

Tin-Based Composite Oxide Confined by Reduced Graphene Oxide as a High-Rate Anode for Sodium-Ion Capacitors

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Table S1. Comparison of XRD Peak FWHM and Crystallite Size between $\text{Sn}_2\text{P}_2\text{O}_7$ and $\text{Sn}_2\text{P}_2\text{O}_7/\text{rGO}$

Peak Position (2θ)	FWHM ($^\circ$) $\text{Sn}_2\text{P}_2\text{O}_7$	FWHM ($^\circ$) $\text{Sn}_2\text{P}_2\text{O}_7/\text{rGO}$	Crystallite Size (nm) $\text{Sn}_2\text{P}_2\text{O}_7$	Crystallite Size (nm) $\text{Sn}_2\text{P}_2\text{O}_7/\text{rGO}$
20.2666	0.0513	0.0107	155.6697	
20.4805	0.0182	0.5395		14.7986
21.3751	0.0129	0.0156		
21.6668	0.0113	0.2683		29.8116
31.2935	0.2282	0.3082		
36.0582	0.2622	0.0194	31.5136	
36.0971	0.0160	0.3171		

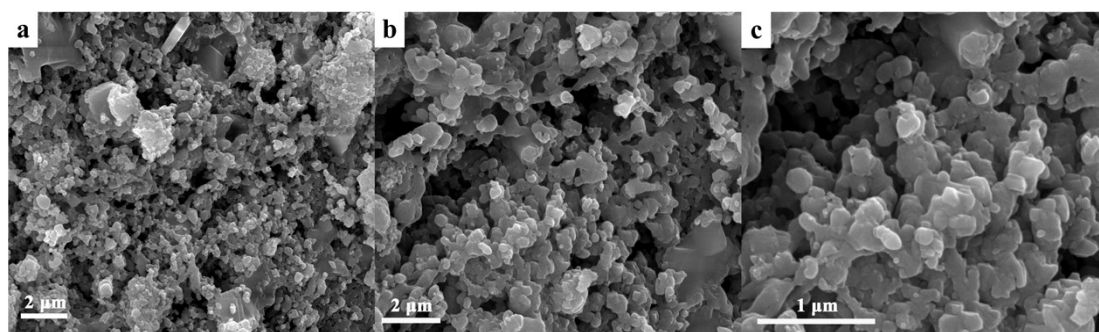


Figure S1. SEM images of $\text{Sn}_2\text{P}_2\text{O}_7$.

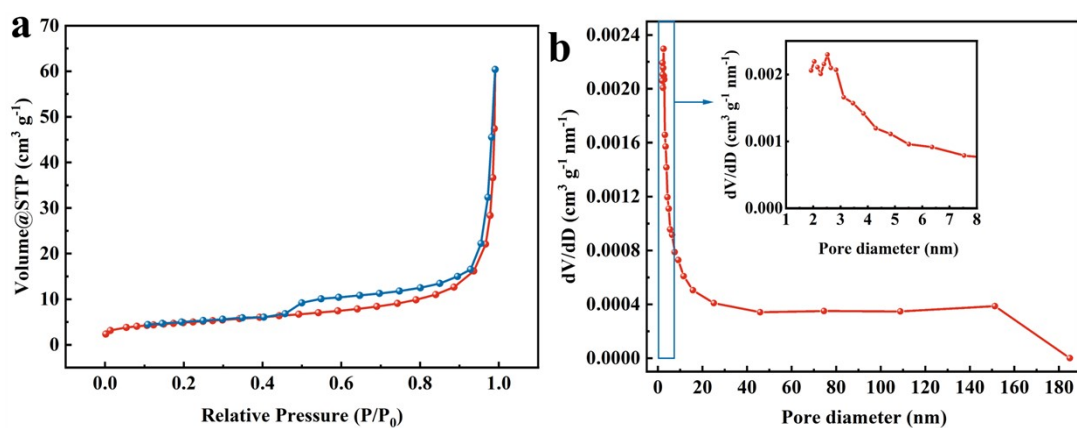


Figure S2. N_2 adsorption-desorption isotherms and corresponding pore size distribution plot of $\text{Sn}_2\text{P}_2\text{O}_7/\text{rGO}$.

