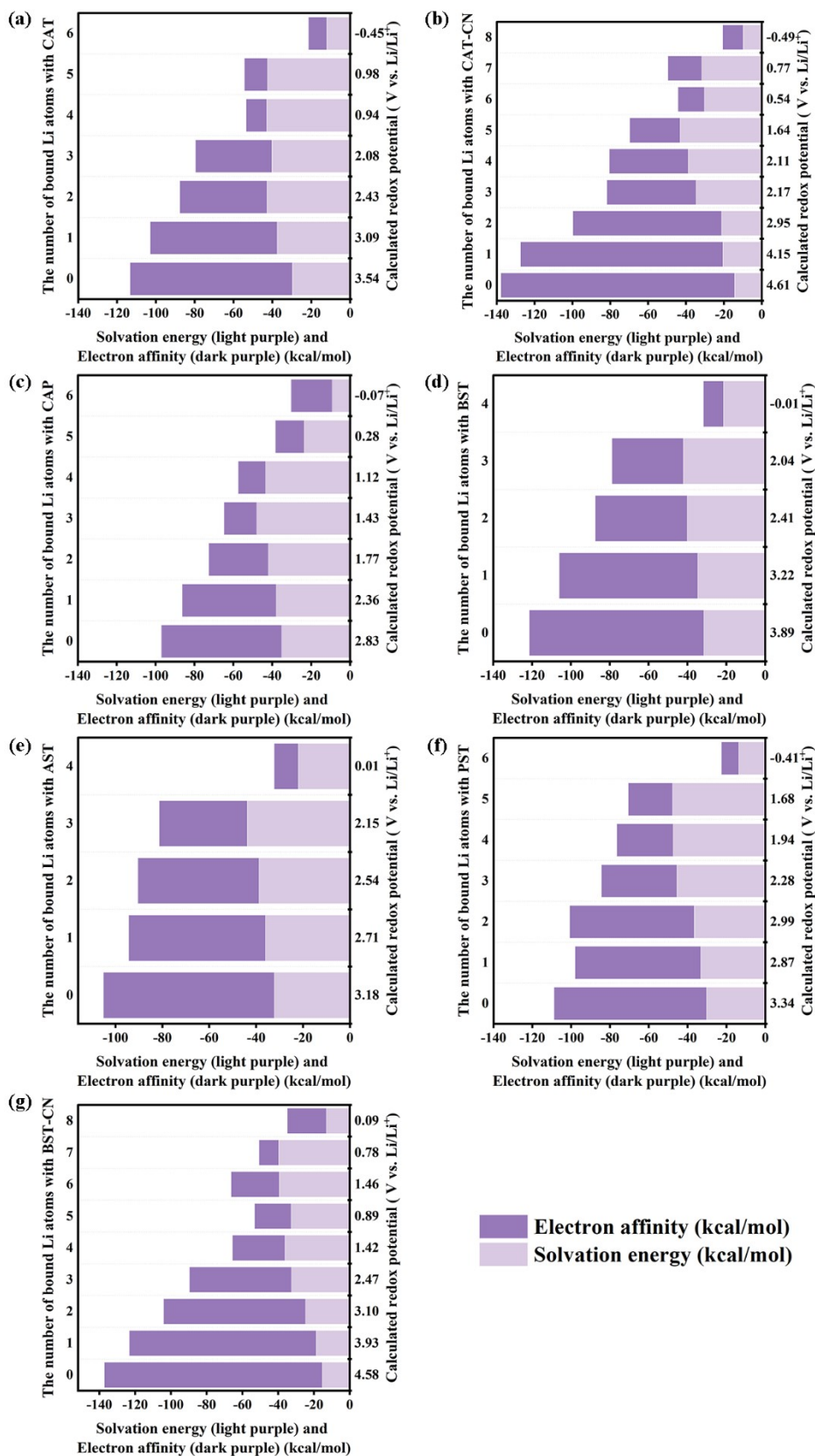


Figure S8. Contributions of solvation energy and electron affinity on redox properties. Solvation energy and electron affinity values for (a) CAT, (b) CAT-CN, (c) CAP, (d) BST, (e) AST, (f) PST and (g) BST-CN during the discharging process.

