

Rational designing of NiMoO₄/carbon nanocomposites for high-performance supercapacitors: an *in situ* carbon incorporation approach

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1. The calculation for the compositions of NiMoO₄ and carbon components in NiMoO₄/C (D50) nanocomposite using TGA curves:

At 600 °C:

Wt. retention percentage (x%) of NiMoO₄ = 97.4%

Wt. retention percentage (y%) of CNS = 0.9%

Wt. retention percentage (composite%) of NiMoO₄/C (D50) = 69.5%

Assume, the proportion of NiMoO₄ and carbon components in the nanocomposite as 'a' and 'b', respectively.

$$a + b = 1 \tag{1}$$

$$x\% \times a + y\% \times b = \text{composite \%} \times 1$$

$$\left(\frac{97.4}{100} \times a\right) + \left(\frac{0.9}{100} \times b\right) = \left(\frac{69.5}{100} \times 1\right)$$

$$0.974a + 0.009b = 0.695 \tag{2}$$

Equation (2) is divided by 0.009

$$\frac{0.974a}{0.009} + \frac{0.009b}{0.009} = \frac{0.695}{0.009}$$

$$108.222 a + b = 77.22 \tag{3}$$

Subtract equation (1) from equation (3), and simplify

$$107.222a + 0 = 76.2$$

$$a = \frac{76.22}{107.222}$$

$$a = 0.711$$

a = 71.1% of NiMoO₄ and b = 28.9% of carbon components.

Therefore, 71.1% of NiMoO_4 and 28.9% of carbon components are calculated to be in NiMoO_4/C (D50) nanocomposite.

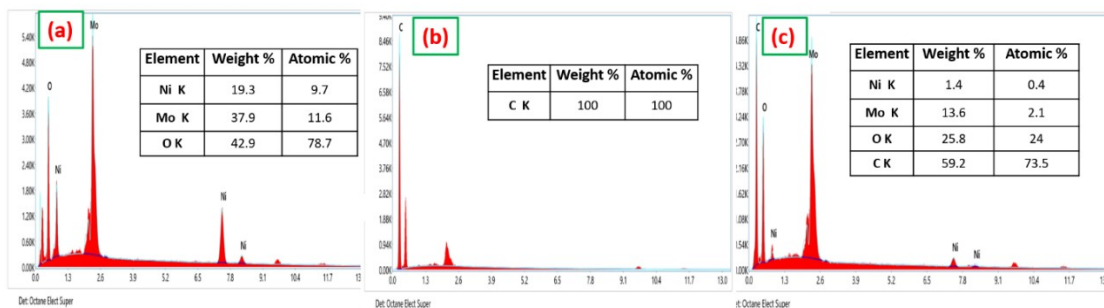


Fig. S1 EDAX analysis of elements present in the samples: (a) NiMoO_4 , (b) CNS and (c) NiMoO_4/C (D50).

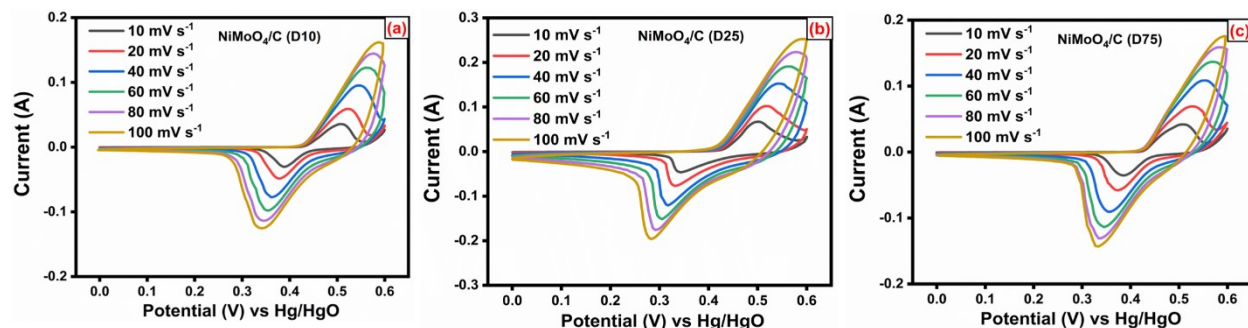


Fig. S2 CV curves of (a) NiMoO_4/C (D10), (b) NiMoO_4/C (D25) and (c) NiMoO_4/C (D75) at different scan rates (10–100 mV s^{-1}).

