

Supporting Information for

General Synthesis of Magnetic Binary Transition Metal Telluride Nanocrystals

Jingxia Wang^a, Bin Wang^a, Yifen Wang^a, Ruixia Yang^a, Lanfang Wang^a, Fang Wang^a,
Xiaohong, Xu^a, and Yang Liu^{b,*}

^aResearch Institute of Materials Science of Shanxi Normal University & Key
Laboratory of Magnetic Molecules and Magnetic Information Materials of Ministry of
Education, Taiyuan 030031, China.

^bDepartment of Materials Science, Fudan University, Shanghai 200433, China.

*Correspondence should be addressed to Yang Liu (liuyang_fd@fudan.edu.cn)

Table S1. Summary of experimental parameters used for the synthesis of metal telluride NCs.

	metal precursor	M:Te molar ratio	Reaction time (min)	Reaction temperature (°C)	Product NCs
Ni-Te	Ni(acac) ₂	1:2	5	250	NiTe ₂
		1:2	5	300	NiTe ₂
		1:2	60	330	NiTe ₂
		1:1.5	5	250	Ni _{0.75} Te
		1:1.5	5	300	Ni _{0.64} Te
		1:1.5	60	330	Ni _{0.53} Te
		1:1	5	250	NiTe
		1:1	5	300	NiTe
		1:1	60	330	Ni _{0.88} Te
	Ni(OAc) ₂	1:2	60	330	NiTe ₂
		1:1.5	60	300	Ni _{0.64} Te
	NiCl ₂	1:2	60	330	NiTe ₂
		1:1.5	60	300	Ni _{0.64} Te
Co-Te	Co(acac) ₂	1:2	60	300	CoTe ₂
		1:2	60	330	CoTe ₂
		1:1.5	5	250	CoTe ₂
		1:1.5	5	300	CoTe ₂
		1:1.5	60	330	Co _{0.64} Te
		1:1	5	250	CoTe ₂
		1:1	5	300	CoTe ₂
		1:1	60	330	CoTe
	Co(OAc) ₂	1:2	60	330	CoTe ₂
		1:1.5	60	330	Co _{0.67} Te
		1:1	60	330	Co _{0.83} Te
	CoCl ₂	1:2	60	330	CoTe ₂

Fe-Te	Fe(acac) ₃	1:2	60	300	FeTe ₂
	Fe(OAc) ₂	1:2	60	330	FeTe ₂
	FeCl ₃	1:2	60	300	FeTe ₂
	FeCl ₂	1:2	60	300	FeTe ₂
Mn-Te	Mn(acac) ₂	1:2	120	330	MnTe ₂
		1:1.5	120	330	MnTe ₂
		1:1.5	60	330	MnO+MnTe ₂
		1:1.5	45	330	MnO+MnTe ₂
		1:1.5	30	330	MnO
		1:1	120	330	MnO+MnTe ₂
	Mn(acac) ₃	1:2	120	330	MnO+MnTe ₂
		1:2	120	300	MnO
	Mn(OAc) ₂	1:2	120	330	MnO
	MnCl ₂	1:2	30	300	No product
	Cr-Te	Cr(CO) ₆	1:1.8	120	350
Cr(acac) ₃		1:1.8	120	350	No product
CrCl ₃		1:1.8	120	350	No product
CrCl ₂		1:1.8	120	350	No product

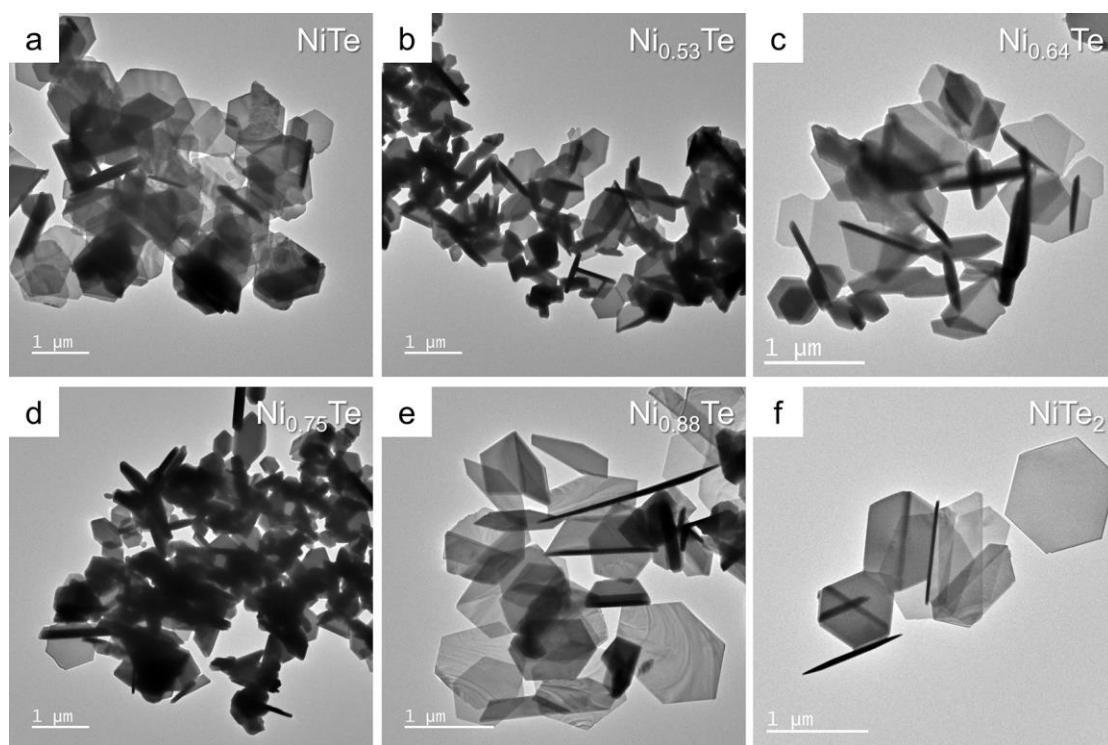


Fig. S1. TEM images different NiTe_{2-x} nanosheets synthesized using Ni(acac)₂ precursor.