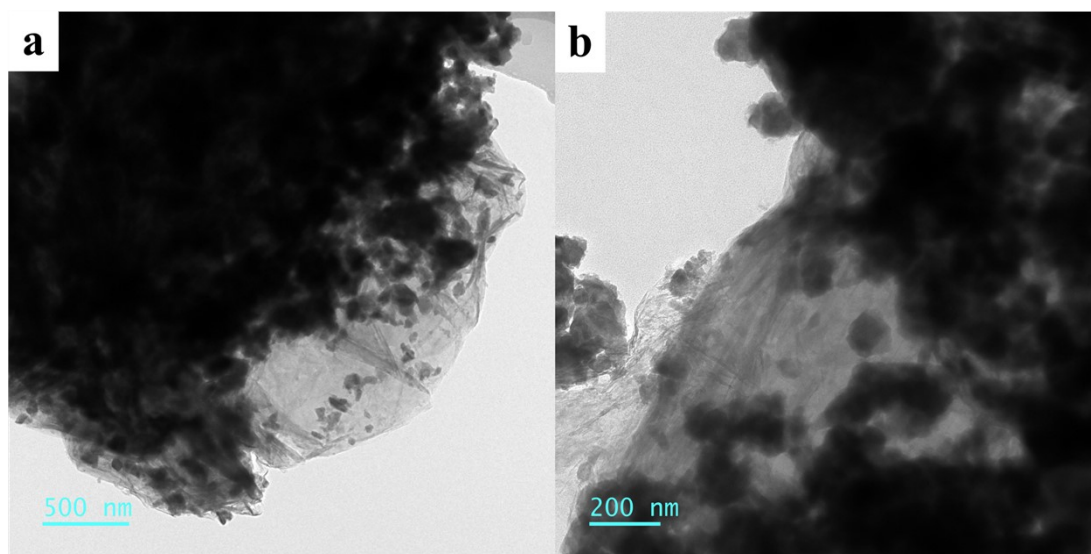


Figure S3. TEM images of $\text{Sn}_2\text{P}_2\text{O}_7/\text{rGO}$ composite.

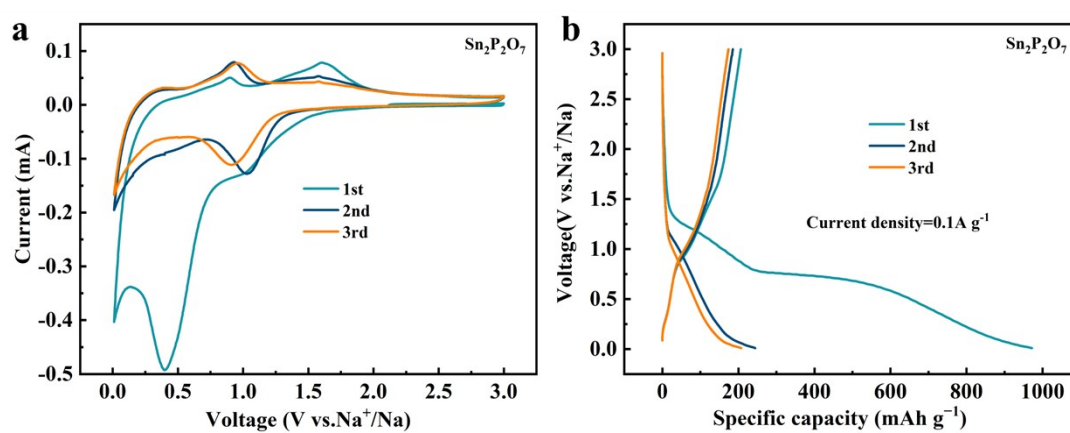


Figure S4. (a) CV curves at a scan rate of 0.1 mV s^{-1} of $\text{Sn}_2\text{P}_2\text{O}_7$ anode. (b) Charge/discharge profiles of the first three cycles at the current density of 0.1 A g^{-1} of $\text{Sn}_2\text{P}_2\text{O}_7$ anode.

