

# Supporting Information

## Single-crystalline Sb<sub>2</sub>S<sub>3</sub> Tubes for High-Performance Broadband Visible Photodetection

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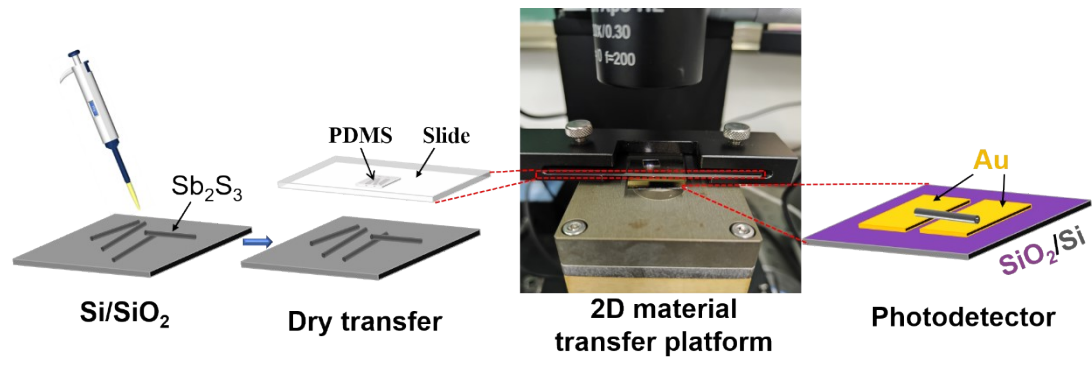


Figure S1. Schematic illustration of the fabrication process of the Sb<sub>2</sub>S<sub>3</sub> microtube-based photodetector device. PDMS: polydimethylsiloxane

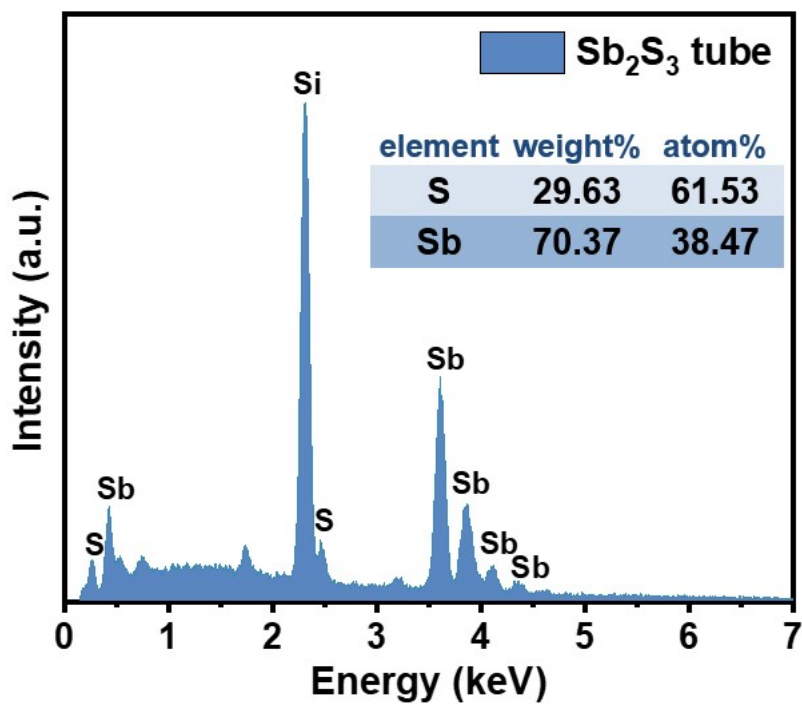


Figure S2. Energy-dispersive X-ray spectroscopy spot images of  $\text{Sb}_2\text{S}_3$ , the inset shows the weight and atomic information for each element.