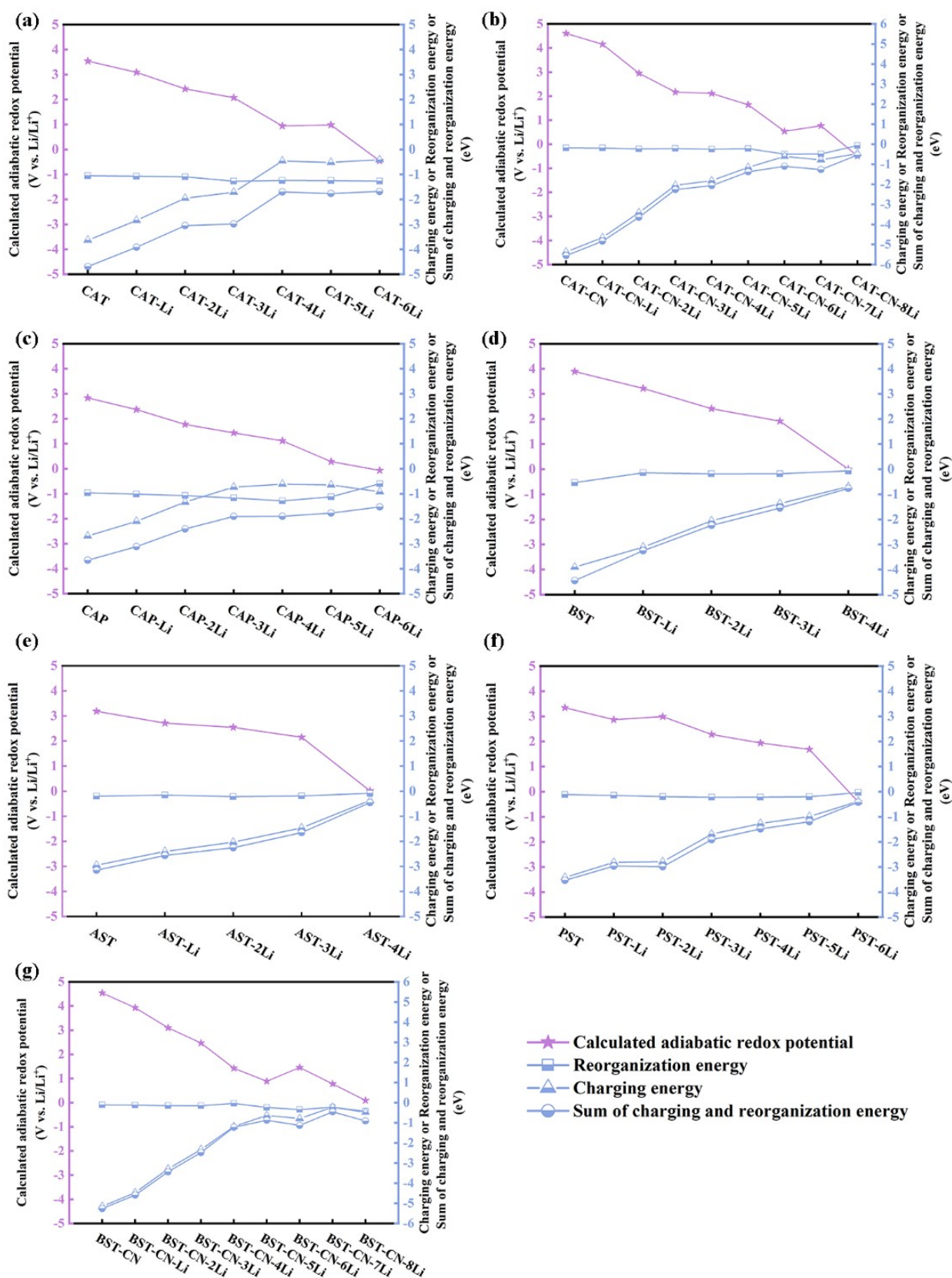


Figure S9. Contributions of charging and reorganization energies. Adiabatic redox potential, charging energy, reorganization energy, and the sum of the charging and reorganization energies for (a) CAT, (b) CAT-CN, (c) CAP, (d) BST, (e) AST, (f) PST and (g) BST-CN.

