

Molecular insights into solid-state photochromism

in bulk and confined *N*-salicylidenes

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Supporting Information

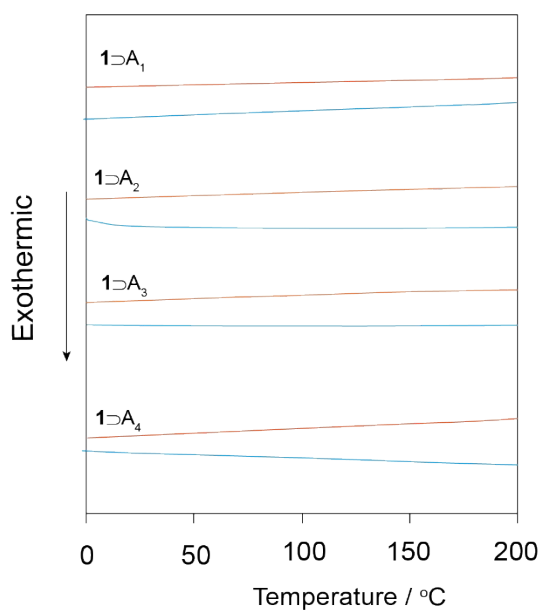


Figure S1. Differential scanning calorimetry traces of 1A_n . Heating curves are shown in orange; cooling curves are shown in blue. No endothermic or exothermic features are observed which would be characteristic of melting or crystallisation of the crystalline *N*-salicylidenes. It can be assumed that excess *N*-salicylidene has been removed via vacuum treatment.

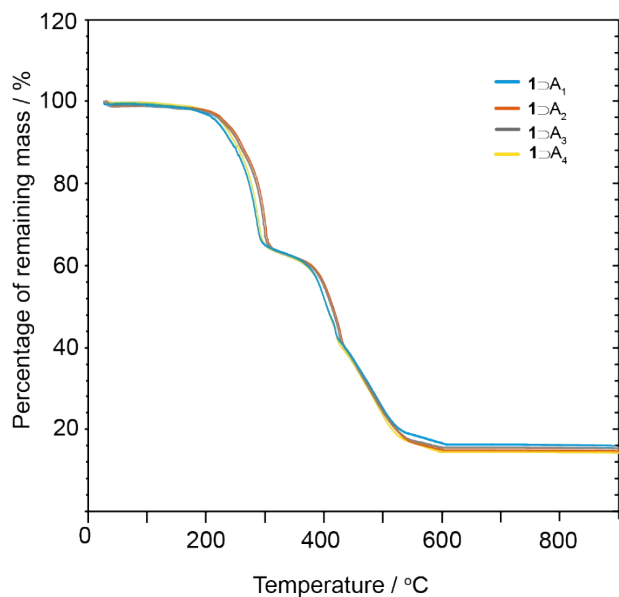


Figure S2. Thermogravimetric analysis traces of 1A_n . Expected final mass for the general formula $1\text{A}_{n(0.5)}$ was calculated to be 15%-16% based on Al_2O_3 . 1A_1 (16.1%), 1A_2 (15.8%), 1A_3 (15.6%), and 1A_4 (14.9%).

Table S1. Crystallographic and calculated geometry optimised parameters for other reported polymorphs of A_n compounds not used in this study.

	A_1	A_1	A_2
Crystal System	Orthorhombic	Orthorhombic	Monoclinic
Space group	$P2_12_12_1$	$Pbc2_1$	$P2_1/c$
Z	4	4	4
Photochromic	Yes	No	No
Experimental torsion angle / °	45.4	6.6	2.3

Table S2. Comparison between calculated and experimental ^{13}C imine chemical shifts at the experimental torsion angle.

Structure	Experimental torsion angle / °	Experimental ^{13}C imine chemical shift (ppm)	Calculated ^{13}C imine chemical shift at experimental torsion angle (ppm)
A_1	47.1	162.1	161.4
A_2	47.0	161.9	161.0
A_3	3.5	155.4	153.2
A_4	5.7	155.2	155.8

Table S3. Torsional angles for anil structures A_1 - A_4 , calculated by geometry optimisations with four methods. Fixed and unfixed refer to whether the unit cell parameters were allowed to relax during the geometry optimisation.

