

NaCaGa₃S₆: A Mixed Alkali Metal and Alkaline Earth Metal Sulfide with a Distinctive [Ga₅S₁₀]_∞ Layered Framework

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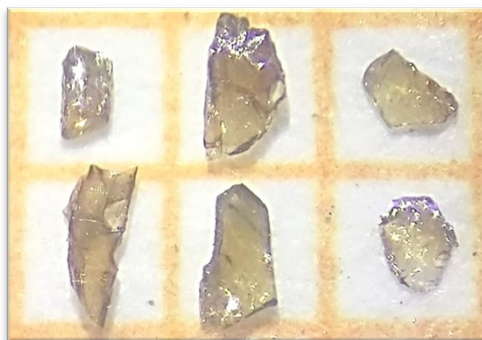


Figure S1. The crystals of NCGS.

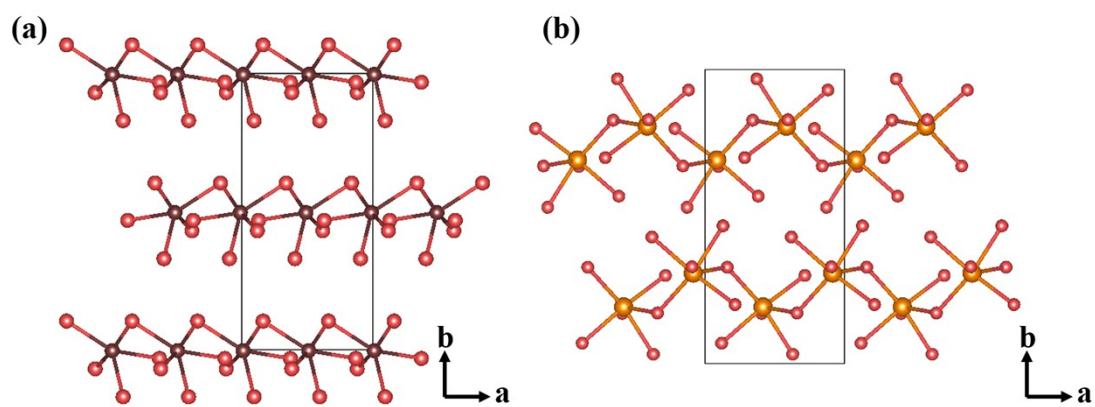


Figure S2. (a) one-dimensional (1D) $[\text{Na}_2\text{S}_8]_\infty$ chains and (b) 1D $[\text{Ca}_2\text{S}_{10}]_\infty$ chains.

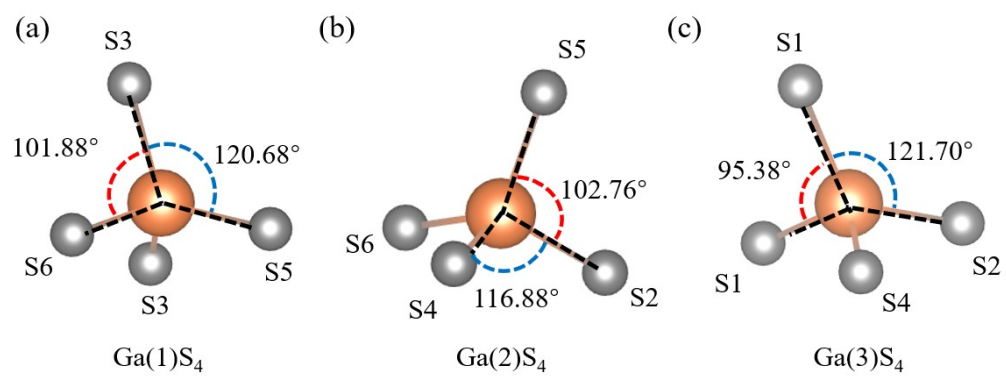


Figure S3. The bond-angle difference of GaS₄ groups in NCGS.

