### Supporting Information

# Aggregation-Induced Emission Silence–Mediated Pathogen Detection Using a Rapidly Degradable Nanographene-Embedded Polymersome

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#### **Preparation of nanographene**

A total of eight steps were sequentially performed to obtain six precompounds and nanographene (nanoG; Figure S19). Detailed descriptions of each step follow.

Synthesis of compound S1: A dual-neck flask was charged with 4-bromobenzonitrile (4.97 g, 27.3 mmol), PPh<sub>3</sub> (0.17 g, 2.5 mol%), PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub> (0.96 g, 5 mol%), CuI (0.17 g, 2.5 mol%), and trimethylsilylacetylene (4.05 g, 41.2 mmol). The flask was degassed and filled with nitrogen. A mixture of THF (60 mL) and Et<sub>3</sub>N (5.9 mL) was added using a syringe and allowed to react for 16 h at room temperature. A yellow solid appeared, and the residue was treated with MeOH/K<sub>2</sub>CO<sub>3</sub> for isolation and then purified through column chromatography on silica gel and eluted with CH<sub>2</sub>Cl<sub>2</sub>-hexane to obtain S1, a colorless solid (2.67 g, 77%).

Synthesis of compound S2: A flask was charged with compound S1 (1 g, 7.87 mmol) and 4-bromobenzonitrile (1.43 g, 7.87 mmol), PPh<sub>3</sub> (41 mg, 0.16 mmol), Pd(PPh<sub>3</sub>)<sub>2</sub>Cl<sub>2</sub> (110 mg, 0.16 mmol), and Et<sub>3</sub>N (30 mL). The mixture was stirred for 10 min with Ar bubbling through the solution, and CuI (60 mg, 0.31 mmol) was then added. The mixture was heated to reflux by stirring under nitrogen for 2 h and filtered. The filtrate was purified using column chromatography on silica gel (hexane/CH<sub>2</sub>Cl<sub>2</sub>, 3:7) to obtain a white solid product (1.3 g, 72%)

Synthesis of compound S3: A solution of 8.0 g (40.0 mmol) of pyrene and 200 mL of tertbutyl chloride was added to 8.0 g (60.0 mmol) of powdered AlCl<sub>3</sub> at 0 °C. After the reaction mixture had been stirred at room temperature for 3 h, it was poured into a large amount of ice/water mixture and extracted with  $CH_2Cl_2$  (twice with 250 mL). The combined  $CH_2Cl_2$ extracts were washed with water (twice with 200 mL) and dried with MgSO<sub>4</sub>, and the solvent was evaporated in vacuum to leave a residue, which was washed with alcohol to yield a colorless solid. The yield was 78%.

Synthesis of compound S4: A solution of compound S3 (1.6 mmol) in  $CH_2Cl_2$  (6.4 mL) and  $CH_3CN$  (6.4 mL) was produced, and  $NaIO_4$  (2.8 g, 13.086 mmol),  $H_2O$  (8 mL), and  $RuCl_3$ ,  $xH_2O$  (0.04 g, 0.192 mmol) were added. The dark brown suspension was heated at 30–40 °C overnight. The reaction mixture was poured into 500 mL of water and extracted with  $CH_2Cl_2$ . The combined extracts were washed with water and brine and dried over anhydrous MgSO<sub>4</sub>, and the solvent was evaporated in vacuo. The crude product was purified through column chromatography to produce the desired compound, a bright orange crystal, in yield of 40%.

Synthesis of compound S5: Iron pentacarbonyl (0.92 mL, 6.83 mmol) was added to a refluxing mixture of 4-*tert* butyl benzyl bromide (2.414 g, 13.14 mmol), sodium hydroxide (2.27 g), benzyltriethylammonium chloride (0.1 g), deionized water (1.3 mL), and dichloromethane (31.5 mL) under a nitrogen atmosphere. The reaction mixture was refluxed with vigorous stirring overnight and extracted with  $CH_2Cl_2$  (twice with 250 mL). The combined  $CH_2Cl_2$  extracts were washed with water (twice with 200 mL) and dried over magnesium sulfate, and the solvent was evaporated under vacuum. The resulting solid was purified by chromatography on silica gel with a mixture of dichloromethane and petroleum ether (3:1) as the eluent. The yield was 65%.

