

SUPPORTING INFORMATION

Unravelling Structure - Luminescence Relationship in Two Dimensional Antimony(III) Doped Cadmium (II) Halide Hybrids

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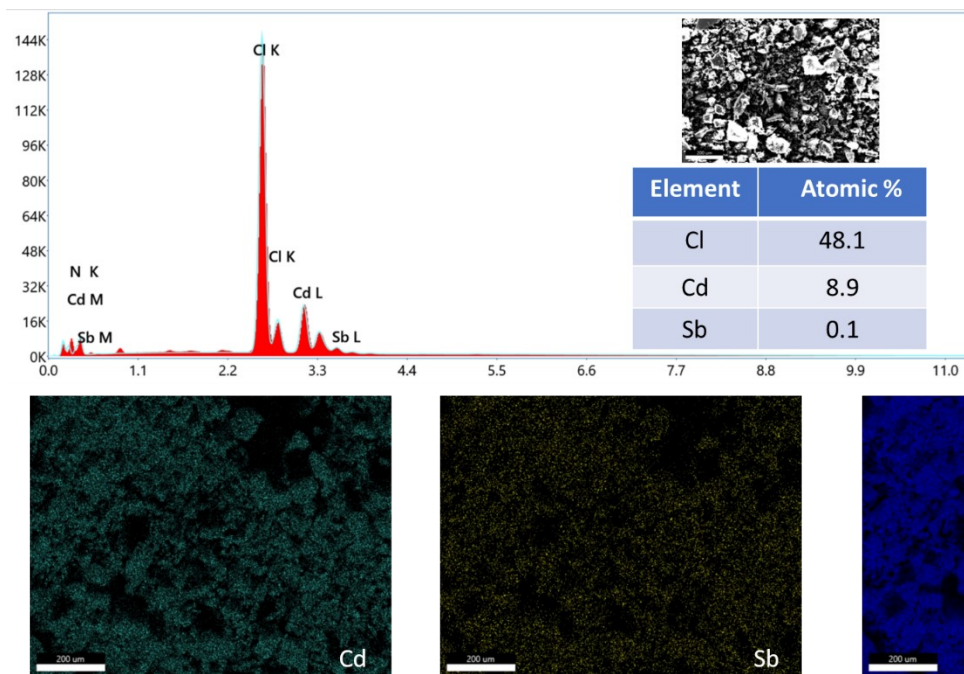


Figure S1. SEM EDS analysis of Sb:DETACdCl hybrid.

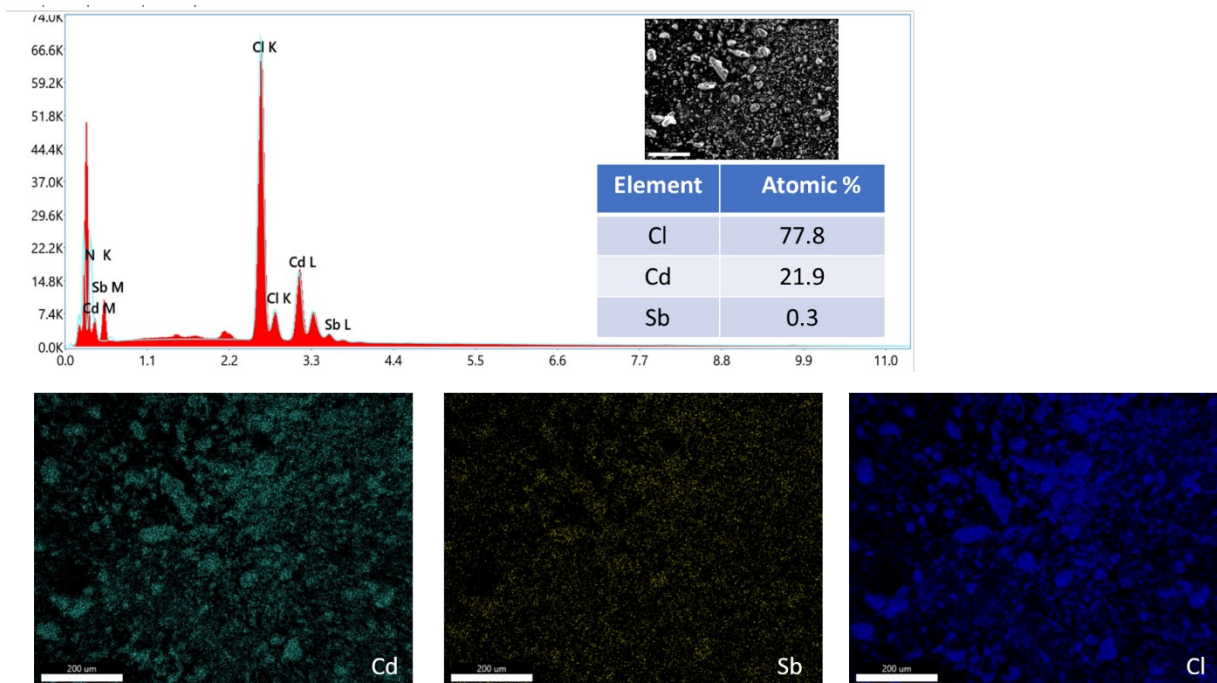


Figure S2. SEM EDS analysis of Sb:PDACdCl hybrid.

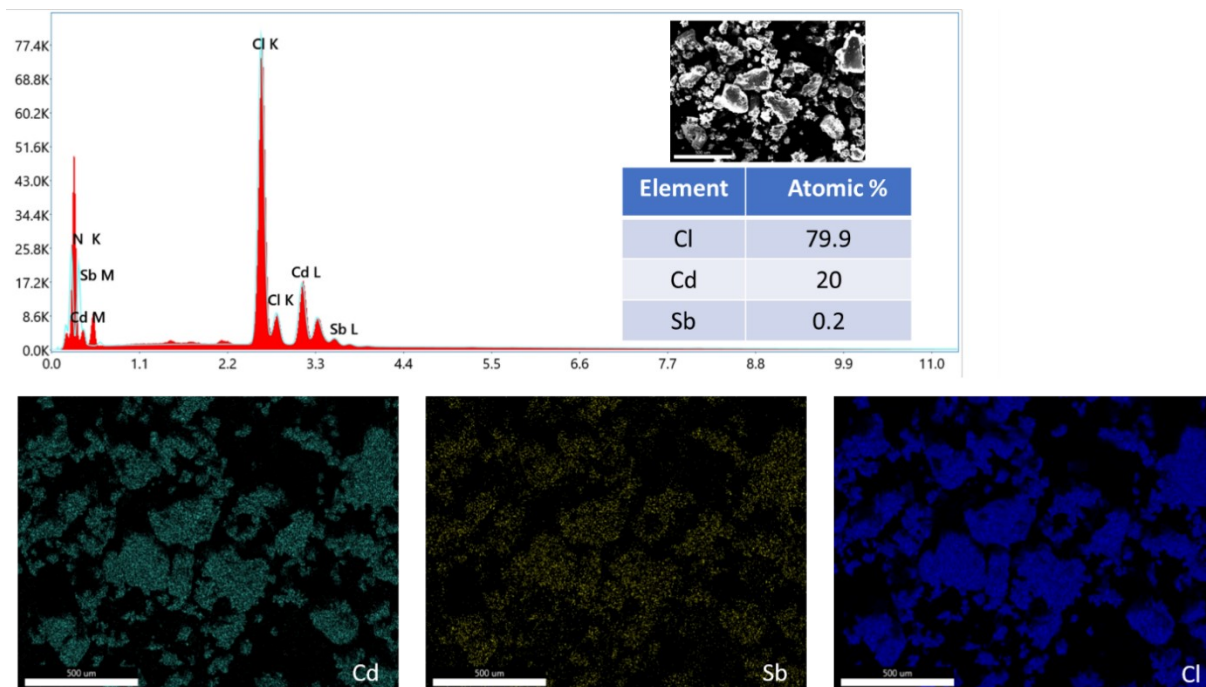


Figure S3. SEM EDS analysis of Sb:DABCdCl hybrid.

Table S1. ICP Analysis of doped hybrids (Sb:DETACdCl, Sb:PDACdCl, Sb:DABCdCl)

Compound	Sb feed percentage	Sb percentage from ICP
Sb: PDACdCl	10%	0.0008
	30%	0.0062
	50%	0.0110
Sb: DETACdCl	10%	0.0028
	30%	0.0033
	50%	0.0365
Sb: DABCdCl	10%	0.0027
	30%	0.0167
	50%	0.0063

Table S2: Comparison of structural and optical properties of host hybrids.

Compound	Space group	Octahedral distortion (d)	Bond angle variance (deg ²)	M-Cl-M (deg)	Φ_{tilt} (deg)	PL Max (nm)	PLE Peak positions (nm)	PLQY(%)
DETACdCl C ₄ H ₁₆ N ₃ Cd Cl ₅	Orthorhombic <i>Pnma</i>	8.51×10^{-5}	2.1785	153.48	11.25	490	274, 330, 380	3.5
PDACdCl C ₅ H ₁₆ N ₂ Cd Cl ₄	Orthorhombic <i>Imma</i>	9.22×10^{-4}	1.4491	168.40	12.49	480	398	0.5
DABCdCl 2(C ₄ H ₁₄ N ₂) Cd ₂ Cl ₃ O	Monoclinic <i>P2₁/c</i>	1.4×10^{-3}	0.6413	165.64	13.03	465	280, 330, 375	3.6

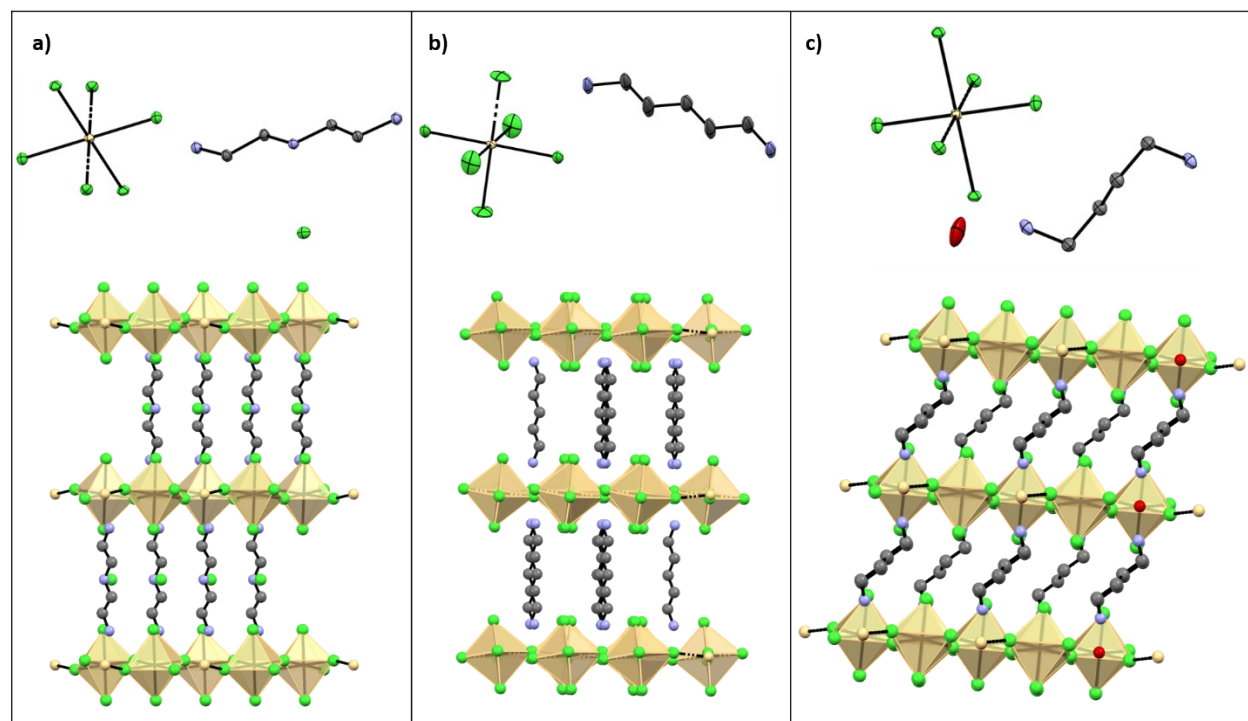


Figure S4: Asymmetric unit and packing diagram of host hybrids- a) DETACdCl, b) PDACdCl, c) DABCdCl.

Table S3: SCXRD parameters of doped hybrids

Identification code	Sb:DETACdCl	Sb:PDACdCl	Sb:DABCdCl
Empirical formula	C ₄ H ₁₆ CdCl ₅ N ₃ Sb	C ₅ H ₁₆ CdCl ₄ N ₂ Sb	C ₈ H ₂₈ Cd ₂ Cl ₈ N ₄ O Sb
Formula weight	395.85	358.40	704.74
Temperature/K	150.00	150.00	150.00
Crystal system	orthorhombic	orthorhombic	monoclinic
Space group	Pnma	Imma	P2 ₁ /c
a/Å	7.1960(4)	7.2902(3)	9.4098(4)
b/Å	24.6826(14)	24.0668(12)	7.5645(3)
c/Å	7.2487(4)	7.4496(4)	7.6395(4)
α/°	90	90	90
β/°	90	90	102.01
γ/°	90	90	90
Volume/Å³	1287.49(12)	1307.05(11)	531.89(4)
Z	4	4	1
ρ_{calc}/cm³	2.042	1.821	2.200
μ/mm⁻¹	2.698	2.448	3.009
F(000)	776.0	704.0	344.0
Crystal size/mm³	0.22 × 0.21 × 0.2	0.21 × 0.2 × 0.19	0.21 × 0.2 × 0.2
Radiation	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)
2θ range for data collection/°	5.858 to 52.764	5.724 to 52.736	6.972 to 52.716
Index ranges	-8 ≤ h ≤ 9, -30 ≤ k ≤ 30, -9 ≤ l ≤ 9	-9 ≤ h ≤ 9, -30 ≤ k ≤ 30, -9 ≤ l ≤ 9	-11 ≤ h ≤ 11, -9 ≤ k ≤ 9, -9 ≤ l ≤ 9
Reflections collected	27418	15463	5946
Independent reflections	1343 [R _{int} = 0.0289, R _{sigma} = 0.0122]	742 [R _{int} = 0.0506, R _{sigma} = 0.0203]	1092 [R _{int} = 0.0366, R _{sigma} = 0.0282]
Data/restraints/parameters	1343/0/65	742/0/43	1092/3/70
Goodness-of-fit on F²	1.107	1.093	1.163
Final R indexes [I ≥ 2σ (I)]	R ₁ = 0.0132, wR ₂ = 0.0379	R ₁ = 0.0318, wR ₂ = 0.0939	R ₁ = 0.0251, wR ₂ = 0.0651
Final R indexes [all data]	R ₁ = 0.0138, wR ₂ = 0.0383	R ₁ = 0.0323, wR ₂ = 0.0946	R ₁ = 0.0260, wR ₂ = 0.0655
Largest diff. peak/hole / e Å⁻³	0.37/-0.33	1.65/-0.90	0.57/-0.95

Table S4: SCXRD parameters of host hybrids

Identification code	DETACdCl	PDACdCl	DABCdCl
Empirical formula	C ₄ H ₁₆ CdCl ₅ N ₃	C ₅ H ₁₆ CdCl ₄ N ₂	C ₈ H ₂₈ Cd ₂ Cl ₈ N ₄ O
Formula weight	395.85	358.40	704.781
Temperature/K	150.00	150.00	150.00
Crystal system	orthorhombic	orthorhombic	monoclinic
Space group	Pnma	Imma	P2 ₁ /c
a/Å	7.1977(3)	7.2884(4)	9.4203(6)
b/Å	24.7093(9)	24.0679(13)	7.5615(4)
c/Å	7.2486(3)	7.4490(4)	7.6317(5)
α/°	90	90	90
β/°	90	90	101.956(2)
γ/°	90	90	90
Volume/Å³	1289.16(9)	1306.68(12)	531.83(6)
Z	4	4	1
ρ_{calc}/g/cm³	2.040	1.822	2.201
μ/mm⁻¹	2.695	2.449	3.010
F(000)	776.0	704.0	343.7
Crystal size/mm³	0.22 × 0.21 × 0.21	0.16 × 0.15 × 0.15	0.16 × 0.15 × 0.15
Radiation	MoKα (λ = 0.71073)	MoKα (λ = 0.71073)	Mo Kα (λ = 0.71073)
2θ range for data collection/°	5.858 to 52.768	5.724 to 52.792	4.42 to 52.74
Index ranges	-9 ≤ h ≤ 8, -30 ≤ k ≤ 30, -9 ≤ l ≤ 8	-9 ≤ h ≤ 9, -30 ≤ k ≤ 30, ≤ l ≤ 9	-9 ≤ h ≤ 11, -9 ≤ k ≤ 9, -9 ≤ l ≤ 9
Reflections collected	15415	17999	8524
Independent reflections	1338 [R _{int} = 0.0313, R _{sigma} = 0.0141]	737 [R _{int} = 0.0262, R _{sigma} = 0.0110]	1082 [R _{int} = 0.0429, R _{sigma} = 0.0266]
Data/restraints/parameters	1338/0/65	737/1/48	1082/3/70
Goodness-of-fit on F²	1.137	1.145	1.007
Final R indexes [I > 2σ(I)]	R ₁ = 0.0140, wR ₂ = 0.0366	R ₁ = 0.0205, wR ₂ = 0.0646	R ₁ = 0.0290, wR ₂ = 0.0880
Final R indexes [all data]	R ₁ = 0.0144, wR ₂ = 0.0369	R ₁ = 0.0208, wR ₂ = 0.0650	R ₁ = 0.0293, wR ₂ = 0.0882
Largest diff. peak/hole / e Å⁻³	0.37/-0.43	0.67/-0.66	0.69/-1.10

