Electrochemical Synthesis Wormcast-like Pd-based Polycrystalline High Entropy Aggregates for Methanol Water Co-electrocatalysis

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2 Figure S1. a) CV curves of Pd, Fe, Co, Ni, and Cu metal salts in 1 M Na₂SO₄ solution; b) Recorded data curves

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5	or current	and vona	ge during	uic syi	Iuncoio	procedure.





Figure S2. Morphology of Pd-Ns. a) SEM image; TEM images for a) bright-field and b) dark-field; c) HR-TEM
image recorded from a selected region in image a); d, e) IFFT images according to the selected regions in image
c); f) Elemental EDX mapping images.



Figure S3. Morphology of FeCoNiCu-pHENs. a) SEM image; TEM images for a) bright-field and b) dark-field;
c) HR-TEM image recorded from a selected region in image a); d, e, f) IFFT images according to the selected
regions in image c); g) Elemental EDX mapping images.







27 Figure S5. Element mass ratios of a) PHENs-MLP, b) PHENs-HLP, c) PHENs-LLP, and d) FeCoNiCu-pHENs.



Figure S6 TEM images of a), b) P-pHENs, c), d) pHENs-LLP, and e), f) pHENs-HLP.





32 Figure S7. XRD patterns of PdFeCoNiCu-pHENs (pHENs-MLP) and glassy carbon substrate.



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Figure S8. CV curves of samples were recorded in 1 M KOH solution. a) pHENs-MLP; b) pHENs-LLP; c)
pHENs-HLP; d) Pd-Ns; e) Pd/C-20_{wt}%; f) Pt/C-20_{wt}%; g) corresponding ECSA of various samples.





Figure S9. CV curve of Pt/C-20_{wt.}% was recorded in 0.5 M H₂SO₄ solution.



42 Figure S10. CV curves of PdFeCoNiCu-pHENs with different applied potentials in an N2-saturated 1.0 M KOH

- 43 solution containing 1.0 M CH₃OH. a) Mass activity and b) specific activity toward MOR.
- 44





46 Figure S11. CV curves of FeCoNiCu-pHENs were recorded in N₂-saturated 1.0 M KOH solution mixing with

- 47 or without $1.0 \text{ M CH}_3\text{OH}$.





Figure S12. Mass activities of commercial catalysts before and after the chronoamperometric test. a) Mass and
b) specific activity of Pt/C-20_{wt}%; c) Mass and d) specific activity of Pd/C-20_{wt}%; e) Mass and f) specific
activity of Pd-Ns.



57 Figure S13. MOR electrocatalytic performances of a) pHENs-MLP, b) Pd-Ns, c) Pd/C-20_{wt.}%, and d) Pt/C-20_{wt.}%

⁵⁸ before and after the CO-stripping test.



61 Figure S14. The calculated model of PdFeCoNiCu-pHENs a) before and b) after the structural optimization.



64 Figure S15. The absorption sites of a) CH₃OH, b) CH₃O, c) CH₂O, d) CHO, and e) CO on PdFeCoNiCu-pHENs.





67 Figure S16. CV curves of a) pHENs-MLP; b) pHENs-LLP; c) pHENs-HLP; d) Pd-Ns; e) Pd/C-20_{wt.}%; f) Pt/C-









Figure S18. HER mass activity of pHENs-MLP before and after 10,000 times LSV repetitive test.



Figure S19. Electroactivities of contrast samples before and after 10000 repeats HER test. a) Mass and b) specific
activity of Pd-Ns; c) Mass and d) specific activity of Pd/C-20_{wt.}%; e) Mass and f) specific activity of Pt/C-20_{wt.}%.





Figure S20. The absorption sites of a) *OH and b) *H on the PdFeCoNiCu-pHENs.



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84 Figure S21. Chronopotentiometry curve of PdFeCoNiCu-pHENs (pHENs-MLP) at a specific current density of

85 100 mA cm⁻².



- 88 Figure S22. Element mass ratio of PdFeCoNiCu-pHENs after MOR-assisted hydrogen generation test. a) EDX
- 89 spectra and b) element mass ratio of PHENs-MLP-used.



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92 Figure S23. XRD patterns of the pHENs-MLP-used and glassy carbon substrate.
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96 Figure S24. XPS characterization of the pHENs-MLP-used. a) Survey curve; High-resolution XPS spectrums of
97 b) Pd 3d, c) Cu 2p, d) Fe 2p, e) Co 2p, and f) Ni 2p.