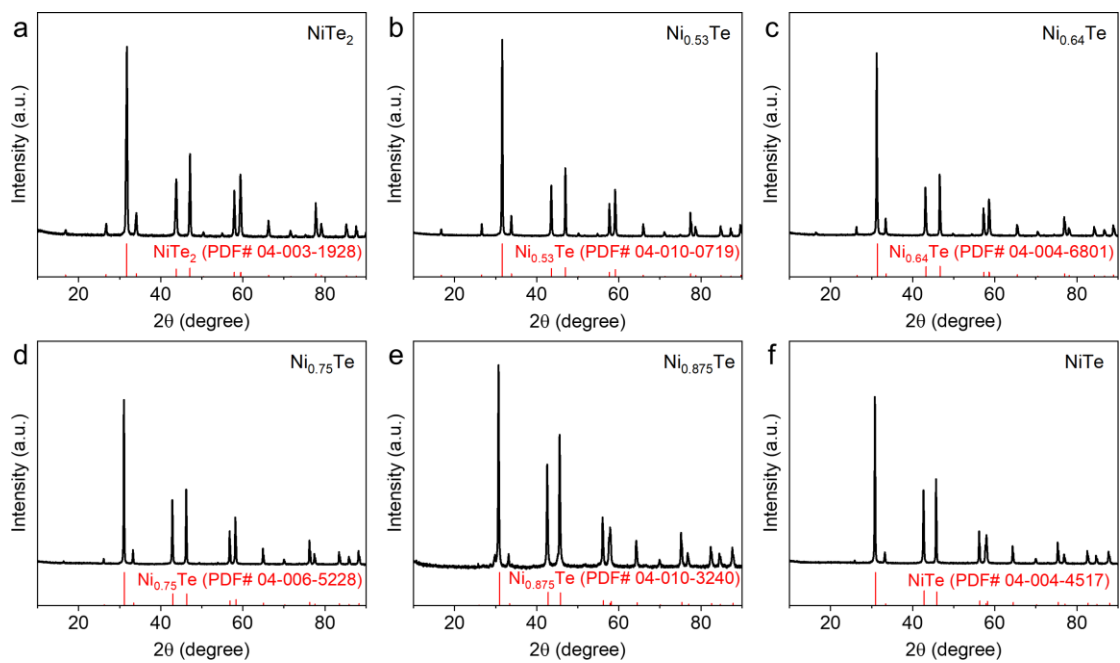


Fig. S2. XRD patterns of NiTe_{2-x} nanosheets: (a) NiTe₂, (b) Ni_{0.53}Te, (c) Ni_{0.64}Te, (d) Ni_{0.75}Te, (e) Ni_{0.88}Te, (f) NiTe.

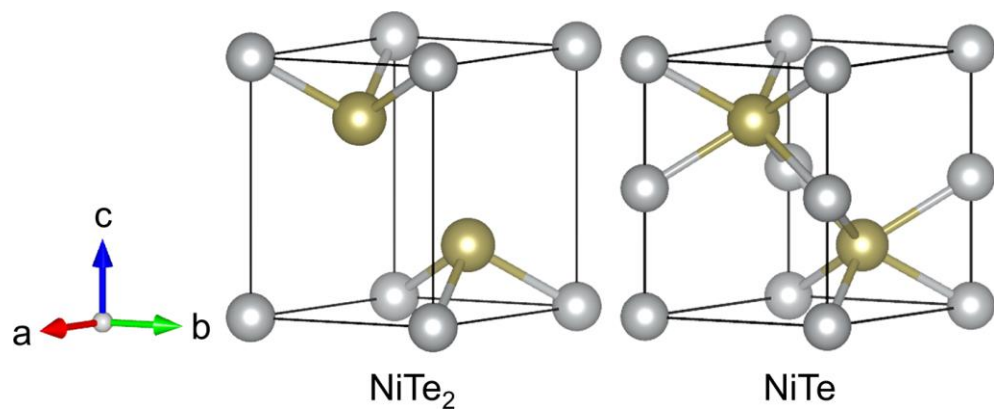


Figure S3. Structural models of NiTe_2 and NiTe .

