

Efficient organic-inorganic heterojunction structure for enhancing the photocatalytic activity of SubPc /Ti₃C₂T_x towards hydrogen production

Xiaoying Yu,^{a,b,c} Tianfang Zheng,^b Yuanlin Li,^b Yanxiang Liu,^b Ping Guo,^{*a,c} Hai-Hua Wang,^{*b} Shin-ichi Sasaki,^{*d} and Xiao-Feng Wang,^{*b}

^aKey Laboratory of Groundwater Resources and Environment Ministry of Education, College of New Energy and Environment, Jilin University, Changchun 130012, P. R. China

^bKey Laboratory of Physics and Technology for Advanced Batteries (Ministry of Education), College of Physics, Jilin University, Changchun 130012, P. R. China

^c Jilin Provincial Key Laboratory of Water Resources and Environment, Jilin University, Changchun 130012, P. R. China

^d Faculty of Bioscience, Nagahama Institute of Bio-Science and Technology, Nagahama, Shiga 526-0829, Japan

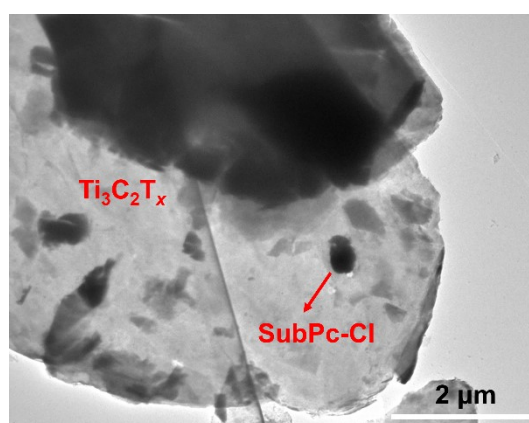


Fig.S1 The TEM image of the SubPc-Cl/Ti₃C₂T_x heterojunction

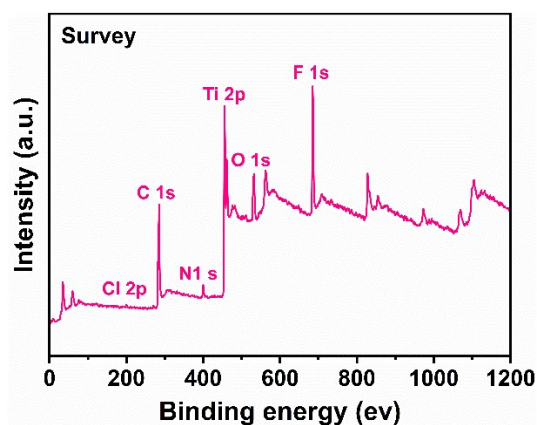


Fig.S2 XPS survey spectra of SubPc-Cl/Ti₃C₂T_x

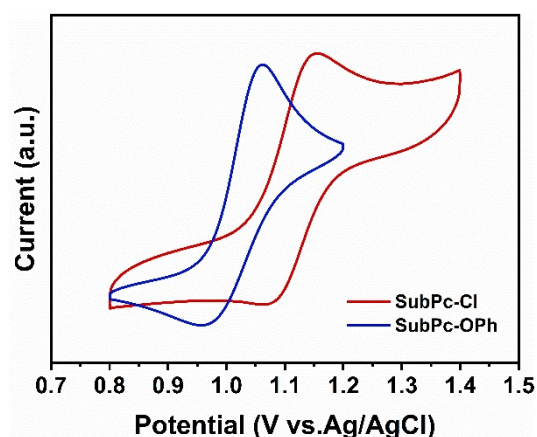


Fig.S3 CVs of SubPc-Cl and SubPc-OPh in CH₂Cl₂ with 0.1 M TBAPF₆ as a supporting electrolyte at a scan rate of 100 mV/s (vs. Ag/AgCl).

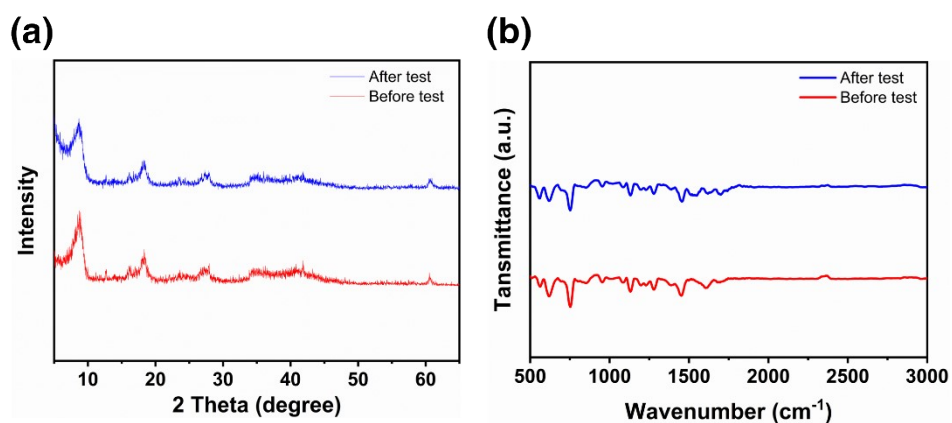


Fig.S4 The XRD and FT-IR spectra of the 2% SubPc-Cl/Ti₃C₂T_x before and after the cycling experiments

