

## Supplementary Information

**Table S1.** Ionic radii for cation ions under corresponding coordination numbers.

Ion	Coordination number (CN)	Ionic radius (Å)
Ga <sup>3+</sup>	6	0.62
Ge <sup>4+</sup>	6	0.39
Zn <sup>2+</sup>	6	0.74
Cr <sup>3+</sup>	6	0.615

**Table S2.** Main parameters of processing and refinement of CaZrTaGa<sub>1-2x</sub>Ge<sub>x</sub>Zn<sub>x</sub>O<sub>7</sub>:0.01Cr<sup>3+</sup> ( $x = 0-0.5$ ).

Sample	$x = 0$	$x = 0.1$	$x = 0.2$	$x = 0.3$	$x = 0.4$	$x = 0.5$
Space group	<i>P2/c</i>	<i>P2/c</i>	<i>P2/c</i>	<i>P2/c</i>	<i>P2/c</i>	<i>P2/c</i>
<i>a</i> (Å)	12.7070	12.6966	12.6778	12.6826	12.6799	12.6759
<i>b</i> (Å)	7.3596	7.3322	7.3446	7.3506	7.3457	7.3489
<i>c</i> (Å)	11.4500	11.4182	11.4216	11.4276	11.4238	11.4264
<i>V</i> (Å <sup>3</sup> )	1051.369	1044.772	1044.104	1046.009	1044.616	1045.133
$\beta$ (°)	100.928	100.615	100.962	100.930	100.964	100.922
$R_{wp}$ (%)	7.34	9.87	6.47	8.69	8.62	8.45
$R_p$ (%)	5.37	5.89	4.37	6.23	5.87	6.75
$\chi^2$	1.821	1.425	1.432	1.627	1.634	1.892

**Table S3.** Emission intensity ratio of 423 K and 298 K ( $I_{423}/I_{298}$ ) and quantum efficiency (IQE and EQE) for some Cr<sup>3+</sup>-doped phosphors with emission wavelength in 760-850 nm, as well as NIR output power and conversion efficiency of the corresponding fabricated *pc*-LEDs.

No.	Phosphor	$\lambda_{em}$ (nm)	FWHM (nm)	IQE (%)	$I_{423}/I_{298}$ (%)	Ref
1	$\text{Na}_3\text{In}_2\text{Li}_3\text{F}_{12}:\text{Cr}^{3+}$	778	121	33	44	1
2	$\text{La}_3\text{Sc}_2\text{Ga}_3\text{O}_{12}:\text{Cr}^{3+}$	818	145	35	60	2
3	$\text{ScBO}_3:\text{Cr}^{3+}$	800	120	65	51	3
4	$\text{BaMgAl}_{10}\text{O}_{17}:\text{Cr}^{3+}$	762	92.6	94	53	4
5	$\text{GaTaO}_4:\text{Cr}^{3+}$	850	266	82.6	73	5
6	$\text{Ca}_2\text{LuZr}_2\text{Al}_3\text{O}_{12}:\text{Cr}^{3+}$	780	117	69.1	60	6
7	$\text{Gd}_3\text{Zn}_{0.8}\text{Ga}_{3.4}\text{Ge}_{0.8}\text{O}_{12}:\text{Cr}^{3+}$	800	105	79.6	40.2	7
8	$\text{Gd}_3\text{In}_2\text{Ga}_3\text{O}_{12}:\text{Cr}^{3+}$	780	124	85.3	87.7	8
9	CZTGGZO: 0.01Cr <sup>3+</sup> phosphor	812	250.39	94.37	88.56	This work
10	CZTGGZO: 0.01Cr <sup>3+</sup> PiG film	812	248.37	90.20	92.34	This work