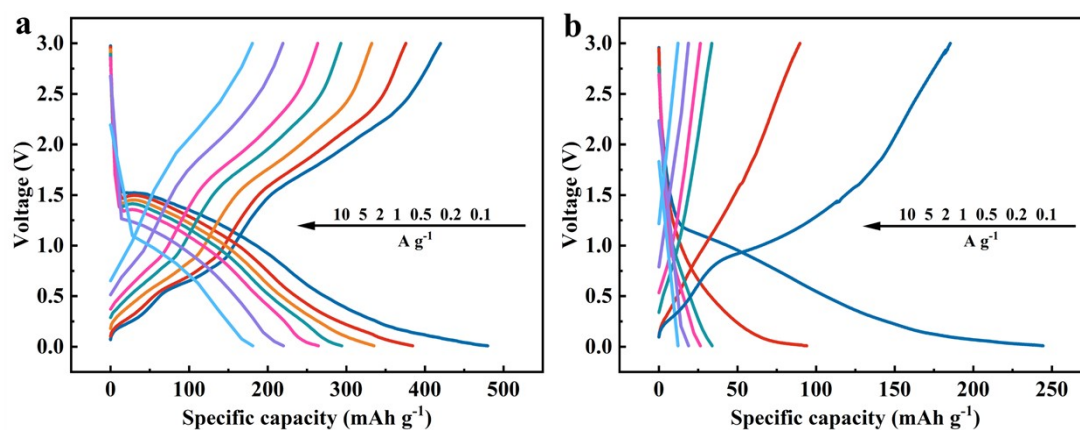


Figure S5. Discharge/charge profiles of (a) $\text{Sn}_2\text{P}_2\text{O}_7/\text{rGO}$ and (b) $\text{Sn}_2\text{P}_2\text{O}_7$ anodes at different current densities from 0.1 to 10 A g^{-1} .

Table S2. Comparison of rate performance of Sn-based compound/carbon composite anode for SIBs.

Anode materials	Specific capacity/current density ($\text{mA h g}^{-1}/\text{A g}^{-1}$)	Ref.
$\text{Sn}_2\text{P}_2\text{O}_7/\text{RGO}$	(480/0.05) (170/10)	[1]
$\text{SnP}_2\text{O}_7/\text{rGO}$	(254.7/0.1) (91.1/10)	[2]
$\text{SnP}_2\text{O}_7@\text{N-C}$	(400/0.1) (210/5)	[3]
$\text{SnP}_2\text{O}_7/\text{NG}$	(423/0.1) (206/2)	[4]
$\text{SnP}_2\text{O}_7@\text{C}$	(403/0.2) (110/2)	[5]
$\text{SnP}_2\text{O}_7@\text{C/CN}$	(342.3/0.1) (106.0/5)	[6]
$\text{Sn}_2\text{P}_2\text{O}_7/\text{rGO}$	(433.3/0.1) (185.7/10)	This work

Table S3. Comparison of cycling stability of Sn-based anode materials for SIBs.