Method	A_1	A_2	A_3	A_4
Experimental	47.1	47.0	3.5	5.7
PBE-D3 Fixed	45.6	43.9	10.7	7.8
PBE-D3 Unfixed	45.6	4.2	10.1	8.6
rSCAN Fixed	46.1	45.7	10.1	8.7
rSCAN Unfixed	46.0	45.7	11.4	9.0

Table S4. Calculated values of V_{free} for A_1 - A_4 and $1\supset A_1$ - $1\supset A_4$

Molecule	Volume of Molecule / \AA^3	Crystalline Unit Cell Volume / \AA^3	Z	V_{free} per molecule / \AA^3	V_{free} per molecule occluded within 1
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A ₁	186.73	1023.38	4	69.1	279.4032	Volume of the cavi
A ₂	200.27	1132.607	4	82.9	265.8632	
A ₃	200.27	1049.972	4	62.2	265.8632	
A ₄	266.13	2388.821	8	32.5	200.0032	

ty of **1** = 233.06 Å³. V_{free} per molecule was calculated using:

$$V_{\text{free}} = (V_{\text{cell}} - Z \cdot V_{\text{molecule}})/Z$$

where V_{cell} is the unit cell volume and V_{molecule} is the volume of a single molecule.

Table S5. Calculated ¹³C chemical shieldings for A₁ and A₃ determined from single-molecule DFT calculations performed on fully optimised structures carbon sites numbered as shown in Figure 2 (main text).

Site	A1 Calculated ¹³ C chemical shielding (ppm)	A3 Calculated ¹³ C chemical shielding (ppm)	Calculated ¹³ C chemical shielding difference (ppm)
C1	29.6	31.6	-2.0
C2	75.8	74.6	1.2
C3	59.9	59.9	0.0
C4	75.9	59.9	16.0
C5	60.3	61.6	-1.3
C6	72.8	72.2	0.6
C7	32.0	33.0	-1.0
C8	42.6	43.2	-0.6
C9	67.5	67.4	0.1
C10	64.1	64.0	0.1
C11	67.2	66.5	0.7
C12	63.8	63.7	0.1
C13	76.3	76.4	-0.1

Torsional angles for fully optimised molecular structures are 34.0° (A₁) and 33.4° (A₃)

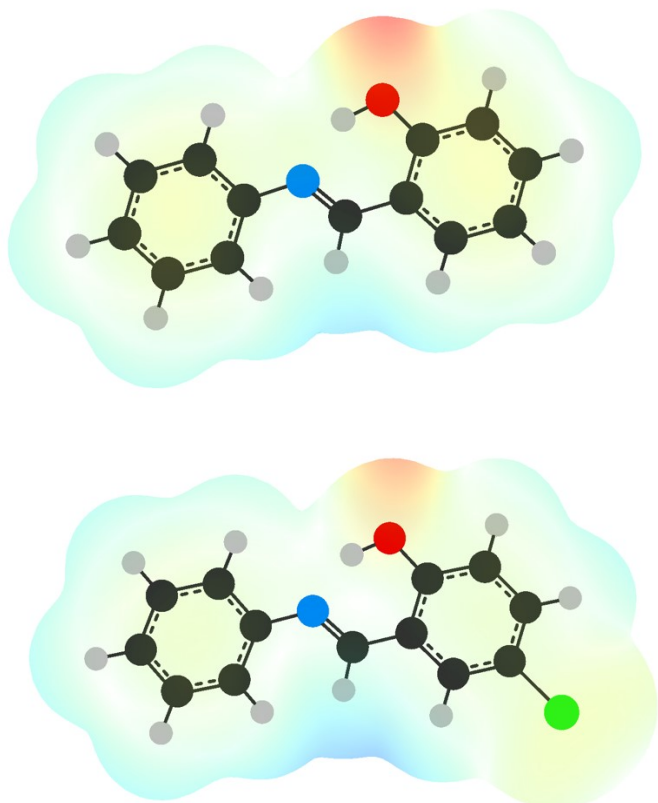


Figure S3. Electron density maps generated from single-molecule DFT calculations on A₁ (top) and A₃ (bottom). Atoms shown are carbon (black), nitrogen (blue), oxygen (red), hydrogen (grey) and chlorine (green).

