

Supplementary Information for

**Metal-Organic Frameworks as Advanced Platforms for Radionuclide Detection†**

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## Abbreviations

- 4-Mpy: 4-Mercaptopyridine  
4,4'-Azbpy: 4,4'-Azobipyridine  
AG: Alginate gel  
Bce: 1,2-Bis(4-carboxyphenoxy)-ethane  
BDAT: 4,4',4'',4'''-([1,1'-Biphenyl]-4,4'-diylbis(azanetriyl))tetrabenzoic acid  
Bdc: Dicarboxylate  
BDC-OH: 2-Hydroxyterephthalic acid  
Bipy: 4,4'-Bipyridine  
BITD: 5'-(4,5-bis(4-carboxyphenyl)-1*H*-imidazol-2-yl)[1,1':3',1"-terphenyl]-4,4"-dicarboxylic acid  
Bpdc: Biphenyl-4,4'-dicarboxylate  
Bpe: 1,2-Bis(4-pyridyl)ethylene  
Bppt: 4-Amino-3,5-bis(4-pyridyl-3-phenyl)-1,2,4-triazole  
Bpy: 2,2'-Bipyridine  
Bpydc: 2,2'-Bipyridine-5,5'-dicarboxylic acid  
BTC: 1,3,5-benzene tricarboxylic acid  
CA: Competitive absorption  
Cada: 9-(4-Carboxy-phenyl)-9*H*-carbazoly-3,6-dicarboxylic acid  
CH<sub>3</sub>CN: Acetonitrile  
COFs: covalent-organic frameworks  
Cpia: 5-(4-Carboxyphenoxy)isophthalic acid  
CT: Charge transfer  
Cyh: Cyclohexane  
DATP: Diaminoterephthalic acid  
DATRz: 3,5-Diamino-1,2,4-triazole  
DBIIA: 5-(1,3-Dioxo-1*H*-benzo[de]isoquinolin-2(3*H*)-yl)isophthalic acid  
DBT: 4,4'-Di(4*H*-1,2,4-triazol-4-yl)-1,1'-biphenyl  
Dcbpy: 4,4'-Dicarboxylate-2,2'-dipyridine anion  
DHTA: 2,5-Dihydroxybenzene-1,4-dicarboxylic acid  
DL-lac: Lactate anion  
DMA: N,N-dimethylacetamide  
DMF: N,N-dimethylformamide  
Dmimpym: 4,6-Di(2-methyl-imidazol-1-yl)-pyrimidine  
DMSO: Dimethyl sulfoxide  
DOG: Dissolved off-gas  
DPA: Dicarboxy-9,10-diphenylanthracene  
DPV: differential pulse voltammetry  
DR: Dissolution-recrystallization

Ebic: 2-Ethyl-1*H*-benzo[d]imidazole-5-carboxylic acid  
EIS: Electrochemical impedance spectroscopy  
ET: Energy transfer  
FDC: 2,5-Furandicarboxylic acid  
FRET: Förster Resonance Energy Transfer  
GCE: Glassy carbon electrodes  
Hcpcp: Hexakis(4-carboxylatophenoxy)cyclotriphosphazene  
Hipamifba: 4-(((4-((carboxymethyl)carbamoyl)phenyl)amino)methyl)benzoic acid  
HOFs: Hydrogen-bonded organic frameworks  
ICP-MS: Inductively coupled plasma mass spectrometry  
IDE: Interdigitated comblike electrode  
Imdc: Imidazole-4,5-dicarboxylic acid  
Ina: Isonicotinate  
IX: Ion exchange  
 $K_{SV}$ : Quenching constant  
LOD: Limit of detection  
LSV: linear sweep voltammetry  
LTP: Low-temperature NH<sub>3</sub> plasma  
MAs: Minor actinides  
MeOH: methanol  
MOFs: Metal-organic frameworks  
Mpy: Mercaptopyridine  
MTBC: [1,1'-biphenyl]-4-carboxylic acid  
Nda: 1,4-Naphthalenedicarboxylic acid  
NH<sub>2</sub>-BDC: 2-Aminobenzene-1,4-dicarboxylic acid  
NI-bpy-44: N-(pyridin-4-yl)-4-(pyridin-4-yl)-1,8-naphthalimide  
NPYC: N-pyridin-4-ylpyridine-4-carboxamide  
P4b: Tetra(pyridin-4-yl)benzene  
PABA: P-aminobenzoic acid  
PAF: Porous aromatic framework  
PATP: 2-((Pyridin-1 ium-2-ylmethyl)ammonio)terephthalate)  
PB: Prussian blue  
PDA: 1,4-Phenylenediacetate  
PET: Photoinduced electron transfer  
Pia: 5-(Pyridin-4-yl)isophthalic acid  
PMA: Pyromellitic acid  
Ppb: Parts-per-billion  
PPh<sub>3</sub>: Triphenylphosphine  
PPPy: Polyphosphonated pyridyl