Anode materials	Capacity (mA h g ⁻¹ /A g ⁻¹)	Cycle life.	Ref.
SnP ₂ O ₇ @C/CN	91.5/2	800	[6]
SnS-AB	290/0.1	50	[7]
Sn/Sb@SSO@PCFs	440/0.05	450	[8]
Sn ₄ P ₃ @CNF	336/1	500	[9]
Sn ₃ Se ₅ @PPy	187.8/5	300	[10]
Sn-Ni@NC	150.3/1	200	[11]
Sn₂P₂O₇/rGO	165.4/1	1000	This work

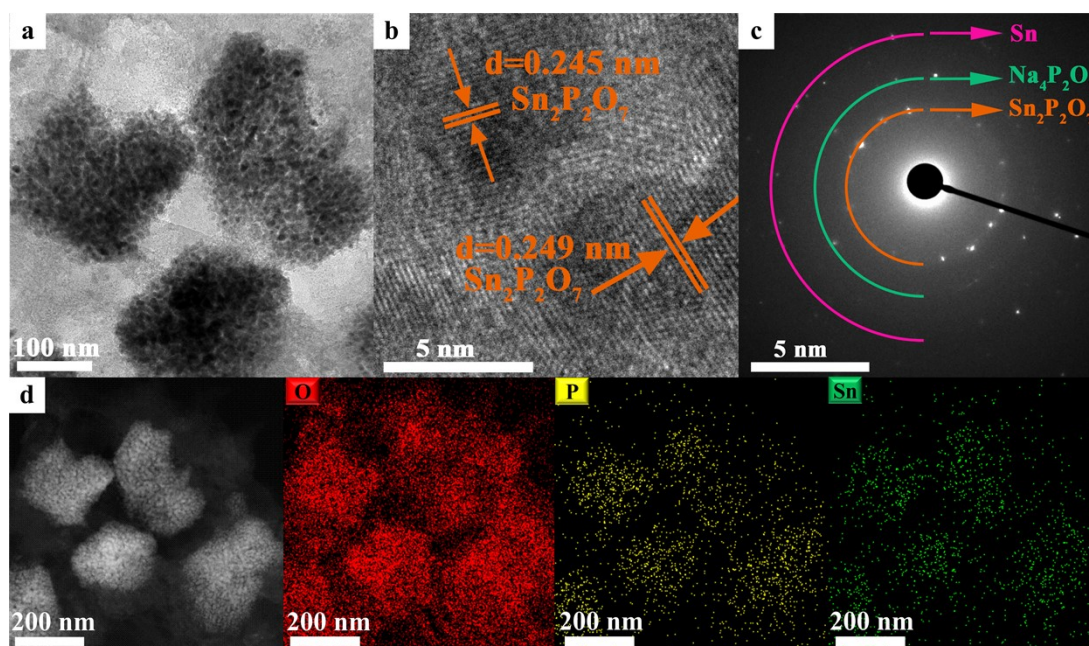


Figure S6. (a) *Ex-situ* TEM images, (b) HRTEM images, (c) SAED patterns and (d) EDS elemental mappings of Sn₂P₂O₇ anode after 1000 cycles at a current density of 1 A g⁻¹.