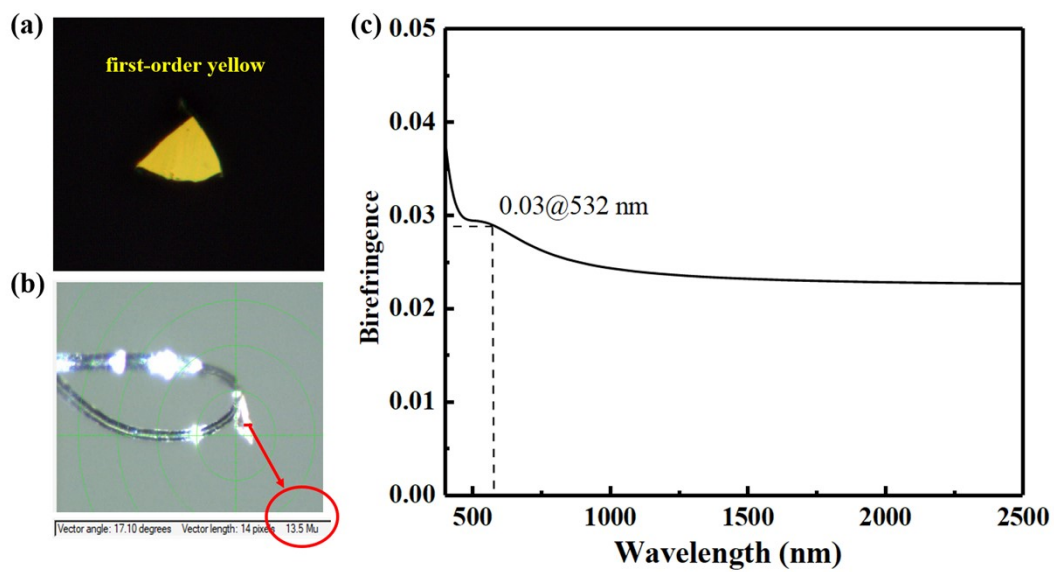


Figure S4. (a) The interference color of NCGS observed in the cross-polarized light, (b) The thickness of NCGS crystal, (c) The calculated birefringence curve for NCGS.

Table S1. Crystal data and structure refinement for NCGS

Empirical formula	NaCaGa ₃ S ₆
Formula weight	206.48
Temperature (K)	297(2)
Crystal system	Orthorhombic
Space group	<i>Pbca</i>
Z	18
<i>a</i> (Å)	6.9613(5)
<i>b</i> (Å)	14.6439(9)
<i>c</i> (Å)	20.1221(15)
<i>V</i> (Å ³)	2051.3(2)
<i>D_c</i> (g cm ⁻³)	3.009
<i>μ</i> (mm ⁻¹)	9.515
<i>F</i> (000)	1760
Radiation	Mo-K _α (<i>λ</i> = 0.71073)
2 <i>θ</i> range(°)	2.024 to 27.502
Reflections collected	17376
Indep. Reflns/ Rint	2350/0.0398
GOOF on <i>F</i> ²	1.142
<i>R_I</i> , <i>wR₂</i> (<i>I</i> > 2 <i>σ</i> (<i>I</i>)) ^a	0.0212, 0.0387
<i>R_I</i> , <i>wR₂</i> (all data)	0.0278, 0.0468
largest diff. peak and hole (e ⁻ Å ⁻³)	0.511, -0.699
^a <i>R</i> ₁ = Σ <i>F_o</i> — <i>F_c</i> /Σ <i>F_o</i> , ^b <i>wR</i> ₂ = Σ <i>w</i> (<i>F_o</i> ² — <i>F_c</i> ²) ² /Σ <i>w</i> (<i>F_o</i> ²) ²] ^{1/2} .	

Table S2. Atomic coordinates ($\times 10^4$), equivalent isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) and bond valence sums (BVS) for NCGS. U_{eq} is defined as one-third of the trace of the orthogonalized U_{ij} tensor.

