## Theoretical Evaluation of Na<sub>2</sub>MgCl<sub>4</sub> Double Chlorite as an Electrolyte for All-Solid-State Sodium Ion Batteries

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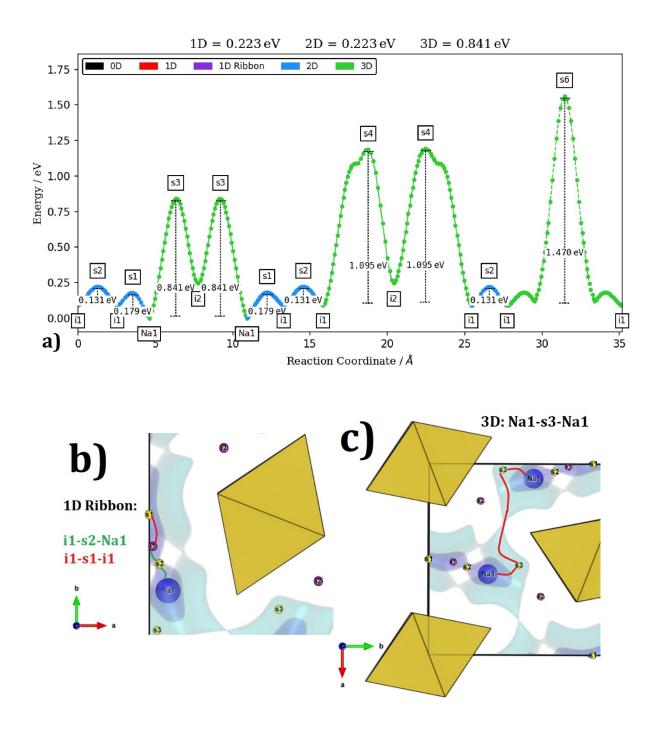
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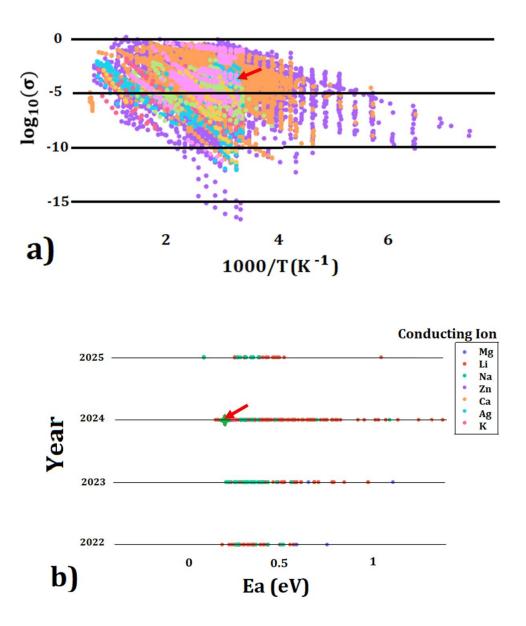
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## **Supplementary Information**



**Fig. S1**: **a)** Energy vs. reaction coordinate profile of Na<sub>2</sub>MgCl<sub>4</sub>, showing the 3D network of Na<sup>+</sup> migration pathways (blue isosurfaces) calculated by bond valence site energy. **b)** 1D ribbons and **c)** 3D migration pathways. The yellow polyhedral in panels **b)** and **c)** represent the [MgCl4] octahedral.



**Fig. S2**: Performance ranking of  $Na_2MgCl_4$  based on **a**) room-temperature conductivity ( $\sigma$ ) and **b**) theoretical activation energy (Ea), as reported in the Dynamic Database for Solid State Electrolyte (DDSE). The red arrow denotes the activation energy and room-temperature conductivity specific to  $Na_2MgCl_4$ .