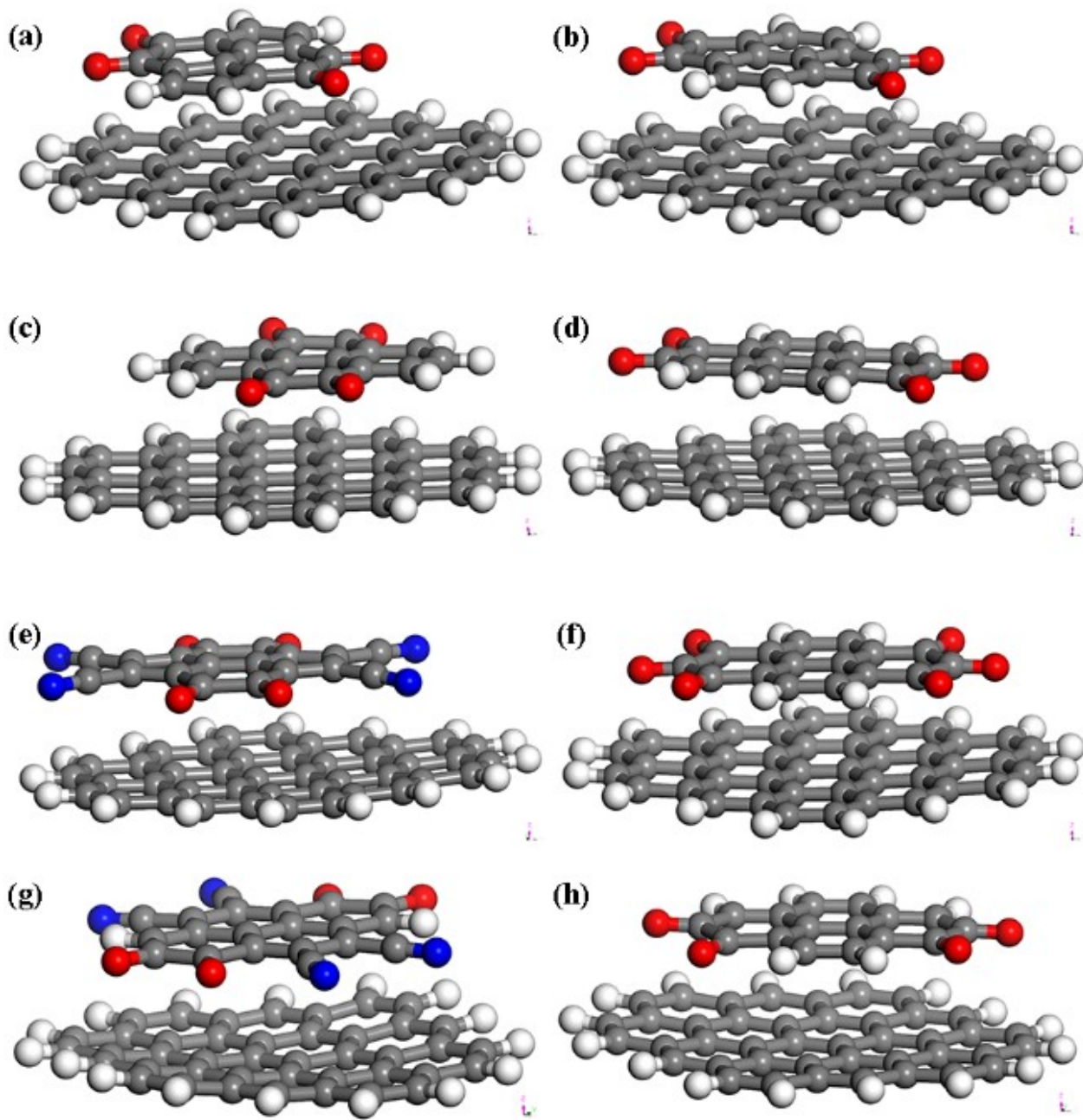


Figure S10. Cathode-graphene dimer: (a) CAT-graphene, (b) CAP-graphene, (c) PTO-graphene, (d) BST-graphene, (e) CAT-CN-graphene, (f) PST-graphene, (g) BST-CN-graphene and (h) AST-graphene.

Table S1. The comparison of the computed redox potentials for a variety of organic compounds with their experimental values. (Diff represents the difference between the computed redox potentials and their experimental values.)

