

## Supporting Information

### Dual Protective Layer on Lithium Metal Anodes for Improved Electrochemical Performance – In-Depth Morphological Characterization

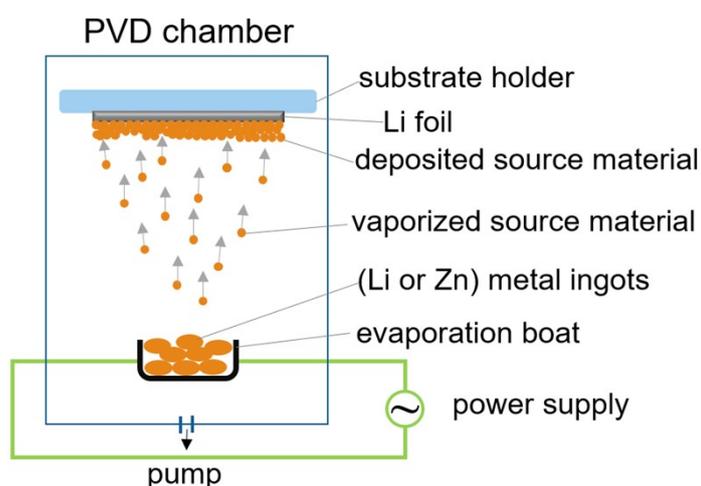
Marlena M. Bela <sup>a</sup>, Maximilian Mense <sup>a</sup>, Sebastian Greiwe<sup>a</sup>, Marian C. Stan<sup>b</sup>, Simon Wiemers-Meyer<sup>a</sup>, Martin Winter <sup>a, b</sup>, Markus Börner <sup>a</sup>

<sup>a</sup> MEET Battery Research Center, University of Münster, Corrensstraße 46, 48149 Münster, Germany

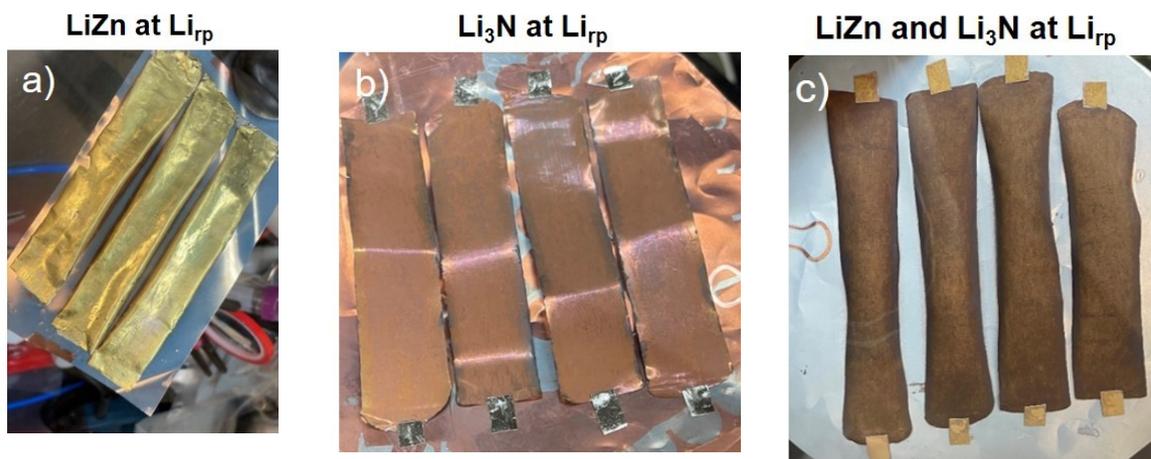
<sup>b</sup> Helmholtz-Institute Münster, IEK-12, Forschungszentrum Jülich GmbH, Corrensstraße 46, 48149 Münster, Germany