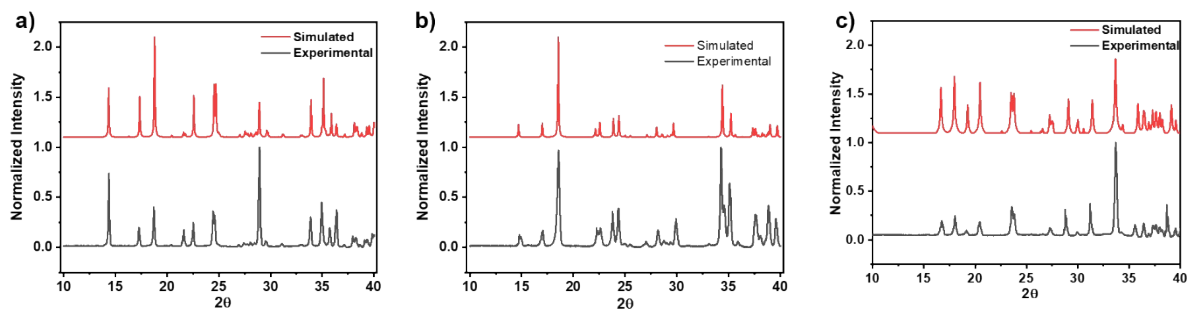


Figure S5: Simulated Vs. experimental PXRD patterns of doped hybrids- a) Sb:DETACdCl, b) Sb:PDACdCl, c) Sb:DABCdCl.

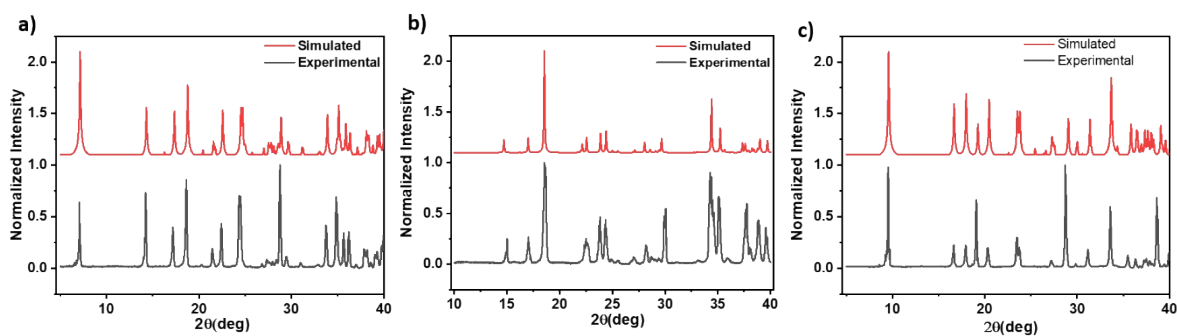


Figure S6: Simulated Vs. experimental PXRD patterns of host hybrids- a) DETACdCl, b) PDACdCl, c) DABCdCl.

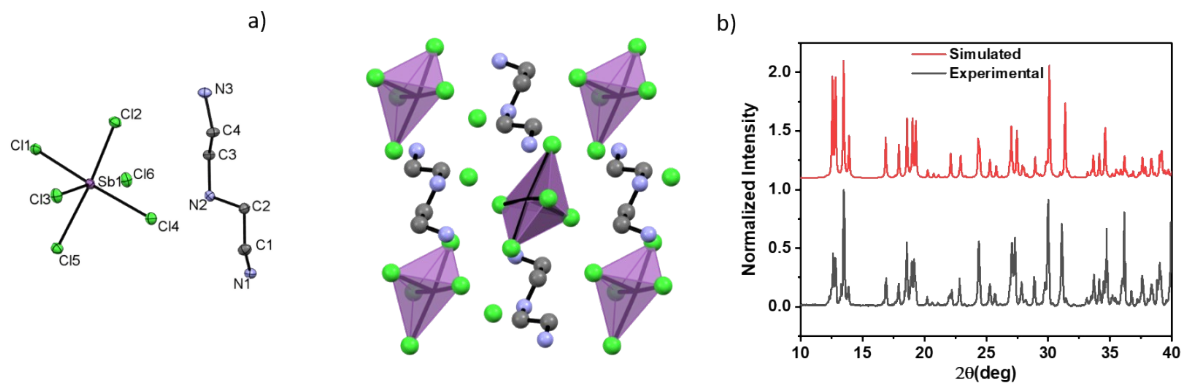


Figure S7: a) SCXRD structure of DETASbCl control hybrid; b) Comparison of simulated and experimental PXRD of DETASbCl control hybrid.

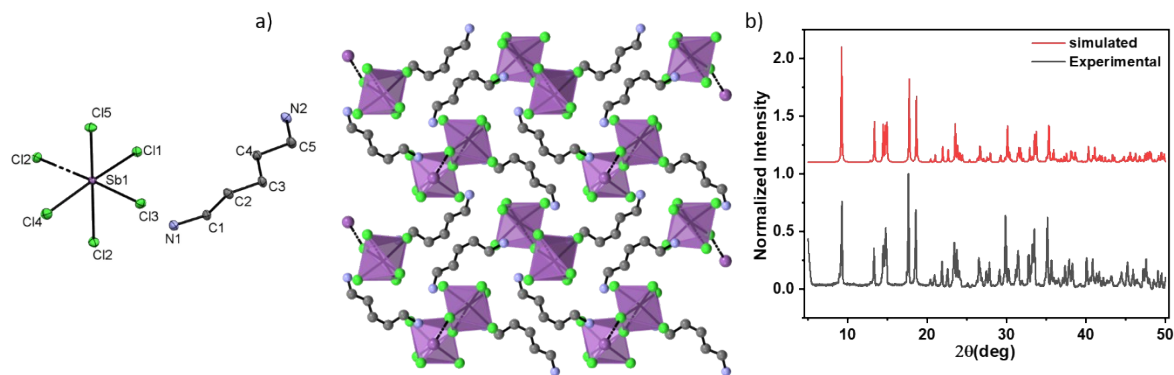


Figure S8: a) SCXRD structure of PDASbCl control hybrid; b) Comparison of simulated and experimental PXRD of PDASbCl control hybrid

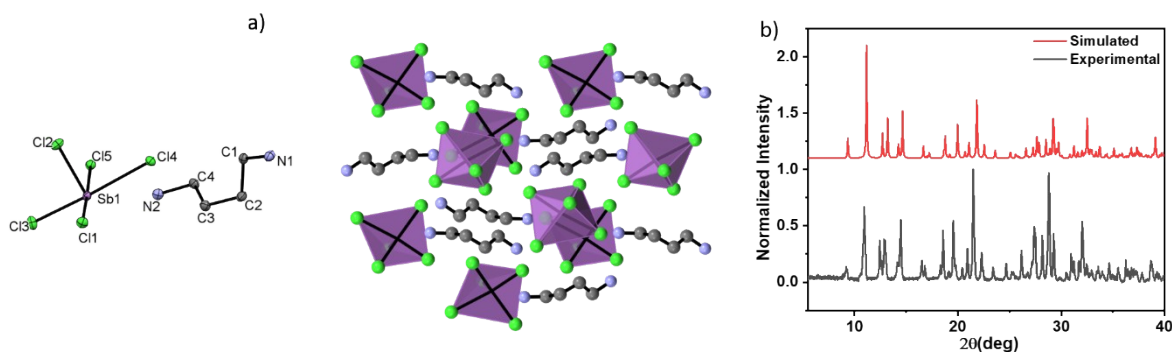


Figure S9: a) SCXRD structure of DABSbCl control hybrid; b) Comparison of simulated and experimental PXRD of DABSbCl control hybrid.

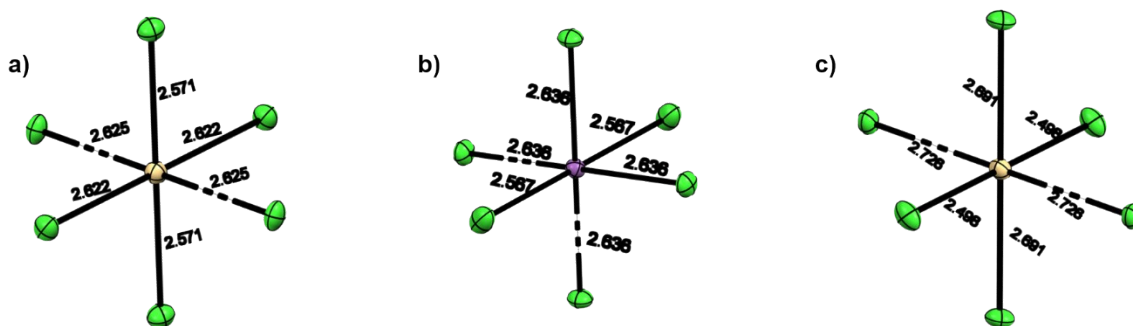


Figure S10. Single crystal XRD structure showing Cd-Cl bond lengths in a) Sb:DETACdCl hybrid, b) Sb:PDACdCl hybrid, c) Sb:DABCdCl hybrid.

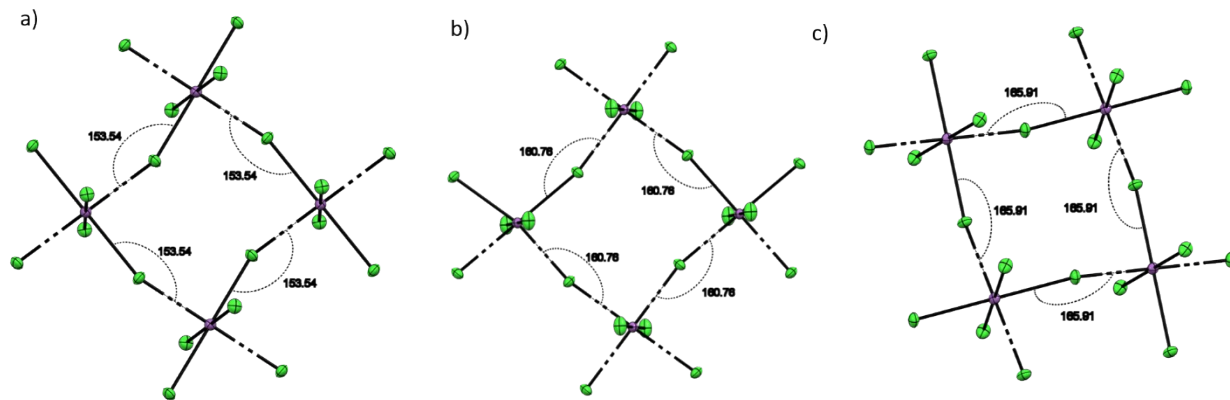


Figure S11: M-X-M bond angles in a) Sb:DETACdCl hybrid, b)Sb:PDACdCl hybrid, c) Sb:DABCdCl hybrid.

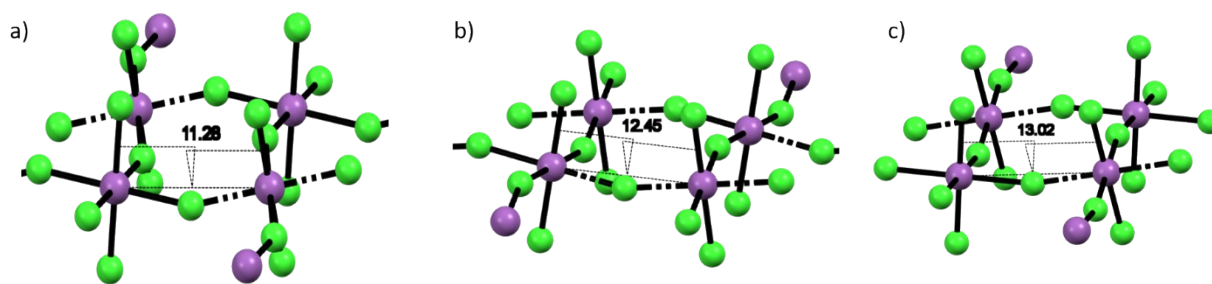


Figure S12: Dihedral angles in a) Sb: DETACdCl hybrid, b) Sb: PDACdCl hybrid, c) Sb:DAbCdCl hybrid.

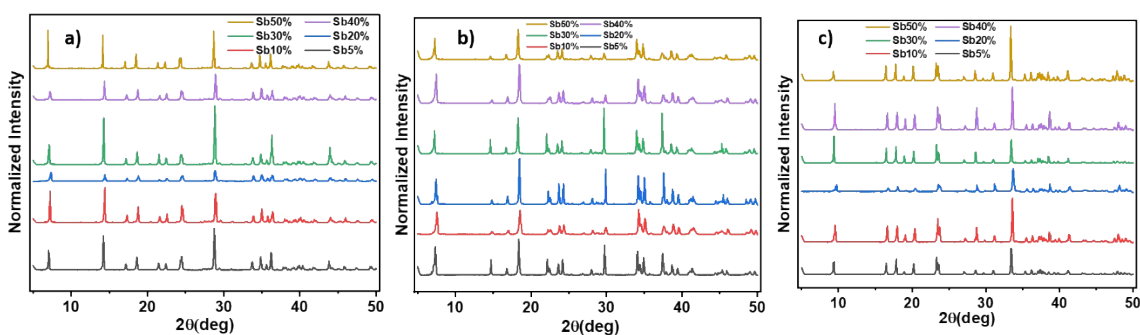


Figure S13: Comparison of PXRD patterns of different Sb feed percentages for a) Sb:DETACdCl hybrid, b) Sb:PDACdCl hybrid, c) Sb:DABCdCl hybrid.

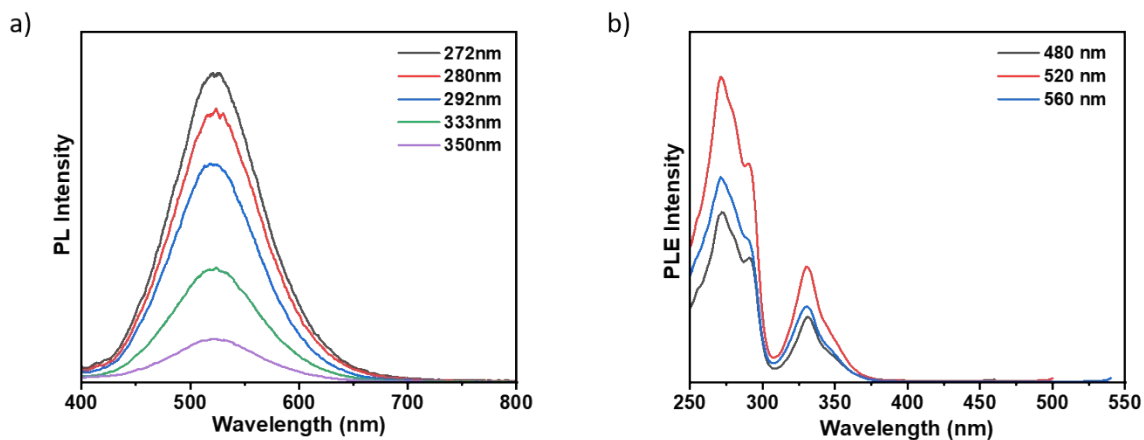


Figure S14: a) Excitation dependent PL, b) Emission dependent PLE of Sb:DETASbCl hybrid.

