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Supporting Information for

# **Synergistic Surface Modulation with Isotropic 2D GA2PbI<sup>4</sup> and Lewis Base Enhances Efficiency and Stability of Perovskite Solar Cells**

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### **Materials**

Methylammonium iodide (MAI), methylammonium bromide (MABr), and formamidinium iodide (FAI) were bought from Greatcell Solar. Lead iodide (PbI $_2$ , 99.99%) and guanidinium iodide (GAI, >97%) were purchased from TCI. Methylamine hydrochloride (MACl) was obtained from Xi'an Polymer Light Technology Corporation. 2,2',7,7'-tetrakis[N,N-di(4 methoxyphenyl)amino]-9,9-spirobifluorene (Spiro-OMeTAD, 99%) were bought from Shenzhen Feiming Technology Corporation. [(2,6-(4,8-bis(5-(2-ethylhexyl)thiophene-2-yl) benzo[1,2-b:4,5-b′]dithiophene))-alt-(5,5-(1′,3′-di-2-thienyl-5′,7′-bis(2-ethylhexyl)benzo[1′,2′ c:4′,5′-c′]dithiophene-4,8-dione)] (PBDB-T, 99%) was purchased from Solarmer. Hydrochloric acid (HCl) was purchased from Shanghai Hushi Corporation. Guanidinium acetate (GAAc, 99%), formamidinium acetate (FAAc, 99%), urea, thioglycolic acid (TGA, 98%),  $SnCl<sub>2</sub>·2H<sub>2</sub>O$ (>99.995%), N, N-dimethylformamide (DMF, anhydrous, 99.8%), dimethyl sulfoxide (DMSO, anhydrous, 99.9%), isopropanol (IPA, anhydrous, 99%), chlorobenzene (CB, 99.8%), acetonitrile (anhydrous, 99.8%), bis(trifluoromethane) sulfonimide lithium salt (99.95%) and 4-tert-butylpyridine (TBP, 96%) were obtained from Sigma-Aldrich.

#### **Device Fabrication**:

*Electron transport layer deposition*: First, FTO glass substrates were cleaned by sequential ultrasonic treatment in detergent, deionized water, acetone, isopropyl alcohol, and ethanol. The cleaned substrates were then treated for 15 minutes with  $O<sub>2</sub>$  plasma. The chemical bath deposition (CBD) solution was prepared by mixing 1250 mg of urea, 1250 μL of HCl, 25 μL of TGA, and 275 mg of  $SnCl<sub>2</sub>·2H<sub>2</sub>O$  into 100 mL of deionized water. The FTO substrates and the CBD solution were loaded onto a glass reaction vessel and reacted at 97 °C for 5 hours. Then, the FTO substrates were taken out from the CBD solution, rinsed with deionized water, and subjected to ultrasonication in IPA for 5 minutes. Finally, it was annealed on a hotplate at 170°C for 1 hour.

*Perovskite absorber deposition*: On the CBD SnO<sub>2</sub> layer, 1.5 M PbI<sub>2</sub> with 5% CsI (dissolved in DMF/DMSO=9/1,  $v/v$ ) was spin-coated at 2,500 rpm for 30 s and annealed at 70 °C for 1 minute in an N<sub>2</sub>-filled glove box. Then, a solution of organic salts (120 mg of FAI, 20 mg of MAI, 3 mg of MABr, and 5 mg of MACl dissolved in 2 mL of IPA) was spin-coated at 1,500 rpm for 30 s and then annealed at 150 °C for 15 minutes in the ambient air (30% relative humidity). After cooling down to room temperature, GAAc with a concentration of 2 mg mL $^{-1}$  in IPA was spin-coated at 4,000 rpm for 30 s. Subsequently, the sample was heated on a hotplate at 100°C for 10 minutes.

*Hole transport layer (HTL) and metal electrode deposition*: The HTL solution consists of 72.3 mg of Spiro-OMeTAD, 17.5 μL of lithium bis(trifluoromethane) sulfonamide salt (520 mg dmL-1 in acetonitrile), and 28.8 μL of 4-tertbutylpyridine in 1 mL of chlorobenzene. The prepared HTL solution was spin-coated on the perovskite films at 4,000 rpm for 30 s. For the thermal stability test, the Spiro-OMeTAD solution was mixed with PBDB-T solution (5mg mL<sup>-1</sup> dissolved in chlorobenzene) at a volume ratio of 4: 1. Finally, A 80 nm Ag was deposited via thermal evaporation to complete the solar cell device.