#### Experimental: Raman spectroscopy

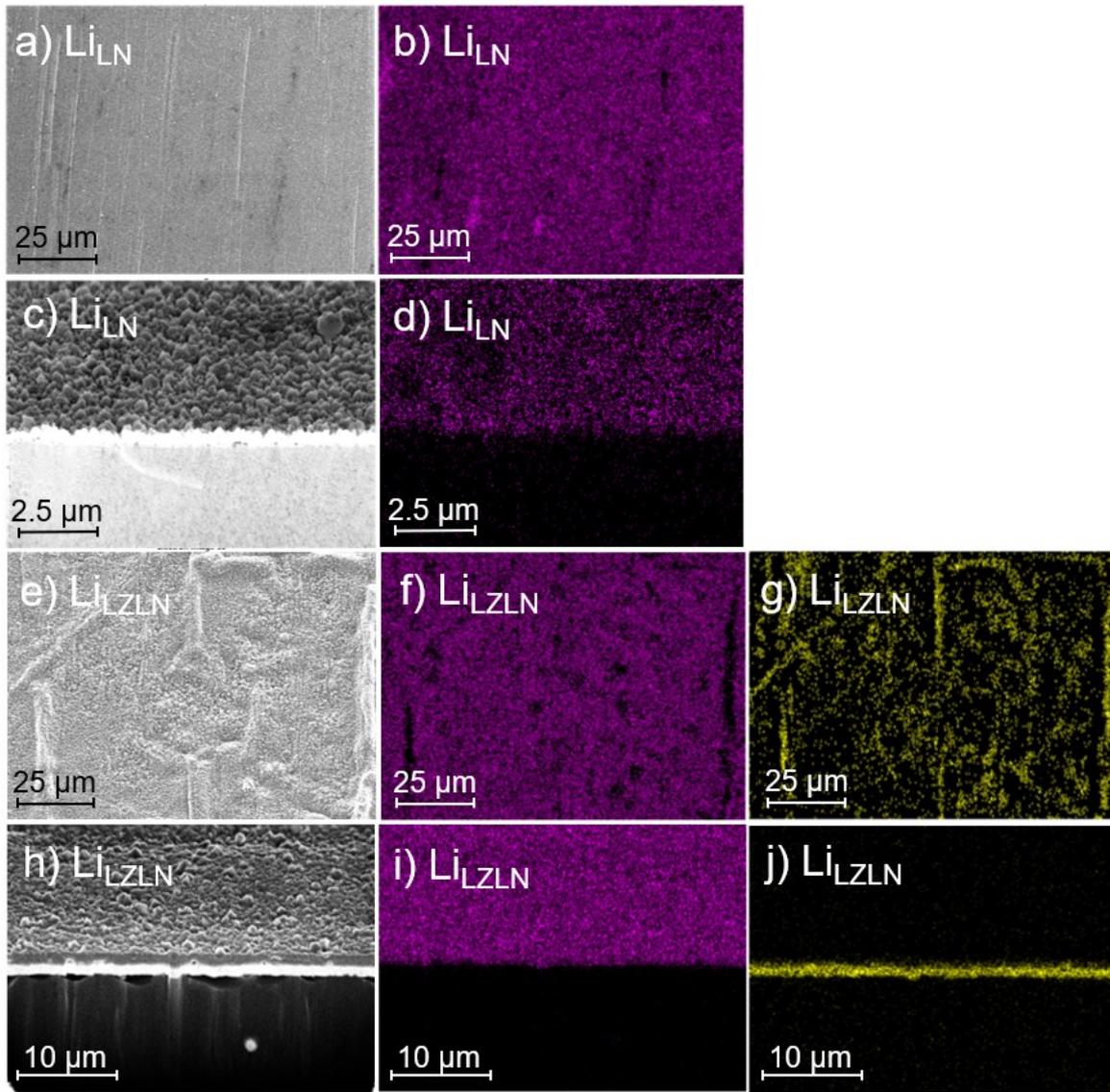
Raman spectroscopy was performed on a LabRAM HR Evolution Raman spectrometer (Horiba Scientific). The following parameters were used for all measurements. A green laser (Nd:YAG, 532 nm) with a 5% laser intensity (corresponding to 1.5 mW), a 50x long distance objective (Olympus), a grating of 300 gr mm<sup>-1</sup>, in a range of 3000 cm<sup>-1</sup> ≥  $\nu$  ≥ 100 cm<sup>-1</sup>. The acquisition time was set to 30 s with an accumulation of 5 s.