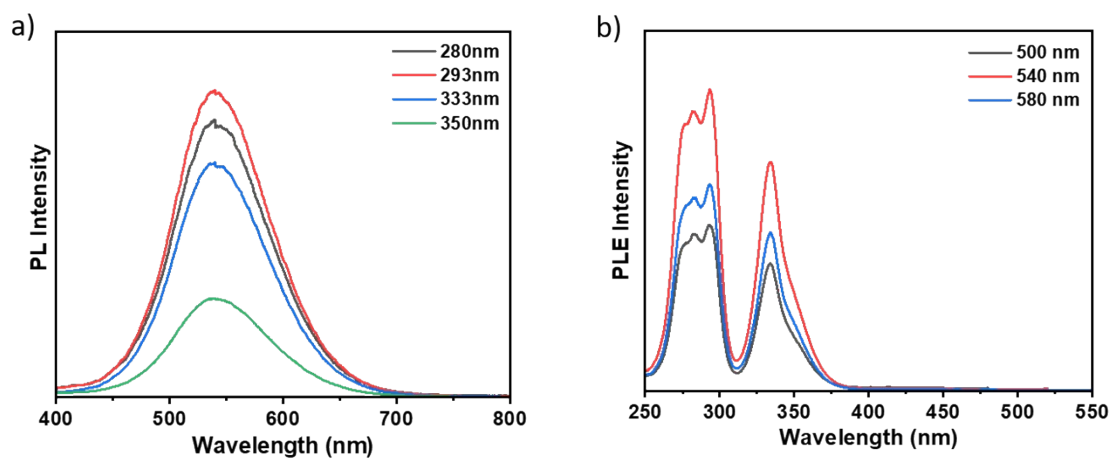


Figure S15: a) Excitation dependent PL, b) Emission dependent PLE of Sb:PDASbCl hybrid.

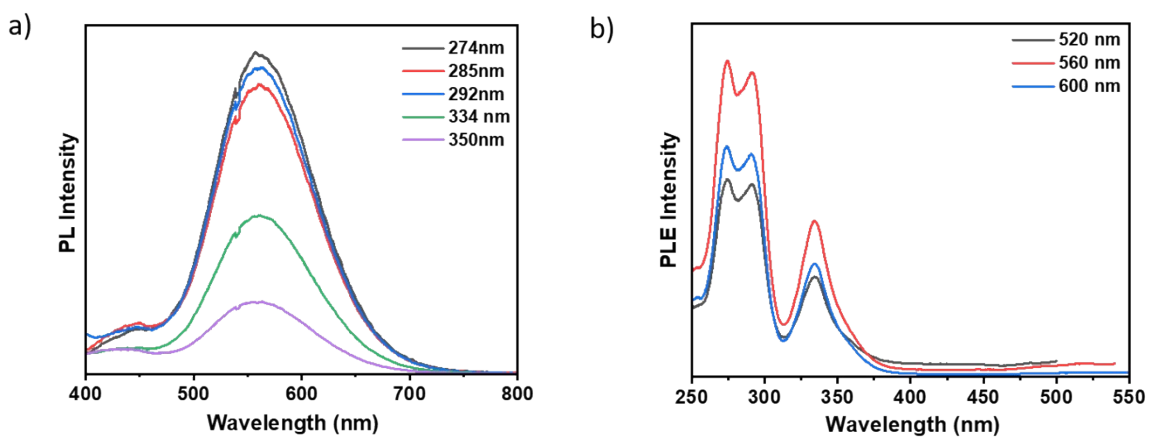


Figure S16: a) Excitation dependent PL, b) Emission dependent PLE of Sb:DABSbCl hybrid.

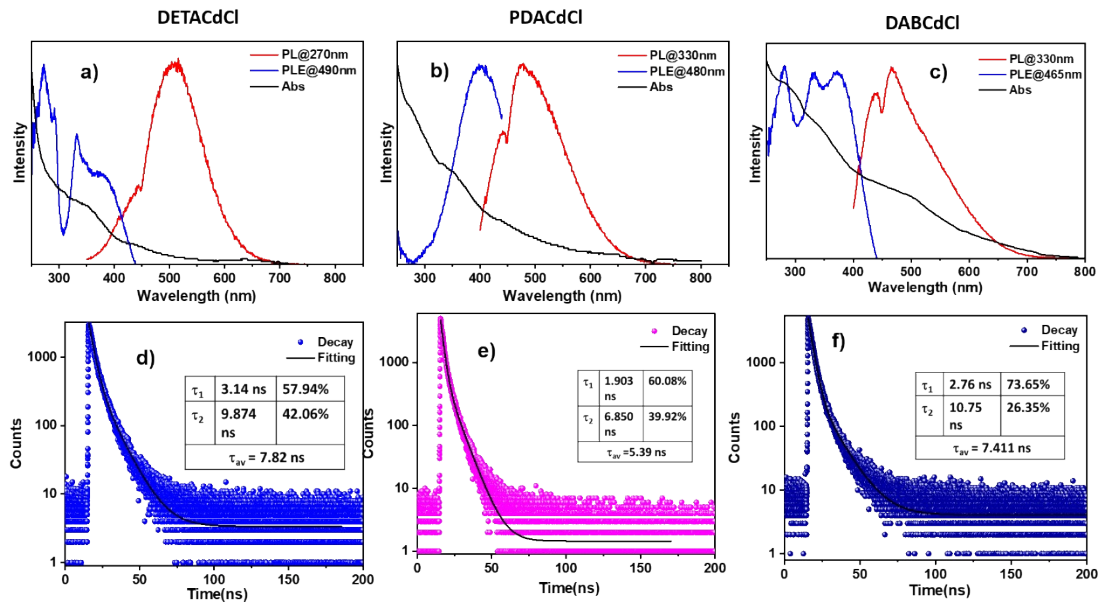


Figure S17: Optical characterization (PL, PLE and Absorption) of host hybrids- a) DETACdCl hybrid, b) PDACdCl hybrid, c) DABCdCl hybrid; Lifetime decay profiles of d) DETACdCl hybrid, e) PDACdCl hybrid, f) DABCdCl hybrid.

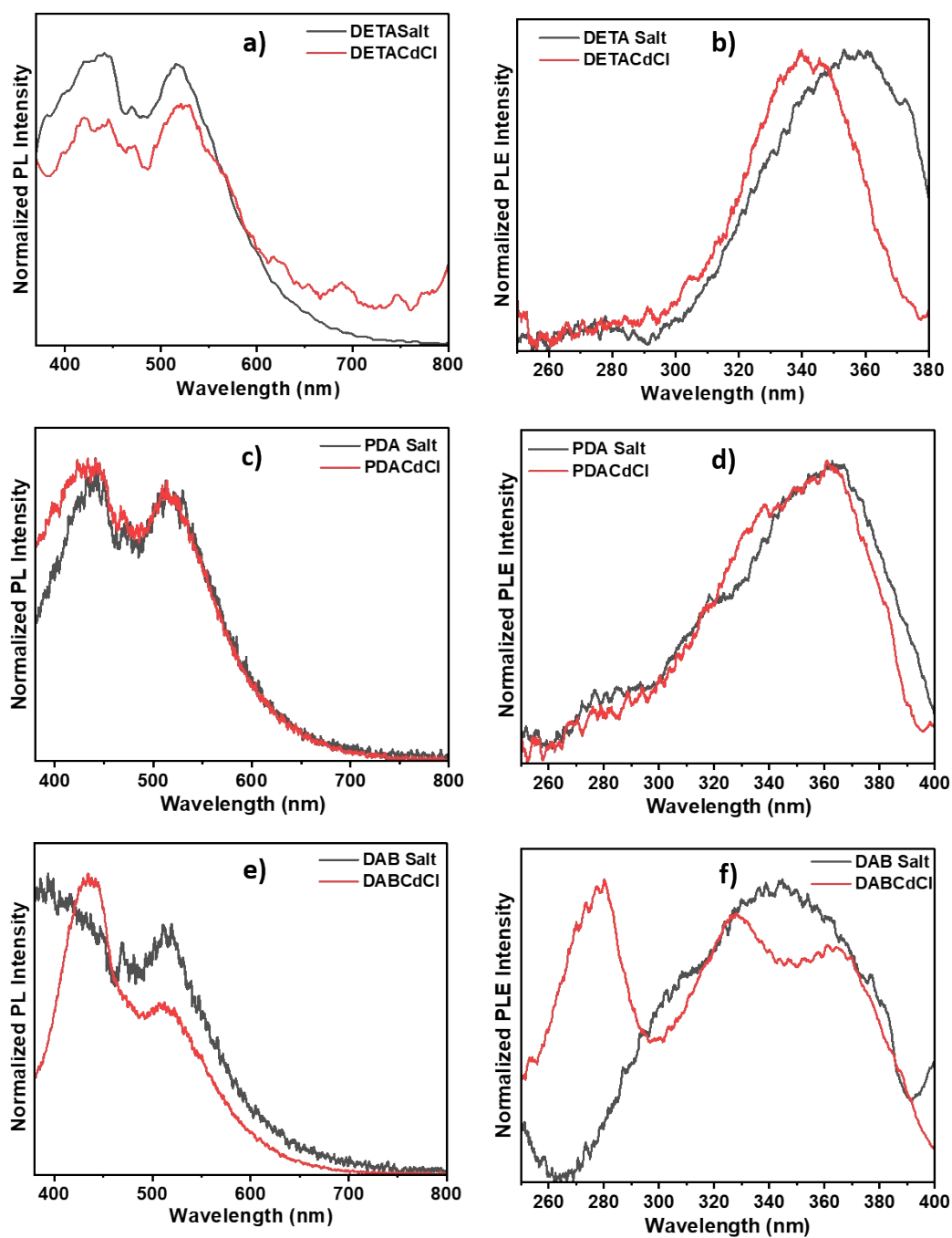


Figure S18: a) PL of DETA salt and DETACdCl excited at 275 nm b) PLE of DETA salt and DETACdCl (Emission: 420 nm) c) PL of PDA salt and PDACdCl excited at 275 nm d) PLE of PDA salt and PDACdCl (Emission: 435 nm) e) PL of DAB salt and DABCdCl excited at 275 nm f) PLE of DAB salt and DABCdCl (Emission: 425 nm)

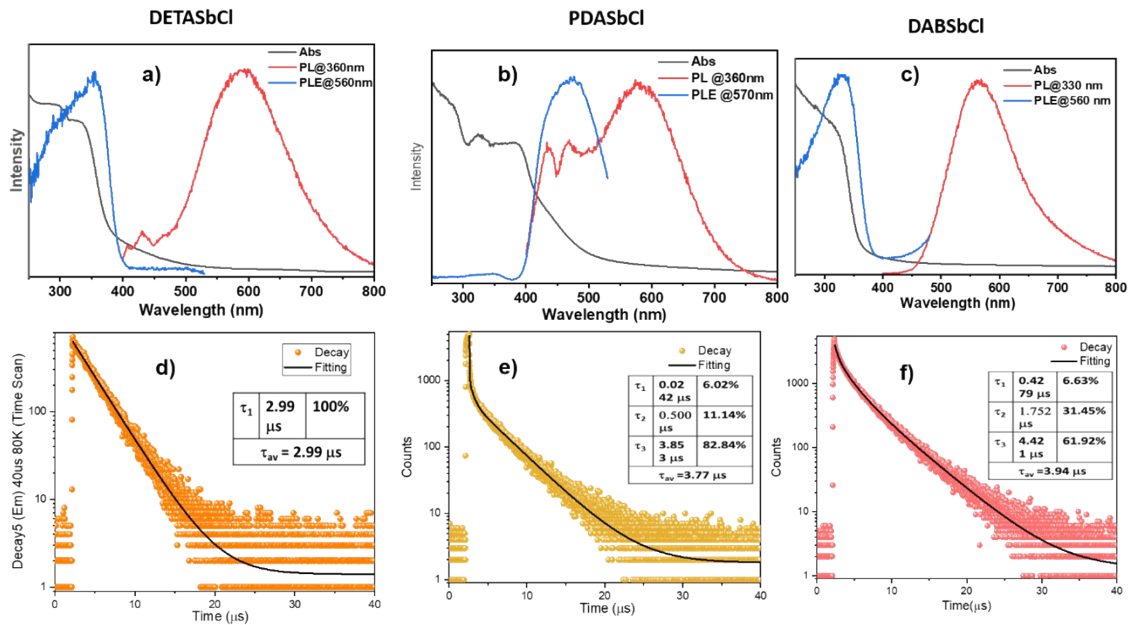


Figure S19: Optical characterization (PL, PLE and Absorption) of Control hybrids- a) DETASbCl hybrid, b) PDASbCl hybrid, c) DABSbCl hybrid; Lifetime decay profiles of d) DETASbCl hybrid, e) PDASbCl hybrid, f) DABSbCl hybrid.

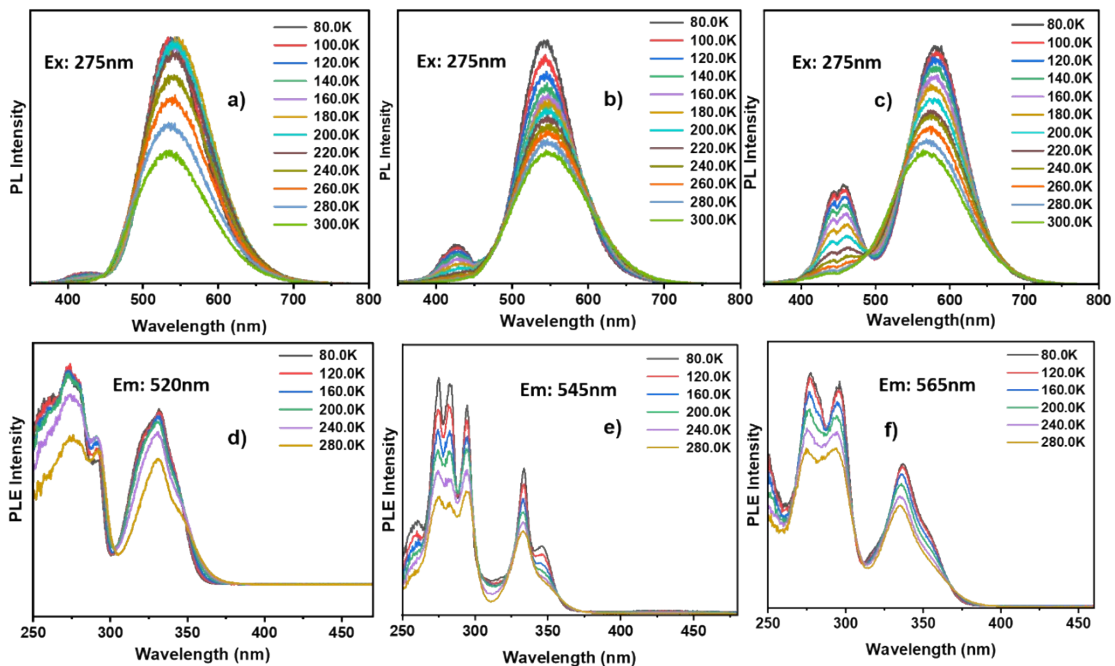


Figure S20: Temperature dependent PL of a) Sb: DETACdCl hybrid, b) Sb:PDACdCl hybrid, c) Sb:DABCdCl hybrid; Temperature dependent PLE of d) Sb:DETACdCl hybrid, e) Sb:PDACdCl hybrid, f) Sb:DABCdCl hybrid.