#### **Characterization**

Photovoltaic characteristics were measured using a Keithley 2400 source meter under simulated AM1.5G irradiation under an Xe lamp simulator (Enli SS-F5). The light intensity was calibrated to 1 sun illumination using a NERL-calibrated silicon photodiode with a KG5 filter. The *J-V* curves of all devices were tested at a scan rate of 40 mV s<sup>-1</sup> with an aperture mask (0.12 cm<sup>2</sup>). The reverse scan was from  $V_{\text{OC}}$  to  $J_{\text{SC}}$  (1.25 V to 0 V), and the forward scan was from  $J_{\text{SC}}$  to  $V_{\text{OC}}$  (0 V to 1.25 V).<sup>13</sup>C NMR spectra were measured using a Bruker Avance III 500 MHz NMR spectrometer. UV-Vis absorption, photoluminescence, and electroluminescent spectra were collected on OceanInsight Spectrometer. Time-resolved photoluminescence and temperaturedependent photoluminescence measurements were carried out using the FLS1000 (Edinburgh Instrument, UK). X-ray diffraction (XRD) patterns were taken with Cu  $K_a$  radiation on a Rigku-SmartLab X-ray diffraction instrument. Grazing-incidence wide-angle X-ray scattering (GIWAXS) patterns were acquired at the XEUSS WAXS/SAXS system (Xenocs, France). Xray photoelectron spectroscopy data were obtained from the Thermo Scientific K-Alpha+. An Al  $K_a$  (1486.7 eV) source gun with 400  $\mu$ m X-ray spot was used. The pass energy was 30 eV and the energy step was 0.05 eV. Ultraviolet photoelectron spectroscopy (UPS) measurements were carried out on a Thermo-Fisher ESCALab  $Xi$ + system at a bias of -5 V using a He-I<sub>α</sub> (21.22 eV) UV light source. PL mapping was performed on Leica TCS SP8 Confocal laser scanning microscopy. The excitation wavelength was 488 nm, the detection wavelength range was 790-800 nm, and the fluorescence signal was probed by a PMT detector with a gain value constant at 400. Scanning electron microscope (SEM) images were obtained from a JEOL-JSM-7610F-plus field-emission scanning electron microscopy. The transient photovoltage (TPV) measurement transient time is 1 μs,potentiostat setting time is 0.2 μs,and the mode of light intensity course is on-off. TPV, Mott-Schottky polt is obtained by Capacitance-voltage test from 0→1.2 V under 10 KHz, and admittance spectroscopy analysis acquired by Capacitancefrequency from 1MHz→1Hz. TPV, Capacitance-voltage, Capacitance-frequency

measurements were carried out on a Zahner electrochemical workstation. The admittance spectroscopy analysis The conductivity of the perovskite films at different temperatures was measured by CHI-660e electrochemical workstation. Operational stability data was tracked on a solar cell stability test system by Suchow DeRui Corporation.

### **Supplementary Note 1**: Calculation of the trap density  $(N_t)$ .

The *N*<sub>t</sub> can be calculated using Capacitance-frequency and Capacitance-voltage measured in

the dark state, and estimated by a formula:  $K_B T eW a\omega$ . Where  $\omega$  is the angular  $N_t(E_\omega) = -\frac{\omega}{K T}$  $\frac{\omega}{K_B T} \times \frac{V_{bi}}{eW}$  $\frac{V_{bi}}{eW} \times \frac{dC}{d\omega}$  $d\omega$ frequency of the ac signal,  $K_B$  is Boltzmann constant, *T* is temperature,  $K_B T = 0.026$  eV in room temperature;  $V_{bi}$  is the built-in voltage in the heterojunction, W is the width of the depletion

$$
A^2 = \frac{2(V_{b_i} - V)}{}
$$

layer; C is capacitance. A Mott-Schottky plot:  $\overline{c^2} = \frac{1}{q \varepsilon \varepsilon_0 N}$  describes a straight line where the intersection on the bias axis determines *V*bi and the slope gives the impurity doping density *N*.

The depletion layer width W can be calculated by  $\sqrt{qN}$ . The energy  $E\omega$  can be estimated  $W = \frac{2\varepsilon\varepsilon_0 V_{bi}}{2\varepsilon}$  $qN$ 

by the formula  $E_{\omega} = K_B T \times In \frac{2\beta_P N_v}{\omega}$ . Where  $\beta_P$  is capture coefficient of hole,  $N_v$  is the effective  $\omega$ density of states in the valence band. The  $\beta_p$  is  $10^{-8}$  cm<sup>3</sup>/s, and the  $N_v$  is 2.524×10<sup>19</sup>/cm<sup>3</sup> for the perovskite film based FAPbI<sub>3</sub>.<sup>[1](#page-15-0), [2](#page-15-1)</sup>

#### **DFT calculations**

To calculate the adsorption energy, the periodic structures were calculated in the Vienna ab initio simulation package (VASP) using the Perdew-Burke-Ernzerhof function with the generalized gradient approximation.[3](#page-15-2) The DFT-D3 method was employed to correct the Van der Waals interactions of the system. The spin polarization calculation was carried out in the geometric optimization. The plane-wave basis cutoff was set as 450 eV, the energy criterion was set as  $10^{-5}$  eV, and the force criterion was set as  $0.02$  eV/Å. The adsorption structure was studied by using a three-layer  $(2 \times 2)$  perovskite supercell and a 15 Å vacuum layer. The bottom layer of perovskite was fixed. A  $3 \times 3 \times 1$  Monkhorst-Pack k-point sampling grid was used to optimize the geometrical structure.