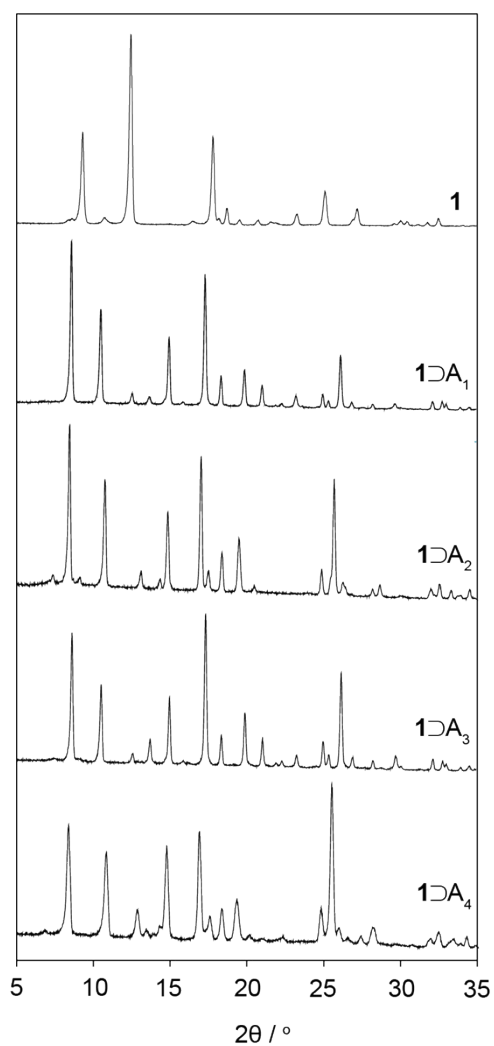


Figure S4a) Comparison of XRPD patterns of $1DA_n$.

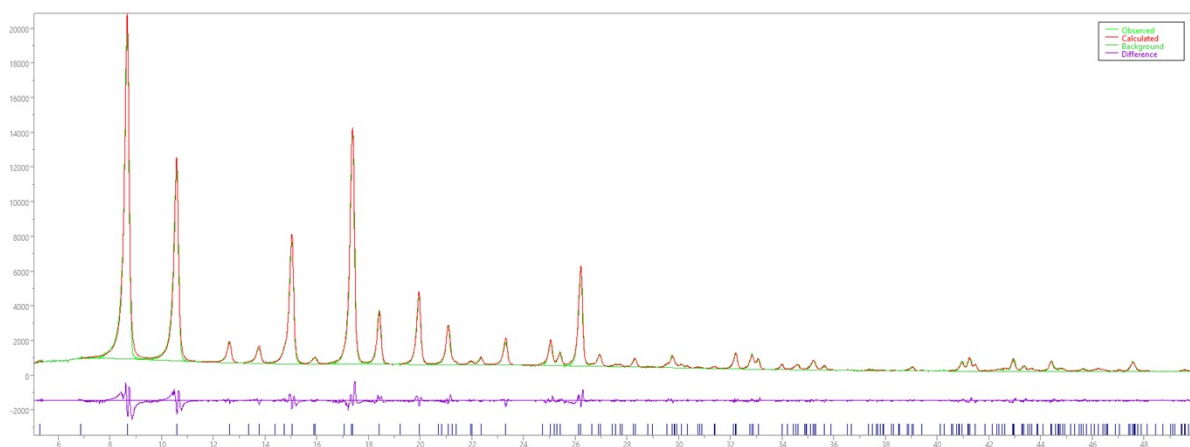


Figure S4b) Le Bail fit of $1DA_1$. Indexing was carried out by N-TREOR09 on EXPO2014. The crystal system was found to be orthorhombic. The lattice parameters were refined to be , $a = 12.87 \text{ \AA}$, $b = 16.71 \text{ \AA}$ and $c = 6.62 \text{ \AA}$, $\alpha = \beta = \gamma = 90^\circ$, $V = 1423.7 \text{ \AA}^3$. The space group was found to be $Imma$. General formula $AlC_{84}H_{68}N_4O_{20}$. The reliability (R) factor based on the powder profile was 5.122 %.

