

## Supporting Information

### **Study of Structural and Composition Redesign to Enhanced the Thermostability and Electrochemical Performance of Co-less Ni-rich $\text{LiNi}_{0.92}\text{Co}_{0.04}\text{Mn}_{0.04}\text{O}_2$ Layered Cathode through Transition-metal Concentration Gradient Strategies**

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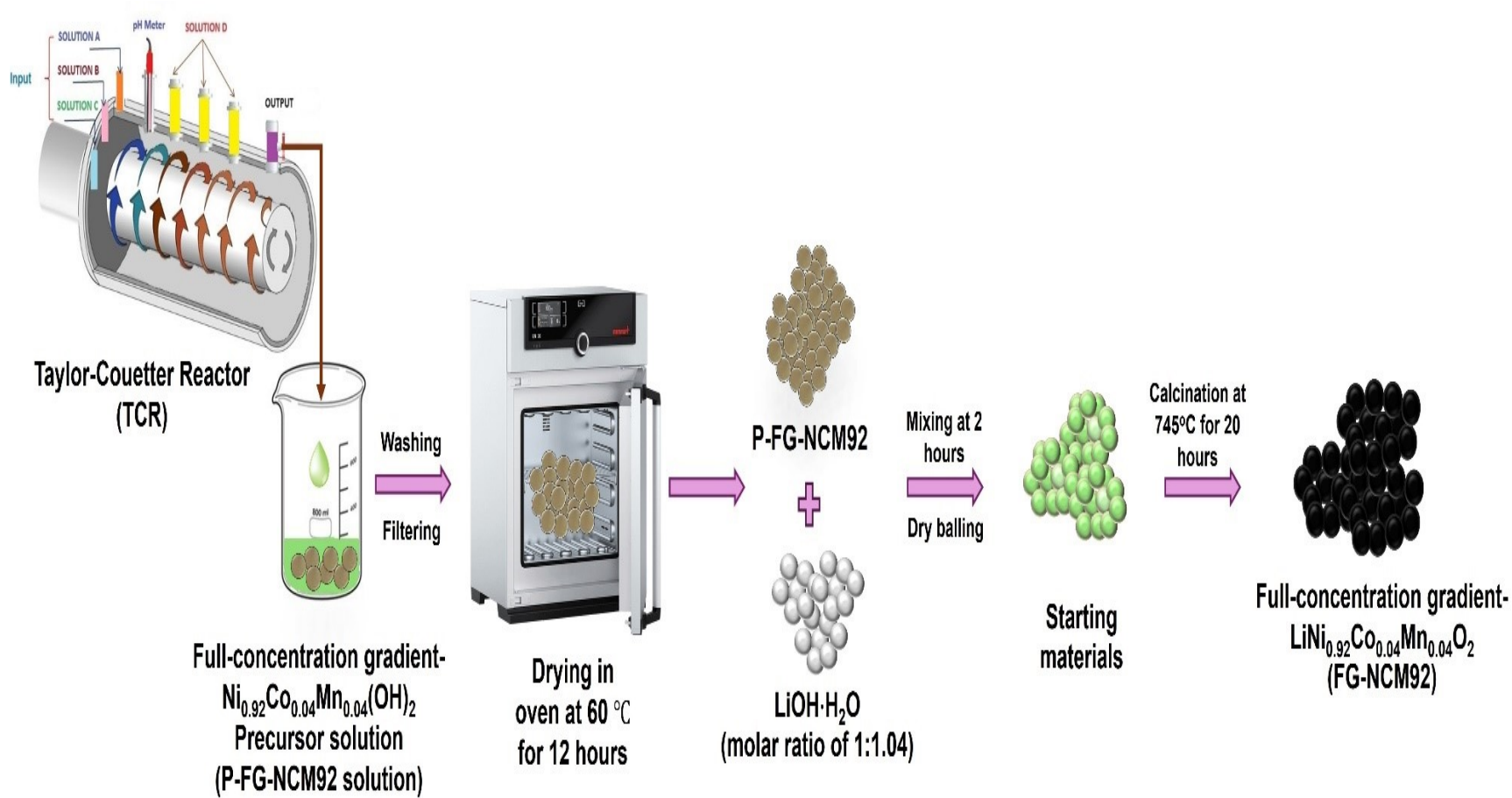
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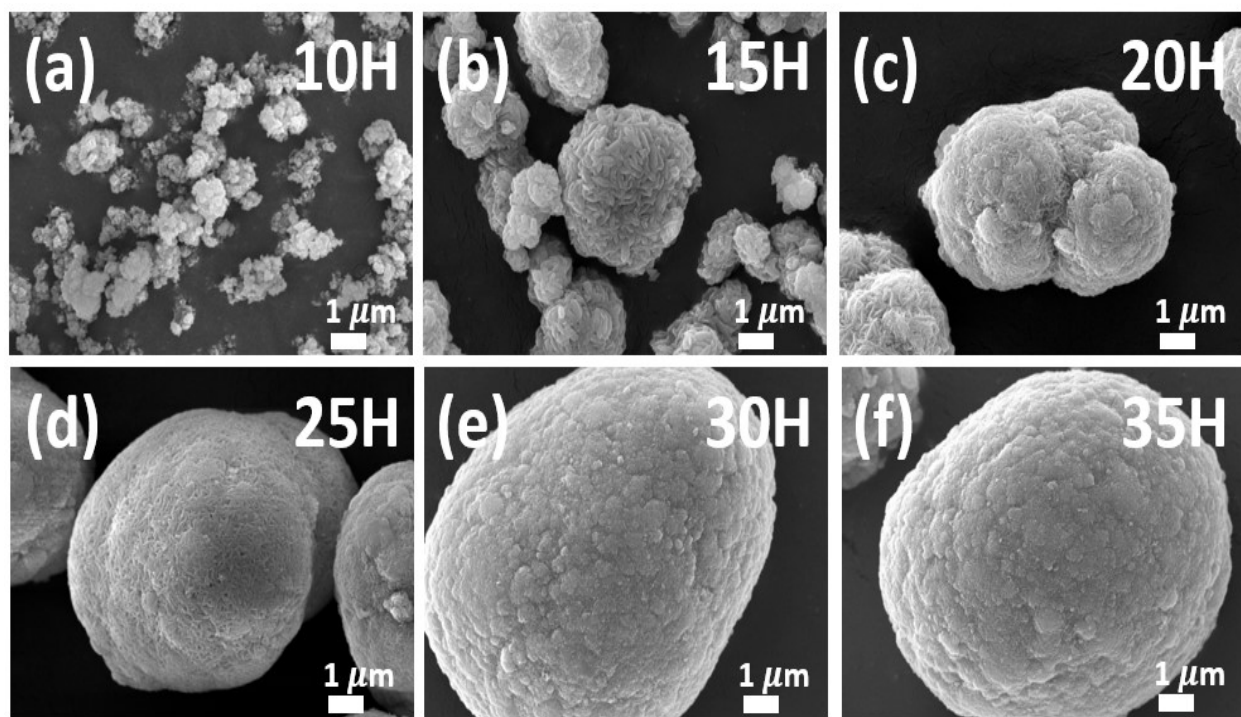
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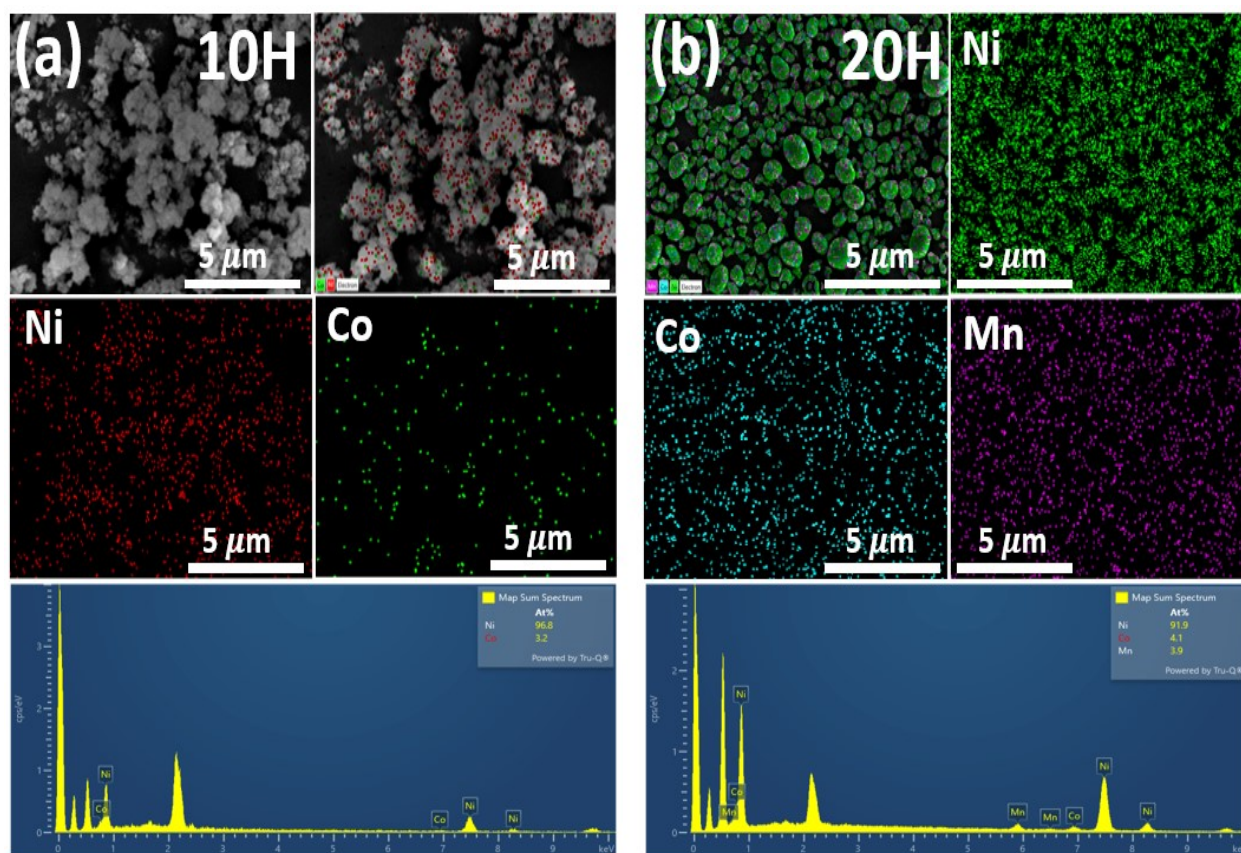
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**Fig. S1.** Schematic diagram of FG-NCM92 synthesis process by using a novel scalable Taylor–Vortex Reactor (TVR).