According to the Tanabe-Sugano plot, the energy level distribution of Cr<sup>3+</sup> at octahedral positions is displayed. Therefore, the corresponding crystal field parameters of Cr<sup>3+</sup> are calculated as follows (Figure S3)<sup>9</sup>:

$$D_q = \frac{E(\uparrow)_2}{10} \quad (\text{S1})$$

$$B = D_q \frac{x^2 - 10x}{15(x - 8)} \quad (\text{S2})$$

$$x = \frac{E(^4T_1 - ^4T_2)}{D_q} \quad (\text{S3})$$

where  $D_q$  is the crystal field strength,  $B$  is the Racah parameter denoting the repulsive force between electrons in the three-dimensional orbitals, and  $E$  denotes the energy difference between the  $^4T_1$  and  $^4T_2$  energy levels determined by the positions of the  $^4T_1(^4F)$  and  $^4F_2(^4F)$  peaks. When  $x = 0.4$  and  $D_q/B = 2.9$ , it indicates that the octahedral site where  $\text{Cr}^{3+}$  is located favors broadband emission of  $\text{Cr}^{3+}$ .

The Kubelka-Munk equation is used to determine the band gap of the sample<sup>2, 10</sup>:

$$F(R) = \frac{(1-R)^2}{2R} \quad (\text{S4})$$

$$Ah\nu = c(h\nu - E_g)^{\frac{1}{2}} \quad (\text{S5})$$

$$[h\nu F(R_\infty)]^2 = A(h\nu - E_g) \quad (\text{S6})$$

where  $E_g$  is the optical band gap,  $c$  is the absorption constant,  $h\nu$  is the photon energy, and  $R$  is the reflection coefficient. The  $E_g$  value was calculated as 1.64 eV for CZTGO: 0.01Cr<sup>3+</sup> as shown in Figure S2c.

The decay curves of the phosphor and PiG film monitored at 827 nm under 461 nm excitation can be fitted according to the following equations<sup>11</sup>:

$$I(t) = I_0 + A_1 \exp\left(-\frac{t}{t_1}\right) + A_2 \exp\left(-\frac{t}{t_2}\right) \quad (\text{S7})$$

Where  $t$  is the time,  $A_1$  and  $A_2$  stand for constants, and  $I(t)$  and  $I_0$  represent the luminous intensity at times  $t$  and 0. Then, the lifetime of the luminescent center can be calculated by the following equation<sup>11, 12</sup>:

$$\tau = \frac{\int_0^\infty tI(t)dt}{\int_0^\infty I(t)dt} \quad (\text{S8})$$

Where  $\tau$  is the fluorescence lifetime.

Equation S9 fits the energy difference  $\Delta E_a$ , as well as activation energy, between the lowest excited state and the crossover point location<sup>13</sup>.