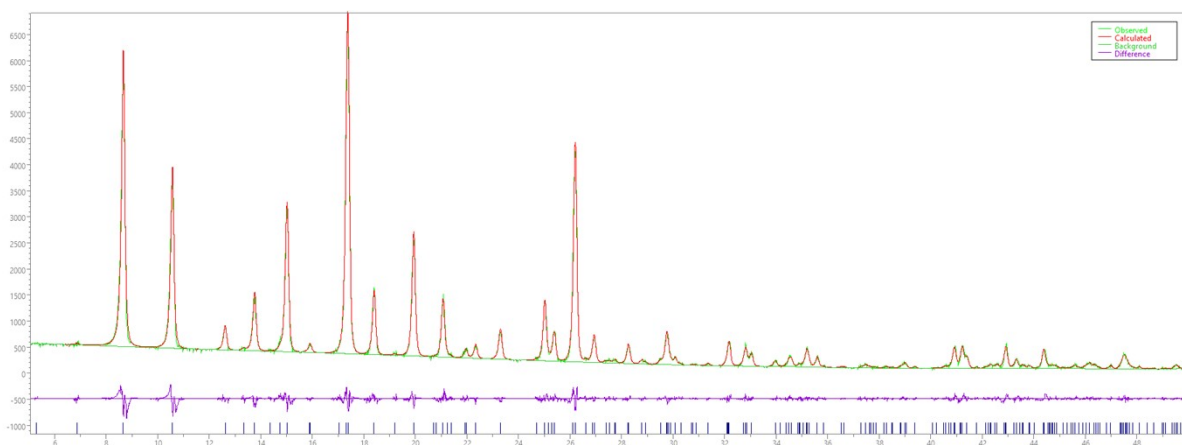


Figure S4c) Le Bail fit of $1D A_3$. Indexing was carried out by N-TREOR09 on EXPO2014. The crystal system was found to be orthorhombic. The lattice parameters were refined to be , $a = 12.87 \text{ \AA}$, $b = 16.74 \text{ \AA}$ and $c = 6.64 \text{ \AA}$, $\alpha = \beta = \gamma = 90^\circ$, $V = 1430.5 \text{ \AA}^3$. The space group was found to be *Imma*. General formula $AlC_{83}H_{68}N_4O_{20}Cl$. The reliability (*R*) factor based on the powder profile was 7.237 %.

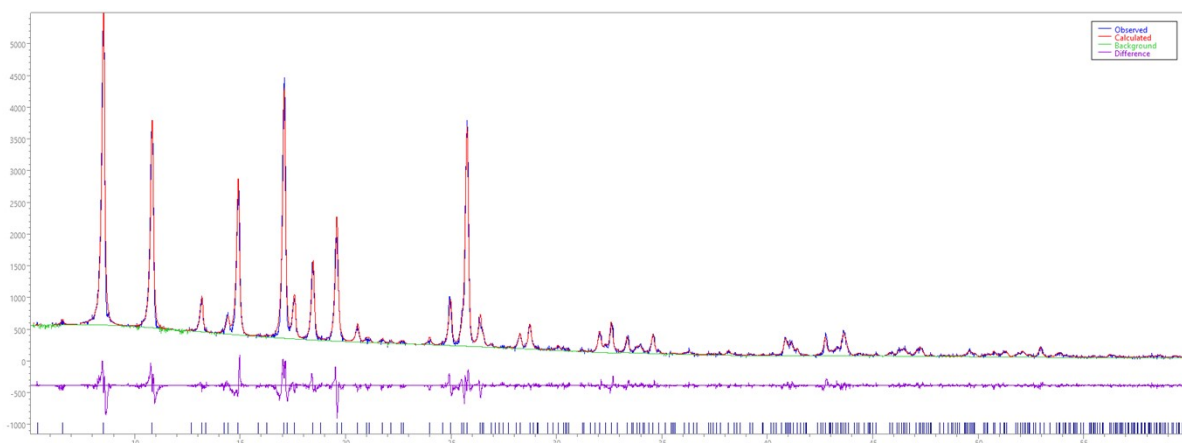


Figure S4d) Le Bail fit of $1D A_2$. Indexing was carried out by N-TREOR09 on EXPO2014. The crystal system was found to be orthorhombic. The lattice parameters were refined to be , $a = 13.44 \text{ \AA}$, $b = 16.33 \text{ \AA}$ and $c = 6.62 \text{ \AA}$, $\alpha = \beta = \gamma = 90^\circ$, $V = 1452.9 \text{ \AA}^3$. The space group was found to be *Imma*. General formula $AlC_{83}H_{68}N_4O_{20}Cl$. The reliability (*R*) factor based on the powder profile was 9.111 %.

