

Supporting information

Synthesis of Indole-Linked β -Cyano-Enones: A Pathway to Indolyl-2-Pyrrolones

Nurabul Mondal, Vidya Kumari, Danish Ali and Lokman H. Choudhury*

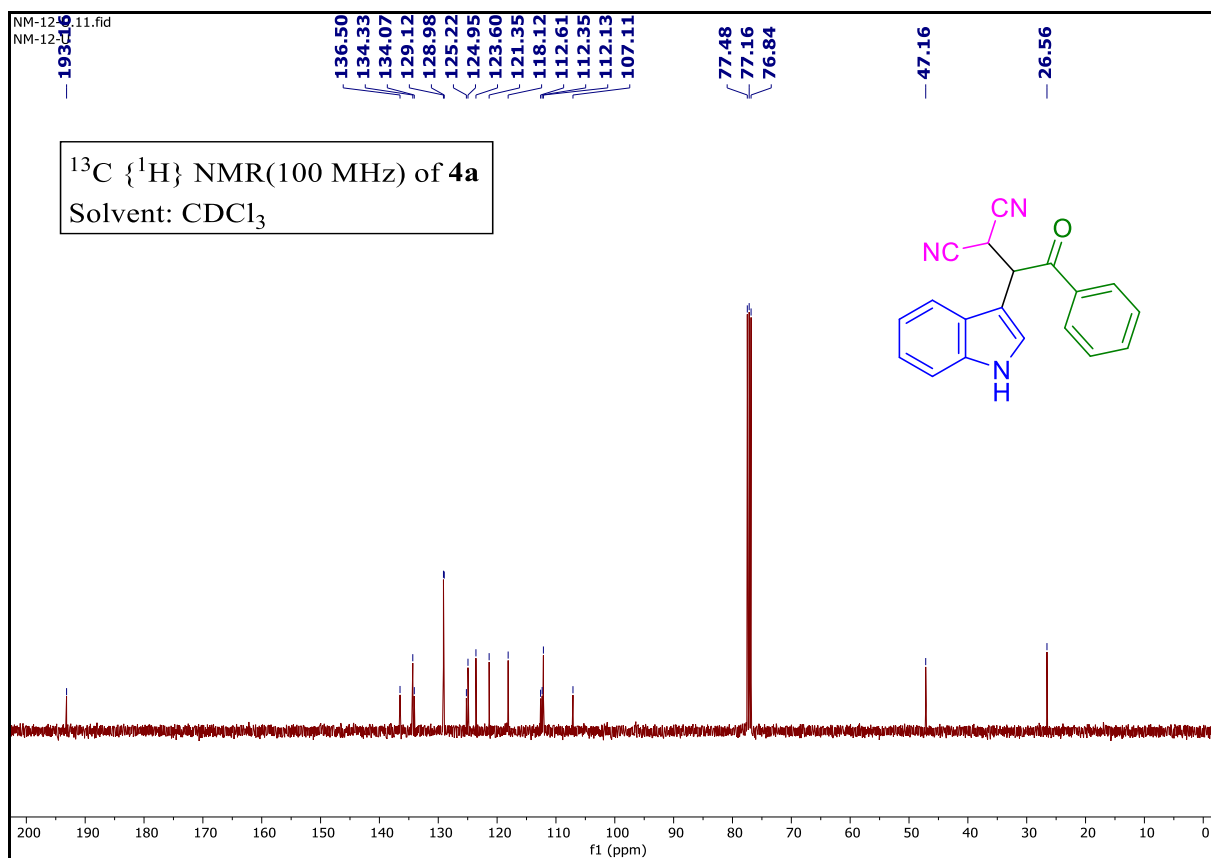
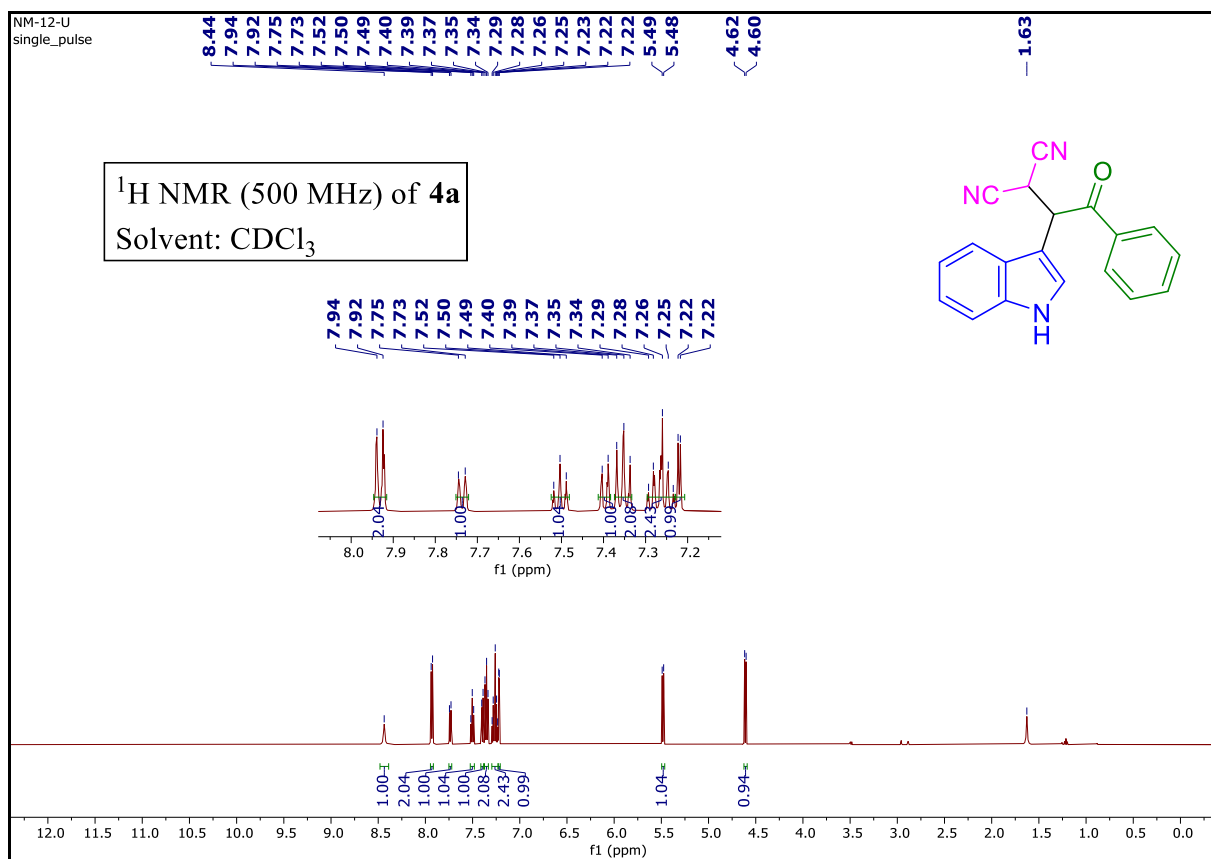
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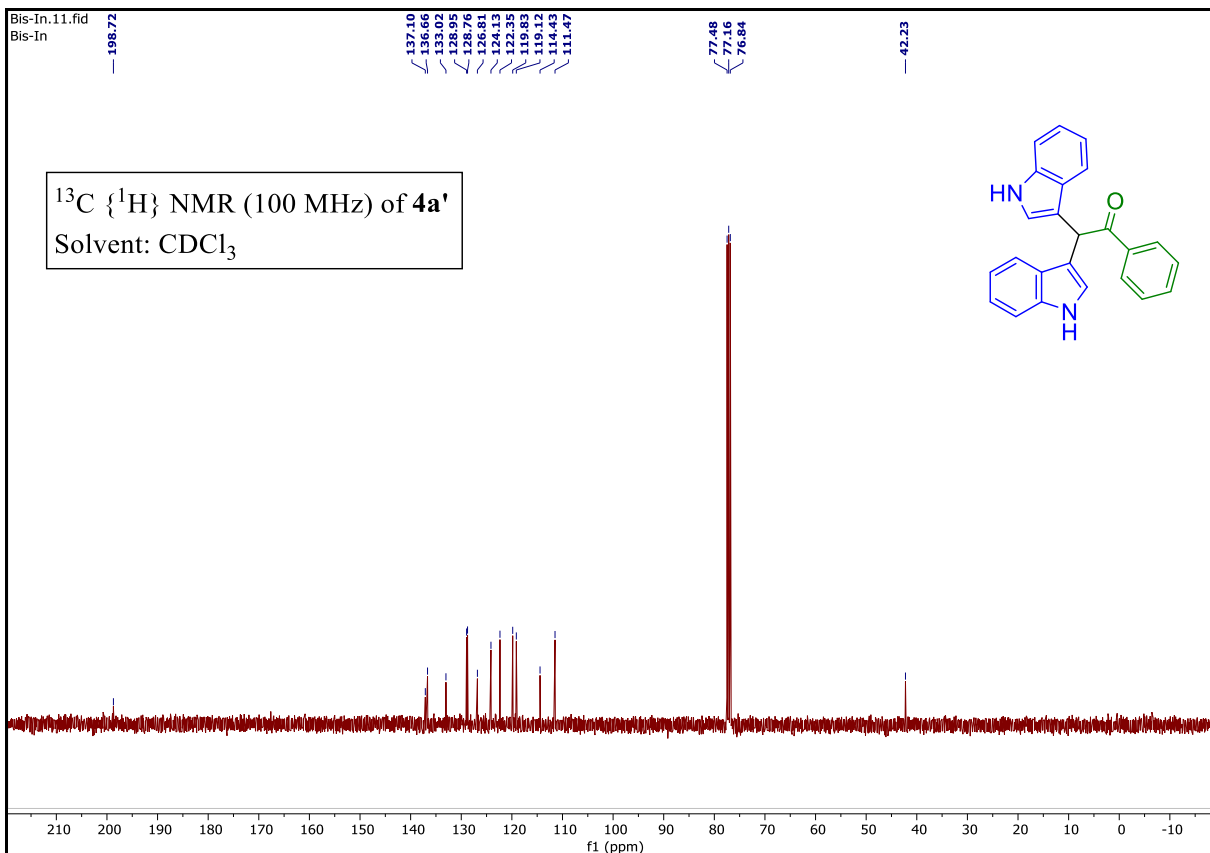
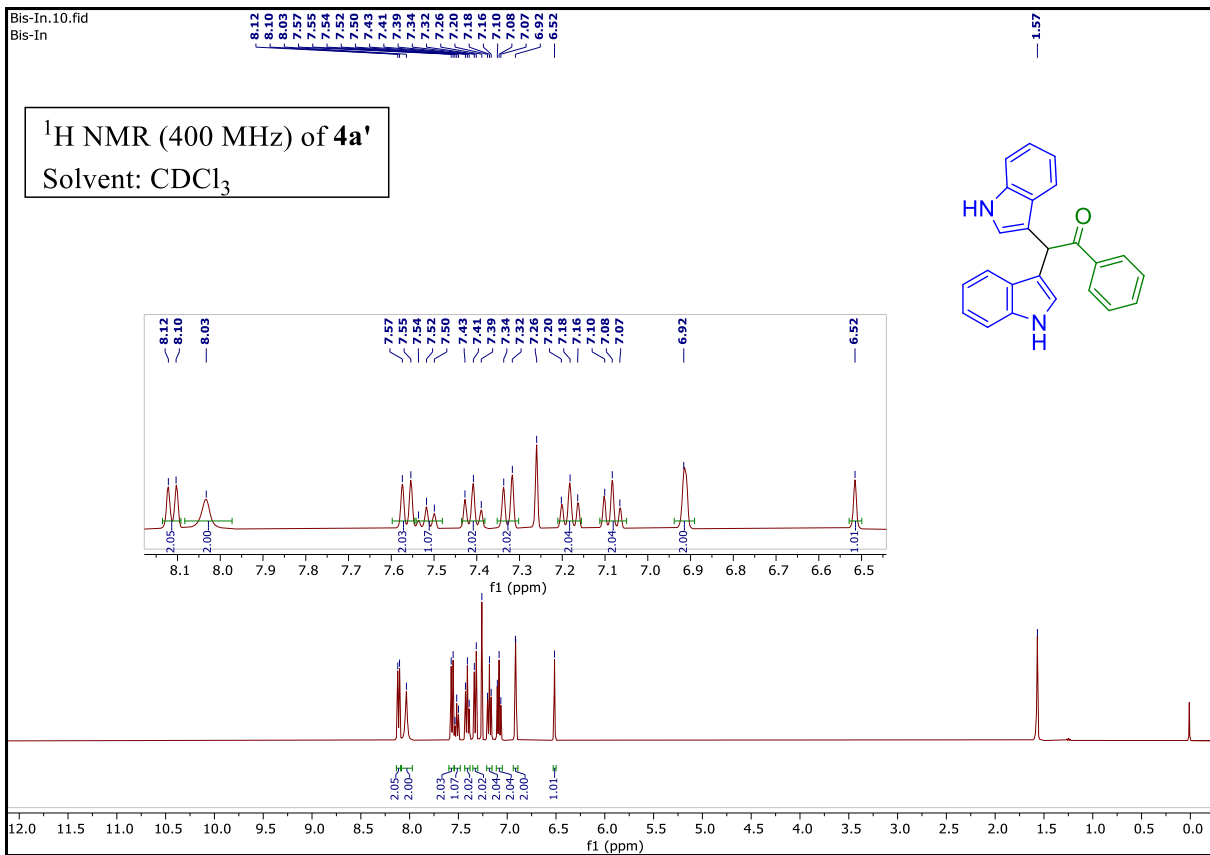
Email: lokman@iitp.ac.in; lokman.iitp@gmail.com