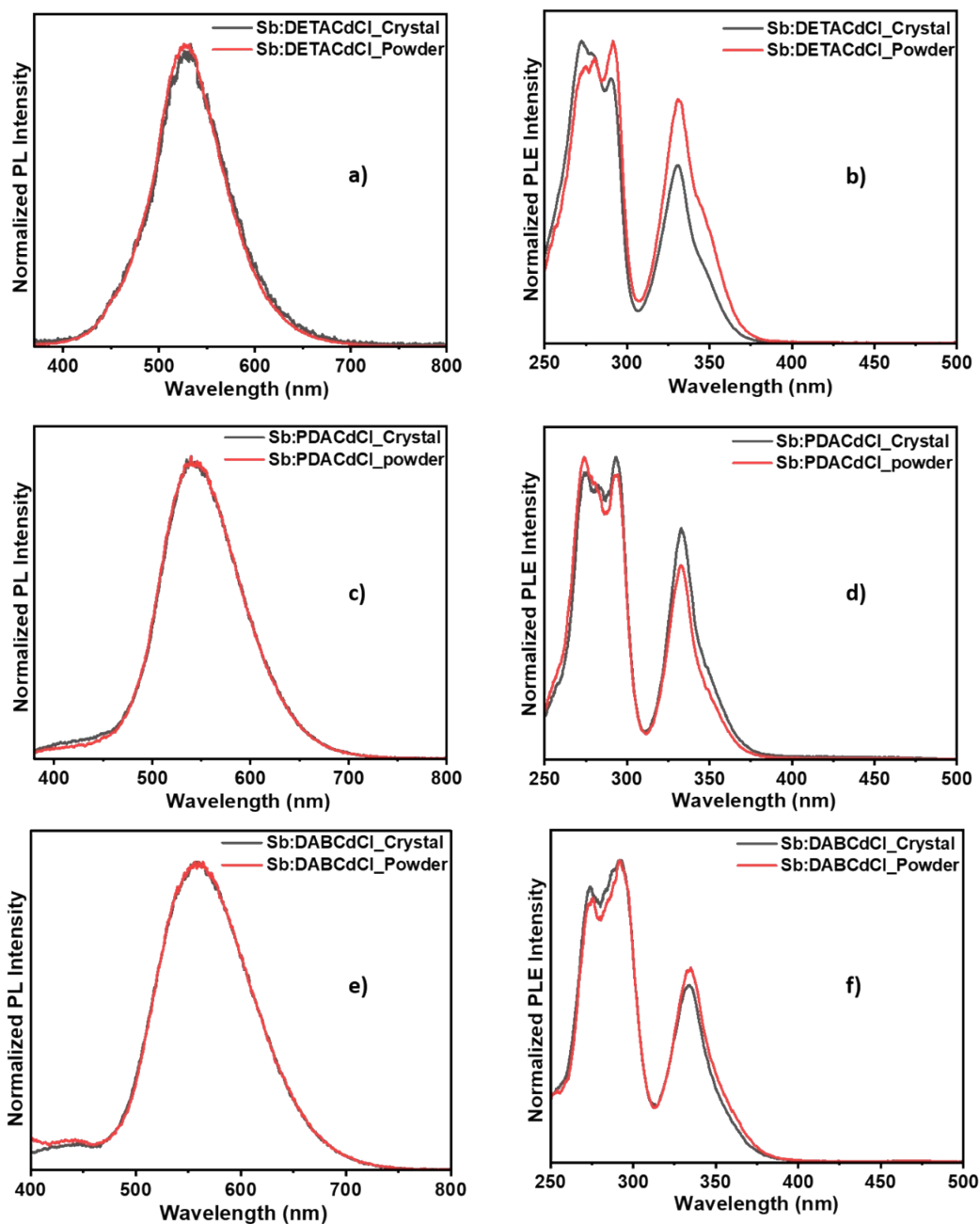


Figure S21: a) PL of Sb:DETACdCl crystal and powder excited at 330 nm b) PLE of Sb:DETACdCl crystal and powder (Emission: 520 nm) c) PL of Sb:PDACdCl crystal and powder excited at 330 nm d) PLE of Sb:PDACdCl crystal and powder (Emission: 540 nm) e) PL of Sb:DABCdCl crystal and powder excited at 330 nm f) PLE of Sb:DABCdCl crystal and powder (Emission: 560 nm).

Table S5: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for DETACdCl

DETACdCl

Table S5-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DETACdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	x	y	z	U(eq)
Cd1	10000	5000	5000	12.48 (7)
Cl1	7006.4 (5)	4865.5 (2)	7011.0 (5)	17.38 (9)
Cl2	10120.2 (5)	3969.3 (2)	4508.4 (6)	18.02 (9)
Cl3	4350.2 (7)	2500	9655.0 (7)	19.18 (11)
N1	5069 (2)	2500	5454 (3)	14.4 (4)
N2	4562 (2)	4001.5 (5)	4828.8 (17)	17.5 (3)
C1	4328 (2)	3007.7 (5)	4628 (2)	15.8 (3)
C2	5249 (2)	3482.3 (6)	5601 (2)	17.2 (3)

Table S5-2 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DETACdCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Cd1	11.87 (10)	13.82 (9)	11.74 (10)	-0.29 (5)	0.29 (5)	-0.09 (5)
Cl1	14.86 (16)	22.47 (17)	14.81 (17)	-1.67 (13)	4.52 (13)	-2.36 (13)
Cl2	20.55 (19)	14.43 (17)	19.09 (18)	-0.40 (14)	-0.17 (13)	-0.93 (12)
Cl3	16.9 (3)	23.5 (2)	17.1 (2)	0	-0.02 (19)	0
N1	15.5 (9)	13.7 (9)	14.0 (8)	0	-1.0 (6)	0
N2	19.8 (6)	14.2 (6)	18.5 (6)	-0.3 (5)	-1.1 (5)	-0.3 (5)
C1	16.7 (7)	13.6 (7)	17.0 (6)	1.0 (5)	-2.3 (6)	-0.2 (5)
C2	20.5 (7)	14.4 (7)	16.8 (7)	-0.1 (6)	-3.4 (6)	0.5 (5)

Table S5-3 Bond Lengths for DETACdCl.

Atom	Atom	Length/ \AA	Atom	Atom	Length/ \AA
Cd1	Cl1	2.6226 (3)	Cd1	Cl2 ¹	2.5732 (4)
Cd1	Cl1 ¹	2.6226 (3)	N1	C1 ⁴	1.4888 (16)
Cd1	Cl1 ²	2.6249 (3)	N1	C1	1.4889 (16)
Cd1	Cl1 ³	2.6249 (3)	N2	C2	1.485 (2)
Cd1	Cl2	2.5731 (4)	C1	C2	1.521 (2)

¹2-X,1-Y,1-Z; ²1/2+X,+Y,3/2-Z; ³2-X,1-Y,-1/2+Z; ⁴+X,1/2-Y,+Z

Table S5-4 Bond Angles for DETACdCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1	Cd1	Cl1 ¹	180.000 (15)	Cl2	Cd1	Cl1 ²	88.310 (12)
Cl1	Cd1	Cl1 ²	88.693 (5)	Cl2 ¹	Cd1	Cl1 ¹	88.807 (11)
Cl1	Cd1	Cl1 ³	91.307 (5)	Cl2	Cd1	Cl1 ³	91.690 (12)
Cl1 ³	Cd1	Cl1 ²	180.0	Cl2 ¹	Cd1	Cl1 ³	88.310 (12)
Cl1 ¹	Cd1	Cl1 ²	91.307 (5)	Cl2	Cd1	Cl2 ¹	179.999 (15)
Cl1 ¹	Cd1	Cl1 ³	88.693 (5)	Cd1	Cl1	Cd1 ⁴	153.475 (14)
Cl2 ¹	Cd1	Cl1	91.193 (11)	C1 ⁵	N1	C1	114.83 (16)
Cl2	Cd1	Cl1	88.807 (11)	N1	C1	C2	107.88 (12)
Cl2	Cd1	Cl1 ¹	91.194 (11)	N2	C2	C1	110.26 (12)
Cl2 ¹	Cd1	Cl1 ²	91.690 (12)				

¹2-X,1-Y,1-Z; ²1/2+X,+Y,3/2-Z; ³3/2-X,1-Y,-1/2+Z; ⁴3/2-X,1-Y,1/2+Z; ⁵+X,1/2-Y,+Z

Table S5-5 Torsion Angles for DETACdCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
N1	C1	C2	N2	179.92 (13)	C1 ¹	N1	C1	C2	178.97 (11)

¹+X,1/2-Y,+Z

Table S5-6 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DETACdCl.

Atom	x	y	z	U(eq)
H1A	6327.9	2499.99	5334.81	17
H1B	4804.68	2500	6681.86	17
H2A	4836.48	4017.58	3604.43	21
H2B	5119.46	4282.3	5423.66	21
H2C	3310.02	4022.91	4984.66	21
H1C	2964	3025.05	4790.04	19
H1D	4607.97	3018.77	3291.42	19
H2D	4971.19	3466.6	6937.77	21
H2E	6613.57	3459.75	5443.93	21

Table S6: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for Sb:DETACdCl

Sb:DETACdCl

Table S6-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:DETACdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	x	y	z	$U(\text{eq})$
Cd1	10000	5000	5000	13.30 (7)
Cl1	7009.6 (4)	4864.7 (2)	7013.7 (4)	18.06 (9)
Cl2	9879.4 (4)	6031.1 (2)	5492.4 (6)	18.72 (9)
Cl3	651.1 (7)	7500	4654.5 (7)	19.90 (11)
N1	4930 (2)	7500	4543 (3)	15.2 (4)
N2	5436 (2)	5997.6 (5)	5170.9 (16)	18.0 (3)
C2	5674 (2)	6992.6 (5)	5373.0 (19)	16.6 (3)
C1	4749.0 (19)	6516.5 (6)	4399 (2)	17.8 (3)
Sb1	10000	5000	5000	13.30 (7)

Table S6-2 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:DETACdCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^2U_{11}+2hka*b*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Cd1	12.58 (10)	14.66 (10)	12.66 (10)	-0.33 (4)	0.36 (4)	-0.01 (4)
Cl1	15.47 (16)	23.29 (17)	15.42 (16)	-1.76 (13)	4.67 (12)	-2.48 (13)
Cl2	21.05 (18)	15.18 (17)	19.93 (18)	-0.29 (15)	-0.20 (12)	-0.81 (11)
Cl3	17.4 (3)	24.9 (2)	17.5 (2)	0	-0.17 (18)	0
N1	16.5 (9)	14.1 (9)	14.9 (8)	0	-0.7 (6)	0
N2	20.7 (6)	15.0 (6)	18.3 (6)	-0.3 (4)	-0.8 (5)	-0.1 (5)
C2	17.5 (7)	14.5 (7)	17.7 (6)	1.1 (5)	-2.8 (5)	0.2 (5)
C1	21.1 (7)	15.5 (7)	17.0 (7)	-0.2 (6)	-3.3 (5)	-0.5 (5)
Sb1	12.58 (10)	14.66 (10)	12.66 (10)	-0.33 (4)	0.36 (4)	-0.01 (4)

Table S6-3 Bond Lengths for Sb:DETACdCl.

Atom	Atom	Length/ \AA	Atom	Atom	Length/ \AA
Cd1	Cl1	2.6216 (3)	Cl1	Sb1	2.6216 (3)
Cd1	Cl1 ¹	2.6246 (3)	Cl2	Sb1	2.5713 (4)
Cd1	Cl1 ²	2.6216 (3)	N1	C2 ⁴	1.4890 (16)
Cd1	Cl1 ³	2.6246 (3)	N1	C2	1.4890 (16)
Cd1	Cl2 ²	2.5713 (4)	N2	C1	1.4826 (19)
Cd1	Cl2	2.5713 (4)	C2	C1	1.5241 (19)

¹3/2-X,1-Y,-1/2+Z; ²2-X,1-Y,1-Z; ³1/2+X,+Y,3/2-Z; ⁴+X,3/2-Y,+Z

Table S6-4 Bond Angles for Sb:DETACdCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1	Cd1	Cl1 ¹	180.0	N1	C2	C1	107.71 (12)
Cl1	Cd1	Cl1 ²	91.328 (5)	N2	C1	C2	110.22 (12)
Cl1 ¹	Cd1	Cl1 ²	88.672 (5)	Cl1	Sb1	Cl1 ¹	180.0
Cl1 ²	Cd1	Cl1 ³	180.000 (15)	Cl1 ²	Sb1	Cl1 ³	180.000 (15)
Cl1	Cd1	Cl1 ³	88.672 (5)	Cl1 ¹	Sb1	Cl1 ²	88.672 (5)
Cl1 ¹	Cd1	Cl1 ³	91.328 (5)	Cl1	Sb1	Cl1 ³	88.672 (5)
Cl2	Cd1	Cl1 ¹	88.787 (11)	Cl1	Sb1	Cl1 ²	91.328 (5)
Cl2	Cd1	Cl1 ²	88.275 (11)	Cl1 ¹	Sb1	Cl1 ³	91.328 (5)
Cl2 ¹	Cd1	Cl1	88.788 (11)	Cl2 ¹	Sb1	Cl1	88.788 (11)
Cl2	Cd1	Cl1	91.211 (11)	Cl2	Sb1	Cl1 ²	88.275 (11)
Cl2	Cd1	Cl1 ³	91.725 (11)	Cl2	Sb1	Cl1	91.211 (11)
Cl2 ¹	Cd1	Cl1 ¹	91.213 (11)	Cl2 ¹	Sb1	Cl1 ¹	91.213 (11)
Cl2 ¹	Cd1	Cl1 ²	91.724 (11)	Cl2	Sb1	Cl1 ¹	88.787 (11)
Cl2 ¹	Cd1	Cl1 ³	88.275 (11)	Cl2 ¹	Sb1	Cl1 ²	91.724 (11)
Cl2	Cd1	Cl2 ¹	179.999 (14)	Cl2 ¹	Sb1	Cl1 ³	88.275 (11)
Cd1	Cl1	Cd1 ⁴	153.540 (14)	Cl2	Sb1	Cl1 ³	91.725 (11)
C2 ⁵	N1	C2	114.49 (16)	Cl2	Sb1	Cl2 ¹	179.999 (14)

¹2-X,1-Y,1-Z; ²3/2-X,1-Y,-1/2+Z; ³1/2+X,+Y,3/2-Z; ⁴3/2-X,1-Y,1/2+Z; ⁵+X,3/2-Y,+Z

Table S6-5 Torsion Angles for Sb:DETACdCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
N1	C2	C1	N2	179.93 (12)	C2 ¹	N1	C2	C1	178.91 (10)

¹+X,3/2-Y,+Z

Table S6-6 Hydrogen Atom Coordinates (Å×10⁴) and Isotropic Displacement Parameters (Å²×10³) for Sb:DETACdCl.

Atom	x	y	z	U(eq)
H1A	5195.22	7500	3315.51	18
H1B	3671.01	7500	4662.67	18
H2A	4891.02	5716.38	4565.79	22
H2B	5148.16	5979.61	6391.76	22

Table S6-6 Hydrogen Atom Coordinates ($\text{\AA}\times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for Sb:DETACdCl.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
H2C	6690.85	5978.36	5028.92	22
H2D	5395.48	6982.11	6709.94	20
H2E	7038.86	6975.38	5209.82	20
H1C	3384.81	6539.23	4557.65	21
H1D	5026.08	6532.04	3062.12	21

Table S6-7 Atomic Occupancy for Sb:DETACdCl.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
Sb1	0.0003				

Table S7: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for DETASbCl

DETASbCl

Table S7-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for DETASbCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
Sb01	7272.0 (2)	7512.4 (2)	7633.9 (2)	14.18 (4)
Cl02	10691.5 (5)	6155.9 (4)	7926.9 (4)	19.06 (8)
Cl03	8582.9 (5)	8919.8 (4)	4856.1 (4)	18.75 (8)
Cl04	4073.1 (5)	8512.8 (4)	7603.5 (4)	21.87 (9)
Cl05	7365.9 (6)	9545.6 (4)	8584.9 (4)	22.07 (9)
Cl06	5873.1 (5)	6488.2 (4)	10154.2 (4)	19.60 (8)
Cl07	7747.0 (6)	4666.8 (4)	6507.7 (4)	22.91 (9)
N008	7969.3 (18)	6893.0 (14)	3161.7 (14)	15.0 (3)
N009	2808.5 (18)	7794.2 (14)	4835.3 (15)	19.5 (3)
N00A	11716.3 (19)	6671.7 (15)	556.3 (15)	20.5 (3)
C00B	6062 (2)	6666.5 (16)	3806.4 (17)	16.2 (3)
C00C	4649 (2)	8053.3 (17)	3937.9 (18)	17.9 (3)
C00D	8374 (2)	7677.4 (17)	1692.5 (17)	18.3 (3)
C00E	10282 (2)	7954.8 (17)	1053.2 (18)	20.6 (3)

Table S7-2 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DETASbCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^2U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Sb01	14.89 (6)	13.75 (6)	15.41 (6)	-0.82 (4)	-6.50 (4)	-4.13 (4)
Cl02	17.91 (19)	18.96 (18)	21.05 (19)	-4.32 (15)	-7.83 (16)	-2.20 (15)
Cl03	20.43 (19)	17.30 (18)	19.50 (18)	-3.97 (15)	-6.88 (15)	-4.42 (15)
Cl04	16.74 (19)	24.5 (2)	24.8 (2)	1.81 (16)	-9.96 (16)	-4.05 (15)
Cl05	29.7 (2)	17.41 (19)	23.6 (2)	-3.14 (15)	-12.26 (17)	-7.17 (16)
Cl06	19.43 (19)	23.77 (19)	18.31 (18)	2.10 (15)	-8.08 (15)	-9.42 (15)
Cl07	29.6 (2)	21.3 (2)	22.1 (2)	-0.72 (16)	-13.20 (17)	-7.13 (16)
N008	14.7 (6)	15.2 (6)	16.2 (6)	-1.3 (5)	-6.9 (5)	-2.7 (5)
N009	15.5 (7)	16.7 (7)	24.7 (7)	-3.0 (6)	-5.5 (6)	-2.4 (5)
N00A	14.3 (7)	27.5 (8)	18.5 (7)	-1.7 (6)	-5.2 (6)	-3.9 (6)
C00B	13.5 (7)	14.9 (7)	18.7 (8)	0.1 (6)	-4.7 (6)	-3.4 (6)
C00C	12.4 (7)	16.2 (8)	22.1 (8)	-0.6 (6)	-3.6 (6)	-2.5 (6)
C00D	15.5 (8)	19.7 (8)	16.7 (8)	2.0 (6)	-5.1 (6)	-2.5 (6)
C00E	20.1 (8)	19.3 (8)	22.4 (8)	-2.4 (7)	-4.5 (7)	-8.3 (7)

Table S7-3 Bond Lengths for DETASbCl.