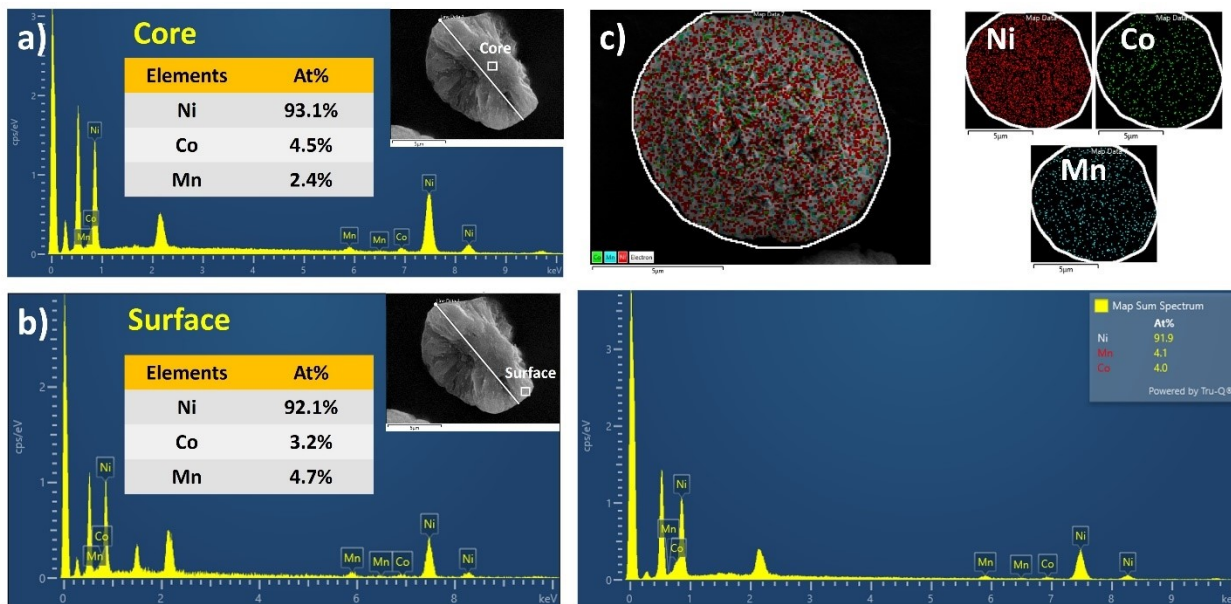


**Fig. S2.** The morphological evolution of P-FG-NCM92 hydroxide particles as a function of reaction time synthesis by using a novel scalable Taylor–Vortex Reactor (TVR): (a). 10 h, (b). 15 h, (c). 20 h, (d). 25 h, (e). 30 h, (f). 35 h, respectively.

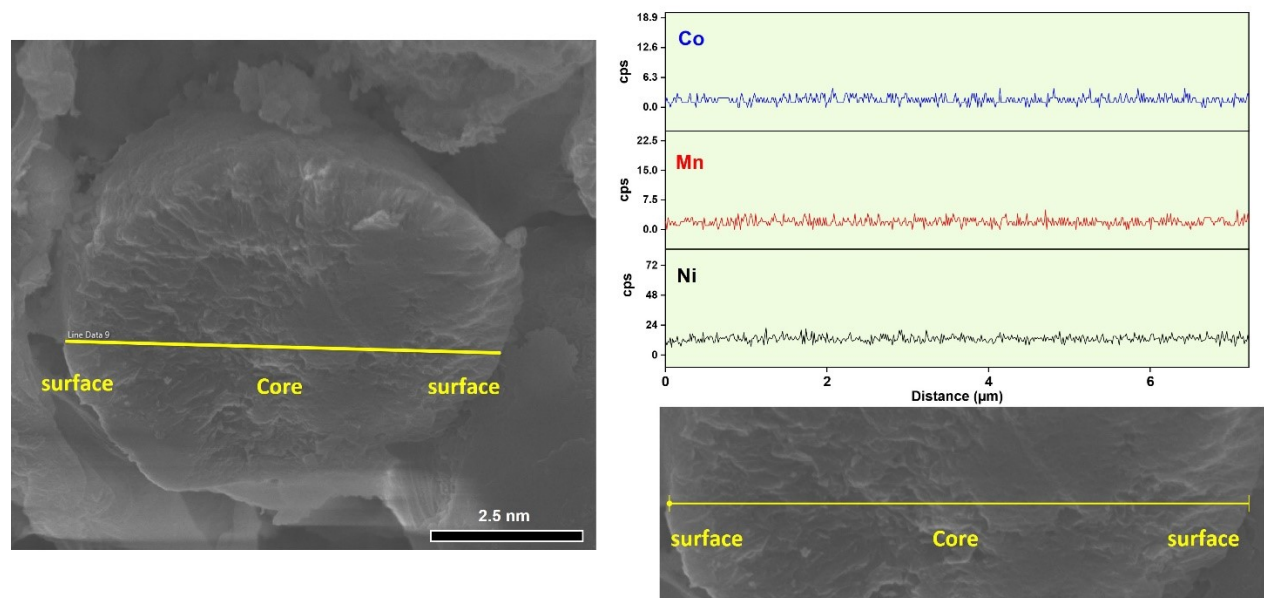


**Fig. S3.** EDX images of FG-Ni<sub>0.92</sub>Co<sub>0.04</sub>Mn<sub>0.04</sub>(OH)<sub>2</sub> precursors synthesized by using TVR reactor