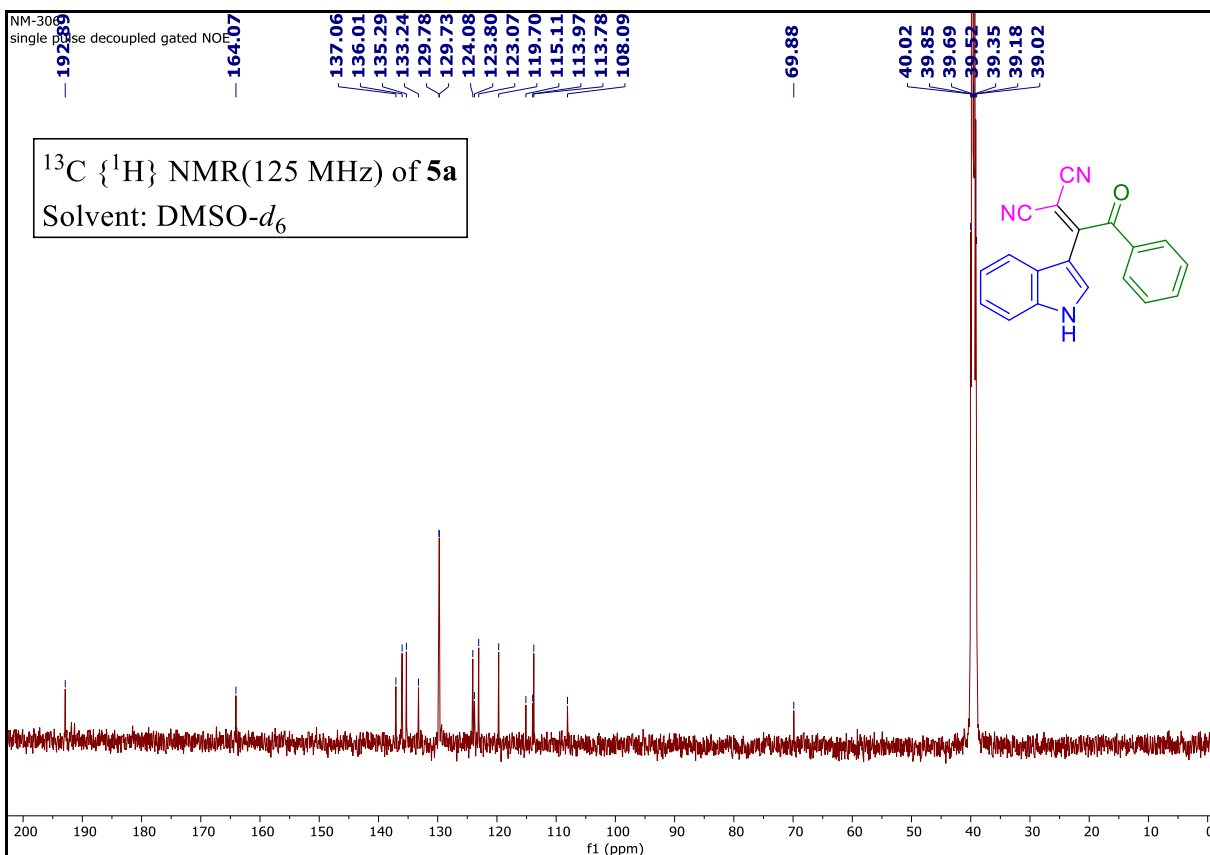
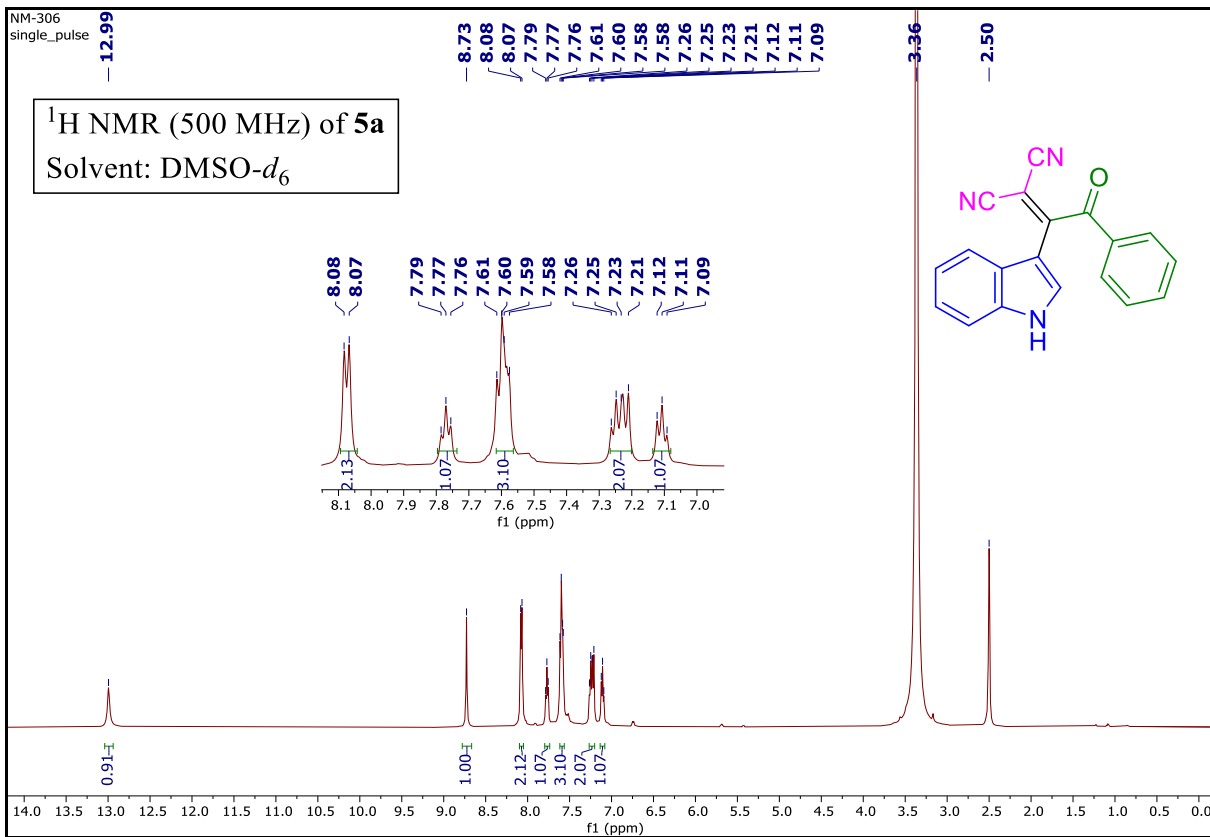
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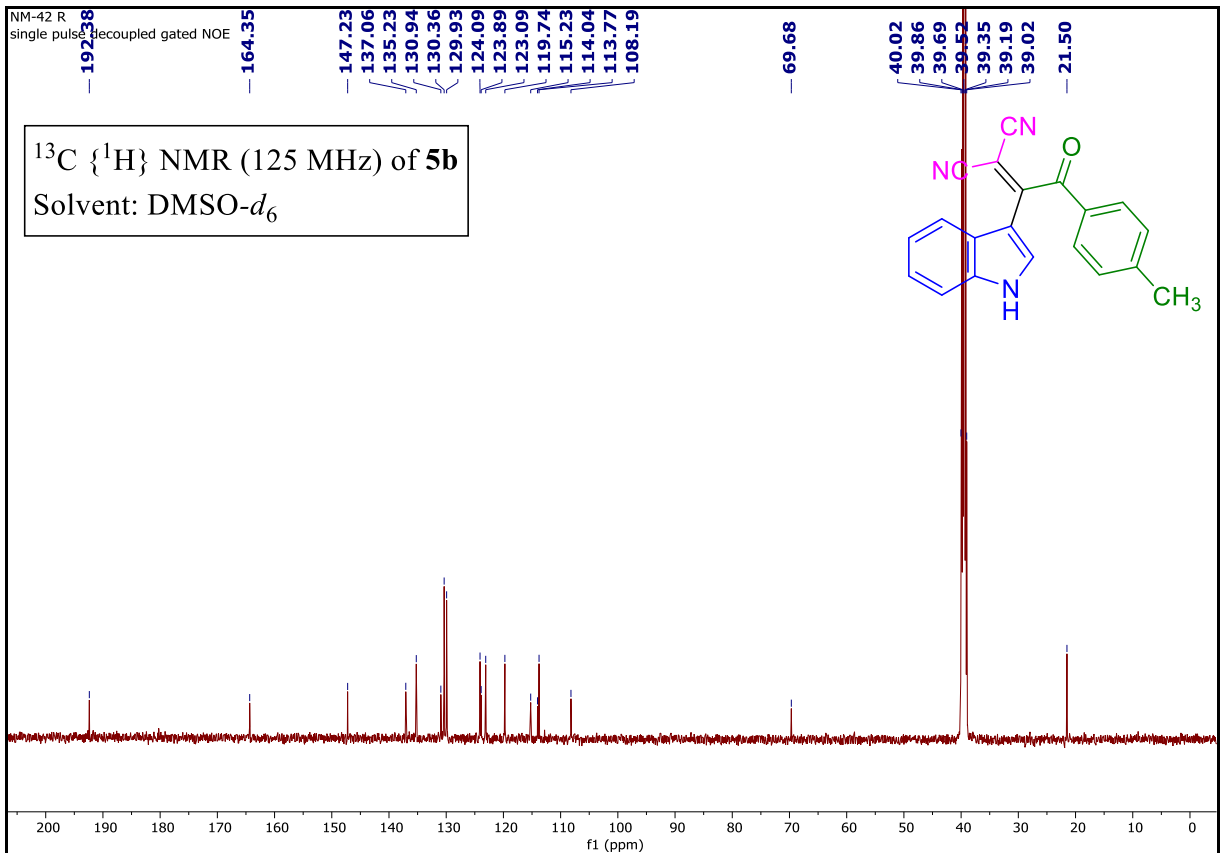
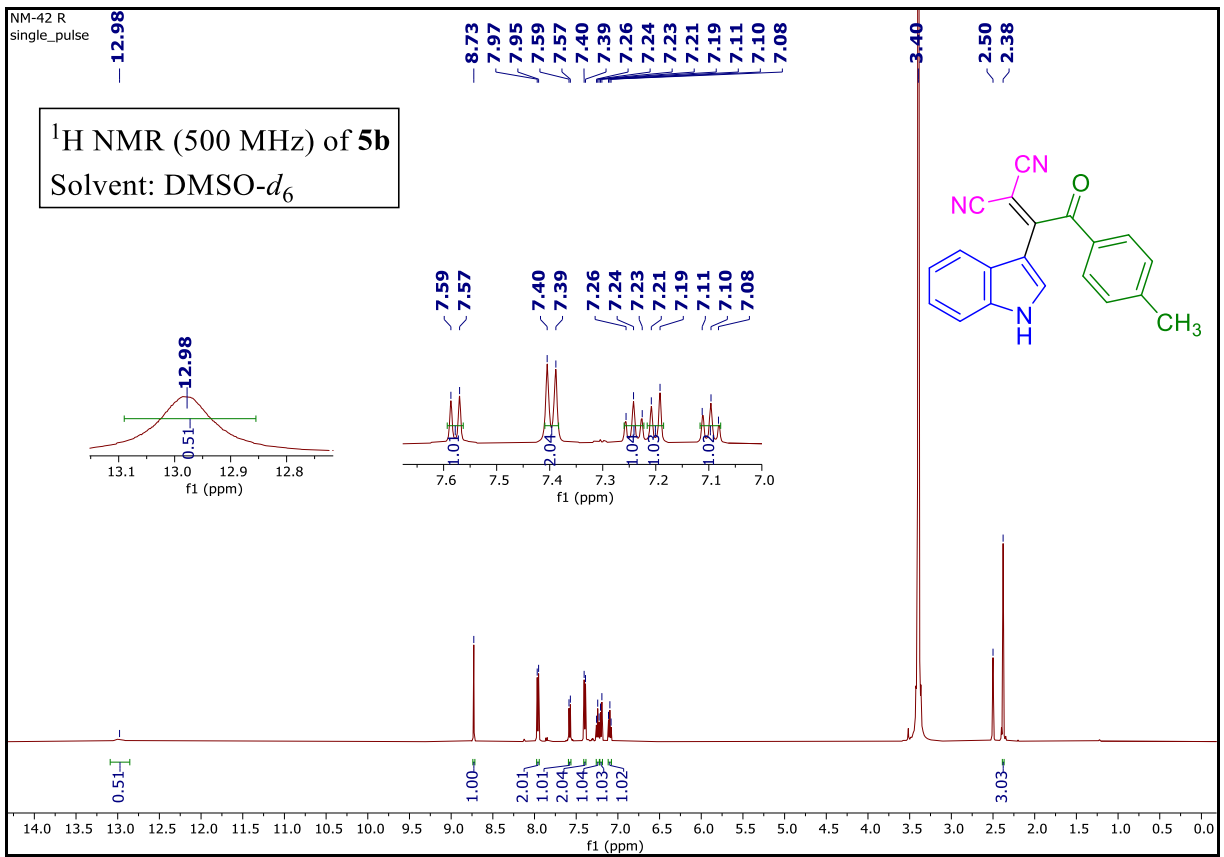
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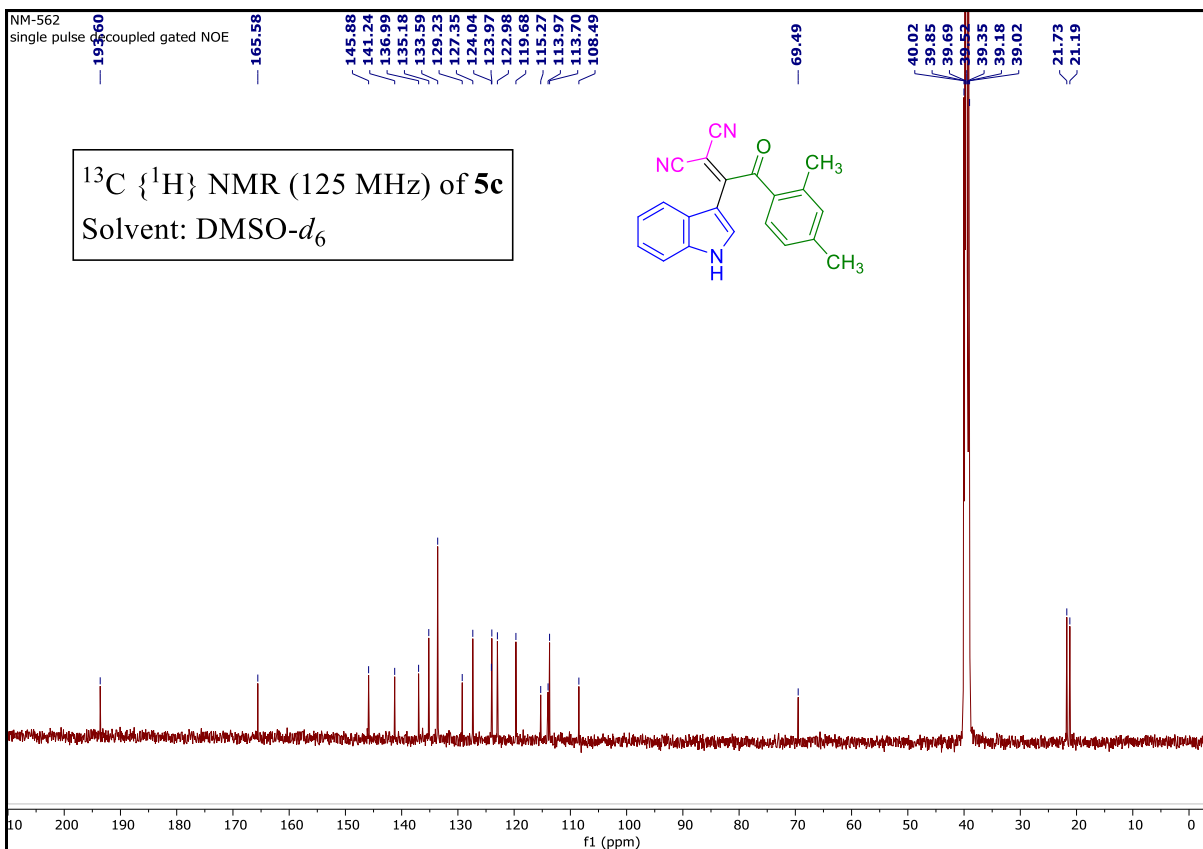
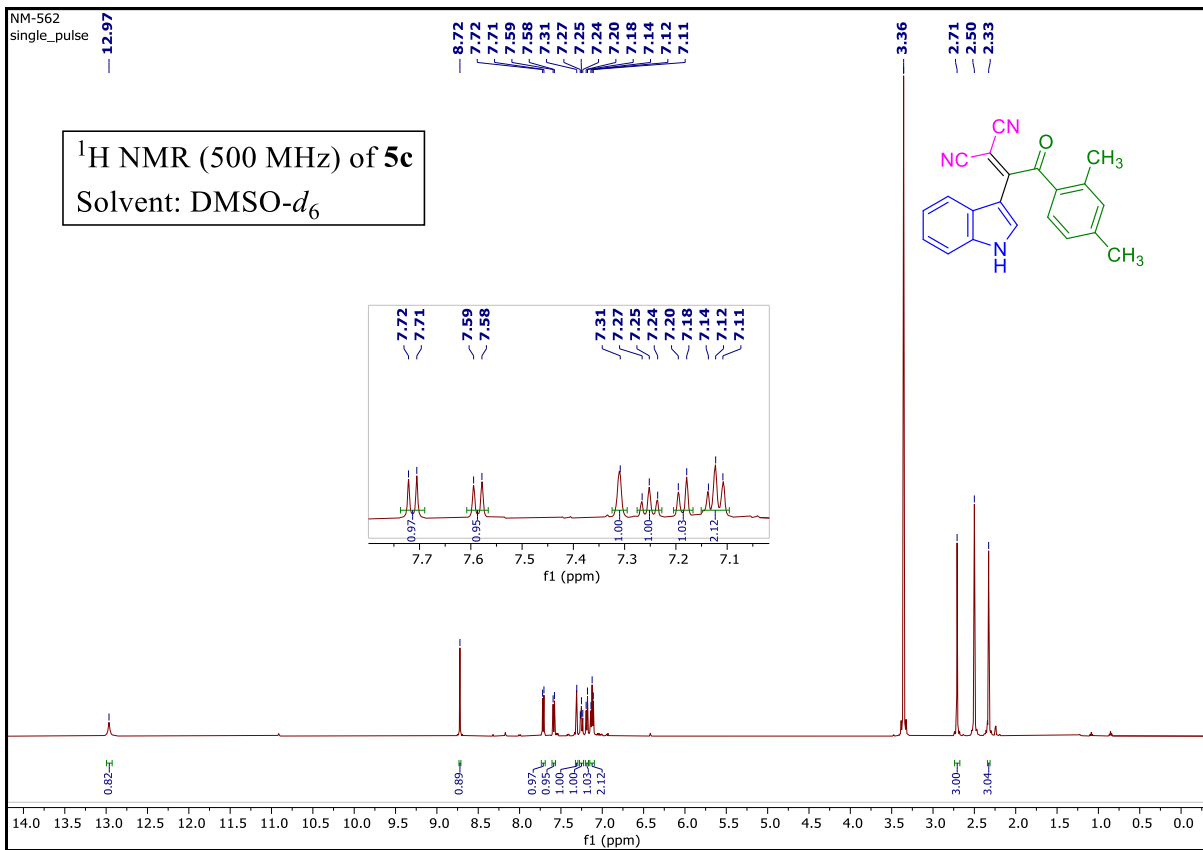
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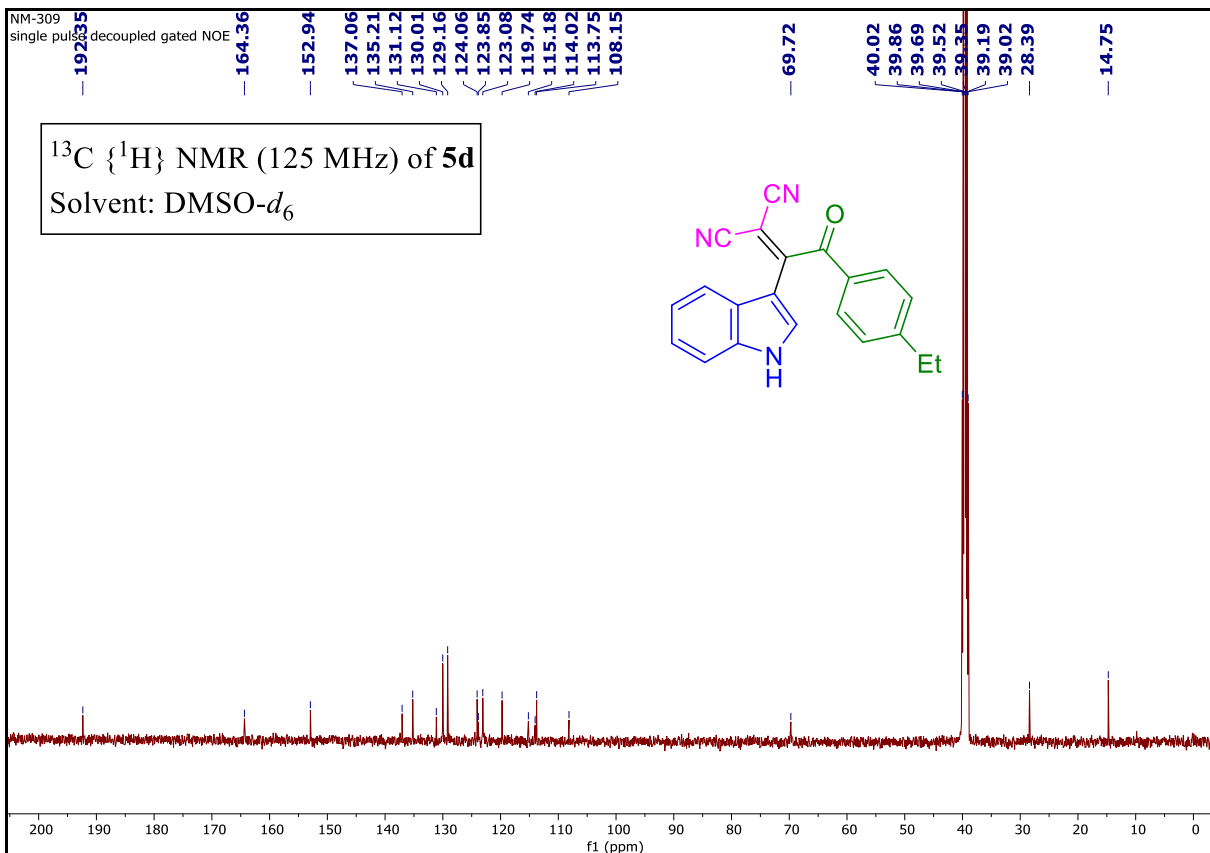
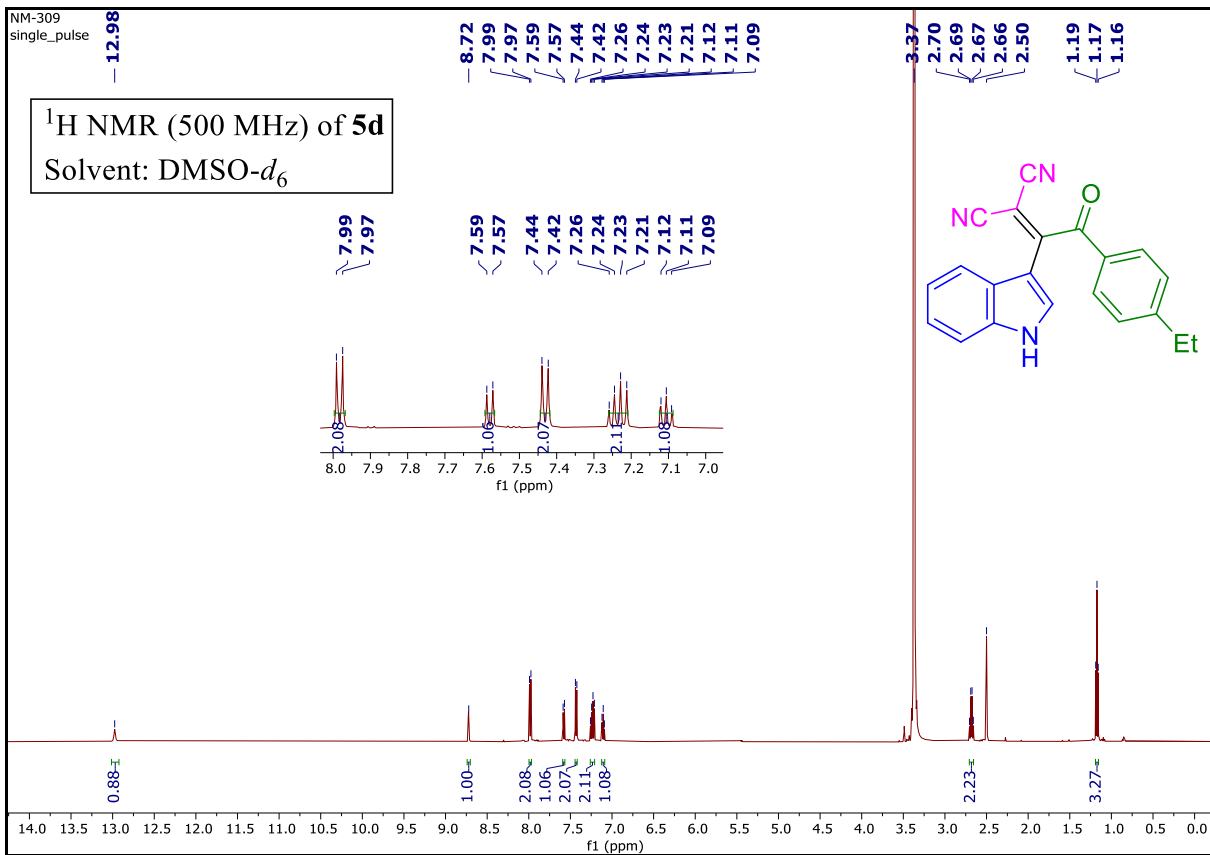


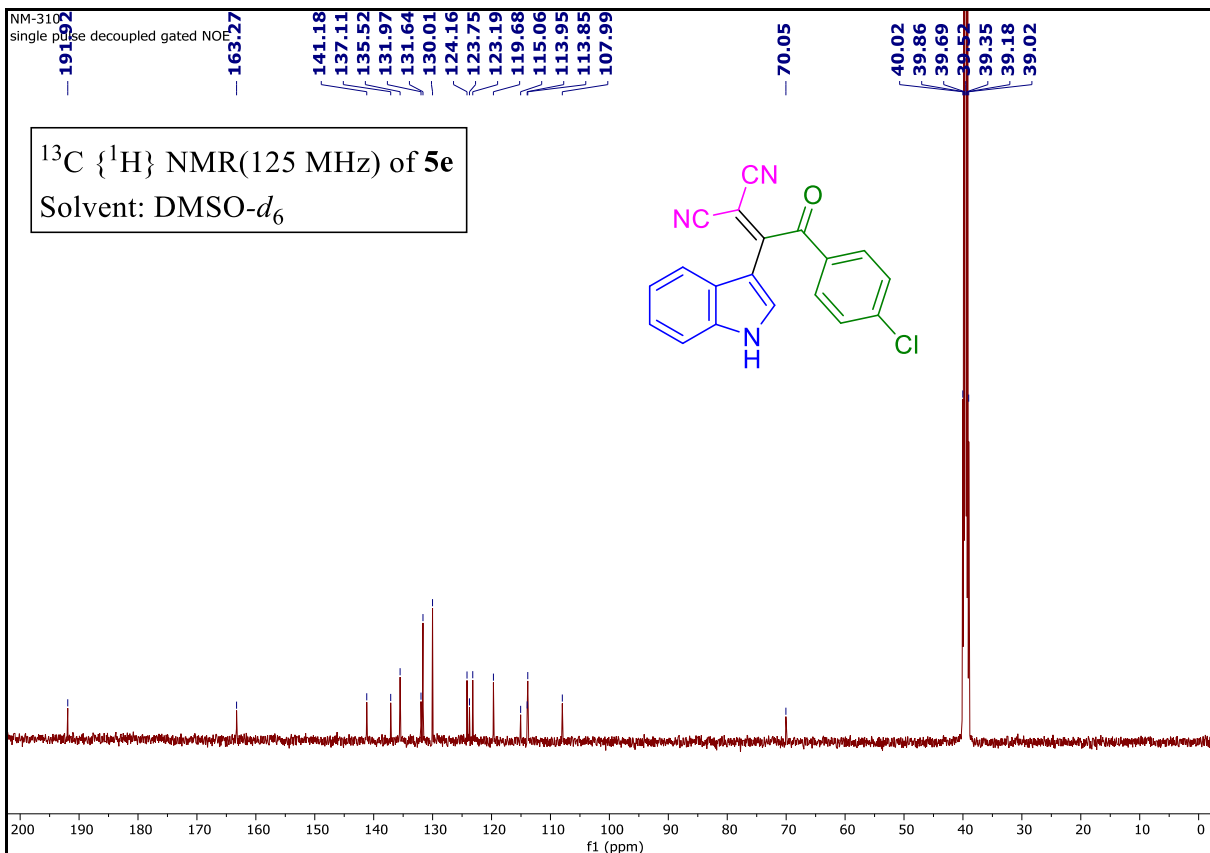
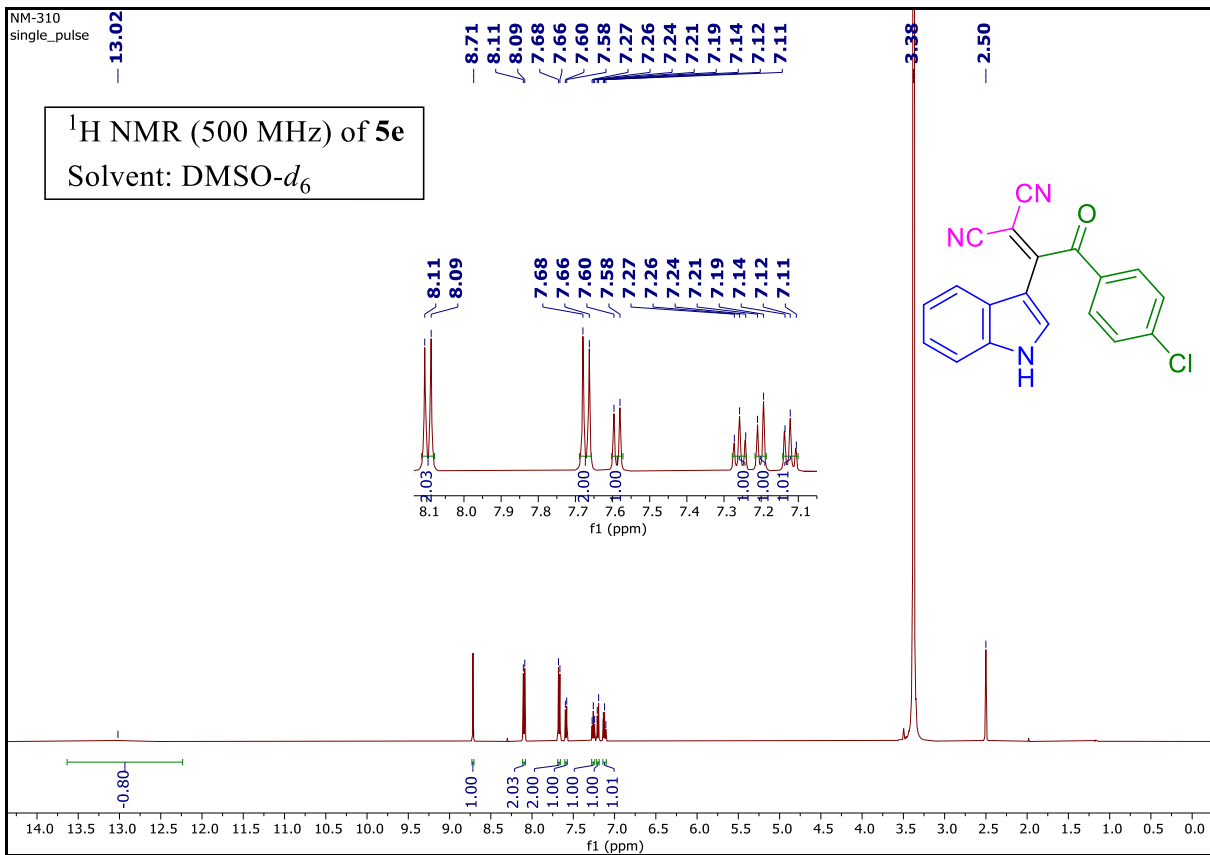


