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Supporting Information

Highly Efficient Organic-Graphene Hybrid Photodetectors *via* Molecular Peripheral Editing

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Methods

1. Measurements

UV-vis absorption spectra

The UV/vis absorption spectra of the TTF-NH $_2$ and TTF-CHO solutions, as well as the TTF-Graphene hybrid films supported on the quartz substrate were recorded with a JASCO V-650 spectrophotometer.

Raman Spectroscopy

Raman spectroscopy investigations were carried out in ambient conditions with a Renishaw inVia spectrometer equipped with a 532 nm laser, and the wavenumber (energy) resolution is about 1.25 cm⁻¹ (\approx 1 meV). The excitation power was kept lower than 1 mW to avoid local heating damage effects.

XPS Measurements

XPS measurements were carried out with a Thermo Scientific K-Alpha X-ray photoelectron spectrometer with Al anode as the X-ray source (X-ray radiation of 1486 eV, Spot sizes of 400 μ m). And basic chamber pressure is about 10⁻⁹ mbar. All XPS spectra were calibrated using the C 1s peak at 284.8 eV as a reference.

2. Equations for the calculation of parameters in devices

2.1 Mobility

$$\mu = \frac{L}{WC_i V_{ds} \partial V_G}$$

2.2 Charge in the carrier density change

$$\Delta n = \frac{C_i \Delta V_D}{e} = 7.99 \times 10^{10} \Delta V_D \text{ cm}^{-2}$$

2.3 Photoresponsivity

$$R = \frac{I_{ph}}{P_{in}}$$

2.4 Specific Detectivity

$$D^* = \frac{\sqrt{AB}}{NEP} = \frac{R}{\sqrt{\frac{2eI_{dark}}{A}}}$$

3. Computational Details

The quantum mechanics calculations were performed using the Vienna Ab initio Simulation Package (VASP) version 5.4.4,^{1,2} employing the projector augmented wave (PAW) method^{3, 4} and a plane wave basis set. We utilized the Perdew, Burke, and Ernzerhof (PBE)⁵ flavor of Density Functional Theory (DFT), supplemented by the post-stage DFT-D3 method to account for London dispersion (van der Waals attraction) with Becke-Johnson damping. The PAW method effectively addressed core-valence interactions. A kinetic energy cutoff of 400 eV was applied for plane wave expansions, and reciprocal space was sampled using the Γ -centered Monkhorst-Pack scheme with a grid of 1 × 1 × 1. Convergence criteria were judiciously set at 1 × 10⁻⁵ eV for energy differences in solving the electronic wave function, while atomic coordinates were diligently converged to within 1 × 10⁻² eV Å⁻¹ for maximal force components.

The gas-phase electronic structure calculations were performed using the Gaussian 16 suite of programs (Revision A.01).⁶ The geometrical structures of the ground-state molecules were optimized by using the B3LYP hybrid functional combined with the 6-31G(d) basis. Compared with the experimental data, it was shown that the energy levels obtained by O3LYP hybrid function is more accurate. Hence, the molecular orbital diagrams were calculated at the O3LYP/ 6-31G(d) level.

In our investigation, we rigorously modeled the graphene surface using 14×14 unit cells, incorporating a 20 Å vacuum thickness to obviate any potential interactions between adjacent slabs. Throughout our calculations, all atomic entities were allowed to relax, thereby ensuring the attainment of a stable configuration.

Cranhana/TTE NH	тте ми	graphana	Binding	
Graphene/1117-Mil2	111-1112	graphene	energy ^a	
-3093.97	-409.24	-2682.48	-2.25	
Graphene/TTF-		•	Binding energy	
СНО	ПИ-СНО	graphene		
-3106.73	-422.12	-2682.48	-2.13	

Table S1. Binding energies (eV) between TTF-derivatives and graphene system.

^a binding energy = E_{total} - $E_{graphene}$ - E_{TTF} , where E_{total} , $E_{graphene}$, and E_{TTF} represent the calculated energies of the total graphene/TTF-derivatives system, graphene and TTF derivatives, respectively.