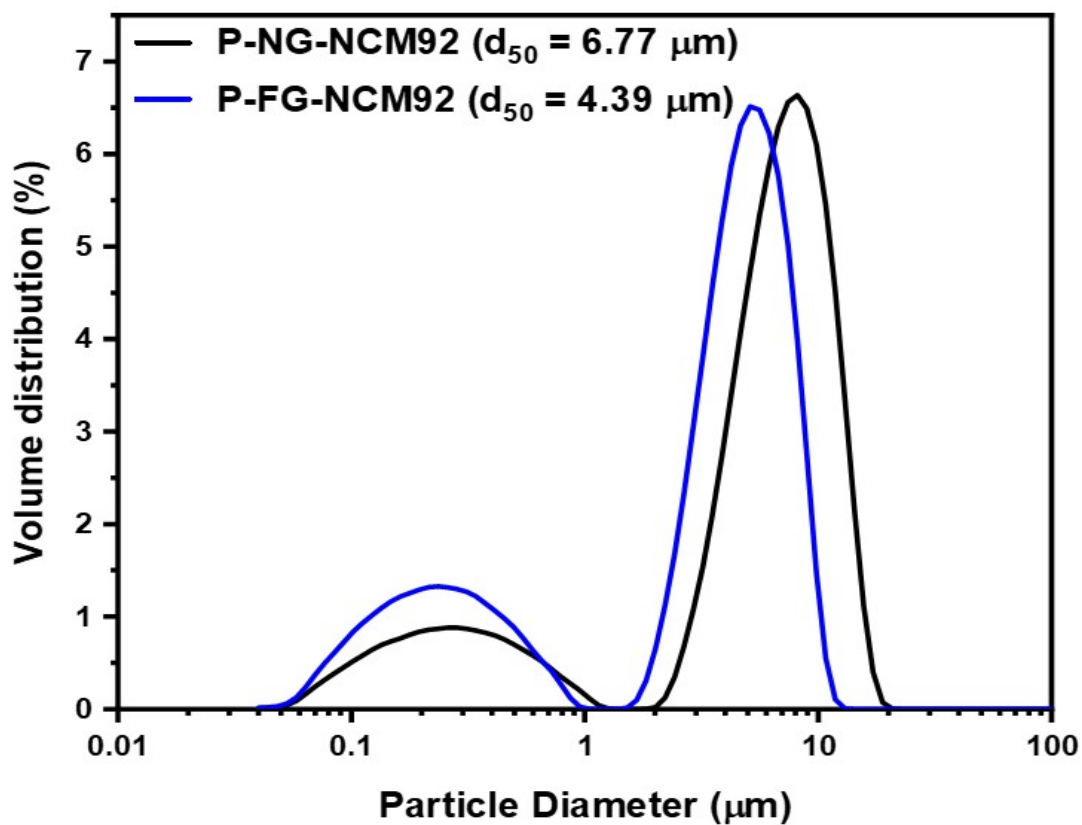
at (a) 10 h, and (b) 20 h.



**Fig. S4.** The EDS graph of (a) area core and (b) area surface in the  $\text{FG-Ni}_{0.92}\text{Co}_{0.04}\text{Mn}_{0.04}(\text{OH})_2$ , and (c) single particle of  $\text{FG-Ni}_{0.92}\text{Co}_{0.04}\text{Mn}_{0.04}(\text{OH})_2$ .

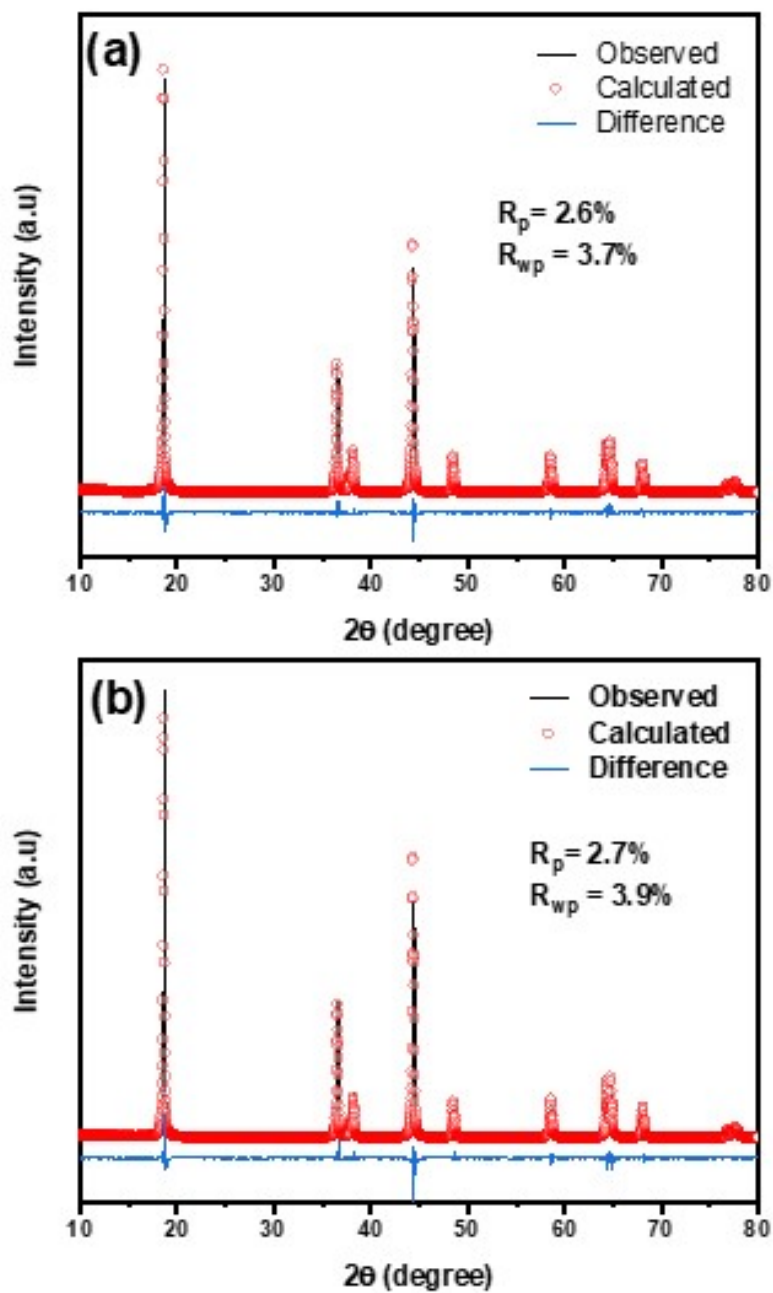


**Fig. S5.** FESEM-EDS-line-scanning of Ni, Co, Mn for P-NG-NCM92 precursors synthesized by a Taylor-Vortex Reactor (TVR).