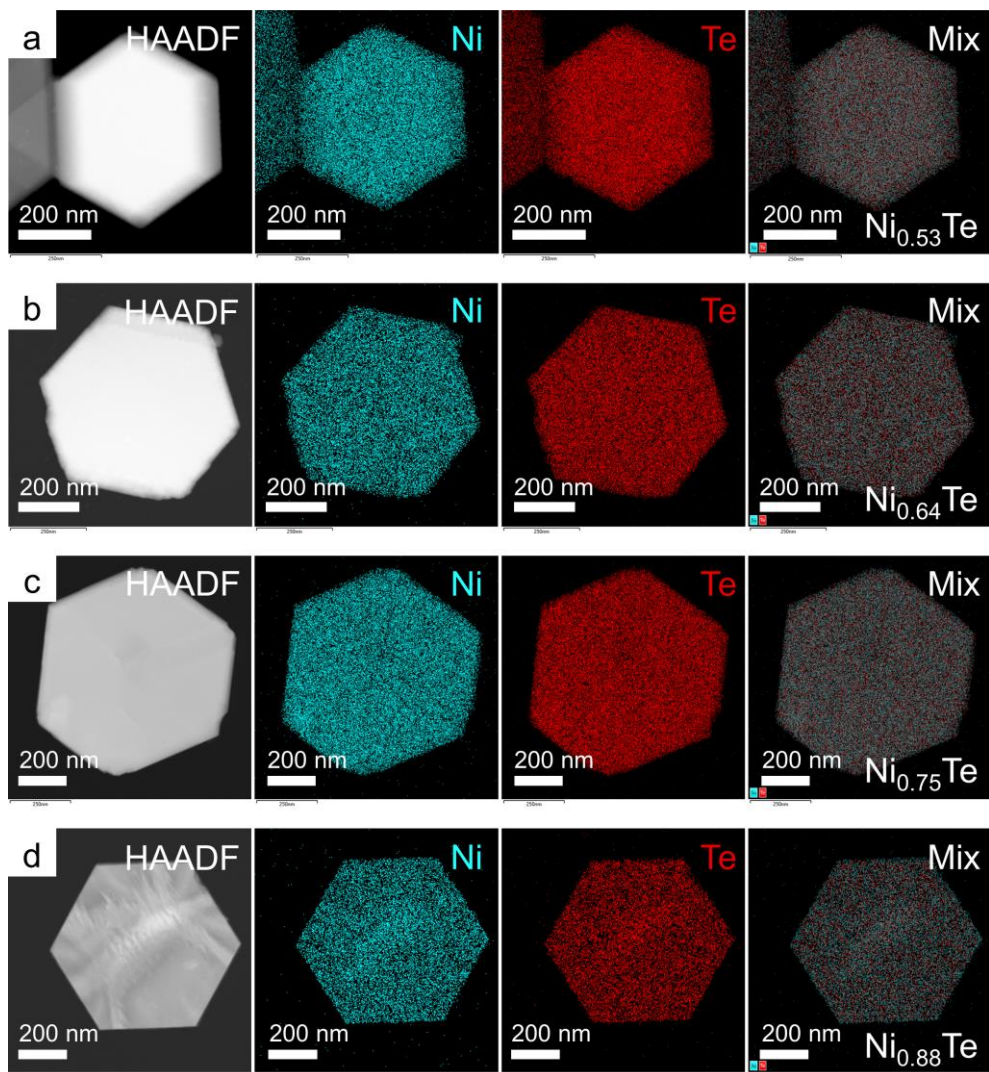


Fig. S4. HAADF and elemental maps of Ni and Te for (a) $\text{Ni}_{0.53}\text{Te}$, (b) $\text{Ni}_{0.64}\text{Te}$, (c) $\text{Ni}_{0.75}\text{Te}$, (d) $\text{Ni}_{0.88}\text{Te}$.

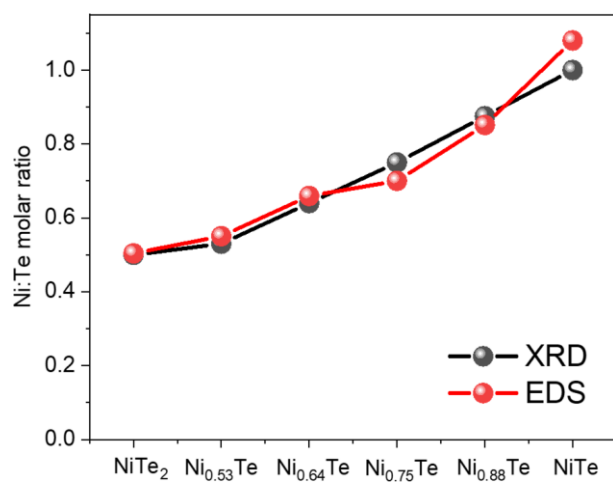


Fig. S5. Comparison of the Ni:Te molar ratio in various NiTe_{2-x} based on quantitative EDS results and XRD patterns.

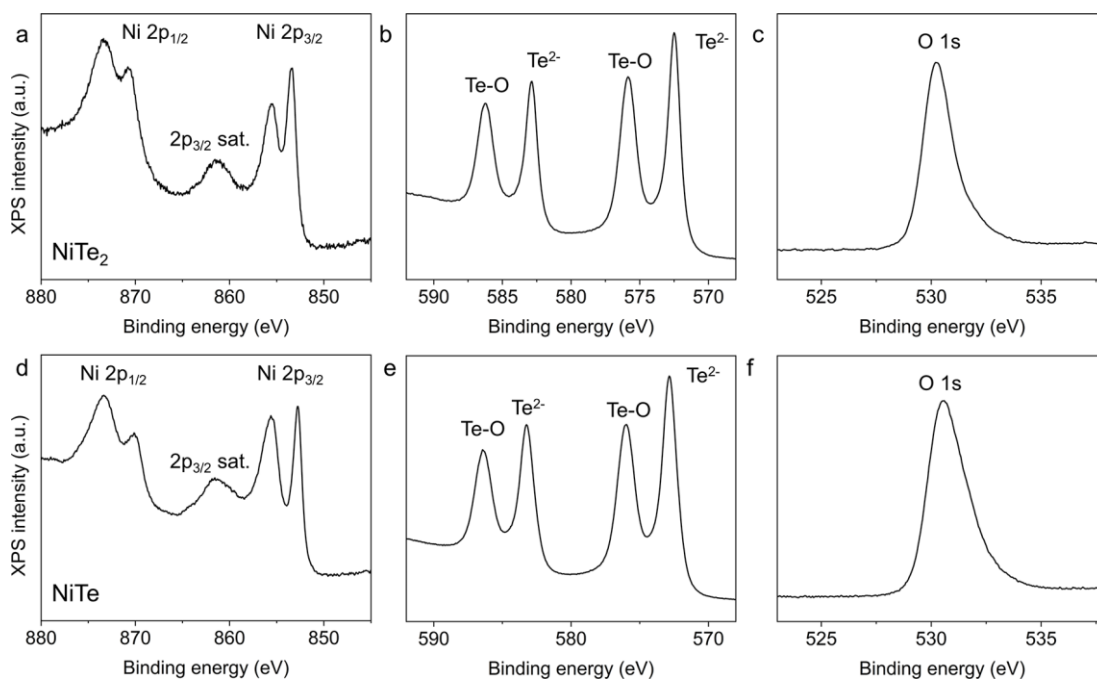


Fig. S6. XPS spectra of (a) Ni 2p, (b) Te 3d, (c) O 1s for NiTe₂ nanosheets. XPS spectra of (d) Ni 2p, (e) Te 3d, (f) O 1s for NiTe nanosheets.