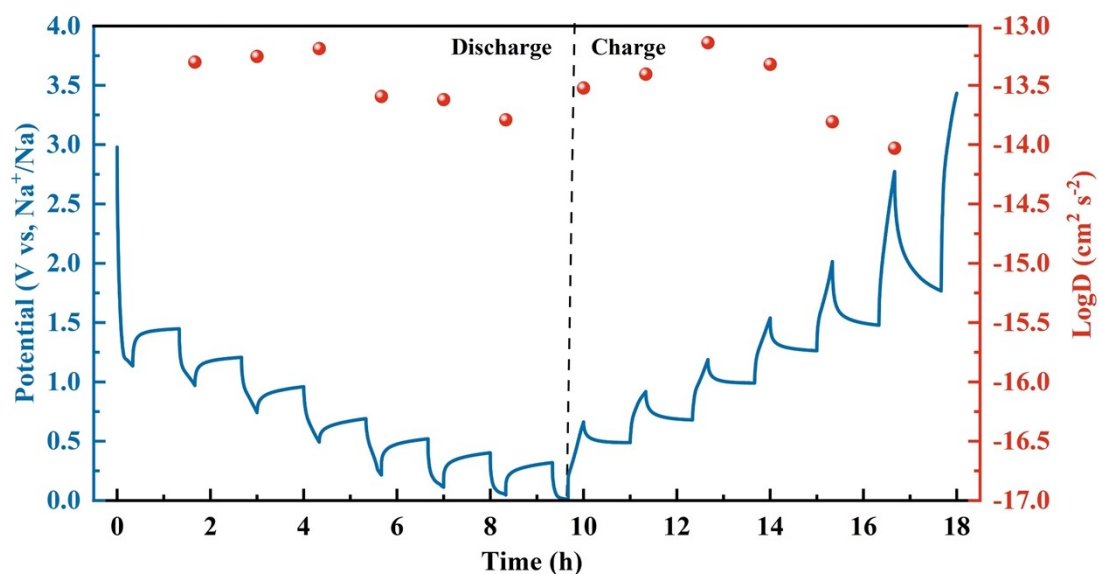


Figure S7. GITT curve and diffusivity versus state of discharge and charge during the selected second cycle of $\text{Sn}_2\text{P}_2\text{O}_7$ anode. (The D_{Na^+} of $\text{Sn}_2\text{P}_2\text{O}_7$ anode ranges from $1 \times 10^{-13.19} \text{ cm}^2 \text{ s}^{-1}$ to $1 \times 10^{-13.79} \text{ cm}^2 \text{ s}^{-1}$ during discharge process, and from $1 \times 10^{-13.14} \text{ cm}^2 \text{ s}^{-1}$ to $1 \times 10^{-14.03} \text{ cm}^2 \text{ s}^{-1}$ during charge process.)

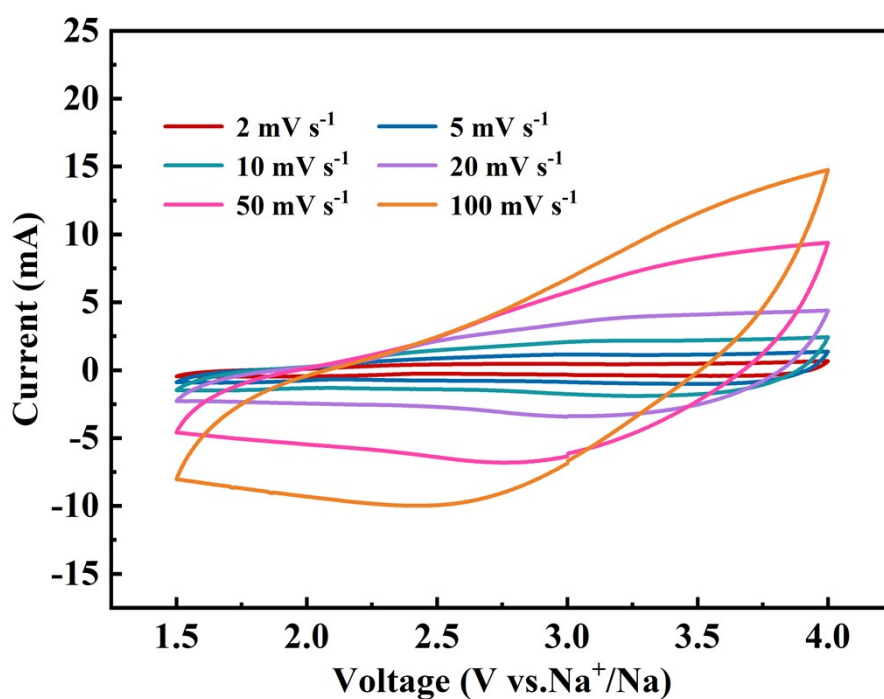


Figure S8. Half-cell performance of the AC vs. Na metal, tested between 1.5–4.0 V. CV curves at different scan rates from 2 mV s^{-1} to 100 mV s^{-1} .

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