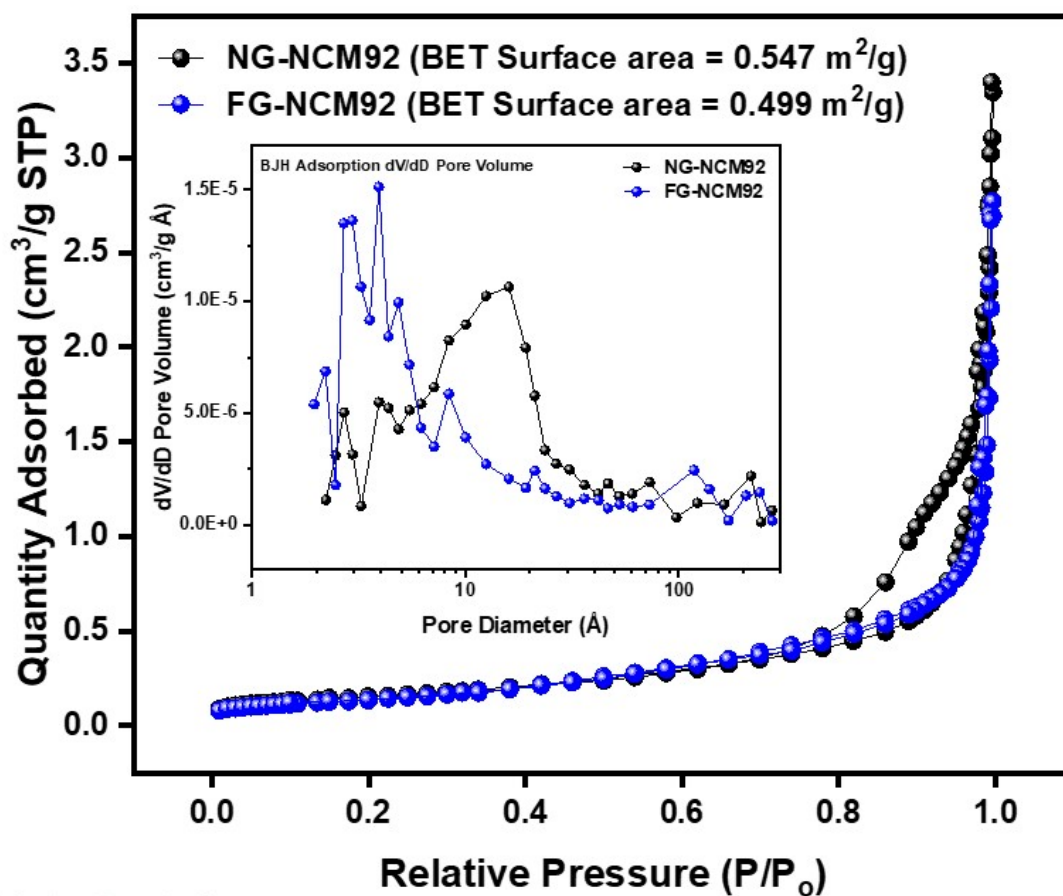
**Fig. S6.** Particle size distribution of (a) P-NG-NCM92 (black line), and (b) P-FG-NCM92 (blue line) hydroxide precursors.



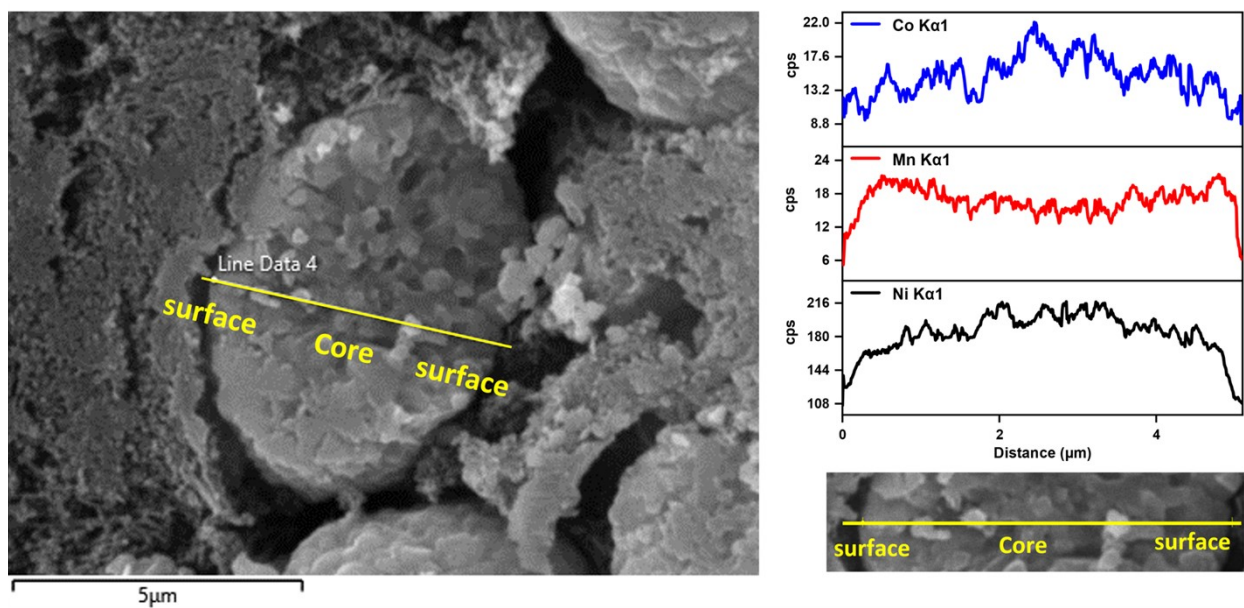


**Fig. S7.** XRD Rietveld refinement results of (a) NG-NCM92, and (b) FG-NCM92 layered oxides cathode materials, respectively.