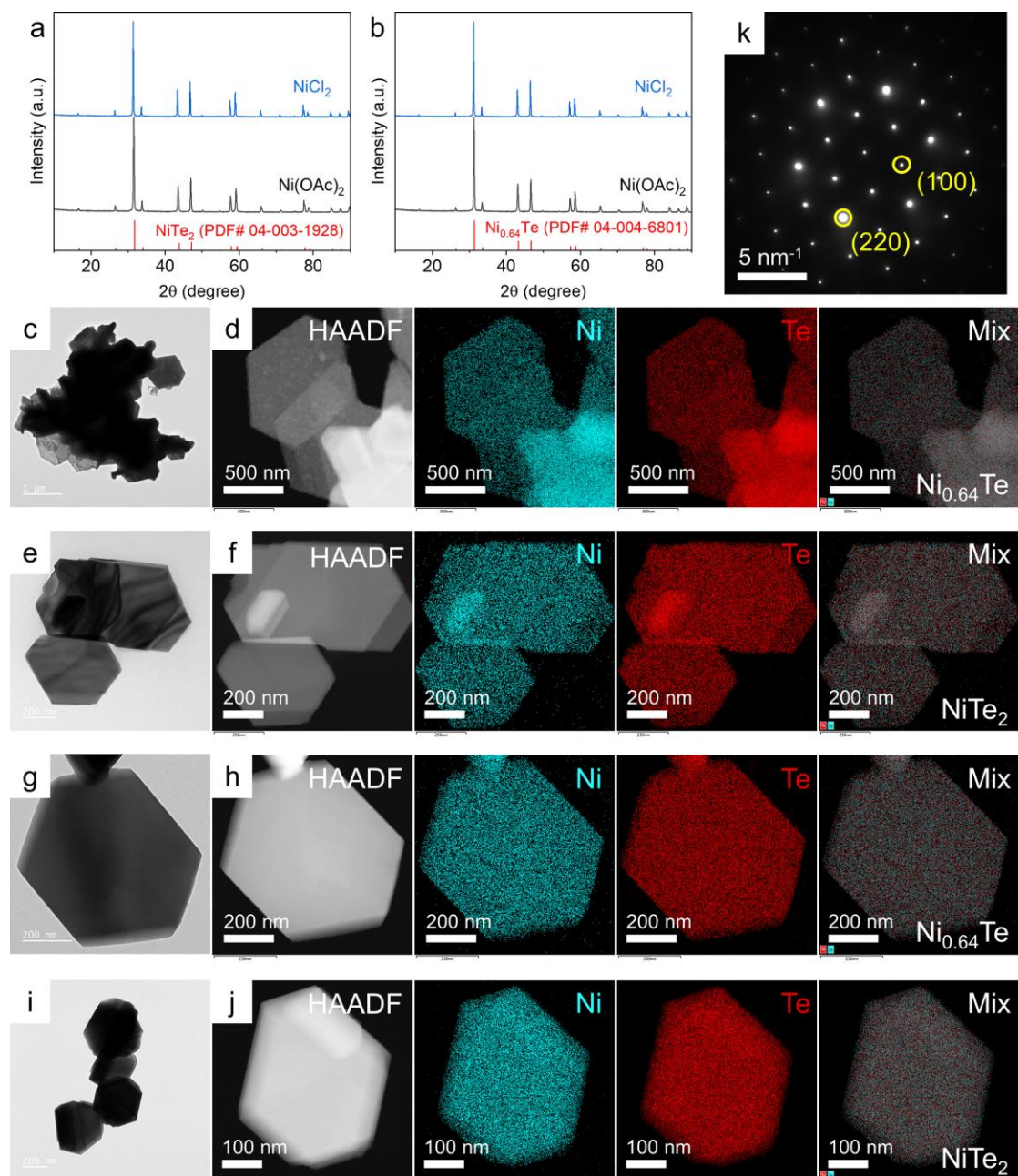


Fig. S7. XRD patterns of NiTe_{2-x} nanosheets synthesized using NiCl₂ and Ni(OAc)₂ precursors: (a) NiTe₂ and (b) Ni_{0.64}Te. (c,d) TEM and HAADF images, and elemental maps of Ni and Te of Ni_{0.64}Te nanosheets synthesized using NiCl₂. (e,f) TEM and HAADF images, and elemental maps of Ni and Te of NiTe₂ nanosheets synthesized using NiCl₂. (g,h) TEM and HAADF images, and elemental maps of Ni and Te of Ni_{0.64}Te nanosheets synthesized using Ni(OAc)₂. (i,j) TEM and HAADF images, and elemental maps of Ni and Te of NiTe₂ nanosheets synthesized using Ni(OAc)₂. (k) Representative SAED pattern of NiTe₂ nanosheets synthesized using NiCl₂.