Ppy: 2-Phenylpyridine  
PQTB: 4,4',4'',4'''-(Pyrazino[2,3-g]quinoxaline-2,3,7,8-tetrayl)tetrabenzoic  
PVTP: 2-Vinyl terephthalic acid  
Pybz: 4-Pyridylbenzoate  
REEs: Rare earth elements  
RGB: Red green blue  
Sal: Salicylaldehyde  
SC: Structure collapse  
SC-SC: Single-crystal-to-single-crystal  
SERS: Surface-enhanced Raman scattering  
SNF: Spent nuclear fuel  
SQ: Static quenching  
TATAB: 2,4,6-Tri[(pcarboxyphenyl) amino]-1,3,5-triazine  
TATIB: 2,2',2''-([1,3,5]-Triazine-2,4,6-triimino)tribenzoic acid  
TBT: 1,3,5-Benzenetrisbenzoate  
TCBPA: Tris(4'-carboxybiphenyl)amine  
Tcbpp: 2,4,6-Tris(1-(4-carboxylatobenzyl)pyridinium-4-yl)pyridine  
TCPE: Tetrakis(4-carboxyphenyl)ethylene acid  
TCPP: Tetras-(4-carboxyphenyl) porphyrin  
TDA: Thiodiglycolic acid  
TDPAT: 5,5',5''-(1,3,5-Triazine 2,4,6-triyltriimino)tris[1,3-benzenedicarboxylic acid]  
TIBTC: 2,4,6-Triiodo-1,3,5-benzene tricarboxylic acid  
Tipa: Tris (4-(1*H*-imidazole-1-yl) phenyl) amine  
TOCNF: TEMPO-oxidized cellulose nanofibers  
TP: 2',5'-dimethyl-[1,1':4',1''-terphenyl]-4,4''-dicarboxylate  
TPE: 1,1,2,2-Tetrakis(4-carboxy) ethene  
TPPYE: Tetrakis(4-pyrimidylphenyl)ethene  
TTHA: 1,3,5-Triazine-2,4,6-triamine hexaacetic acid  
Tz: 1,2,4-Triazolate  
U.S. EPA: United States Environmental Protection Agency  
WHO: World Health Organization  
ZIF: Zeolitic imidazolate framework

