## Reducing the hydrogen transfer barrier by introduction of Ru *via* constructed Ir-Ru-WO<sub>2.72</sub> bridge for highly CO-tolerant hydrogen oxidation

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Figure S1. (a-b) SEM images of PS template.



Figure S2. (a-b) SEM images of WO<sub>2.72</sub>-C at different magnifications.



Figure S3. N<sub>2</sub> adsorption-desorption isotherms of WO<sub>2.72</sub>-C and IrRu-WO<sub>2.72</sub>-C.



Figure S4. (a-b) TEM images of  $WO_{2.72}$ -C at different magnifications and the corresponding (c) HRTEM images. The d $\approx$ 0.38 nm is assigned to the (010) plane of monoclinic  $WO_{2.72}$ .



Figure S5. (a-d) Bright field (BF) STEM images of WO<sub>2.72</sub>-C at different angles,



Figure S6. (a-d) SEM images of IrRu-WO<sub>2.72</sub>-C at different magnifications.



**Figure S7.** (a-c) TEM images of IrRu-WO<sub>2.72</sub>-C at different magnifications. (d) The size distribution histograms of IrRu alloy clusters for IrRu-WO<sub>2.72</sub>-C.



Figure S8. AC HAADF-STEM images (a, b) and liner elemental scanning profiles of

IrRu clusters (c).



Figure S9. Digital photos of (a) WO<sub>3</sub>, (b)WO<sub>2.72</sub>, (c) WO<sub>2.72</sub>-C and (d) IrRu-WO<sub>2.72</sub>-C.



Figure S10. The Tauc plots of UV–vis spectra for (a) IrRu-WO<sub>2.72</sub>-C, (b) WO<sub>2.72</sub>-C, (c) WO<sub>2.72</sub> and (d) WO<sub>3</sub>.



Figure S11. High-resolution XPS spectra of W 4f for IrRu-WO<sub>2.72</sub>-C, Ru-WO<sub>2.72</sub>-C and

WO<sub>2.72</sub>-C.



Figure S12. High-resolution XPS spectra of Ru 3p for IrRu-WO<sub>2.72</sub>-C and Ru-WO<sub>2.72</sub>-C

C.



Figure S13. High-resolution XPS spectra of Ir 4d for IrRu-WO<sub>2.72</sub>-C and Ir-WO<sub>2.72</sub>-C.



Figure S14. High-resolution XPS spectra of O 1s for IrRu-WO<sub>2.72</sub>-C, Ru-WO<sub>2.72</sub>-C and

WO<sub>2.72</sub>-C.



Figure S15. ESR spectra of IrRu-WO<sub>2.72</sub>-C, Ir-WO<sub>2.72</sub>-C, Ru-WO<sub>2.72</sub>-C and WO<sub>2.72</sub>-C.



Figure S16. High-resolution XPS spectra of C 1s and Ru 3d for IrRu-WO<sub>2.72</sub>-C, Ir-WO<sub>2.72</sub>-C, Ru-WO<sub>2.72</sub>-C and WO<sub>2.72</sub>-C.



Figure S17. HOR polarization curves of IrRu-WO<sub>2.72</sub>-C in N<sub>2</sub> or H<sub>2</sub> saturated 0.5 M  $H_2SO_4$ .



**Figure S18.** HOR polarization curves of (a) 20 wt.% Pt/C, (b) Ru-WO<sub>2.72</sub>-C, (c) Ir-WO<sub>2.72</sub>-C and (d) WO<sub>2.72</sub>-C at different rotating speeds.



**Figure S19.** HOR polarization curves (a) and Tafel plots of kinetic current densities (b) for IrRu-WO<sub>2.72</sub>-C, 0.5 IrRu-WO<sub>2.72</sub>-C and 1.5 IrRu-WO<sub>2.72</sub>-C.



Figure S20. HOR polarization curves of (a) 0.5 IrRu-WO<sub>2.72</sub>-C and (b) 1.5 IrRu-WO<sub>2.72</sub>-C at different rotating speeds. (c) Koutecky–Levich plots of the catalysts at 0.15 V vs RHE.



Figure S21. HOR polarization curves (a) and Tafel plots of kinetic current densities (b)

for IrRu-WO<sub>2.72</sub>-C, IrRu(2:1)-WO<sub>2.72</sub>-C and IrRu(1:2)-WO<sub>2.72</sub>-C.



