# Supplementary material

# Engineered Amine Oxidase for Efficient Oxidative Dehydroaromatization of

# 1,2,3,4-Tetrahydroquinolines toward Quinolines in Aqueous Media

Mengmeng He,<sup>[a]</sup> Xiaoyang Yue,\*<sup>[a][b]</sup> Jianqiao Liu,<sup>[a][b]</sup> Guanhua Liu, <sup>[a][b]</sup> Liya Zhou,

<sup>[a][b]</sup> Ying He, <sup>[a][b]</sup> Li Ma, <sup>[a][b]</sup> Yunting Liu\*<sup>[a][b]</sup> and Yanjun Jiang\*<sup>[a][b]</sup>

[a] School of Chemical Engineering and Technology, Hebei University of Technology,

Tianjin 300401, China

[b] National-Local Joint Engineering Laboratory for Energy Conservation in Chemical Process Integration and Resources Utilization, Hebei University of Technology, Tianjin 300401, China

\*E-mail: xiaoyang.yue@hebut.edu.cn; \*E-mail: ytliu@hebut.edu.cn; \*E-mail: yanjunjiang@hebut.edu.cn

Chemicals and materials 4 -
Enzyme library mining
Protein expression and purification of model enzyme and candidate enzymes6
Enzyme library screening7
Mutagenesis
Enzymatic properties of VsAO9
Enzymatic properties of M210
Enzyme activity assay11
Molecular docking12
Determination of kinetic parameters
Time-course experiments14
Preparative-scale oxidative dehydroaromatization of 5a and 18b15
Analytical methods16
Figure S1. The phylogenetic tree of amine oxidases used in this study17
Figure S2. Multiple sequence alignments of these AOs used in this study
Figure S3. SDS-PAGE analysis of expression level of AOs
Figure S4. SDS-PAGE analysis of expression level of VsAO and its mutant21
Figure S5. Effect of temperature and pH on the enzyme activity and stability of <i>Vs</i> AO.
Figure S6. (a) first round of mutation screening. (b) second round of mutation screening.
Figure S7. Surface potential simulation of (i, ii) VsAO and (iii, iv) M224
Figure S8. Effect of temperature and pH on the enzyme activity and stability of M2.25
Figure S9. Kinetic parameters for the oxidative dehydroaromatization of substrate 1a
by VsAO and its mutants
Table S1. List of protein information of enzymes used in this study.
Table S2. Amino acid sequence of MAO5 <sup>[1]</sup>
Table S3. The physical and chemical properties of candidates predicted by Expasy and
NovoPro
Table S4. Screening for the forms of biocatalyst a,b,c    30
Table S5. Optimization of reaction conditions for oxidative dehydroaromatization of
substrate 1a by <i>Vs</i> AO <sup><i>a,b,c,d,e</i></sup>
Table S6. Gene sequence of wild-type VsAO    32      2 / 70    32

## Contents

Table S7. Primers used for site-directed mutagenesis.    33
Table S8. Kinetic parameters for the oxidative dehydroaromatization of substrate 1a by
VsAO and its mutants. <sup><i>a,b</i></sup>
Table S9. Comparison between classical chemical catalytic method and biocatalytic
method for the oxidative dehydroaromatization of THQ
Table S10. Optimization of the type and content of co-solvents for oxidative
dehydroaromatization of substrate 5a by VsAO <sup>a,b</sup>
<sup>1</sup> H and <sup>13</sup> C NMR spectra of the products recorded in CDCl <sub>3</sub> or DMSO- <i>d</i> <sub>6</sub> 38
NMR spectra of products45
Reference

### Chemicals and materials

All chemicals and reagents were purchased from commercial suppliers (MREDA, Kermel, Aladdin, Macklin, Solarbio and Sangon Biotech) at least of reagent grade and used without further purification unless otherwise noted. Potassium phosphate buffer (50 mM, pH=7.5) was used as a buffering system unless otherwise specified. *E. coli* BL21 (DE3) cells were grown using Luria–Bertani (LB) medium. pET-28a (+) was used as the cloning and expression vector for all enzymes described in this study.

#### **Enzyme library mining**

The candidate enzymes were mined by BLAST in NCBI database using MAO5 as template. Then, the sequences with identity less than 30% and more than 80% were removed. The physical and chemical properties of candidates, for instance, molecular weight, theoretical pI, amino acid composition, atomic composition, extinction coefficient, estimated half-life, instability index, aliphatic index and grand average of hydropathicity of proteins can be estimated in this database only by inputting the protein sequence, estimated online database were using Expasy (https://web.expasy.org/protparam/). The solubility of the candidates was estimated using Protein-Sol (https://protein-sol.manchester.ac.uk/)<sup>[1]</sup> and the genes which were estimated by the program Protein-Sol to be insolubly expressed in E. coli were eliminated. Multiple sequence alignments were performed using the MEGA server (https://www.megasoftware.net/show\_eua) displayed and using ESPript (https://espript.ibcp.fr/ESPript/ESPript/index.php). 6 sequences from each branch were selected for gene synthesis and characterization. The protein sequences encoding candidate enzymes and model enzyme (Supporting Information, Table S1) were codonoptimized and chemically synthesized (Sangon Biotech, China). The candidate amine oxidase from *Gammaproteobacteria* bacterium. Pseudomonas genes sp., Saccharospirillum sp., Thalassomonas sp. RHCI1, Luminiphilus sp. and Vibrio sp. JCM 19236. All enzyme genes were inserted into the expression vector pET-28a (+) and subsequently transformed into E. coli BL21 (DE3) for expression.

#### Protein expression and purification of model enzyme and candidate enzymes

Recombinant *E. coli* BL21 (DE3) strain were cultured in 10 mL LB medium containing 50 µg/mL kanamycin (final concentration) at 37°C and 200 rpm overnight. Then, 0.5 mL preculture was inoculated in 50 mL TB medium with 50 µg/mL kanamycin, cultured at 37°C and 200 rpm until OD600 reached 0.6~0.8, and then induced by addition of IPTG (final concentration of 0.2 mM) at 25°C for further 20 h. After medium removal by centrifugation (8000 rpm, 5 min, 4°C), the cell pellets were washed with potassium phosphate buffer (50 mM, pH 7.5) twice and resuspended in the same buffer. The cells were disrupted by high-pressure homogenization, and cell debris was removed by centrifugation (10000 rpm, 20 min, 4°C). Unless otherwise stated, the crude lysate in this study was obtained by this method. The crude lysates were used for SDS-PAGE analysis.