**Figure S1:** Schematic illustration of the PVD chamber. Interchangeable tungsten or tantalum boats are installed at the bottom of the chamber for the respective Li or Zn metal to be vaporized. Li metal foil at the substrate holder is coated with the vaporized source material.

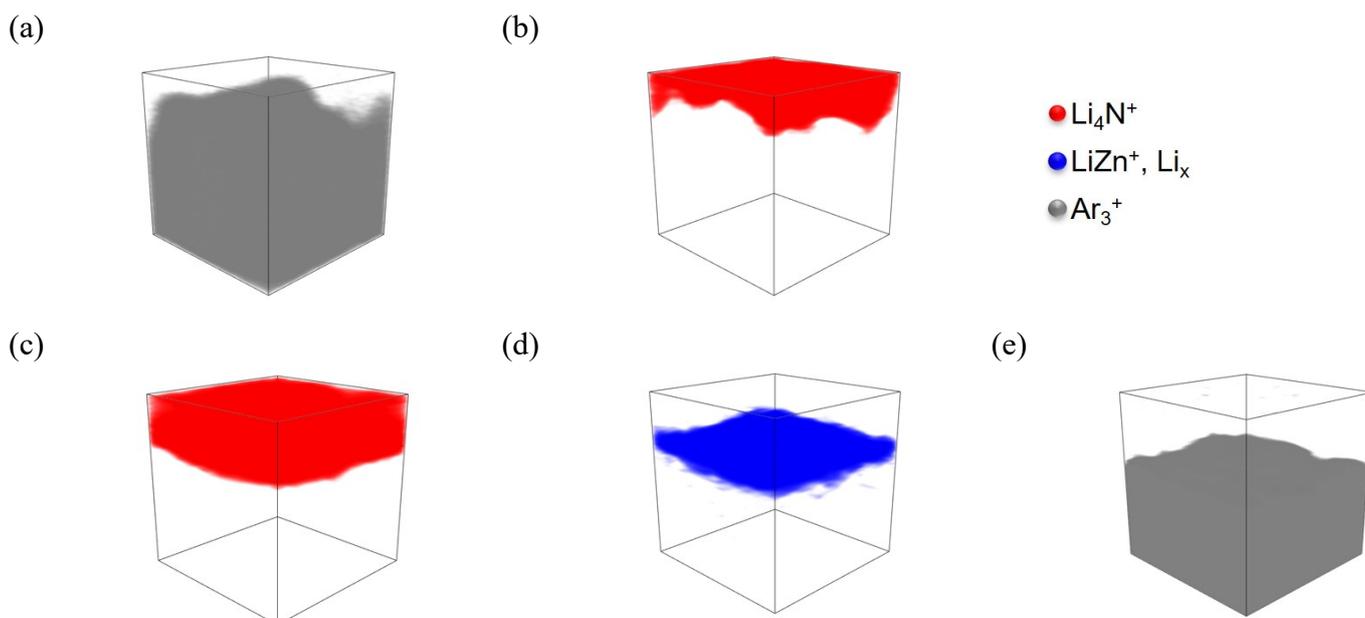


**Figure S2:** Optical images after thermal evaporation to form a) LiZn at Li<sub>rp</sub> b) Li<sub>3</sub>N at Li<sub>rp</sub> and c) LiZn and Li<sub>3</sub>N at Li<sub>rp</sub> (dual layer, LZLN). The images were taken inside an argon-filled glovebox.

