Supporting Information

Design of Cobalt Phosphate/Nickel hosphate Film with Improved Electrochromic Performance

Xiaotong Chi, Dairong Chen, Ting Wang* and Xiuling Jiao *

National Engineering Research Center for Colloidal Materials School of Chemistry

and Chemical Engineering, Shandong University, Jinan 250100, P. R.

E-mail: t54wang@sdu.edu.cn, jiaoxl@sdu.edu.cn

Supplementary Table

materials	optical modulation	coloration efficiency (cm ² C ⁻¹)	switching time tc/tb	cycle stability	reference
NiO	78.5% at 550nm	51.8	3.92s / 3.43s	1000	[1]
NiO	51% at 550nm	40	7s /5s	3000	[2]
NiCoO ₂	64.2% at 550nm	33.6	6.8 / 6.7s	200	[3]
NiHCF	17.5% at 400nm	/	21.5min / 8.5s	55	[4]
Li-Ti-NiO	56% at 580nm	21.8	0.9s / 1.0s	300	[5]
Sn-NiO	65.1% at 550nm	39.3	1.3s / 1.4s	500	[6]
rGO-NiO	53% at 630nm	30.5	3.4s / 5.3s	1000	[7]
OL-NiO	54.27% at 450nm	48.5	3.2s / 2.7s	/	[8]
NiHPO ₄	90.8% at 500nm	75.4	7.1/9.6	300	[9]
CoHPO₄/NiHPO₄	65.55% at 500nm	140.37	2.0/8.5	1400	This work

Table S1. EC properties of the NiO films with porous nanostructures.

Table S2. Anode and cathode peak current density *vs* (scan rate)^{1/2} plot obtained from CV curves, corresponding linear fits of the CoHP/NiHP-1 film, CoHP/NiHP-2 film, CoHP/NiHP-3 film, CoHP/NiHP-4 film, CoHP/NiHP-5 film CoHP/NiHP-6 film.

	CoHP/NiHP-1 film	CoHP/NiHP-2 film	CoHP/NiHP-3 film	CoHP/NiHP-4 film	CoHP/NiHP-5 film	CoHP/NiHP-6 film
Oxidative linear fits	I=-0.456+0.77 × $v^{1/2}$ R ² =0.99952	l=-1.1818+1.28 × v ^{1/2} R ² =0.99985	I=-1.15+1.49 × $v^{1/2}$ R ² =0.99913	$I=-0.48+1.64 \times v^{1/2}$ $R^{2}=0.99981$	$I=0.252+1.59 \times v^{1/2}$ $R^{2}=0.99942$	I=0.756+1.52 × v ^{1/2} R ² =0.99945
Reductive linear fits	I=0.392-0.66 × $v^{1/2}$ R ² =0.99976	I=-0.925-1.04 × $v^{1/2}$ R ² =0.99967	$I=1.03-1.23 \times v^{1/2}$ $R^{2}=0.99917$	l=0.45-1.28 × v ^{1/2} R ² =0.99972	I=-0.018-1.23 × v ^{1/2} R ² =0.99911	I=-0.796-1.19 × $v^{1/2}$ R ² =0.99942

Table S3. Anode and cathode peak current density vs (scan rate) $^{1/2}$ plot obtained

from CV curves, corresponding linear fits and diffusion coefficient of the CoHP film,

NiHP film and CoHPO₄/NiHPO₄ film (CoHP/NiHP-4 film).

	CoHP film	NiHP film	CoHP/NiHP-4 film
Oxidative linear fits	$I=-2.31+1.37 \times v^{1/2}$ $R^{2}=0.99904$	I=-1.05+1.49 × $v^{1/2}$ R ² =0.99979	$I=-0.48+1.64 \times v^{1/2}$ $R^{2}=0.99981$
Reductive linear fits	I=1.85-1.25 × $v^{1/2}$ R ² =0.99937	I=-0.51-1.08 × $v^{1/2}$ R ² =0.99935	I=0.45-1.28 × $v^{1/2}$ R ² =0.99972

Table S4. The transmittance of initial state, bleached state and color state of theCoHP/NiHP-1 film, CoHP/NiHP-2 film, CoHP/NiHP-3 film, CoHP/NiHP-4 film,CoHP/NiHP-5 film CoHP/NiHP-6 film.