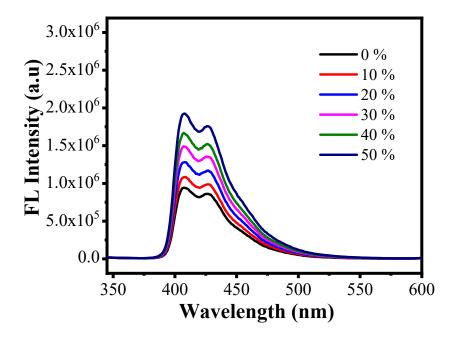
*Synthesis of compound* **S6**: To a suspension of 2,7-di-tert-butyl-pyrene-4,5,9,10-tetraone (118 mg, 0.32 mmol) and 1,3-bis(4-tert butylphenyl) propan-2-one (204 mg, 0.68 mmol) in dry ethanol (65 mL), DBU (0.18 mL, 1.21 mmol) was added dropwise under argon atmosphere. The mixture was stirred for 40 min at 78 °C. The formed solid was collected through vacuum filtration, washed with cold ethanol (twice with 2 mL), and dried under vacuum to yield a pale green solid. The yield was 70%.

Synthesis of compound NanoG: Compound S6 (500 mg, 0.528 mmol) and Compound S2 (250 mg, 1.05 mmol) were added to a dry Schlenk flask equipped with a magnetic stir bar, and 2 mL of diphenylether was added to the flask. The reaction mixture was refluxed for 6 h at 260 °C. The crude product was subjected to column chromatography (DCM/hexane = 2:3 as an eluent). The yield of nanoG was 38%.

Types	Materials	Pagthogens	<b>Detection method</b>	LOD	References
Nanoparticle-based biosensors	MNP-2nd DNA probe/target DNA/1st DNA probe-Au-NPs- barcode DNA	Salmonella	Fluorescence	1 ng/mL	[1]
	Dye-doped silica NPs	<i>E. coli</i> O157:H7	Flow cytometry	1 CFU/mL	[2]
	PGNF	E. coli	Electrochemistry	10 CFU/mL	[3]
	GQD-AuNP	Staphylococcus aureus	Fluorescence	1 nM	[4]
	Tyr/MNPs-CNTs/GCE	E. coli	FIA system	10 CFU/mL	[5]
	MNP-SMCC-DNA probe <sub>3</sub> -target DNA Fluorescence DNA probe <sub>2</sub> - AuNP-DNA probe <sub>1</sub>	Shigella species	Fluorescence	10 <sup>2</sup> CFU/mL	[6]
	FDNPs	E. coli	Fluorescence	1.1×10 <sup>5</sup> CFU/μL	[7]
	TPE-DNA@GO	SARS-CoV-2	AIE recovery	Ct value of 20	[8]
	GSH-AuAg NCs	Baumannii	AIE silence	2.3×10 <sup>3</sup> CFU/mL	[9]
	MSN-PGEDA-CB[7]-TPE	Staphylococcus aureus, E. coli	AIE silence	2.5×10 <sup>6</sup> CFU/mL	[10]
	NPGHP	E.coli, Pseudomonas aeruginosa, Staphylococcus aureus, Bacillus subtilis	AIE effect	1×10 <sup>4</sup> CFU/mL	[11]
	NanoGs-embedded PLGA polymersome	Methicillin-resistant Staphylococcus aureus, E. coli	AIE silence	2.709×10 <sup>8</sup> CFU/mL	This work

Table S1. Nanoparticle-based AIE biosensors for pathogenic bacteria detection.

