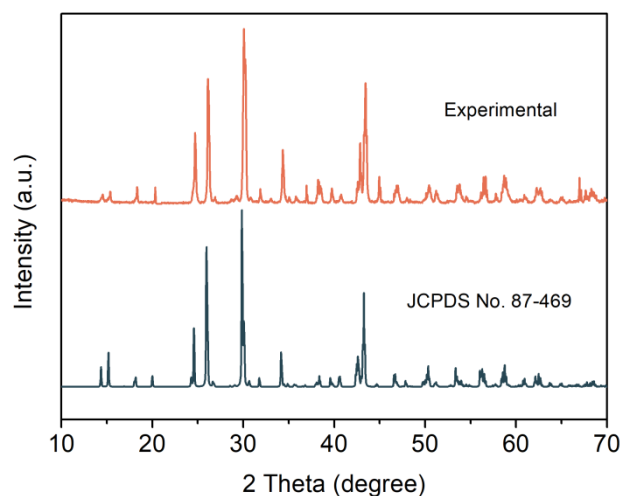


**Growth, characterization and theoretical analysis of  $\alpha$ -SrGeO<sub>3</sub> as a candidate  
mid-IR stimulated Raman scattering crystal**

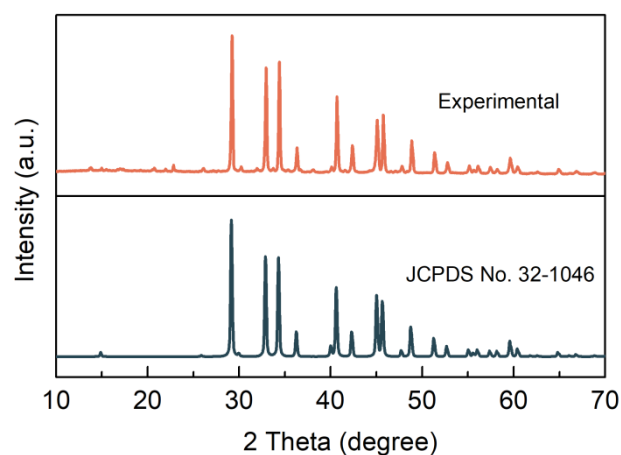
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Synthesis of polycrystalline raw materials for growth of the  $\alpha$ -SrGeO<sub>3</sub> crystal

**Fig. S1** Experimental and standard patterns of  $\alpha$ -SrGeO<sub>3</sub> [Joint Committee on Powder Diffraction Standards (JCPDS) card No. 87-469].



**Fig. S2** Experimental and standard patterns of NaBO<sub>2</sub> (JCPDS No. 32-1046).

Polycrystalline raw materials of  $\alpha$ -SrGeO<sub>3</sub> and NaBO<sub>2</sub> were synthesized via high-temperature solid-state reactions. All of the chemical reagents were commercially available and used without further treatment: (1) SrCO<sub>3</sub>, 99.0%, Sinopharm Chemical Reagent Co., Ltd.; (2) GeO<sub>2</sub>, 99.99%, Xiangkang Technology Development Co., Ltd., China; (3) Na<sub>2</sub>CO<sub>3</sub>, 99.8%, Sinopharm Chemical Reagent Co., Ltd.; (4) H<sub>3</sub>BO<sub>3</sub>, 99.5%, Yunling Chemical Reagent Co., China. The chemical reagents in stoichiometric ratios were weighed, mixed, and fully ground in a mortar. Then, the mixture was transferred to a corundum crucible and heated in a furnace. For  $\alpha$ -SrGeO<sub>3</sub>, the mixture was heated at 1000°C for 24 h with an intermediate grinding. For NaBO<sub>2</sub>, the mixture was heated at 600°C for the first 12 hours. Then, the obtained intermediate product was reground and calcined at 650°C for the second 12 hours.

The polycrystalline products were identified by the XRD technology. The results are shown in Figs. S1 and S2, responding to polycrystalline  $\alpha$ -SrGeO<sub>3</sub> and NaBO<sub>2</sub>, respectively. The consistency of the experimental and the standard patterns indicates that the synthesized products are  $\alpha$ -SrGeO<sub>3</sub> and NaBO<sub>2</sub>.