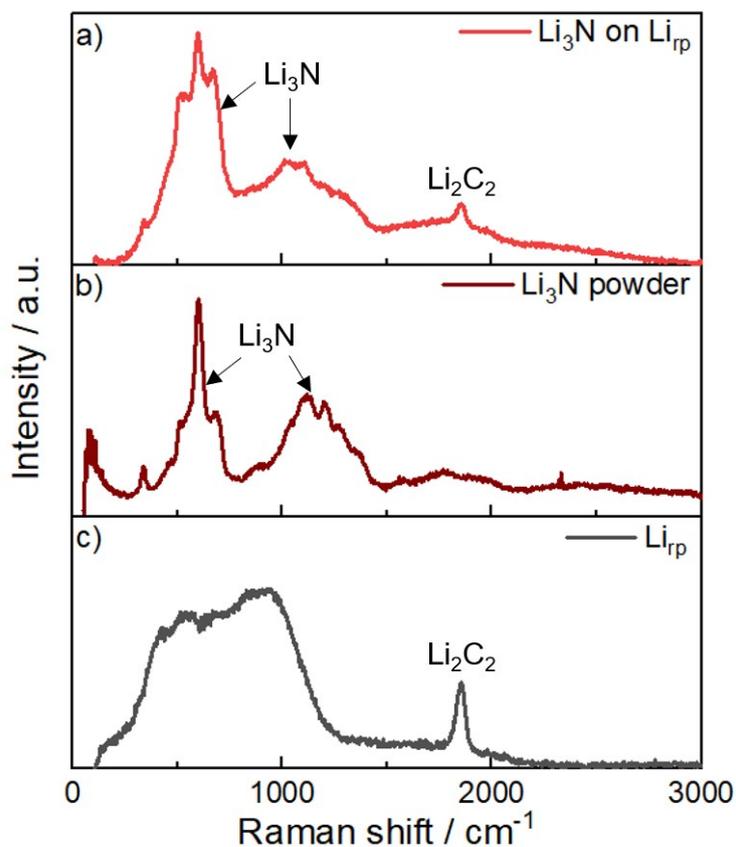


**N** **Zn** Fig

**ure S3:** SEM images and corresponding elemental mapping of a-d)  $\text{Li}_3\text{N}$  layer on  $\text{Li}_{\text{ip}}$  and e-j) LZLN on  $\text{Li}_{\text{ip}}$  before cycling. a-b) Top-view and c-d) cross-section image of the  $\text{Li}_3\text{N}$  layer on  $\text{Li}_{\text{ip}}$ . e-g) Top-view and h-j) cross-section image of the LZLN layer on  $\text{Li}_{\text{ip}}$ .



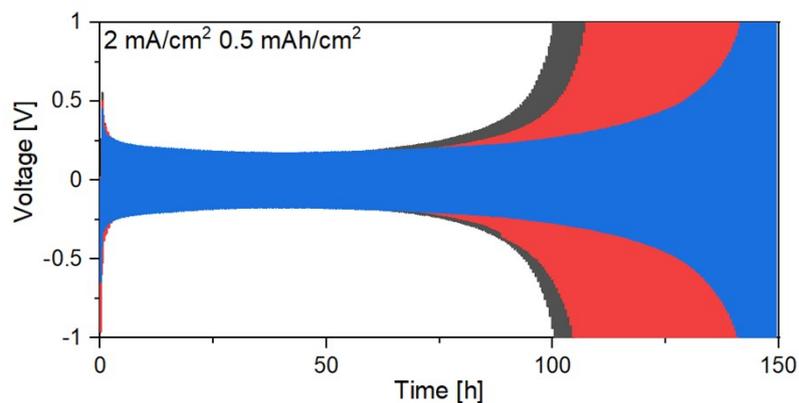
**Figure S4:** 3D reconstruction of corresponding ToF-SIMS depth profiles in an area of  $100 \times 100 \mu\text{m}^2$  from a-b)  $\text{Li}_3\text{N}$  at  $\text{Li}_{\text{rp}}$  and c-e) LZLN at  $\text{Li}_{\text{rp}}$ . Individual spatial distributions of selected secondary ions are marked in red:  $\text{Li}_4\text{N}^+$ , blue:  $\text{LiZn}^+$  and  $\text{Li}_x$  ( $x = 7-9$ ) and grey:  $\text{Ar}_3^+$ . The multicolor overlay is represented in Figure 3.



