

Supporting information for Porosity and fluid pathway development during cadmium sequestration by calcium carbonate replacement

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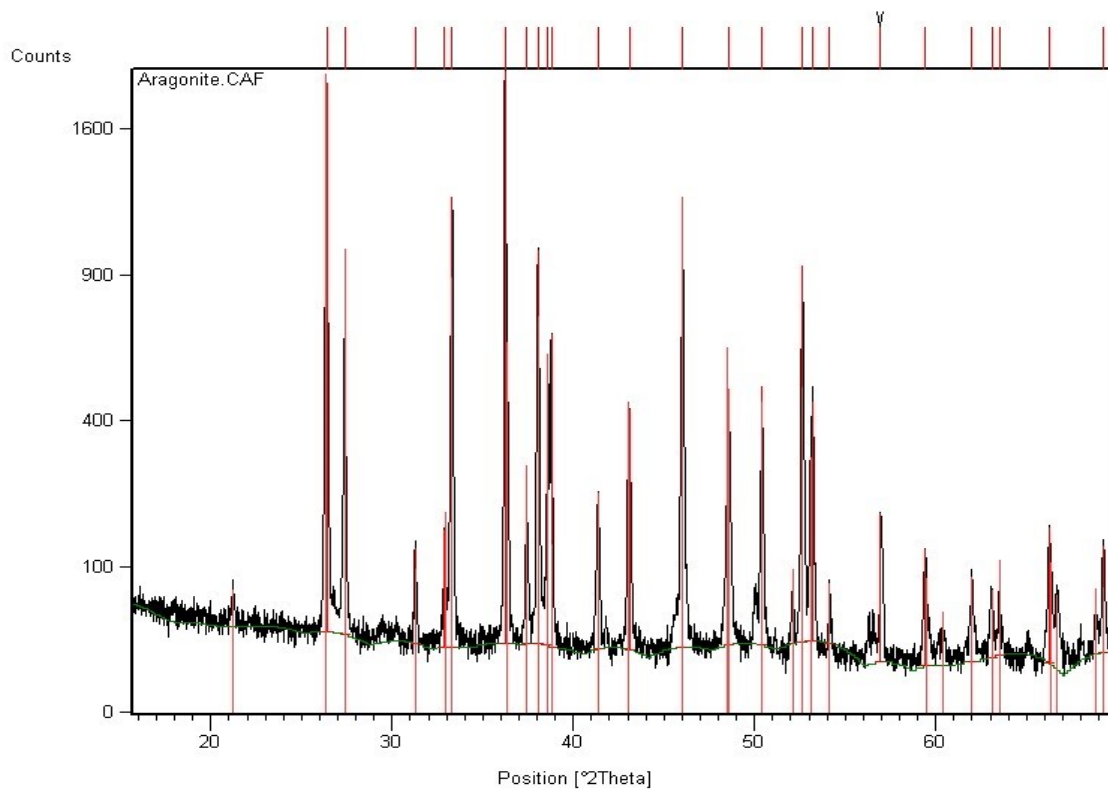