**Table S1.** Convergence test for the total energy of  $\alpha$ -SrGeO<sub>3</sub> with respect to the  $k$ -point grids (the energy cutoff is fixed at 1000 eV)

$k$ -point	total energy (eV)
$1 \times 1 \times 1$	-26906.6557
$1 \times 2 \times 1$	-26906.8228
$1 \times 2 \times 2$	-26906.8363

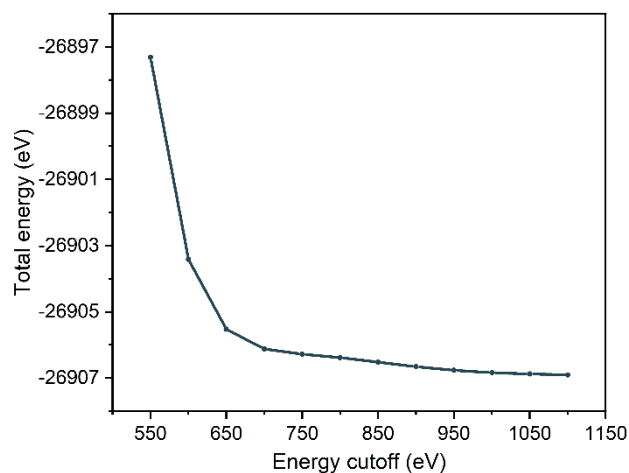
**Fig. S3** Convergence test for the total energy of  $\alpha$ -SrGeO<sub>3</sub> with respect to the energy cutoffs. The  $k$ -point grid is fixed at  $1 \times 2 \times 1$ .

Table S1 and Fig. S3 display the results of convergence tests for the total energy of  $\alpha$ -SrGeO<sub>3</sub> with respect to the  $k$ -point grids and the energy cutoffs, respectively. The results indicate that a  $k$ -point grid of  $1 \times 2 \times 1$  and an energy cutoff of 1000 eV are sufficient to ensure the total energy convergence to within 1 meV/atom.