## Supplementary Tables

**Table S1.** Examples of MOFs for the detection of I<sub>2</sub>.

No.	MOFs	Radionuclide	Detection method	LOD	Solvent	Mechanism	Refs.
1	{[Zn <sub>3</sub> (DL-lac) <sub>2</sub> (pybz) <sub>2</sub> ] <sub>n</sub> ·2.5DMF}	I <sub>2</sub>	Direct measurement	/	Cyh	SC-SC and CT	S1
2	[Cu <sub>6</sub> (pybz) <sub>8</sub> (OH) <sub>2</sub> ]·I <sub>5</sub> <sup>-</sup> ·I <sub>7</sub> <sup>-</sup>	I <sub>2</sub>	Direct measurement	/	Cyh	SC-SC and CT	S2
3	{[Zn(ebic) <sub>2</sub> ]} <sub>n</sub>	I <sub>2</sub>	Direct measurement	/	Cyh	CT	S3
4	ZIF-70@IDE	I <sub>2</sub>	EIS	467.6 μg/kg	Vapor	CT	S4
5	ZIF-8-coated IDEs	I <sub>2</sub>	EIS	/	Vapor	CT	S5
6	MFM-300(X) (X = Al, Fe, In, or Sc)	I <sub>2</sub>	EIS	/	Vapor	CT	S6
7	Ln-BTC (Ln = Eu, Tb, Dy, Er, or Yb)	I <sub>2</sub>	EIS	/	Vapor	CT	S7
8	Tb(Cu <sub>4</sub> I <sub>4</sub> )(ina) <sub>3</sub> (DMF)	I <sub>2</sub>	LSV	0.01 g/L	Vapor	CT	S8
9	Ln-BTC (Ln = Eu or Tb)	I <sub>2</sub>	Turn off	/	Vapor	PET	S7
10	Cd <sub>2</sub> (BDC) <sub>2</sub> (NI-bpy-44) <sub>2</sub> (1)						
11	Cd <sub>2</sub> (BDC-Br) <sub>2</sub> (NI-bpy-44) <sub>2</sub> (2)	I <sub>2</sub>	Turn off	/	Vapor	CT	S9
12	Cd <sub>2</sub> (BDC-NO <sub>2</sub> ) <sub>2</sub> (NI-bpy-44) <sub>2</sub> (3)						
13	Cd <sub>2</sub> (bpdc) <sub>2</sub> (NI-bpy-44) <sub>2</sub> (4)						
14	Cd(bppt) <sub>2</sub> (ClO <sub>4</sub> ) <sub>2</sub> ·H <sub>2</sub> O	I <sub>2</sub>	Turn off	/	EtOH	Interaction	S10
15	[Pb <sub>6</sub> (μ <sub>4</sub> -O) <sub>2</sub> (pia) <sub>4</sub> ] <sub>4</sub> ·4DMF	I <sub>2</sub>	Turn off	/	Vapor/Water	Interaction and ET	S11
16	Ag(4-Mpy) <sub>2</sub> (PPh <sub>3</sub> ) <sub>2</sub> (MSOF-1)	I <sub>2</sub>	Turn off	/	Vapor	CT	S12
17	MIL-53(Al)-TDC	I <sub>2</sub>	Fluorochromic	2 wt%	Vapor	ET	S13
18	ZIF-90	I <sub>2</sub>	Colorimetric	/	Vapor	CT	S14