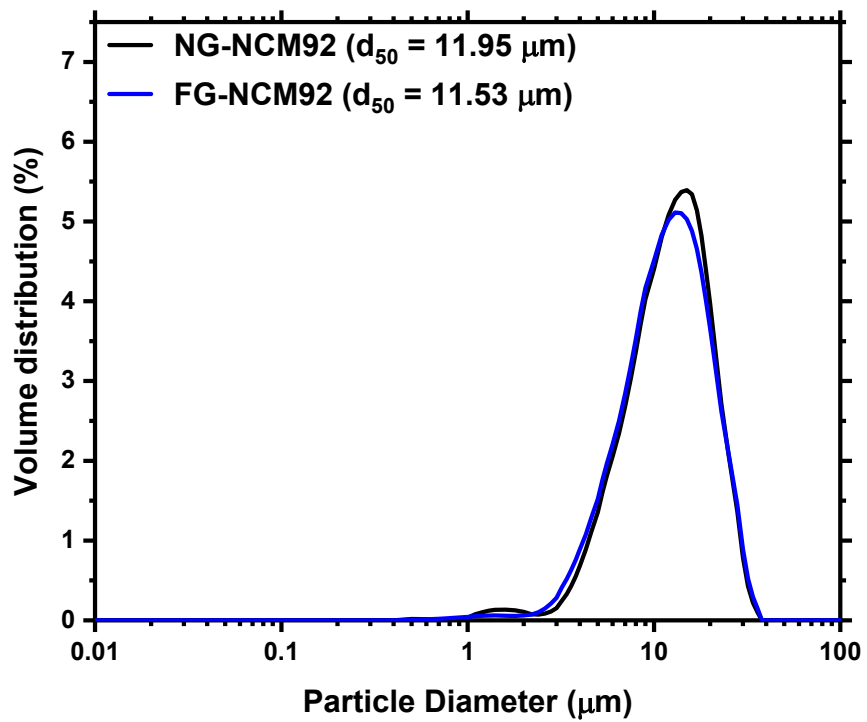




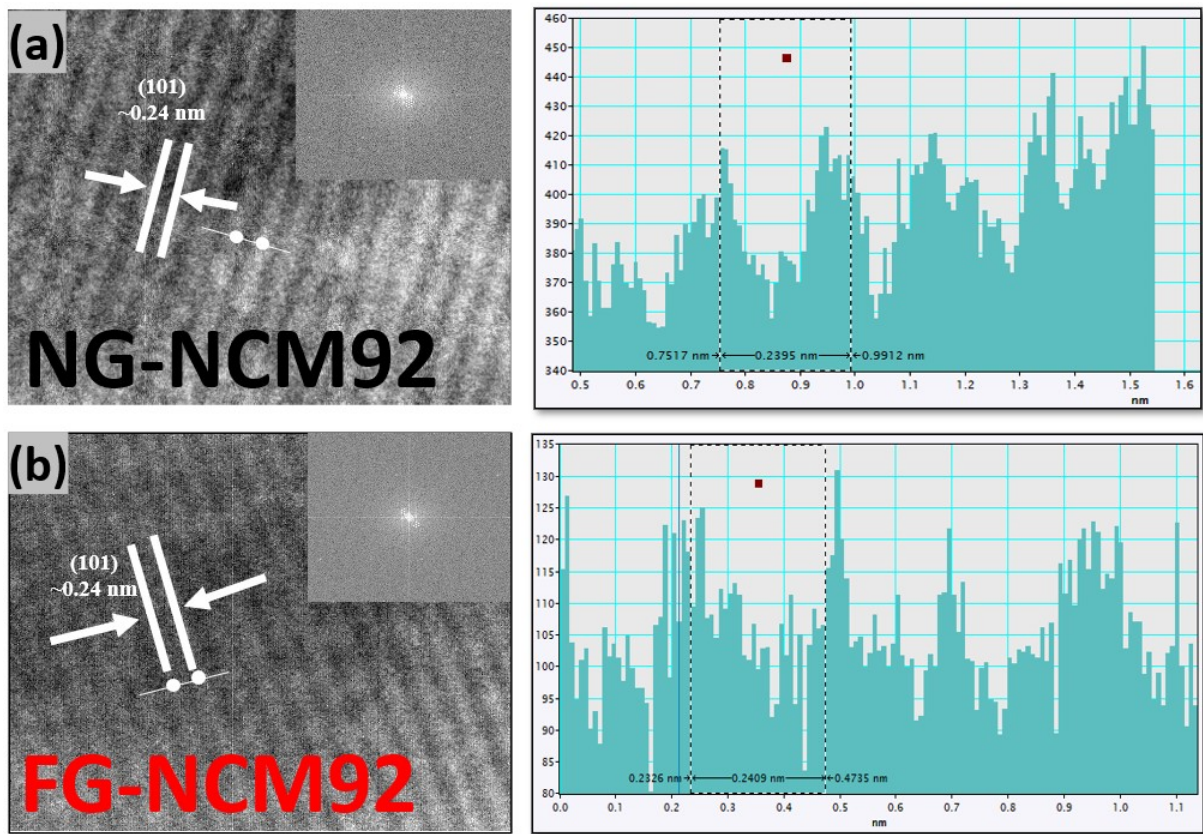
**Fig. S8.** BET specific surface area and pore size distribution (shown in the inset) of NG-NCM92 and FG-NCM92 layered oxide cathode materials.



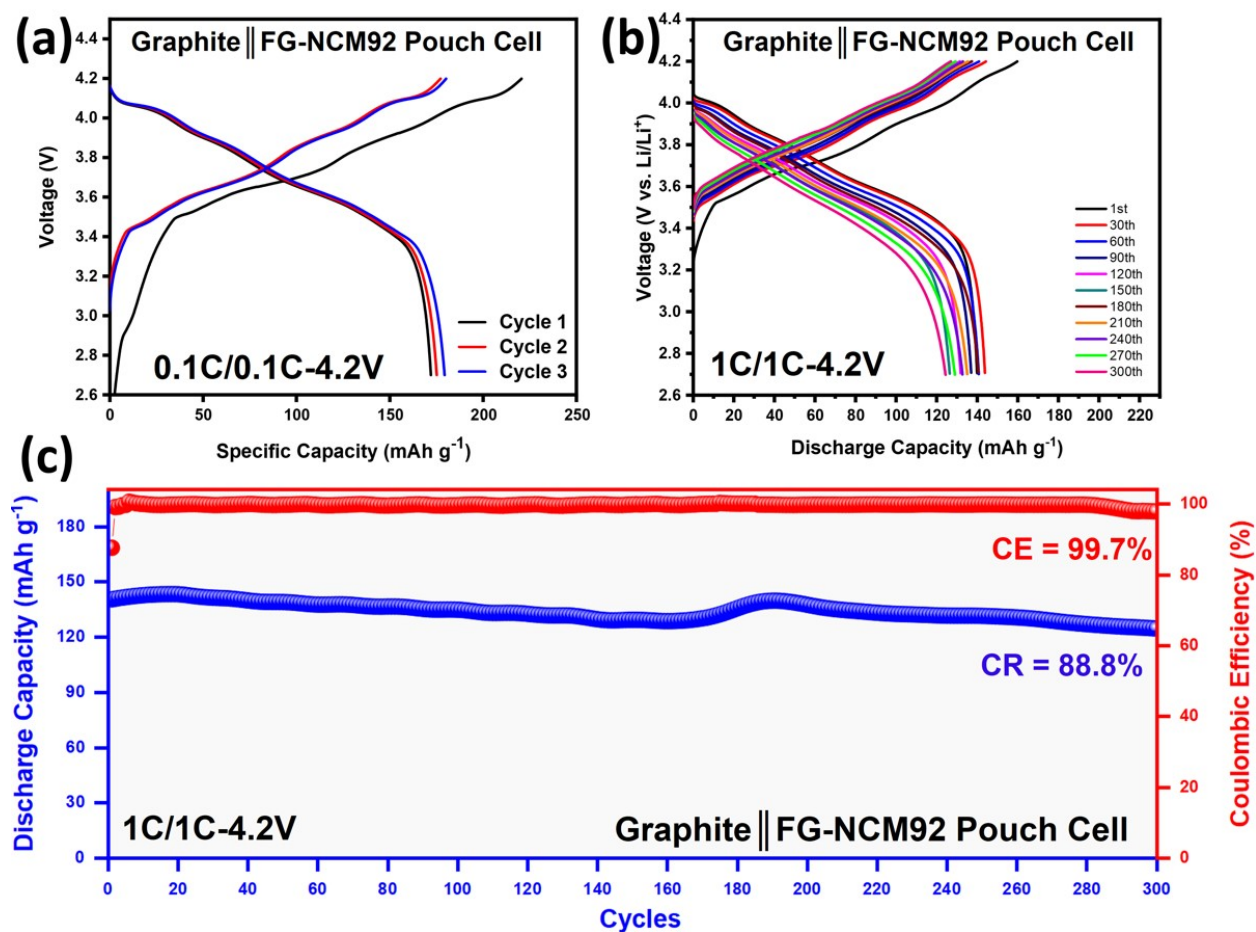
**Fig. S9.** SEM-EDS-line-scanning of Ni, Co, Mn for FG-NCM92 cathode.