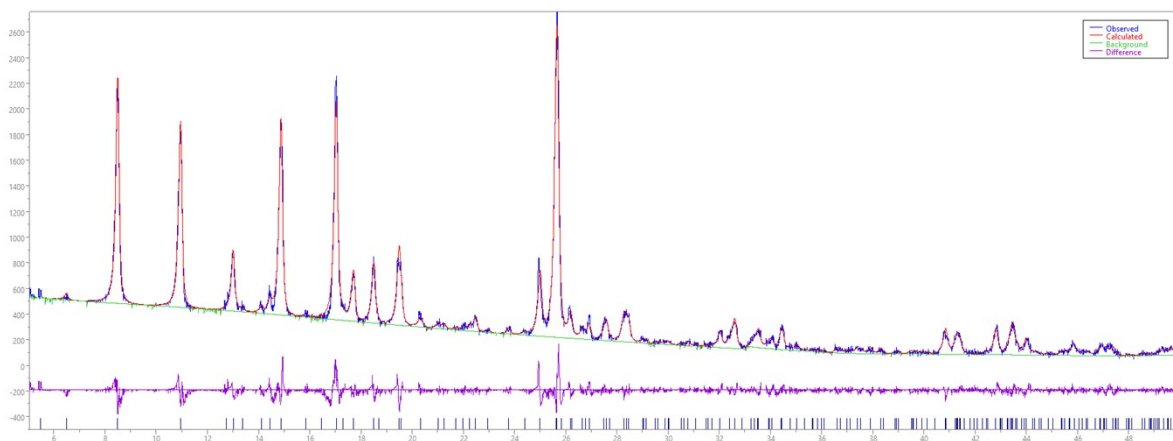


Figure S4e) Le Bail fit of 1D_4 . Indexing was carried out by N-TREOR09 on EXPO2014. The crystal system was found to be orthorhombic. The lattice parameters were refined to be , $a = 13.63 \text{ \AA}$, $b = 16.16 \text{ \AA}$ and $c = 6.63 \text{ \AA}$, $\alpha=\beta=\gamma=90^\circ$, $V = 1460.3 \text{ \AA}^3$. The space group was found to be $Imma$. General formula $\text{AlC}_{83}\text{H}_{68}\text{N}_4\text{O}_{20}\text{Cl}$. The reliability (R) factor based on the powder profile was 10.265 %.

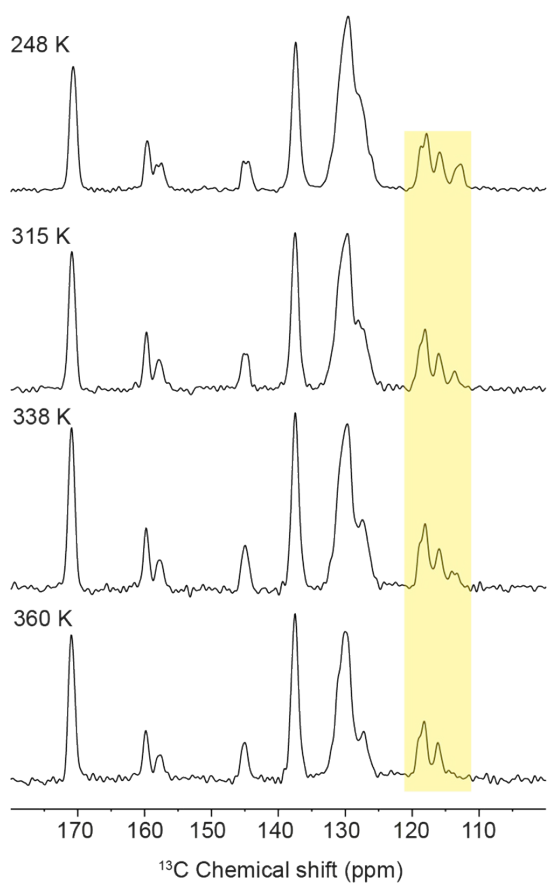


Figure S5. Variable-temperature ^{13}C CPMAS NMR spectra of 1D_1 .