**Table S2.** Examples of MOFs for the detection of  $\text{UO}_2^{2+}$ .

No.	MOFs	Radion uclide	Detection method	$K_{\text{SV}} (\text{M}^{-1})$	LOD (g/L)	LOD (M)	Solvent	Mechanism	Refs.
1	$[\text{Tb}_4(\text{C}_{29}\text{O}_8\text{H}_{17})_2(\text{NO}_3)_4(\text{DMF})_4(\text{H}_2\text{O})_4] \cdot 4\text{H}_2\text{O} \cdot 8.5\text{DMF}$ (YTU-100)	$\text{UO}_2^{2+}$	Turn off	$8.10 \times 10^4$	$7.50 \times 10^{-7}$	$2.78 \times 10^{-9}$	Water	FRET	S15
2	$[\text{Tb}(\text{BPDC})_2 \cdot (\text{CH}_3)_2\text{NH}_2$ (DUT-101)]	$\text{UO}_2^{2+}$	Turn off	$1.03 \times 10^4$	$8.34 \times 10^{-6}$	$3.10 \times 10^{-8}$	Water	FRET and CT	S16
3	$[\text{EuNa}_{0.5}(\text{bce})_2] \cdot \text{N}_{a_{0.5}} \cdot \text{DMF} \cdot 3.5\text{H}_2\text{O}$	$\text{UO}_2^{2+}$	Turn off	$3.66 \times 10^4$	$1.34 \times 10^{-5}$	$4.97 \times 10^{-8}$	Water	CA and ET	S17
4	$[\text{Eu}_2(\text{NH}_2\text{-BDC})_{2.5}(\text{CH}_3\text{COO})(\text{DMA})(\text{H}_2\text{O})] \cdot \text{DMA}@{\text{Sal}}$	$\text{UO}_2^{2+}$	Turn off	$6.70 \times 10^2$	$3.51 \times 10^{-3}$	$1.30 \times 10^{-5}$	Water	CA	S18
5	$[\text{Co}_2(\text{dmimpym})(\text{n-da})_2]_n$	$\text{UO}_2^{2+}$	Turn off	$1.10 \times 10^4$	$3.56 \times 10^{-3}$	$1.32 \times 10^{-5}$	DMF	CT	S19
6	$[\text{Eu}_2(\text{MTBC})(\text{OH})_2(\text{DMF})_3(\text{H}_2\text{O})_4] \cdot 2\text{DMF} \cdot 7\text{H}_2\text{O}$	$\text{UO}_2^{2+}$	Turn off	$3.63 \times 10^3$	$3.09 \times 10^{-4}$	$1.14 \times 10^{-6}$	DMF/ ${}^2\text{O}$ (v/v=1/1)	ET and CA	S20
7	$[\text{Eu}_2(\text{TATAB})_2] \cdot 4\text{H}_2\text{O} \cdot 6\text{DMF}$	$\text{UO}_2^{2+}$	Turn off	$8.40 \times 10^4$	$2.43 \times 10^{-4}$	$9.00 \times 10^{-7}$	DMF	CT and ET	S21
8	$[\text{Eu}(\text{TIBTC})(\text{DMF})_3]_n$	$\text{UO}_2^{2+}$	Turn off	$4.00 \times 10^3$	$5.54 \times 10^{-3}$	$1.10 \times 10^{-5}$	Water	CT and ET	S22
9	TOCNF@Eu-MOF	$\text{UO}_2^{2+}$	Turn off	$8.21 \times 10^4$	$3.56 \times 10^{-4}$	$1.32 \times 10^{-6}$	Water	Interaction and ET	S23
10	$[\text{In}_2(\text{OH})_2(\text{H}_2\text{TTHA})(\text{H}_2\text{O})_2]_n$	$\text{UO}_2^{2+}$	Turn off	$4.80 \times 10^4$	$4.20 \times 10^{-4}$	$1.56 \times 10^{-6}$	Water	Interaction	S24
11	$\text{Zr}_6\text{O}_4(\text{OH})_4(\text{OH})_6(\text{H}_2\text{O})_6(\text{TCPE})_{1.5} \cdot (\text{H}_2\text{O})_{24}(\text{DMF})_9$ (Zr-TCPE)	$\text{UO}_2^{2+}$	Turn off	$3.78 \times 10^3$	$6.70 \times 10^{-7}$	$2.48 \times 10^{-9}$	Water	CA	S25
12	$\{[\text{Me}_2\text{NH}_2]_{0.5}[\text{Co}(\text{DATRz})_{0.5}(\text{NH}_2\text{BDC})] \cdot x\text{G}\}_n$	$\text{UO}_2^{2+}$	Turn off	$7.95 \times 10^4$	$3.09 \times 10^{-5}$	$1.30 \times 10^{-7}$	Water	IX FRET and	S26
13	UiO-66-NH <sub>2</sub>	$\text{UO}_2^{2+}$	Turn off	$9.94 \times 10^4$	$5.40 \times 10^{-5}$	$2.00 \times 10^{-7}$	Water	Interaction	S27
14	LTP@UiO-66-NH <sub>2</sub>	$\text{UO}_2^{2+}$	Turn off	$1.81 \times 10^5$	$2.16 \times 10^{-5}$	$8.00 \times 10^{-8}$	Water	and SQ	S27
15	$\text{Zn}_2(\text{PMA})(\text{NPYC})(\text{H}_2\text{O})_2 \cdot 2\text{H}_2\text{O}$ (HNU-50)	$\text{UO}_2^{2+}$	Turn off	/	$3.24 \times 10^{-6}$	$1.20 \times 10^{-8}$	Water	Interaction and CA	S28
16	Tb-MOF-76	$\text{UO}_2^{2+}$	Turn off	/	$5.40 \times 10^{-1}$	$2.00 \times 10^{-3}$	Water	Interaction and ET	S29