Ni-NTA affinity chromatography was performed to purify the filtered lysate. After pre-equilibrating the Ni-NTA column with buffer A (20 mM NaH<sub>2</sub>PO<sub>4</sub>, 500 mM NaCl, and 20 mM imidazole at pH 7.4), the filtered lysate was loaded into the Ni-NTA column. Subsequently, bound protein was recovered with gradient elution buffer (20 mM NaH<sub>2</sub>PO<sub>4</sub>, 500 mM NaCl, pH 7.4) containing 30 mM, 60 mM, 300 mM and 500 mM imidazole for 5-10 column volumes, respectively. Fractions containing target protein was verified by SDS-PAGE analysis and concentrated by ultrafiltration.

# **Enzyme library screening**

Reaction mixture comprising potassium phosphate buffer (50 mM, pH 7.5), substrate 1a (5 mM), purified AOs (2.0 mg/mL), in a total volume of 5 mL was incubated at 37°C with 200 rpm shaking for 24 h. The sample was then analyzed by HPLC method.

### **Mutagenesis**

The site-directed mutagenesis was conducted using recombinant plasmids containing VsAO genes (Table S6) as templates with the following program: 98 °C for 5 min, (98 °C for 2 min, 55 °C for 30 s, 68 °C for 60 s) with 15 cycles, 68 °C for 5 min 4 °C hold. 50  $\mu$ L PCR reaction mixture contains reverse upstream and downstream primers (1.5  $\mu$ L each), 2  $\mu$ L recombinant plasmid, 25  $\mu$ L  $K_{OD}$  DNA Polymerase and 20  $\mu$ L distilled water. The PCR products were digested by *NcoI* and *XhoI*, and transformed into *E. coli*. BL21 (DE3) cells. The sequences were verified by DNA sequencing, and the oxidative dehydroaromatization activity of developed *Vs*AO variants was detected via HPLC method. The primers used in this work are shown in Table S7.

### Enzymatic properties of VsAO

The effect of pH on enzyme activity was determined at diverse pH levels (5.0–10.5) in the following buffers (50 mM): Citrate buffer (pH 5.0, 5.5, 6.0), potassium phosphate buffer (pH 6.0, 6.5, 7.0, 7.5, 8.0) and Gly-NaOH buffer (pH 8.0, 8.5, 9.0, 9.5, 10.0, 10.5). The pH stability of *Vs*AO was determined by preincubating the purified *Vs*AO in different pH (7.5, 10.0) buffers, followed by measuring the residual activity at 37°C. The effect of temperature on enzyme activity was determined at different temperatures ranging from 20 to 70°C. The thermostability of *Vs*AO was determined by preincubating the purified *Vs*AO at different temperatures (30°C, 50°C) followed by measuring the residual activity at 30°C. The highest enzyme activity was taken as 100%.

### **Enzymatic properties of M2**

The effect of pH on enzyme activity was determined at diverse pH levels (4.0–11.0) in the following buffers (50 mM): Citrate buffer (pH 4.0, 4.5, 5.0, 5.5, 6.0), potassium phosphate buffer (pH 6.0, 6.5, 7.0, 7.5, 8.0) and Gly-NaOH buffer (pH 8.0, 8.5, 9.0, 9.5, 10.0, 10.5, 11.0). The pH stability of M2 was determined by preincubating the purified M2 in different pH (7.5, 10.0) buffers, followed by measuring the residual activity at 30°C. The effect of temperature on enzyme activity was determined at different temperatures ranging from 20 to 70°C. The thermostability of M2 was determined by preincubating the purified M2 at different temperatures (35°C, 50°C) followed by measuring the residual activity at 30°C. The section activity at 30°C. The highest enzyme activity was determined by measuring the residual activity at 30°C. The highest enzyme activity was taken as 100%.

# Enzyme activity assay

The activities of *Vs*AO were measured by HPLC at 254 nm. The activity assay was carried out in 1 mL reaction mixture containing potassium phosphate buffer (50 mM, pH 7.5), substrate **1a** (5 mM) and appropriate purified enzyme solution at 30°C. 1 U was defined as the amount of enzyme required to catalyze the conversion of 1  $\mu$ mol of substrate per minute under specified assay conditions.

# Molecular docking

AutoDock 4 was used for the docking of substrate 1a into the predicted structures. A grid box of  $8 \times 8 \times 8$  with spacing of 1.0 Å, which encompassed the active sites of *Vs*AO, was set as the search space to explore suitable substrate-binding modes. The PDBQT format was employed for both the input and output of molecular structures.

### **Determination of kinetic parameters**

Purified  $V_sAO$  and its mutants were kinetically characterized to determine Michaelis Menten-parameters for oxidative dehydroaromatization of 2-methyl-1,2,3,4tetrahydroquinoline (2-MeTHQ, **1a**). The substrate concentrations were varied from 0.5-10 mM while keeping the other reaction conditions same. Reactions were performed at 30°C in 1 mL volume and contained 50 mM potassium phosphate buffer (pH 7.5), substrate **1a** and appropriate purified enzyme solution. Due to limited substrate solubility, activity at higher concentrations could not be explored.

#### **Time-course experiments**

The oxidative dehydroaromatization reaction of **5a** was carried out in a 1000 mL three-neck flask equipped with a magnetic stirrer. Substrate **5a** (756 mg, 5 mmol) was initially dissolved in 50 mL of DMSO as substrate solution. Then, 50 mL of substrate solution and freeze-dried whole cells containing *Vs*AO or M2(16 g/L) was added to 450 mL of the potassium phosphate buffer (50 mM, pH=7.5). The final concentration of substrate **5a** was 10 mM in the reaction. The pH condition was kept at 7.5 and a thermostat-controlled water bath was used to maintain the reaction at 30°C. The reaction was monitored by HPLC.