Atom	BVS ^a	x	y	z	$U_{eq}(\text{\AA})$
Na(1)	1.146	4895(2)	4966(1)	1845(1)	33(1)
Ca(1)	1.911	10847(1)	8076(1)	4170(1)	18(1)
Ga(1)	2.904	8286(1)	7375(1)	2670(1)	12(1)
Ga(2)	2.972	5204(1)	6130(1)	3810(1)	12(1)
Ga(3)	2.909	8097(1)	4503(1)	4856(1)	13(1)
S(1)	1.830	8813(1)	5694(1)	5532(1)	15(1)
S(2)	1.897	6289(1)	4684(1)	3933(1)	16(1)
S(3)	1.813	5941(1)	8294(1)	2275(1)	16(1)
S(4)	1.858	6590(1)	3279(1)	5333(1)	14(1)
S(5)	1.938	3376(1)	6035(1)	2894(1)	16(1)
S(6)	2.030	7842(2)	7004(1)	3748(1)	31(1)

^aBond valence state was calculated using the empirical formula $V_i = \sum S_{ij} = \sum \exp[(r_0 - r_{ij})/0.37]$, where S_{ij} is the bond valence associated with bond lengths r_{ij} and r_0 .

Table S3. Selected bond distances (Å) and angles (°).

Na(1)-S(5)	2.8326(18)	Ga(2)-S(5)	2.2452(9)
Na(1)-S(5)#1	2.9327(18)	Ga(2)-S(2)	2.2626(9)
Na(1)-S(1)#2	2.9540(18)	Ga(2)-S(4)#8	2.2971(9)
Na(1)-S(2)#3	2.9871(19)	Ga(3)-S(2)	2.2593(9)
Na(1)-S(3)#4	3.078(2)	Ga(3)-S(1)	2.2663(9)
Ca(1)-S(6)	2.7498(12)	Ga(3)-S(4)	2.2869(9)
Ca(1)-S(1)#5	2.8050(11)	Ga(3)-S(1)#7	2.3069(9)
Ca(1)-S(2)#6	2.8253(11)	Ga(2)-S(5)	2.2452(9)
Ca(1)-S(4)#7	2.8498(11)	Ga(2)-S(2)	2.2626(9)
Ca(1)-S(4)#6	2.9051(12)	Ga(2)-S(4)#8	2.2971(9)
Ca(1)-S(3)#1	2.9267(12)	Ga(3)-S(2)	2.2593(9)
Ca(1)-Ga(1)	3.6530(8)	Ga(3)-S(1)	2.2663(9)
Ca(1)-Ga(3)#6	3.7161(8)	Ga(3)-S(4)	2.2869(9)
Ga(1)-S(6)	2.2578(10)	Ga(3)-S(1)#7	2.3069(9)
Ga(1)-S(3)	2.2596(9)	Ga(2)-S(5)	2.2452(9)