Atom Atom	Length/ \AA	Atom Atom	Length/ \AA
Sb01 Cl02	2.8306 (4)	N008 C00D	1.494 (2)
Sb01 Cl03	2.8230 (4)	N009 C00C	1.488 (2)
Sb01 Cl04	2.5201 (4)	N00A C00E	1.482 (2)
Sb01 Cl05	2.4100 (4)	C00B C00C	1.521 (2)
Sb01 Cl06	2.4904 (4)	C00D C00E	1.518 (2)
N008 C00B	1.4981 (19)		

Table S7-4 Bond Angles for DETASbCl.

Atom Atom Atom	Angle/ $^\circ$	Atom Atom Atom	Angle/ $^\circ$
Cl03 Sb01 Cl02	98.139 (12)	Cl06 Sb01 Cl03	174.276 (13)
Cl04 Sb01 Cl02	173.306 (12)	Cl06 Sb01 Cl04	87.594 (13)
Cl04 Sb01 Cl03	88.203 (12)	C00D N008 C00B	113.49 (12)
Cl05 Sb01 Cl02	88.714 (13)	N008 C00B C00C	111.24 (13)
Cl05 Sb01 Cl03	86.090 (13)	N009 C00C C00B	108.99 (13)
Cl05 Sb01 Cl04	93.878 (15)	N008 C00D C00E	113.84 (13)
Cl05 Sb01 Cl06	90.320 (14)	N00A C00E C00D	112.38 (13)
Cl06 Sb01 Cl02	86.220 (13)		

Table S7-5 Torsion Angles for DETASbCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
N008	C00B	C00C	N009	169.15 (13)	C00B	N008	C00D	C00E	178.04 (13)
N008	C00D	C00E	N00A	73.19 (18)	C00D	N008	C00B	C00C	-58.02 (17)

Table S7-6 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DETASbCl.

Atom	x	y	z	U(eq)
H00A	8143.48	7375.77	3774.07	18
H00B	8776.11	6044.38	3090.74	18
H00C	2584.56	7129.29	4448.14	23
H00D	1943.2	8603.95	4845.35	23
H00E	2772.09	7491	5761.13	23
H00F	11410.32	6244.74	-23.46	25
H00G	11812.04	6074.83	1342.31	25
H00H	12806.96	6907.97	44.98	25
H00I	5877.42	6158.92	4782.27	19
H00J	5892.01	6078.87	3184.94	19
H00K	4947.49	8712.6	4401.17	21
H00L	4658.33	8477.83	2958.76	21
H00M	7467.25	8585.88	1764.09	22
H00N	8224.49	7135.24	1019.04	22
H00O	10344.63	8671.72	219.35	25
H00P	10523.12	8336.08	1797.09	25

Table S8: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for PDACdCl

PDACdCl

Table S8-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for PDACdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	x	y	z	U(eq)
Cd1	5000	5000	5000	12.63 (16)
Cl1	5000	6053.2 (3)	4467.5 (12)	25.3 (2)
Cl2	7500	5110 (20)	7500	50 (20)
N1	5000	5945.5 (16)	205 (4)	44.7 (13)
C3	5000	7500	-659 (8)	45.6 (16)

Table S8-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for PDACdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	$U(\text{eq})$
C2	5000	6974.2 (19)	206 (6)	70 (3)
C1	5000	6459.7 (16)	-666 (5)	51.6 (13)
Cl3	7783 (7)	5140.1 (19)	7211 (6)	15.5 (10)

Table S8-2 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for PDACdCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Cd1	11.4 (2)	15.2 (2)	11.3 (2)	0.49 (9)	0	0
Cl1	42.6 (5)	15.8 (4)	17.7 (4)	0.6 (3)	0	0
Cl2	50 (30)	60 (30)	50 (40)	0	-40 (30)	0
N1	100 (4)	13.5 (17)	20.5 (15)	0.3 (11)	0	0
C3	93 (5)	19 (3)	25 (3)	0	0	0
C2	168 (9)	15 (2)	26 (2)	-1.2 (15)	0	0
C1	117 (4)	17.0 (17)	20.5 (18)	1.6 (15)	0	0
Cl3	14.7 (17)	19.3 (12)	12.5 (16)	-0.1 (6)	-3.1 (12)	-0.5 (6)

Table S8-3 Bond Lengths for PDACdCl.

Atom Atom	Length/ \AA	Atom Atom	Length/ \AA
Cd1 Cl1 ¹	2.5657 (8)	Cd1 Cl3	2.635 (5)
Cd1 Cl1	2.5658 (8)	Cd1 Cl3 ⁶	2.653 (5)
Cd1 Cl2 ²	2.618 (5)	Cd1 Cl3 ⁷	2.635 (5)
Cd1 Cl2 ¹	2.618 (5)	N1 C1	1.397 (5)
Cd1 Cl2	2.618 (6)	C3 C2	1.420 (5)
Cd1 Cl2 ³	2.618 (6)	C3 C2 ⁸	1.420 (5)
Cd1 Cl3 ¹	2.635 (5)	C2 C1	1.398 (6)
Cd1 Cl3 ⁴	2.653 (5)	Cl3 Cl3 ⁴	0.596 (10)
Cd1 Cl3 ⁵	2.635 (5)		

¹1-X,1-Y,1-Z; ²-1/2+X,+Y,3/2-Z; ³3/2-X,1-Y,-1/2+Z; ⁴3/2-X,+Y,3/2-Z; ⁵+X,1-Y,1-Z; ⁶-1/2+X,1-Y,-1/2+Z; ⁷1-X,+Y,+Z; ⁸1-X,3/2-Y,+Z

Table S8-4 Bond Angles for PDACdCl.

Atom Atom Atom	Angle/ $^\circ$	Atom Atom Atom	Angle/ $^\circ$
Cl1 ¹ Cd1 Cl1	180.0	Cl2 ³ Cd1 Cl3 ⁵	98.38 (16)

Table S8-4 Bond Angles for PDACdCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1	Cd1	Cl2	90.7 (12)	Cl2 ²	Cd1	Cl3 ⁵	81.62 (16)
Cl1 ¹	Cd1	Cl2 ¹	90.7 (12)	Cl2 ³	Cd1	Cl3 ⁶	6.7 (4)
Cl1 ¹	Cd1	Cl2 ²	89.3 (12)	Cl2 ³	Cd1	Cl3 ⁴	81.62 (16)
Cl1	Cd1	Cl2 ²	90.7 (12)	Cl2 ²	Cd1	Cl3 ⁷	6.7 (4)
Cl1 ¹	Cd1	Cl2 ³	90.7 (12)	Cl2 ¹	Cd1	Cl3 ⁵	173.3 (3)
Cl1 ¹	Cd1	Cl2	89.3 (12)	Cl2 ²	Cd1	Cl3 ⁶	173.3 (4)
Cl1	Cd1	Cl2 ¹	89.3 (12)	Cl2 ¹	Cd1	Cl3 ⁴	6.7 (3)
Cl1	Cd1	Cl2 ³	89.3 (12)	Cl2	Cd1	Cl3 ⁶	85.50 (19)
Cl1	Cd1	Cl3 ⁴	90.26 (10)	Cl2 ²	Cd1	Cl3 ⁴	98.38 (16)
Cl1	Cd1	Cl3 ⁵	89.74 (10)	Cl2 ²	Cd1	Cl3 ¹	85.50 (19)
Cl1	Cd1	Cl3 ⁶	91.70 (10)	Cl2	Cd1	Cl3 ⁵	6.7 (3)
Cl1 ¹	Cd1	Cl3 ⁵	90.26 (10)	Cl2 ¹	Cd1	Cl3 ⁶	94.50 (19)
Cl1 ¹	Cd1	Cl3 ⁶	88.30 (10)	Cl2 ¹	Cd1	Cl3 ⁷	85.50 (19)
Cl1	Cd1	Cl3 ⁷	88.30 (10)	Cl2 ³	Cd1	Cl3 ⁷	173.3 (4)
Cl1 ¹	Cd1	Cl3 ⁷	91.70 (10)	Cl2 ¹	Cd1	Cl3 ¹	6.7 (4)
Cl1	Cd1	Cl3	88.29 (10)	Cl2 ³	Cd1	Cl3 ¹	94.50 (19)
Cl1 ¹	Cd1	Cl3	91.71 (10)	Cl2	Cd1	Cl3 ⁴	173.3 (3)
Cl1	Cd1	Cl3 ¹	91.70 (10)	Cl3 ⁶	Cd1	Cl3	79.3 (2)
Cl1 ¹	Cd1	Cl3 ¹	88.30 (10)	Cl3 ⁷	Cd1	Cl3	100.7 (2)
Cl1 ¹	Cd1	Cl3 ⁴	89.74 (10)	Cl3 ¹	Cd1	Cl3 ⁷	79.3 (2)
Cl2	Cd1	Cl2 ²	88.2 (2)	Cl3 ¹	Cd1	Cl3	180.0
Cl2	Cd1	Cl2 ¹	180.0	Cd1	Cl2	Cd1 ⁸	169 (2)
Cl2	Cd1	Cl2 ³	91.8 (2)	C2	C3	C2 ⁹	126.0 (5)
Cl2 ¹	Cd1	Cl2 ³	88.2 (2)	C1	C2	C3	125.3 (4)
Cl2 ²	Cd1	Cl2 ³	180.0	N1	C1	C2	124.7 (3)
Cl2 ¹	Cd1	Cl2 ²	91.8 (2)	Cd1	Cl3	Cd1 ⁸	160.4 (2)
Cl2	Cd1	Cl3 ⁷	94.50 (19)	Cl3 ⁵	Cl3	Cd1	85.3 (10)
Cl2	Cd1	Cl3 ¹	173.3 (4)				

¹1-X,1-Y,1-Z; ²-1/2+X,+Y,3/2-Z; ³3/2-X,1-Y,-1/2+Z; ⁴-1/2+X,1-Y,-1/2+Z; ⁵3/2-X,+Y,3/2-Z; ⁶+X,1-Y,1-Z; ⁷1-X,+Y,+Z; ⁸3/2-X,1-Y,1/2+Z; ⁹1-X,3/2-Y,+Z

Table S8-5 Torsion Angles for PDACdCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C3	C2	C1	N1	180.000 (2)	C2 ¹	C3	C2	C1	180.000 (3)

¹1-X,3/2-Y,+Z

Table S8-6 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for PDACdCl.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
H1A	3832.17	5813.14	265.73	67
H1B	5455.9	5986.49	1334.37	67
H1C	5711.94	5702.13	-418.58	67
H3B	3913.77	7500	-1457.2	55
H3A	6086.23	7500	-1457.2	55
H2B	6086.78	6972.71	1003.62	83
H2A	3913.22	6972.71	1003.61	83
H1D	3912.68	6459.74	-1462.42	62
H1E	6087.32	6459.74	-1462.42	62

Table S8-7 Atomic Occupancy for PDACdCl.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
Cl2	0.1	H1A	0.5	H1B	0.5
H1C	0.5	H3B	0.5	H3A	0.5
H2B	0.5	H2A	0.5	H1D	0.5
H1E	0.5	Cl3	0.45		

Table S9: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for Sb:PDACdCl

Sb:PDACdCl

Table S9-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:PDACdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
Cd01	5000	5000	0	13.0(2)

Table S9-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:PDACdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	x	y	z	U(eq)
Cl02	5000	6053.8 (5)	531.1 (17)	25.3 (3)
N004	5000	5946 (2)	4803 (6)	46.3 (19)
C005	5000	7500	5644 (12)	48 (3)
C006	5000	6975 (3)	4787 (9)	70 (4)
C007	5000	6460 (2)	5661 (8)	54 (2)
Sb01	5000	5000	0	13.0 (2)
Cl1	7781 (3)	4863.0 (6)	2217 (3)	18.2 (5)

Table S9-2 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:PDACdCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Cd01	12.4 (3)	13.5 (3)	13.2 (3)	-0.49 (13)	0	0
Cl02	41.8 (7)	14.2 (5)	19.8 (6)	-0.9 (4)	0	0
N004	103 (6)	14 (3)	22 (2)	2.9 (17)	0	0
C005	100 (8)	16 (4)	28 (4)	0	0	0
C006	165 (13)	16 (4)	29 (3)	-2 (2)	0	0
C007	125 (7)	15 (2)	22 (3)	0 (2)	0	0
Sb01	12.4 (3)	13.5 (3)	13.2 (3)	-0.49 (13)	0	0
Cl1	15.4 (16)	21.9 (6)	17.3 (16)	0.3 (6)	-5.8 (9)	0.7 (6)

Table S9-3 Bond Lengths for Sb:PDACdCl.

Atom Atom	Length/ \AA	Atom Atom	Length/ \AA
Cd01 Cl02 ¹	2.5669 (11)	C005 C006	1.415 (7)
Cd01 Cl02	2.5669 (11)	C005 C006 ⁸	1.415 (7)
Cd01 Cl1 ²	2.650 (3)	C006 C007	1.400 (9)
Cd01 Cl1 ³	2.650 (3)	Sb01 Cl1 ¹	2.636 (3)
Cd01 Cl1 ¹	2.636 (3)	Sb01 Cl1 ⁷	2.636 (3)
Cd01 Cl1 ⁴	2.650 (3)	Sb01 Cl1 ⁶	2.650 (3)
Cd01 Cl1 ⁵	2.636 (3)	Sb01 Cl1 ²	2.650 (3)
Cd01 Cl1 ⁶	2.650 (3)	Sb01 Cl1	2.636 (3)
Cd01 Cl1	2.636 (3)	Sb01 Cl1 ⁴	2.650 (3)
Cd01 Cl1 ⁷	2.636 (3)	Sb01 Cl1 ⁵	2.636 (3)
Cl02 Sb01	2.5669 (11)	Sb01 Cl1 ³	2.650 (3)
N004 C007	1.394 (8)	Cl1 Cl1 ²	0.588 (3)

¹1-X,1-Y,-Z; ²3/2-X,+Y,1/2-Z; ³-1/2+X,+Y,1/2-Z; ⁴3/2-X,1-Y,-1/2+Z; ⁵1-X,+Y,+Z; ⁶-1/2+X,1-Y,-1/2+Z; ⁷+X,1-Y,-Z; ⁸1-X,3/2-Y,+Z

Table S9-4 Bond Angles for Sb:PDACdCl.