#### Preparative-scale oxidative dehydroaromatization of 5a and 18b

The preparative-scale oxidative dehydroaromatization reaction of 5a and 18a was carried out in a 1000 mL three-neck flask equipped with a magnetic stirrer. Substrate 5a (756 mg, 5 mmol) or 18a (838 mg, 5 mmol) was initially dissolved in 50 mL of DMSO as substrate solution. Then, 50 mL of substrate solution and freeze-dried whole cells containing M2(16 g/L) was added to 450 mL of the potassium phosphate buffer (50 mM, pH=7.5). The final concentration of substrate 5a and 18a was 10 mM in the reaction. The pH condition was maintained at 7.5 and a thermostat-controlled water bath was used to keep the reaction at 30°C. The reaction was monitored by HPLC. Upon the completion of the reactions, 500 mL of samples was collected, 1500 mL of ethyl acetate was added, and the organic layers were dried over MgSO<sub>4</sub> and filtered. The residue was purified by column chromatography using a gradient of hexane and ethyl acetate. The mixture was concentrated so that products 5b and 18b was obtained and analyzed described the ESI. using NMR, as in

### **Analytical methods**

**Method 1** HPLC was performed with a Daicel Chiralcel OJ-H column (nhexane/isopropanol mixtures as solvent) or Daicel Chiralcel AD column (nhexane/isopropanol/diethylamine mixtures as solvent) at 30°C at 254 nm UV detector.

**Method 2** <sup>1</sup>H NMR and <sup>13</sup>C NMR spectra were recorded on a Bruker AV 400 spectrometer at 400 MHz with CDCl<sub>3</sub>/DMSO-d6 as solvent. Chemical shifts were reported in ppm relative to internal TMS for <sup>1</sup>H NMR data, respectively. Data are presented in the following space: chemical shift, multiplicity, coupling constant in hertz (Hz), and signal area integration in natural numbers.

### **Supplementary Data**

The phylogenetic tree of amine oxidases



Figure S1. The phylogenetic tree of amine oxidases used in this study.

The phylogenetic tree was performed using the MEGA 11.0 and was displayed by ITOL (<u>https://itol.embl.de/</u>).

### Multiple sequence alignments





Figure S2. Multiple sequence alignments of these AOs used in this study.

Corresponding sources and GenBank accession numbers were given in Table S1. Multiple sequence alignments were performed using the MEGA server (https://www.megasoftware.net/show\_eua) and displayed using ESPript (https://espript.ibcp.fr/ESPript/ESPript/index.php).

### **SDS-PAGE** analysis



# Figure S3. SDS-PAGE analysis of expression level of AOs.

Lane 1, protein markers. Lane 2, crude extract of GbAO. Lane 3, crude extract of PsAO. Lane 4, crude extract of SsAO. Lane 5, crude extract of TsAO. Lane 6, crude extract of MAO5. Lane 7, crude extract of LsAO. Lane 8, crude extract of VsAO. Seven proteins are soluble which have similar molecular mass of approximately 50 KDa.



Figure S4. SDS-PAGE analysis of expression level of VsAO and its mutant.

Lane 1, protein markers. Lane 2, crude extract of *Vs*AO. Lane 3, purified *Vs*AO. Lane 4, crude extract of M1. Lane 5, purified M1. Lane 6, crude extract of M2. Lane 7, purified M2. Four proteins are soluble which have the same molecular mass of approximately 50 KDa.

Enzymatic properties of VsAO



**Figure S5.** Effect of temperature and pH on the enzyme activity and stability of *Vs*AO. (a) Effect of pH on the enzyme activity was determined by measuring the activity at various buffers (pH 5.0-10.5) at 37°C. The value at pH 7.5 was taken as 100%. (b) Effect of pH on the enzyme stability was evaluated by measuring the residual activity after incubation of the enzyme in different buffers (pH 7.5, 10.0) at 25°C. (c) Effect of temperature on the enzyme activity was determined by measuring the activity at various temperatures (20-70°C) at pH 7.5. The value at 30°C was set as 100%. (d) Effect of temperature on the enzyme stability was determined by measuring the remaining activity after incubation of the enzyme at 30 and 50°C. All experiments were performed in triplicate. Each data point represents the mean  $\pm$  the standard deviation of three measurements.

# Mutagenesis



Figure S6. (a) first round of mutation screening. (b) second round of mutation screening.



Figure S7. Surface potential simulation of (i, ii) VsAO and (iii, iv) M2.

**Enzymatic properties of M2** 



**Figure S8.** Effect of temperature and pH on the enzyme activity and stability of M2.

(a) Effect of pH on the enzyme activity was determined by measuring the activity at various buffers (pH 4.0-11.0) at 30°C. The value at pH 7.5 was taken as 100%. (b) Effect of pH on the enzyme stability was evaluated by measuring the residual activity after incubation of the enzyme in different buffers (pH 7.5, 10.0) at 25°C. (c) Effect of temperature on the enzyme activity was determined by measuring the activity at various temperatures (20-70°C) at pH 7.5. The value at 35°C was set as 100%. (d) Effect of temperature on the enzyme stability was determined by measuring the remaining activity after incubation of the enzyme at 35 and 50°C. All experiments were performed in triplicate. Each data point represents the mean  $\pm$  the standard deviation of three measurements.

### **Kinetic parameters**



Figure S9. Kinetic parameters for the oxidative dehydroaromatization of substrate 1a by *Vs*AO and its mutants.

(a) Effect of substrate concentration on the oxidative dehydroaromatization reaction initial rate. (b) Kinetic parameters for the oxidative dehydroaromatization of substrate 1a by VsAO. (c) Kinetic parameters for the oxidative dehydroaromatization of substrate 1a by M1. (d) Kinetic parameters for the oxidative dehydroaromatization of substrate 1a by M2. All experiments were performed in triplicate. Each data point represents the mean  $\pm$  the standard deviation of three measurements.

Enzyme	Source	GenBank Accession Number	Protein sequence identity (%)
GbAO	Gammaproteobacteria bacterium	MCL5043555.1	72
PsAO	Pseudomonas sp.	HCW97116.1	75
SsAO	Saccharospirillum sp.	MCH8530221.1	55
TsAO	Thalassomonas sp. RHCI1	WP_281556337.1	57
LsAO	Luminiphilus sp.	MDB4491165.1	57
VsAO	Vibrio sp. JCM 19236	GAM70821.1	55
MAO5	Pseudomonas monteilii ZMU-T01	WP_013972159.1	100

 Table S1. List of protein information of enzymes used in this study.