Molecule	Theoretical redox potential (V vs. Li/Li ⁺)	Experimental redox potential (V vs. Li/Li ⁺)	Diff	Reference
1,4-benzoquinone	2.78	3.10	0.32	[17]
1,4-benzoquinone dimer	2.99	2.90	-0.09	[18]
1,4-naphthoquinone	2.52	2.60	0.08	[17]
9,10-anthraquinone	2.19	2.20	0.01	[17]
2-aminoanthraquinone	2.03	2.10	0.07	[17]
2,6-diaminoanthraquinone	1.84	2.00	0.16	[17]
Anthraquinone-2-carboxylic acid	2.23	2.30	0.07	[17]
1,4-bezenedicarbonitrile	1.45	1.70	0.25	[19]
Pyrenetetrone	2.62	2.59	0.03	[20]
Thiophene	-0.26	-0.10	0.16	[20]
Furan	-1.09	-1.10	-0.01	[20]
Benzopyrrole	-0.25	-0.20	0.05	[20]
Anthraquinone-1,5-disulfonic acid sodium	2.30	2.40	0.1	[21]
5,7,12,14-Pentacenetetrone	2.65	2.60	-0.05	[22]
Thianthrene-5,7,12,14-tetrone	2.82	2.90	0.08	[22]
Benzo [1,2-b:4,5-b']dithiophene-4,8-dione	2.52	2.50	-0.02	[23]
Benzofuro [5,6-b]furan-4,8-dione	2.55	2.60	0.05	[23]
Phenazine	1.79	1.90	0.11	[24]
Chloranil	3.31	3.40	0.09	[25]
2,3-Dicyano-chloranil	3.86	3.70	-0.16	[25]
Fluoranil	3.45	3.60	0.15	[25]
(t-Bu) ₃ TOT	3.06	3.10	0.04	[26]
Br ₃ TOT	3.41	3.60	0.19	[26]

Molecule	Theoretical redox potential (V vs. Na/ Na ⁺)	Experimental redox potential (V vs. Na / Na ⁺)	Diff	Reference
9,10-anthraquinone	1.85	1.9	0.05	[27]
1,4-benzoquinone	2.44	2.8	0.36	[27]
1,4-naphthoquinone	2.18	2.3	0.12	[27]
2-aminoanthraquinone	1.69	1.8	0.11	[27]
2,6-diaminoanthraquinone	1.50	1.7	0.2	[27]
Anthraquinone-2-carboxylic acid	1.89	2.0	0.11	[27]
1,4-bezenedicarbonitrile	1.11	1.4	0.29	[19]
Thiophene	-0.60	-0.4	0.2	[28]
Furan	-1.43	-1.4	0.03	[28]
Benzopyrrole	-0.59	-0.5	0.09	[28]
Pyrenetetrone	2.11	2.2	0.09	[28]

Table S2. Atomic coordinates of AST, BST, PTO, CAP, CAT, PST, BST-CN, BST-NH₂, CAT-NH₂ and CAT-CN.

