Supplementary Information

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Year	DOI	Remarks	Termination	Material	Capacitance(F/g)	Specific Capacitance(mAh/g)	Volumetric Specific Capacitance(F/cm^3)	Specific surface area (m2/g)	Percentages of Terminations	Electrolyte ChemicalFormula	Cation ion mobility(cm^2/Vs) x10^4	Anion ion mobility (cm^2/Vs) x10^4	Electrolyte Concentration(M)	Scan Rate (mV/s)	Current Density (A/g)	Cell Configuration(three/two electrode system)	Potential Window (V)	Sheet thickness (mm)	Density of sheet (g/cm^3)	Intercalation(charge per gram)	wt% Ti3C2 in electrode		
2015	https://doi.org/10.1007/s40145-015-0143-3			Ti3C2 exfoliated by HF in KOH	71.2		119.8	22.35		KOH	7.61	20.5	3	1	2.5	2		0.15					
					60.4			22.35		KOH	7.61	20.5	3	1	5	2	0.9	0.15					
2018	https://doi.org/10.1016/j.electacta.2018.09.148		-F, -OH/-O	Ti3C2/Ni2CO3(OH)2	695.2	173.8				KOH	7.61	20.5	6	20	1	4	0.6						
			-F	Ti3C2	35.2	8.8				KOH	7.61	20.5	6	20	1	3	0.6						
2014	https://doi.org/10.1016/j.elecom.2014.07.026			Exfoliated by HF (ex-Ti3C2)	317.4	264.5				1 M LiPF6 in a mixture of ethylene carbonate (EC)/dimethyl carbonate (DMC) /ethylmethyl carbonate (EMC) in a 1: 1:1 volume ratio.							3						
				intercalated with DMSO after exfoliation (in-Ti3C2)		123.6																	
2014	https://doi.org/10.1016/j.elecom.2014.09.002		-OH/-O/-F	Ti3C2-intercalated with KOH	350			98		H2SO4	36.2	8.27	1	1	5	3	0.6		1.37				
				Ti3C2-intercalated with Potassium acetate	230					H2SO4	36.2	8.27	1	1		3	0.6		1.51				
				Ti3C2Tx	88					H2SO4	36.2	8.27	1	1		3	0.6		0.79				
2018	https://doi.org/10.1016/j.ceramint.2018.06.086			Ti3C2/MnO2	254			88.6		KOH	7.61	20.5	6	50	0.5		0.5						
2019	https://doi.org/10.1016/j.ceramint.2019.01.016			Ti3C2 Ti3C2/polypyrrole composites	47					KOH	7.61	20.5	6	50	0.5		0.5						
				Ti3C2	133.91					Na2SO4	5.2	8.27	1	2		3	0.55						
				Ti3C2/PPy-1	156.64					Na2SO4	5.2	8.27	1	2		3	0.55				1.0234		
				Ti3C2/PPy-2	184.36					Na2SO4	5.2	8.27	1	2		3	0.55				0.5117		
				Ti3C2/PPy-3	159.57					Na2SO4	5.2	8.27	1	2		3	0.55				0.2559		
				Ti3C2/PPy-4	135.69					Na2SO4	5.2	8.27	1	2		3	0.55				0.1279		
				Ti3C2/PPy-5	132.87					Na2SO4	5.2	8.27	1	2		3	0.55				0.0639		
				Ti3C2	112.02					Na2SO4	5.2	8.27	1	5		3	0.55						
				Ti3C2/PPy-1	132.32					Na2SO4	5.2	8.27	1	5		3	0.55				1.0234		
				Ti3C2/PPy-2	149.1					Na2SO4	5.2	8.27	1	5		3	0.55				0.5117		
				Ti3C2/PPy-3	128.63					Na2SO4	5.2	8.27	1	5		3	0.55				0.2559		
				Ti3C2/PPy-4	116.87					Na2SO4	5.2	8.27	1	5		3	0.55				0.1279		
				Ti3C2/PPy-5	108.15					Na2SO4	5.2	8.27	1	5		3	0.55				0.0639		
				Ti3C2	105.43					Na2SO4	5.2	8.27	1	10		3	0.55						
				Ti3C2/PPy-1	121.77					Na2SO4	5.2	8.27	1	10		3	0.55				1.0234		