# AMINO ACID SEQUENCE OF MAO5

MRIAIIGSGIAGLSCAHLLSRKHEVTVFEAEKWIGGHTHTLDVIWQGERHAI DTGFIVFNDWTYPHFIRLLDHLKVASRPTEMSFSVHDPVTGLEYNGHDLNT LFAQRRNLVSPGFWGMLRDILRFNRQALADLDNQRIDANATLGAYLRAQR YGQRFIDHYIVPMGSAIWSMSRADMLGFPLAFFVRFCRNHGLLSVNQRPQ WRVIEGGSRSYVSPLCRPFAGRIRLDCKVFRVSRDEGGVTLTSAAGTERFD NVVFACHSDQALALLEAPSVQEREVLGAIGYASNDVVLHTDTRLLPRRKR AWASWNYRLGGPQQAPAALTYNMNILQGIQAPTTFCVSLNQTALVDPAQII ARFQYDHPQYSLAACAAQARQAQLQGQRHSYFCGAYWGNGFHEDGVVS ALKVAAHFGEHL

Enzyme	<b>Predicted</b> solubility	Theoretical pI	Instability index	Estimated half-life (hour)
VsAO	0.243	5.51	36.52	>10
LsAO	0.103	6.02	37.39	>10
TsAO	0.139	8.46	39.86	>10
SsAO	0.212	6.01	40.72	>10
PsAO	0.124	6.20	45.51	>10
GbAO	0.123	5.64	37.14	>10
MAO5	0.083	8.91	35.34	>10

**Table S3.** The physical and chemical properties of candidates predicted by Expasy andProtein-Sol.

Catalyst form	Conversion <sup>b</sup> (%)	Yield <sup>c</sup> (%)
Crude extract	60	48
Wet whole cells	67	56
Freeze-dried whole cells	68	58

**Table S4.** Screening for the forms of biocatalyst a,b,c.

<sup>*a*</sup>*Reaction conditions* (5.0 mL): **1a** (5.0 mM) and catalyst in KH<sub>2</sub>PO<sub>4</sub>-K<sub>2</sub>HPO<sub>4</sub> buffer (pH 7.5, 50 mM), and incubated at 37°C and shaken at 200 rpm for 24 h. <sup>*b*</sup>The conversion was determined by HPLC. Conversion =  $[n(mol)_{initial substrate} - n(mol)_{residual substrate}/n(mol)_{initial substrate}] \times 100\%$ . <sup>*c*</sup>The yield was determined by HPLC. Yield =  $[n(mol)_{generated product}/n(mol)_{initial substrate}] \times 100\%$ .

E.4.	D. St. (50 . M)	Temperature	Concentration of	Conversion <sup>b</sup>	Yield <sup>c</sup>
Entry	Butter (50 mM)	(°C)	biocatalyst (g/L)	(%)	(%)
1	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.0)	37	8	60	48
2	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	37	8	67	55
3	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 8.0)	37	8	59	45
4	Gly-NaOH (pH 8.0)	37	8	51	40
5	Gly-NaOH (pH 8.5)	37	8	48	38
6	Gly-NaOH (pH 9.0)	37	8	46	35
7	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	20	8	62	53
8	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	25	8	69	56
9	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	8	71	60
10	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	37	8	67	55
11	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	40	8	58	44
12	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	4	60	48
13	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	8	71	60
14	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	12	83	70
15	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	16	91	76
16	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	20	79	72
17	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	30	69	62
18	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	40	67	56
19 <sup>d</sup>	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	16	92	78
20 <sup>e</sup>	KH <sub>2</sub> PO <sub>4</sub> -K <sub>2</sub> HPO <sub>4</sub> (pH 7.5)	30	16	99	83

**Table S5.** Optimization of reaction conditions for oxidative dehydroaromatization of substrate 1a by *Vs*AO<sup>*a,b,c,d,e*</sup>.

<sup>*a*</sup>*Reaction conditions* (5.0 mL): **1a** (5 mM) and lyophilized whole cells in KH<sub>2</sub>PO<sub>4</sub>-K<sub>2</sub>HPO<sub>4</sub> or Gly-NaOH buffer (50 mM). Reactions were incubated at 30°C and shaken at 200 rpm for 24 h. <sup>*b*</sup>The conversion was determined by HPLC. Conversion =  $[n(mol)_{initial substrate} - n(mol)_{residual substrate}/n(mol)_{initial substrate}] \times 100\%$ . <sup>*c*</sup>The yield was determined by HPLC. Yield =  $[n(mol)_{generated product}/n(mol)_{initial substrate}] \times 100\%$ . <sup>*d*</sup>Catalase was added. <sup>*e*</sup> Reaction was extended to 48 h.

 Table S6. Gene sequence of wild-type VsAO

# GENE SEQUENCE OF WILD-TYPE VsAO

ATGAACATCGCGATCATCGGTACCGGTATCAGCGGCCTGACCTGCGCGT ACCGTCTGCACCAGGAACACGAAGTTACCCTGTTCGAAGCGAACGACTA CATCGGCGGTCATACCGCGACCGTTGATGTGACCCTGGATGGTAAAGAA TACGCGGTTGACACCGGTTTCATCGTTTACAACGACCGCACCTACCCGA AATGAGCTTCTCTGTTCGCAACGATGCGAACGGCCTGGAATACAACGGC CACTCCCTGTCTAACCTGTTCGCGCAGAAACGTAACTGGCTGAACCCGA AATTCTACTCCTTCATCCTGGAAATCCTGCGTTTTAACAAACTGGCGAA AGAATTCGCGGATCAGGATATCGATGCGGATAGCACCCTGGGTGAATTC CTGGAAGAACACACTTTCGGCGATTACTTTTGCGATAACTACATCCTGC CGATGGGCGCGGCGATTTGGTCCTCTACCCTGGCGGATATGCGTGGCTT CCCGCTGCGTTTCTTCCTGCAGTTCTTCCTGAACCACGGCCTGCTGGATA TCAAAAACCGTCCGCAGTGGTACGTTGTTAAAGGTGGTAGCCGTGCGTA TATCGATCCGCTGACCCAGGGTTTCCGTGATAACATTCGTCTGAGCTCCC CGGTTCACAAAGTTGAACGTTCCGATACCGGCATCCGTCTGCAGGTGAA CGGCGAATGGCACGATTTCGATCAGGTGATTTTCGCCTGCCACAGCGAT CAGGCGCTGGCTATCCTGGGCGAACAGGCGAGCTCCGCGGAACGTGAA GTTCTGGGCGATATGGCGTACCAGGAAAACGAAGTTGTTCTGCATACCG ATACCCAGCTGCCGAAACGTAAACTGGCCTGGGCGAGCTGGAACT ATTACCTGAAAGGCGGCGCGGGACCAGGAAACCCGTCTGCCGACCCTGA CCTACAACATGAACATCCTGCAGCACATCCAGTCTGAACACACCTTCTG TGTTTCTCTGAACAACAGCTCTATGATCTCTTCTGATAAAATCCTGCGTT CTTTCAACTACCACCACCCGGTTTTCACCCGTGCTAGTATCGCGGCGCA GAAAAAGCTGGCCTGGTTCAGGGCAAACAGGGTACCTGGTTCTGCGG CGCGTACTGGCGTAACGGTTTTCACGAAGATGGTGTTGTTAGCGCACTG GAAGTTGTTAAAGGTATCCAGGCTTCTCAGCCGGCAGCTGCGCCGTTCG AACAGAAAGGCGCGCGCGCTCGAG