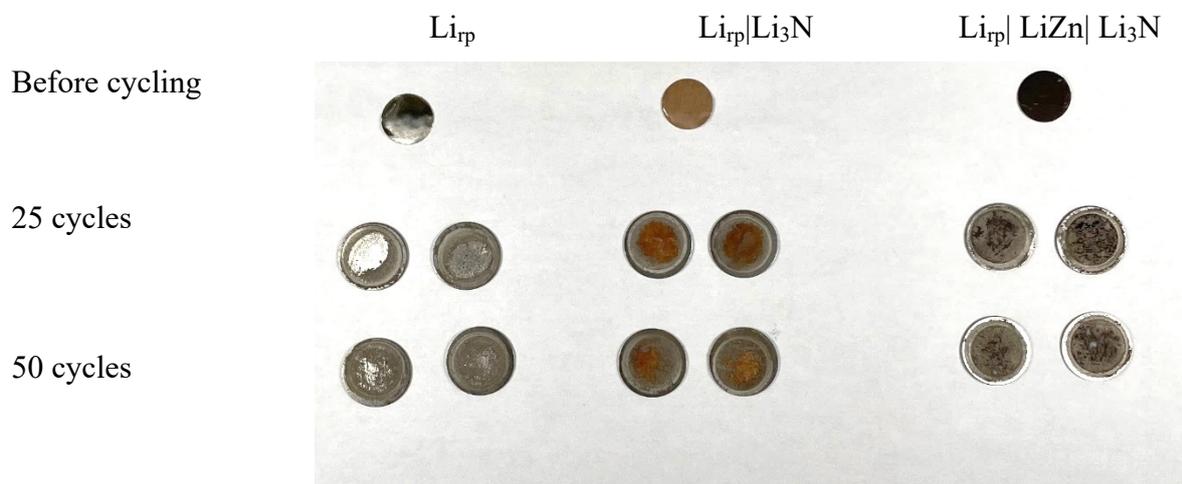
**Figure S5:** Raman spectra of (a)  $\text{Li}_{\text{rp}}$  coated with 100 nm Li via thermal evaporation and subsequent reaction with  $\text{N}_2$  in the PVD chamber, (b)  $\text{Li}_3\text{N}$  powder and (c)  $\text{Li}_{\text{rp}}$  as reference.

**Table S1:** Cycle number of symmetric cells with  $\text{Li}_{\text{rp}}$ ,  $\text{Li}_{\text{LN}}$  or  $\text{Li}_{\text{LZLN}}$  negative electrodes at the cut-off voltage ( $\pm 1.5$  V) for different current densities.

Current density	$\text{Li}_{\text{rp}} \parallel \text{Li}_{\text{rp}}$	$\text{Li}_{\text{LN}} \parallel \text{Li}_{\text{LN}}$	$\text{Li}_{\text{LZLN}} \parallel \text{Li}_{\text{LZLN}}$
$0.5 \text{ mA cm}^{-2}$	484	600	770
$1.0 \text{ mA cm}^{-2}$	200	290	360
$2.0 \text{ mA cm}^{-2}$	102	110	145



**Figure S6:** Galvanostatic cycling at a fixed current density of  $2.0 \text{ mA/cm}^2$  with a capacity of  $0.5 \text{ mAh/cm}^2$  of  $\text{Li}_{\text{rp}}$  (grey),  $\text{Li}_3\text{N}$  (red) and LZLN symmetric  $\text{Li} \parallel \text{Li}$  cells.



**Figure S7:** Optical images of  $\text{Li}_{\text{rp}}$ ,  $\text{Li}_3\text{N}$  and LZLN before cycling and after 25 and 50 cycles. Cell disassembly, washing with dimethyl carbonate and photo capturing were pursued in the dry room.