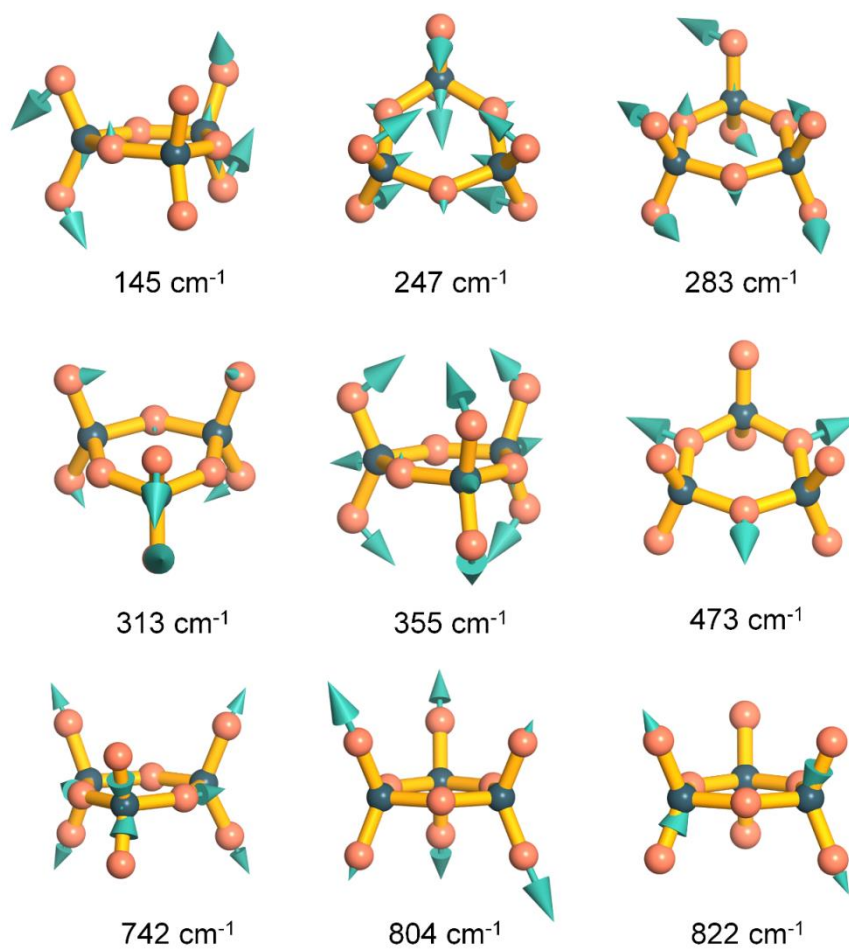
# Electronic Supplementary Information

**Table S2.** Vibrational modes and their corresponding frequencies of the  $\alpha$ -SrGeO<sub>3</sub> crystal

No.	$\omega$ (cm <sup>-1</sup> )	mode	No.	$\omega$ (cm <sup>-1</sup> )	mode	No.	$\omega$ (cm <sup>-1</sup> )	mode
1	-0.03	B <sub>u</sub> *	31	147.52	B <sub>u</sub>	61	360.34	A <sub>u</sub>
2	-0.02	B <sub>u</sub> *	32	151.11	B <sub>g</sub>	62	360.63	B <sub>g</sub>
3	-0.02	A <sub>u</sub> *	33	154.72	B <sub>u</sub>	63	375.14	A <sub>g</sub>
4	65.49	B <sub>u</sub>	34	172.10	A <sub>u</sub>	64	376.61	B <sub>u</sub>
5	73.79	B <sub>g</sub>	35	176.42	B <sub>u</sub>	65	392.33	B <sub>u</sub>
6	78.64	A <sub>u</sub>	36	181.01	B <sub>g</sub>	66	423.60	B <sub>g</sub>
7	82.40	A <sub>u</sub>	37	192.22	B <sub>u</sub>	67	471.68	A <sub>u</sub>
8	84.23	B <sub>u</sub>	38	193.25	A <sub>g</sub>	68	473.36	A <sub>g</sub>
9	88.77	A <sub>g</sub>	39	195.24	A <sub>u</sub>	69	493.50	A <sub>g</sub>
10	94.22	B <sub>g</sub>	40	197.11	B <sub>g</sub>	70	494.66	B <sub>u</sub>
11	97.51	B <sub>g</sub>	41	243.90	A <sub>u</sub>	71	497.05	A <sub>u</sub>
12	98.44	A <sub>u</sub>	42	246.69	A <sub>g</sub>	72	498.48	B <sub>g</sub>
13	100.51	B <sub>u</sub>	43	254.64	A <sub>u</sub>	73	736.43	B <sub>u</sub>
14	103.76	B <sub>u</sub>	44	256.09	A <sub>g</sub>	74	740.55	B <sub>g</sub>
15	104.39	A <sub>g</sub>	45	262.67	B <sub>u</sub>	75	741.98	A <sub>g</sub>
16	109.51	A <sub>u</sub>	46	264.34	A <sub>u</sub>	76	744.57	A <sub>u</sub>
17	112.76	B <sub>u</sub>	47	265.45	B <sub>u</sub>	77	775.75	B <sub>g</sub>
18	113.98	B <sub>g</sub>	48	269.10	A <sub>u</sub>	78	775.78	B <sub>u</sub>
19	114.05	A <sub>g</sub>	49	270.77	A <sub>g</sub>	79	803.59	A <sub>g</sub>
20	119.54	B <sub>u</sub>	50	279.63	B <sub>g</sub>	80	805.99	B <sub>u</sub>
21	119.97	A <sub>u</sub>	51	281.72	B <sub>g</sub>	81	807.40	A <sub>u</sub>
22	121.05	B <sub>g</sub>	52	283.42	A <sub>g</sub>	82	816.15	A <sub>u</sub>
23	129.32	A <sub>g</sub>	53	302.86	B <sub>u</sub>	83	821.40	B <sub>g</sub>
24	132.53	A <sub>u</sub>	54	310.57	B <sub>g</sub>	84	821.86	A <sub>g</sub>
25	133.96	A <sub>g</sub>	55	312.83	A <sub>g</sub>	85	823.30	B <sub>u</sub>
26	137.84	B <sub>g</sub>	56	313.03	B <sub>u</sub>	86	850.73	B <sub>u</sub>
27	142.62	A <sub>u</sub>	57	316.44	A <sub>u</sub>	87	850.91	B <sub>g</sub>
28	142.88	B <sub>u</sub>	58	320.24	B <sub>g</sub>	88	852.71	A <sub>g</sub>
29	144.93	A <sub>g</sub>	59	354.01	A <sub>u</sub>	89	853.37	A <sub>u</sub>
30	145.19	B <sub>g</sub>	60	354.53	A <sub>g</sub>	90	883.73	B <sub>g</sub>

\*Acoustic modes

The Raman-active modes are highlighted in red.



**Fig. S4** Nine characteristic Raman vibrational modes of the  $\alpha$ -SrGeO<sub>3</sub> crystal. All of the modes are the A<sub>g</sub> mode; all of the frequencies are the computational values given in Table S2.