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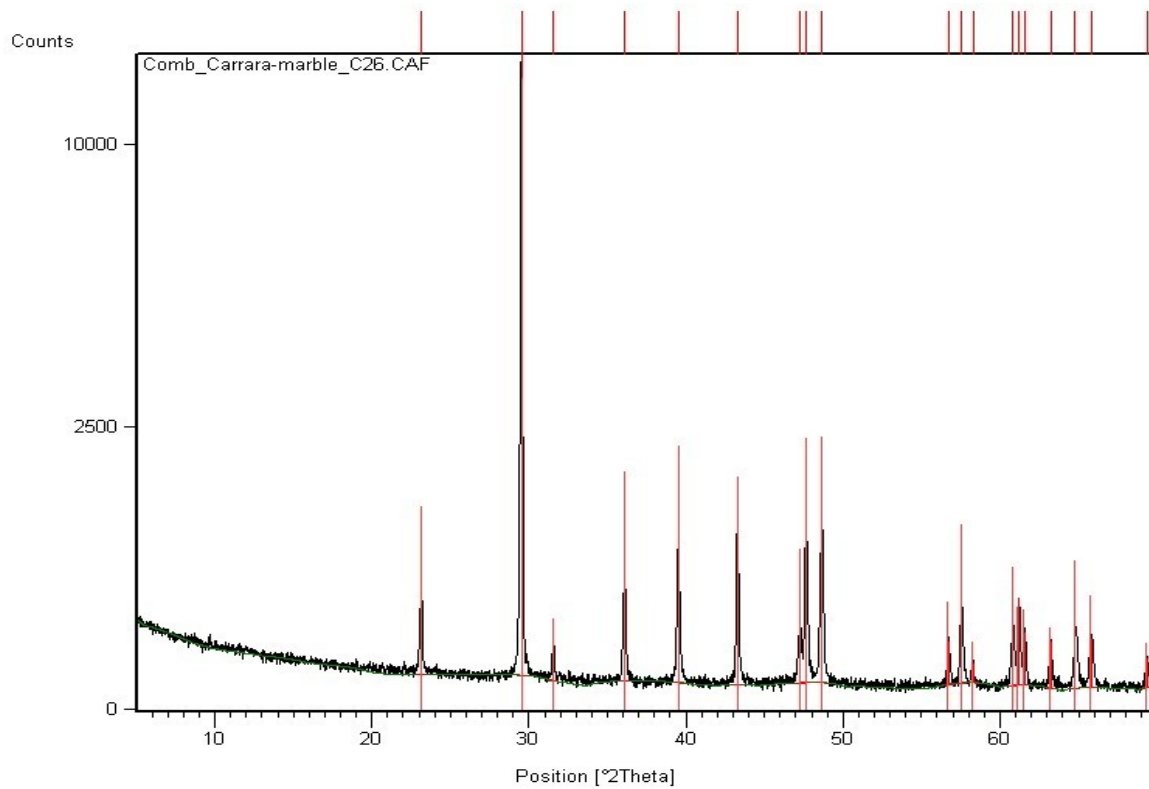


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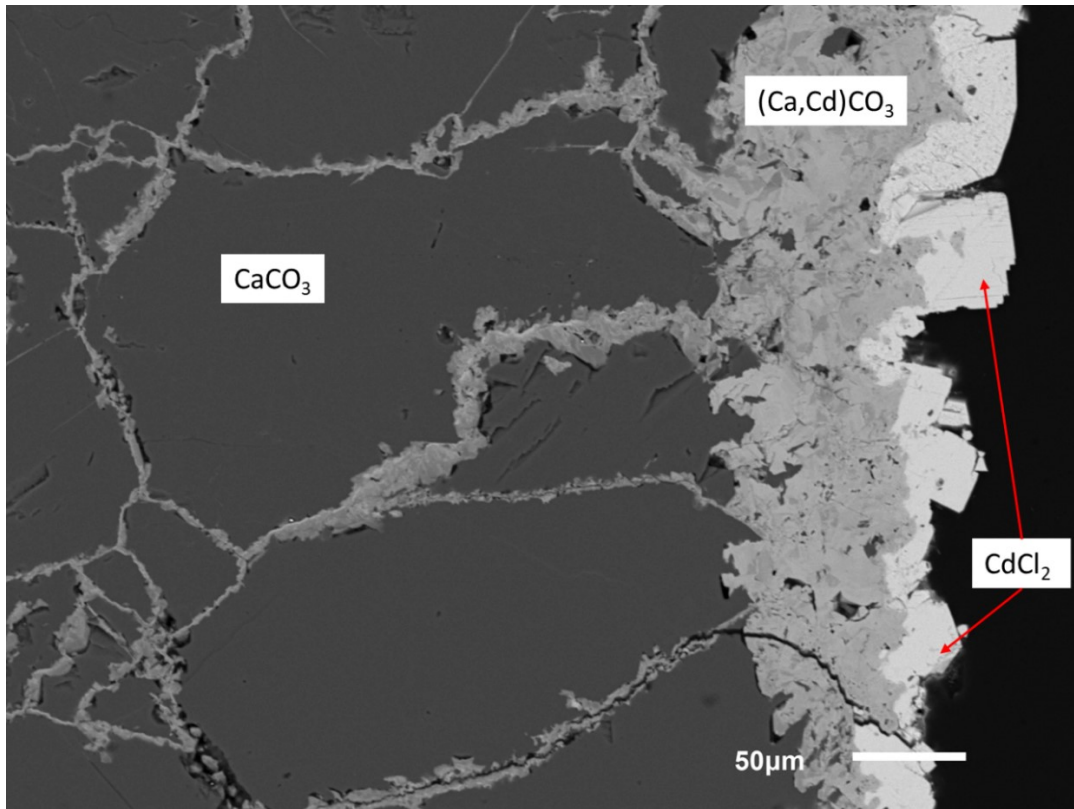