Atom Atom Atom	Angle/°	Atom Atom Atom	Angle/°
Cl02 ¹ Cd01 Cl02	180.0	Cl02 Sb01 Cl1 ⁶	91.55 (4)
Cl02 ¹ Cd01 Cl1 ²	90.14 (4)	Cl02 Sb01 Cl1 ⁷	88.45 (4)
Cl02 Cd01 Cl1 ³	90.14 (4)	Cl02 ¹ Sb01 Cl1 ⁵	90.14 (4)
Cl02 Cd01 Cl1	91.55 (4)	Cl02 ¹ Sb01 Cl1 ⁷	91.55 (4)
Cl02 Cd01 Cl1 ⁴	90.14 (4)	Cl02 Sb01 Cl1 ²	89.86 (4)
Cl02 ¹ Cd01 Cl1 ³	89.86 (4)	Cl02 Sb01 Cl1 ⁵	89.86 (4)
Cl02 Cd01 Cl1 ⁵	89.86 (4)	Cl02 Sb01 Cl1 ³	90.14 (4)
Cl02 ¹ Cd01 Cl1 ¹	91.55 (4)	Cl02 ¹ Sb01 Cl1 ⁴	89.86 (4)
Cl02 Cd01 Cl1 ⁶	91.55 (4)	Cl02 Sb01 Cl1 ¹	88.45 (4)
Cl02 Cd01 Cl1 ²	89.86 (4)	Cl02 ¹ Sb01 Cl1 ³	89.86 (4)
Cl02 ¹ Cd01 Cl1 ⁷	91.55 (4)	Cl02 Sb01 Cl1 ⁴	90.14 (4)
Cl02 Cd01 Cl1 ¹	88.45 (4)	Cl02 ¹ Sb01 Cl1	88.45 (4)
Cl02 Cd01 Cl1 ⁷	88.45 (4)	Cl02 ¹ Sb01 Cl1 ⁶	88.45 (4)
Cl02 ¹ Cd01 Cl1 ⁵	90.14 (4)	Cl02 ¹ Sb01 Cl1 ²	90.14 (4)
Cl02 ¹ Cd01 Cl1	88.45 (4)	Cl02 Sb01 Cl1	91.55 (4)
Cl02 ¹ Cd01 Cl1 ⁶	88.45 (4)	Cl02 ¹ Sb01 Cl1 ¹	91.55 (4)
Cl02 ¹ Cd01 Cl1 ⁴	89.86 (4)	Cl1 ¹ Sb01 Cl1 ²	12.77 (6)
Cl1 ⁶ Cd01 Cl1 ⁷	180.00 (7)	Cl1 Sb01 Cl1 ⁴	12.77 (6)
Cl1 ⁷ Cd01 Cl1 ⁴	92.079 (16)	Cl1 ¹ Sb01 Cl1 ³	92.079 (16)
Cl1 ¹ Cd01 Cl1 ⁶	79.44 (6)	Cl1 ² Sb01 Cl1 ³	104.77 (6)
Cl1 ⁶ Cd01 Cl1 ⁵	167.23 (6)	Cl1 ¹ Sb01 Cl1	180.0
Cl1 ⁷ Cd01 Cl1 ⁵	12.77 (6)	Cl1 ² Sb01 Cl1 ⁵	75.23 (6)
Cl1 Cd01 Cl1 ⁵	92.078 (16)	Cl1 Sb01 Cl1 ³	87.922 (16)
Cl1 ⁷ Cd01 Cl1 ³	167.23 (6)	Cl1 ⁷ Sb01 Cl1 ²	87.921 (16)
Cl1 ⁴ Cd01 Cl1 ⁵	104.77 (6)	Cl1 ² Sb01 Cl1 ⁴	180.00 (7)
Cl1 ³ Cd01 Cl1 ⁵	180.00 (7)	Cl1 ¹ Sb01 Cl1 ⁵	87.921 (16)
Cl1 ¹ Cd01 Cl1 ³	92.079 (16)	Cl1 ⁶ Sb01 Cl1	100.57 (6)
Cl1 ¹ Cd01 Cl1 ⁷	100.56 (6)	Cl1 Sb01 Cl1 ²	167.23 (6)
Cl1 ⁷ Cd01 Cl1 ²	87.921 (16)	Cl1 ⁶ Sb01 Cl1 ²	92.079 (16)
Cl1 ¹ Cd01 Cl1 ⁴	167.23 (6)	Cl1 ⁷ Sb01 Cl1 ³	167.23 (6)
Cl1 ⁶ Cd01 Cl1 ³	12.77 (6)	Cl1 ⁴ Sb01 Cl1 ³	75.23 (6)
Cl1 ² Cd01 Cl1 ³	104.77 (6)	Cl1 ¹ Sb01 Cl1 ⁶	79.44 (6)
Cl1 Cd01 Cl1 ²	167.23 (6)	Cl1 ³ Sb01 Cl1 ⁵	180.00 (7)
Cl1 ⁷ Cd01 Cl1	79.43 (6)	Cl1 ⁷ Sb01 Cl1	79.43 (6)
Cl1 Cd01 Cl1 ⁴	12.77 (6)	Cl1 ⁶ Sb01 Cl1 ⁵	167.23 (6)
Cl1 ¹ Cd01 Cl1 ²	12.77 (6)	Cl1 ¹ Sb01 Cl1 ⁷	100.56 (6)
Cl1 ² Cd01 Cl1 ⁴	180.00 (7)	Cl1 Sb01 Cl1 ⁵	92.078 (16)

Table S9-4 Bond Angles for Sb:PDACdCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1	Cd01	Cl1 ³	87.922 (16)	Cl1 ⁷	Sb01	Cl1 ⁵	12.77 (6)
Cl1 ¹	Cd01	Cl1	180.0	Cl1 ⁴	Sb01	Cl1 ⁵	104.77 (6)
Cl1 ²	Cd01	Cl1 ⁵	75.23 (6)	Cl1 ¹	Sb01	Cl1 ⁴	167.23 (6)
Cl1 ⁶	Cd01	Cl1	100.57 (6)	Cl1 ⁶	Sb01	Cl1 ³	12.77 (6)
Cl1 ⁶	Cd01	Cl1 ²	92.079 (16)	Cl1 ⁶	Sb01	Cl1 ⁷	180.00 (7)
Cl1 ⁴	Cd01	Cl1 ³	75.23 (6)	Cl1 ⁶	Sb01	Cl1 ⁴	87.921 (16)
Cl1 ¹	Cd01	Cl1 ⁵	87.921 (16)	Cl1 ⁷	Sb01	Cl1 ⁴	92.079 (16)
Cl1 ⁶	Cd01	Cl1 ⁴	87.921 (16)	Cd01	Cl1	Cd01 ⁹	160.76 (6)
C006	C005	C006 ⁸	126.4 (8)	Cl1 ⁴	Cl1	Cd01	85.0 (5)
C007	C006	C005	125.5 (6)	Cl1 ⁴	Cl1	Cd01 ⁹	82.3 (5)
N004	C007	C006	125.0 (5)	Cl1 ⁴	Cl1	Sb01	85.0 (5)
Cl02 ¹	Sb01	Cl02	180.0				

¹1-X,1-Y,-Z; ²-1/2+X,1-Y,-1/2+Z; ³-1/2+X,+Y,1/2-Z; ⁴3/2-X,+Y,1/2-Z; ⁵3/2-X,1-Y,-1/2+Z; ⁶1-X,+Y,+Z; ⁷+X,1-Y,-Z; ⁸1-X,3/2-Y,+Z; ⁹3/2-X,1-Y,1/2+Z

Table S9-5 Torsion Angles for Sb:PDACdCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C005	C006	C007	N004	180.000 (4)	C006 ¹	C005	C006	C007	180.000 (5)

¹1-X,3/2-Y,+Z

Table S9-6 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:PDACdCl.

Atom	x	y	z	U(eq)
H00A	6150.23	5799.03	4833.32	69
H00B	4640.47	5989.23	3642.24	69
H00C	4209.3	5712.55	5374.9	69
H00E	6085.67	7500	6442.07	57
H00D	3914.33	7500	6442.06	57
H00G	3913.6	6973.51	3989.83	84
H00F	6086.4	6973.51	3989.83	84
H00I	6086.78	6461.36	6458.33	65
H00H	3913.22	6461.36	6458.33	65

Table S9-7 Atomic Occupancy for Sb:PDACdCl.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
H00A	0.5	H00B	0.5	H00C	0.5
H00E	0.5	H00D	0.5	H00G	0.5
H00F	0.5	H00I	0.5	H00H	0.5
Sb01	0.0001	Cl1	0.5		

Table S10: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for PDASbCl

PDASbCl

Table S10-1 Crystal data and structure refinement for PDASbCl.

Identification code	PDASbCl
Empirical formula	C ₅ H ₁₆ Cl ₅ N ₂ Sb
Formula weight	403.20
Temperature/K	100.00
Crystal system	orthorhombic
Space group	P2 ₁ 2 ₁ 2 ₁
a/Å	7.5550(5)
b/Å	13.3233(10)
c/Å	13.7812(10)
α/°	90
β/°	90
γ/°	90
Volume/Å ³	1387.18(17)
Z	4
ρ _{calc} /cm ³	1.931
μ/mm ⁻¹	2.917
F(000)	784.0
Crystal size/mm ³	0.21 × 0.2 × 0.19
Radiation	MoKα (λ = 0.71073)
2θ range for data collection/°	5.912 to 52.94
Index ranges	-9 ≤ h ≤ 9, -16 ≤ k ≤ 16, -17 ≤ l ≤ 17
Reflections collected	31654
Independent reflections	2864 [R _{int} = 0.0329, R _{sigma} = 0.0165]
Data/restraints/parameters	2864/0/120
Goodness-of-fit on F ²	1.116
Final R indexes [I ≥ 2σ (I)]	R ₁ = 0.0094, wR ₂ = 0.0231
Final R indexes [all data]	R ₁ = 0.0096, wR ₂ = 0.0232
Largest diff. peak/hole / e Å ⁻³	0.23/-0.31

Flack parameter

-0.010(5)

Table S10-2 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for PDASbCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	x	y	z	U(eq)
Sb01	4828.1 (2)	8674.4 (2)	8970.2 (2)	11.13 (4)
Cl02	5557.0 (6)	7300.4 (3)	7592.2 (3)	15.92 (10)
Cl03	2352.5 (6)	9423.4 (4)	8002.9 (4)	18.79 (10)
Cl04	7638.8 (7)	7919.5 (4)	10092.4 (3)	18.43 (10)
Cl05	6937.4 (7)	9804.6 (4)	8225.2 (4)	19.47 (11)
Cl06	4322.9 (7)	9809.8 (4)	10471.4 (4)	21.64 (11)
N007	9874 (2)	4672.2 (11)	3935.2 (12)	18.1 (3)
N008	9714 (2)	7416.3 (13)	8080.6 (11)	18.0 (3)
C009	10000 (3)	6059.7 (14)	5129.8 (13)	15.3 (4)
C00A	10911 (3)	7043.6 (15)	5398.7 (14)	17.2 (4)
C00B	10139 (3)	7522.3 (14)	6312.0 (13)	16.9 (4)
C00C	10525 (3)	6917.3 (15)	7218.1 (13)	15.9 (4)
C00D	10719 (3)	5657.9 (15)	4182.8 (14)	18.7 (4)

Table S10-3 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for PDASbCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Sb01	11.19 (6)	11.08 (6)	11.10 (6)	0.24 (4)	-0.36 (5)	-0.46 (4)
Cl02	17.4 (2)	14.4 (2)	15.9 (2)	-0.63 (17)	0.87 (18)	-0.64 (17)
Cl03	16.4 (2)	19.1 (2)	20.9 (2)	-0.3 (2)	-4.92 (19)	3.22 (18)
Cl04	17.3 (2)	21.4 (2)	16.6 (2)	0.93 (18)	-1.25 (19)	4.74 (19)
Cl05	19.6 (2)	19.6 (2)	19.2 (2)	4.0 (2)	-2.01 (19)	-8.68 (19)
Cl06	19.1 (3)	22.2 (2)	23.6 (3)	-9.2 (2)	-0.1 (2)	2.54 (19)
N007	22.1 (8)	15.5 (7)	16.7 (7)	-2.7 (6)	-0.5 (9)	0.6 (7)
N008	17.2 (8)	22.8 (8)	14.1 (7)	-2.3 (7)	-0.1 (7)	1.4 (7)
C009	16.7 (9)	15.5 (8)	13.8 (8)	-1.3 (7)	0.9 (8)	-1.3 (8)
C00A	22.8 (11)	16.0 (10)	12.7 (9)	-0.1 (8)	2.9 (8)	-4.0 (8)
C00B	21.1 (10)	15.0 (9)	14.6 (8)	-1.4 (7)	-0.4 (8)	-0.8 (8)
C00C	16.0 (10)	19.1 (9)	12.6 (9)	-2.0 (7)	1.3 (7)	0.8 (8)
C00D	24.0 (11)	16.2 (9)	15.8 (10)	-2.8 (8)	2.9 (8)	-4.6 (8)

Table S10-4 Bond Lengths for PDASbCl.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Sb01	Cl02	2.6947 (5)	N008	C00C	1.494 (2)
Sb01	Cl03	2.5041 (5)	C009	C00A	1.526 (3)
Sb01	Cl04	2.8129 (5)	C009	C00D	1.512 (3)
Sb01	Cl05	2.4209 (5)	C00A	C00B	1.527 (3)
Sb01	Cl06	2.5911 (5)	C00B	C00C	1.515 (3)
N007	C00D	1.500 (2)			

Table S10-5 Bond Angles for PDASbCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl02	Sb01	Cl04	89.444 (15)	Cl06	Sb01	Cl02	171.657 (16)
Cl03	Sb01	Cl02	92.762 (16)	Cl06	Sb01	Cl04	83.163 (16)
Cl03	Sb01	Cl04	177.409 (16)	C00D	C009	C00A	110.57 (16)
Cl03	Sb01	Cl06	94.725 (17)	C009	C00A	C00B	112.74 (17)
Cl05	Sb01	Cl02	89.378 (17)	C00C	C00B	C00A	112.57 (16)
Cl05	Sb01	Cl03	91.033 (18)	N008	C00C	C00B	109.88 (15)
Cl05	Sb01	Cl04	87.631 (17)	N007	C00D	C009	110.69 (16)
Cl05	Sb01	Cl06	94.156 (18)				

Table S10-6 Torsion Angles for PDASbCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C009	C00A	C00B	C00C	68.1 (2)	C00A	C00B	C00C	N008	179.15 (17)
C00A	C009	C00D	N007	177.92 (17)	C00D	C009	C00A	C00B	175.80 (17)

Table S10-7 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for PDASbCl.

Atom	x	y	z	U(eq)
H00A	10057.28	4533.36	3296.49	22
H00B	8690.65	4706.78	4053.67	22
H00C	10361.14	4178.49	4305.09	22
H00D	8515.82	7428.2	8011.95	22
H00E	10127.97	8055.7	8128.57	22
H00F	10002.53	7069.04	8626.67	22

Table S10-7 Hydrogen Atom Coordinates ($\text{\AA}\times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for PDASbCl.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
H00G	10194.73	5560.03	5650.63	18
H00H	8709.51	6172.67	5068.52	18
H00I	10795.63	7520.37	4851.37	21
H00J	12188.33	6915.38	5501.58	21
H00K	10637.61	8204.79	6389.25	20
H00L	8841.6	7588.6	6234.76	20
H00M	10036.62	6231.82	7146.51	19
H00N	11820.94	6861.28	7310.78	19
H00O	10485.24	6147.06	3657.45	22
H00P	12016.77	5569.08	4236.4	22

Table S11: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for DABCdCl

DABCdCl

Table S11-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for DABCdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
Cd1	5000	5000	0	12.35 (18)
Cl1	5349.3 (10)	2613.0 (10)	2669.5 (10)	16.5 (2)
Cl2	2323.0 (8)	4475.2 (12)	-591.4 (11)	18.8 (2)
N1	7315 (4)	5311 (4)	5670 (5)	19.3 (7)
C1	8606 (4)	4261 (5)	6419 (5)	22.5 (7)
C2	9633 (4)	4133 (4)	5138 (5)	19.6 (7)
O1	5000	5000	5000	56 (4)

Table S11-2 Anisotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for DABCdCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U ₁₁	U ₂₂	U ₃₃	U ₁₂	U ₁₃	U ₂₃
Cd1	12.4 (2)	12.3 (3)	12.5 (2)	-1.13 (10)	2.90 (15)	0.14 (9)
Cl1	18.7 (4)	15.1 (4)	16.3 (4)	1.8 (3)	5.5 (3)	5.4 (3)
Cl2	13.3 (4)	16.5 (4)	26.1 (4)	-0.7 (3)	3.2 (3)	0.1 (3)
N1	20.0 (16)	14.0 (12)	26.8 (18)	2.7 (13)	11.8 (14)	-0.3 (14)
C1	19.8 (17)	24.9 (19)	23.2 (17)	2.4 (14)	5.6 (14)	5.2 (14)
C2	20.7 (16)	15.6 (17)	23.5 (17)	0.8 (15)	7.0 (13)	-0.6 (12)
O1	25 (6)	55 (9)	75 (11)	-4 (4)	-18 (6)	-2 (4)

Table S11-3 Bond Lengths for DABCdCl.