AST			BST			PTO					
X	Y	Z	X	Y	Z	X	Y	Z			
C1	0.702	0.204	0	O1	3.783	-1.767	0	O1	0	1.365	-3.574
C2	-0.702	0.204	0	O2	-3.783	1.767	0	O2	0	-1.365	-3.574
C3	1.441	1.472	0	O3	4.778	0.802	0	O3	0	1.365	3.574
C4	-1.392	-1.025	0	O4	-4.778	-0.802	0.001	O4	0	-1.365	3.574
C5	1.392	-1.025	0	C5	0.717	0.098	0	C5	0	0.736	0
C6	-1.441	1.472	0	C6	-0.717	-0.098	0	C6	0	-0.736	0
C7	0.676	2.696	0	C7	1.548	-1.004	0	C7	0	1.46	-1.207
C8	-0.695	-2.23	0	C8	-1.548	1.004	0	C8	0	-1.46	-1.207
C9	-0.676	2.696	0	C9	1.253	1.465	0	C9	0	1.46	1.207
C10	0.695	-2.23	0	C10	-1.253	-1.465	0	C10	0	-1.46	1.207
C11	2.804	1.505	0	C11	1.023	-2.337	0	C11	0	0.77	-2.518
C12	-2.881	-1.054	0	C12	-1.023	2.337	0	C12	0	-0.77	-2.518
C13	2.881	-1.054	0	C13	-0.315	-2.559	0	C13	0	0.77	2.518
C14	-2.804	1.505	0	C14	0.315	2.559	0	C14	0	-0.77	2.518
C15	3.621	0.299	0	C15	3.021	-0.825	0	C15	0	2.858	-1.199
C16	-3.621	0.299	0	C16	-3.021	0.825	0	C16	0	-2.858	-1.199
H17	1.225	3.634	0	C17	2.603	1.702	0	C17	0	2.858	1.199
H18	-1.263	-3.155	0	C18	-2.603	-1.702	0	C18	0	-2.858	1.199
H19	-1.225	3.634	0	C19	3.573	0.623	0	C19	0	3.561	0
H20	1.263	-3.155	0	C20	-3.573	-0.623	0	C20	0	-3.561	0
H21	-3.338	2.451	0	H21	1.736	-3.155	0	H21	0	3.37	-2.157
H22	3.338	2.451	0	H22	-1.736	3.155	0	H22	0	-3.37	-2.157
O23	3.509	-2.09	0	H23	-0.711	-3.57	0	H23	0	3.37	2.156
O24	-3.509	-2.09	0	H24	0.711	3.57	0	H24	0	-3.37	2.156
O25	4.838	0.323	0	H25	2.994	2.716	0	H25	0	4.647	0
O26	-4.838	0.323	0	H26	-2.994	-2.716	0	H26	0	-4.647	0
CAP			CAT			PST					
X	Y	Z	X	Y	Z	X	Y	Z			
O1	-3.843	1.472	0	O1	3.565	1.372	0	O1	-3.49	-2.373	-0.001
O2	3.843	1.472	0	O2	3.565	-1.372	0	O2	3.49	-2.373	-0.004
O3	-3.843	-1.472	0	O3	-3.565	1.372	0	O3	-3.49	2.373	-0.002
O4	3.843	-1.472	0	O4	-3.565	-1.372	0	O4	3.49	2.373	-0.003
C5	-0.69	0	0	C5	0	0.669	0	O5	-4.875	0	0.005
C6	0.69	0	0	C6	0	-0.669	0	O6	4.875	0	0.007
C7	-1.413	1.205	0	C7	1.187	1.476	0	C7	-0.716	0	0
C8	1.413	1.205	0	C8	-1.187	1.476	0	C8	0.716	0	0
C9	-1.413	-1.205	0	C9	1.187	-1.476	0	C9	-1.402	-1.251	0
C10	1.413	-1.205	0	C10	-1.187	-1.476	0	C10	1.402	-1.251	-0.001
C11	-2.847	0.798	0	C11	2.5	-0.799	0	C11	-1.402	1.251	-0.001

