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Supplementary material



20 nm

Fig. S1. TEM image and microdiffraction patterns of the reference sample.



Fig. S2. Mössbauer spectrum of the reference sample at a temperature of 300 K. Shaded areas show the partial components obtained during processing.

Table S1. Mössbauer parameters of the sample in the temperature range of 4–300 K. IS is the chemical shift relative to α -Fe, H_{hf} is the hyperfine field on iron nuclei, QS is the quadrupole splitting, W is the Mössbauer line full width at half maximum, and A is the relative site occupancy.

	IS, mm/s	Н,	QS, mm/s	W, mm/s	A, %	Origin		
	±0.005	kOe, ±3	±0.01	±0.01	±0.03	Origin		
300 К								
D1	0.363		0.56	0.34	0.45			
D2	0.365		0.97	0.35	0.36	Fe₂O₃• <i>n</i> H₂O		
D3	0.371		1.37	0.37	0.19			



Fig. S3. FC and ZFC temperature dependences of the magnetization of the reference sample in a magnetic field of 1000 Oe.

Table S2. Averaging of the normalized atomic concentration in the sample according to the EDS data

Element	Series	Intensity	Concentration	Concentration	(3 Sigma)				
		[a.u.]	[wt %]	[at %]	[wt %]				
Element Series Intensity Concentration Concentration (3 Sigma) [a.u.] [wt %] [at %] [wt %] [wt %] Selected area (1) Iron K-series 39898 42.78 20.86 3.99 Phosphorus K-series 22919 18.7 16.44 1.8 Oxygen K-series 22307 33.66 57.31 3.18 Sulfur K-series 317 0.26 0.22 0.13 Magnesium K-series 5955 4.6 5.15 0.53 Total: 100 100 100 Selected area (2) Iron K-series 30165 55.47 41.22 5.17 Phosphorus K-series 261 0.39 0.52 0.16 Oxygen K-series 334 0.86 2.24 0.24									
Iron	K-series	39898	42.78	20.86	3.99				
Phosphorus	K-series	22919	18.7	16.44	1.8				
Oxygen	K-series	22307	33.66	57.31	3.18				
Sulfur	K-series	317	0.26	0.22	0.13				
Magnesium	K-series	5955	4.6	5.15	0.53				
Total:			100	100					
Selected area (2)									
Iron	K-series	30165	55.47	41.22	5.17				
Phosphorus	K-series	261	0.39	0.52	0.16				
Oxygen	K-series	334	0.86	2.24	0.24				
Sulfur	K-series	32373	43.28	56.01	4.04				
Total:			100	100					

Table S3. Mössbauer parameters of the sample from 4 to 300 K. δ is the chemical shift relative to α -Fe, H_{hf} is the hyperfine field on iron nuclei, Δ is the quadrupole splitting, W is the Mössbauer line FWHM, dW is the line broadening, and A is the relative site occupancy.

	δ, mm/s	H _{hf} ,	Δ/2, mm/s	W, mm/s	dW, mm/s	A, a.u	Origin	
	±0.005	kOe, ±3	±0.01	±0.01	±0.01	±0.03		
300 K								
S1	0.437	304	-0.15	0.74	0.47	0.05	Fe_3S_4	
D1	1.214		2.51	0.39		0.10		
D2	1.266		2.92	0.22		0.05	re ₃ (PO ₄) ₂ ·on ₂ O	
D3	1.242		1.84	0.29		0.13	FeCO ₃	
D4	0.365		0.54	0.42		0.29		
D5	0.372		0.67	0.44		0.14	$Fe_2O_3 \cdot nH_2O$	
D6	0.34		1.03	0.47		0.24		
				250 K				
S1	0.447	306	-0.15	0.74	0.47	0.05	Fe_3S_4	
S2	0.388	311	-0.15	0.10	2.85	0.25	$Fe_2O_3 \cdot nH_2O$	
D1	1.351		2.47	0.40		0.09		
D2	1.353		2.96	0.22		0.04	Fe ₃ (PO ₄) ₂ ·8n ₂ O	
D3	1.284		1.94	0.34		0.11	FeCO ₃	
D4	0.378		0.50	0.40		0.16		
D5	0.377		0.82	0.46		0.18	$Fe_2O_3 \cdot nH_2O$ (SPM)	
D6	0.336		1.15	0.40		0.12		
				200 K				
S1	0.459	314	-0.12	0.30	0.00	0.05	Fe ₃ S ₄	
S2	0.449	410	-0.27	1.07	0.00	0.11		
S3	0.567	333	0.00	0.10	3.64	0.27	Fe ₂ O ₃ • <i>n</i> H ₂ O	
D1	1.364		2.38	0.42		0.10		
D2	1.441		2.86	0.20		0.03	163(104)2.0120	
D3	1.317		1.91	0.31		0.09	FeCO ₃	
D4	0.414		0.51	0.43		0.15		
D5	0.342		0.98	0.47		0.08	(SDM)	
D6	0.379		1.16	0.39		0.12	(38101)	
				140 K				
S1	0.527	322	0.03	0.52	0.0	0.04	Fe_3S_4	
S2	0.407	434	-0.02	0.65	0.50	0.16	$Fe_2O_3 \cdot nH_2O$	
S3	0.348	367	-0.38	0.19	2.79	0.29		
D1	1.417		2.60	0.41		0.08		
D2	1.306		3.28	0.18		0.04		
D3	1.348		2.07	0.33		0.10	FeCO ₃	
D4	0.396		0.64	0.59		0.18	$Fe_2O_3 \cdot nH_2O$	
D5	0.418		0.91	0.37		0.10	(SPM)	