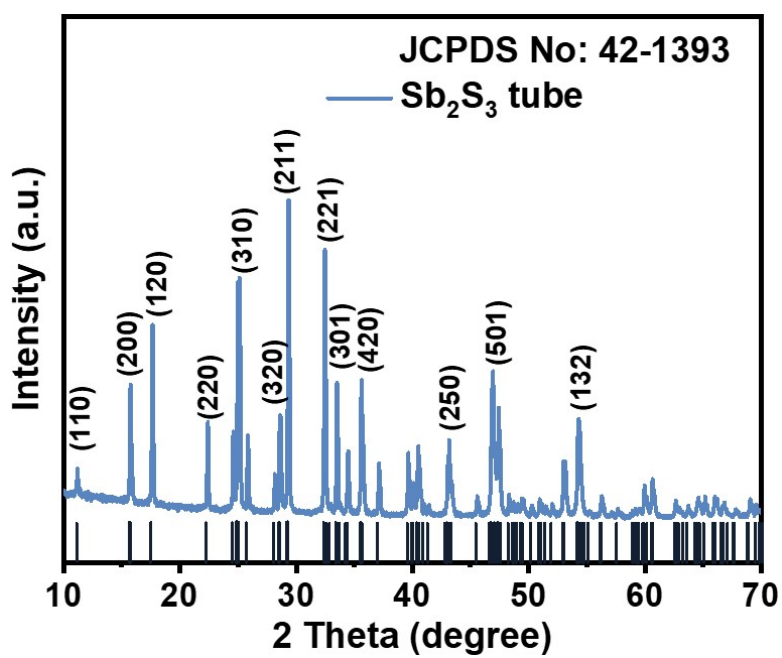


Figure S3. X-ray diffraction pattern of the as-synthesized  $\text{Sb}_2\text{S}_3$  microtubes.

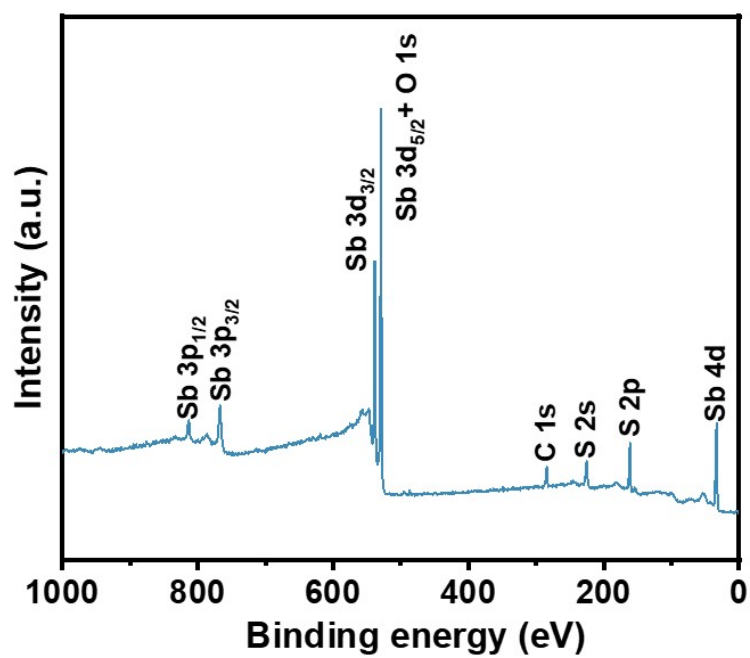


Figure S4. Survey X-ray photoelectron spectroscopy spectrum of the as-received  $\text{Sb}_2\text{S}_3$  microtubes.