Figure S22. HOR polarization curves of (a) IrRu(2:1)-WO<sub>2.72</sub>-C and (b) IrRu(1:2)-WO<sub>2.72</sub>-C at different rotating speeds. (c) Koutecky-Levich plots of the catalysts at 0.15 V *vs.* RHE.



Figure S23. High-resolution XPS spectra of W 4f (a), Ir 4f (b), Ru 3p (c) and O1s (d)

for IrRu-WO<sub>2.72</sub>-C and after stability test named as IrRu-WO<sub>2.72</sub>-C-after.



Figure S24. SEM image of IrRu-WO<sub>2.72</sub>-C after stability test.



Figure S25. HOR polarization curves in  $H_2$  and 1,000 ppm CO/ $H_2$  saturated electrolytes for (a) 20 wt.% Pt/C, IrRu-WO<sub>2.72</sub>-C and (b) Ir-WO<sub>2.72</sub>-C.



Figure S26. Side view of model structure for WO<sub>2.72</sub>-C.



Figure S27. Side view of different H adsorption model structure on IrRu-WO<sub>2.72</sub>-C.



Figure S28. Side view of different H adsorption model structure on Ir-WO<sub>2.72</sub>-C.



Figure S29.Schematic representation of the HOR mechanism and active hydrogen

transfer	pathway	on	the	Ir-WO <sub>2.72</sub> -C	surface.
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Figure S30. Power density curves of IrRu-WO<sub>2.72</sub>-C, Ir-WO<sub>2.72</sub>-C, Ru-WO<sub>2.72</sub>-C and WO<sub>2.72</sub>-C.

Table	<b>S1</b> .	The	Ir and	Ru	loading	gs of	samp	les	tested	by	ICP-	OES.
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Samples	Percentage of Ir (wt.%)	Percentage of Ru (wt.%)
IrRu-WO <sub>2.72</sub> -C	1.19	1.65
Ir-WO <sub>2.72</sub> -C	1.65	/
Ru-WO <sub>2.72</sub> -C	/	2.43

Samples	The electrical conductivity values (S cm <sup>-1</sup> )			
WO <sub>3</sub>	1.30×10 <sup>-5</sup>			
IrRu-WO <sub>2.72</sub> -C	9.29×10 <sup>-5</sup>			
WO <sub>2.72</sub> -C	0.22			
IrRu-WO <sub>2.72</sub> -C	2.41			

 Table S2. The electrical conductivity values of as-prepared catalysts.

Catalyst	HOR current density (mA/cm <sup>2</sup> )	The mass loading	Papers
Ru@TiO <sub>2</sub>	~ 2.9	$25.07 \mu g_{Ru} \text{ cm}^{-2}$	Nat. Catal. <sup>1</sup>
IrP <sub>2</sub> -rGO	~ 2.5	$8.84 \ \mu g_{Ir} \ cm^{-2}$	ACS Appl. Mater. Interfaces <sup>2</sup>
PdRu-WO <sub>x</sub> /C	~ 3.1	49.44 $\mu g_{Pd}  cm^{-2}$	Catal. Today <sup>3</sup>
Ni <sub>x</sub> Mo <sub>1-x</sub> O <sub>2</sub>	~ 0.95	/	ACS Energy Lett. <sup>4</sup>
Rh-Rh <sub>2</sub> O <sub>3</sub> NPs/C	~3.3	$10.20 \ \mu g_{Rh} \ cm^{-2}$	J. Mater. Chem. A <sup>5</sup>
Ir <sub>NP</sub> @Ir <sub>SA</sub> -N-C	~2.7	$5.61 \mu g_{Ir} \text{ cm}^{-2}$	Angew. Chem., Int. Ed.
IrRu-N-C	~3.2	3.06 µg <sub>Ir</sub> cm <sup>-2</sup>	Proc. Natl. Acad. Sci. U.S.A <sup>7</sup>
IrRu-WO <sub>2.72</sub> -C	~3.35	$5.41 \mu g_{Ir} \text{ cm}^{-2}$	This work

 Table S3. HOR performances in acidic media in the latest reported literature.

Catalyst	HOR current density (mA cm <sup>-2</sup> )	mass loading of catalyst (mg cm <sup>-2</sup> )	mass loading of precious metal (μg <sub>Ir/Pt</sub> cm <sup>-2</sup> )
IrRu-WO <sub>2.72</sub> -C	~3.35	0.51	5.41
Ir-WO <sub>2.72</sub> -C	~3.12	0.51	8.62
20 wt.% Pt-C	~2.68	0.51	102.04

 Table S4. The performance comparison and corresponding catalyst loading.

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