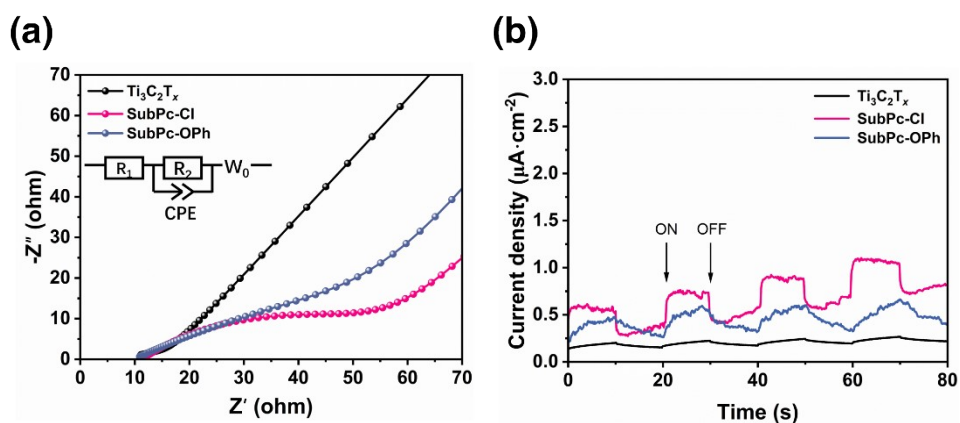


Fig.S5 The EIS Nyquist plots and TPC responses of pure SubPc (SubPc-Cl and SubPc-OPh) and $\text{Ti}_3\text{C}_2\text{T}_x$ samples

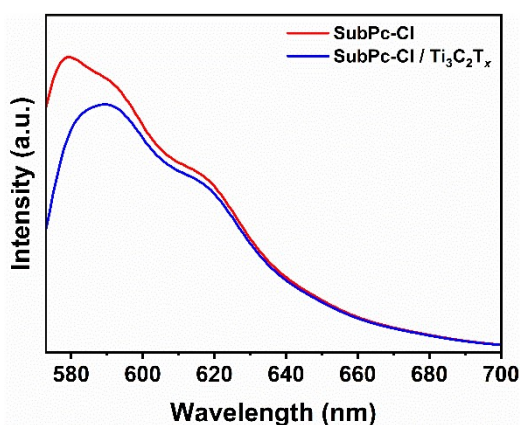


Fig.S6 The PL spectra of SubPc-Cl and SubPc-Cl/ $\text{Ti}_3\text{C}_2\text{T}_x$

Table S1. Comparison of Dye@MXene photocatalysts for hydrogen production

Photocatalyst	Light-source	Sacrificial reagent	H_2 production ($\mu\text{mol g}^{-1} \text{h}^{-1}$)	Rrf.
SubPc/ $\text{Ti}_3\text{C}_2\text{T}_x$	350 W Xenon lamp	AA	105.0	This work
Car@ $\text{Ti}_3\text{C}_2\text{T}_x$	350 W Xenon lamp	AA	84.4	1
SQ@ $\text{Ti}_3\text{C}_2\text{T}_x$	350 W Xenon lamp	AA	28.6	2
BChl@ $\text{Ti}_3\text{C}_2\text{T}_x$	350 W Xenon lamp	AA	51.0	3
Chl@ $\text{Ti}_3\text{C}_2\text{T}_x$	350 W Xenon lamp	AA	52 ± 5	4

Table S2. H_2 production ($\mu\text{mol g}^{-1} \text{h}^{-1}$)

Mass ratio with $\text{Ti}_3\text{C}_2\text{T}_x$ (%)	SubPc-Cl	SubPc-Oph
0.5	50.0	23.0
2	105.0	92.0
4	80.0	56.0

8	41.0	33.0
---	------	------

Table S3. Cycling test results of SubPc-Cl/Ti₃C₂T_x photocatalyst

H ₂ production of 2% SubPc-Cl/Ti ₃ C ₂ T _x	
Run1	419.84
Run2	402.33
Run3	380.62
Run4	350.81
Run5	340.14

Reference

1. Y. Liu, Y. Li, Y.-T. Xu, H. Xu, Y. Gao, X.-F. Wang, R. Fujii and S. Sasaki, *J. Alloy. Compd.*, 2024, 997, 174798.
2. Y. Liu, Y. Li, A. Li, Y. Gao, X.-F. Wang, R. Fujii and S. Sasaki, *J. Colloid Interface Sci.*, 2023, 633, 218-225.
3. Y. Li, Y. Sun, Y. Liu, T. Zheng, A. Li, G. G. Levchenko, W. Han, A. V. Pashchenko, S. Sasaki, H. Tamiaki and X.-F. Wang, *J. Colloid Interface Sci.*, 2024, 654, 1001-1009.
4. Y. Li, X. Chen, Y. Sun, X. Meng, Y. Dall'Agnese, G. Chen, C. Dall'Agnese, H. Ren, S. Sasaki, H. Tamiaki and X.-F. Wang, *Adv. Mater. Interfaces*, 2020, 7, 1902080.