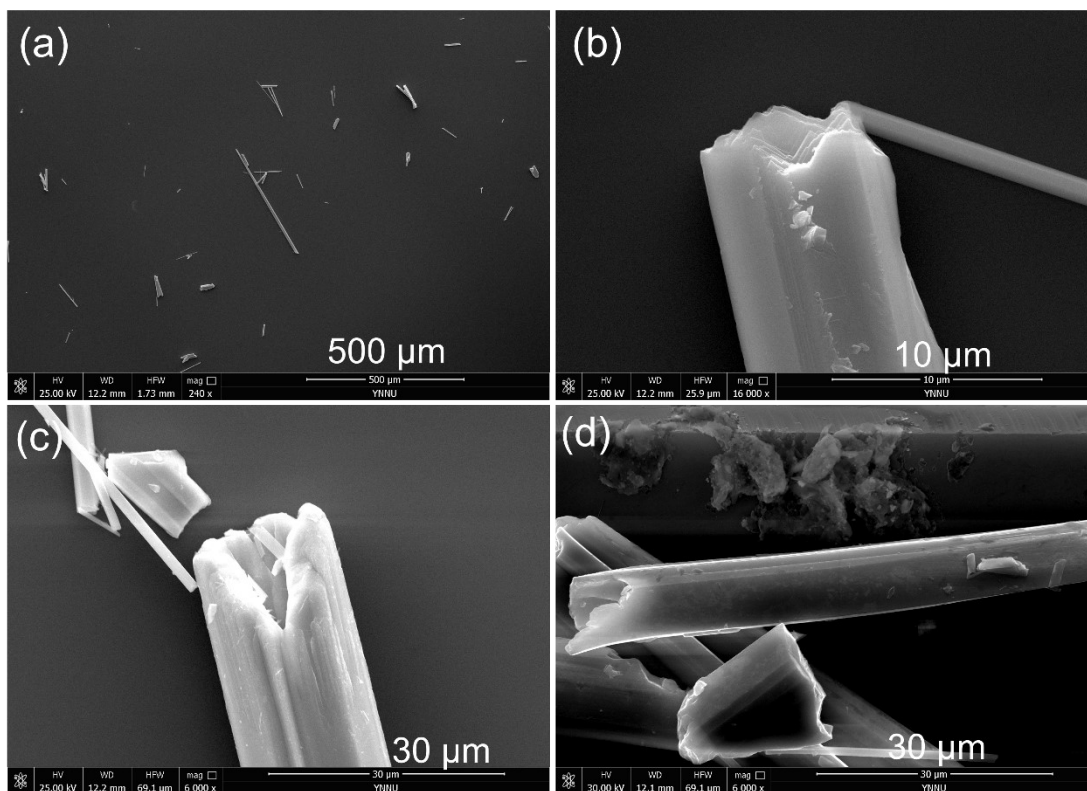


Figure S5. SEM images of the  $\text{Sb}_2\text{S}_3$  products synthesized after hydrothermal reaction for 24 h with the addition of different amounts of EDTA: (a, b) 1 mmol, (c) 3 mmol, and (d) 4 mmol.

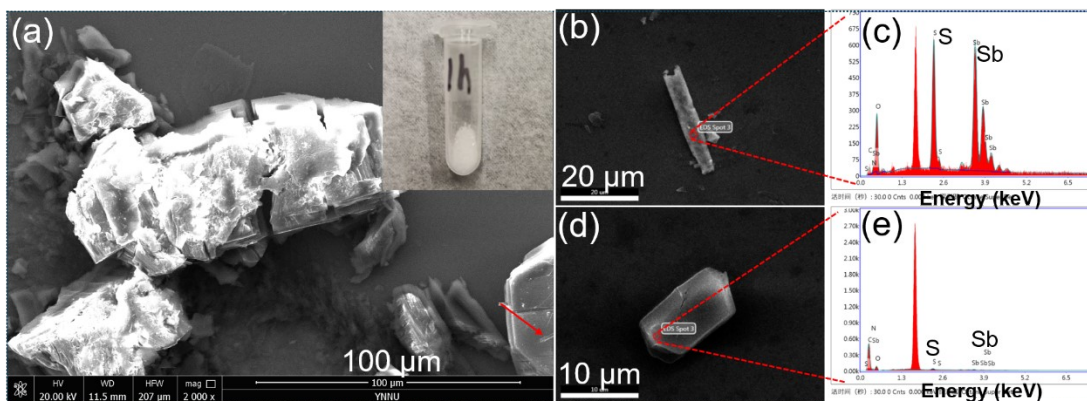


Figure S6. SEM images (a, b, and d) and EDX spectra (c, e) of the product synthesized by hydrothermal reaction for 1 h.

