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Supporting Information

for

Unraveling Surface Polarization in Hydrothermally Derived AgFeO₂

Nanosheets for Enhanced Photoelectrochemical Performance

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$$Ag^{+} + H_2O \rightarrow [AgOH_2]^{+} \rightarrow AgOH + H^{+} \rightarrow Ag_2O$$
 (Eqs. 1)

$$Ag_2O + 3 H_2O \rightarrow 2 [Ag(OH)_2]^- + 2 H^+$$
 (Eqs. 2)

$$Fe^{3+} + 6 H_2O \rightarrow [Fe(OH_2)_6]^{3+} \rightarrow [Fe(OH)_6]^{3-} + 6 H^+$$
 (Eqs. 3)

$$[Ag(OH)_2]^- + [Fe(OH)_6]^{3-} + 4 H^+ \rightarrow AgFeO_2 + 6 H_2O \qquad (Eqs. 4)$$



Figure S1. (a) XRD patterns; (b-e) SEM images of AgFeO₂ nanosheets with different NaOH additions at 2.1, 2.7, 4.0 and 8.0 g, respectively.



Figure S2. Influence of different NaOH concentrations on the width (A1-D1) and thickness (A1-D4) of AgFeO₂ nanosheets.



Figure S3. XRD patterns of AgFeO₂ nanosheets under (a) different reaction temperatures and (b) reaction time.



Figure S4. SEM images of AgFeO₂ nanosheet sizes regulated by different (a-c) reaction temperatures at 190, 210 and 230 $^{\circ}$ C; (d-f) reaction time for 12, 15 and 24 h.



Figure S5. Effects of different reaction temperatures on the width (a2-c2) and thickness (A2-C2) of AgFeO₂ nanosheets.



Figure S6. Effects of different reaction times on the width (a3-c3) and thickness (A3-C3) of AgFeO₂ nanosheets.



Figure S7. XRD pattern of AgFeO₂ nanosheets with different thickness.



Figure S8. (a5-e5) width and (A5-E5) thickness of AgFeO₂ nanosheets with different thicknesses.



Figure S9. (a) EIS curve; (b) M-S curve; (c) LSV curve; (d) Tafel slope diagram; (e) UV-VIS curve of AgFeO₂ electrodes with different thickness.



Figure S10. XRD patterns; (b-e) SEM of AgFeO₂ nanosheets regulated by different (001) polar surfaces.



Figure S11. (a4-d4) width and (A4-D4) thickness of AgFeO₂ nanosheets with different (001) polar exposed surfaces.



Figure S12. (a) EIS curves; (b) M-S curve; (c) the LSV curve; (d) Tafel slope diagram; (e) UV-Vis curve of AgFeO₂ electrodes with different (001) polarity exposure surfaces.

Reaction	Reaction	NaOH concentration	Nanosheet	Nanosheet	Crystalline	Sample					
temperature	time		width	thickness	phase	labeling					
190°C	12 h	1.5 g (1.875 mol/L)	-	-	No pure phase is formed	-					
190°C	12 h	1.8 g (2.250 mol/L)	-	-	No pure phase is formed	-					
190℃	12 h	2.1 g (2.625 mol/L)	100 nm	40 nm	3R	AFO-100					
190℃	12 h	2.7 g (3.375 mol/L)	270 nm	190 nm	3R	AFO-270					
190℃	12 h	4.0 g (5.000 mol/L)	470 nm	230 nm	3R	AFO-470					
190℃	12 h	8.0 g (10.00 mol/L)	450 nm	200 nm	3R	AFO-450					
180°C	12 h	2.4 g (3.000 mol/L)	-	-	No pure phase is formed						
190℃	12 h	2.4 g (3.000 mol/L)	250 nm	180 nm	3R	AFO-250					
210°C	12 h	2.4 g (3.000 mol/L)	280 nm	220 nm	3R	AFO-280					
230°C	12 h	2.4 g (3.000 mol/L)	320 nm	250 nm	3R	AFO-320					
190°C	6 h	2.1 g (2.625 mol/L)	-	-	No pure phase is formed						
190°C	9 h	2.1 g (2.625 mol/L)	-	-	No pure phase is formed						
190℃	15 h	2.1 g (2.625 mol/L)	380 nm	210 nm	3R	AFO-380					
190℃	24 h	2.1 g (2.625 mol/L)	540 nm	270 nm	2H, 3R	AFO-540					
190°C	48 h	2.1 g (2.625 mol/L)	-	-	No pure phase is formed						
210°C	12 h	2.7 g (3.375 mol/L)	300 nm	140 nm	3R	AFO-300					
230°C	12 h	8.0 g (10.00 mol/L)	820 nm	200 nm	3R	AFO-820					
230°C	48 h	8.0 g (10.00 mol/L)	1000 nm	350 nm	3R	AFO-1000					

 Table S1. The regulation of hydrothermal reaction temperature, time and NaOH concentration on the size of AgFeO2 nanosheets.

Table S2. The BET area and electrochemically active surface area (ECSA) of typical AgFeO₂ nanosheets.

Characteristics	Label	Thickness	BET area	BET area	Normalized	ECSA
	Laber	(nm)	(m^2/g)	Thickness	$ECSA(cm^2)$	Thickness
Comparable exposed area	AFO-100	40	3.8246	0.0956	0.7412	0.0185
	AFO-300	140	5.4547	0.0390	0.7456	0.0053
	AFO-820	200	6.9024	0.0345	0.7450	0.0038
	AFO-470	230	6.5179	0.0283	0.7432	0.0032
	AFO-1000	350	4.7389	0.0135	0.9729	0.0028
Comparable thickness	AFO-820	200	6.9024	0.0345	0.7450	0.0038
	AFO-470	230	6.5179	0.0283	0.7432	0.0032
	AFO-450	200	5.6059	0.0280	0.6295	0.0031
	AFO-270	190	3.6666	0.0193	0.5673	0.0030