Ga(1)-S(5)#1	2.2673(9)	Ga(2)-S(2)	2.2626(9)
Ga(1)-S(3)#1	2.2889(9)	Ga(2)-S(4)#8	2.2971(9)
Ga(2)-S(6)	2.2417(10)	Ga(3)-S(2)	2.2593(9)
Ga(2)-S(5)	2.2452(9)	S(5)-Na(1)-S(5)#1	83.09(5)
Ga(2)-S(2)	2.2626(9)	S(5)-Na(1)-S(1)#2	163.94(8)
Ga(2)-S(4)#8	2.2971(9)	S(5)#1-Na(1)-S(1)#2	94.77(5)
Ga(3)-S(2)	2.2593(9)	S(5)-Na(1)-S(2)#3	98.83(6)
Ga(3)-S(1)	2.2663(9)	S(5)#1-Na(1)-S(2)#3	149.56(7)
Ga(3)-S(4)	2.2869(9)	S(1)#2-Na(1)-S(2)#3	75.03(4)
Ga(3)-S(1)#7	2.3069(9)	S(5)-Na(1)-S(3)#4	86.60(5)
Na(1)-S(5)	2.8326(18)	S(5)#1-Na(1)-S(3)#4	118.56(6)
Na(1)-S(5)#1	2.9327(18)	S(1)#2-Na(1)-S(3)#4	108.18(6)
Na(1)-S(1)#2	2.9540(18)	S(2)#3-Na(1)-S(3)#4	91.86(5)
Na(1)-S(2)#3	2.9871(19)	S(6)-Ca(1)-S(1)#5	173.03(4)
Na(1)-S(3)#4	3.078(2)	S(6)-Ca(1)-S(2)#6	91.32(4)
Ca(1)-S(6)	2.7498(12)	S(1)#5-Ca(1)-S(2)#6	83.58(3)
Ca(1)-S(1)#5	2.8050(11)	S(6)-Ca(1)-S(4)#7	100.79(4)
Ca(1)-S(2)#6	2.8253(11)	S(1)#5-Ca(1)-S(4)#7	84.93(3)
Ca(1)-S(4)#7	2.8498(11)	S(2)#6-Ca(1)-S(4)#7	165.73(4)
Ca(1)-S(4)#6	2.9051(12)	S(6)-Ca(1)-S(4)#6	82.12(3)
Ca(1)-S(3)#1	2.9267(12)	S(1)#5-Ca(1)-S(4)#6	101.05(3)
Ca(1)-Ga(1)	3.6530(8)	S(2)#6-Ca(1)-S(4)#6	75.12(3)
Ca(1)-Ga(3)#6	3.7161(8)	S(4)#7-Ca(1)-S(4)#6	98.90(3)
Ga(1)-S(6)	2.2578(10)	S(6)-Ca(1)-S(3)#1	76.84(3)
Ga(1)-S(3)	2.2596(9)	S(1)#5-Ca(1)-S(3)#1	97.23(3)
Ga(1)-S(3)#1	2.2889(9)	S(4)#7-Ca(1)-S(3)#1	114.22(3)
Ga(2)-S(6)	2.2417(10)	S(4)#6-Ca(1)-S(3)#1	143.34(4)
S(6)-Ca(1)-Ga(1)	38.11(2)	S(2)-Ga(3)-Ca(1)#9	49.38(2)

