

## Supporting information

### **Oxygen Vacancy-Enriched NiO Nanozymes Achieved by Facile Annealing in Argon for Detection of L-Cys**

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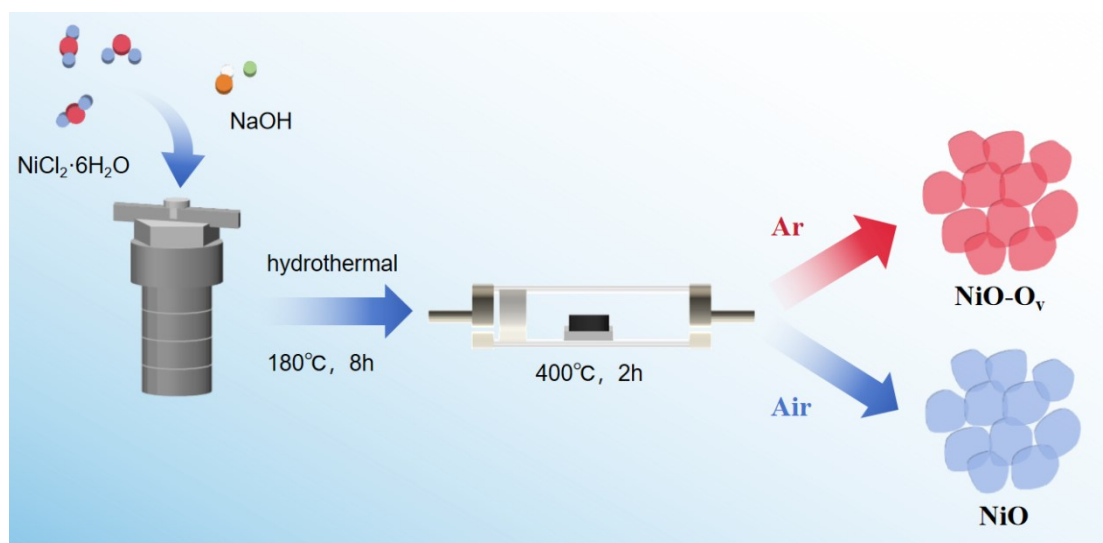
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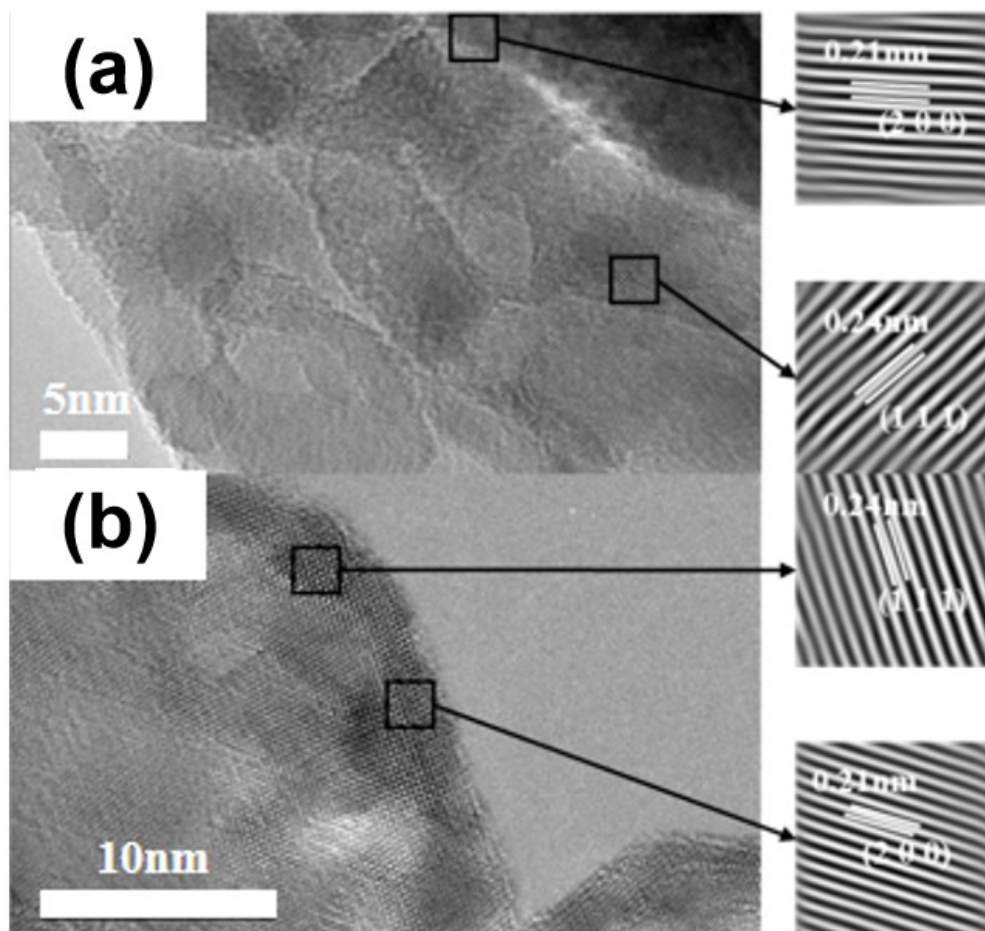
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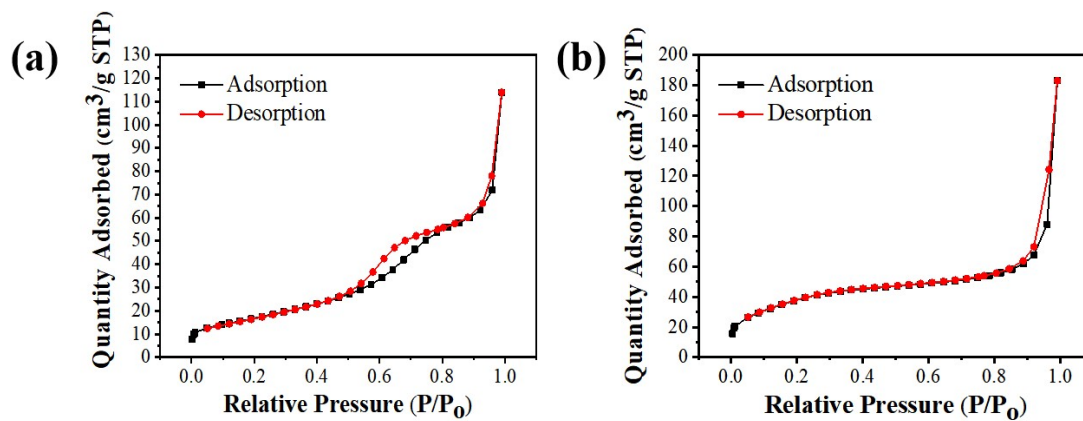
## 1. Supplementary Figures