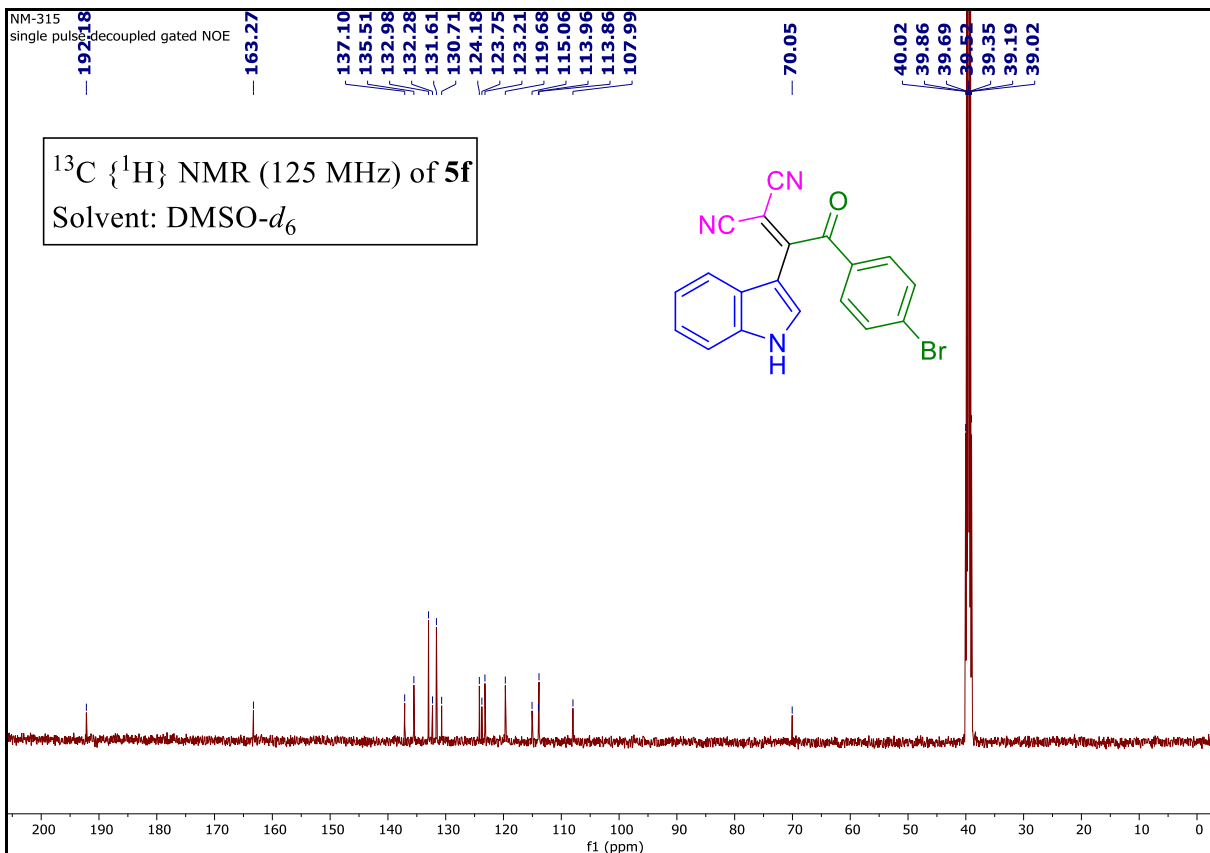
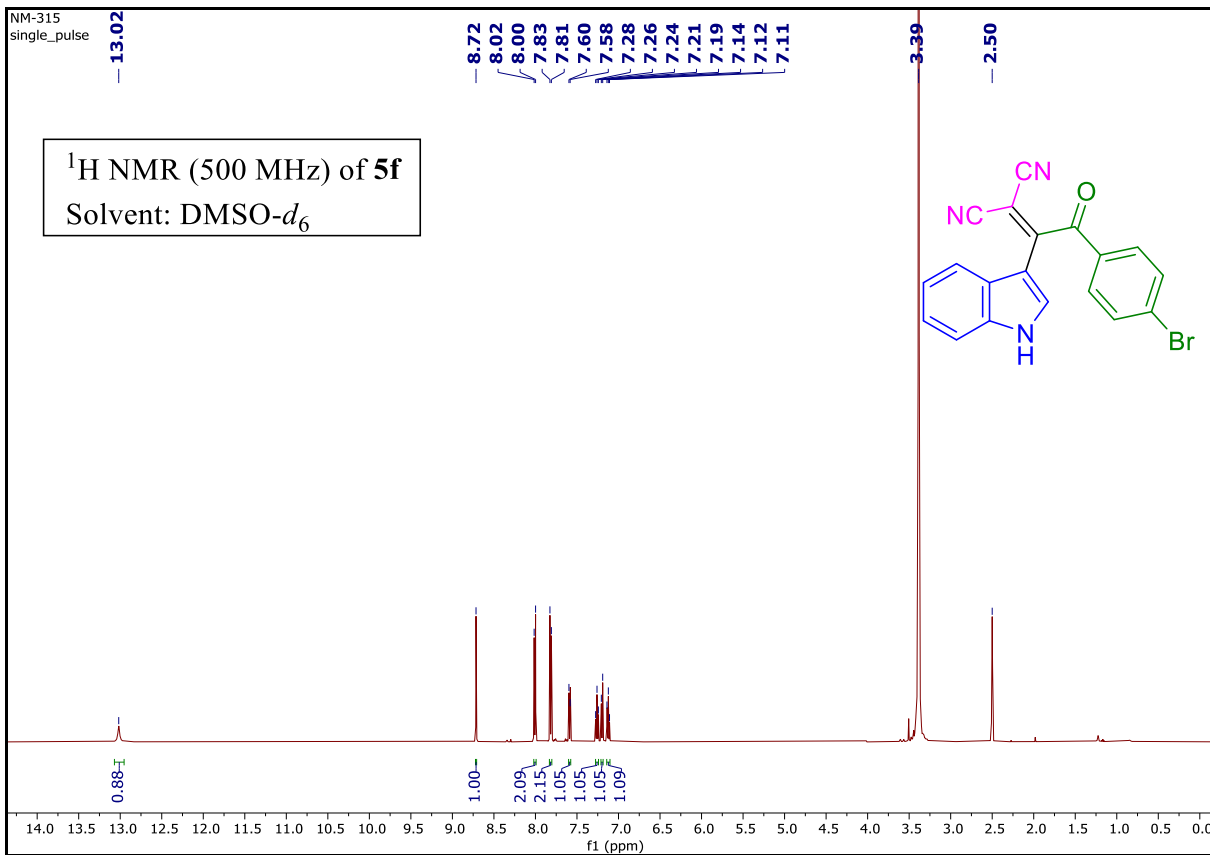


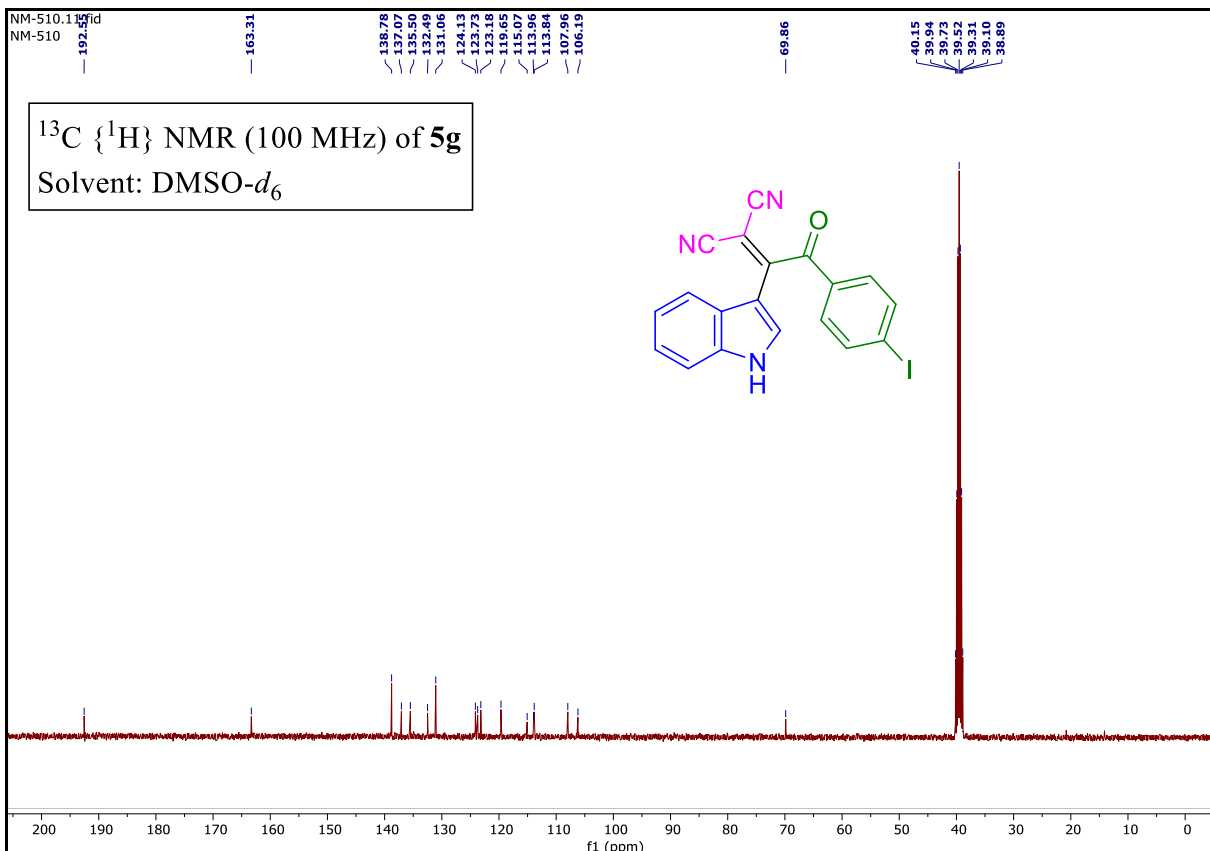
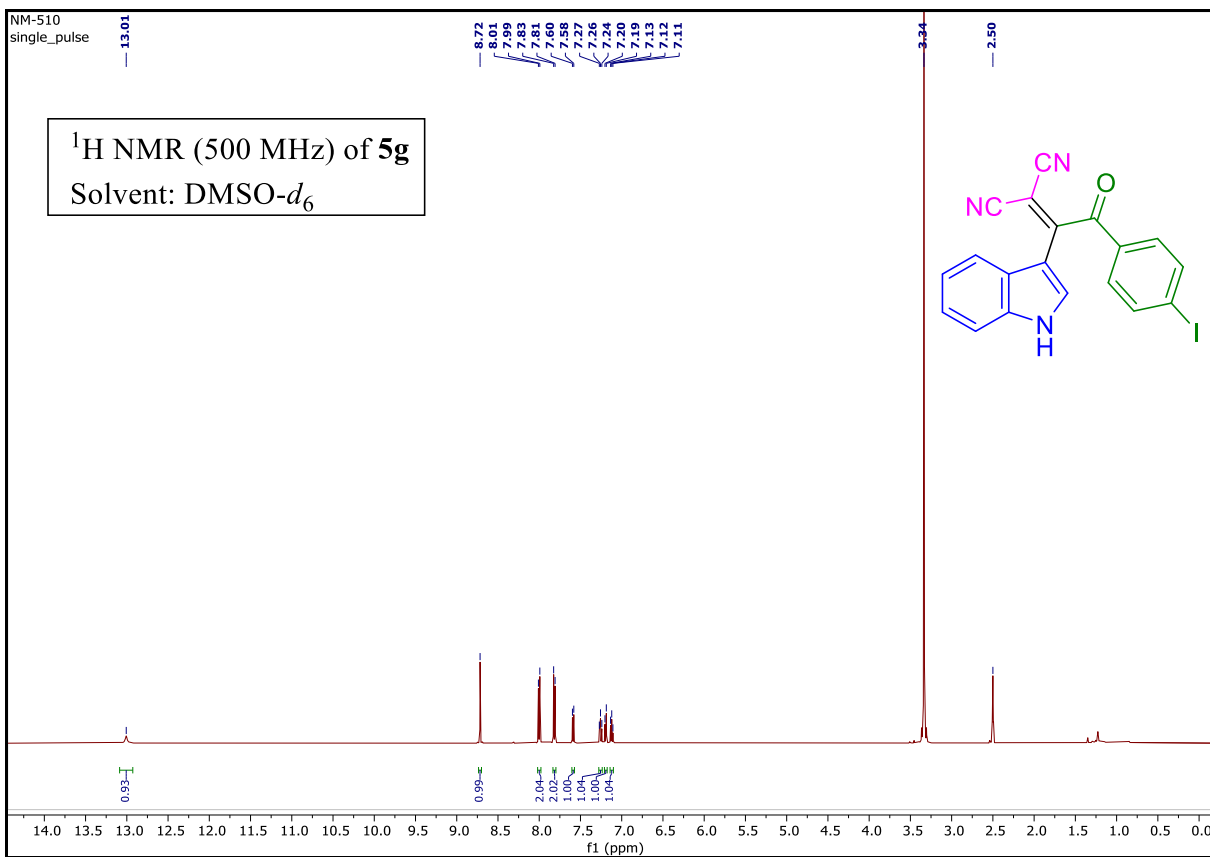


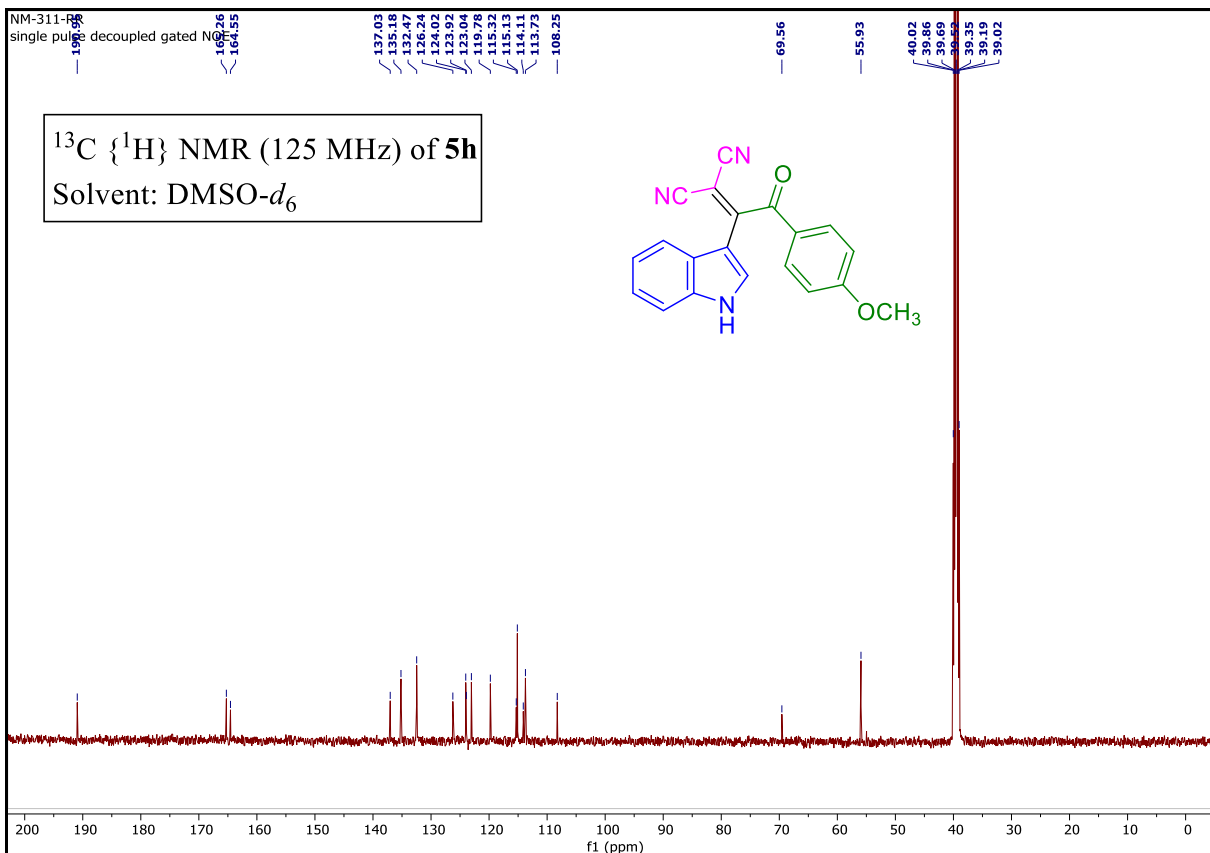
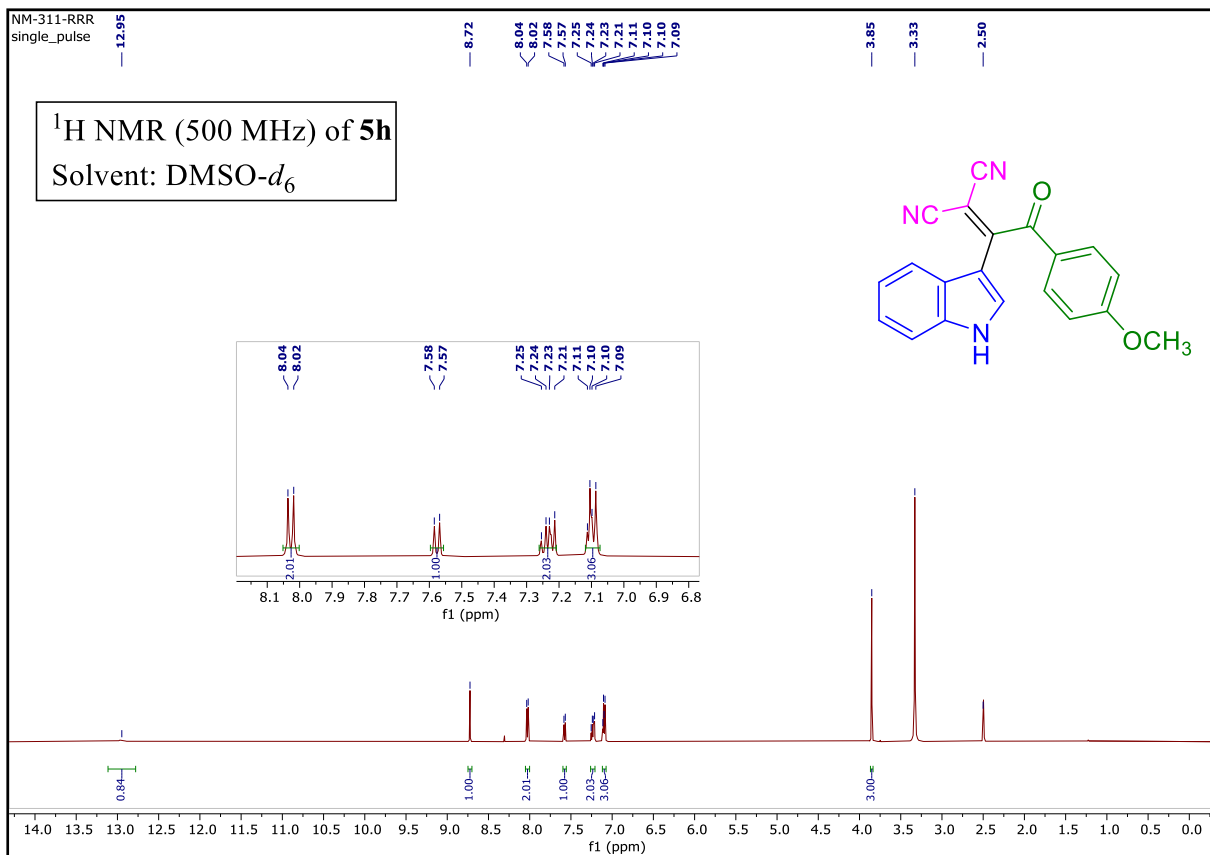


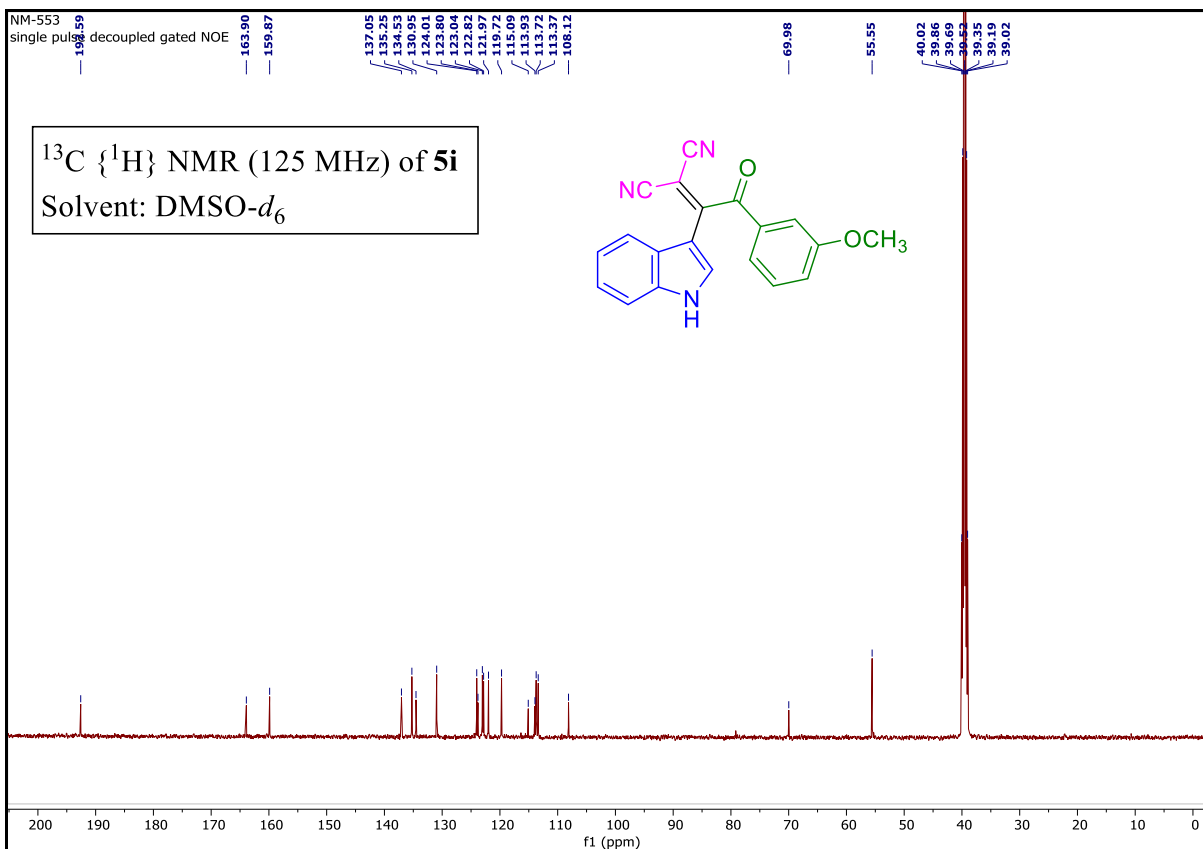
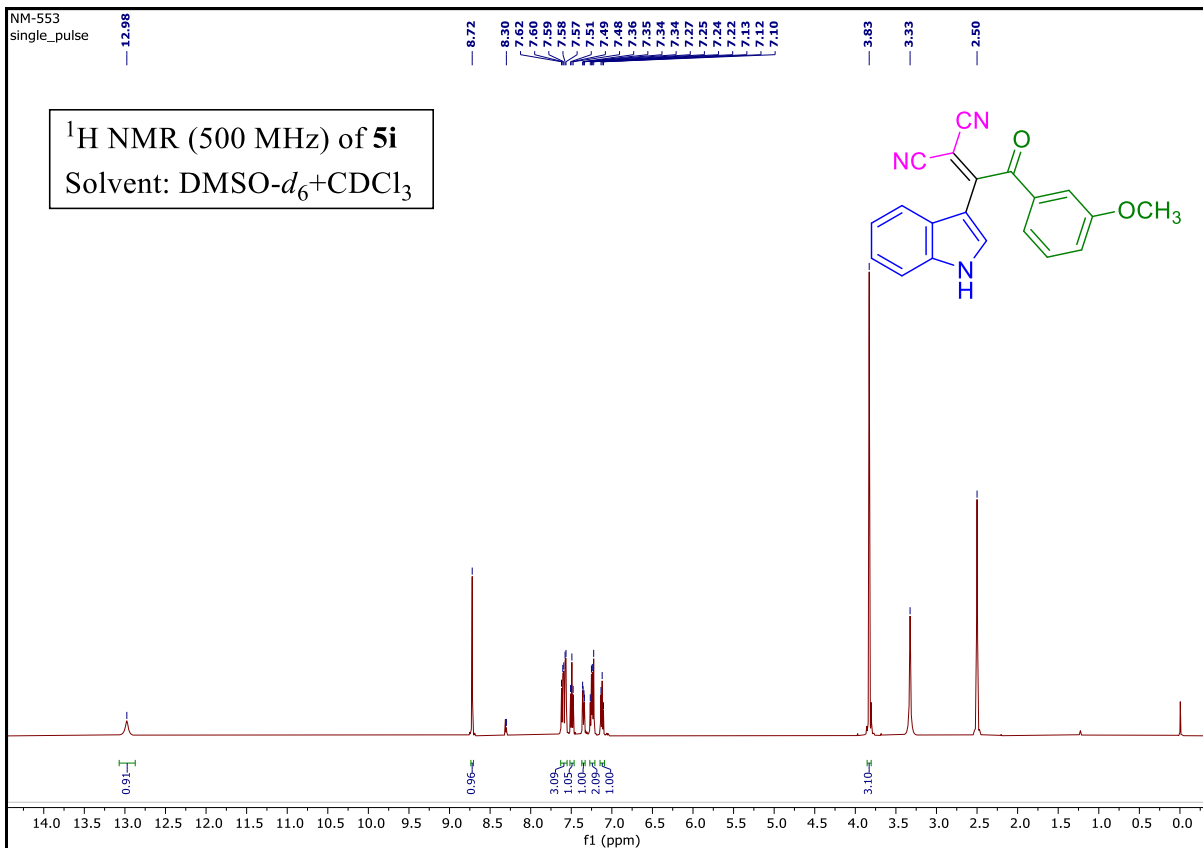