Figure S3: Back-scattered electron image of a Carrara marble sample reacted with similar conditions to Carrara – 05. Crystals of CdCl_2 are visible on the sample surface as a result of precipitation during cooling at the end of the experiments.

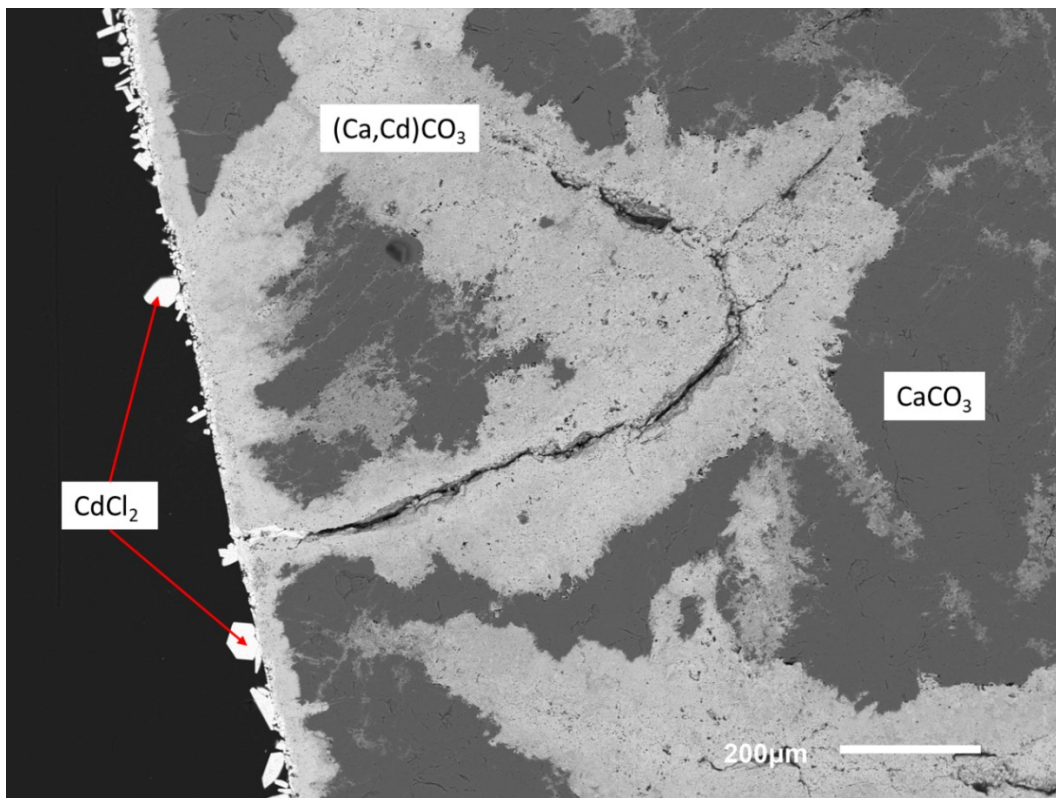


Figure S4: Back-scattered electron image of an aragonite sample reacted with similar to Aragonite – 02. Crystals of CdCl_2 are visible on the sample surface as a result of precipitation during cooling at the end of the experiments.