	CoHP/NiHP-1 film	CoHP/NiHP-2 film	CoHP/NiHP-3 film	CoHP/NiHP-4 film	CoHP/NiHP-5 film	CoHP/NiHP-6 film
T _{initial}	94.86%	93.90%	91.45%	89.53%	88.72%	88.13%
T _{bleached}	88.25%	87.62%	87.33%	87.03%	85.94%	85.39%
T _{colored}	63.25%	42.57%	28.99%	21.48%	20.09%	19.15%
ΔΤ	25.00%	45.05%	58.34%	65.55%	65.85%	66.24%

Supplementary Figures



Fig. S1 The two-step electrochemical deposition CV curves while prepared (between -1.2 and 0.2 V *vs.* Ag/AgCl, 20 mV s⁻¹) and corresponding top-view SEM images of (a-b) CoHP film, (c-d) NiHP film and (e-f) CoHP/NiHP-5 film.



Fig. S2 XRD patterns of (a) purchased clean ITO glass and (b) CoHP film, NiHP film and CoHP/NiHP film.



Fig. S3 (a) XPS spectrum of the NiHP film, (b-d) Ni 2p, O 1s and P 2p spectrum of the NiHP film.



Fig. S4 (a) CV curve comparison between CoHP/NiHP-1 film, CoHP/NiHP-2 film, CoHP/NiHP-3 film, CoHP/NiHP-4 film, CoHP/NiHP-5 film CoHP/NiHP-6 film at the scan rate of 20 mV s-1, (b) Anode and cathode peak current density vs (scan rate)^{1/2} plot obtained from CV curves and corresponding linear fits of the CoHP/NiHP-1 film, CoHP/NiHP-2 film, CoHP/NiHP-3 film, CoHP/NiHP-4 film, CoHP/NiHP-5 film CoHP/NiHP-6 film.



Fig. S5 CV curves of (a) CoHP film, (b) NiHP film (c) CoHP/NiHP-1 film,(d) CoHP/NiHP-

2 film,(e) CoHP/NiHP-3 film,(f) CoHP/NiHP-4 film, (g)CoHP/NiHP-5 film and (h)

CoHP/NiHP-6 film at various scan rates from5 mV s⁻¹ to 25 mV s⁻¹.



Fig. S6 GCD curves of (a) CoHP film and (b) NiHP film in 0.1 M KOH under multiple

discharging current densities.



Fig. S7 Initial state of CoHP film, NiHP film and CoHP/NiHP film.



Fig. S8 (a)The initial state transmittance and (b) the bleached state and the colored

state of CoHP/NiHP film with different numbers of electrodeposited CV cycles.



Fig. S9 Optical memory at 500 nm of the CoHP/NiHP film at stimulation voltages of

0.7 V (vs. Ag/AgCl) for 60 s, power off for 43200 s.



Fig. S10 300 cycles cyclic stability measurement of (a) CoHP film and (b) NiHP film at



500 nm by chronoamperometry and in-situ spectroscopic response.

Fig. S11 1400 cycles cyclic stability measurement of CoHP/NiHP film at 500 nm by

chronoamperometry and in-situ spectroscopic response.

Reference:

- 1. Y. Tian, Z. Li, S. Dou, X. Zhang, J. Zhang, L. Zhang, L. Wang, X. Zhao and Y. Li, *Surf. Coat. Technol.*, 2018, 337, 63-67.
- 2. Y. Ren, X. Zhou, H. Zhang, L. Lei and G. Zhao, J. Mater. Chem. C, 2018, 6, 4952-4958.
- 3. P. Lei, J. Wang, P. Zhang, S. Liu, S. Zhang, Y. Gao, J. Tu and G. Cai, *J. Mater. Chem. C*, 2021, 9, 14378-14387.
- 4. W. Xue, Y. Zhang, F. Liu, Y. Dou, M. Yan and W. Wang, Research, 2023, 6, 0227.

- 5. J. Zhou, G. Luo, Y. Wei, J. Zheng and C. Xu, *Electrochim. Acta*, 2015, 186, 182-191.
- 6. Y. Zhao, X. Zhang, X. Chen, W. Li, L. Wang, Z. Li, J. Zhao, F. Endres and Y. Li, *Electrochim. Acta*, 2021, 367, 137457.
- 7. J. Xue, H. Xu, S. Wang, T. Hao, Y. Yang, X. Zhang, Y. Song, Y. Li and J. Zhao, *Appl. Surf. Sci.*, 2021, 565, 150512.
- 8. K.-H. Kim, S.-J. Jeong, B.-R. Koo and H.-J. Ahn, *Appl. Surf. Sci.*, 2021, 537, 147902.
- 9. P. Lei, J. Wang, Y. Gao, C. Hu, S. Zhang, X. Tong, Z. Wang, Y. Gao and G. Cai, *Nano-Micro Lett.*, 2023, 15, 34.