17	ZIF-90-PABA-Eu	$\text{UO}_2^{2+}$	Turn off	/	$1.50 \times 10^{-5}$	$5.56 \times 10^{-8}$	Water	CA and SQ	S30
18	[Tb(1,3,5-benzenetrisbenzoate)] <sub>n</sub> (Tb-TBT)	$\text{UO}_2^{2+}$	Turn off	/	$2.03 \times 10^{-7}$	$7.50 \times 10^{-10}$	Water	ET	S31
19	Tb-TATAB	$\text{UO}_2^{2+}$	Turn off	/	$9.00 \times 10^{-7}$	$3.33 \times 10^{-9}$	Water	Interaction and ET	S32
20	Tb@MOF-808-TDA	$\text{UO}_2^{2+}$	Turn off	/	$8.20 \times 10^{-5}$	$3.04 \times 10^{-7}$	Water	PET	S33
21	Tb-MOF/Tb-AG	$\text{UO}_2^{2+}$	Turn off	/	$1.20 \times 10^{-9}$	$4.44 \times 10^{-12}$	Water	IX	S34
22	Tb-TDPAT	$\text{UO}_2^{2+}$	Turn off	/	$3.38 \times 10^{-7}$	$1.25 \times 10^{-9}$	Water	ET	S35
23	$\text{Zn}_4(\text{Hcpcp})(\text{OH})_2(\text{H}_2\text{O})_4$	$\text{UO}_2^{2+}$	Turn off	$7.44 \times 10^3$	/	/	Water	CT	S36
24	$\text{Cd}_3(\text{Cada})_2(\text{bipy})(\text{H}_2\text{O})_2 \cdot \text{H}_2\text{O}$	$\text{UO}_2^{2+}$	Turn off	$2.67 \times 10^4$	/	/	Water	FRET	S37
25	$\text{C}_{40}\text{H}_{22}\text{N}_4\text{O}_8\text{Cd}$ (HUST-36)	$\text{UO}_2^{2+}$	Turn off	$2.00 \times 10^5$	/	/	Water	CT	S38
26	IHEP-24	$\text{UO}_2^{2+}$	Turn off	$1.25 \times 10^2$	/	/	Water	CT	S39
27	Eu-PPPy	$\text{UO}_2^{2+}$	Turn on	/	$3.24 \times 10^{-3}$	$1.20 \times 10^{-5}$	Water	FRET	S40
28	$[\text{UO}_2(\text{O}_2)_{0.5}(\text{DPA})][\text{H}_2\text{N}(\text{CH}_3)_2]$ (ECUT-177)	$\text{UO}_2^{2+}$	Turn on	/	$6.59 \times 10^{-7}$	$2.44 \times 10^{-9}$	Water	Interaction and IX	S41
29	C-GSH-AuNCs@UiO-66-NH <sub>2</sub>	$\text{UO}_2^{2+}$	Turn on	/	$4.13 \times 10^{-6}$	$1.53 \times 10^{-8}$	Water	CT	S42
30	Eu-DATP	$\text{UO}_2^{2+}$	Fluorochromic	/	$7.29 \times 10^{-7}$	$2.70 \times 10^{-9}$	Water	ET	S43
31	UiO-67-bpydc@Eu <sup>3+</sup>	$\text{UO}_2^{2+}$	Fluorochromic	/	$8.10 \times 10^{-7}$	$3.00 \times 10^{-9}$	Water	Interaction and ET	S44
32	ZIF-L-hemin	$\text{UO}_2^{2+}$	Colorimetric	/	$1.88 \times 10^{-5}$	$7.90 \times 10^{-8}$	Water	Interaction	S45
33	FA@ZIF-8	$\text{UO}_2^{2+}$	SERS	/	$2.70 \times 10^{-5}$	$1.00 \times 10^{-7}$	Water	Interaction	S46
34	PtRu/UiO66-NH <sub>2</sub>	$\text{UO}_2^{2+}$	LSV, and DPV	/	$5.70 \times 10^{-6}$	$2.40 \times 10^{-9}$	Water	CT	S47