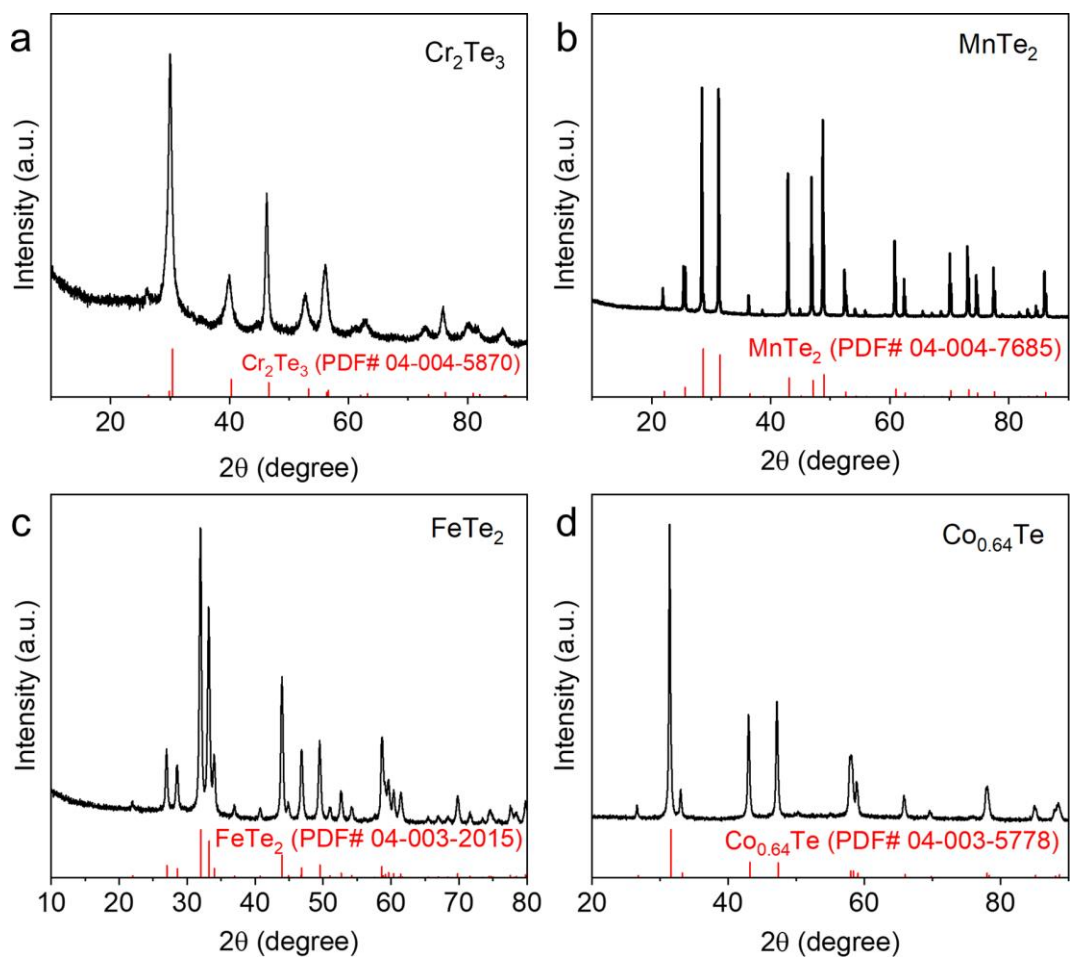


Fig. S8. XRD patterns of (a) Cr_2Te_3 , (b) MnTe_2 , (c) FeTe_2 , and (d) $\text{Co}_{0.64}\text{Te}$.

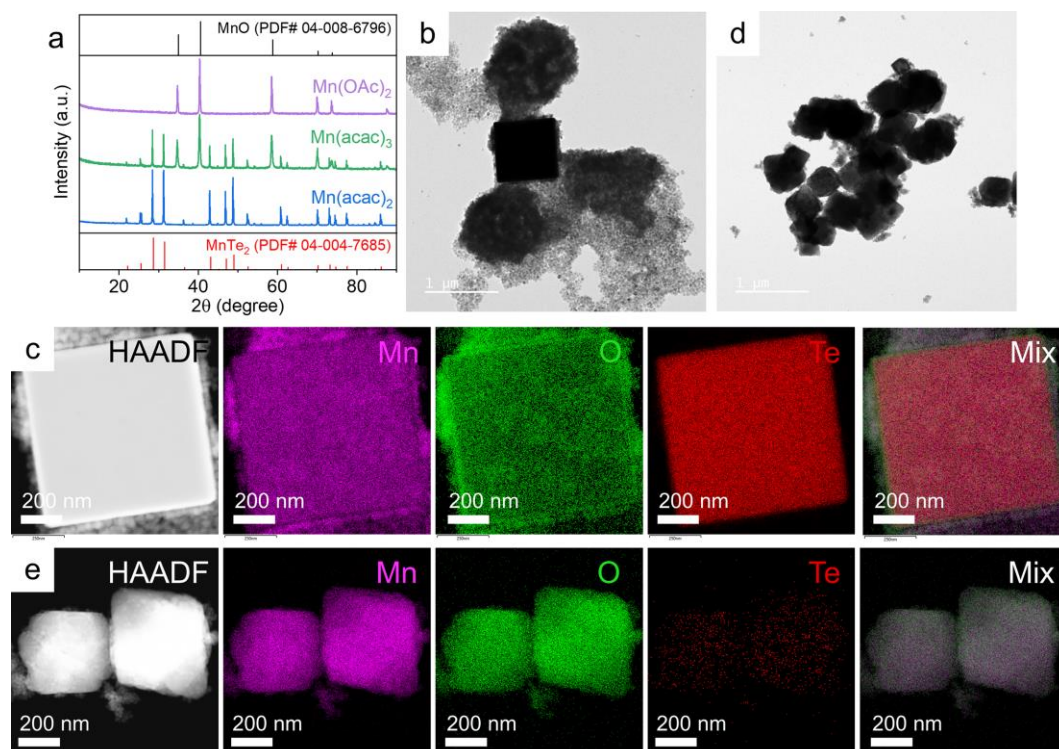


Fig. S9. (a) XRD patterns of the synthesis of the MnTe₂ nanocube using Mn(OAc)₂, Mn(acac)₃, and Mn(acac)₂ precursors: (b-c) TEM and HAADF images, and elemental maps of Mn, O, and Te of the mixture of MnTe₂ nanocube and MnO nanoparticles synthesized using Mn(acac)₃. (d-e) TEM and HAADF images, and elemental maps of Mn, O, and Te of the large MnO nanoparticles synthesized using Mn(OAc)₂.

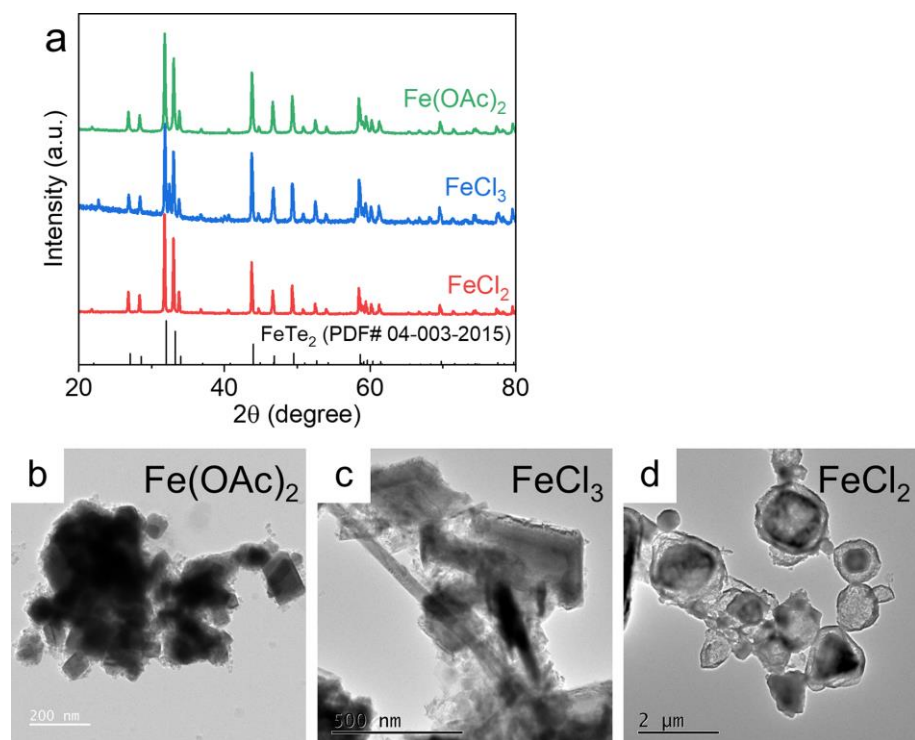


Fig. S10. (a) XRD patterns and TEM images of FeTe_2 NCs synthesized using (b) Fe(OAc)_2 , (c) FeCl_3 , and (d) FeCl_2 precursors, respectively.