 Table S7. Primers used for site-directed mutagenesis.

PRIMER	SEQUENCE (5' TO 3')
V58F-FP	CGGTTTCATCTTTTACAACGACCGC
V58F-RP	GGTCGTTGTAAAAGATGAAACCGGTG
V58W-FP	CGGTTTCATCTGGTACAACGACC
V58W-RP	GTCGTTGTACCAGATGAAACCGG
N127R-FP	TCCTGCGTTTTAGAAAACTGGCGA
N127R-RP	TTCGCCAGTTTTCTAAAACGCAGGAT
N127W-FP	TCCTGCGTTTTTGGAAACTGGC
N127W-RP	CCAGTTTCCAAAAACGCAGGATT
H365F-FP	TCAACTACCACTTTCCGGTTTTCACC
H365F-RP	GGTGAAAACCGGAAAGTGGTAGTTG
H365W-FP	CAACTACCACTGGCCGGTTTTC
H365W-RP	AAAACCGGCCAGTGGTAGTTGA
H365D-FP	CAACTACCACGATCCGGTTTTCAC
H365D-RP	TGAAAACCGGATCGTGGTAGTTGAA
G398D-FP	CTGGCGTAACGATTTTCACGAAGAT
G398D-RP	CTTCGTGAAAATCGTTACGCCAGT
F399G-FP	GTAACGGTGGACACGAAGATGG
F399G-RP	ACCATCTTCGTGTCCACCGTTA
F399L-FP	CGTAACGGTCTTCACGAAGATGG
F399L-RP	CATCTTCGTGAAGACCGTTACGC
F399D-FP	CGTAACGGTGATCACGAAGATGG
F399D-RP	CATCTTCGTGATCACCGTTACGC
N127K FD	AAATCCTGCGTTTTAAGAAACTGGCGAAAG
1112/18-11	A
N127K-RP	ATTCTTTCGCCAGTTTCTTAAAACGCAGGAT

33 / 70

N127Y-FP	AAATCCTGCGTTTTTATAAACTGGCGAAAGA		
	AATTCTTTCGCCAGTTTATAAAAACGCAGGA		
N127Y-RP	Т		
L299M-FP	GATACCCAGCTGATGCCGAAACGTAAA		
L299M-RP	AGTTTACGTTTCGGCATCAGCTGGGTA		
F67L-FP	ACCTACCCGAACCTTATGCAGATGATGT		
F67L-RP	CATCATCTGCATAAGGTTCGGGTAGGTG		
F67M-FP	ACCTACCCGAACATGATGCAGATGATGT		
F67M-RP	ATCATCTGCATCATGTTCGGGTAGGTGC		
R88S-FP	GAGCTTCTCTGTTTCTAACGATGCGAACGG		
R88S-RP	CGTTCGCATCGTTAGAAACAGAGAAGCTCA		
M71F-FP	CTTCATGCAGATGTTTTCTGAAATCGGCGT		
M71F-RP	AACGCCGATTTCAGAAAACATCTGCATGAA		
M71L-FP	ACTTCATGCAGATGCTTTCTGAAATCGGCGT		
	AACGCCGATTTCAGAAAGCATCTGCATGAA		
M/IL-KP	G		
F368S-FP	ACCCGGTTTCTACCCGTGCTAGTAT		
F368S-RP	TAGCACGGGTAGAAACCGGGT		
F368T-FP	CACCCGGTTACCACCCGTGCTAGTATC		
F368T-RP	ATACTAGCACGGGTGGTAACCGGGT		
F368C-FP	ACCCGGTTTGCACCCGTGCTAGTAT		
F368C-RP	GATACTAGCACGGGTGCAAACCGGGT		
F368D-FP	ACCCGGTTGATACCCGTGCTA		
F368D-RP	ATACTAGCACGGGTATCAACCGGG		
N397S-FP	TACTGGCGTTCCGGTTTTCACG		
N397S-FP N397S-RP	TACTGGCGTTCCGGTTTTCACG TGAAAACCGGAACGCCAGTACG		
N397S-FP N397S-RP N397T-FP	TACTGGCGTTCCGGTTTTCACG TGAAAACCGGAACGCCAGTACG TACTGGCGTACCGGTTTTCACGA		

34 / 70

N397C-FP	TACTGGCGTTGCGGTTTTCACG
N397C-RP	AAAACCGCAACGCCAGTACG
N397D-FP	TACTGGCGTGATGGTTTTCACG
N397D-RP	AAAACCATCACGCCAGTACGC

Enzyme	V <sub>max</sub>	$k_{\text{cat}}$	$K_{\rm M}$	$k_{\text{cat}}/K_{\text{M}}$	Activity <sup>b</sup>
		(mm)	(IIIIVI)		(0/g)
VsAO	0.488	7.9	8.5	0.9	17.4
M1(F368D)	0.836	14.5	13.5	1.1	27.2
M2 (F368D/N127K)	0.733	13.7	11.2	1.2	40.3

**Table S8.** Kinetic parameters for the oxidative dehydroaromatization of substrate 1a byVsAO and its mutants.<sup>*a,b*</sup>

<sup>*a*</sup>Kinetic parameters were determined in potassium phosphate buffer (50 mM, pH 7.5) containing purified AOs and varied concentration of 1a at 30°C. <sup>*b*</sup>The activity assay was carried out in 1 mL reaction mixture containing potassium phosphate buffer (50 mM, pH 7.5), substrate **1a** (5 mM) and appropriate purified enzyme solution at 30°C.