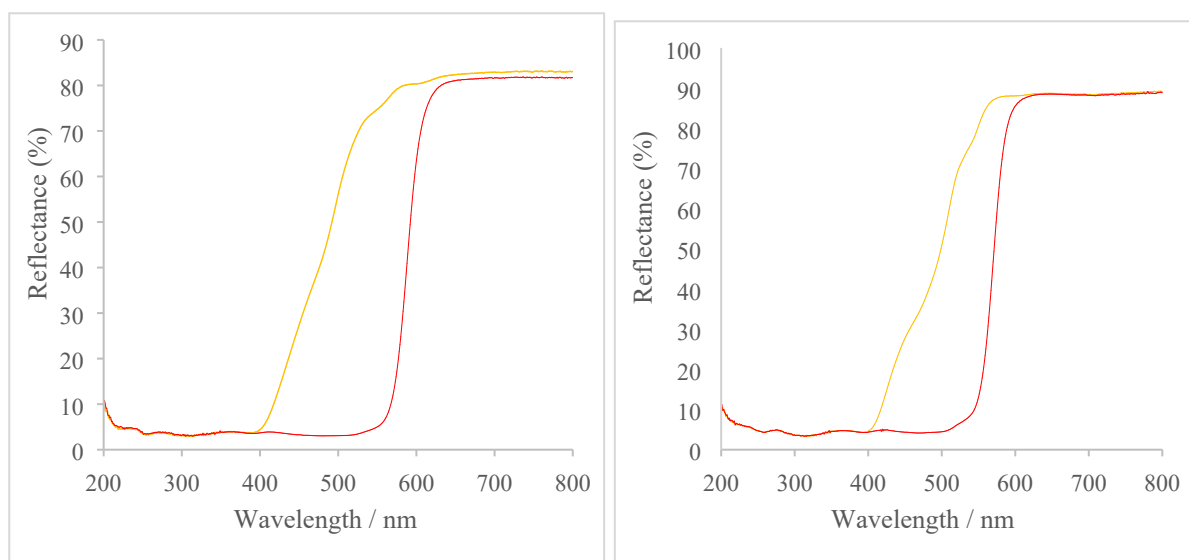


Figure S6. UV-vis reflectance spectra before irradiation (yellow) and after irradiation (red) of A₁ (left) and A₂ (right).