MNP: magnetic nanoparticle; DNA: deoxyribonucleic acid; NP: nanoparticle; PGNF: Ptnanoparticle-coated gold nanoporous film; GQD: graphene quantum dot; Tyr: tyrosinase; CNT: carbon nanotube; GCE: glassy carbon electrode; SMCC: sulfosuccinimidyl 4nmaleimidomethyl cyclohexane-1- carboxylate; FDNP: fluorescent dextran nanoparticle; TPE: tetraphenylethene; GO: graphene oxide nanosheet; GSH: L-glutathione; NC: nanocluster; MSN: mesoporous silica nanoparticle; PGEDA: ethylenediamine-modified polyglycerol methacrylate; CB[7]: cucurbit[7]uril; NPGHP: nanoengineered peptide-grafted hyperbranched polymer; FIA: flow injection assay.



**Figure S1.** Fluorescent spectra of nanoG dispersed in cosolvent at various water–THF ratios. Increasing the THF content increased the AIE effect of the nanoG.

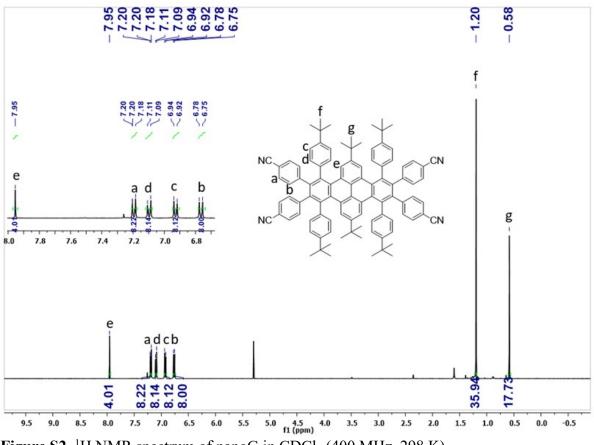
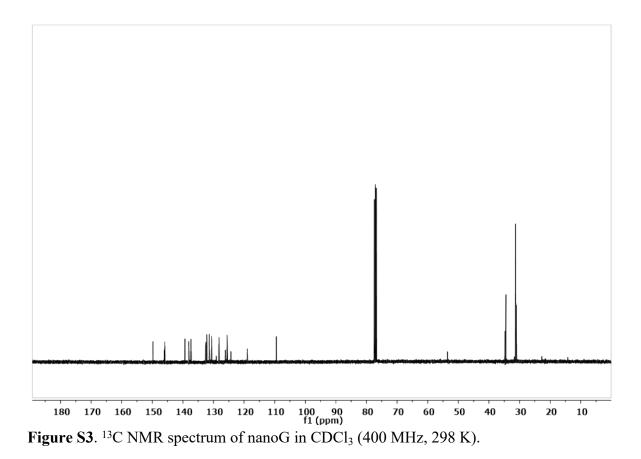


Figure S2. <sup>1</sup>H NMR spectrum of nanoG in CDCl<sub>3</sub> (400 MHz, 298 K).



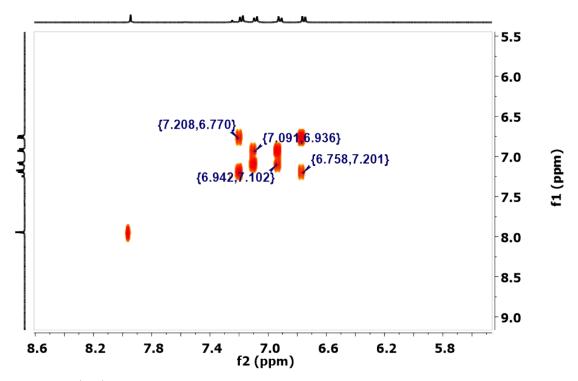


Figure S4. <sup>1</sup>H-<sup>1</sup>H COSY spectrum of nanoG in CDCl<sub>3</sub> (400 MHz, 298 K).