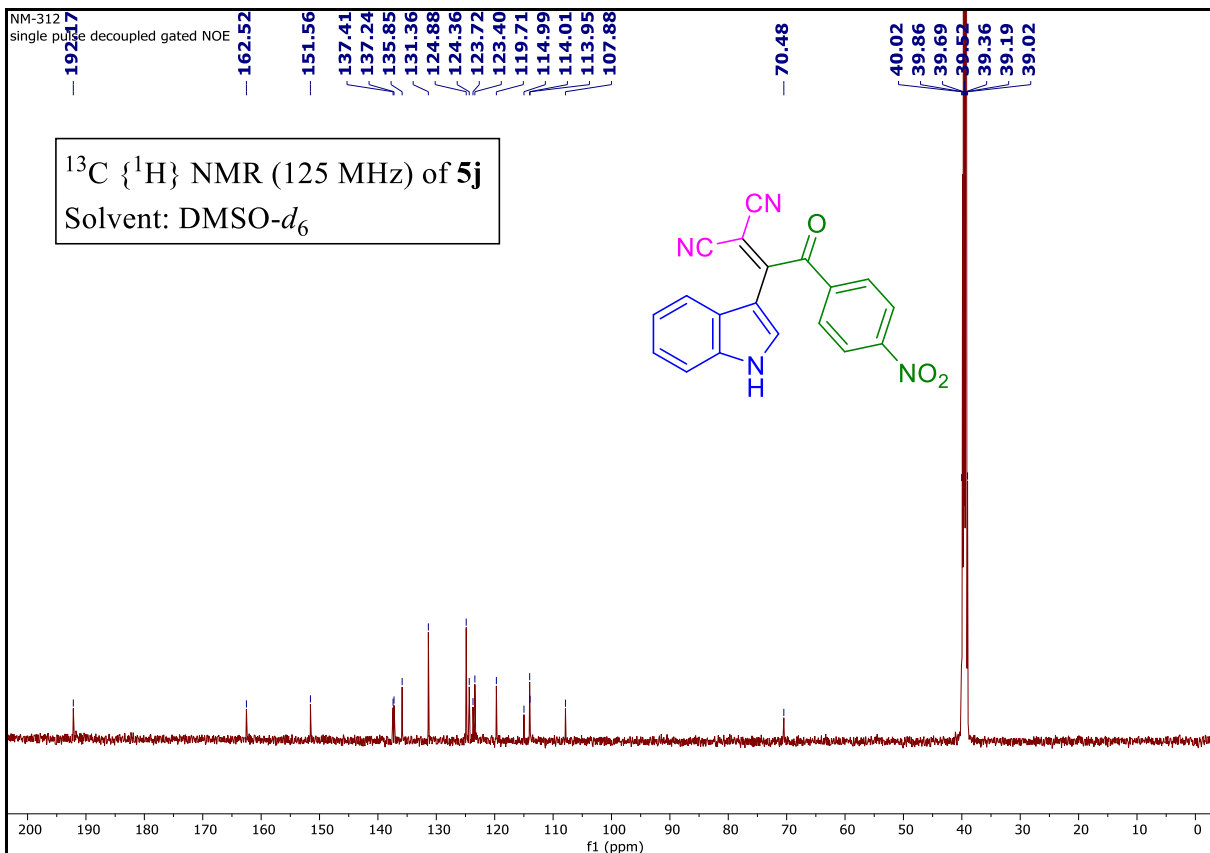
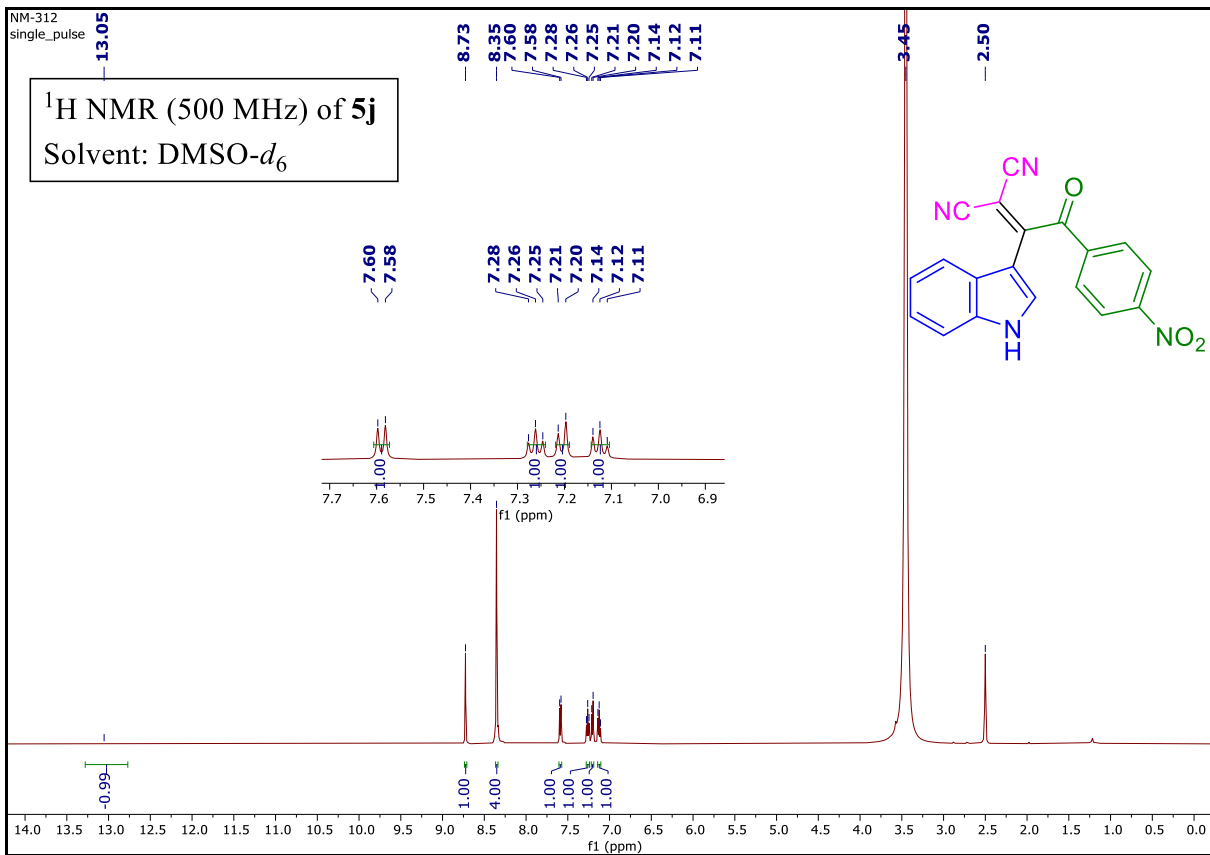


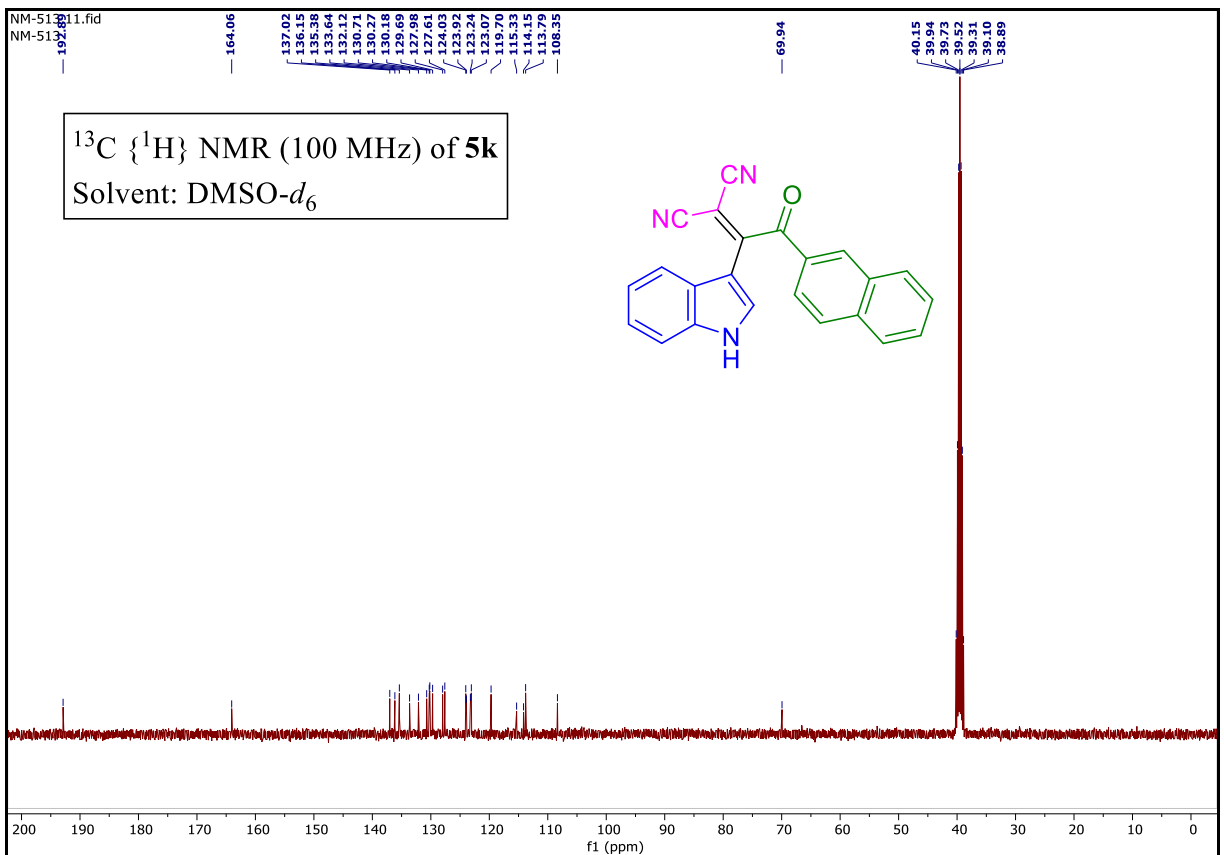
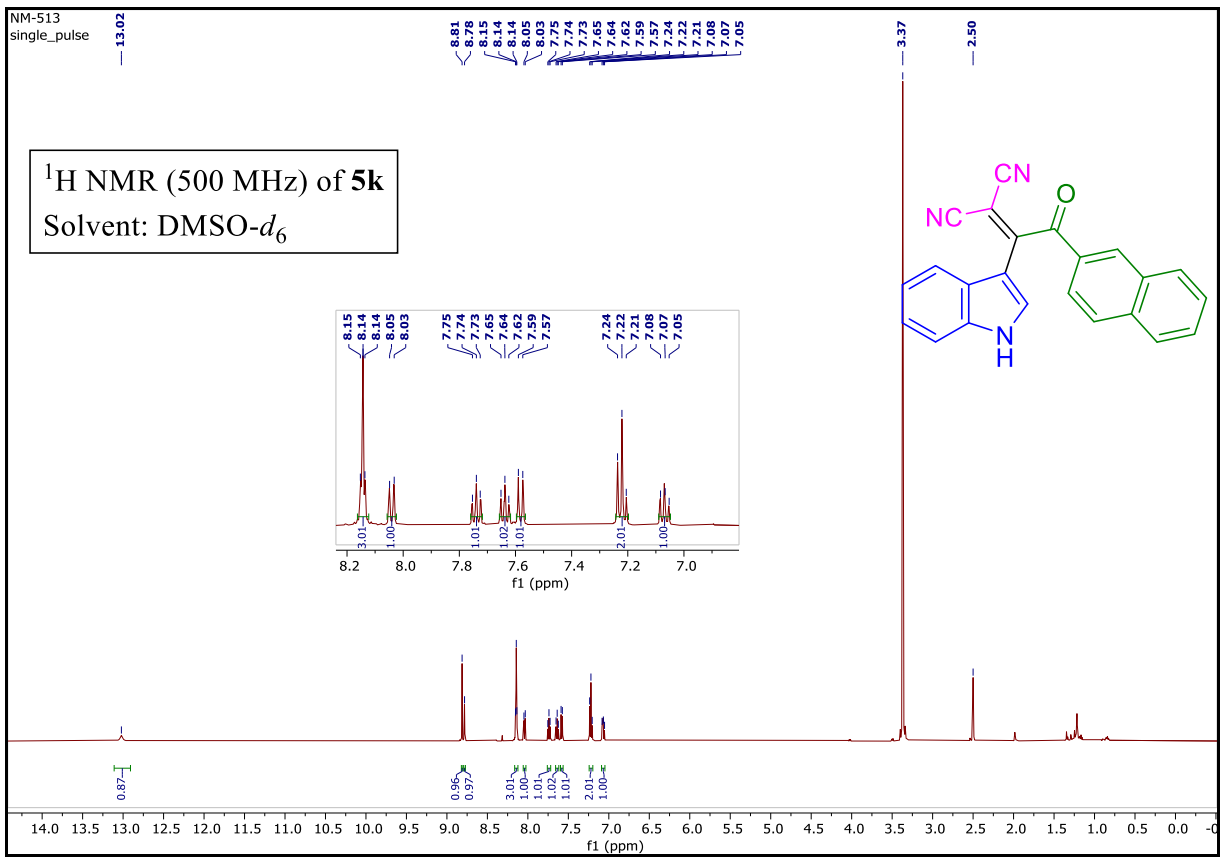


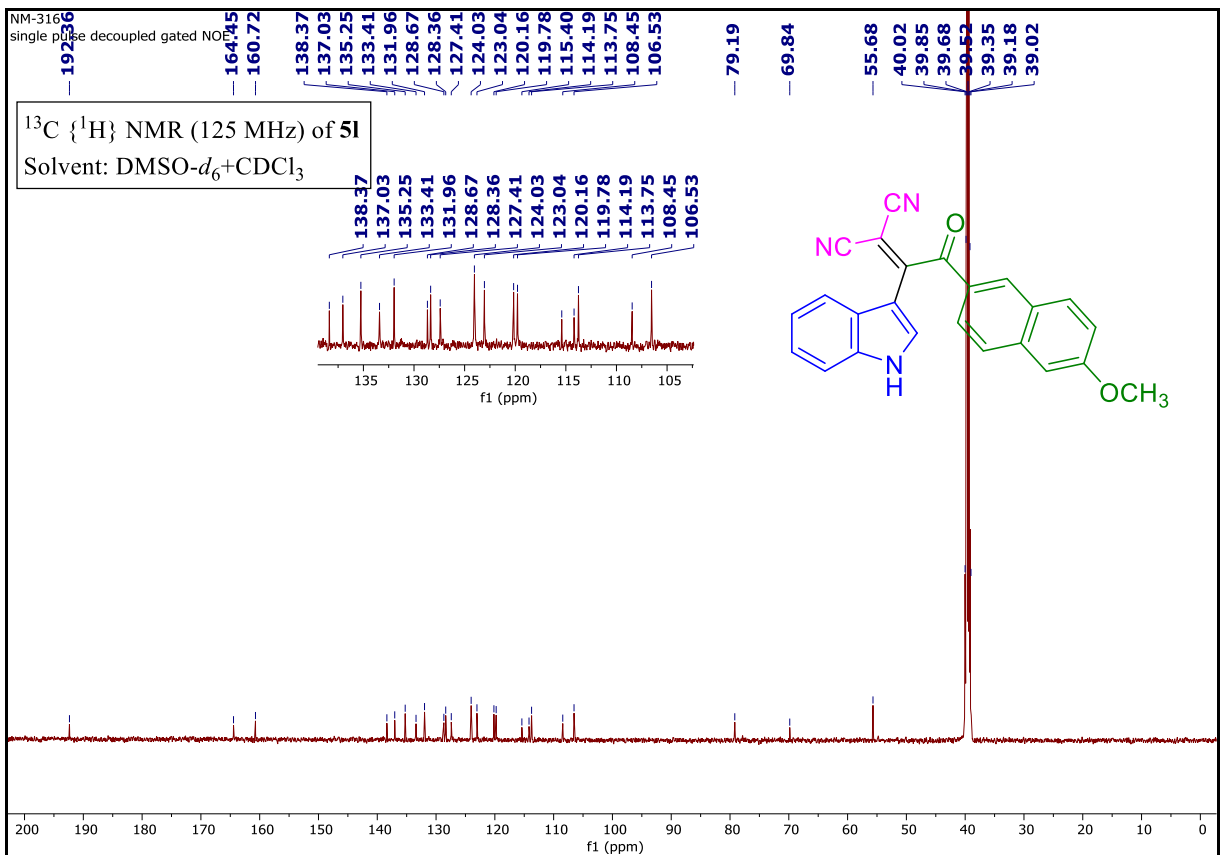
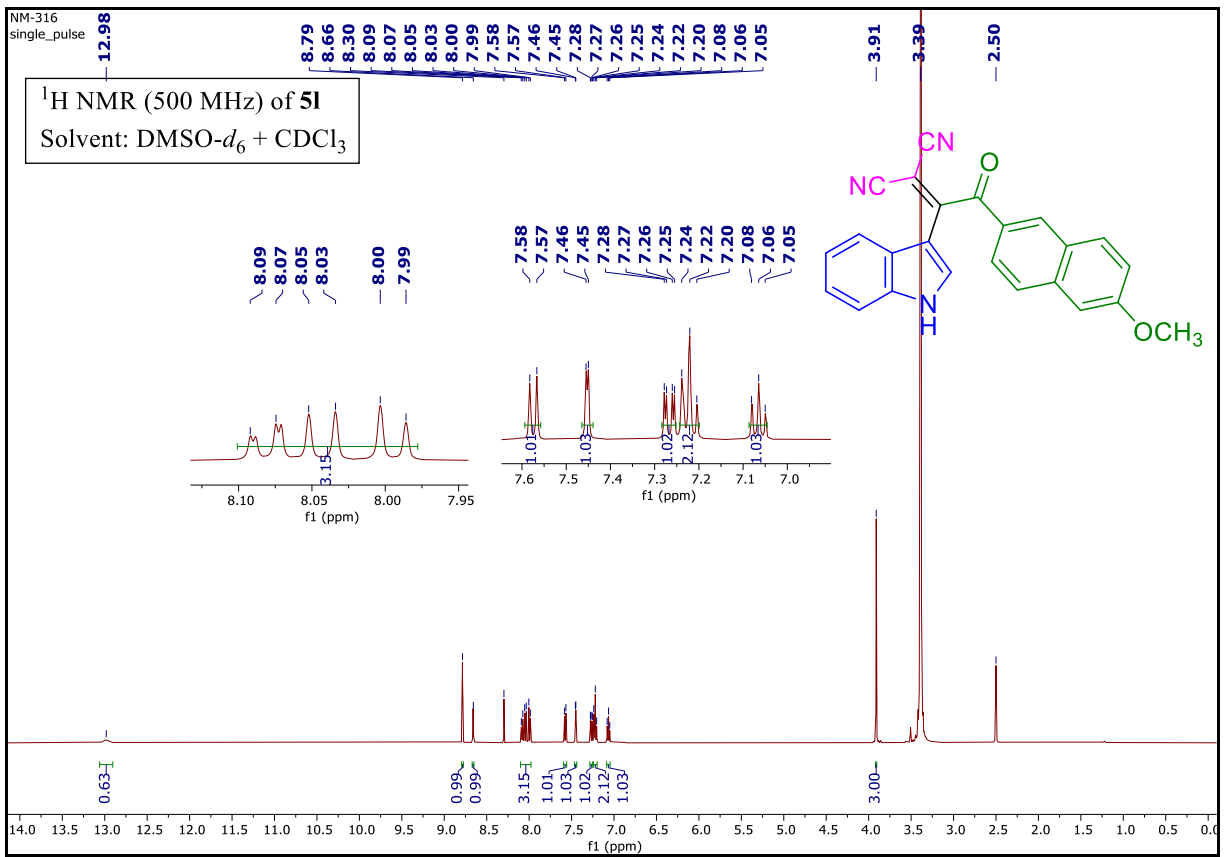


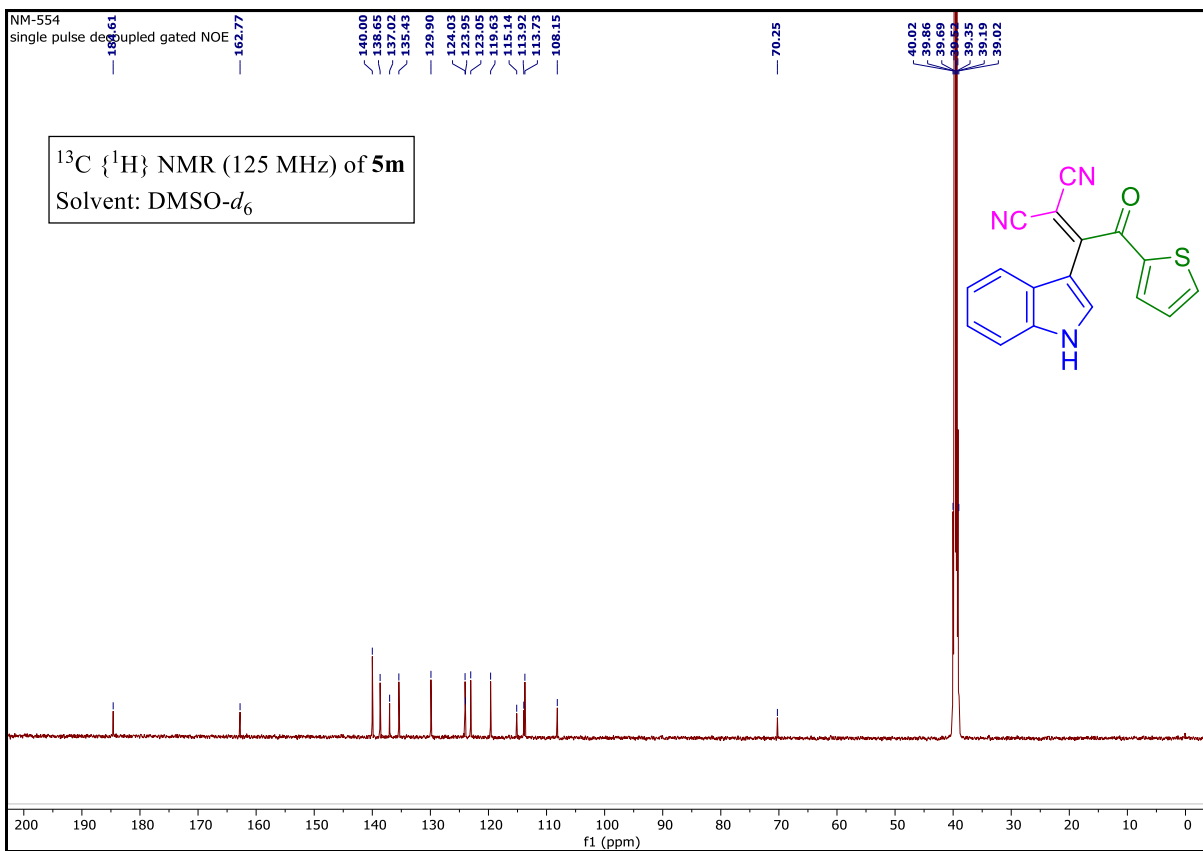
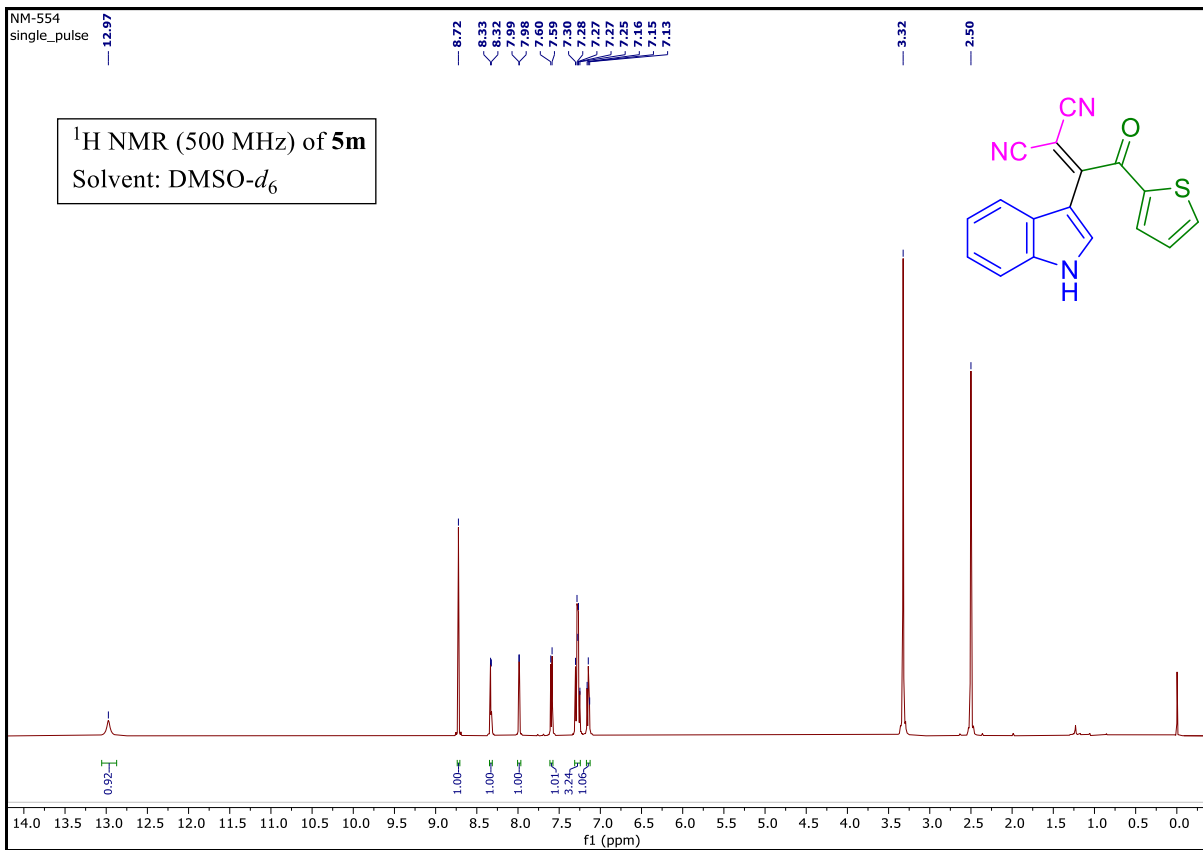