**Table S9.** Comparison between classical chemical catalytic method and biocatalytic

 method for the oxidative dehydroaromatization of THQ.

	Oxidative dehydroaromatization		
Catalyst	<b>Reaction condition</b>	Yield	Reference
NĽDe	<i>t</i> -amyl alcohol, TEMPO, <i>t</i> -BuOK, 95°C, 24 h,	000/	[3]
NIBr <sub>2</sub>	250 mM, 0.025 mmol catalyst, 0.1 MPa $O_2$	99%	[3]
Co Phon@C	n-decane, <i>t</i> -BuOK, 150°C, 36 h, 250 mM,	020/	[4]
Co-Phen@C	0.03 mmol catalyst, 0.1 MPa Ar.	93%0	[ ']
NDCU	H <sub>2</sub> O, 0.1 MPa air, 120°C, 12 h, 133 mM, 5	0.40/	[5]
NPCH	mg catalyst	94%	[- ]
	acetonitrile, 2.5 bar O <sub>2</sub> , 150°C, 12 h, 100	07 (0/	[6]
Co/MC	mM, 20 mg catalyst	97.6%	[0]
	<i>o</i> -xylene, <i>t</i> -BuOK, 140°C, 36 h, 250 mM, 0.1	020/	[7]
Acceptorless method	MPa N <sub>2</sub>		[,]
M2 (montaint of V AO)	KPi buffer, air, 30°C, 12 h, 5 mM, 16 g/L	000/	
wiz (mutant of <i>vs</i> AO)	lyophilized whole cells	88%0	i nis work

<b>F</b>	Type and Content of Co-	Yield <sup>b</sup> (%
Entry	solvent(v/v)	
1	PBS buffer	64
2	10% DMSO	73
3	10% Dodecane	54
4	10% Ethanol	51
5	10% Isooctane	40
6	10% DMF	61
7	10% Oleic acid	15
8	10% 2-MeTHF	13
9	10% EtOAc	24
10	10% MTBE	37
11	5% DMSO	70
12	10% DMSO	73
13	15% DMSO	71
14	20% DMSO	69

**Table S10.** Optimization of the type and content of co-solvents for oxidative dehydroaromatization of substrate 5a by  $VsAO^{a,b}$ .

<sup>*a*</sup>*Reaction conditions* (5.0 mL): **5a** (10 mM), various co-solvent, freeze-dried whole cells (16 g/L), 5.0 mL KH<sub>2</sub>PO<sub>4</sub>-K<sub>2</sub>HPO<sub>4</sub> buffer (pH 7.5, 50 mM). Reactions were incubated at 30°C and shaken at 200 rpm for 24 h. <sup>*b*</sup>The yield was determined by HPLC. Yield =  $[n(mol)_{generated product}/n(mol)_{initial substrate}] \times 100\%$ .

### **Characterization and NMR of products**

<sup>1</sup>H and <sup>13</sup>C NMR spectra of the products recorded in CDCl<sub>3</sub> or DMSO-*d*<sub>6</sub>

2-Methylquinoline 1b

1b

<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.03 (d, *J* = 8.3 Hz, 2H), 7.76 (d, *J* = 8.1 Hz, 1H), 7.68 (t, *J* = 7.7 Hz, 1H), 7.47 (t, *J* = 7.6 Hz, 1H), 7.28 (s, 1H), 2.75 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 159.05, 147.98, 136.23, 129.49, 128.73, 127.56, 126.58, 125.74, 122.05, 25.44.

6-Fluoro-2-Methylquinoline 2b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*)  $\delta$  8.02 – 7.90 (m, 2H), 7.41 (td, *J* = 8.8, 2.7 Hz, 1H), 7.37 – 7.30 (m, 1H), 7.24 (s, 1H), 2.69 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*)  $\delta$  160.12 (d, *J* = 246.8 Hz), 158.40 (d, *J* = 2.7 Hz), 145.01, 135.70 (d, *J* = 5.1 Hz), 131.11 (d, *J* = 9.1 Hz), 127.12 (d, *J* = 9.8 Hz), 122.88, 119.62 (d, *J* = 25.6 Hz), 110.64 (d, *J* = 21.4 Hz), 25.29.

6-Bromo-2-Methylquinoline **3b** 



<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 7.92 – 7.82 (m, 3H), 7.70 (dd, *J* = 8.9, 2.2 Hz, 1H), 7.24 (s, 1H), 2.70 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 159.63, 146.54, 135.30, 132.98, 130.52, 129.65, 127.77, 122.97, 119.54, 25.46.

Quinoline **4b** 



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.91 (d, *J* = 4.3 Hz, 1H), 8.12 (t, *J* = 9.0 Hz, 2H), 7.80 (d, *J* = 8.2 Hz, 1H), 7.70 (t, *J* = 7.7 Hz, 1H), 7.53 (t, *J* = 7.6 Hz, 1H), 7.37 (dd, *J* = 8.4, 4.2 Hz, 1H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 150.44, 148.38, 136.05, 129.54, 129.48, 128.35, 127.83, 126.57, 121.10.

6-Fluoroquinoline 5b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*)  $\delta$  8.73 (d, J = 4.2 Hz, 1H), 7.96 (dd, J = 9.2, 6.6 Hz, 2H), 7.33 (td, J = 8.8, 2.7 Hz, 1H), 7.30 – 7.21 (m, 2H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*)  $\delta$  160.44 (d, J = 248.4 Hz), 149.67 (d, J = 2.7 Hz), 145.39, 135.47 (d, J = 5.4 Hz), 131.98 (d, J = 9.2 Hz), 128.94 (d, J = 10.2 Hz), 121.80, 119.79 (d, J = 26.0 Hz), 110.73 (d, J = 21.4 Hz).

6-Bromoquinoline 6b





<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.89 (d, J = 4.2 Hz, 1H), 8.03 (d, J = 8.3 Hz, 1H), 7.95 (d, J = 8.8 Hz, 2H), 7.75 (dd, J = 9.0, 2.1 Hz, 1H), 7.39 (dd, J = 8.4, 4.2 Hz, 1H).<sup>13</sup>C NMR (151 MHz, Chloroform-*d*) δ 150.63, 146.74, 134.91, 132.84, 131.15, 129.71, 129.24, 121.79, 120.39.