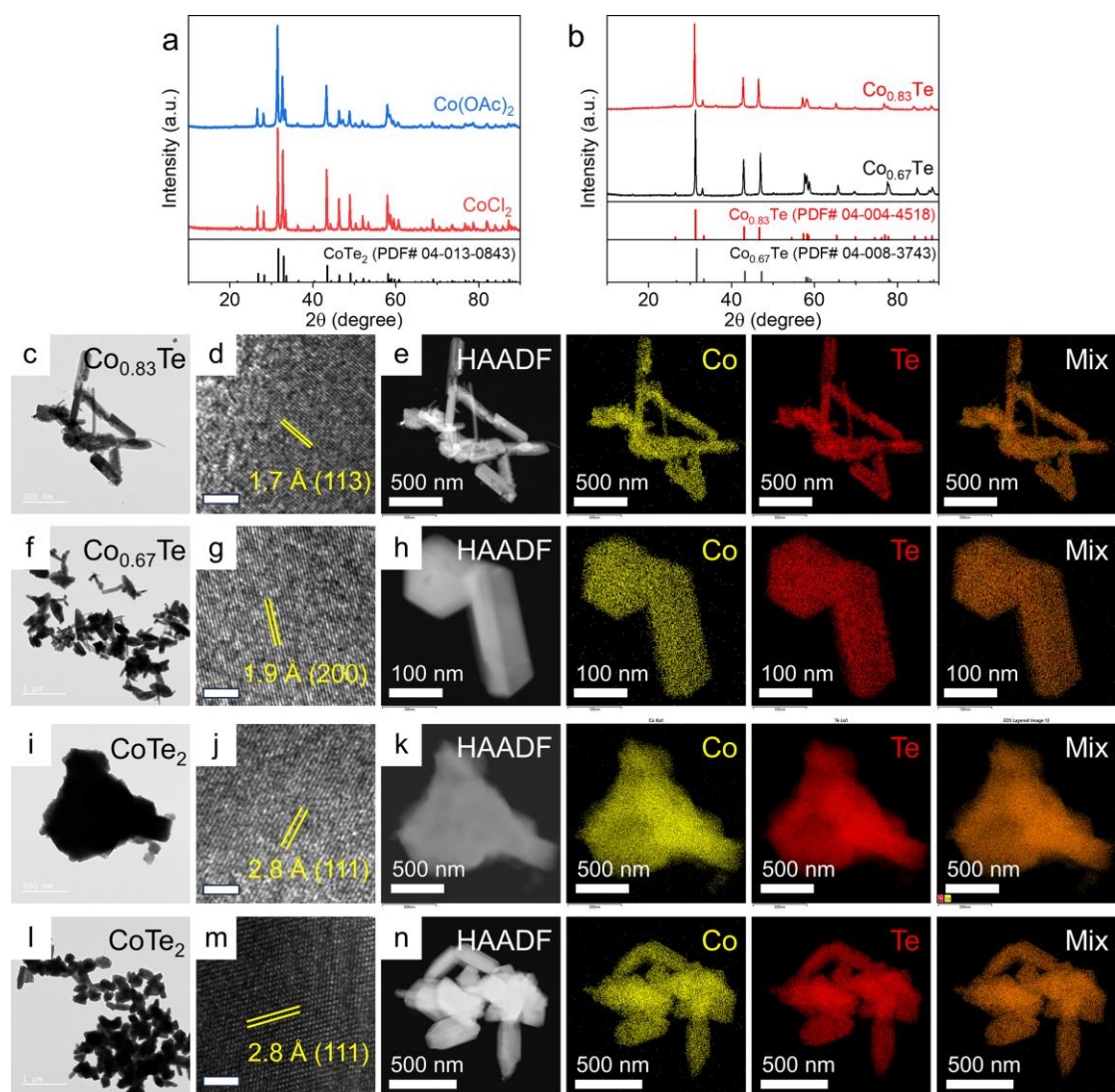


Fig. S11. XRD patterns of CoTe_{2-x} nanoplates synthesized using CoCl_2 and Co(OAc)_2 precursors: (a) CoTe_2 and (b) $\text{Co}_{0.64}\text{Te}$ and $\text{Co}_{0.83}\text{Te}$. (c-e) TEM, HRTEM and HAADF images, and elemental maps of Co and Te of $\text{Co}_{0.83}\text{Te}$ nanoplates synthesized using Co(OAc)_2 . (f-h) TEM, HRTEM and HAADF images, and elemental maps of Co and Te of $\text{Co}_{0.67}\text{Te}$ nanoplates synthesized using Co(OAc)_2 . (i-k) TEM, HRTEM and HAADF images, and elemental maps of Co and Te of CoTe_2 nanoplates synthesized using CoCl_2 . (l-n) TEM, HRTEM and HAADF images, and elemental maps of Co and Te of CoTe_2 nanoplates synthesized using Co(OAc)_2 . Undefined scale bars: 2 nm.