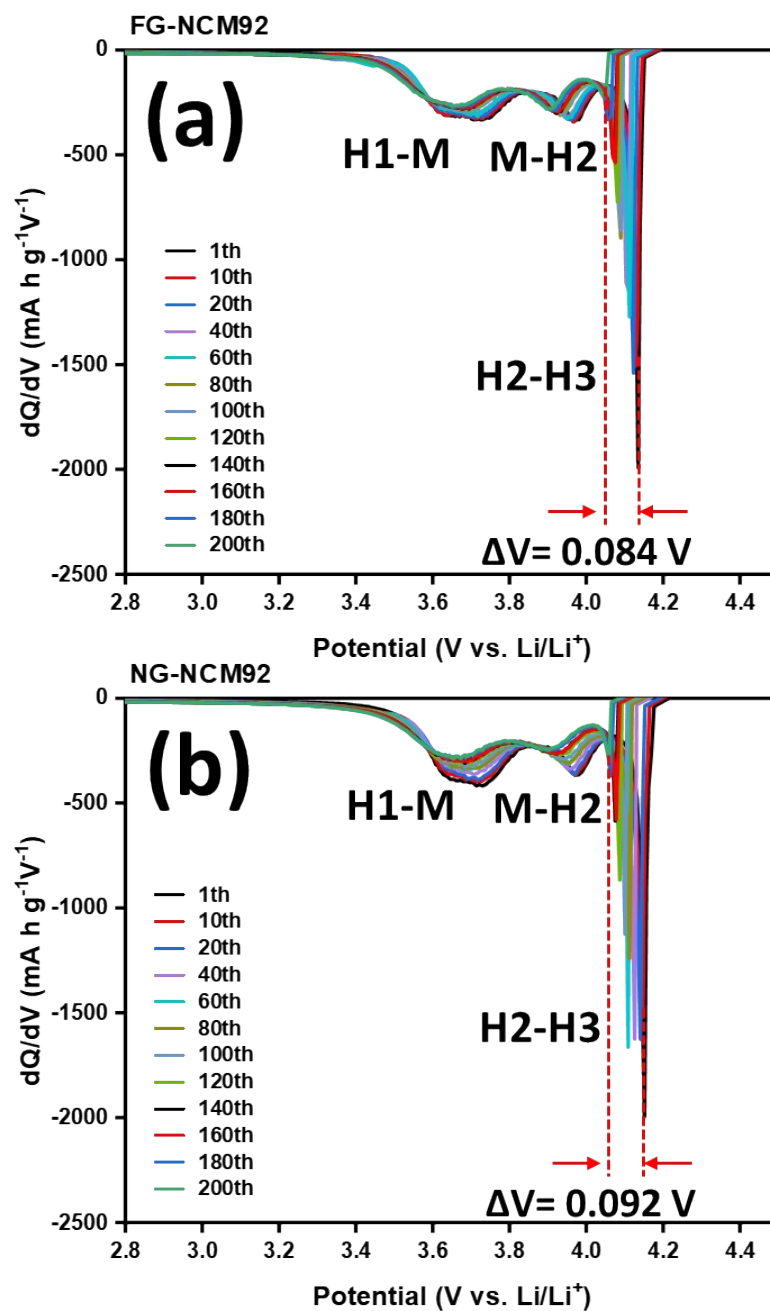
**Fig. S10.** Particle size distribution of NG-NCM92 (black line) and FG-NCM92 (blue line) cathode oxide materials.



**Fig. S11.** The FFT and IFFT analysis of (a) NG-NCM92, and (b) FG-NCM92 layered oxide cathode materials.

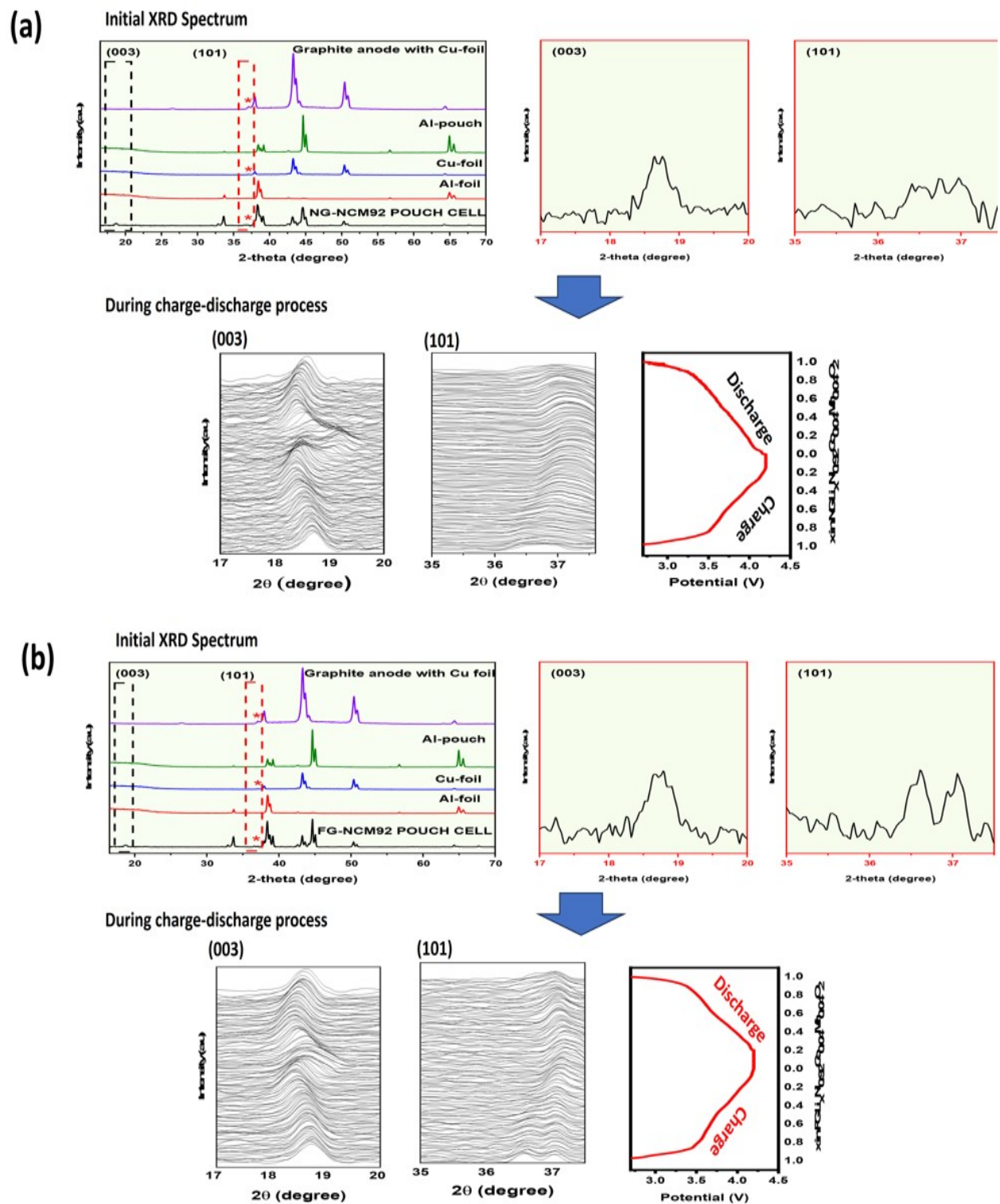


**Fig. S12.** The electrochemical performance of FG-NCM92//Graphite Pouch cells: (a). The charge and discharge profiles at 0.1C/0.1C, (b). The charge and discharge profiles at 1C/1C, (c). Long-term cycling performance of FG-NCM92//Graphite based on Pouch-type full-cells (Size: 3 × 5 cm<sup>2</sup>).