	Responsivity (A/W)	Detectivity (Jones)	Response time	References
RG6	460	1010	<100 ms	[7]
C8-BTBT	1.57×10 ⁴	-	<25 ms	[8]
Ru-Complex	1×10^{5}	-	2.8 s	[9]
COF _{ETBC-TAPT}	3.2×10^{7}	6×10 ¹³	1.14 ms	[10]
PTCDA/pentacene	105	-	<30 µs	[11]
PTCDI-C8	2×10^{5}	10 ¹⁶	~10 ms	[12]
rubrene	107	9×10 ¹¹	~100 ms	[13]
MOF	>106	6.9×10 ¹⁴	<150 ms	[14]
TTF-CA	105	1013	8 ms	[15]
PTCDA/C8-BTBT	5.76×10 ⁵	-	470 ms	[16]
PTAA/ CH3NH3PbI3-xCl x	105	10 ¹³	310 ms	[17]
PbS QDs	107	7×10^{13}	10 ms	[18]
BUBD-1	~10 ⁶	>10 ¹²	0.3 ms	[19]
TTF-NH ₂	1.8×10^{7}	1.1×10 ¹⁵	267 ms	This work
TTF-CHO	2.0×10^{6}	3.2×10^{14}	381 ms	This work

Table S2. The figures of merit of several reported graphene-based hybrid,graphene/TTF-NH2, and graphene/TTF-CHO photodetectors.

Graphene/	Power density (W/cm ²)	7.6E-6	9.2E-6	1.5E-5	2.1E-5	3.0E-5
TTF-NH ₂	Responsivity (A/W)	1.1E7	9.4E6	7.0E6	5.7E6	4.8E6
	Detectivity (Jones)	6.5E14	5.5E14	4.1E14	3.4E14	2.8E14
Graphene/ TTF-NH ₂	Power density (W/cm ²)	6.7E-6	9.0E-6	1.7E-5	4.6E-5	1.1E-4
	Responsivity (A/W)	1.2E6	1.09E6	8.7E5	4.3E5	1.3E5
	Detectivity (Jones)	1.5E14	1.3E14	1.1E14	5.3E13	1.6E13
Graphene/ TTF-NH ₂	Power density (W/cm ²)	1.1E-6	6.1E-6	7.6E-6	9.2E-6	1.5E-5
	Responsivity (A/W)	1.8E7	1.3E7	1.2E7	1.1E7	8.0E6
	Detectivity (Jones)	1.1E15	7.6E14	7.0E14	6.5E14	4.8E14
Graphene/ TTF-NH ₂	Power density (W/cm ²)	3.8E-6	7.6E-6	9.2E-6	1.5E-5	2.1E-5
	Responsivity (A/W)	2.1E7	1.7E7	1.6E7	1.3E7	1.1E7
	Detectivity (Jones)	1.1E15	9.6E14	8.8E14	7.2E14	5.8E14
Graphene/	Power density (W/cm ²)	3.8E-6	7.6E-6	1.5E-5	2.1E-5	3.0E-5
TTF-CHO	Responsivity (A/W)	2.0E6	1.5E6	1.0E6	9.6E5	7.8E5

Table S3. The figures of merit of graphene/TTF- NH_2 and graphene/TTF-CHO photodetectors under 300 nm irradiation.

	Detectivity (Jones)	3.2E14	2.4E14	1.7E14	1.5E14	1.3E14
Graphene/	Power density (W/cm ²)	3.8E-6	7.6E-5	1.5E-5	2.1E-5	3.0E-5
TTF-CHO	Responsivity (A/W)	1.1E6	9.4E5	8.3E5	7.0E5	6.2E5
	Detectivity (Jones)	1.2E14	1.0E14	9.2E13	7.7E13	6.8E13



Figure S1. Monolayer graphene visualized by AFM, showing a thickness of 0.69 nm.

The AFM measurement of the selected graphene flake for device fabrication shows a thickness of 0.69 nm, demonstrating the monolayer nature of graphene.



Figure S2. C1s XPS spectra of (A) CVD graphene, (B) CVD graphene/TTF-NH₂ hybrids, and (C) CVD graphene/TTF-CHO hybrids; (D) N1s XPS spectra of CVD graphene/TTF-NH₂ hybrids; Cl2p XPS spectra of (E) CVD graphene/TTF-NH₂ hybrids and (F) CVD graphene/TTF-CHO hybrids.