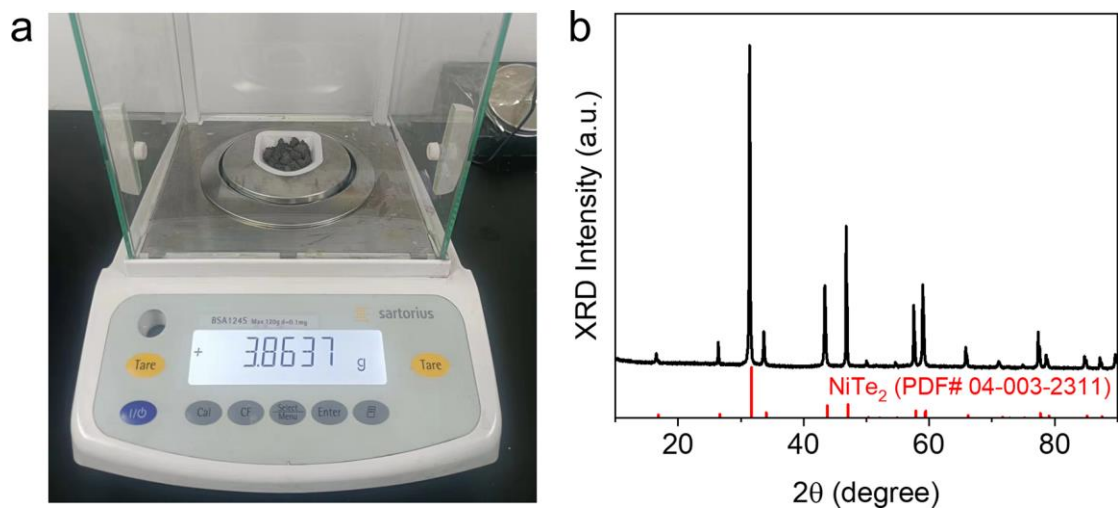


Fig. S12. (a) Photograph of NiTe_2 nanosheets powders produced by large-scale synthesis in a single batch. (b) XRD pattern of NiTe_2 nanosheets from scaled-up synthesis.

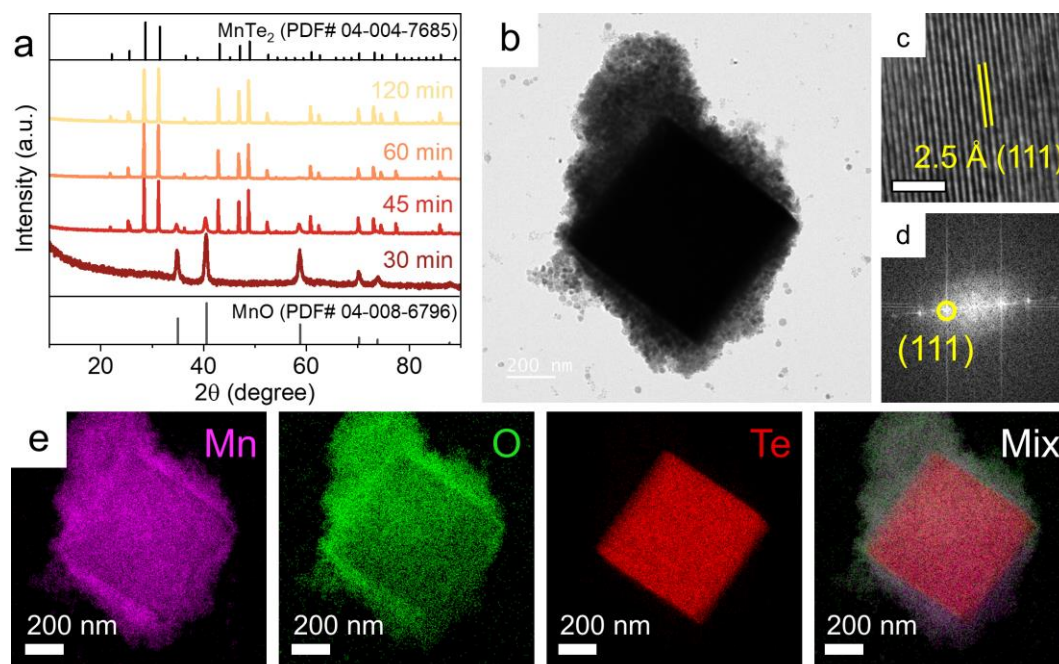


Fig. S13. XRD patterns of the transformation from MnO to MnTe₂ using Mn:Te ratio of 1:1.5. (b) TEM, HRTEM image, corresponding FFT pattern and elemental maps of Mn, O, and Te of MnO to MnTe₂ mixtures obtained at 45 min of the reaction. Undefined scale bars: 2 nm.

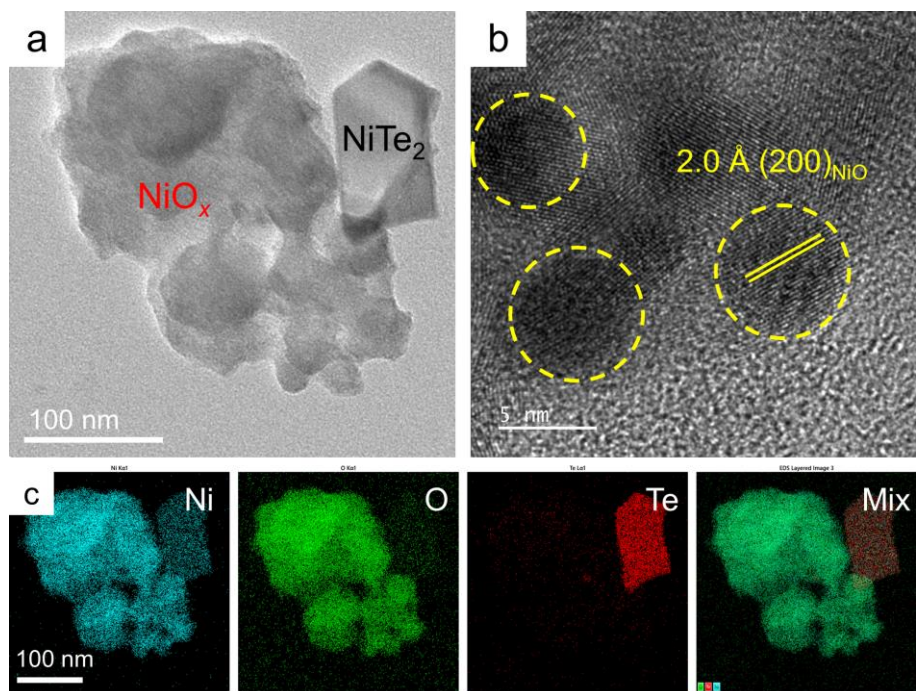


Fig. 14. (a) TEM image of the mixture of NiO_x and NiTe_{2-x} NCs at the initial stage for the synthesis of NiTe₂, (b) HRTEM image of NiO_x intermediates, (c) elemental maps of Ni, O, and Te for the mixture of NiO_x and NiTe_{2-x} NCs.