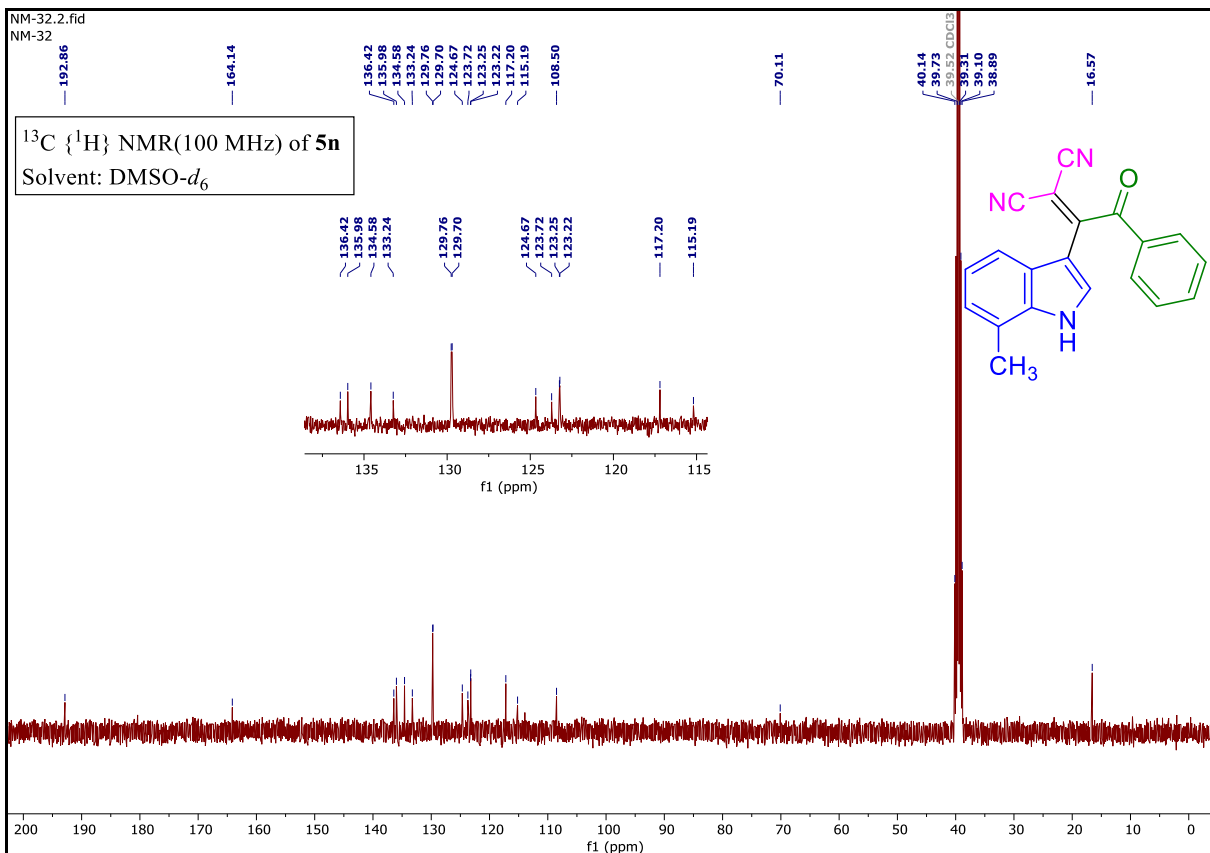
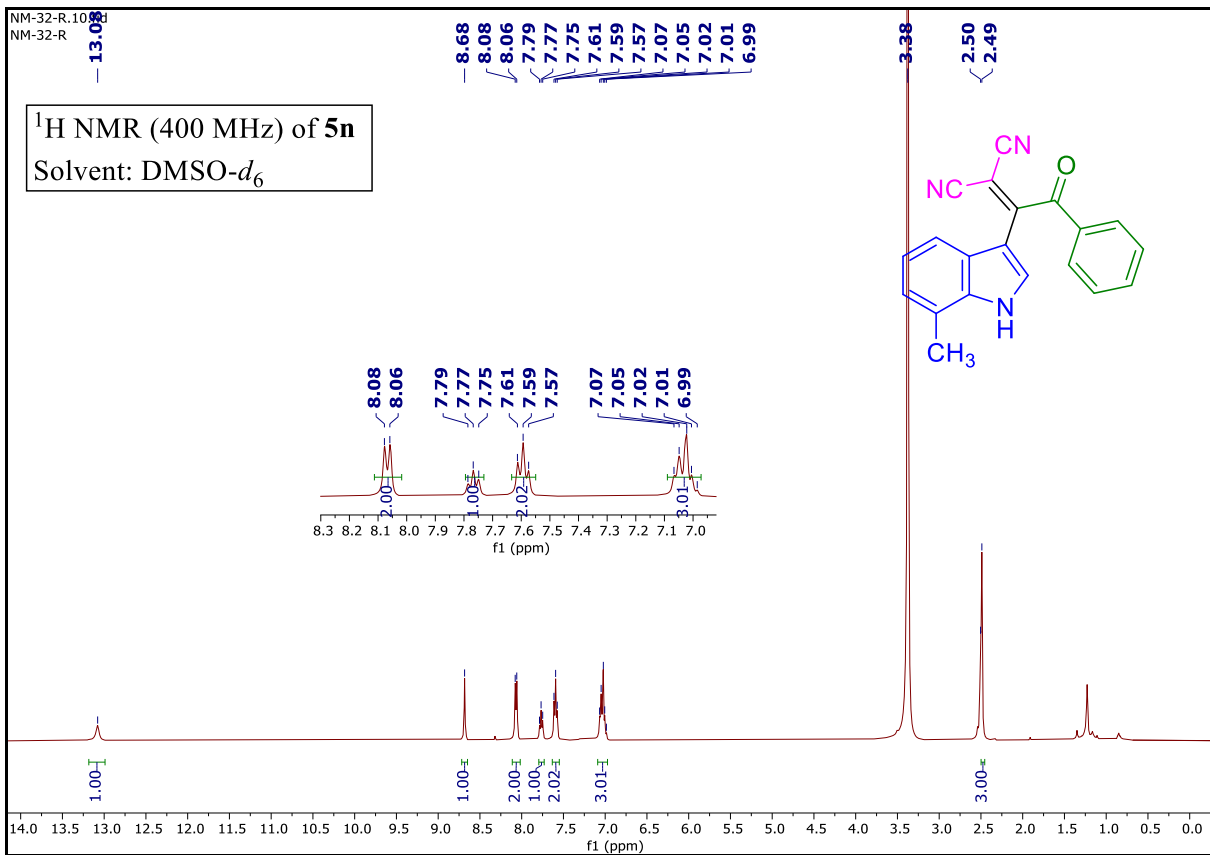


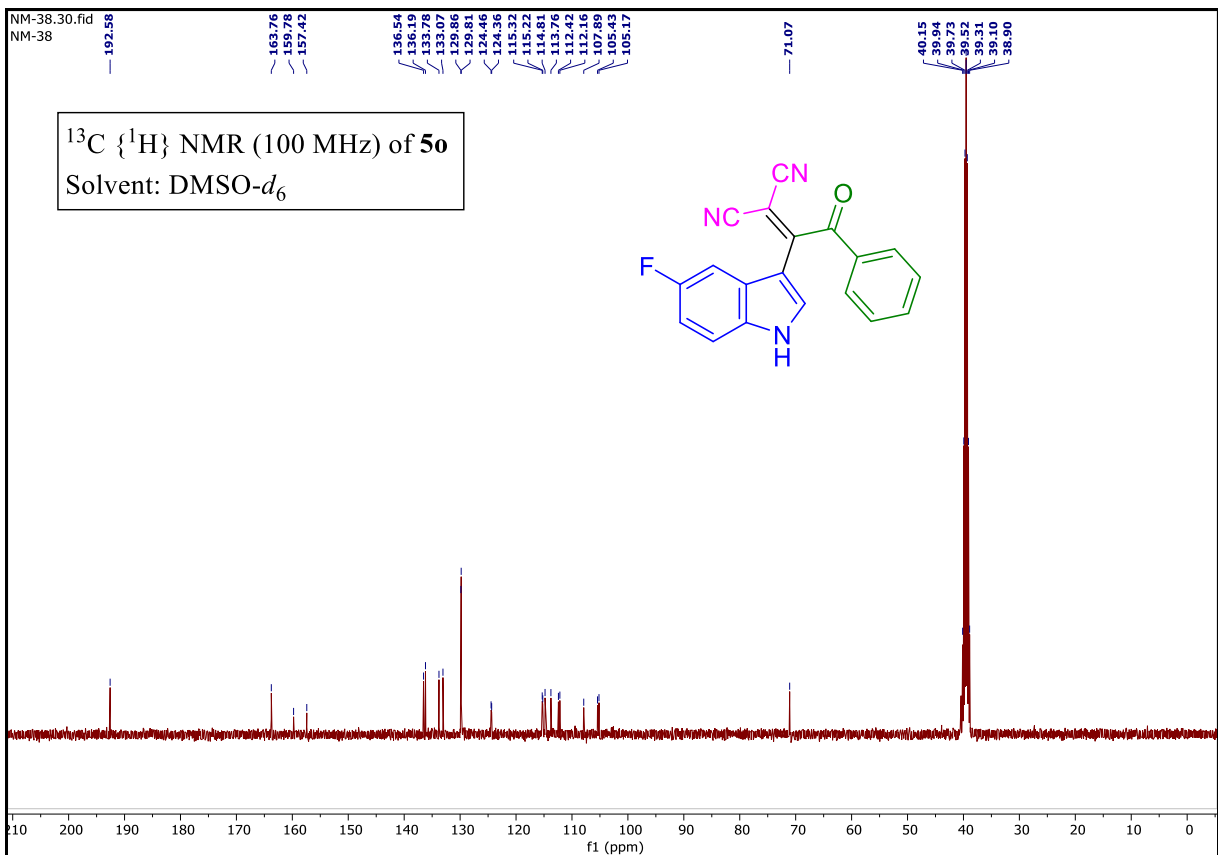
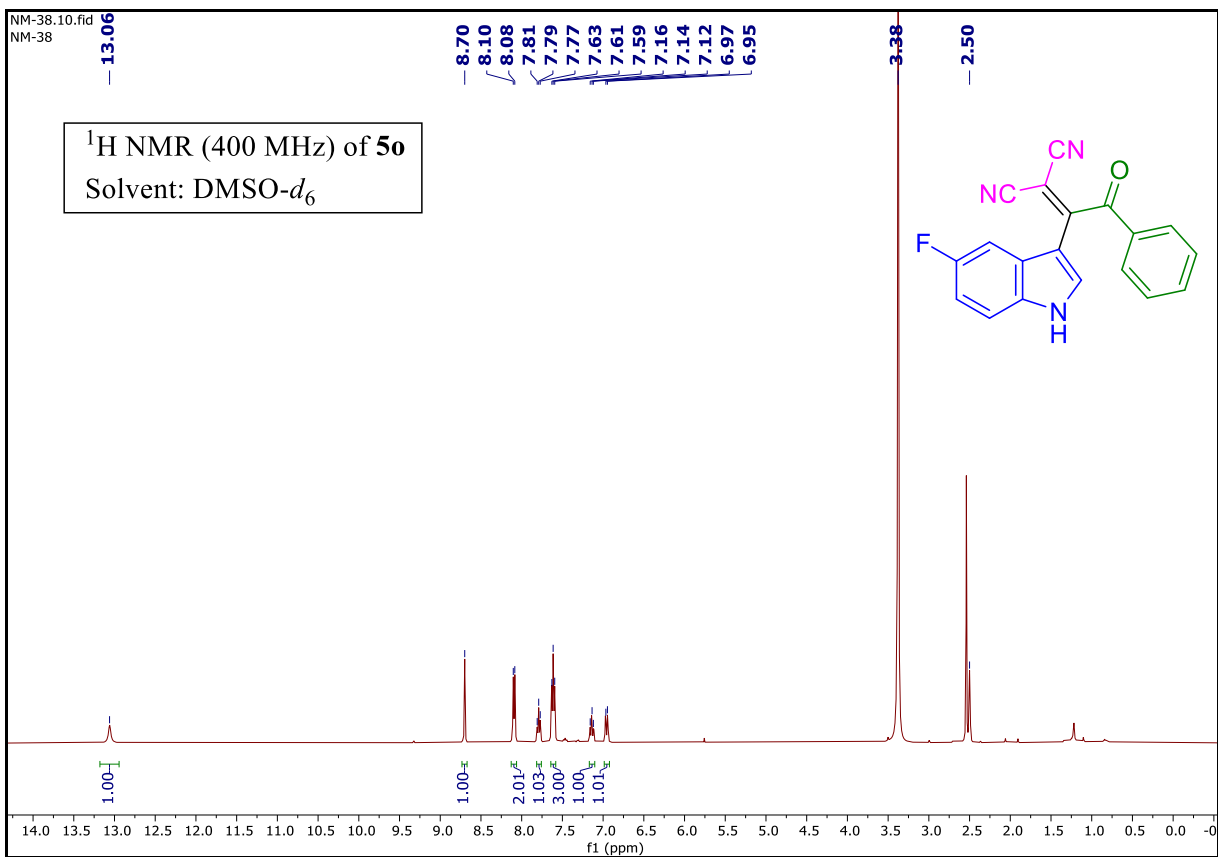


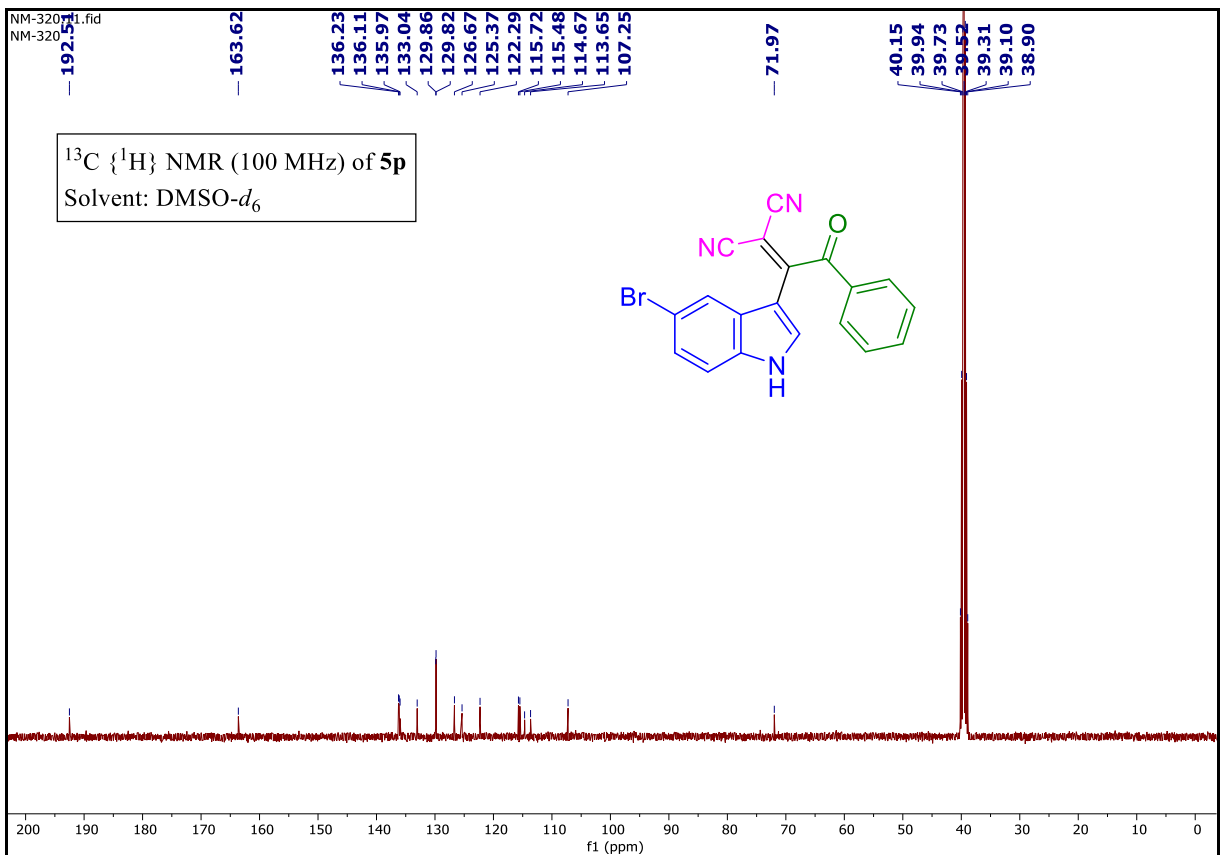
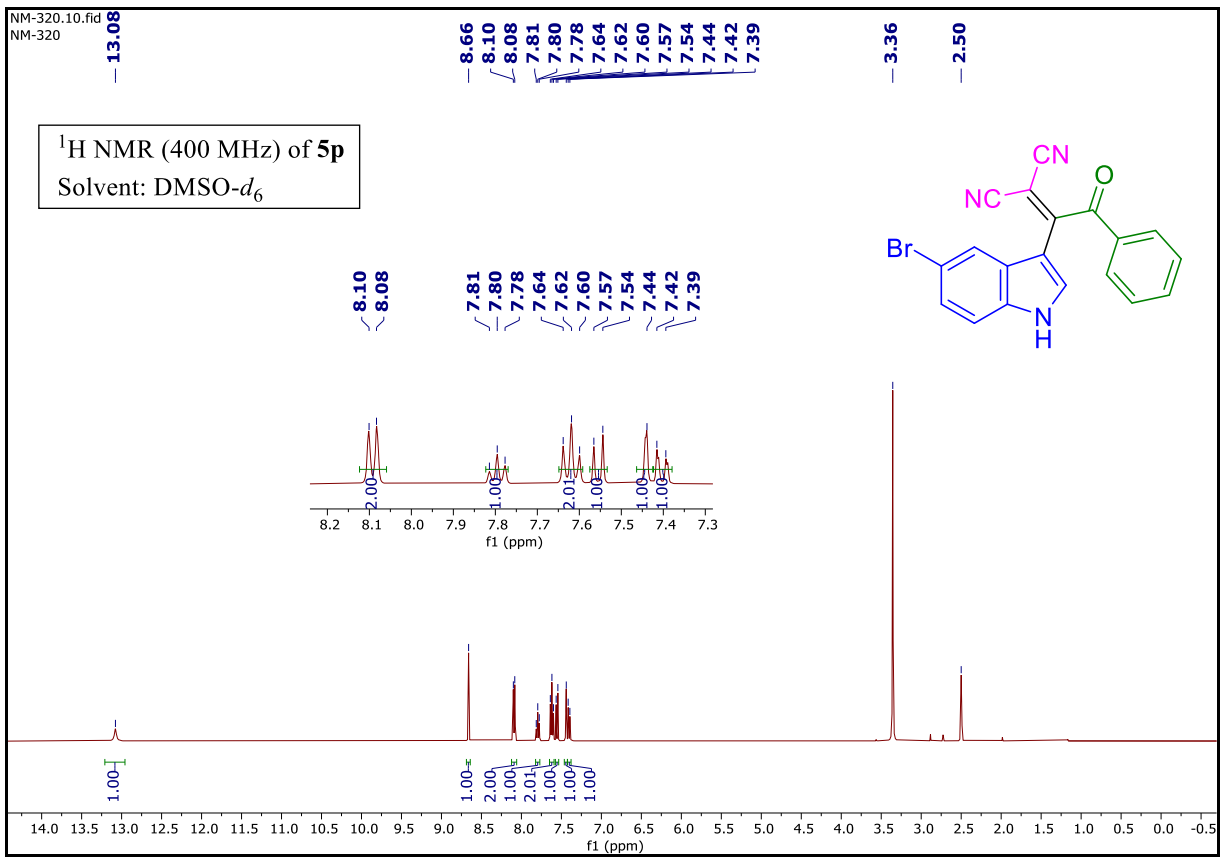


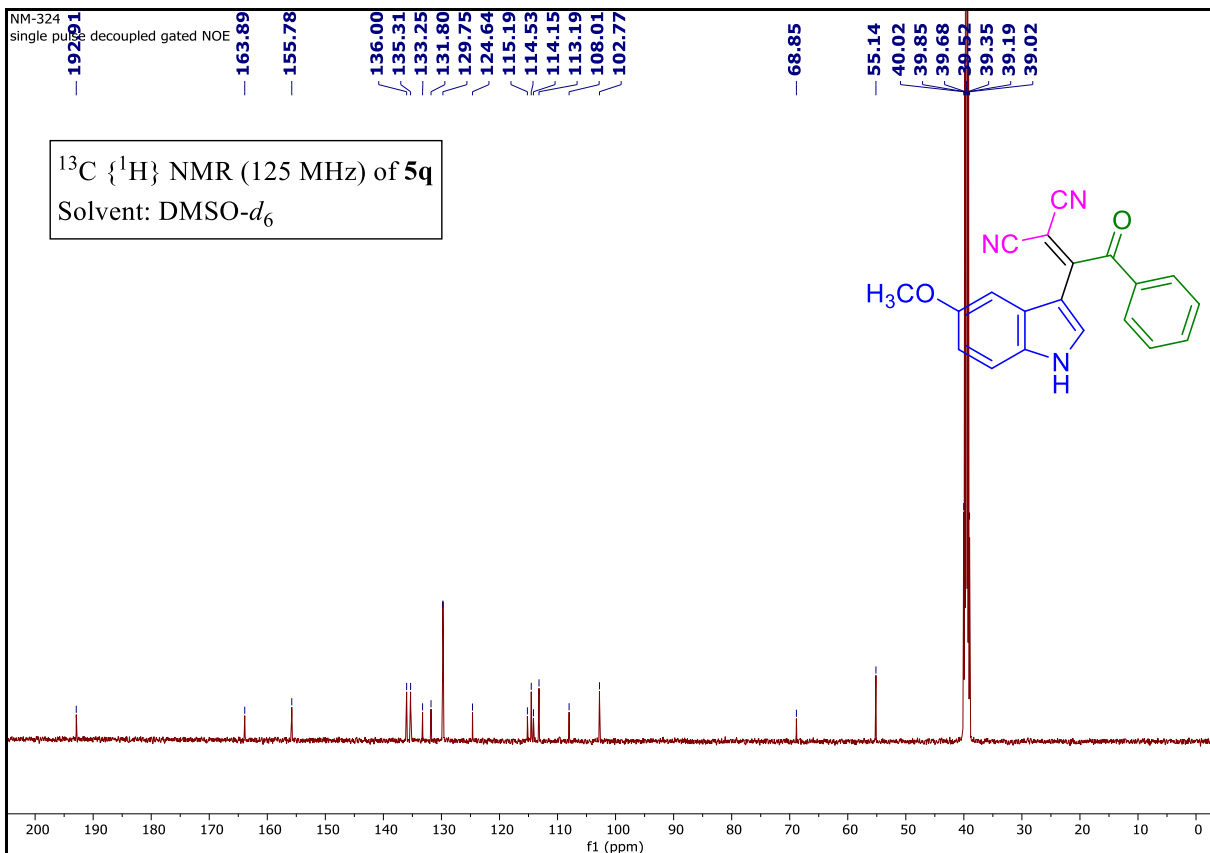
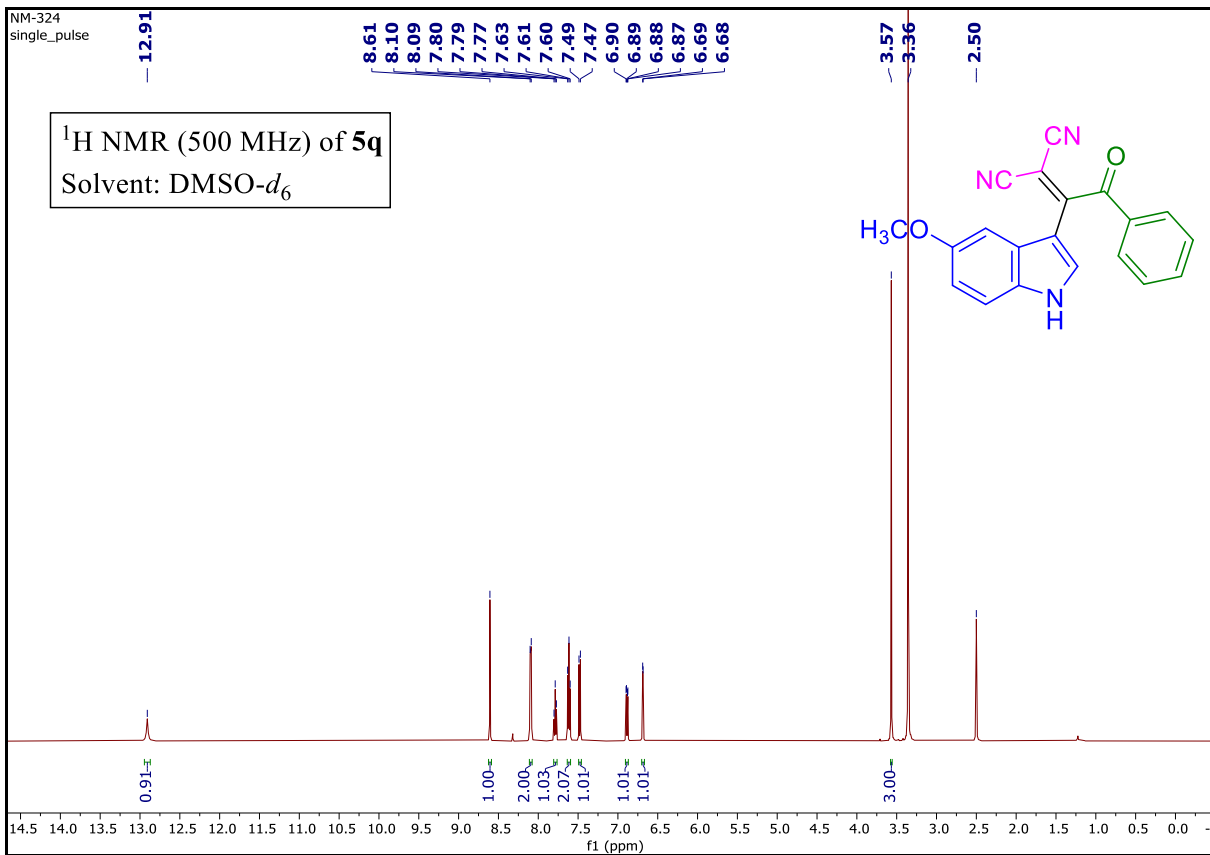