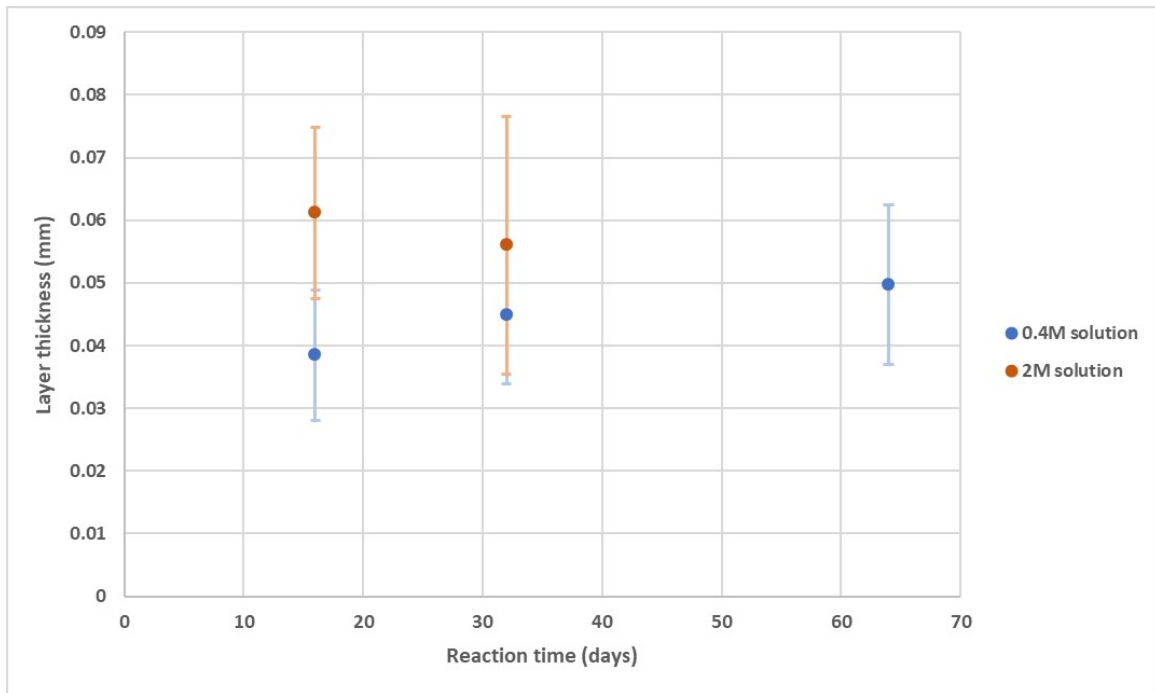


Figure S5: Evolution of the thickness of the surface reaction layer on the aragonite samples with reaction time and solution concentration (Aragonite – 01 to 05). The standard deviation of one hundred measurements on each sample is indicated by the error bars.