SEM image (Figure S6a) reveals the product obtained after hydrothermal reaction for 1 h mainly consists of some irregular aggregates with size larger than 100  $\mu\text{m}$ . Only a very small amount of microrods was produced, as indicated by the red arrow in Figure R1a. An additional SEM image of one microrod (Figure S6b) and the corresponding EDX spectrum (Figure S6c) unravel it mainly consists of Sb and S elements with an atomic ratio of approximately 0.93. In contrast, the SEM image (Figure S6d) and the associated EDX spectrum (Figure S6e) reveal that the irregular large microparticles are almost free of Sb/S and are primarily composed of an unknown intermediate product consisting of some C, N, and O elements. This result is consistent with the very bright image in Figure S6a (mainly caused by poor conductivity of the sample surface leading to discharge during SEM testing). Meantime, the digital photograph (inset, Figure S6a) also shows that the amount of product collected is minimal and appears white. This observation discloses that the reaction for the synthesis of  $\text{Sb}_2\text{S}_3$  in 1 h is incomplete.

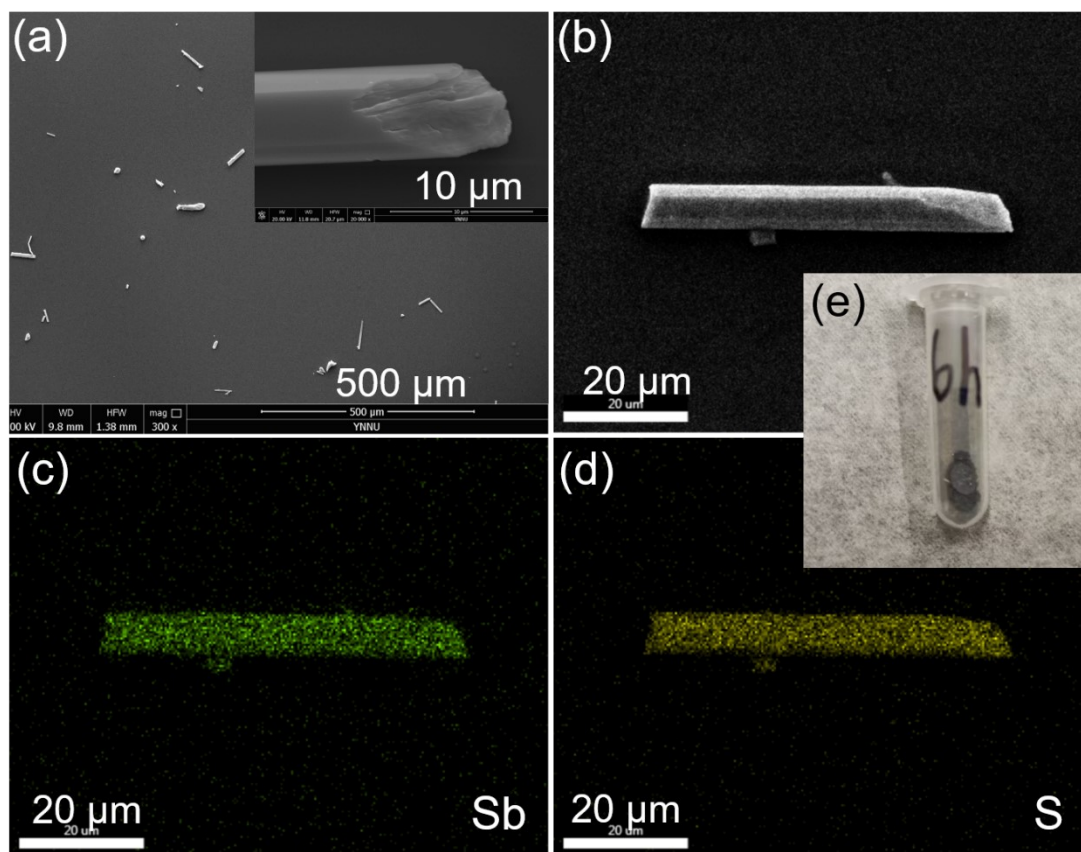


Figure S7. SEM images (a, b) and EDX mapping images (c, d), and (e) an optical photo of the product synthesized by hydrothermal reaction for 6 h.