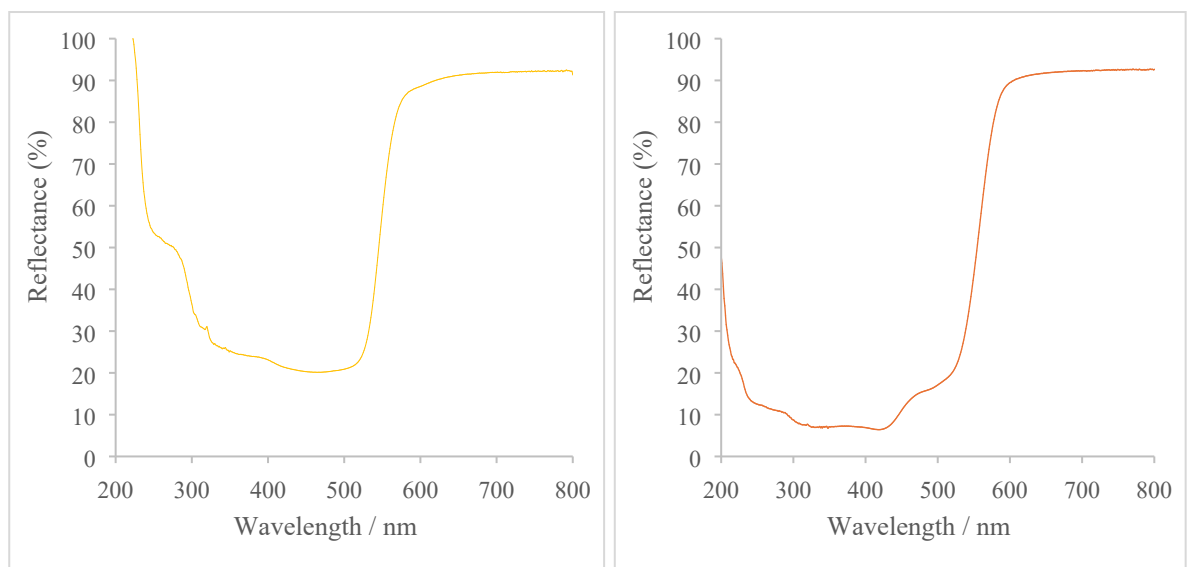


Figure S7. UV-vis reflectance spectra of bulk crystalline A₃ (left) and A₄ (right). UV-vis reflectance profiles remain the same after irradiation.