S(1)#5-Ca(1)-Ga(1)	135.75(3)	S(1)-Ga(3)-Ca(1)#9	144.87(3)
S(2)#6-Ca(1)-Ga(1)	80.65(3)	S(4)-Ga(3)-Ca(1)#9	51.42(3)
S(4)#7-Ca(1)-Ga(1)	113.56(3)	S(1)#7-Ga(3)-Ca(1)#9	119.55(3)
S(4)#6-Ca(1)-Ga(1)	114.16(3)	Ga(3)-S(1)-Ga(3)#7	84.62(3)
S(3)#1-Ca(1)-Ga(1)	38.77(2)	Ga(3)-S(1)-Ca(1)#10	117.43(4)
S(6)-Ca(1)-Ga(3)#6	82.73(3)	Ga(3)#7-S(1)-Ca(1)#10	147.01(4)
S(1)#5-Ca(1)-Ga(3)#6	95.92(3)	Ga(3)-S(1)-Na(1)#11	110.60(5)
S(2)#6-Ca(1)-Ga(3)#6	37.37(2)	Ga(3)#7-S(1)-Na(1)#11	88.74(4)
S(4)#7-Ca(1)-Ga(3)#6	136.38(3)	Ca(1)#10-S(1)-Na(1)#11	104.07(5)
S(4)#6-Ca(1)-Ga(3)#6	37.980(19)	Ga(3)-S(2)-Ga(2)	112.66(4)
S(3)#1-Ca(1)-Ga(3)#6	108.93(3)	Ga(3)-S(2)-Ca(1)#9	93.24(3)
Ga(1)-Ca(1)-Ga(3)#6	95.989(19)	Ga(2)-S(2)-Ca(1)#9	128.53(4)
S(6)-Ga(1)-S(3)	112.49(4)	Ga(3)-S(2)-Na(1)#1	88.82(4)
S(6)-Ga(1)-S(5)#1	106.06(4)	Ga(2)-S(2)-Na(1)#1	95.38(4)
S(3)-Ga(1)-S(5)#1	111.03(4)	Ca(1)#9-S(2)-Na(1)#1	130.26(5)
S(6)-Ga(1)-S(3)#1	101.88(4)	Ga(1)-S(3)-Ga(1)#3	102.51(3)
S(3)-Ga(1)-S(3)#1	104.50(4)	Ga(1)-S(3)-Ca(1)#3	107.52(4)
S(5)#1-Ga(1)-S(3)#1	120.68(4)	Ga(1)#3-S(3)-Ca(1)#3	88.03(3)
S(6)-Ga(1)-Ca(1)	48.73(3)	Ga(1)-S(3)-Na(1)#12	114.07(5)
S(3)-Ga(1)-Ca(1)	118.43(3)	Ga(1)#3-S(3)-Na(1)#12	106.70(5)
S(5)#1-Ga(1)-Ca(1)	130.04(3)	Ca(1)#3-S(3)-Na(1)#12	130.85(4)
S(3)#1-Ga(1)-Ca(1)	53.20(3)	Ga(3)-S(4)-Ga(2)#8	105.62(3)
S(6)-Ga(2)-S(5)	116.99(4)	Ga(3)-S(4)-Ca(1)#7	113.92(4)
S(6)-Ga(2)-S(2)	105.49(4)	Ga(2)#8-S(4)-Ca(1)#7	109.83(4)
S(5)-Ga(2)-S(2)	102.76(3)	Ga(3)-S(4)-Ca(1)#9	90.60(3)
S(6)-Ga(2)-S(4)#8	105.79(4)	Ga(2)#8-S(4)-Ca(1)#9	108.99(4)
S(5)-Ga(2)-S(4)#8	109.36(4)	Ca(1)#7-S(4)-Ca(1)#9	125.21(3)
S(2)-Ga(2)-S(4)#8	116.87(3)	Ga(2)-S(5)-Ga(1)#3	111.93(4)
S(2)-Ga(3)-S(1)	121.70(3)	Ga(2)-S(5)-Na(1)	115.76(5)
S(2)-Ga(3)-S(4)	100.44(3)	Ga(1)#3-S(5)-Na(1)	96.64(5)
S(1)-Ga(3)-S(4)	116.88(3)	Ga(2)-S(5)-Na(1)#3	110.75(5)
S(2)-Ga(3)-S(1)#7	104.81(4)	Ga(1)#3-S(5)-Na(1)#3	121.93(5)
S(1)-Ga(3)-S(1)#7	95.38(3)	Na(1)-S(5)-Na(1)#3	98.44(6)
S(4)-Ga(3)-S(1)#7	118.18(3)	Ga(2)-S(6)-Ga(1)	107.64(4)
Ga(1)-S(6)-Ca(1)	93.16(4)	Ga(2)-S(6)-Ca(1)	158.73(5)

Symmetry transformations used to generate equivalent atoms:

#1 $x+1/2, y, -z+1/2$ #2 $-x+3/2, -y+1, z-1/2$ #3 $x-1/2, y, -z+1/2$
#4 $-x+1, y-1/2, -z+1/2$ #5 $x+1/2, -y+3/2, -z+1$ #6 $-x+3/2, y+1/2, z$
#7 $-x+2, -y+1, -z+1$ #8 $-x+1, -y+1, -z+1$ #9 $-x+3/2, y-1/2, z$
#10 $x-1/2, -y+3/2, -z+1$ #11 $-x+3/2, -y+1, z+1/2$
#12 $-x+1, y+1/2, -z+1/2$