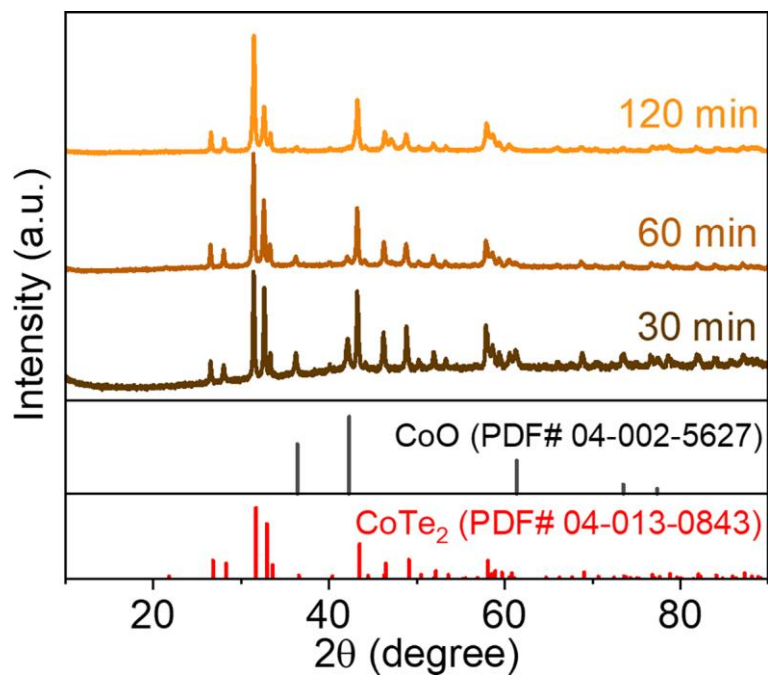


Fig. S15. XRD patterns aliquotes extracted from the synthesis of CoTe₂ nanoplates at different reaction times.

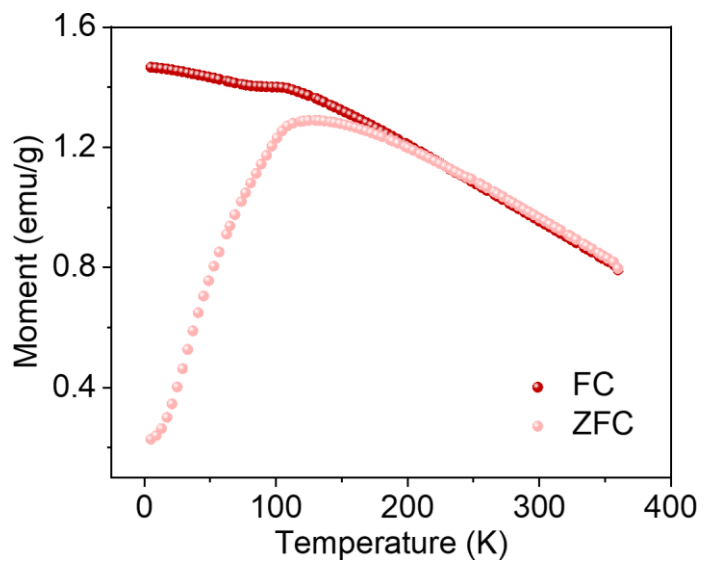


Fig. S16. M-T curve of FeTe₂.

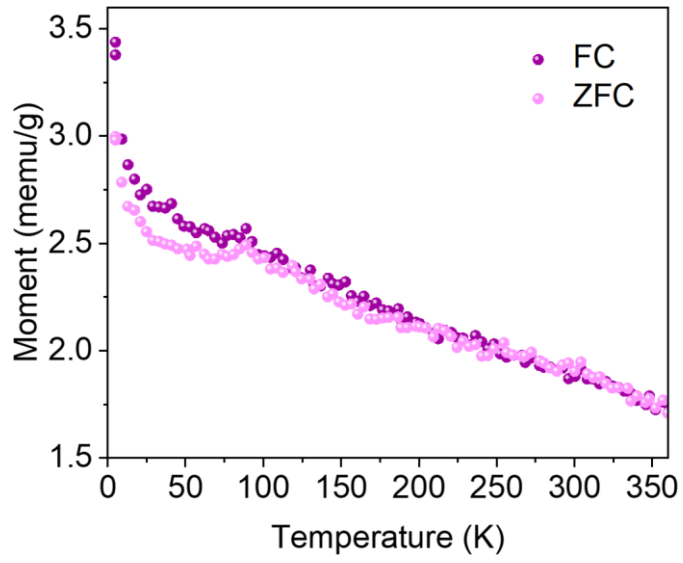


Fig. S17. M-T curve of MnTe₂.

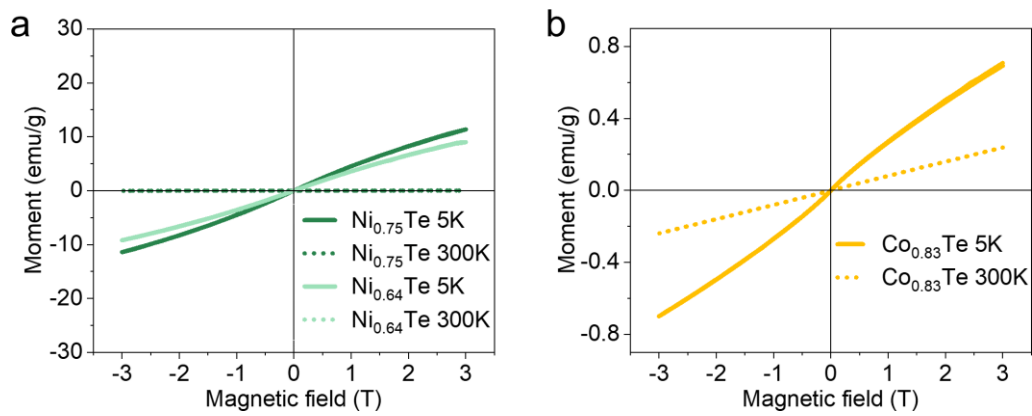


Fig. S18. M-T curve of (a) $\text{Ni}_{0.75}\text{Te}$ and $\text{Ni}_{0.64}\text{Te}$, (b) $\text{Co}_{0.83}\text{Te}$.