**Table S3.** Examples of MOFs for the detection of Th<sup>4+</sup>.

No.	MOFs	Radion uide	Detection method	$K_{SV} (M^{-1})$	LOD (g/L)	LOD (M)	Solvent	Mechanism	Refs.
1	[Eu <sub>2</sub> (NH <sub>2</sub> -BDC) <sub>2.5</sub> (CH <sub>3</sub> COO)(DMA)(H <sub>2</sub> O)]·DMA@Sal	Th <sup>4+</sup>	Turn off	7.52×10 <sup>3</sup>	2.69×10 <sup>-4</sup>	1.16×10 <sup>-6</sup>	Water	CA	S18
2	[Eu <sub>2</sub> (FDC) <sub>3</sub> (DMA) <sub>2</sub> ]·4H <sub>2</sub> O	Th <sup>4+</sup>	Turn off	6.68×10 <sup>4</sup>	8.10×10 <sup>-3</sup>	3.49×10 <sup>-5</sup>	Water	FRET and CA	S48
3	DBT-DHTA-Cd	Th <sup>4+</sup>	Turn off	/	8.00×10 <sup>-5</sup>	1.67×10 <sup>-7</sup>	Water	PET	S49
4	[Eu <sub>2</sub> (MTBC)(OH) <sub>2</sub> (DMF) <sub>3</sub> (H <sub>2</sub> O) <sub>4</sub> ]·2DMF·7H <sub>2</sub> O (ThP-1)	Th <sup>4+</sup>	Fluorochromic	/	2.42×10 <sup>-5</sup>	1.04×10 <sup>-7</sup>	Water	Adsorption and CT	S50
5	Eu <sub>x</sub> Tb <sub>1-x</sub> -BDC-OH	Th <sup>4+</sup>	Fluorochromic	/	6.11×10 <sup>-4</sup>	2.63×10 <sup>-6</sup>	DMF	DR	S51

**Table S4.** Examples of MOFs for the detection of Cs<sup>+</sup> and Sr<sup>2+</sup>.

No.	MOFs	Radion uide	Detection method	$K_{SV} (M^{-1})$	LOD (g/L)	LOD (M)	Solvent	Mechanism	Refs.
1	Cd-PQTB (HUST-31)	Cs <sup>+</sup>	Turn on	/	2.66×10 <sup>-2</sup>	2×10 <sup>-4</sup>	Water	CT	S52
2	Zn-TPE-PB (PB-COP)	Cs <sup>+</sup>	Turn off	/	7.38×10 <sup>-5</sup>	5.5×10 <sup>-7</sup>	ethanol	SC and PET	S53
1	Al-TATIB	Sr <sup>2+</sup>	Turn on	-8.35×10 <sup>3</sup>	4.7×10 <sup>-4</sup>	5.37×10 <sup>-6</sup>	DMF	CT	S54
2	Mg(DBIIA)(DMSO) <sub>2</sub> (1)				6.08×10 <sup>-4</sup>	6.94×10 <sup>-6</sup>			
3	Mn(DBIIA)(DMSO) <sub>2</sub> (2)				2.54×10 <sup>-5</sup>	2.9×10 <sup>-7</sup>			
4	Mg <sub>0.89</sub> Mn <sub>0.11</sub> (DBIIA)(DMSO) <sub>2</sub> (3)	Sr <sup>2+</sup>	Turn on	/	1.49×10 <sup>-4</sup>	1.7×10 <sup>-6</sup>	EtOH	DR	S55
5	Mg <sub>0.75</sub> Mn <sub>0.25</sub> (BBIIA)(DMSO) <sub>2</sub> (4)				5.69×10 <sup>-5</sup>	6.5×10 <sup>-7</sup>			
6	Mg <sub>0.37</sub> Mn <sub>0.63</sub> (DBIIA)(DMSO) <sub>2</sub> (5)				3.24×10 <sup>-5</sup>	3.7×10 <sup>-7</sup>			
7	Al-TCPP	Sr <sup>2+</sup>	LSV	/	2.63×10 <sup>-9</sup>	3×10 <sup>-11</sup>	Water	Interaction	S56