From Figure S7a that in the product synthesized by hydrothermal reaction for 6 h, the amount of microrods has significantly increased, in addition to the large particles with irregular shapes. The SEM image in Figure S7b, along with the corresponding EDX mapping images in Figures S7c and S7d, shows a single microrod composed of Sb and S. The elements are uniformly distributed with an atomic ratio of approximately 0.59, confirming the microrod to be  $\text{Sb}_2\text{S}_3$ . In addition, the optical photo of the sample also appears as dark grey, consistent with the color of  $\text{Sb}_2\text{S}_3$ .



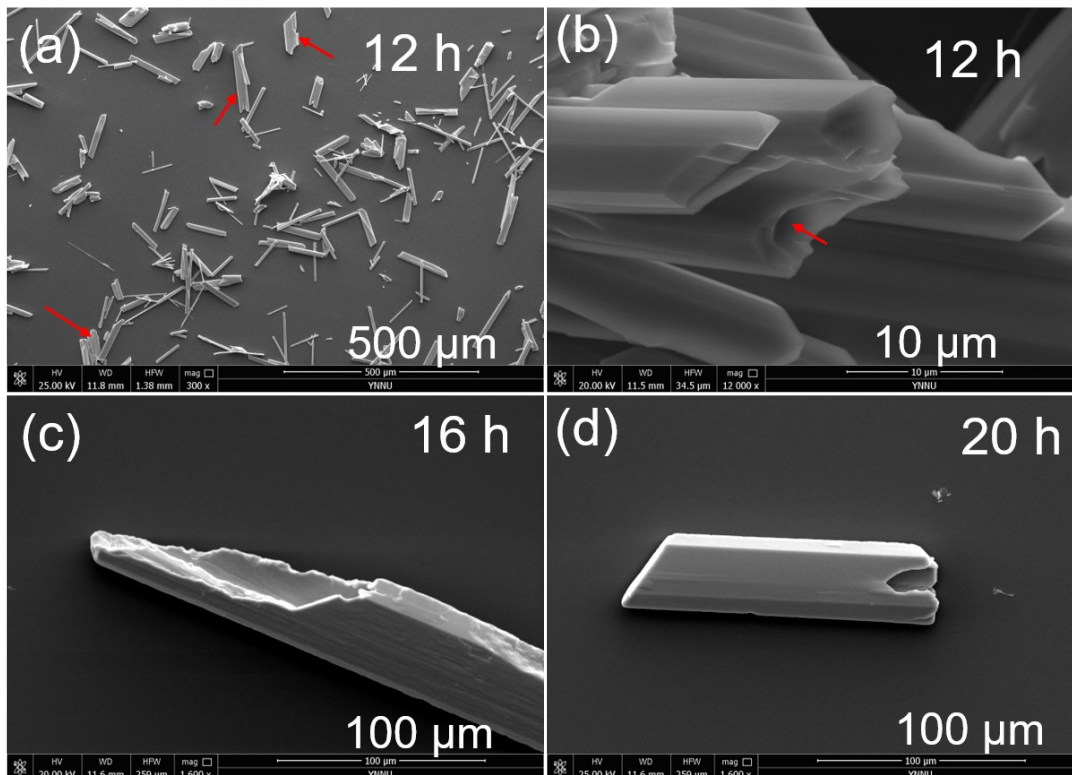


Figure S8. SEM images of the  $\text{Sb}_2\text{S}_3$  products synthesized by hydrothermal reaction for different durations: (a, b) 12 h, (c) 16 h, and (d) 20 h.

SEM image (Figure S8a) reveals that extending the reaction time to 12 h, the resulting sample contains many microrods; simultaneously, some flattened microbelts are formed (as indicated by the red arrows), which may be caused by oriented attachment between the microrods. From the high-magnification SEM image in Figure S8b, these microrods do not yet display a hollow structure, but they do show a tendency to start hollowing out (as indicated by the red arrow). However, for samples reacted for 16 h and 20 h, there is a clear formation of microtubes, as shown in Figures S8c and S8d.