				Ti3C2/PPy-2	138.79					Na2SO4	5.2	8.27	1	10		3	0.55				0.5117		
				Ti3C2/PPy-3	120.11					Na2SO4	5.2	8.27	1	10		3	0.55				0.2559		
				Ti3C2/PPy-4	108.25					Na2SO4	5.2	8.27	1	10		3	0.55				0.1279		
				Ti3C2/PPy-5	98.79					Na2SO4	5.2	8.27	1	10		3	0.55				0.0639		
				Ti3C2	99.48					Na2SO4	5.2	8.27	1	20		3	0.55						
				Ti3C2/PPy-1	112.78					Na2SO4	5.2	8.27	1	20		3	0.55				1.0234		
				Ti3C2/PPy-2	129.75					Na2SO4	5.2	8.27	1	20		3	0.55				0.5117		
				Ti3C2/PPy-3	112.96					Na2SO4	5.2	8.27	1	20		3	0.55				0.2559		
				Ti3C2/PPy-4	100.19					Na2SO4	5.2	8.27	1	20		3	0.55				0.1279		
				Ti3C2/PPy-5	87.57					Na2SO4	5.2	8.27	1	20		3	0.55				0.0639		
				Ti3C2	90.21					Na2SO4	5.2	8.27	1	50		3	0.55						
				Ti3C2/PPy-1	98.61					Na2SO4	5.2	8.27	1	50		3	0.55				1.0234		
				Ti3C2/PPy-2	117.97					Na2SO4	5.2	8.27	1	50		3	0.55				0.5117		
				Ti3C2/PPy-3	102.28					Na2SO4	5.2	8.27	1	50		3	0.55				0.2559		
				Ti3C2/PPy-4	88.39					Na2SO4	5.2	8.27	1	50		3	0.55				0.1279		
				Ti3C2/PPy-5	74.86					Na2SO4	5.2	8.27	1	50		3	0.55				0.0639		
				Ti3C2	80.08					Na2SO4	5.2	8.27	1	100		3	0.55						
				Ti3C2/PPy-1	83.18					Na2SO4	5.2	8.27	1	100		3	0.55				1.0234		
				Ti3C2/PPy-2	104.51					Na2SO4	5.2	8.27	1	100		3	0.55				0.5117		
				Ti3C2/PPy-3	90.73					Na2SO4	5.2	8.27	1	100		3	0.55				0.2559		
				Ti3C2/PPy-4	82.22					Na2SO4	5.2	8.27	1	100		3	0.55				0.1279		
				Ti3C2/PPy-5	62.13					Na2SO4	5.2	8.27	1	100		3	0.55				0.0639		
2021	https://doi.org/10.1016/j.apsusc.2020.148272			N-Ti3C2/NiCo2S4	1849					KOH	7.61	20.5	3	2		3	0.5						
				N-Ti3C2/NiCo2S4	1697					KOH	7.61	20.5	3	5		3	0.5						
				N-Ti3C2/NiCo2S4	1611					KOH	7.61	20.5	3	10		3	0.5						
				N-Ti3C2/NiCo2S4	1418					KOH	7.61	20.5	3	20		3	0.5						
				N-Ti3C2/NiCo2S4	1045					KOH	7.61	20.5	3	40		3	0.5						
				N-Ti3C2/NiCo2S4	254					KOH	7.61	20.5	3		0.5	3	0.5						
				N-Ti3C2/NiCo2S4	282.5					KOH	7.61	20.5	3		1	3	0.5						

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2022	https://doi.org/10.1007/s13391-022-00337-9	Review paper. Citing the next 2 papers																					
2020	https://doi.org/10.1039/D0TA00687D			Ti3C2Tx flakes layered with organic photovoltaics			450			Solid organic ionogel					1								
2018	https://doi.org/10.1021/acsnano.8b02908			Ti3C2Tx (5 µm thick) /elastomer composites stretched to:	502					Solid organic ionogel				2									
				0%																			
				50%	118		395																
				80%	117		390																
					108		362																
2022	https://doi.org/10.1002/smm2.1130	Review paper:																					
2021	https://doi.org/10.1016/j.apsusc.2021.149710	Ti3C2Tx MXene/carbon nanofiber by electrospinning.		Small-sized mxene sheets (SMX/C)	120					H2SO4	36.2	8.27	1	5		3	0.7	0.00075					