6-Chloroquinoline 7b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.78 (d, *J* = 4.2 Hz, 1H), 7.92 (dd, *J* = 8.7, 4.7 Hz, 2H), 7.64 (d, *J* = 2.4 Hz, 1H), 7.52 (dd, *J* = 8.9, 2.4 Hz, 1H), 7.32 – 7.23 (m, 1H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 150.65, 146.73, 135.23, 132.43, 131.21, 130.53, 128.95, 126.51, 122.00.

6-Phenylquinoline 8b



<sup>1</sup>**H NMR** (600 MHz, Chloroform-*d*) δ 8.14 (s, 1H), 7.94 (s, 1H), 7.67 (s, 1H), 7.53 (s, 1H), 7.44 (s, 1H), 7.37 (s, 2H), 7.31 (s, 2H), 7.09 (s, 1H), 6.93 (s, 1H).<sup>13</sup>**C NMR** <sup>13</sup>**C** NMR (151 MHz, Chloroform-*d*) δ 150.21, 147.39, 145.83, 136.13, 132.18, 131.42, 129.69, 128.84, 127.63, 127.44, 127.27, 125.96, 125.33, 121.38, 120.10. 6-Methoxyquinoline **9b** 



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.74 (dd, *J* = 4.0, 1.7 Hz, 1H), 8.00 (dd, *J* = 10.6, 8.9 Hz, 2H), 7.33 (ddd, *J* = 14.1, 8.6, 3.4 Hz, 2H), 7.03 (d, *J* = 2.8 Hz, 1H), 3.89 (d, *J* = 1.4 Hz, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 157.63, 147.79, 144.35, 134.64, 130.72, 129.22, 122.15, 121.23, 105.07, 55.38.

6-Hydroxyquinoline **10b** 



<sup>1</sup>**H NMR** (400 MHz, DMSO- $d_6$ )  $\delta$  9.19 (s, 1H), 7.75 (d, J = 4.1 Hz, 1H), 7.22 (d, J = 8.3 Hz, 1H), 6.96 (d, J = 9.0 Hz, 1H), 6.52 – 6.39 (m, 2H), 6.26 (d, J = 2.6 Hz, 1H).<sup>13</sup>**C NMR** (151 MHz, DMSO- $d_6$ )  $\delta$  155.43, 147.02, 143.01, 134.03, 130.31, 129.25, 121.91, 121.31, 108.29.

3-Methylquinoline 11b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.75 (d, *J* = 2.3 Hz, 1H), 8.06 (d, *J* = 8.4 Hz, 1H), 7.88 (s, 1H), 7.71 (d, *J* = 8.2 Hz, 1H), 7.67 – 7.58 (m, 1H), 7.53 – 7.44 (m, 1H),

2.49 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 152.37, 146.57, 134.69, 130.47, 129.17, 128.46, 128.17, 127.15, 126.56, 18.73.

4-Methylquinoline 12b



<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.82 (d, J = 4.3 Hz, 1H), 8.17 (d, J = 8.5 Hz, 1H), 8.03 (d, J = 8.4 Hz, 1H), 7.76 (t, J = 7.7 Hz, 1H), 7.61 (t, J = 7.7 Hz, 1H), 7.27 (s, 1H), 2.74 (s, 3H).<sup>13</sup>C NMR (151 MHz, Chloroform-*d*) δ 150.16, 148.03, 144.25, 130.05, 129.09, 128.30, 126.28, 123.81, 121.85, 18.61.

5-Methylquinoline 13b



13b

<sup>1</sup>**H NMR** (600 MHz, Chloroform-*d*) δ 8.89 (dd, *J* = 4.1, 1.6 Hz, 1H), 8.29 (dd, *J* = 8.5, 1.5 Hz, 1H), 7.96 (d, *J* = 8.4 Hz, 1H), 7.58 (dd, *J* = 8.4, 7.0 Hz, 1H), 7.38 (dd, *J* = 8.5, 4.1 Hz, 1H), 7.34 (d, *J* = 7.0 Hz, 1H), 2.65 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-d) δ 149.77, 148.47, 134.44, 132.28, 129.00, 127.64, 127.50, 126.87, 120.52, 18.38.

6-Methylquinoline 14b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*)  $\delta$  8.83 (d, *J* = 4.2 Hz, 1H), 8.01 (dd, *J* = 16.9, 8.4 Hz, 2H), 7.57 – 7.49 (m, 2H), 7.33 (dd, *J* = 8.3, 4.2 Hz, 1H), 2.52 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*)  $\delta$  149.44, 146.89, 136.28, 135.25, 131.65, 129.07, 128.27, 126.53, 120.98, 21.48.

7-Methylquinoline 15b

42 / 70



#### 15b

<sup>1</sup>**H NMR** (600 MHz, Chloroform-*d*) δ 8.85 (dd, J = 4.0, 1.8 Hz, 1H), 8.07 (dd, J = 8.3, 1.7 Hz, 1H), 7.88 (s, 1H), 7.68 (d, J = 8.2 Hz, 1H), 7.35 (dd, J = 8.3, 1.7 Hz, 1H), 7.29 (ddd, J = 8.2, 4.3, 1.3 Hz, 1H), 2.55 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 150.20, 148.43, 139.59, 135.55, 128.66, 128.31, 127.28, 126.24, 120.13, 21.72.

8-Methylquinoline 16b



16b

<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.98 – 8.92 (m, 1H), 8.13 (dd, *J* = 8.1, 1.8 Hz, 1H), 7.66 (d, *J* = 8.2 Hz, 1H), 7.56 (d, *J* = 7.0 Hz, 1H), 7.48 – 7.35 (m, 2H), 2.83 (s, 3H).<sup>13</sup>C NMR (151 MHz, Chloroform-*d*) δ 149.29, 147.45, 137.16, 136.34, 129.68, 128.34, 126.35, 125.93, 120.87, 18.20.

7-Methoxyquinoline 17b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.80 (dd, *J* = 4.3, 1.9 Hz, 1H), 8.04 (d, *J* = 8.2 Hz, 1H), 7.66 (d, *J* = 9.0 Hz, 1H), 7.41 (d, *J* = 2.5 Hz, 1H), 7.21 (ddd, *J* = 21.5, 8.6, 3.4 Hz, 2H), 3.93 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 160.71, 150.46, 149.91, 135.71, 128.79, 123.55, 119.81, 118.96, 107.28, 55.48.