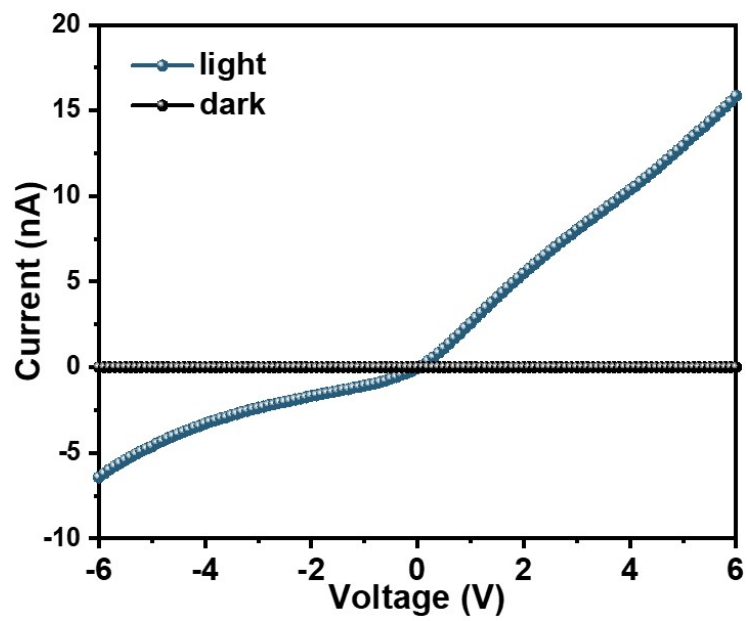


Figure S9.  $I$ - $V$  plot of the device in natural light with a bias voltage of 6 V

Table S1. Adsorption energy and surface energy of  $\text{Sb}_2\text{S}_3$  on different surfaces

$\text{Sb}_2\text{S}_3$ slab	X(Å)	Y(Å)	A(Å)	$E_{\text{surface}}$ (eV)	$E_{\text{ads}}$ (eV)
(001)	11.12	11.47	127.55	0.023	-0.74
(010)	3.86	11.12	42.92	0.030	-0.55
(100)	11.47	3.86	44.27	0.027	-0.64

Table S2. photodetection property comparison of our work with recent literature

Device structure	$\lambda$ (nm)	$R_{\lambda}(A W^{-1})$	$EQE$ (%)	$D^*$ (Jones)	Rise/decay (ms)	Refs.
Sb <sub>2</sub> S <sub>3</sub> film	530	0.035	-	$3.18 \times 10^{10}$	23.2/58	1
TiO <sub>2</sub> /Sb <sub>2</sub> S <sub>3</sub>	530	0.057	-	$5.30 \times 10^{10}$	0.6/11.8	1
Sb <sub>2</sub> S <sub>3</sub> microrods	560	5.10	1130.68	$2.16 \times 10^{10}$	4.03/4.08	2
Sb <sub>2</sub> S <sub>3</sub> nanowires	532	65	$1.5 \times 10^4$	$2.1 \times 10^{14}$	76/82	3
Sb <sub>2</sub> S <sub>3</sub> tubes	808	0.0085	-	$1.33 \times 10^6$	22/24	4
Sb <sub>2</sub> S <sub>3</sub> /PbI <sub>2</sub>	445	156.3	$4.4 \times 10^4$	$3.2 \times 10^{13}$	8.5/8.7	5
Sb <sub>2</sub> S <sub>3</sub> /PbI <sub>2</sub>	730	120.74	$2.06 \times 10^4$	$2.45 \times 10^{12}$	-	5
Sb <sub>2</sub> Se <sub>3</sub> NTs	830	4.39	655	$9.63 \times 10^{10}$	27/27	6
<b>Sb<sub>2</sub>S<sub>3</sub> microtubes</b>	<b>722</b>	<b>28.99</b>	<b>13021</b>	<b><math>7.6 \times 10^{10}</math></b>	<b>9.8/9.2</b>	<b>This work</b>

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