Figure S3. Work function plots of (A) CVD graphene/TTF-NH₂ and (B) CVD graphene/TTF-CHO measured by Photoelectron Yield Spectroscopy in Ambient conditions (PYSA).



Figure S4. Molecular configurations calculated by density functional theory (DFT) of pristine (A) TTF-NH₂ and (C) TTF-CHO; Optimized molecular configurations (i.e., configurations with the lowest energy) of (B) TTF-NH₂ and (D) TTF-CHO when adsorbed onto the graphene surface.



Figure S5. Normalized UV-Vis absorption spectra of (A) TTF-NH₂ and (B) TTF-CHO solution with different concentrations; (C) Absorbance at 300 nm as a function of the concentration of TTF-NH₂ ($R^2 = 0.9956$) and TTF-CHO ($R^2 = 0.9995$) solutions.



Figure S6. (A) UV-vis absorption spectra of pristine CVD-graphene, TTF-NH₂, and graphene/TTF-NH₂. (B) UV-vis absorption spectra of pristine CVD-graphene, TTF-CHO, and graphene/TTF-CHO.



Figure S7. PXRD patterns of pristine HOPG and TTF-derivatives drop-casted onto the

HOPG surface.



Figure S8. Transfer characteristics of the pristine graphene FET under dark and light irradiation with different wavelength.



Figure S9. Cyclic voltammetry curves of (A) TTF-NH₂, (B) TTF-CHO, and (C) ferrocenium/ferrocene.



Figure S10. (A-C) Transfer curves of different graphene/TTF-NH₂ hybrid devices under light illumination (V_{ds} =50 mV); (D-F) Corresponding photoresponsivity and specific detectivity change in response to 300 nm illumination under varying power densities.



Figure S11. (A) Transfer curves of graphene/TTF-CHO hybrid device, where TTF-CHO was spin-coated onto the graphene FET, under different power densities of 300 nm illumination (V_{ds} =50 mV); (B) Energy diagrams of the graphene/TTF-CHO hybrids; (C) Photocurrent curves with varying illumination power densities and wavelengths; (D) Photoresponsivity of graphene/TTF-CHO transistor under different wavelengths. The black line is the corresponding absorbance of the TTF-CHO film; (E) Temporal photocurrent response of the device under dark and 450 nm irradiation; (F) Photoresponsivity and specific detectivity change in response to 300 nm illumination under varying power densities.



Figure S12. (A) Transfer curves of graphene/TTF-CHO hybrid device, where TTF-CHO was drop-cast onto the graphene FET at a concentration of 0.5 mM at 70 °C, under different power densities of 300 nm illumination ($V_{ds} = 50 \text{ mV}$); Corresponding (B) photoresponsivity and specific detectivity performance under 300 nm illumination, (C) temporal photocurrent under dark and 450 nm irradiation, and (D) photoresponsivity under different wavelengths of the graphene/TTF-CHO transistor.



Figure S13. Long-term stability tests based on the photoresponsitivity of graphene/TTF-NH₂(A) and graphene/TTF-CHO(B) hybrid devices.



Figure S14. AFM images of TTF-NH₂ films on graphene: (A) after the first dropcasting (R_{RMS} :5.1 nm, 1.0 ×1.0 µm²), (B) after the second drop-casting (R_{RMS} :10.3 nm, 1.0 ×1.0 µm²), and (C) after the third drop-casting (R_{RMS} :11.8 nm, 1.0 ×1.0 µm²); AFM images of spin-coated TTF-CHO films: (D) 3.2 mM solution at a spin rate of 600 rpm (R_{RMS} :1.2 nm, 1.0 ×1.0 µm²), (E) 8.0 mM solution at a spin rate of 600 rpm (R_{RMS} :1.5 nm, 1.0 ×1.0 µm²), and (F) 8.0 mM solution at a spin rate of 400 rpm (R_{RMS} :12.1 nm, 1.0 ×1.0 µm²).



Figure S15. (A) Transfer curves of graphene/TTF-NH₂ hybrid device after step-bystep drop-casting process. (B) Photoresponsivity changes with thickness accumulation during step-by-step drop-casting. (C) Transfer curves and (D) Photoresponsivity changes of the graphene/TTF-CHO hybrid device at different concentrations and spincoating speeds of the TTF-CHO solution.

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