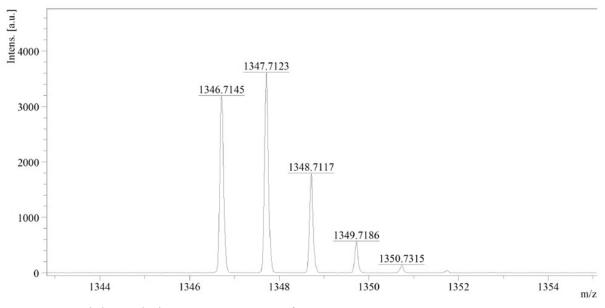


Figure S5. High-resolution mass spectrum of nanoG.

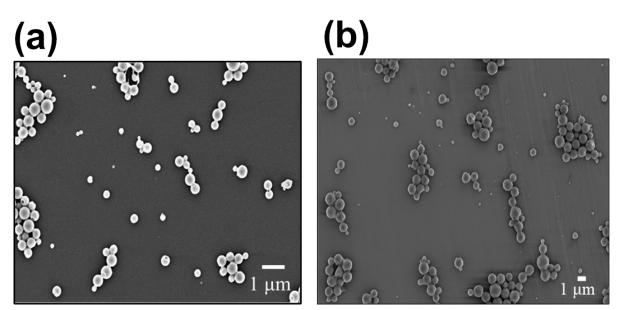
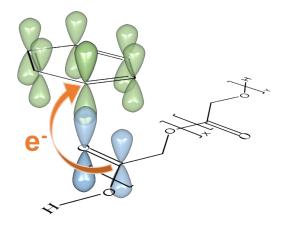


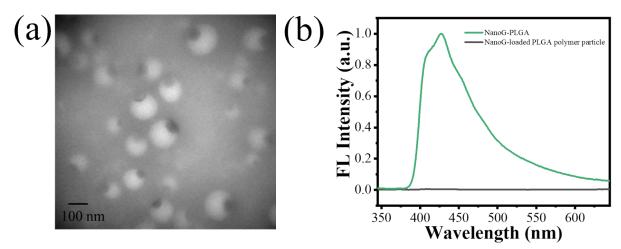
Figure S6. SEM images of (c) PLGA and (d) nanoG-embedded PLGA polymersomes.

	Zeta potential (mV)	Size (nm)
PLGA	-15.43	825
NanoG-PLGA	-19.92	865

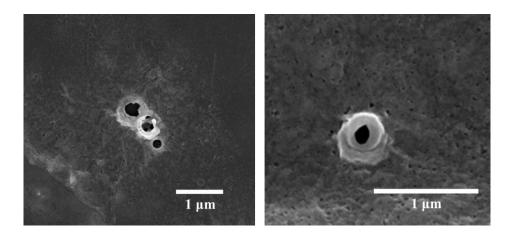
**Figure S7.** Dynamic light scattering and zeta potential of PLGA polymersome with and without nanoG.



**Figure S8.** Illustration of the  $\pi$ - $\pi$  interaction between aromatic benzene and the carboxylic group of PLGA, which enables electron donation from PLGA to the aromatic-ring-rich nanoG.

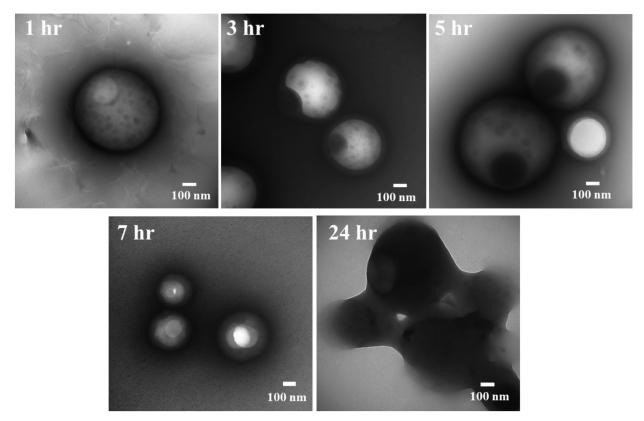


**Figure S9**. (a) The TEM image of nanoG-loaded PLGA polymer particle prepared by single emulsion method. The dark contrast area in the spheres indicated the nanoG aggregation in the polymer particle. (b) Fluorescence spectra of nanoG-embedded PLGA polymersomes and nanoG-loaded PLGA polymer particles upon excitation at 330 nm.