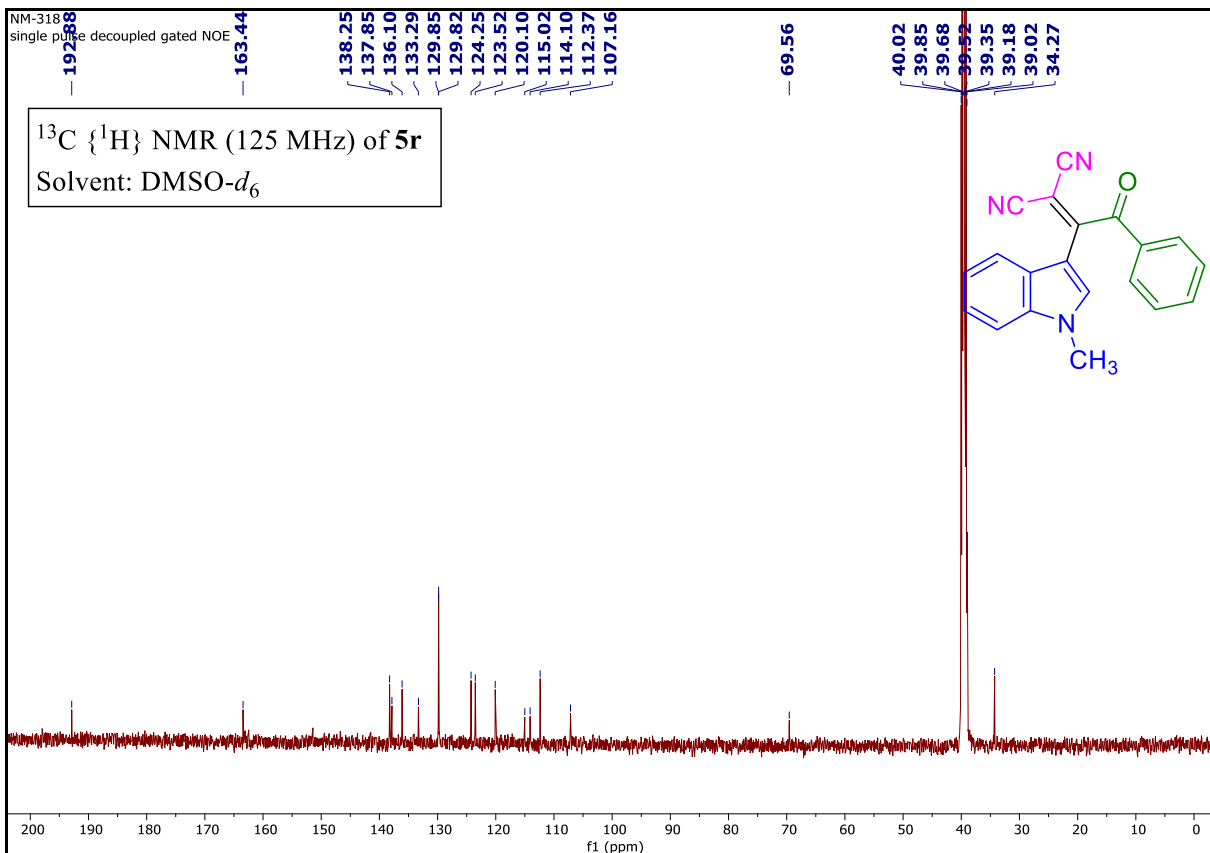
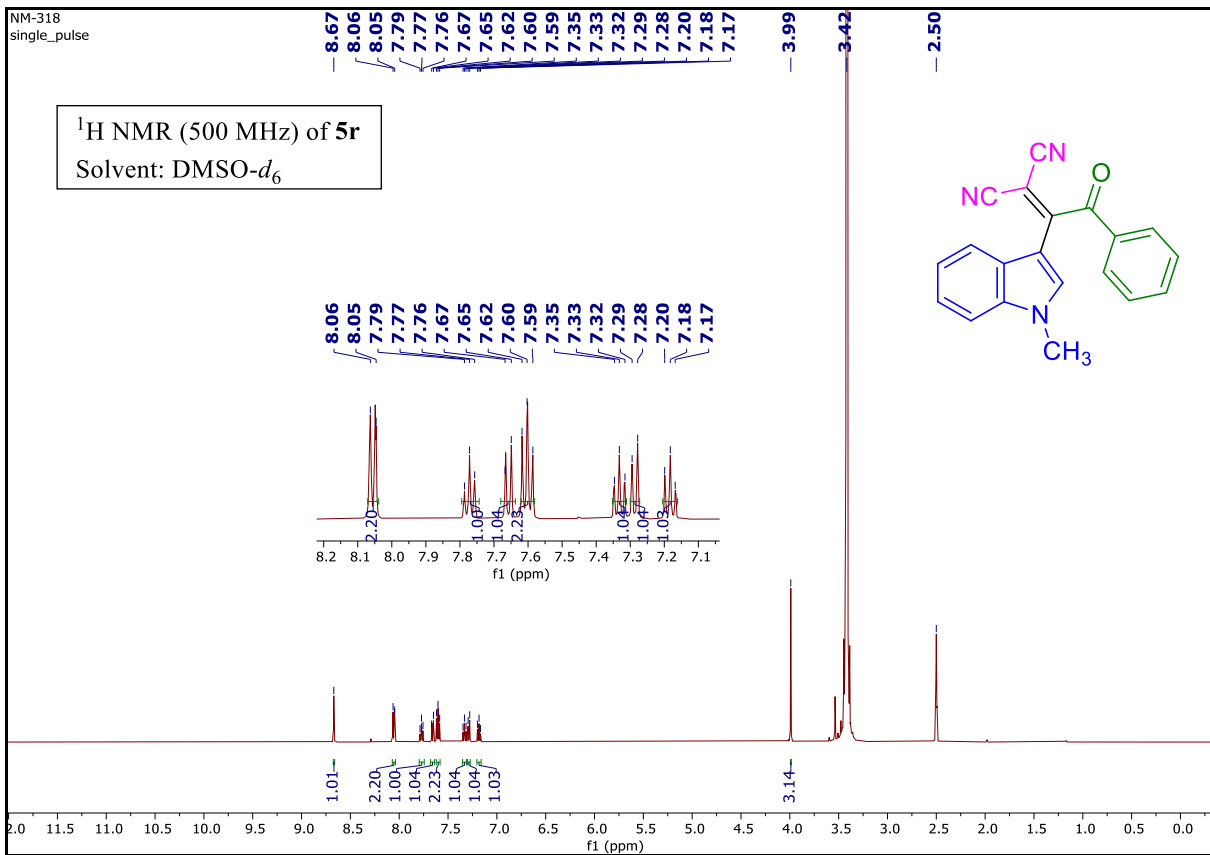


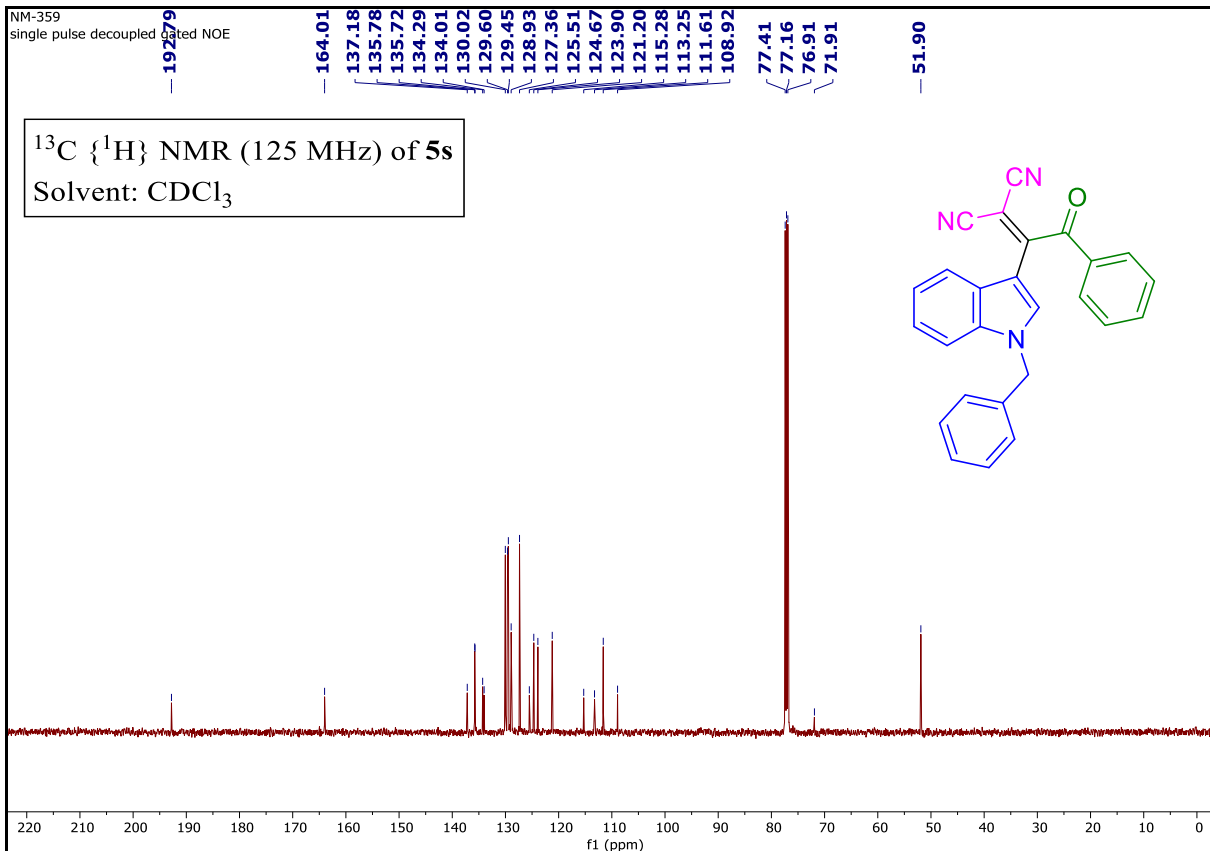
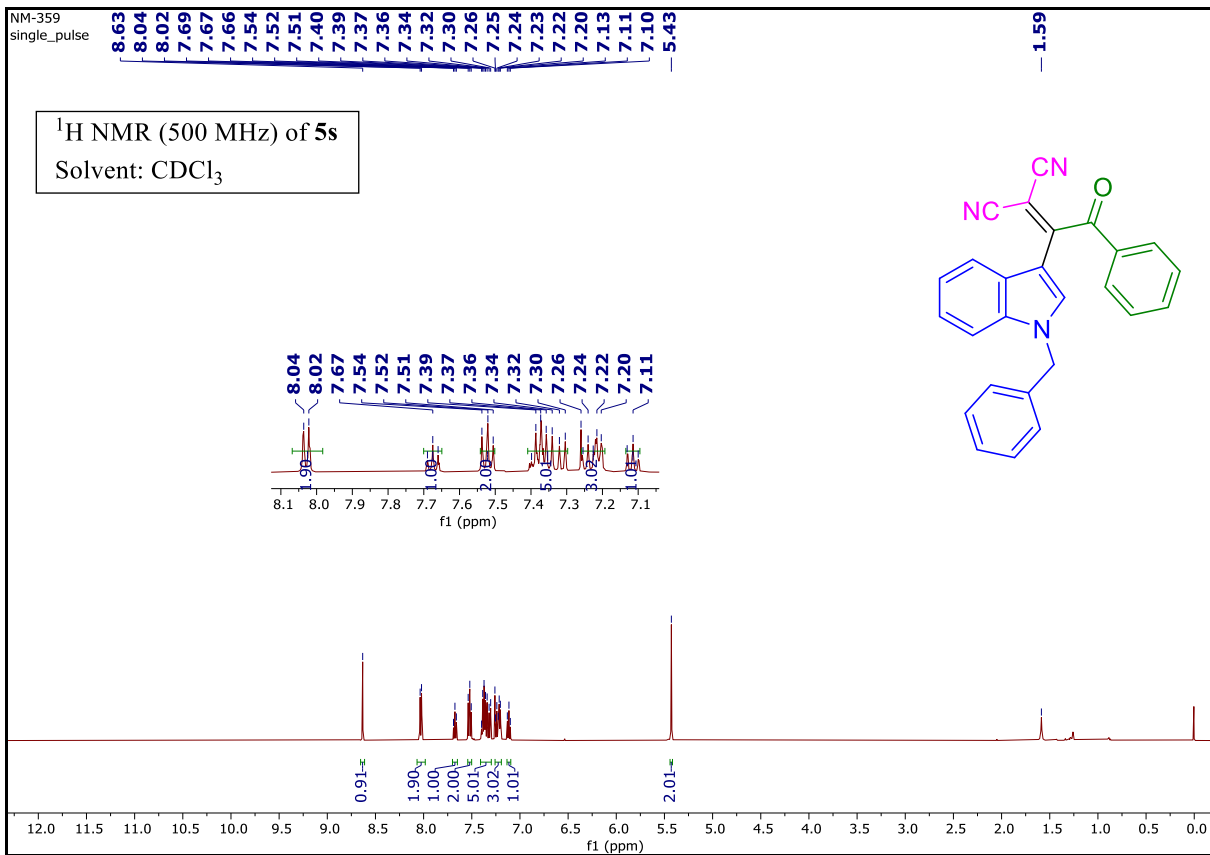


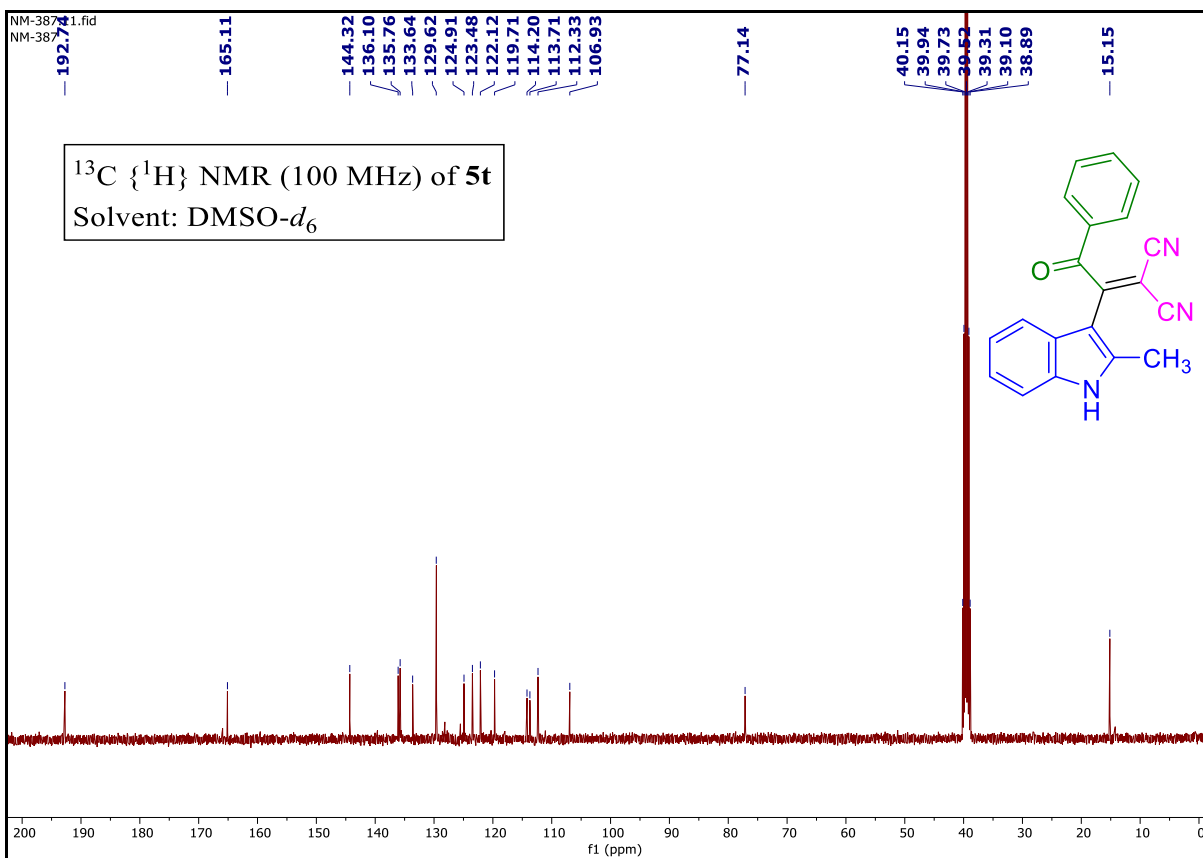
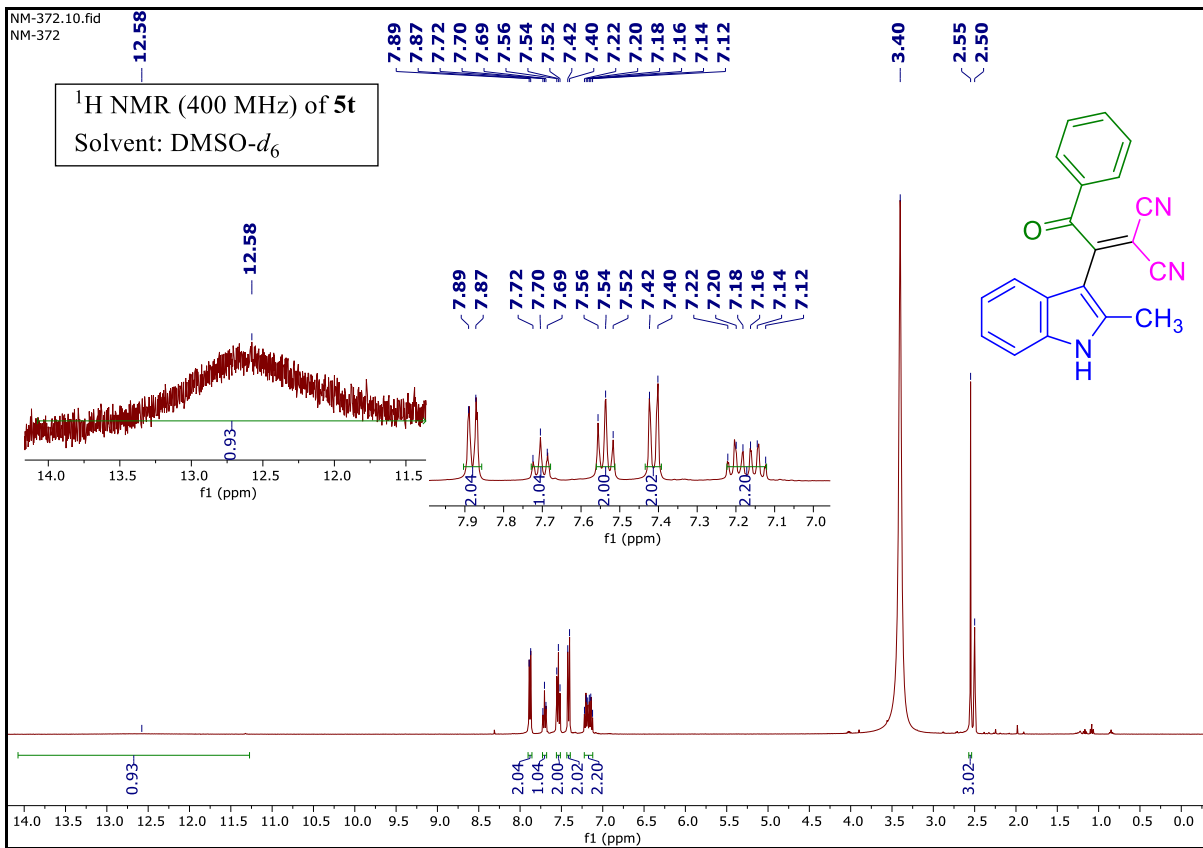


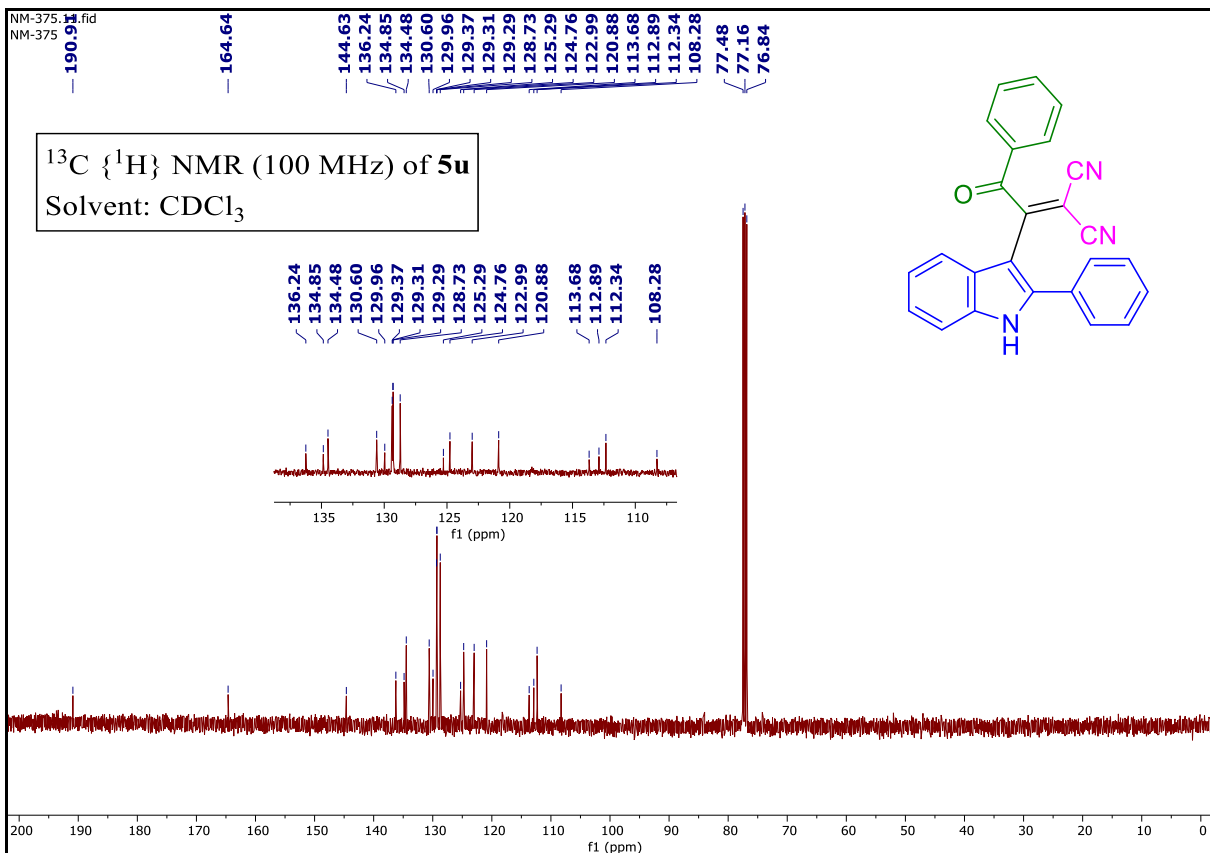
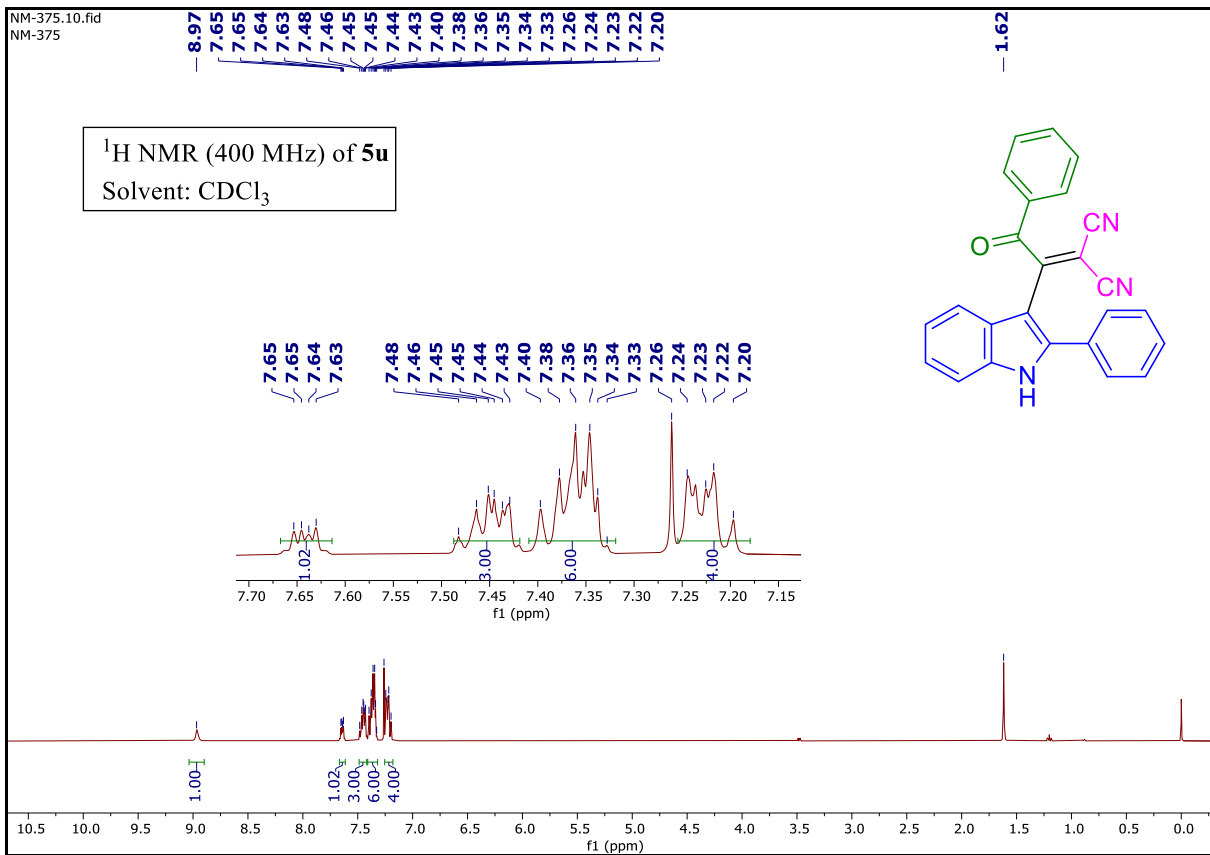


