## **Supplementary information**

## Revealing the surface oxidation mechanism and performance evolution of Nd-Fe-B sintered magnets

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**Fig. S1** Electrochemical equivalent circuit in impedance measurement for (a) the original Nd-Fe-B magnet, and (b) the oxidized magnets.



**Fig. S2** Mechanical performance of (a) 0.5 h and (b) 3 h oxidized Nd-Fe-B magnets as a function of oxidation temperature, including Modulus and Hardness.



**Fig. S3** Surface morphologies of (a) 550 °C-0.5 h, (b) 550 °C-3 h, (c) 650 °C-0.5 h and (d) 650 °C-3 h oxidized Nd-Fe-B magnets.



**Fig. S4** Low-magnification surface morphologies of (a-d) 0.5 h and (e-f) 3 h oxidized Nd-Fe-B magnets at different temperatures: (a) 250 °C-0.5 h, (b) 450 °C-0.5 h, (c, d) 650 °C-0.5 h, (e) 250 °C-3 h, (f) 450 °C-3 h and (g, h) 650 °C-3 h.



**Fig. S5** Cross-sectional BSE SEM image and corresponding EPMA elemental mappings of the 650 °C-0.5 h oxidized magnet.



**Fig. S6** Focused ion beam-assisted preparation of the TEM sample from the different areas in the 650 °C-3 h oxidized magnet. (a) Region of interest (ROI 1) from the interface between the outmost surface zone A and the internal oxidation zone B in **Fig.** 7(a). (b) ROI 2 of the continuous and coarse grain boundary in the internal oxidation layer.



Fig. S7 STEM-EDS mappings for the continuous grain boundary region obtained from Fig. 9(a).

Samples	E <sub>corr</sub> (V)	$I_{\rm corr}$ ( $\mu {\rm A} \cdot {\rm cm}^{-2}$ )	$R_{\rm s}$ ( $\Omega \cdot {\rm cm}^2$ )	$\begin{array}{c} Q_{c} \\ (F \cdot cm^{-2} \cdot s^{n-1}) \end{array}$	n <sub>c</sub>	$R_{\rm c}$ ( $\Omega \cdot { m cm}^2$ )	$\begin{array}{c} Q_{\rm dl} \\ ({\rm F}{\cdot}{\rm cm}^{-2}{\cdot}{\rm s}^{{\rm n}{\rm -1}}) \end{array}$	n <sub>dl</sub>	$R_{\rm ct}$ ( $\Omega \cdot {\rm cm}^2$ )	L (H·cm²)	$R_{\rm L}$ ( $\Omega \cdot {\rm cm}^2$ )
Original	-0.911	31.2	28.6	/	/	/	$2.82 \times 10^{-4}$	0.710	1450	1558	594
250 °C	-0.852	10.8	30.7	1.06×10 <sup>-4</sup>	0.773	12.2	$6.87 \times 10^{-5}$	0.788	2527	/	/
350 °C	-0.854	7.3	31.6	2.45×10 <sup>-4</sup>	0.812	22.1	4.87×10 <sup>-9</sup>	0.775	4584	/	/
450 °C	-0.808	9.3	31.9	1.25×10 <sup>-4</sup>	0.770	62.14	1.06×10 <sup>-4</sup>	0.763	3328	/	/
550 °C	-0.808	10	32.5	1.06×10 <sup>-4</sup>	0.766	277.8	2.26×10 <sup>-4</sup>	0.769	2748	/	/
650 °C	-0.877	21.4	32.5	$1.98 \times 10^{-4}$	0.830	543.0	$1.37 \times 10^{-4}$	0.882	1475		

**Table S1.** Electrochemical parameters of the original and 0.5 h oxidized Nd–Fe–B magnets under different oxidation temperatures in 3.5% NaCl solution.

**Table S2.** Electrochemical parameters of the original and 3 h oxidized Nd–Fe–Bmagnets under different oxidation temperatures in 3.5% NaCl solution.

Samples	$E_{\rm corr}$ (V)	$I_{\rm corr}$ ( $\mu {\rm A} \cdot {\rm cm}^{-2}$ )	$R_{\rm s}$ ( $\Omega \cdot { m cm}^2$ )	$\frac{Q_{\rm c}}{({\rm F}\cdot{\rm cm}^{-2}\cdot{\rm s}^{\rm n-1})}$	n <sub>c</sub>	$R_{\rm c}$ ( $\Omega \cdot {\rm cm}^2$ )	$Q_{\rm dl}$ (F·cm <sup>-2</sup> ·s <sup>n-1</sup> )	n <sub>dl</sub>	$R_{\rm ct}$ ( $\Omega$ ·cm <sup>2</sup> )	L (H·cm <sup>2</sup> )	$R_{\rm L}$ ( $\Omega \cdot { m cm}^2$ )
Original	-0.911	31.2	28.6	/	/	/	2.82×10-4	0.710	1450	1558	594
250 °C	-0.820	7.6	29.1	1.30×10 <sup>-4</sup>	0.797	18.2	8.07×10 <sup>-4</sup>	0.838	3356	/	/
350 °C	-0.840	7.9	35.5	1.36×10-4	0.742	23.3	3.75×10-4	0.768	3273	/	/
450 °C	-0.881	26.0	33.5	3.78×10-4	0.787	418.9	2.34×10-4	0.861	1469	/	/
550 °C	-0.884	29.8	37.8	1.71×10 <sup>-4</sup>	0.718	100.9	9.46×10-4	0.726	1059	/	/
650 °C	-0.895	38.1	33.6	5.25×10 <sup>-4</sup>	0.764	134.6	1.13×10 <sup>-4</sup>	0.785	547	/	/