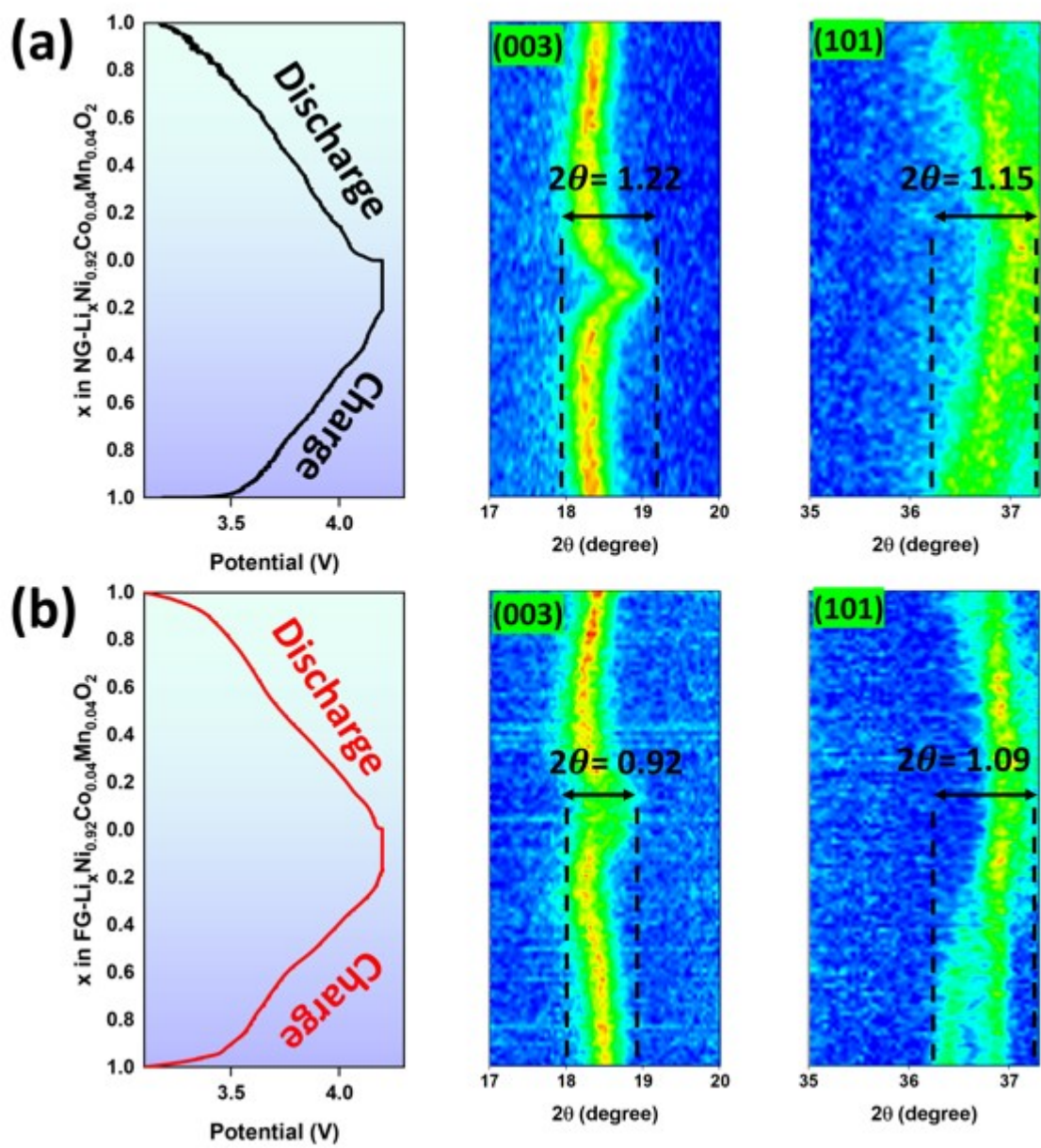
**Fig. S13.** Differential capacity ( $dQ/dV$ ) curves of (a) FG-NCM92, and (b) NG-NCM92 cathodes at 1C/1C in the voltage window of 2.8–4.3 V at RT for 200 cycles.



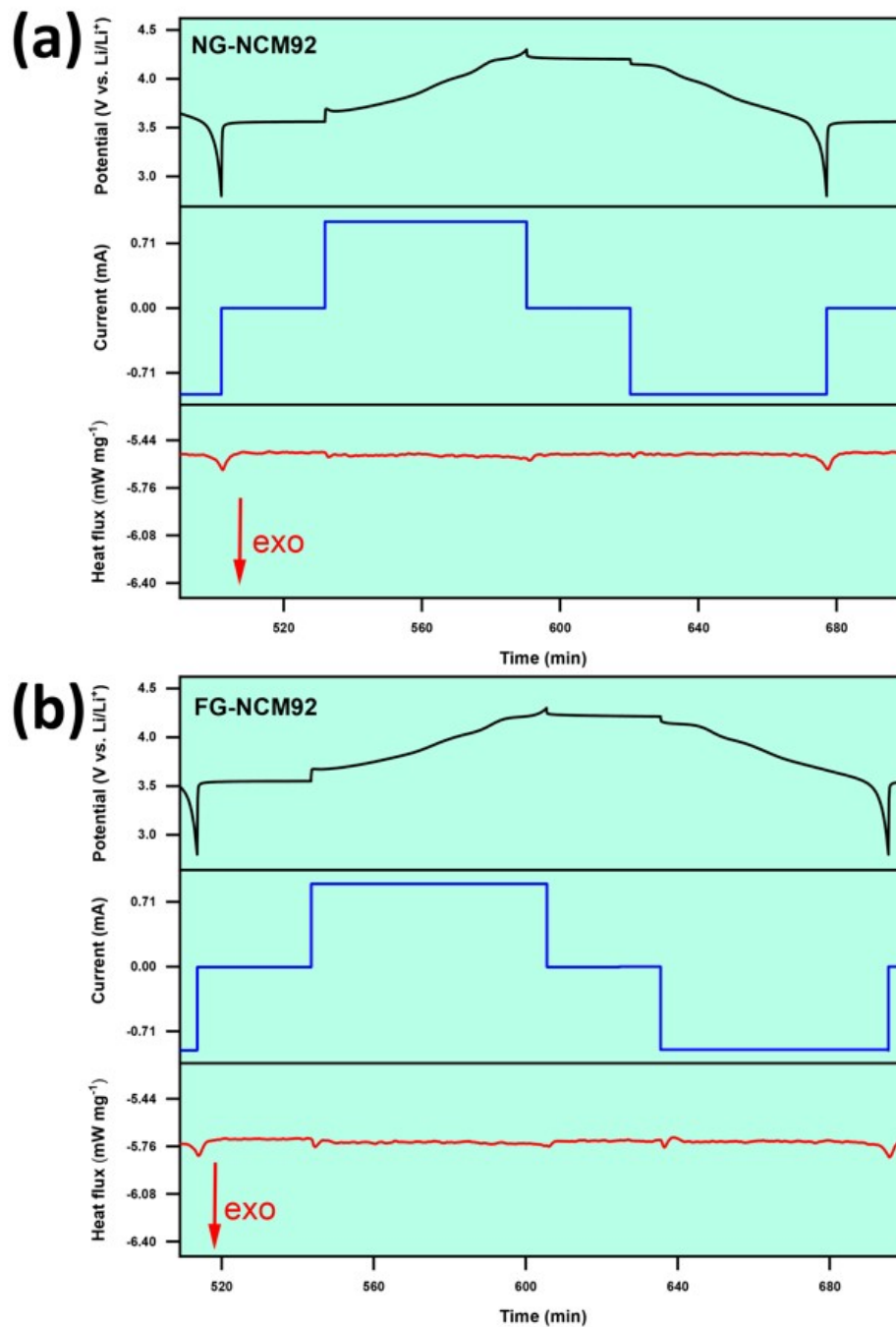


**Fig. S14.** Observed initial in-situ XRD pattern and XRD pattern at selected  $2\theta$  regions during initial cycles of charge/discharge process for (a) NG-NCM92, and (b) FG-NCM92 electrodes, respectively.