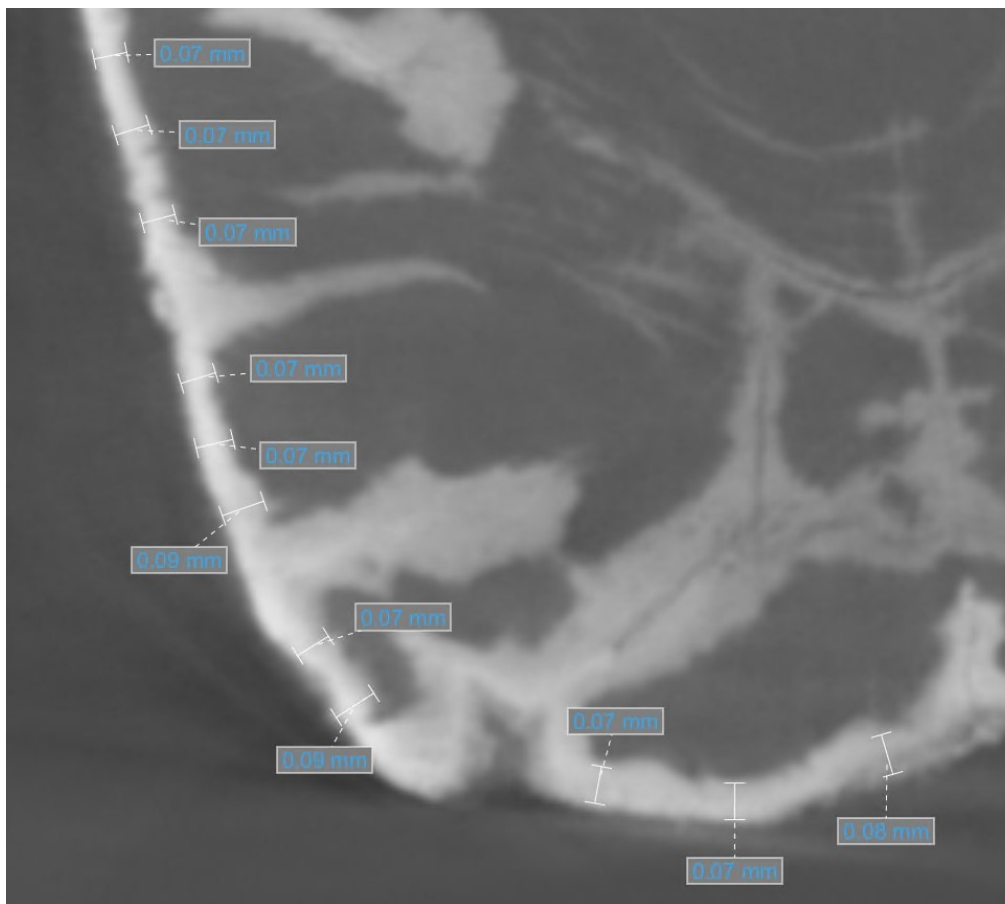


Figure S6: Sample Aragonite-04 reacted for 16 days in a 2M CdCl_2 solution tomograph slice with measurements of the thickness of the reacted layer surrounding the sample. Note also the invasive dissolution-precipitation reaction due to reaction-induced fracturing in this sample.