The binding energies were calculated by the following equation:

*E*=*E*tot−*E*substrate−*E*mole,

where  $E_{\text{tot}}$ ,  $E_{\text{substrate}}$ , and  $E_{\text{mole}}$  are the total electronic energies of the total system, the perovskite substrate, and the ion, respectively. The structure of them was based on the optimized structure of the system.



Fig. S1 XRD patterns of the GA<sub>2</sub>PbI<sub>4</sub>, GAFAPbI<sub>4</sub>, and GAPbI<sub>3</sub> perovskite films.



**Fig.** S2 XRD patterns of the 2D GA<sub>2</sub>PbI<sub>[4](#page-15-3)</sub> films and the single crystal ICSD card.<sup>4</sup>



**Fig. S3** (a) GIWAXS profile of the control and GAAc-treated perovskite films. (b) The azimuthal integration profile of the  $(021)$  diffraction of  $GA_2PbI_4$  from the GIWAXS pattern.



Fig. S4 UV-vis absorption spectrum of spin-coated (a) Control and (b)  $GA_2PbI_4$  perovskite film on the glass substrate.



**Fig. S5** Molecular structure of (a) GAI and (b) FAAc.



**Fig. S6** XRD patterns of the control, GAI, FAAc, and GAAc-treated perovskite films.



**Fig. S7** GIWAXS patterns of (a) GAI and (b) FAAc-teated perovskite films.



**Fig. S8** Surface recombination velocity of the control, GAI, FAAc, and GAAc treated perovskite films.



**Fig. S9** PL-Mapping images of the (a) control, (b) GAI, (c) FAAc, and (d) GAAc treated perovskite films.



**Fig. S10** SCLC measurements of the electron-only device with the device architecture of ITO/SnO2/perovskite/PCBM/Ag based on the (a) control, (b) GAI, (c) FAAc, and (d) GAAc treated perovksites.



**Fig. S11** SCLC measurements of the hole-only devices with the architecture of ITO/PEDOT: PSS/perovskite/PTAA/Ag based on the (a) control, (b) GAI, (c) FAAc, and (d) GAAc treated perovksites.



**Fig. S12** Energy band diagram of solar cell devices based on the control, GAI, FAAc, and GAAc treated perovksites.



**Fig. S13** The FWHMs of the (a) control and (b) GAAc treated perovskites as a function of temperature.



Fig. S14 EQE and integrated  $J_{SC}$  of the control, GAI, FAAc, and GAAc treated PSCs.



**Fig. S15** Statistic diagram box of the photovoltaic parameters of the control, GAI, FAAc, and GAAc treated PSCs. (a)  $V_{OC}$ , (b)  $J_{SC}$ , (c) FF, and (d) PCE.



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## 福建省计量科学研究院<br>FUJIAN METROLOGY INSTITUTE (国家光伏产业计量测试中心) National PV Industry Measurement and Testing Center

S.

报告编号: 22Q3-00044 Report No

检测结果/说明: Results of Test and additional explanation

Standard Test Condition (STC): Total Irradiance: 1000 W/m<sup>2</sup>  $\mathbf{1}$ Temperature: 25.0 °C Spectral Distribution: AM1.5G

Measurement Data and I-V/P-V Curves under STC  $\overline{c}$ 

Forward Scan



#### Reverse Scan



Mismatch factor: 1.004



**Fig. S16** Certified efficiencies by FUJIAN METROLOGY INSTITUTE, National PV Industry Measurement and Testing Center.



**Fig. S17** EL spectrum of the GAAc treated PSC device work as an LED.

Sample	$\tau_1$ (ns)	$\tau_2$ (ns)	A <sub>1</sub>	A <sub>2</sub>	$\tau_{\text{ave}}$ (ns)
Control	228.90	972.19	583.27	307.43	742.69
GAI	253.94	1327.69	399.05	468.77	1177.34
FAAc	268.87	1600.73	444.03	469.39	1418.13
GAAc	285.27	2192.80	370.77	545.21	2037.76

**Table S1.** Carrier lifetime of the control, GAI, FAAc and GAAc treated perovskite films.

**Table S2.** Integrated *JSC* of the control, GAI, FAAc and GAAc treated PSCs.

Sample	Control	GAI	FAAc	GAAc
Integrated $J_{SC}$ (mA cm <sup>-2</sup> )	24.76	24.88	24.78	24.91





## **Reference**

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