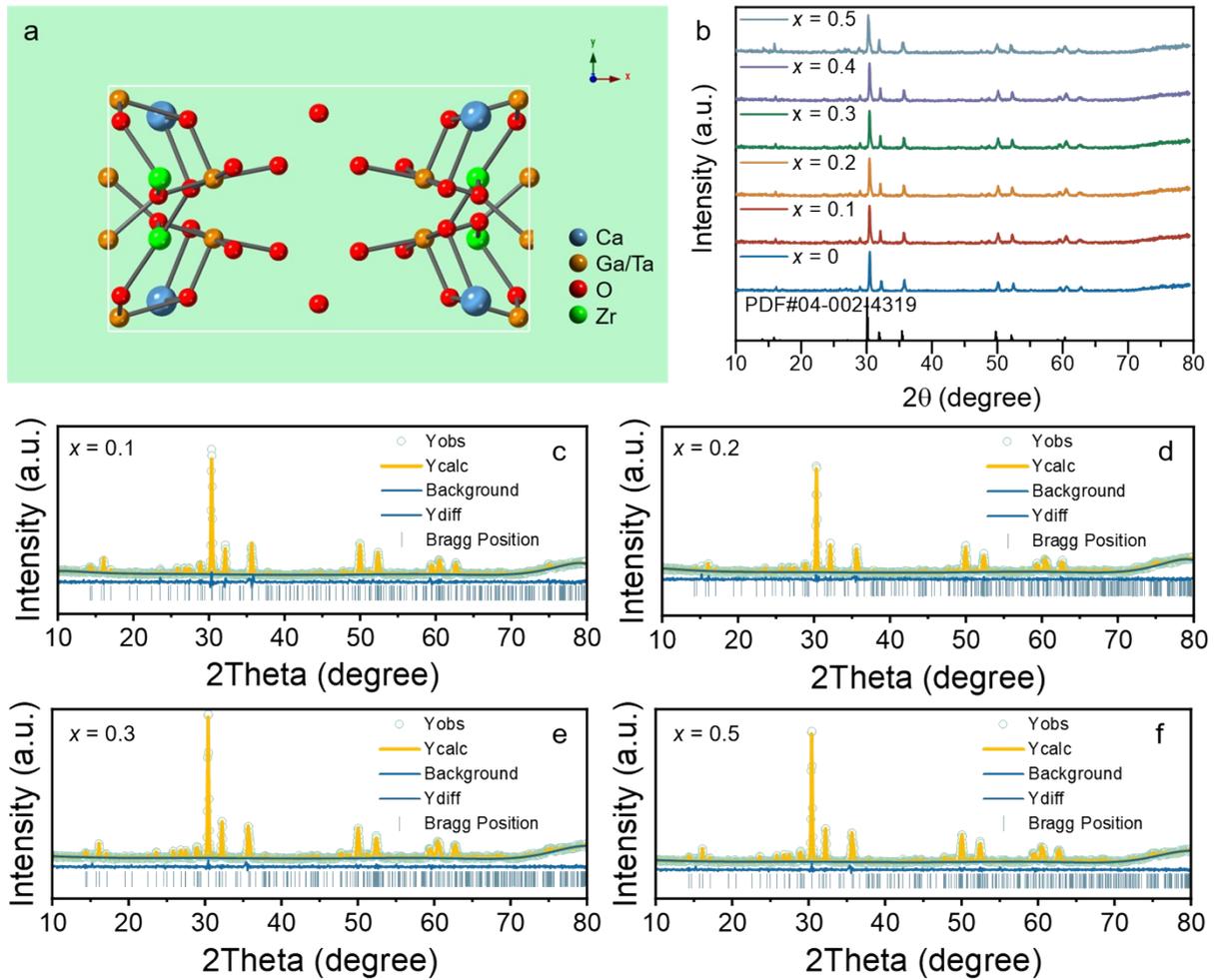
$$I_T = \frac{I_0}{1 + c * \exp(-\frac{E_a}{kT})} \quad (S9)$$

where  $C$  is a constant that is independent of temperature,  $k$  is the Boltzmann constant, and  $I_0$  and  $I_t$  are, respectively, the starting and the intensity of luminescence at temperature  $T$ .

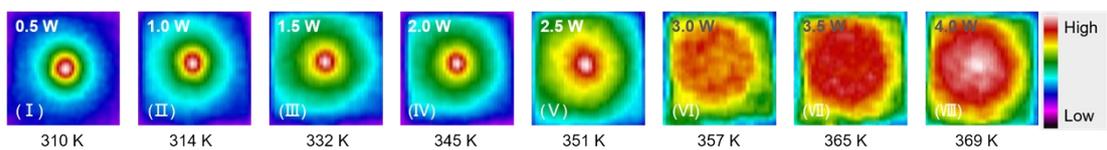
The luminescent spectra for QE measurement ( $S_1$  part) were only recorded till 850 nm due to the range limitation of the instrument. The missing  $S_2$  part (850–1100) should be taken into consideration for the actual IQE. Therefore, the actual IQE can be calculated via the following equation<sup>14</sup>:

$$IQE = IQE_m \times (S_1 + S_2) / S_1 \quad (S4)$$

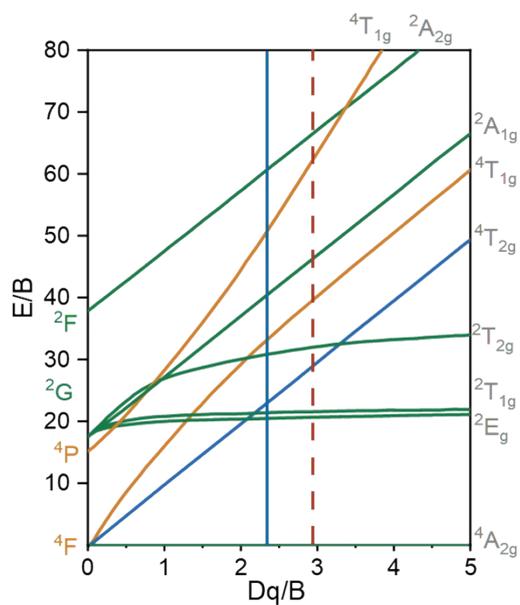
where IQE and  $IQE_m$  are the actual IQE and measured IQE.  $S_1$  (550–850 nm) and  $S_2$  (850–1100 nm) are integrated intensities, which are obtained from PL spectra. The actual IQEs of  $CaZrTaGa_{0.2}Ge_{0.4}Zn_{0.4}O_7:0.01Cr^{3+}$  phosphor and PiG film are determined to be 55.67% and 54.12%, respectively.



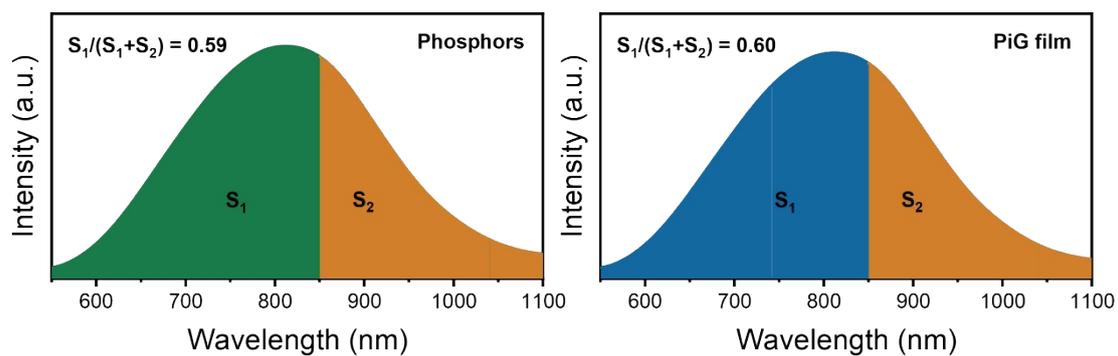
**Figure S1.** (a) Projection of the crystal structure of  $\text{CaZrTaGaO}_7$  along the z axis. (b) XRD patterns of the  $\text{CaZrTaGa}_{1-2x}\text{Ge}_x\text{Zn}_x\text{O}_7:0.01\text{Cr}^{3+}$  samples ( $x = 0-0.5$ ). (c) - (f) Rietveld refinement results of the synthesized  $\text{CaZrTaGa}_{1-2x}\text{Ge}_x\text{Zn}_x\text{O}_7:0.01\text{Cr}^{3+}$  ( $x = 0.1, 0.2, 0.3, 0.5$ ).



**Figure S2.** (d) Infrared thermal images of 20 s are taken at a laser power of 1-4 W under blue LD conditions.



**Figure S3.** The Tanabe–Sugano diagram.



**Figure S4.** Comparison between IQE measuring spectra and PL spectra for  $\text{CaZrTaGa}_{0.2}\text{Ge}_{0.4}\text{Zn}_{0.4}\text{O}_7:0.01\text{Cr}^{3+}$  phosphor and PiG film.

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