C12	2.847	0.798	0	C12	-2.5	-0.799	0	C12	1.402	1.251	0
C13	-2.847	-0.798	0	C13	-2.5	0.799	0	C13	-0.699	-2.442	-0.001
C14	2.847	-0.798	0	C14	2.5	0.799	0	C14	0.699	-2.442	-0.001
C15	-0.713	2.402	0	C15	0.735	2.774	0	C15	-0.699	2.442	-0.001
C16	0.713	2.402	0	C16	-0.735	2.774	0	C16	0.699	2.442	-0.001
C17	-0.713	-2.402	0	C17	0.735	-2.774	0	C17	-2.886	-1.324	0
C18	0.713	-2.402	0	C18	-0.735	-2.774	0	C18	2.886	-1.324	-0.001
H19	-1.245	3.349	0	H19	1.348	3.668	0	C19	-2.886	1.324	-0.001
H20	1.245	3.349	0	H20	-1.348	3.668	0	C20	2.886	1.324	0
H21	-1.245	-3.349	0	H21	1.348	-3.668	0	C21	-3.673	0	0.002
H22	1.245	-3.349	0	H22	-1.348	-3.668	0	C22	3.673	0	0.003
								H23	-1.26	-3.371	-0.001
								H24	1.26	-3.371	-0.001
								H25	-1.26	3.371	-0.001
								H26	1.26	3.371	-0.001
BST-CN				BST-NH₂				CAT-NH₂			
	X	Y	Z		X	Y	Z		X	Y	Z
O1	3.584	-2.173	0.001	O1	-4.026	1.2	0	O1	1.395	-3.544	-0.004
O2	-3.584	2.173	0.001	O2	4.026	-1.2	0	O2	-1.395	-3.544	-0.004
O3	2.006	-4.413	0.001	O3	-4.656	-1.424	0	O3	1.395	3.544	0.004
O4	-2.006	4.413	0.001	O4	4.656	1.424	0	O4	-1.395	3.544	0.004
C5	0.33	-0.645	-0.001	C5	-0.707	-0.189	0	C5	0.67	0	0
C6	-0.33	0.645	-0.001	C6	0.707	0.189	0	C6	-0.67	0	0
C7	1.712	-0.726	0	C7	-1.681	0.788	0	C7	1.455	-1.179	-0.024
C8	-1.712	0.726	-0.001	C8	1.681	-0.788	0	C8	1.455	1.179	0.024
C9	-0.495	-1.861	0	C9	-1.059	-1.624	0	C9	-1.455	-1.179	-0.024
C10	0.495	1.861	0	C10	1.059	1.624	0	C10	-1.455	1.18	0.024
C11	2.527	0.463	0	C11	-1.321	2.204	0	C11	-0.8	-2.476	-0.011
C12	-2.527	-0.463	0	C12	1.321	-2.204	0	C12	-0.8	2.476	0.011
C13	1.938	1.706	0	C13	0.017	2.583	0	C13	0.8	2.476	0.011
C14	-1.938	-1.706	0	C14	-0.017	-2.583	0	C14	0.8	-2.476	-0.011
C15	2.381	-2.056	0	C15	-3.117	0.389	0	C15	2.778	-0.743	0.002
C16	-2.381	2.056	-0.001	C16	3.117	-0.389	0	C16	2.778	0.742	-0.002
C17	0.052	-3.113	0	C17	-2.382	-2.028	0	C17	-2.778	-0.743	0.002
C18	-0.052	3.113	0	C18	2.382	2.028	0	C18	-2.778	0.743	-0.002
C19	1.491	-3.314	-0.001	C19	-3.471	-1.109	0	N19	-3.896	1.506	0.017
C20	-1.491	3.314	-0.001	C20	3.471	1.109	0	H20	-3.75	2.503	-0.069
H21	-0.571	-4.003	0	H21	-2.641	-3.081	0	H21	-4.75	1.149	-0.383
H22	0.571	4.003	0	H22	2.641	3.081	0	N22	-3.896	-1.506	-0.017
C23	2.74	2.885	0	N23	2.267	-3.152	0	H23	-4.75	-1.149	0.383
N24	3.351	3.873	0.001	H24	3.243	-2.893	0	H24	-3.749	-2.503	0.069
C25	3.958	0.413	0	N25	-0.263	-3.986	0	N25	3.896	1.506	0.017
N26	5.113	0.535	0.001	H26	-0.78	-4.286	-0.821	H26	4.75	1.149	-0.382

C27	-3.958	-0.413	0	N27	0.263	3.986	0	H27	3.749	2.503	-0.069
N28	-5.113	-0.535	0	H28	0.781	4.286	0.821	N28	3.896	-1.506	-0.017
C29	-2.74	-2.885	0	N29	-2.267	3.152	0	H29	3.749	-2.504	0.069
N30	-3.351	-3.873	0	H30	1.932	-4.107	0	H30	4.75	-1.149	0.383
				H31	-0.781	-4.285	0.821				
				H32	-3.243	2.893	0				
				H33	0.781	4.286	-0.821				
				H34	-1.932	4.107	0				
CAT-CN											
	X	Y	Z								
O1	-1.377	-3.564	0								
O2	1.377	-3.564	0								
O3	-1.377	3.564	0								
O4	1.377	3.564	0								
C5	-0.673	0	0								
C6	0.673	0	0								
C7	-1.47	-1.188	0								
C8	-1.47	1.188	0								
C9	1.47	-1.188	0								
C10	1.47	1.188	0								
C11	0.796	-2.508	0								
C12	0.796	2.508	-0.001								
C13	-0.796	2.508	-0.001								
C14	-0.796	-2.508	0								
C15	-2.775	-0.748	0								
C16	-2.775	0.748	0								
C17	2.775	-0.748	0								
C18	2.775	0.748	0								
C19	3.957	1.524	0								
C20	3.957	-1.524	0								
C21	-3.957	1.524	0								
C22	-3.957	-1.524	0								
N23	4.948	2.131	0								
N24	4.948	-2.131	0								
N25	-4.948	2.131	0								
N26	-4.948	-2.131	0								

Reference

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