**Table S5.** Examples of MOFs for the detection of  $\text{ReO}_4^-$ .

No.	MOFs	Rddion uclide	Detection method	$K_{SV} (\text{M}^{-1})$	LOD (g/L)	LOD (M)	Solvent	Mechanism	Refs.
1	$[\text{Zr}_6\text{O}_4(\text{OH})_4(\text{NH}_3^+ -\text{BDC})_6]\text{Cl}_6 \cdot x\text{H}_2\text{O}$ (MOR-1)				$3.6 \times 10^{-4}$	$1.44 \times 10^{-6}$			
2	$\text{H}_{16}[\text{Zr}_6\text{O}_{16}(\text{H}_2\text{PA TP})_4]\text{Cl}_8 \cdot x\text{H}_2\text{O}$ (MOR-2)	$\text{ReO}_4^-$	Turn off	/			Water	CT	S57
3	$\{[\text{Cu}(\text{hipamifba})(4',4''-\text{azbpy})] \cdot 2\text{CH}_3\text{OH} \cdot 2\text{H}_2\text{O}\}_n$ (1)			$3.16 \times 10^6$	$3.4 \times 10^{-3}$	$1.36 \times 10^{-5}$			
4	$\{[\text{Zn}(\text{hipamifba})(4',4''-\text{azbpy})] \cdot 2\text{CH}_3\text{OH} \cdot 2\text{H}_2\text{O}\}_n$ (2)	$\text{ReO}_4^-$	Turn off				Water	PET and SQ	S58
5	$[\text{Ag}(1,2,4,5-\text{p}4\text{b})](\text{SbF}_6)$ (TJNU-302)	$\text{ReO}_4^-$	Turn off	$3.47 \times 10^3$	$2.25 \times 10^{-2}$	$9 \times 10^{-5}$	Water	CT	S59
6	$\text{Ag}_3\text{Tipa}_2$ (NCU-2)	$\text{ReO}_4^-$	Turn off	/	$1.67 \times 10^{-5}$	$6.67 \times 10^{-8}$	Water	Adsorption, CT, and IX	S60
7	$\text{Ir}-(\text{ppy})_2(\text{bpy})^+$ - PAF (Ir-PAF)	$\text{ReO}_4^-$	Turn on	/	$5.569 \times 10^{-4}$	$2.23 \times 10^{-6}$	Water	Interaction and IX	S60
8	Ag-TPPYE (ZJU-X8)	$\text{ReO}_4^-$	Fluorochro- mic	/	$1.02 \times 10^{-2}$	$4.08 \times 10^{-5}$	Water	CT and IX	S61