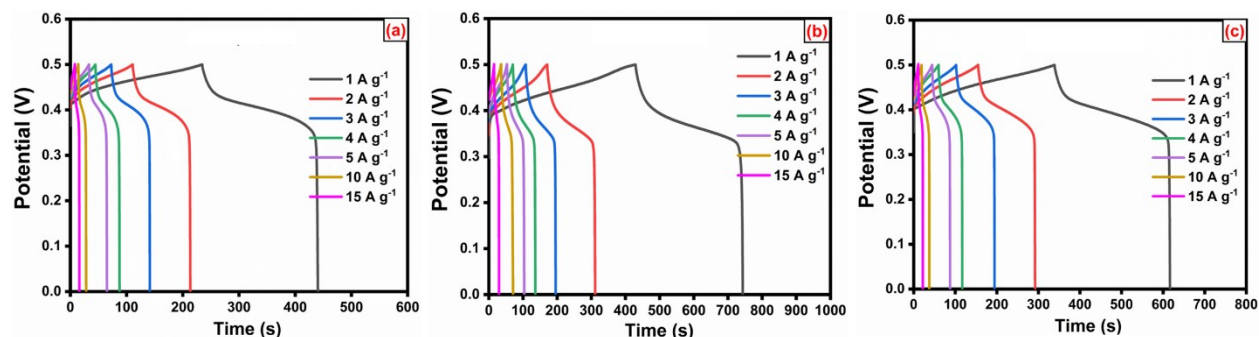


Fig. S3 GCD curves at different current densities (1–15 A g^{-1}): (a) NiMoO_4/C (D10), (b) NiMoO_4/C (D25) and (c) NiMoO_4/C (D75).

Table S1. Three electrode specific capacitance of NiMoO₄, CNS and NiMoO₄/C based nanocomposites at current density of 1 A g⁻¹ using 3 M KOH as electrolyte.

S. No.	Electrode Materials	Specific capacitance (F g ⁻¹)
1	NiMoO ₄	520
2	CNS	75
3	NiMoO ₄ /C (D10)	436
4	NiMoO ₄ /C (D25)	583
5	NiMoO₄/C (D50)	940
6	NiMoO ₄ /C (D75)	508

Table: S2 Electrical parameters of NiMoO₄, CNS, and NiMoO₄/C (D50) electrode materials estimated using ZSimpWin circuit fitting software for EIS experimental data.

Electrode materials	R _s (Ω.cm ²)	R _{pore} (mΩ.cm ²)	R _{CT} (Ω.cm ²)	Q _c (S.cm ⁻² . s ⁿ)	n	Q _{dl} (S.cm ⁻² . s ⁿ)	n	W (S.cm ⁻² . s ^{0.5})	χ ²
NiMoO ₄	1.15	0.95	13.02	0.0032	0.78	0.0229	0.32	4.091×10 ⁻¹²	0.0032
CNS	1.07	57.7	5.58×10 ⁻⁵	0.0012	0.70	0.0005	0.90	1.034×10 ⁻⁴	0.0007
NiMoO ₄ /C (D50)	1.16	5.48	2.96	0.0024	0.82	0.0236	0.34	6.465×10 ⁻¹²	0.0033

R_s - solution resistance, R_{pore} - coating pore resistance, R_{ct} - charge transfer resistance, CPE - constant phase element, n - exponent of CPE, W -Warburg impedance and χ² - chi-square value.

Table S3 Comparison of the present work with previously reported NiMoO₄-based nanomaterials.

S. No.	Electrode Materials	Method	Specific capacitance (F g ⁻¹)	Cyclic stability (capacity retention % at no. of cycles)	Ref.
1	Mn-doped NiMoO ₄ /rGO	Solvothermal	689	96% at 200 cycles	1
2	NiMoO ₄ /rGO nanocomposite	Hydrothermal	1400	91% at 2000 cycles	2
3	Hierarchical carbon sphere @NiMoO ₄	Two-step hydrothermal	268	88% at 2000 cycles	3
4	NiMoO ₄ /C Composite	Hydrothermal	325	56% at 10000 cycles	4
5	NiMoO ₄ /carbon composites	Solvothermal	805	92% at 6000 cycles	5
6	g-C ₃ N ₄ /NiMoO ₄	Chemical precipitation	398	70% at 4000 cycles	6
7	NiMoO ₄ /reduced graphene oxide composite	Hydrothermal method	680	68% at 4000 cycles	7
8	NiMoO ₄ nanorods and hierarchical nanospheres	Facile hydrothermal	974	91% at 5000 cycles	8
9	Carbon nanofibers/NiMoO ₄ nanoparticles	Electrospinning	1438	92% at 3000 cycles	9
10	Ni-doped MnMoO ₄	One-step hydrothermal method	2315	96 % at 5000 cycles	10
11	NiMoO ₄ /rGO nanocomposite	Hydrothermal method	1516	96% at 10000 cycles	11
12	NiCo ₂ S ₄ /NiMoO ₄	Hydrothermal method	2323	90% at 10000 cycles	12
13	NiMoO ₄ @CoMoO ₄	Hydrothermal method	1282	76% at 1000 cycles	13
14	NiMoO₄/C (D50)	<i>In situ</i> Hydrothermal carbonization	940	71% at 5000 cycles	<i>This report</i>

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