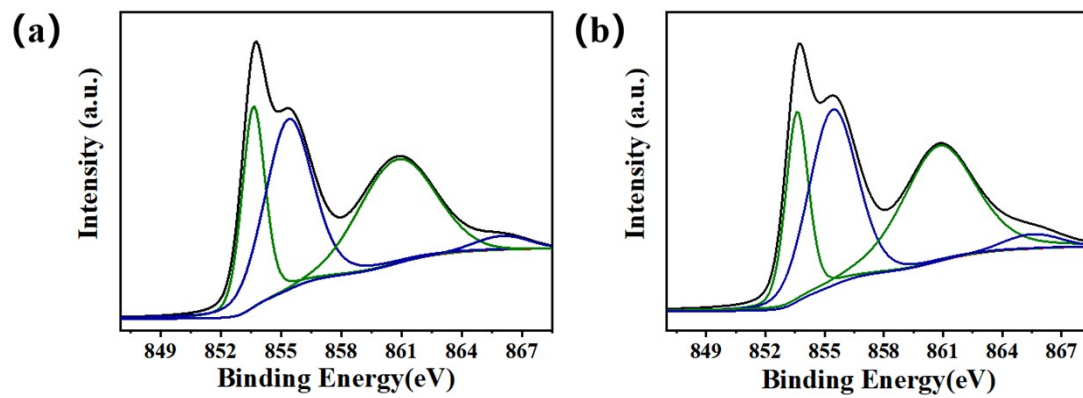
**Figure S1.** The synthetic route of NiO and NiO-O<sub>v</sub> samples.



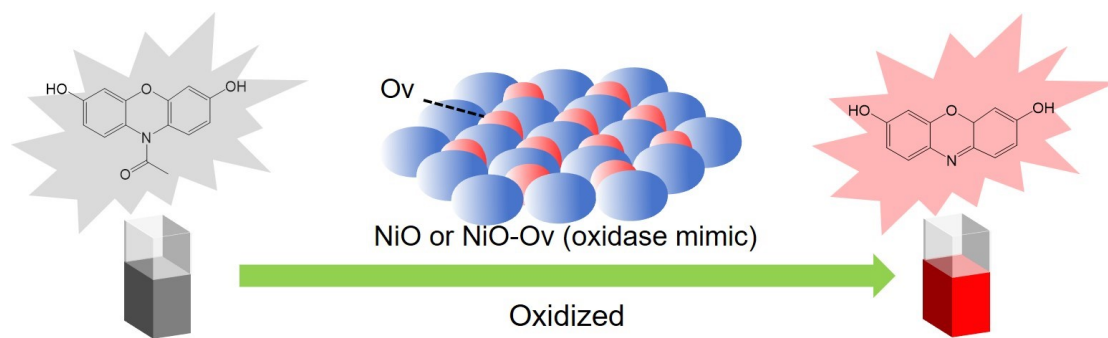
**Figure S2.** High-resolution TEM of (a) NiO; (b) NiO-Ov (The black boxes correspond to the Fourier-transformed lattice stripes)