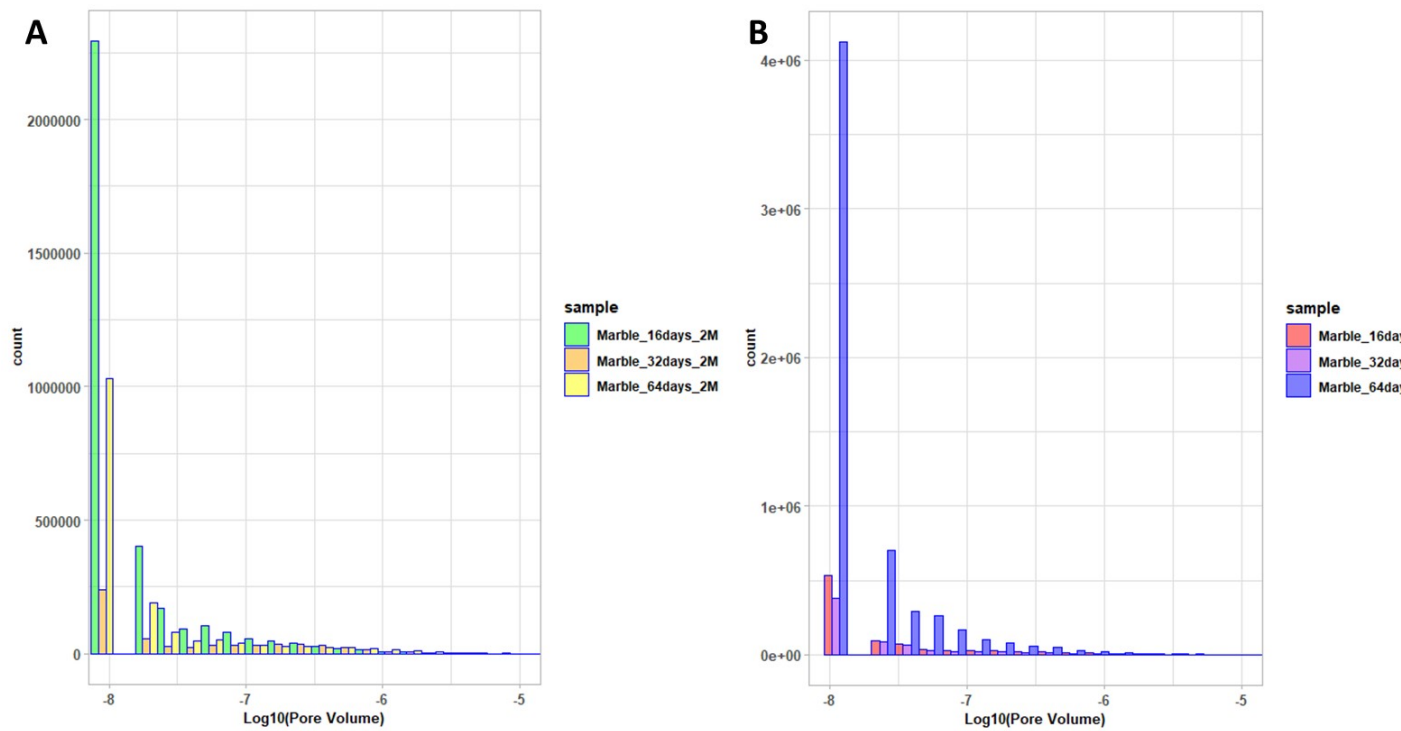


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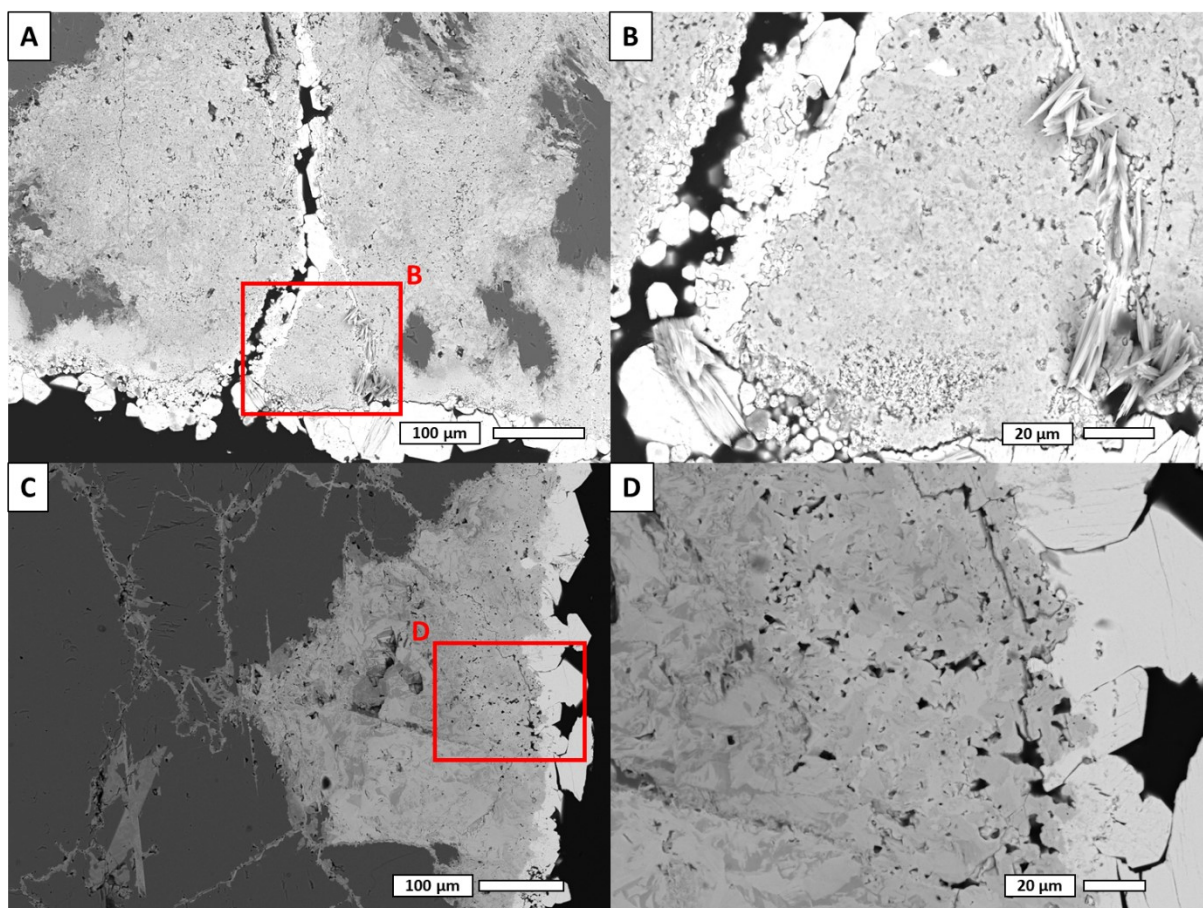


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