Atom	Atom	Length/Å	Atom	Atom	Length/Å
Cd1	Cl1 ¹	2.6905 (7)	Cd1	Cl2 ¹	2.4999 (8)
Cd1	Cl1	2.6905 (7)	N1	C1	1.466 (5)
Cd1	Cl1 ²	2.7236 (7)	C1	C2	1.514 (5)
Cd1	Cl1 ³	2.7236 (7)	C2	C2 ⁴	1.517 (7)
Cd1	Cl2	2.4999 (8)			

¹1-X,1-Y,-Z; ²+X,1/2-Y,-1/2+Z; ³1-X,1/2+Y,1/2-Z; ⁴2-X,1-Y,1-Z

Table S11-4 Bond Angles for DABCdCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1	Cd1	Cl1 ¹	180.0	Cl2 ¹	Cd1	Cl1	90.53 (3)
Cl1	Cd1	Cl1 ²	90.261 (7)	Cl2	Cd1	Cl1	89.47 (3)
Cl1 ³	Cd1	Cl1 ¹	90.261 (7)	Cl2 ¹	Cd1	Cl1 ³	88.81 (3)
Cl1 ³	Cd1	Cl1 ²	180.0	Cl2	Cd1	Cl1 ¹	90.53 (3)
Cl2 ¹	Cd1	Cl1 ¹	89.47 (3)	Cl2	Cd1	Cl2 ¹	180.0
Cl2	Cd1	Cl1 ³	91.19 (3)	C2	C1	N1	111.9 (3)
Cl2	Cd1	Cl1 ²	88.81 (3)	C2 ⁴	C2	C1	113.9 (4)
Cl2 ¹	Cd1	Cl1 ²	91.19 (3)				

¹1-X,1-Y,-Z; ²1-X,1/2+Y,1/2-Z; ³+X,1/2-Y,-1/2+Z; ⁴2-X,1-Y,1-Z

Table S11-5 Torsion Angles for DABCdCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
N1	C1	C2	C2 ¹	68.7 (4)	C1	C2	C2 ¹	C1 ¹	-180.0 (4)

¹2-X,1-Y,1-Z

Table S11-6 Hydrogen Atom Coordinates ($\text{\AA}\times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for DABCdCl.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
H1d	9118 (4)	4808 (5)	7553 (5)	27.0 (9)
H1e	8303 (4)	3056 (5)	6695 (5)	27.0 (9)
H2a	9085 (4)	3709 (4)	3964 (5)	23.5 (8)
H2b	10389 (4)	3243 (4)	5600 (5)	23.5 (8)
H1a	7520 (40)	6300 (20)	5420 (50)	16 (10)
H1b	6940 (50)	4850 (50)	4740 (30)	17 (13)
H1c	6960 (60)	5410 (70)	6530 (50)	38 (15)

Table S11-7 Atomic Occupancy for DABCdCl.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
O1	0.500000				

Table S12: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for Sb:DABCdCl

Sb:DABCdCl

Table S12-1 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2\times 10^3$) for Sb:DABCdCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
Cd1	5000	0	5000	13.35 (13)
Cl1	5343.9 (9)	2389.3 (11)	7664.5 (12)	17.2 (2)
Cl2	2321.9 (9)	523.6 (12)	4409.4 (13)	20.4 (2)
N1	2696 (4)	4681 (4)	4329 (5)	20.4 (7)
C1	368 (4)	5864 (5)	4862 (6)	21.7 (8)
C2	1385 (4)	5746 (6)	3581 (6)	24.7 (9)
Sb1	5000	0	5000	13.35 (13)
O2	5000	5000	5000	69 (4)

Table S12-2 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:DABCdCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^2U_{11}+2hka*b*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Cd1	12.58 (19)	13.6 (2)	14.2 (2)	-0.15 (13)	3.58 (13)	1.21 (13)
Cl1	19.9 (4)	15.6 (4)	17.4 (4)	-5.5 (3)	6.7 (3)	-2.4 (3)
Cl2	13.5 (4)	18.7 (4)	28.7 (5)	-0.5 (4)	3.4 (3)	0.7 (3)
N1	19.8 (16)	14.6 (16)	30.0 (19)	-0.3 (14)	12.6 (14)	2.4 (13)
C1	22.0 (19)	19 (2)	25 (2)	-0.8 (16)	7.8 (16)	1.1 (16)
C2	21.0 (19)	26 (2)	27 (2)	5.4 (17)	4.5 (16)	2.1 (16)
Sb1	12.58 (19)	13.6 (2)	14.2 (2)	-0.15 (13)	3.58 (13)	1.21 (13)
O2	60 (8)	72 (9)	69 (9)	7 (7)	0 (7)	-9 (7)

Table S12-3 Bond Lengths for Sb:DABCdCl.

Atom	Atom	Length/ \AA	Atom	Atom	Length/ \AA
Cd1	Cl1 ¹	2.6908 (8)	Cl1	Sb1	2.6908 (8)
Cd1	Cl1 ²	2.7256 (8)	Cl2	Sb1	2.4976 (9)
Cd1	Cl1 ³	2.7256 (8)	N1	C2	1.485 (5)
Cd1	Cl1	2.6908 (8)	C1	C1 ⁴	1.515 (8)
Cd1	Cl2	2.4976 (9)	C1	C2	1.507 (6)
Cd1	Cl2 ¹	2.4975 (9)			

¹1-X,-Y,1-Z; ²1-X,-1/2+Y,3/2-Z; ³+X,1/2-Y,-1/2+Z; ⁴-X,1-Y,1-Z

Table S12-4 Bond Angles for Sb:DABCdCl.

Atom	Atom	Atom	Angle/ $^\circ$	Atom	Atom	Atom	Angle/ $^\circ$
Cl1 ¹	Cd1	Cl1	180.0	N1	C2	C1	111.8 (3)
Cl1 ²	Cd1	Cl1 ³	180.0	Cl1	Sb1	Cl1 ²	90.197 (6)
Cl1 ¹	Cd1	Cl1 ³	90.197 (6)	Cl1 ¹	Sb1	Cl1 ³	90.197 (6)
Cl1	Cd1	Cl1 ²	90.197 (6)	Cl1 ²	Sb1	Cl1 ³	180.0
Cl1 ¹	Cd1	Cl1 ²	89.803 (6)	Cl1 ¹	Sb1	Cl1 ²	89.803 (6)
Cl1	Cd1	Cl1 ³	89.803 (6)	Cl1 ¹	Sb1	Cl1	180.0
Cl2 ¹	Cd1	Cl1 ³	88.88 (3)	Cl1	Sb1	Cl1 ³	89.803 (6)
Cl2 ¹	Cd1	Cl1 ¹	89.32 (3)	Cl2	Sb1	Cl1 ²	88.88 (3)
Cl2	Cd1	Cl1	89.32 (3)	Cl2	Sb1	Cl1 ¹	90.68 (3)
Cl2	Cd1	Cl1 ¹	90.68 (3)	Cl2 ¹	Sb1	Cl1 ¹	89.32 (3)
Cl2	Cd1	Cl1 ²	88.88 (3)	Cl2 ¹	Sb1	Cl1 ²	91.12 (3)
Cl2	Cd1	Cl1 ³	91.12 (3)	Cl2 ¹	Sb1	Cl1	90.68 (3)

Table S12-4 Bond Angles for Sb:DABCdCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl2 ¹	Cd1	Cl1	90.68 (3)	Cl2 ¹	Sb1	Cl1 ³	88.88 (3)
Cl2 ¹	Cd1	Cl1 ²	91.12 (3)	Cl2	Sb1	Cl1	89.32 (3)
Cl2 ¹	Cd1	Cl2	180.00 (4)	Cl2	Sb1	Cl1 ³	91.12 (3)
Cd1	Cl1	Cd1 ⁴	165.91 (4)	Cl2 ¹	Sb1	Cl2	180.00 (4)
C2	C1	C1 ⁵	114.2 (4)				

¹1-X,-Y,1-Z; ²1-X,-1/2+Y,3/2-Z; ³+X,1/2-Y,-1/2+Z; ⁴1-X,1/2+Y,3/2-Z; ⁵-X,1-Y,1-Z

Table S12-5 Torsion Angles for Sb:DABCdCl.

A	B	C	D	Angle/°	ABCD	Angle/°
C1 ¹	C1	C2	N1	-68.8 (5)		

¹-X,1-Y,1-Z

Table S12-6 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:DABCdCl.

Atom	x	y	z	U(eq)
H1D	-388.43	6756.77	4409.02	26
H1E	921.87	6283.54	6034.76	26
H2A	867.7	5204.15	2447.18	30
H2B	1686.08	6951.99	3309.04	30
H1A	3220 (30)	4650 (80)	3450 (30)	53 (18)
H1B	3140 (50)	5210 (50)	5370 (30)	27 (13)
H1C	2530 (50)	3528 (18)	4560 (60)	32 (13)

Table S12-7 Atomic Occupancy for Sb:DABCdCl.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
Sb1	0.0001	O2	0.5		

Table S13: Summary of SCXRD structural analysis and tables of bond lengths, bond angles, dihedral angles for DABSbCl

DABSbCl

Table S13-1 Crystal data and structure refinement for DABSbCl.

Identification code	DABSbCl
Empirical formula	C ₄ H ₁₄ Cl ₅ N ₂ Sb
Formula weight	389.17
Temperature/K	100.00
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	9.8543(4)
b/Å	10.2894(4)
c/Å	12.9617(5)
α/°	90
β/°	106.6220(10)
γ/°	90
Volume/Å ³	1259.33(9)
Z	4
ρ _{calc} /g/cm ³	2.053
μ/mm ⁻¹	3.209
F(000)	752.0
Crystal size/mm ³	? × ? × ?
Radiation	MoKα (λ = 0.71073)
2θ range for data collection/°	4.314 to 52.802
Index ranges	-12 ≤ h ≤ 12, -12 ≤ k ≤ 12, -16 ≤ l ≤ 16
Reflections collected	28477
Independent reflections	2586 [R _{int} = 0.0339, R _{sigma} = 0.0157]
Data/restraints/parameters	2586/0/111
Goodness-of-fit on F ²	1.111
Final R indexes [I >= 2σ (I)]	R ₁ = 0.0122, wR ₂ = 0.0292
Final R indexes [all data]	R ₁ = 0.0132, wR ₂ = 0.0297
Largest diff. peak/hole / e Å ⁻³	0.37/-0.34

Table S13-2 Fractional Atomic Coordinates (×10⁴) and Equivalent Isotropic Displacement Parameters (Å²×10³) for DABSbCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	x	y	z	U(eq)
Sb01	3134.7 (2)	9166.2 (2)	4324.0 (2)	11.66 (4)
Cl02	6043.2 (4)	9489.5 (3)	3539.0 (3)	14.96 (7)
Cl03	11446.8 (4)	3712.1 (3)	8061.9 (3)	16.27 (7)
Cl04	4512.2 (4)	7370.6 (3)	5337.1 (3)	17.78 (8)
Cl05	1089.6 (4)	8539.5 (4)	4830.6 (3)	19.47 (8)
Cl06	2420.7 (4)	7862.2 (3)	2703.3 (3)	19.91 (8)

Table S13-2 Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DABSbCl. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U(eq)
N007	8706.1 (14)	5448.7 (13)	7587.3 (10)	17.7 (3)
N008	6024.2 (14)	6455.0 (12)	3001.8 (10)	16.1 (3)
C009	6700.7 (17)	5657.9 (14)	3977.8 (12)	15.6 (3)
C00A	7818.1 (16)	5474.6 (14)	6441.4 (11)	15.2 (3)
C00B	7999.4 (16)	6360.5 (15)	4663.5 (12)	16.3 (3)
C00C	8754.0 (17)	5631.0 (15)	5690.3 (12)	17.9 (3)

Table S13-3 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DABSbCl. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Sb01	10.91 (6)	11.33 (6)	12.32 (6)	0.83 (3)	2.67 (4)	-0.21 (3)
Cl02	16.79 (18)	13.39 (16)	14.77 (16)	0.41 (13)	4.65 (13)	1.26 (13)
Cl03	14.64 (17)	15.25 (16)	18.37 (17)	0.10 (13)	3.84 (14)	1.08 (13)
Cl04	17.50 (18)	14.33 (16)	20.17 (18)	3.15 (13)	3.23 (14)	3.52 (13)
Cl05	12.90 (17)	25.11 (19)	20.60 (18)	5.26 (14)	5.11 (14)	-2.04 (14)
Cl06	26.3 (2)	16.04 (17)	15.60 (17)	-2.53 (13)	3.05 (15)	-3.34 (14)
N007	14.3 (7)	23.6 (7)	14.8 (6)	-0.2 (5)	3.3 (5)	3.5 (5)
N008	16.3 (6)	15.3 (6)	15.8 (6)	-2.1 (5)	3.2 (5)	-1.3 (5)
C009	18.0 (8)	13.3 (7)	15.6 (7)	-0.2 (6)	4.9 (6)	-0.4 (6)
C00A	13.6 (7)	15.8 (7)	15.0 (7)	0.7 (6)	2.1 (6)	1.4 (6)
C00B	16.3 (8)	17.2 (7)	15.1 (7)	0.0 (6)	4.0 (6)	-1.3 (6)
C00C	13.7 (8)	21.3 (8)	18.9 (7)	1.4 (6)	5.3 (6)	3.8 (6)

Table S13-4 Bond Lengths for DABSbCl.

Atom Atom	Length/ \AA	Atom Atom	Length/ \AA
Sb01 Cl04	2.4415 (4)	N008 C009	1.4950 (19)
Sb01 Cl05	2.3802 (4)	C009 C00B	1.517 (2)
Sb01 Cl06	2.4213 (4)	C00A C00C	1.529 (2)
N007 C00A	1.4930 (19)	C00B C00C	1.524 (2)

Table S13-5 Bond Angles for DABSbCl.

Atom Atom Atom	Angle/ $^\circ$	Atom Atom Atom	Angle/ $^\circ$
Cl05 Sb01 Cl04	91.776 (13)	N007 C00A C00C	110.28 (12)

Table S13-5 Bond Angles for DABSbCl.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl05	Sb01	Cl06	91.741 (13)	C009	C00B	C00C	113.35 (12)
Cl06	Sb01	Cl04	91.501 (13)	C00B	C00C	C00A	112.05 (12)
N008	C009	C00B	109.37 (12)				

Table S13-6 Torsion Angles for DABSbCl.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
N007	C00A	C00C	C00B	149.41 (13)	C009	C00B	C00C	C00A	-62.92 (17)
N008	C009	C00B	C00C	179.16 (12)					

Table S13-7 Hydrogen Atom Coordinates ($\text{\AA} \times 10^4$) and Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for DABSbCl.

Atom	x	y	z	U(eq)
H00A	9359.52	4801.78	7678.04	21
H00B	8144.03	5303.18	8023.53	21
H00C	9156.12	6225.21	7758.81	21
H00D	5236.54	6040.77	2594.88	19
H00E	6646.76	6564.74	2608.35	19
H00F	5777.56	7244.85	3207.96	19
H00G	6976.79	4799.26	3757.92	19
H00H	6019.16	5516.34	4399	19
H00I	7138.91	6205.95	6332.42	18
H00J	7270.35	4656.84	6269.43	18
H00K	7712	7226.85	4858.97	20
H00L	8670.57	6495.16	4232.51	20
H00M	9625.25	6108.36	6071.94	21
H00N	9037.3	4761.1	5499.05	21

Sb:DABCdCl_Dehydrated

Table S14-1 Crystal data and structure refinement for Sb:DABCdCl_Dehydrated.

Identification code	Sb:DABCdCl_Dehydrated
Empirical formula	C ₄ H ₁₄ CdCl ₄ N ₂ Sb
Formula weight	344.37
Temperature/K	100.00

Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	9.3984(5)
b/Å	7.5570(4)
c/Å	7.6313(4)
α/°	90
β/°	102.104(2)
γ/°	90
Volume/Å ³	529.95(5)
Z	2
ρ _{calc} /g/cm ³	2.158
μ/mm ⁻¹	3.014
F(000)	336.0
Crystal size/mm ³	0.18 × 0.17 × 0.17
Radiation	MoKα (λ = 0.71073)
2θ range for data collection/°	6.982 to 61.014
Index ranges	-13 ≤ h ≤ 13, -10 ≤ k ≤ 10, -9 ≤ l ≤ 10
Reflections collected	10427
Independent reflections	1619 [R _{int} = 0.0513, R _{sigma} = 0.0331]
Data/restraints/parameters	1619/0/53
Goodness-of-fit on F ²	1.091
Final R indexes [I ≥ 2σ (I)]	R ₁ = 0.0248, wR ₂ = 0.0611
Final R indexes [all data]	R ₁ = 0.0276, wR ₂ = 0.0619
Largest diff. peak/hole / e Å ⁻³	0.91/-0.88

Table S14-2 Fractional Atomic Coordinates (×10⁴) and Equivalent Isotropic Displacement Parameters (Å²×10³) for Sb:DABCdCl₂ Dehydrated. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	x	y	z	U(eq)
Cd01	5000	5000	5000	12.20 (8)
Cl2	4647.6 (6)	7379.5 (6)	2319.3 (7)	15.65 (12)
Cl1	2313.5 (6)	4476.1 (7)	4414.6 (8)	16.97 (12)
N1	2716 (2)	314 (3)	4314 (3)	19.1 (4)
C2	368 (2)	-870 (3)	4864 (3)	16.3 (4)
C1	1390 (2)	-754 (3)	3570 (3)	18.0 (4)
Sb1	5000	5000	5000	12.20 (8)

Table S14-3 Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for Sb:DABCdCl₂ Dehydrated.
The Anisotropic displacement factor exponent takes the form: -
 $2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
Cd01	13.52 (12)	11.23 (12)	11.85 (13)	0.12 (7)	2.68 (8)	-0.75 (7)
Cl2	18.7 (2)	13.7 (2)	15.2 (2)	3.38 (19)	5.02 (19)	1.25 (17)
Cl1	14.5 (2)	14.3 (2)	21.5 (3)	0.5 (2)	2.2 (2)	-0.43 (18)
N1	16.0 (9)	16.4 (9)	25.9 (12)	0.1 (8)	6.6 (8)	-0.4 (7)
C2	17.3 (10)	13.7 (10)	18.5 (11)	1.9 (9)	5.1 (8)	-0.8 (8)
C1	17.7 (10)	18.9 (11)	17.9 (11)	-3.3 (9)	5.3 (9)	-1.9 (8)
Sb1	13.52 (12)	11.23 (12)	11.85 (13)	0.12 (7)	2.68 (8)	-0.75 (7)

Table S14-4 Bond Lengths for Sb:DABCdCl₂ Dehydrated.