**Figure S3.** BET test curve comparison: (a) NiO; (b) NiO-O<sub>v</sub>.



**Figure S4.** XPS spectra of different samples of Ni<sub>2p</sub>: (a) NiO; (b) NiO-O<sub>v</sub>.



**Figure S5.** Schematic diagram of Fluorescent turn-on process for catalysis of NiO or NiO-Ov nanozyme on AR to resorufin.

## 2. Supplementary Tables

**Table S1.** Structural parameters of NiO samples calcined in different atmospheres

<b>Sample</b>	<b>2<math>\theta</math> (°)</b>	<b>FWHM</b>	<b>Grain size<sup>a</sup> (nm)</b>	<b>BET<sup>b</sup> (m<sup>2</sup>/g)</b>	<b>Pore size (nm)</b>
NiO	43.240	0.641	2.97	62.01	9.39
NiO-O <sub>v</sub>	43.220	0.820	2.32	136.55	12.37

<sup>a</sup> calculated by the method of Scherrer's equation based on the diffraction of the (200) peak of NiO at  $2\theta = 43.1^\circ$ .

<sup>b</sup> calculated by the method of specific surface area.

**Table S2.** The summary of XPS results of different NiO samples.

Sample	Binding energy (eV)					Atomic ratio(%)		EPR area
	Ni 2p <sub>3/2</sub>		O 1s			Ni <sup>3+</sup> / Ni <sup>2+</sup>	O <sub>β</sub> / (O <sub>α</sub> + O <sub>β</sub> + O <sub>γ</sub> )	
	Ni <sup>2+</sup>	Ni <sup>3+</sup>	O <sub>α</sub>	O <sub>β</sub>	O <sub>γ</sub>			
NiO	853.6	855.43	529.15	531.18	532.14	1.72	21.4	8.1×10 <sup>4</sup>
NiO-O <sub>v</sub>	853.63	855.45	529.18	531.08	532.91	1.85	32.8	1.1×10 <sup>5</sup>



**Table S3.** Comparison of kinetic parameters ( $K_m$  and  $V_{max}$ ) corresponding to different nanozymes

<b>Substrate</b>	<b>Nanozymes</b>	<b><math>K_m</math> (<math>\mu\text{M}</math>)</b>	<b><math>V_{max}</math> (<math>\text{nM s}^{-1}</math>)</b>	<b>Ref.</b>
Amplex Red	Bi–Au NPs	89.3	15.0	1
	ZiF-67	5.28	28.2	2
	FeP@C	2.30	—	3
	Au/AgCl	17	4.6	4
	MFNP1:1	34.2	244	5
	CoOxH-GO	4.87	0.839	6
	NiO- $\text{O}_V$	2.83	26.7	this work

**Table S4. Detection range and detection limit of L-Cys by different systems**

<b>System</b>	<b>Method</b>	<b>Linear range (<math>\mu\text{M}</math>)</b>	<b>Detection limit (nM)</b>	<b>Ref.</b>
$\text{Co}_4\text{S}_3$	fluorescence	0.25-2.5	75	7
AuNRs/Au-Ag NCs	fluorescence	5-100	$1.73 \times 10^3$	8
AuNCs-AuNPs	fluorescence	1.5-35	$1.4 \times 10^3$	9
Si-CDs	fluorescence	20-100	410	10
CDs	chemiluminescence	10-100	$8.8 \times 10^3$	11
OV- $\text{Mn}_3\text{O}_4$	colorimetric	5-800	$1.31 \times 10^3$	12
$\text{Gd}(\text{OH})_3$	colorimetric	0.2-75	$2.6 \times 10^3$	13
2D $\text{Co}_3\text{S}_4$	colorimetric	0.2-100	$2.7 \times 10^3$	14
$\text{VS}_4$	colorimetric	5-100	$2.5 \times 10^3$	15
$[\text{Ag}_2(\text{bit})_2]_2[\text{Mo}_8\text{O}_{26}]$	colorimetric	1-100	220	16
rGO-GP	colorimetric	2-30	100	17
$\text{MnO}_2$ nanobelts	colorimetric	0-35	100	18
$\text{MnO}_2@\text{Co}_3\text{O}_4$	colorimetric	1.25-25	$1.1 \times 10^3$	19
$\text{NiO-O}_\text{V}$	fluorescence	0.05-2	37.8	this work

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