Electronic Supplementary Information (ESI) for: -

ESIPT-Active Columnar Liquid Crystal: Organic Dyes and Quantum Dots Assisted Fluorescence Modulation

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Figure S1. 1D intensity vs 2 θ profiles obtained for the Col_h/p6mm phase of PBSAN-14 as a function of temperature.

Table S1. Periodic parameter of host PBSAN-14 at two different temperatures

Figure S2. Molecular structure of (a) methyl red dye and (b) rubrene.

Figure S3. Room-temperature absorption and photoluminescence emission spectrum of different organic and inorganic dopants used in the study; (a) MRC dye, (b) rubrene, (c) carbon QDs and (d) perovskite QDs.



Figure S1: 1D intensity vs 2 θ profiles obtained for the Col_h/p6mm phase of PBSAN-14 as a function of temperature.

Table S1: Periodic param	eter of host PBS	SAN-14 at two differe	nt temperatures derived
using XRD plot shown ab	ove ¹ .		

Temperature	Spacing Observed	Spacing Calculated	Miller Indices	Periodic
(⁰ C)	d _{obs} (Å)	d_{cal} (Å)	hkl	Parameters
80 ºC	41.03	41.02	100	<i>a</i> = 47.37
	24.79	23.68	110	<i>S</i> = 1943.20
	20.56	20.51	200	V = 8647.63
	15.60	15.52	210	Z = 2.70
105 ⁰C	41.10	41.09	100	<i>a</i> = 47.45
	24.20	23.72	110	<i>S</i> = 1949.85
	20.61	20.54	200	V = 8891.35
	15.61	15.53	210	<i>Z</i> = 2.73

a – Lattice Parameter (Å); S – Lattice Area (Å²); V – Molecular Volume (Å³);

Z – Number of molecules per column slice



Figure S2: Molecular structure of (a) methyl red dye and (b) rubrene.



Figure S3: Room-temperature absorption and photoluminescence emission spectrum of different organic and inorganic dopants used in the study; (a) MRC dye, (b) rubrene, (c) carbon QDs and (d) perovskite QDs. The dopants have been dispersed in dichloromethane (concentration: 0.01 mg/mL).

References-

1. M. B. Kanakala and C. V. Yelamaggad, *Journal of Molecular Liquids*, 2021, **332**, 115879.