		To investigate the effects of MXene sheet size the MXene/CNF hybrid, the freestanding hybrid cut and assembled into a symmetric supercapacitor.		Medium-sized mxene sheets (MMX/C)	113					H2SO4	36.2	8.27	1	5		3	0.7	0.0025					
				Large-sized mxene sheets (LMX/C)	106					H2SO4	36.2	8.27	1	5		3	0.7	0.007					
				Small-sized mxene sheets (SMX/C)	90					H2SO4	36.2	8.27	1	300		3	0.7	0.00075					
				Medium-sized mxene sheets (MMX/C)	69					H2SO4	36.2	8.27	1	300		3	0.7	0.0025					
				Large-sized mxene sheets (LMX/C)	51					H2SO4	36.2	8.27	1	300		3	0.7	0.007					
				Small-sized mxene sheets (SMX/C)	76					H2SO4	36.2	8.27	1	100		3	0.7	0.00075					
2020	https://doi.org/10.1016/j.nanoen.2020.104971			flexible and asymmetric pseudocapacitor using a Ti3C2Tx/MXene film with wavy architecture	340		1293			H2SO4	36.2	8.27	3	10		2	0.85	0.006	3.8				
					299		1136			H2SO4	36.2	8.27	3	1000		2	0.85	0.006	3.8				
					208		790			H2SO4	36.2	8.27	3	5000		2	0.85	0.006	3.8				
2017	https://doi.org/10.1038/nenergy.2017.105			Ti3 C2 O0.84(6) (OH)0.06(2) F0.25(8)	450					H2SO4	36.2	8.27	3	100		2	1	0.00009					
					380					H2SO4	36.2	8.27	3	2		2	1	0.003					
					370					H2SO4	36.2	8.27	3	2		2	1	0.013					
				templating macroporous MXene flakes with polymethyl methacrylate (PMMA) microspheres	310					H2SO4	36.2	8.27	3	10			1	0.013					
					210					H2SO4	36.2	8.27	3	10000			1	0.013					
					100					H2SO4	36.2	8.27	3	40000			1	0.013					
2019	https://doi.org/10.1038/s41560-019-0339-9	Ti3C2 electrodes tested in three 1M LiTFSI organic electrolytes with different solvents		DMSO	130					lithium bis (trifluoromethyl sulfonyl) amine (LiTFSI)			1	2		3	2.1		0.46				
				acetonitrile (ACN)	110					lithium bis (trifluoromethyl sulfonyl) amine (LiTFSI)			1	2		3	2.2		0.37				
				propylene carbonate (PC)	195					lithium bis (trifluoromethyl sulfonyl) amine (LiTFSI)			1	2		3	2.4		0.93				
2014	https://doi.org/10.1038/nature13970			Ti3C2Tx Clay	245					H2SO4	36.2	8.27	1	2			0.6	0.005					
2020	https://doi.org/10.1016/j.nanoen.2020.104734																						
	min				35.2	8.8	119.8	22.35			5.2	8.27	1	1	0.5	2	0.5	0.00009	0.79	0.37			
	max				1849	264.5	1293	98			36.2	20.5	6	40000	5	4	3	0.15	3.8	0.93		min	max
	std				363.6073579	106.6090482	413.5561518	41.14236462			13.90473538	4.817404844	1.345019213	4896.335716	1.782632261	0.3564261874	0.4477507472	0.04566600335	1.431801895	0.300721355		0.04566600335	4896.335716
	avg				249.1307792	142.675	616.975	57.825			14.63956522	10.5742029	1.777777778	823.8450704	1.8	2.893939394	0.7215277778	0.02101789474	2.511666667	0.5866666667		0.02101789474	823.8450704