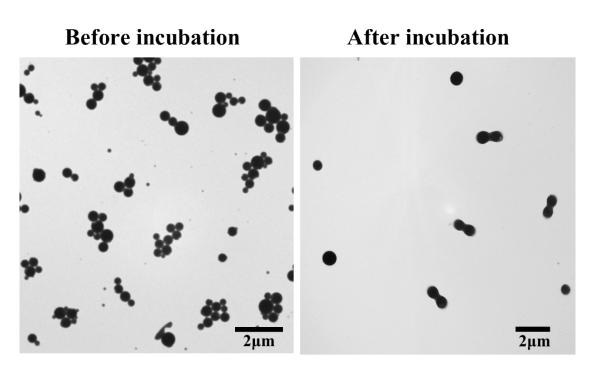


**Figure S10.** SEM images of nanoG-embedded PLGA polymersomes incubated with bacteria for 24h.

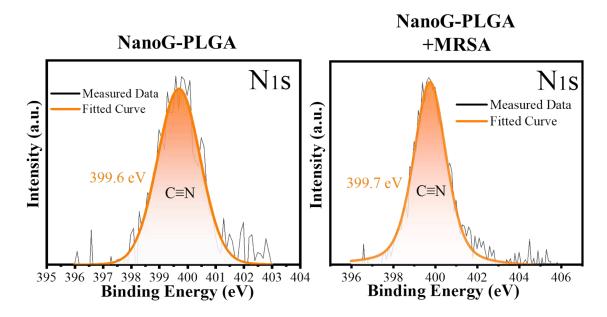
## PLGA+ MRSA



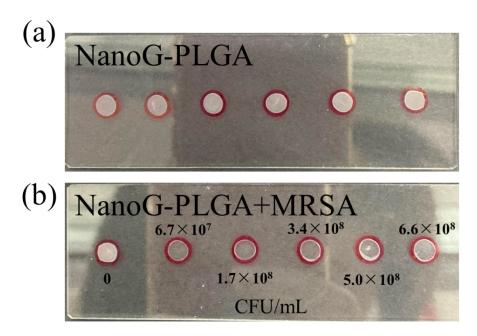
**Figure S11.** TEM images of PLGA polymersome incubated with MRSA at  $6.7 \times 10^7$  CFU/mL at 37 °C for various incubation times (0, 1, 3, 5, or 24 h).



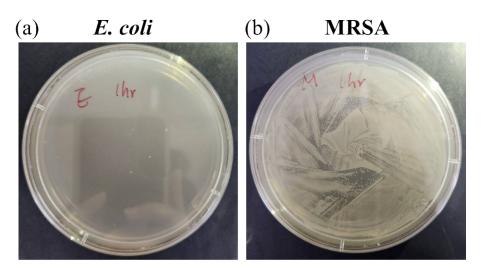
**Figure S12.** TEM images of nanoG-embedded PLGA polymersome dispersed in cell-free medium at 37 °C for 24 h.



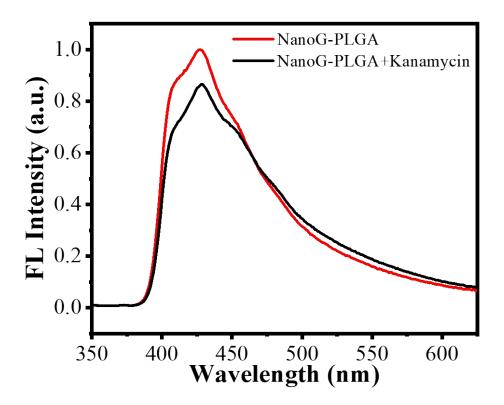
**Figure S13.** XPS analysis results of nanoG-embedded PLGA polymersome before and after incubation with MRSA in  $6.7 \times 10^7$  CFU/mL at 37 °C for 24 h.