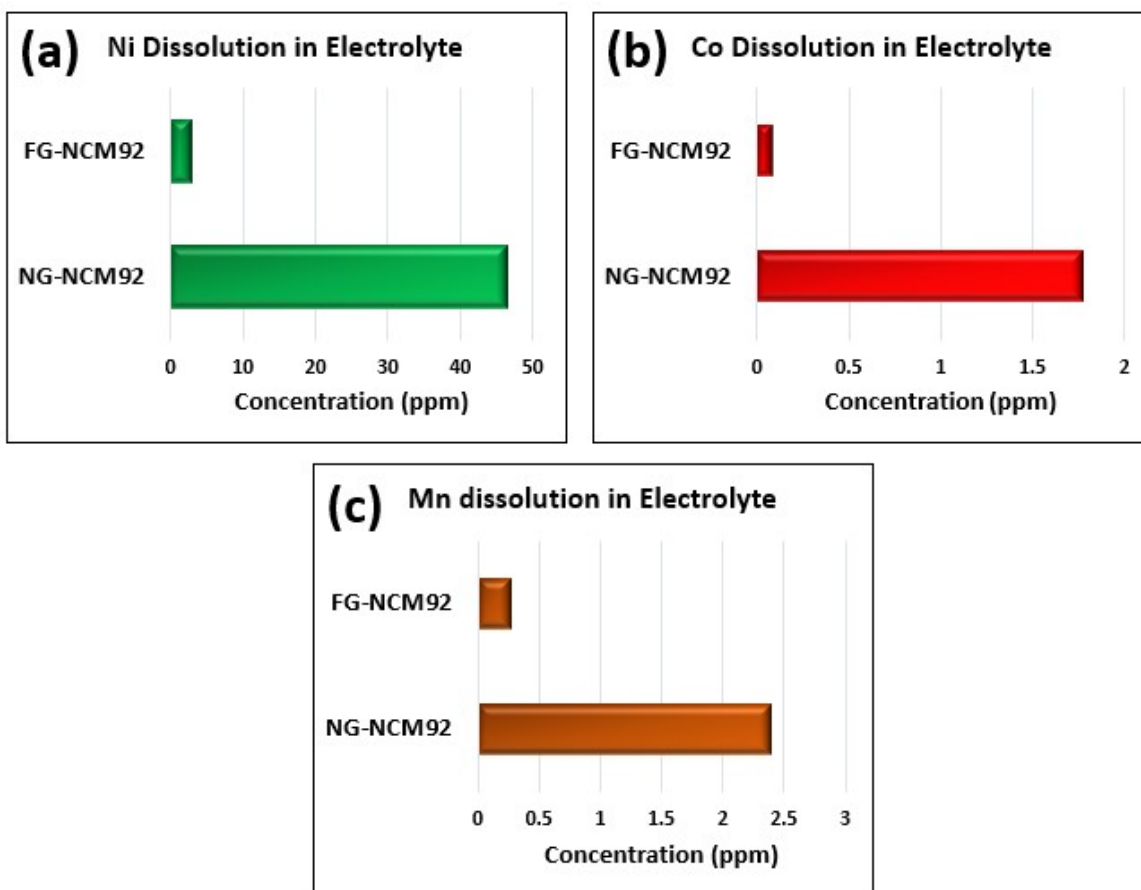




**Fig. S15.** In-situ XRD patterns at selected  $2\theta$  regions during second cycles for (a). NG-NCM92, (b). FG-NCM92 samples, respectively.



**Fig. S16.** Operando micro-calorimetry results for CR2032 coin cells containing: (a) NG-NCM92, and (b) FG-NCM92 electrodes at 1C/1C (in voltage window of 2.8–4.3 V (vs. Li/Li<sup>+</sup>)) in isothermal conditions at 35°C with heating rate of 0.5°C min<sup>-1</sup> (Potential (black line), current (blue line), and heat flux (red line)).



**Fig. S17.** The transition metal dissolution properties of NG-NCM92 and FG-NCM92 cathodes, which were measured by inductively coupled plasma mass spectrometry (ICP-MS): (a). Ni, (b). Co, (c). Mn.

**Table S1.** Rietveld refinement results of NG-NCM92 and FG-NCM92 layered oxide materials.

Sample	a (Å)	c (Å)	V (Å <sup>3</sup> )	c/a	FWHM of (003)	R = I <sub>(003)</sub> /I <sub>(104)</sub>	Crystallite size (nm)	R <sub>wp</sub> <sup>d</sup> (%)	GoF <sup>e</sup>
NG-NCM92	2.874	14.193	101.544	4.938	0.162	1.78	49.62	3.66	1.84
FG-NCM92	2.873	14.195	101.451	4.942	0.156	1.82	51.46	3.92	2.07

**Table S2.** The in-situ XRD lattice parameters for NG-NCM92 and FG-NCM92 layered oxide materials during initial cycles.

	NG-NCM92			FG-NCM92		
	Lithiated state	Delithiated state	$\Delta_{\max}$ (%)	Lithiated state	Delithiated state	$\Delta_{\max}$ (%)
	<i>c</i> -axis (Å)	14.527	14.387	0.97	14.433	14.361
<i>a</i> -axis (Å)	2.850	2.833	0.61	2.846	2.831	0.35
Unit-cell volume (Å <sup>3</sup> )	101.50	100.11	1.37	100.74	99.873	0.86

**Table S3.** Fitting results of EIS spectra of NG-NCM92 and FG-NCM92 electrodes before cycling and after 200 cycles.

Sample	Before Cycling		After 200 Cycles	
	NG-NCM92	FG-NCM92	NG-NCM92	FG-NCM92
R <sub>b</sub> (Ω)	3.76	1.55	4.63	3.05
R <sub>CEI</sub> (Ω)	—	—	6.17	6.10
R <sub>int</sub> (Ω)	—	—	20.50	18.00
R <sub>ct</sub> (Ω)	138.50	118.50	610.00	581.80

**Table S4.** The Total Exothermic Heat Generation ( $Q_t$ ) and heat reduction of the NG-NCM92//Li coin-type cell and FG-NCM92//Li coin-type during charge/discharge processes at 35 °C.

<b>Electrode</b>	<b>Total heat generation</b>		<b>Heat reduction</b>	
	<b><math>Q_t</math> (J g<sup>-1</sup>)</b>		<b>(%) vs. NG-NCM92</b>	
	Charge	Discharge	Charge	Discharge
NG-NCM92	-18.94	-26.59	—	—
FG-NCM92	-17.57	-23.04	7.2	13.4

\*A negative  $Q_t$  value indicates an exothermic heat release