7-Chloroquinoline 18b





<sup>1</sup>**H NMR** (600 MHz, Chloroform-*d*) δ 8.78 (d, *J* = 4.4 Hz, 1H), 7.98 (q, *J* = 3.3, 2.9 Hz, 2H), 7.59 (d, *J* = 8.7 Hz, 1H), 7.35 (dd, *J* = 8.7, 2.1 Hz, 1H), 7.29 – 7.22 (m, 1H).<sup>13</sup>**C** 

**NMR** (151 MHz, Chloroform-*d*) δ 151.33, 148.57, 135.91, 135.33, 129.06, 128.45, 127.67, 126.65, 121.29.

7-Bromoquinoline 19b



19b

<sup>1</sup>**H NMR** (600 MHz, Chloroform-*d*) δ 8.91 (dt, *J* = 3.9, 1.8 Hz, 1H), 8.30 (d, *J* = 1.9 Hz, 1H), 8.12 (dd, *J* = 8.3, 1.8 Hz, 1H), 7.68 (dd, *J* = 8.6, 1.8 Hz, 1H), 7.63 (dt, *J* = 8.7, 2.0 Hz, 1H), 7.45 – 7.36 (m, 1H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 151.38, 148.93, 136.12, 131.95, 130.31, 129.18, 127.02, 123.71, 121.53.

4-Hydroxyquinoline 20b



<sup>1</sup>**H NMR** (400 MHz, DMSO- $d_6$ )  $\delta$  10.87 (s, 1H), 7.15 (d, J = 8.0 Hz, 1H), 6.95 (d, J = 7.2 Hz, 1H), 6.62 (dd, J = 30.0, 7.9 Hz, 2H), 6.34 (t, J = 7.5 Hz, 1H), 5.10 (d, J = 6.9 Hz, 1H).<sup>13</sup>**C NMR** (151 MHz, DMSO- $d_6$ )  $\delta$  176.87, 139.99, 139.31, 131.53, 125.79, 124.90, 122.98, 118.19, 108.65.

8-Hydroxyquinoline 21b



<sup>1</sup>**H NMR** (600 MHz, DMSO-d6)  $\delta$  9.84 (s, 1H), 8.83 (dd, J = 4.1, 1.7 Hz, 1H), 8.27 (dd, J = 8.3, 1.7 Hz, 1H), 7.50 (dd, J = 8.3, 4.1 Hz, 1H), 7.46 – 7.40 (m, 1H), 7.37 (dd, J = 8.2, 1.4 Hz, 1H), 7.12 (dd, J = 7.6, 1.4 Hz, 1H).<sup>13</sup>**C NMR** (151 MHz, DMSO-d6)  $\delta$  153.26, 148.05, 138.47, 135.97, 128.74, 127.44, 121.74, 117.68, 111.21.

6-Bromo-4-Methylquinoline 22b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.77 – 8.71 (m, 1H), 8.10 (t, *J* = 1.7 Hz, 1H), 7.94 (dd, *J* = 9.0, 1.3 Hz, 1H), 7.73 (dt, *J* = 9.0, 1.8 Hz, 1H), 7.21 (d, *J* = 4.3 Hz, 1H), 2.63 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 150.51, 146.61, 143.77, 132.77, 131.82, 129.67, 126.45, 122.70, 120.64, 18.75.

1-Methylisoquinoline 23b



<sup>1</sup>**H NMR** (400 MHz, Chloroform-*d*) δ 8.37 (d, *J* = 5.8 Hz, 1H), 8.08 (d, *J* = 8.4 Hz, 1H), 7.77 (d, *J* = 8.2 Hz, 1H), 7.69 – 7.61 (m, 1H), 7.61 – 7.52 (m, 1H), 7.48 (d, *J* = 5.8 Hz, 1H), 2.95 (s, 3H).<sup>13</sup>**C NMR** (151 MHz, Chloroform-*d*) δ 158.57, 141.77, 135.91, 129.92, 127.51, 127.19, 127.02, 125.60, 119.28, 22.36.

1-Phenylisoquinoline 24b



<sup>1</sup>H NMR (400 MHz, Chloroform-*d*) δ 8.62 (d, J = 5.6 Hz, 1H), 8.11 (d, J = 8.5 Hz, 1H), 7.88 (d, J = 8.2 Hz, 1H), 7.74 – 7.62 (m, 4H), 7.53 (dt, J = 10.1, 6.3 Hz, 4H). <sup>13</sup>C NMR (151 MHz, Chloroform-*d*) δ 160.66, 142.08, 139.46, 136.82, 129.95, 129.85, 128.52, 128.26, 127.53, 127.10, 126.91, 126.66, 119.84

# NMR spectra of products

2-Methylquinoline 1b



<sup>6-</sup>Fluoro-2-Methylquinoline 2b



6-Bromo-2-Methylquinoline **3b** 







6-Fluoroquinoline **5b** 



6-Bromoquinoline 6b





6-Phenylquinoline 8b



6-Methoxyquinoline 9b







3-Methylquinoline 11b



4-Methylquinoline 12b



5-Methylquinoline 13b









7-Methoxyquinoline 17b



7-Chloroquinoline 18b







4-Hydroxyquinoline 20b



8-Hydroxyquinoline 21b









1-Phenylisoquinoline 24b



### Reference

1. M. Hebditch, M. A. Carballo-Amador, S. Charonis, R. Curtis, J. Warwicker, *Bioinformatics*, 2017, **33**, 3098-3100.

- 2. G. Deng, N. Wan, L. Qin, B. Cui, M. An, W. Han, Y. Chen, ChemCatChem, 2018, 10, 2374-2377.
- 3. S. Bera, A. Ber, Org. Lett., 2020, 22, 6458-6463.
- 4. G. Jaiswal, M. Subaramanian, M. K. Sahoo, E. Balaraman, ChemCatChem, 2019, 11, 2449-2457.

5. K. Sun, H. Shan, R. Ma, P. Wang, H. Neumann, G.-P. Lu, M. Beller, *Chem. Sci.*, 2022, **13**, 6865-6872.

6. C. Liao, X. Li, K. Yao, Z. Yuan, Q. Chi, Z. Zhang, ACS Sustainable Chem. Eng., 2019, 7, 13646-13654.

7. T. Liu, K. Wu, L. Wang, Z. Yu, Adv. Synth. Catal., 2019, 361, 3958-3964.