**Figure S14.** Photographs of AIE thin films without UV light exposure. (a) Total of 10  $\mu$ L of nanoG-embedded PLGA polymersome was deposited on a glass slide to form AIE thin-film test zones (marked by red circles). (b) AIE thin-film test zones after dropping 10  $\mu$ L of buffer solution containing MRSA at different concentrations (6.7×10<sup>7</sup>, 1.7×10<sup>8</sup>, 3.4×10<sup>8</sup>, 5.0×10<sup>8</sup>, 6.7×10<sup>8</sup> CFU/mL).



**Figure S15.** Colony assays of (a) *E. coli* and (b) MRSA with kanamycin (9  $\mu$ g/mL). A total of 100  $\mu$ L of the bacterial solution containing antibiotic was added to an LB agar plate and incubated at 37 °C for 24 h.



**Figure S16.** Fluorescence spectra of nanoG-embedded PLGA polymersome before and after incubation with kanamycin at 37 °C for 1 h.

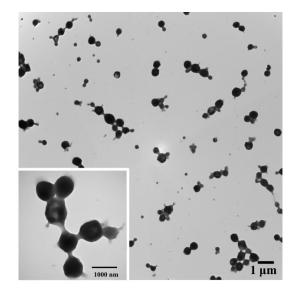
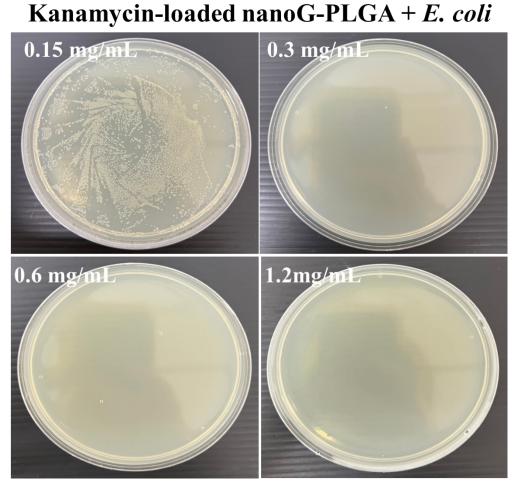


Figure S17. TEM image of kanamycin-loaded nanoG-embedded PLGA polymersome.



**Figure S18.** Colony assay of *E. coli* with kanamycin-loaded nanoG-PLGA at different concentrations. A total of 100  $\mu$ L of the bacterial solution containing antibiotic was added to an LB agar plate and incubated at 37 °C for 24 h.

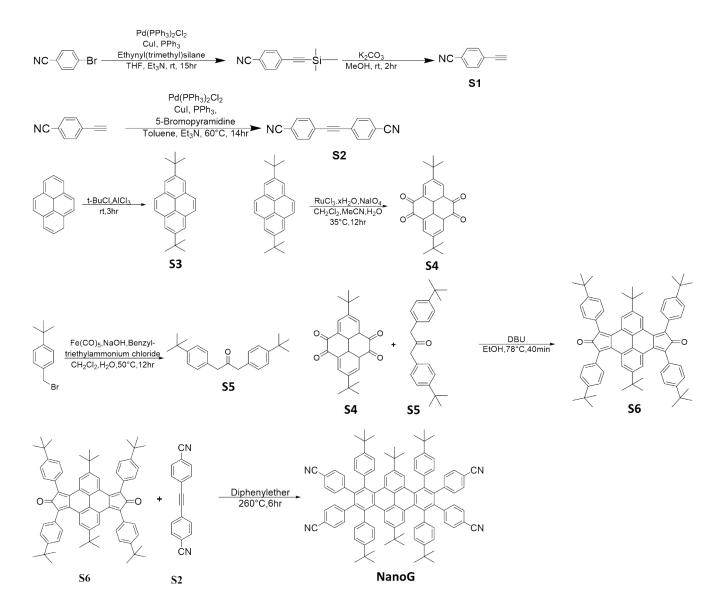


Figure S19. NanoG synthesis procedure.

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