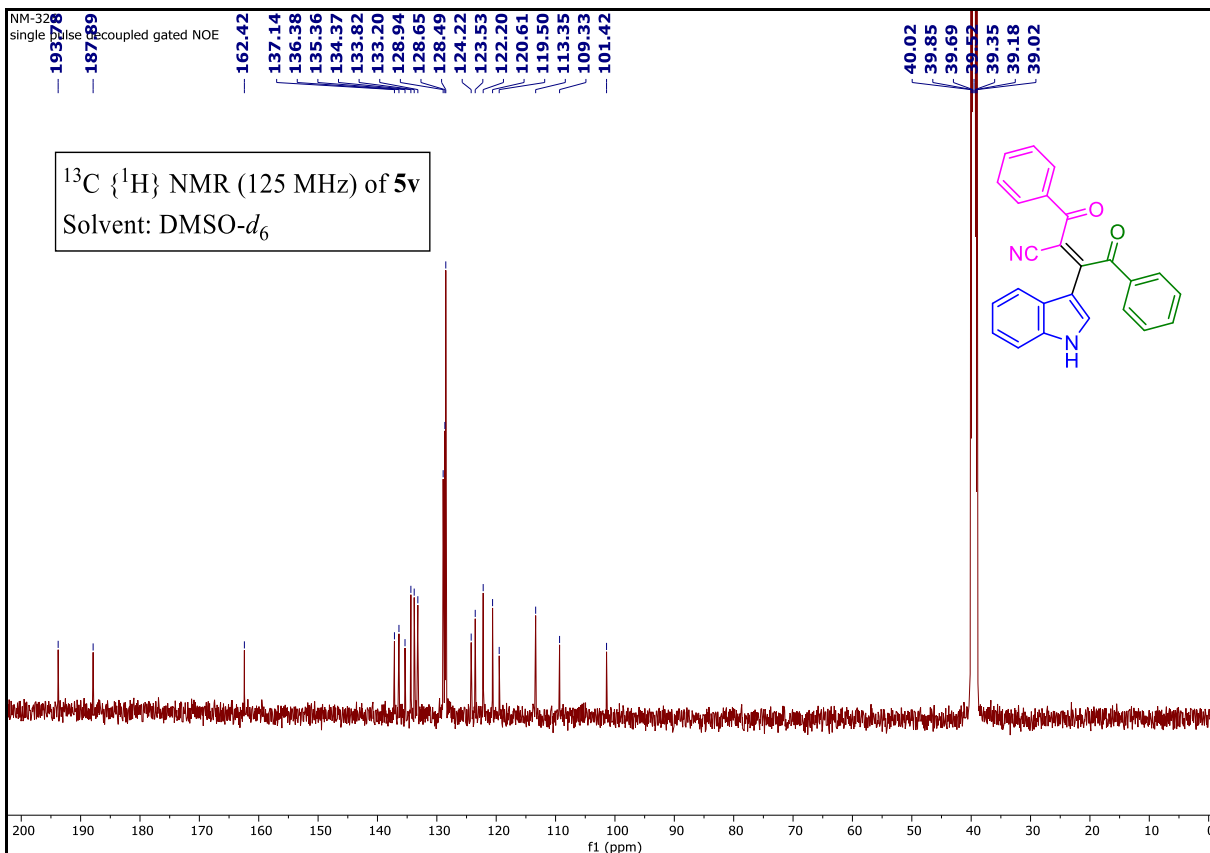
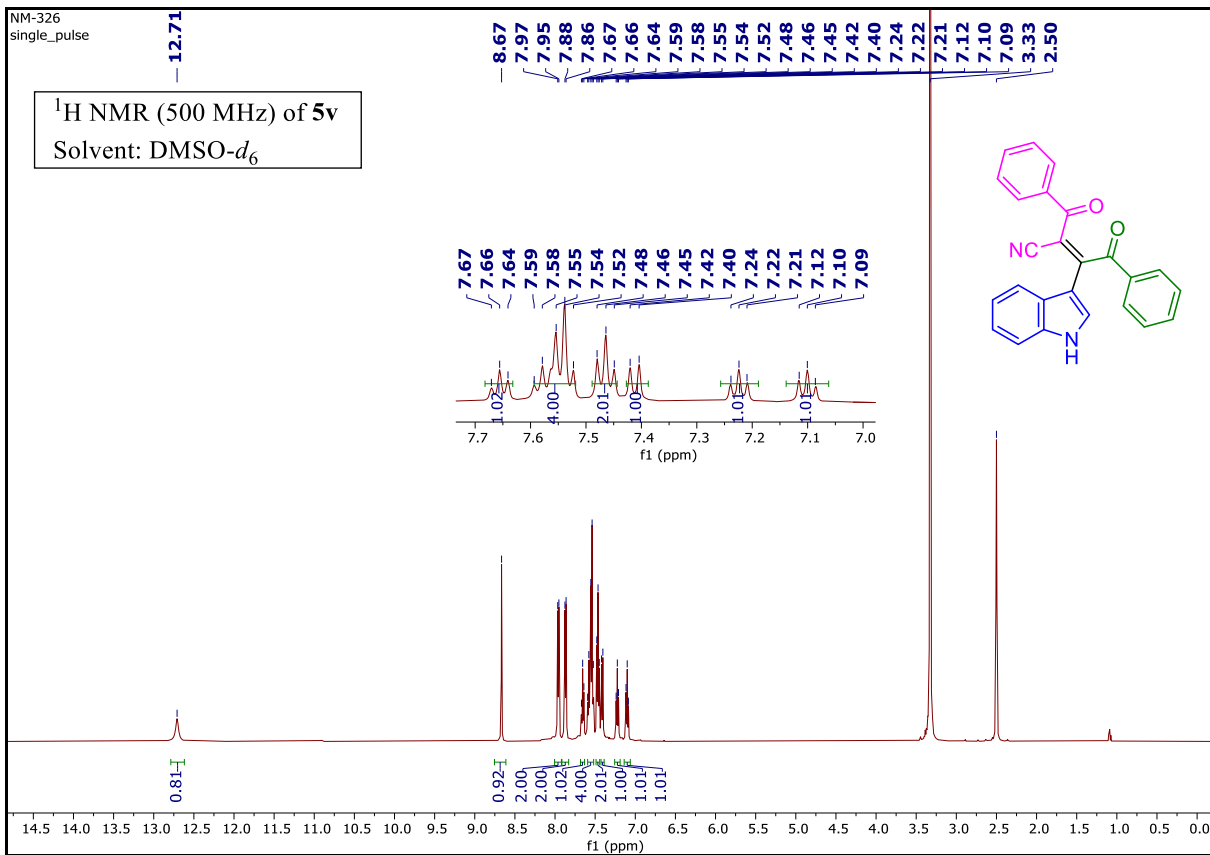


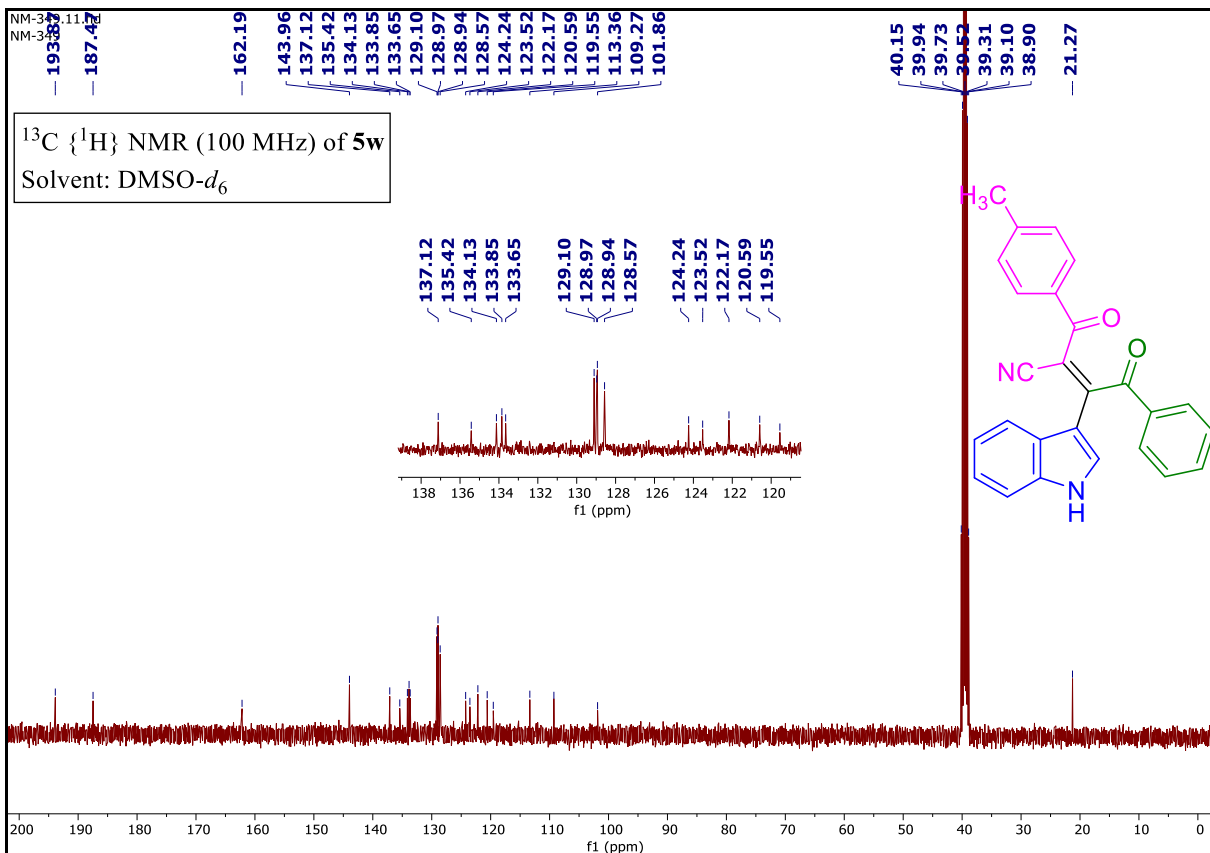
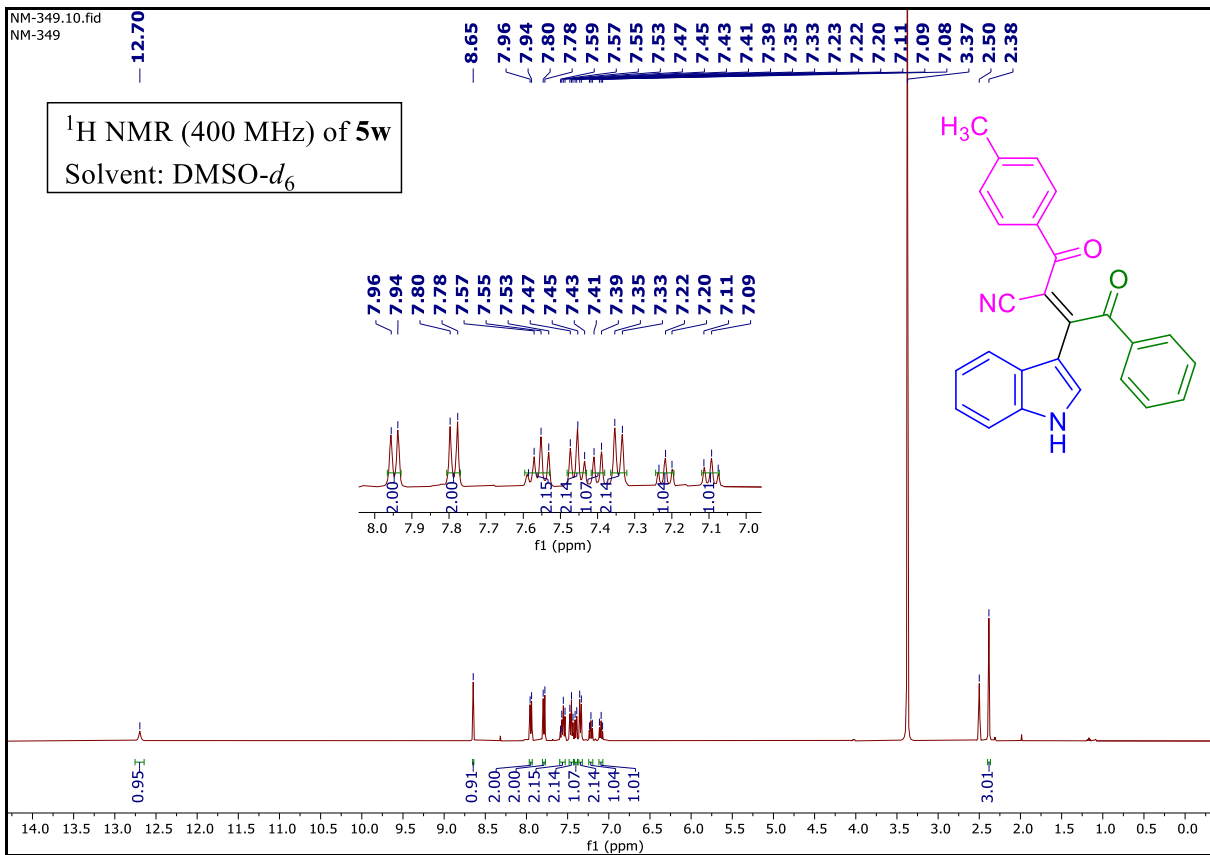


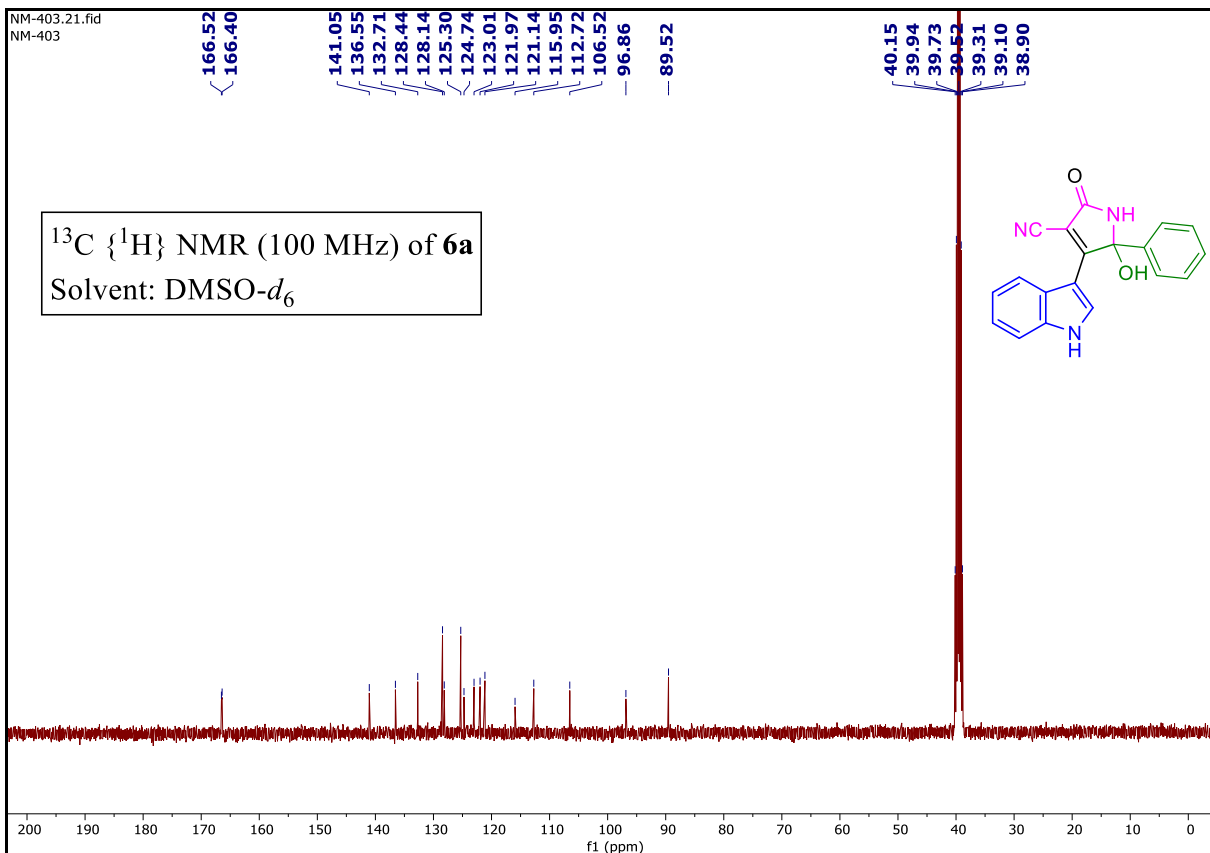
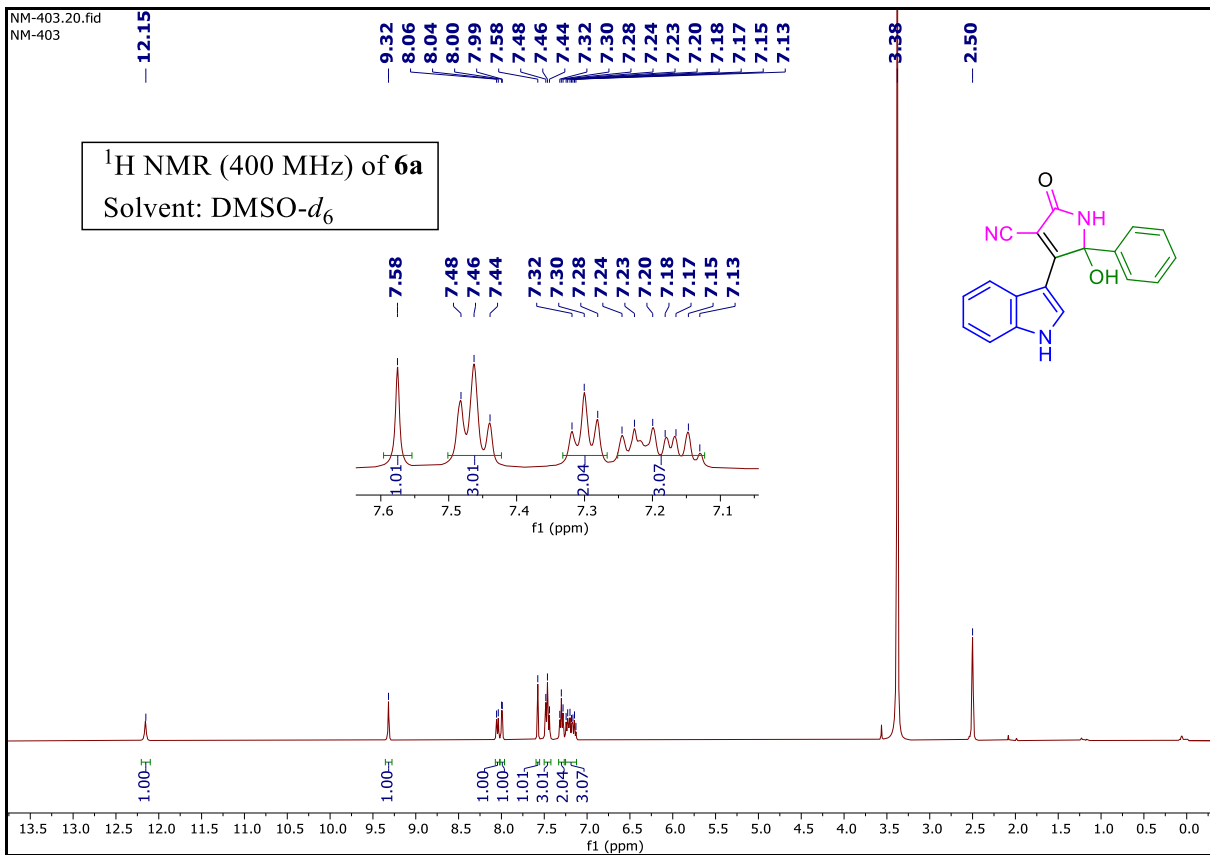