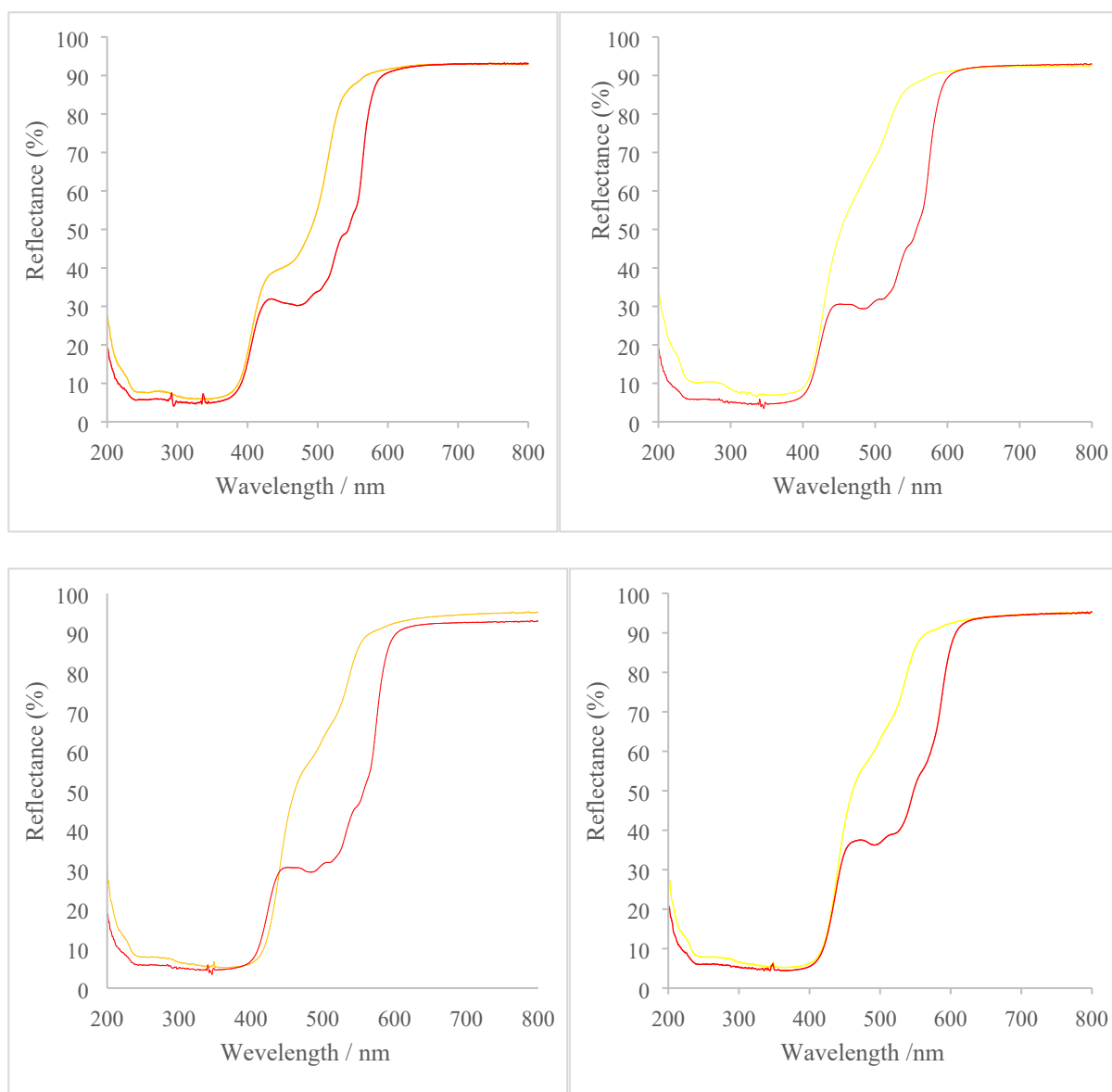


Figure S8. UV-vis reflectance spectra before irradiation (yellow) and after irradiation (red) of $1\text{D}A_1$ (upper left), $1\text{D}A_2$ (upper right), $1\text{D}A_3$ (lower left), and $1\text{D}A_4$ (lower right).

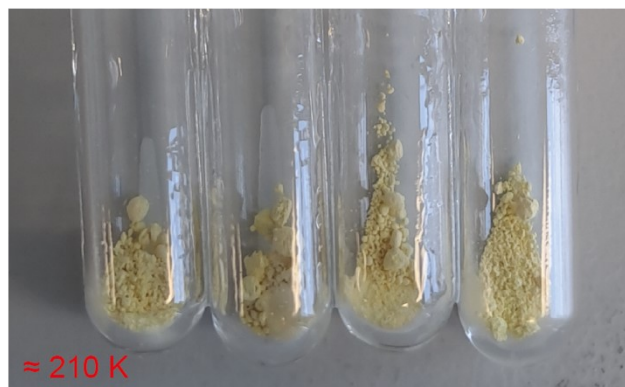
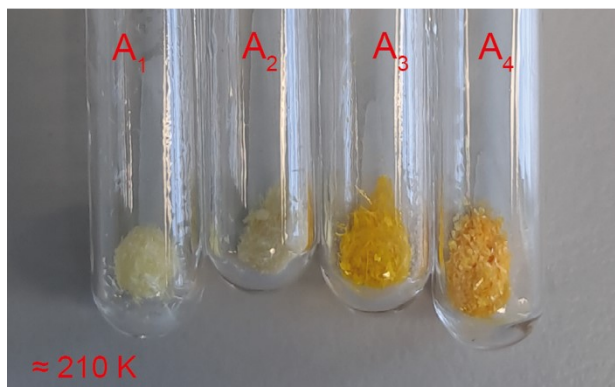
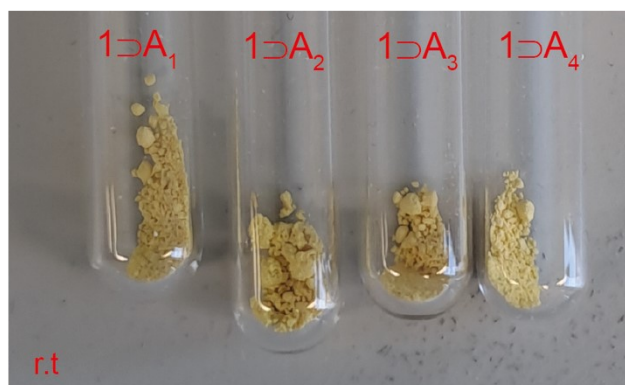
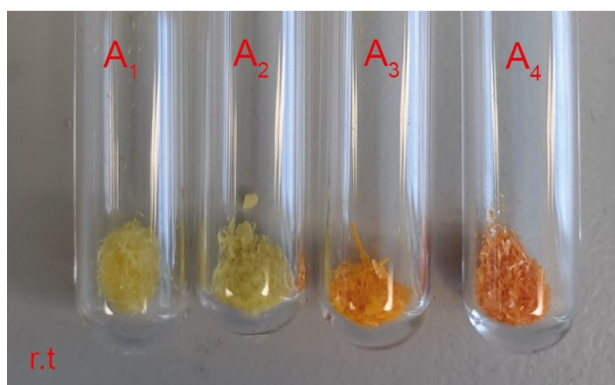


Figure S9. (Left) thermochromic properties of A_1 - A_4 at room temperature (top) and 210 K (bottom). (Right) thermochromic properties of $1>A_1$ - $1>A_4$ at room temperature (top) and 210 K (bottom).