**Table S6.** Examples of MOFs for the detection of I<sup>-</sup> and IO<sub>3</sub><sup>-</sup>.

No.	MOFs	Rddion uclide	Detection method	K <sub>SV</sub> (M <sup>-1</sup> )	LOD (g/L)	LOD (M)	Solvent	Mechanism	Refs.
1	Zr <sub>6</sub> O <sub>4</sub> (OH) <sub>4</sub> (OH) <sub>6</sub> (H <sub>2</sub> O) <sub>6</sub> (TCPE) <sub>1.5</sub> ·(H <sub>2</sub> O) <sub>24</sub> (DMF) <sub>9</sub> (Zr-TCPE)	I <sup>-</sup>	Turn off	2.92×10 <sup>3</sup>	8.7×10 <sup>-7</sup>	6.86×10 <sup>-9</sup>	Water	ET	S25
2	Cd <sub>2.5</sub> (PDA)(tz) <sub>3</sub>	I <sup>-</sup>	Turn off	1.8×10 <sup>4</sup>	8×10 <sup>-5</sup>	6.3×10 <sup>-7</sup>	Water	CA	S62
3	[Tb(cpia)(H <sub>2</sub> O) <sub>2</sub> ] <sub>n</sub> ·nH <sub>2</sub> O	I <sup>-</sup>	Turn off	1.23×10 <sup>4</sup>	2.9×10 <sup>-4</sup>	2.29×10 <sup>-6</sup>	Water	CA	S63
4	[Eu <sub>2</sub> Zn(dcbpy) <sub>3</sub> (H <sub>2</sub> O) <sub>4</sub> ](NO <sub>3</sub> ) <sub>2</sub> ·12H <sub>2</sub> O <sub>n</sub> (1·NO <sub>3</sub> <sup>-</sup> )	I <sup>-</sup>	Turn off	2.3×10 <sup>4</sup>	6×10 <sup>-5</sup>	4.73×10 <sup>-7</sup>	Water	CT and CA	S64
5	[Tb <sub>2</sub> Zn(dcbpy) <sub>3</sub> (H <sub>2</sub> O) <sub>4</sub> ](NO <sub>3</sub> ) <sub>2</sub> ·12H <sub>2</sub> O <sub>n</sub> (2·NO <sub>3</sub> <sup>-</sup> )			1.8×10 <sup>5</sup>	1×10 <sup>-6</sup>	7.88×10 <sup>-9</sup>			
6	Eu <sub>4</sub> (μ <sub>3</sub> -OH) <sub>4</sub> (tcbpp) <sub>2</sub> (H <sub>2</sub> O) <sub>9</sub> ·Cl <sub>2</sub>	I <sup>-</sup>	Turn off	/	/	/	Water	CT	S65
7	{[Zn <sub>4</sub> Eu <sub>2</sub> (imdc) <sub>4</sub> (SO <sub>4</sub> )(H <sub>2</sub> O) <sub>8</sub> ] <sub>n</sub> ·4H <sub>2</sub> O}	I <sup>-</sup>	Turn off	/	/	/	DMF	Interaction	S66
8	LVMOF-1-Cl	I <sup>-</sup>	Turn off / and colorimet ric	/	/	/	Water	FRET	S67
9	Co <sub>2</sub> (bpe) <sub>2.5</sub> (NO <sub>3</sub> ) <sub>4</sub> (CH <sub>3</sub> O)(Co-bpe)	I <sup>-</sup>	Turn off / and colorimet ric	6.64×10 <sup>3</sup>	3.43×10 <sup>-5</sup>	2.7×10 <sup>-7</sup>	MeOH	CT	S68
10	Tb@Cu-BTC	I <sup>-</sup>	Turn on	/	5.96×10 <sup>-4</sup>	4.7×10 <sup>-6</sup>	Water	Guest- induced emission	S69
11	[Dy(cpia)(H <sub>2</sub> O) <sub>2</sub> ] <sub>n</sub> ·nH <sub>2</sub> O	I <sup>-</sup>	Fluorochr omic	1.17×10 <sup>4</sup> (574 nm)	3.05×10 <sup>-6</sup>	2.4×10 <sup>-8</sup>	Water	CA	S70
1	[Th <sub>6</sub> (μ <sub>3</sub> -OH) <sub>8</sub> (H <sub>2</sub> O) <sub>3</sub> (DMF)(TCBPA) <sub>4</sub> (HCOO) <sub>4</sub> ] <sub>n</sub> ·(NO <sub>3</sub> ) <sub>2</sub> (Th-SINAP-200)	IO <sub>3</sub> <sup>-</sup>	Turn off	2.2×10 <sup>4</sup>	1.07×10 <sup>-7</sup>	6.12×10 <sup>-10</sup>	Water	SC-SC and ET	S71
2	[Th <sub>6</sub> (μ <sub>3</sub> -O) <sub>4</sub> (μ <sub>3</sub> -OH) <sub>4</sub> (H <sub>2</sub> O) <sub>4</sub> (DMF) <sub>2</sub> (BDAT) <sub>2</sub> (HCOO) <sub>4</sub> ] <sub>n</sub> ·(DMF) <sub>13</sub> (H <sub>2</sub> O) <sub>2</sub> <sub>4</sub> (Th-BDAT)	IO <sub>3</sub> <sup>-</sup>	Turn off	3.86×10 <sup>4</sup>	2.39×10 <sup>-4</sup>	1.37×10 <sup>-6</sup>	Water	Adsorption, CA, and ET	S72
3	[Th <sub>6</sub> (μ <sub>3</sub> -O) <sub>4</sub> (μ <sub>3</sub> -OH) <sub>4</sub> (H <sub>2</sub> O) <sub>6</sub> (BITD)] <sub>n</sub>	IO <sub>3</sub> <sup>-</sup>	Turn off	6.65×10 <sup>3</sup>	3.53×10 <sup>-4</sup>	2.02×10 <sup>-6</sup>	Water	Adsorption	S73

	$\text{)}_2(\text{HCOO})_4\text{]}\cdot 39.5$ DMF·48.5H <sub>2</sub> O (Th-BITD-1)						and ET
4	$[\text{Th}_6(\mu_3-\text{O})_4(\mu_3-\text{OH})_4(\text{H}_2\text{O})_6(\text{BITD})\text{)}_2(\text{HCOO})_4\text{]}\cdot 16\text{D}$ MF·19H <sub>2</sub> O (Th-BITD-2)		$2.97\times 10^3$	$1.84\times 10^{-3}$	$1.05\times 10^{-5}$		
5	$[\text{Zr}_6(\mu_3-\text{OH})_4(\mu_3-\text{O})_4(\text{OH})_4(\text{BITD})_2(\text{H}_2\text{O})_4\text{]}\cdot (\text{DMF})_6(\text{H}_2\text{O})_{20}$ (Zr-BITD)	IO <sub>3</sub> <sup>-</sup>	Turn off	$8.02\times 10^3$	$3.59\times 10^{-4}$	$2.05\times 10^{-6}$	Water Adsorption, CA, and ET S74

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