Atom	Atom	Length/ \AA	Atom	Atom	Length/ \AA
Cd01	Cl2 ¹	2.6915 (5)	Cl2	Sb1	2.6915 (5)
Cd01	Cl2 ²	2.7221 (5)	Cl1	Sb1	2.5016 (5)
Cd01	Cl2	2.6915 (5)	N1	C1	1.493 (3)
Cd01	Cl2 ³	2.7221 (5)	C2	C2 ⁴	1.521 (4)
Cd01	Cl1 ¹	2.5016 (5)	C2	C1	1.517 (3)
Cd01	Cl1	2.5016 (5)			

¹1-X,1-Y,1-Z; ²1-X,-1/2+Y,1/2-Z; ³X,3/2-Y,1/2+Z; ⁴-X,-Y,1-Z

Table S14-5 Bond Angles for Sb:DABCdCl₂ Dehydrated.

Atom	Atom	Atom	Angle/ $^\circ$	Atom	Atom	Atom	Angle/ $^\circ$
Cl2 ¹	Cd01	Cl2	180.0	N1	C1	C2	112.06 (19)
Cl2 ¹	Cd01	Cl2 ²	89.760 (5)	Cl2 ¹	Sb1	Cl2 ²	89.760 (5)
Cl2	Cd01	Cl2 ²	90.240 (5)	Cl2	Sb1	Cl2 ³	89.760 (5)
Cl2 ²	Cd01	Cl2 ³	180.000 (17)	Cl2 ¹	Sb1	Cl2	180.0
Cl2 ¹	Cd01	Cl2 ³	90.240 (5)	Cl2	Sb1	Cl2 ²	90.240 (5)
Cl2	Cd01	Cl2 ³	89.760 (5)	Cl2 ¹	Sb1	Cl2 ³	90.240 (5)
Cl1 ¹	Cd01	Cl2 ²	88.590 (17)	Cl2 ²	Sb1	Cl2 ³	180.000 (17)
Cl1	Cd01	Cl2 ²	91.410 (18)	Cl1	Sb1	Cl2 ²	91.410 (18)
Cl1	Cd01	Cl2 ¹	89.354 (18)	Cl1 ¹	Sb1	Cl2 ³	91.410 (17)
Cl1 ¹	Cd01	Cl2 ¹	90.646 (18)	Cl1	Sb1	Cl2 ³	88.590 (18)
Cl1 ¹	Cd01	Cl2	89.354 (18)	Cl1	Sb1	Cl2 ¹	89.354 (18)
Cl1	Cd01	Cl2	90.646 (18)	Cl1	Sb1	Cl2	90.646 (18)
Cl1 ¹	Cd01	Cl2 ³	91.410 (17)	Cl1 ¹	Sb1	Cl2 ¹	90.646 (18)
Cl1	Cd01	Cl2 ³	88.590 (18)	Cl1 ¹	Sb1	Cl2 ²	88.590 (17)

Table S14-5 Bond Angles for Sb:DABCdCl_Dehydrated.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
Cl1 ¹	Cd01	Cl1	180.0	Cl1 ¹	Sb1	Cl2	89.354 (18)
Cd01	Cl2	Cd01 ⁴	165.44 (2)	Cl1 ¹	Sb1	Cl1	180.0
C1	C2	C2 ⁵	114.0 (2)				

¹1-X,1-Y,1-Z; ²1-X,-1/2+Y,1/2-Z; ³+X,3/2-Y,1/2+Z; ⁴1-X,1/2+Y,1/2-Z; ⁵-X,-Y,1-Z

Table S14-6 Torsion Angles for Sb:DABCdCl_Dehydrated.

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C2 ¹	C2	C1	N1	68.9 (3)					

¹-X,-Y,1-Z

Table S14-7 Hydrogen Atom Coordinates (Å×10⁴) and Isotropic Displacement Parameters (Å²×10³) for Sb:DABCdCl_Dehydrated.

Atom	x	y	z	U(eq)
H1A	3304.37	342.72	3507.97	23
H1B	3201.91	-186.77	5352.8	23
H1C	2448.66	1435.65	4535.49	23
H2A	924.52	-1288.16	6038.49	20
H2B	-391.24	-1762.82	4412.24	20
H1D	1686.82	-1962.16	3296.62	22
H1E	868.91	-210.68	2435.77	22

Table S14-8 Atomic Occupancy for Sb:DABCdCl_Dehydrated.

Atom	Occupancy	Atom	Occupancy	Atom	Occupancy
Sb1	0.0003				

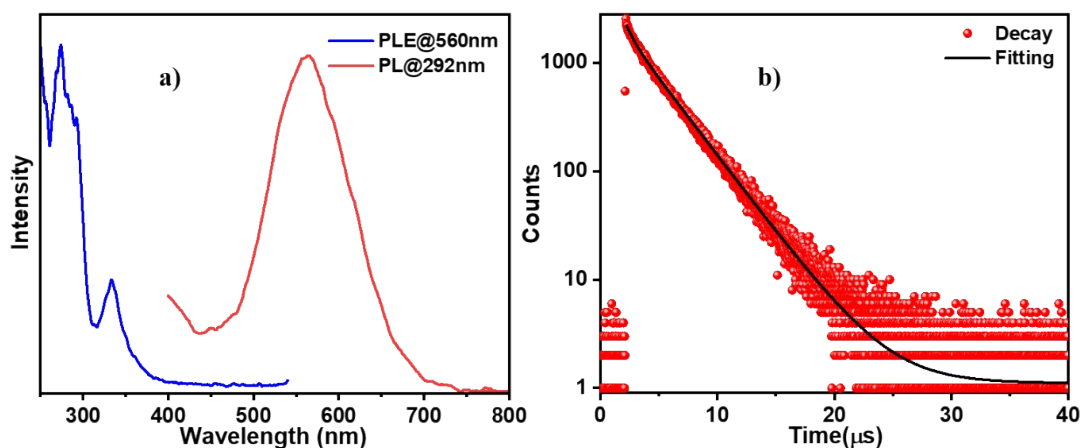


Figure S22: a) PL / PLE of dehydrated Sb:DABCdCl b) Lifetime Decay profile of dehydrated Sb:DABCdCl (Ex:310 nm Em:560 nm).

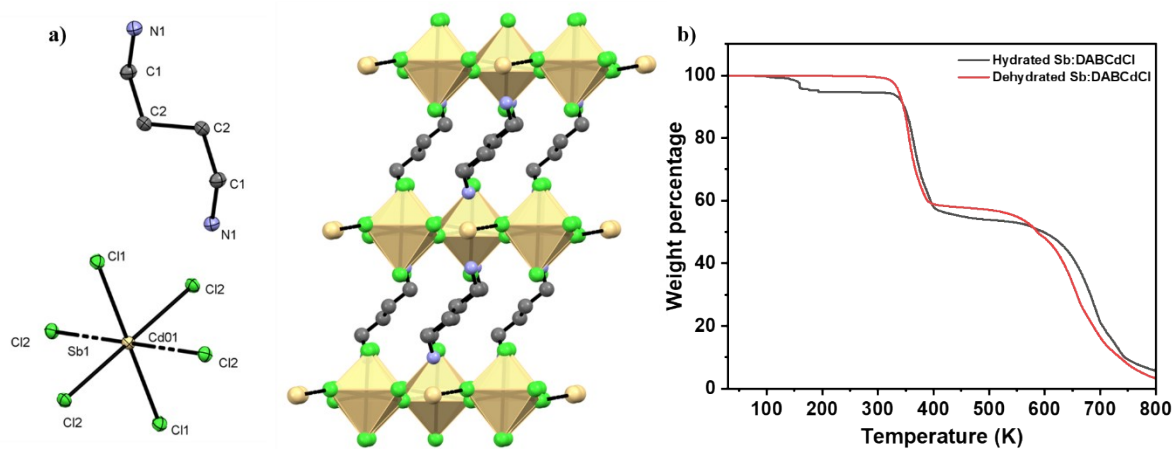


Figure S23: a) SCXRD structure of dehydrated Sb:DABCdCl hybrid b) TGA of Sb:DABCdCl [Black-Hydrated Sb:DABCdCl, Red- Dehydrated Sb:DABCdCl]

Table S14: Distortion parameters and optical properties of Hydrated and dehydrated Sb:DABCdCl

Compound	Space group	Octahedral distortion (d)	Bond angle variance (deg ²)	M-Cl-M (deg)	ϕ_{tilt} (deg)	PL Max (nm)	PLE Peak positions (nm)	PLQY(%)	Lifetime (μs)
Sb:DABCdCl 2(C ₄ H ₁₄ N ₂) Cd ₂ Cl ₈ O (Hydrated)	Monoclinic <i>P2₁/c</i>	1.4x10 ⁻³	0.6409	165.91	13.02	560	275, 291, 333	96	3.1 μs
Sb:DABCdCl C ₄ H ₁₄ N ₂ CdCl ₄ (Dehydrated)	Monoclinic <i>P2₁/c</i>	1.5x10 ⁻³	0.8956	165.44	12.99	560	275, 291, 333	39	3.02 μs

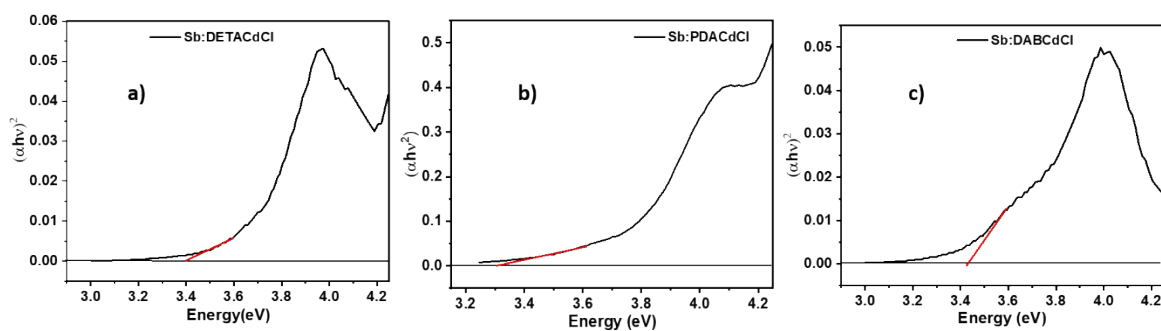


Figure S24: Tauc plot analysis of the doped hybrids- a) Sb:DETACdCl, b) Sb:PDACdCl, c) Sb:DABCdCl

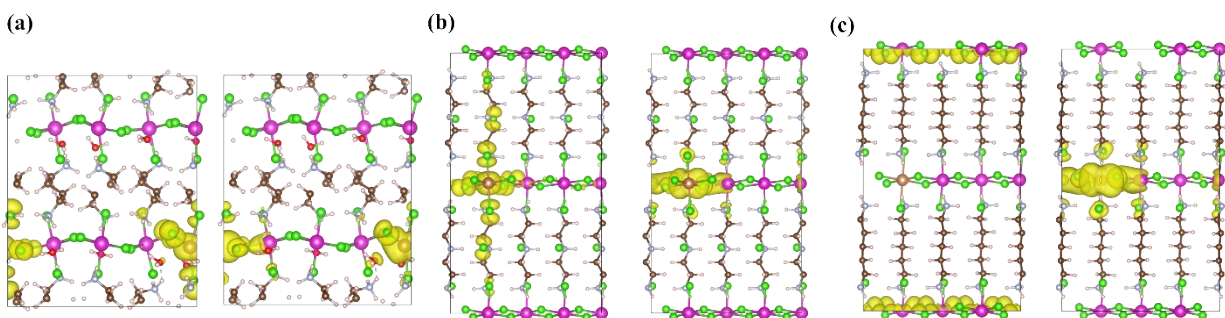


Figure S25: Band edge charge density localization of (a) Sb:DABCdCl (b) Sb:DETACdCl (c) Sb:PDACdCl.

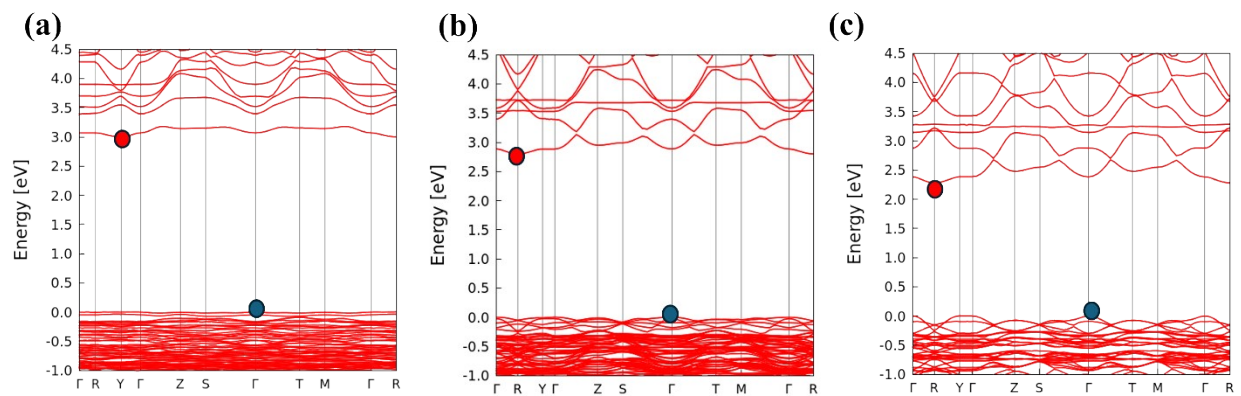


Figure S26: Band structure of (a) Sb:DABCdCl (b) Sb:DETACdCl (c) Sb:PDACdCl.

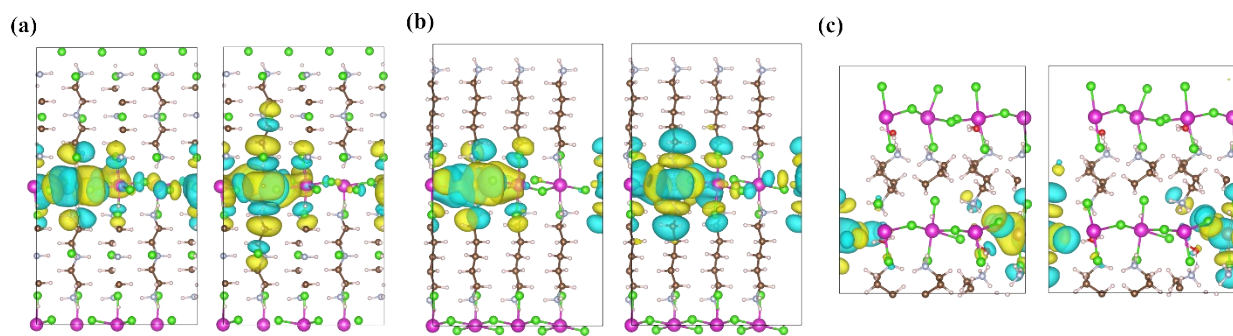


Figure S27: STE formation in the studied structure and their Kohn-Sham (KS) orbitals of the electron and hole in triplet state of (a) Sb:DETACdCl (b) Sb:PDACdCl and (c) Sb:DABCdCl.