	δ, mm/s	H _{hf} ,	Δ/2, mm/s	W, mm/s	dW, mm/s	A, a.u	Origin	
	±0.005	kOe, ±3	±0.01	±0.01	±0.01	±0.03	0.18	
110 К								
S1	0.567	321	0.01	0.37	0.0	0.04	Fe ₃ S ₄	
S2	0.525	445	-0.14	0.67	0.56	0.27	FeaOa • nHaO	
S3	0.518	325	-0.13	0.10	3.74	0.22		
D1	1.422		2.35	0.31		0.07	Fe ₂ (PO ₄) ₂ ,8H ₂ O	
D2	1.357		3.10	0.35		0.05		
D3	1.370		2.04	0.30		0.09	FeCO ₃	
D4	0.411		0.72	0.44		0.11	Fe ₂ O ₃ • <i>n</i> H ₂ O	
D5	0.466		0.88	0.61		0.15	(SPM)	
				80 K				
S1	0.562	323	0.11	0.43	0.07	0.05	Fe ₃ S ₄	
S2	0.503	465	-0.08	0.53	0.43	0.22		
S3	0.474	392	-0.13	0.48	1.97	0.30	Fe ₂ O ₃ • <i>n</i> H ₂ O	
D1	1.472		2.55	0.40		0.08		
D2	1.372		3.24	0.24		0.04	Fe3(FO4)2.0H2O	
D3	1.399		2.10	0.37		0.09	FeCO ₃	
D4	0.412		0.64	0.46		0.11	Fe ₂ O ₃ • <i>n</i> H ₂ O	
D5	0.459		1.06	0.50		0.12	(SPM)	
				50 K				
S1	0.612	326	0.05	0.44	0.10	0.04	Fe ₃ S ₄	
S2	0.519	483	-0.10	0.49	0.24	0.19		
S3	0.481	347	-0.34	0.10	3.89	0.21	Fe ₂ O ₃ • <i>n</i> H ₂ O	
S4	0.490	443	0.00	0.45	0.74	0.18		
D1	1.511		2.64	0.69		0.09		
D2	1.371		3.27	0.17		0.04	Fe3(PO4)2.0H2O	
D3	1.429		2.21	0.52		0.08	FeCO ₃	
D4	0.393		0.66	0.58		0.09	Fe ₂ O ₃ • <i>n</i> H ₂ O	
D5	0.449		0.99	0.55		0.10	(SPM)	
				4.2 K				
S1	0.624	327	0.00	0.42	0.16	0.05	Fe ₃ S ₄	
S2	0.510	490	-0.17	0.38	0.39	0.28		
S3	0.512	512	-0.08	0.42	0.11	0.18	Fe ₂ O ₃ • <i>n</i> H ₂ O	
S4	0.501	461	0.03	0.51	0.96	0.24		
S5	1.231	223	-2.44	1.33	0.40	0.08		
S6	1.423	136	2.58	0.26	0.46	0.07	re3(rU4)2.9H2O	
S7	1.359	169	2.05	0.20	1.00	0.12	FeCO ₃	