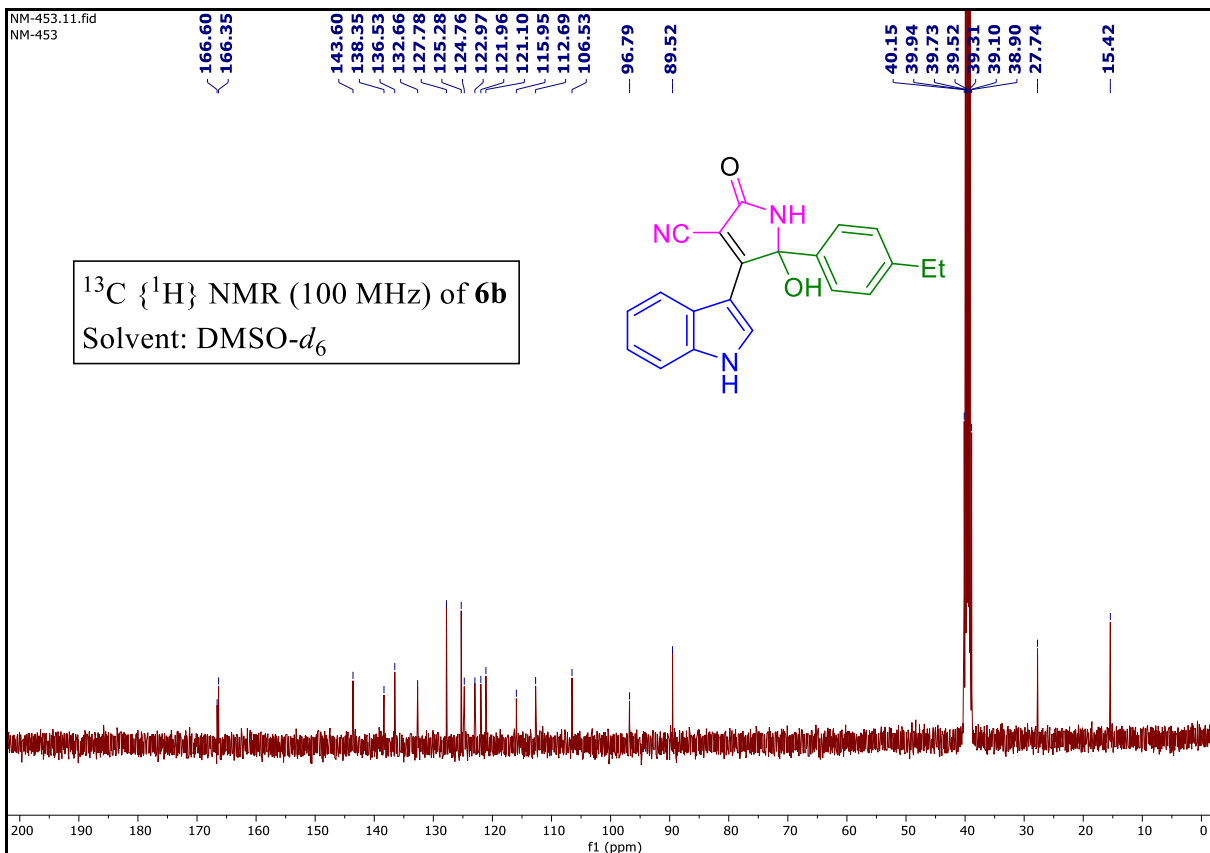
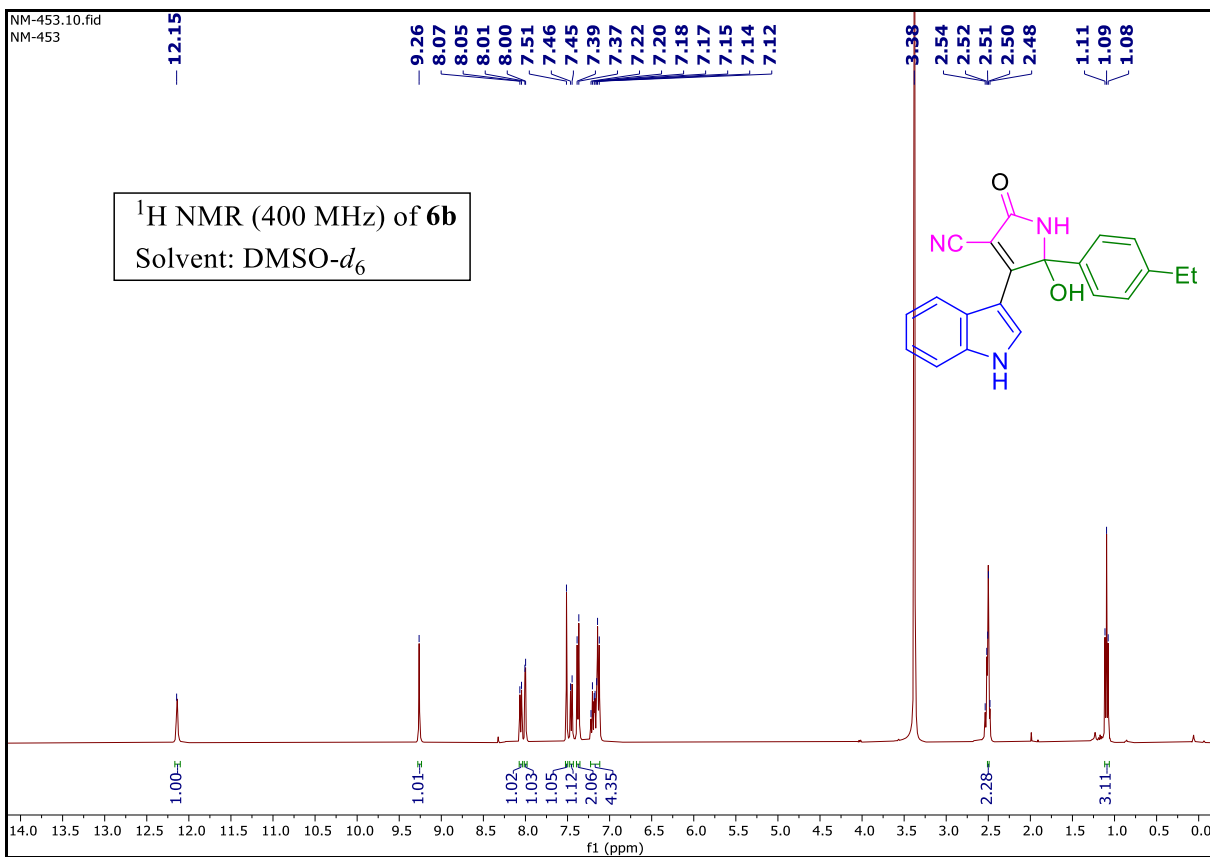


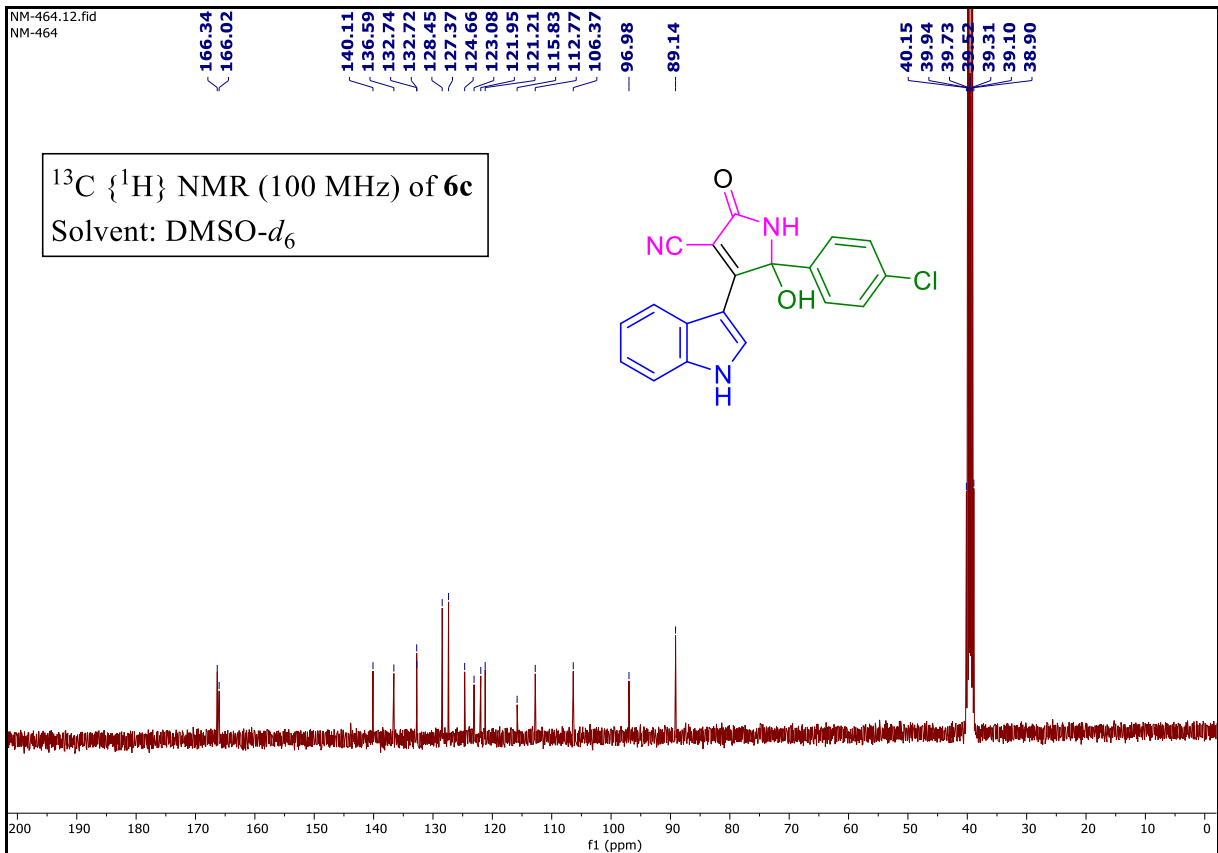
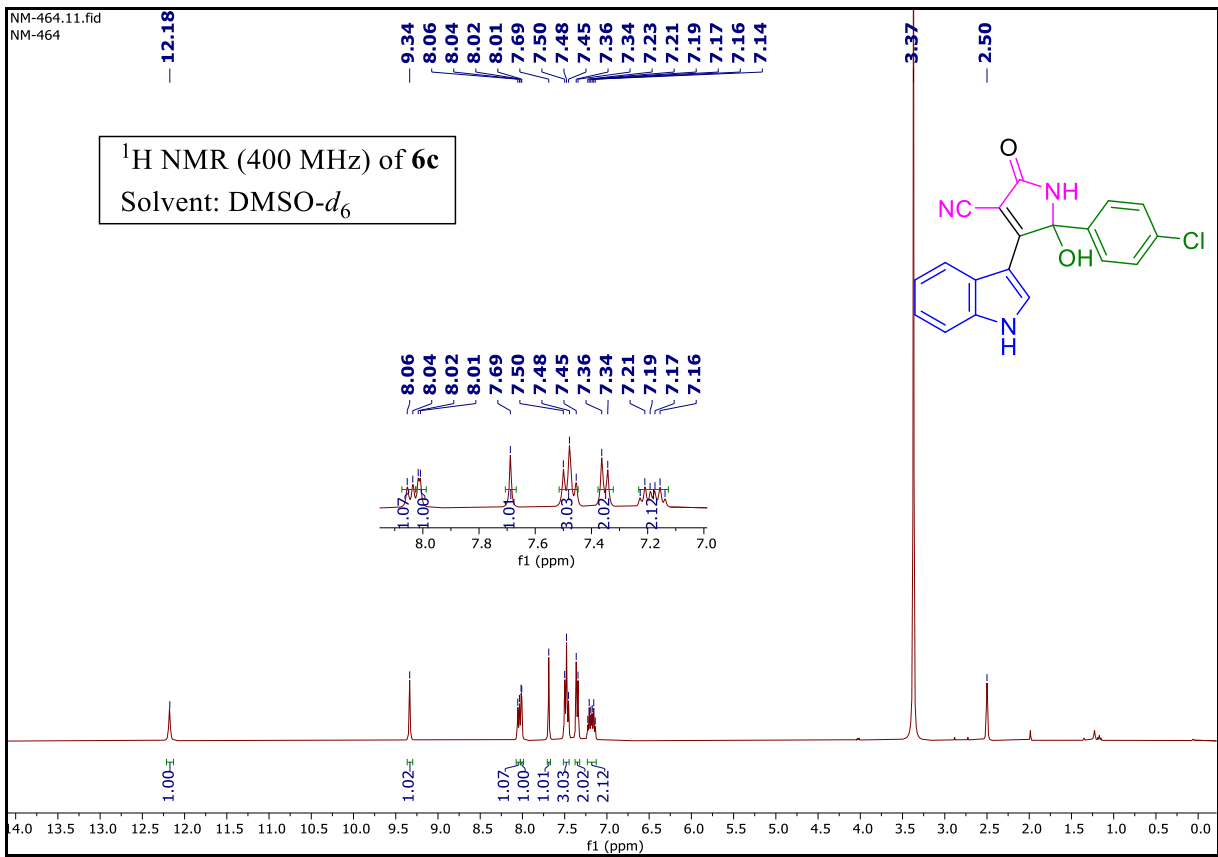


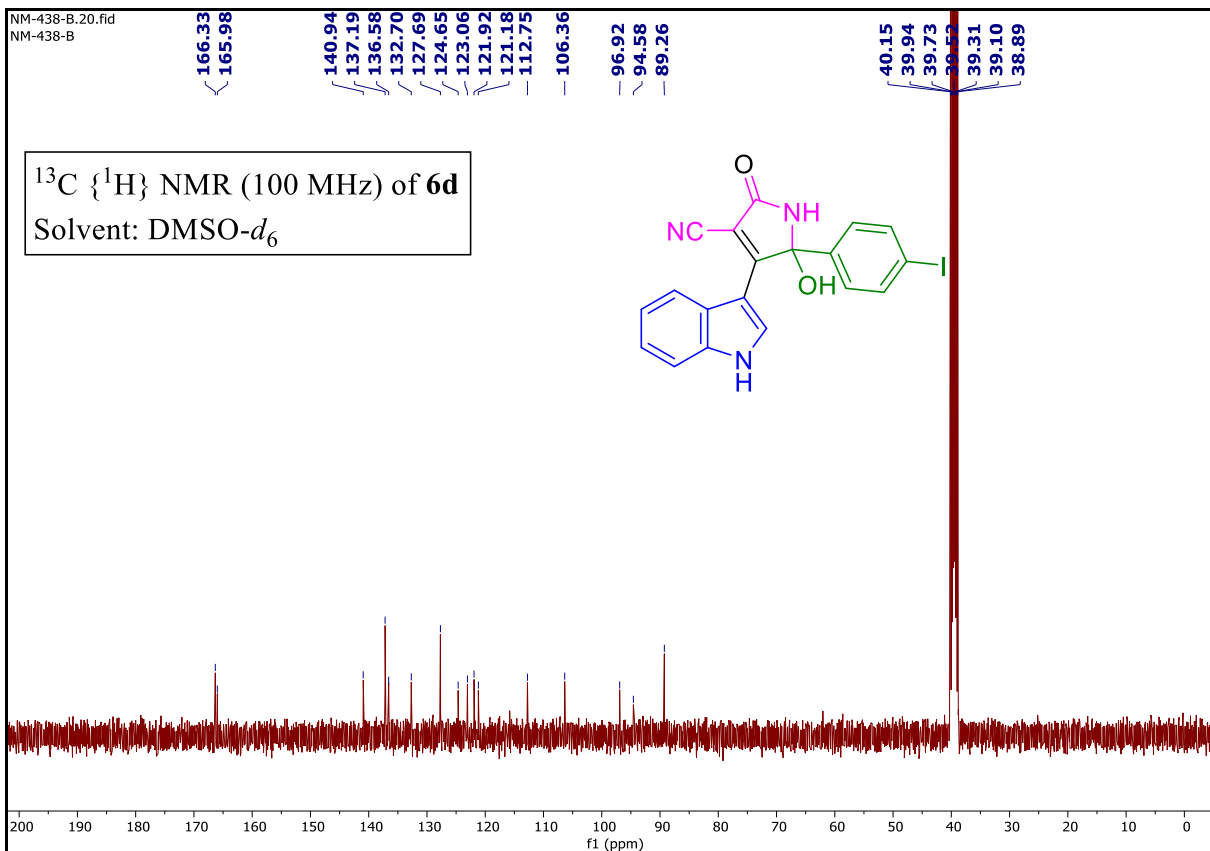
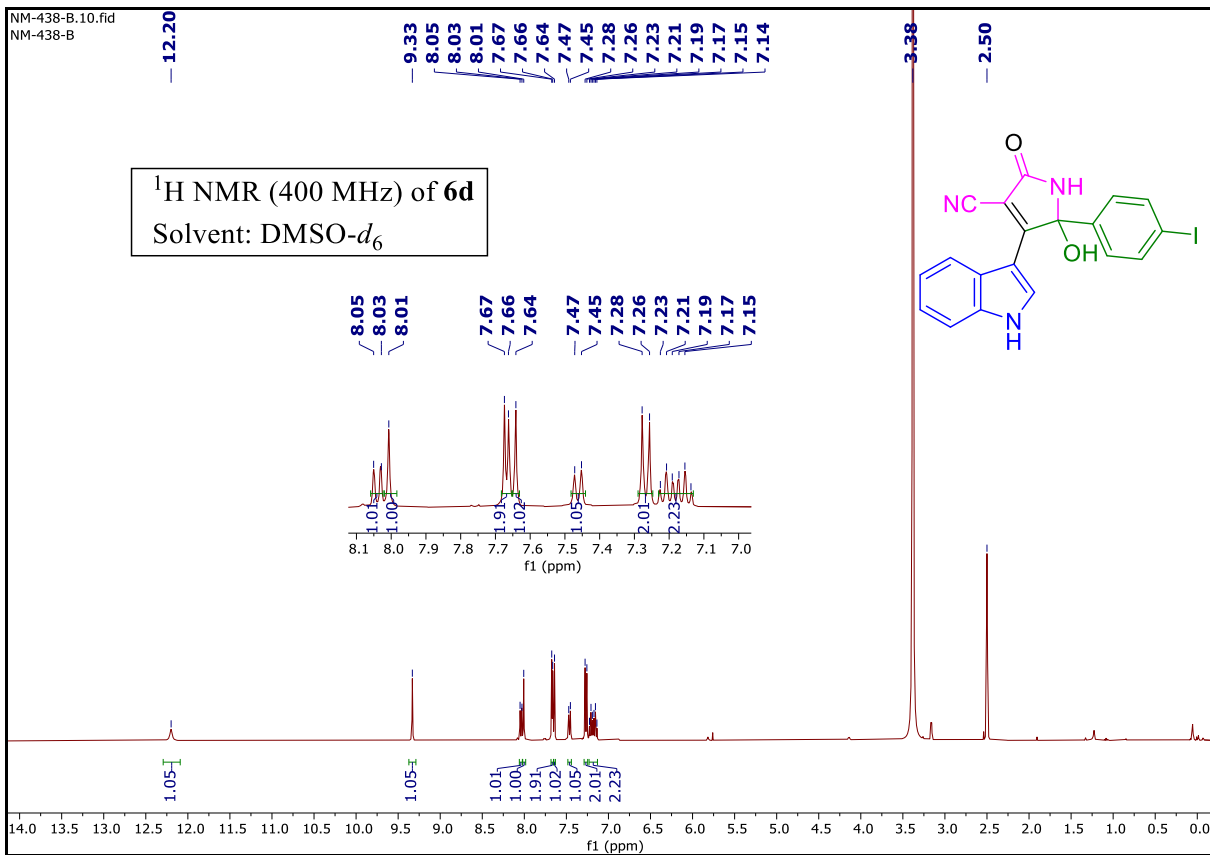


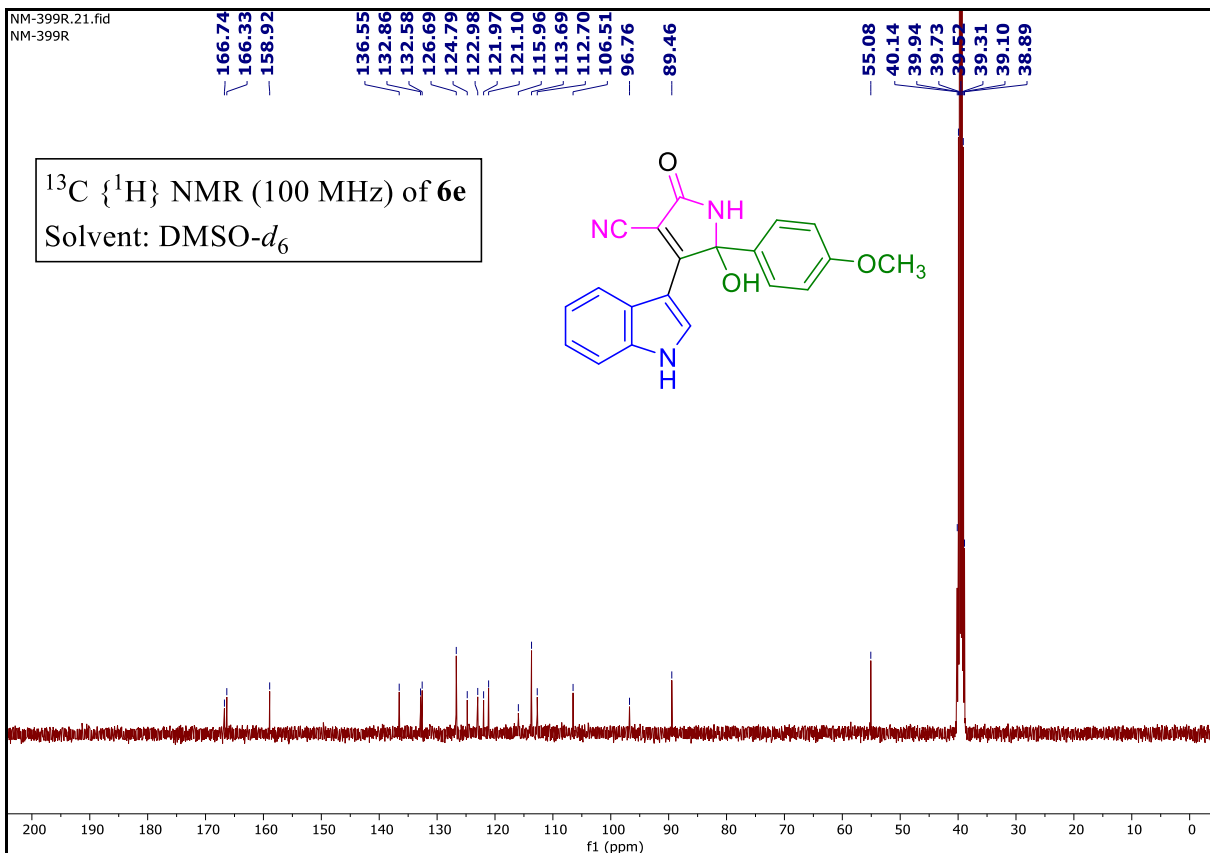
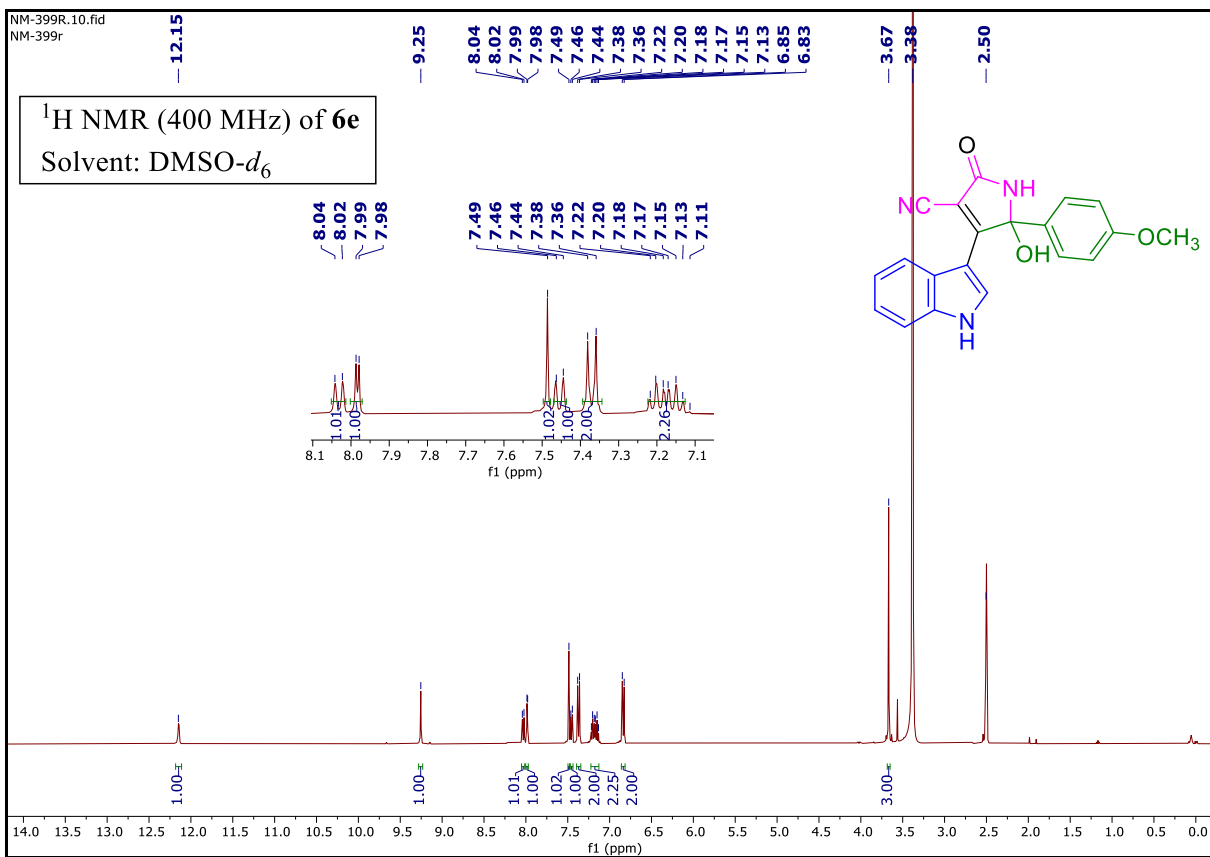


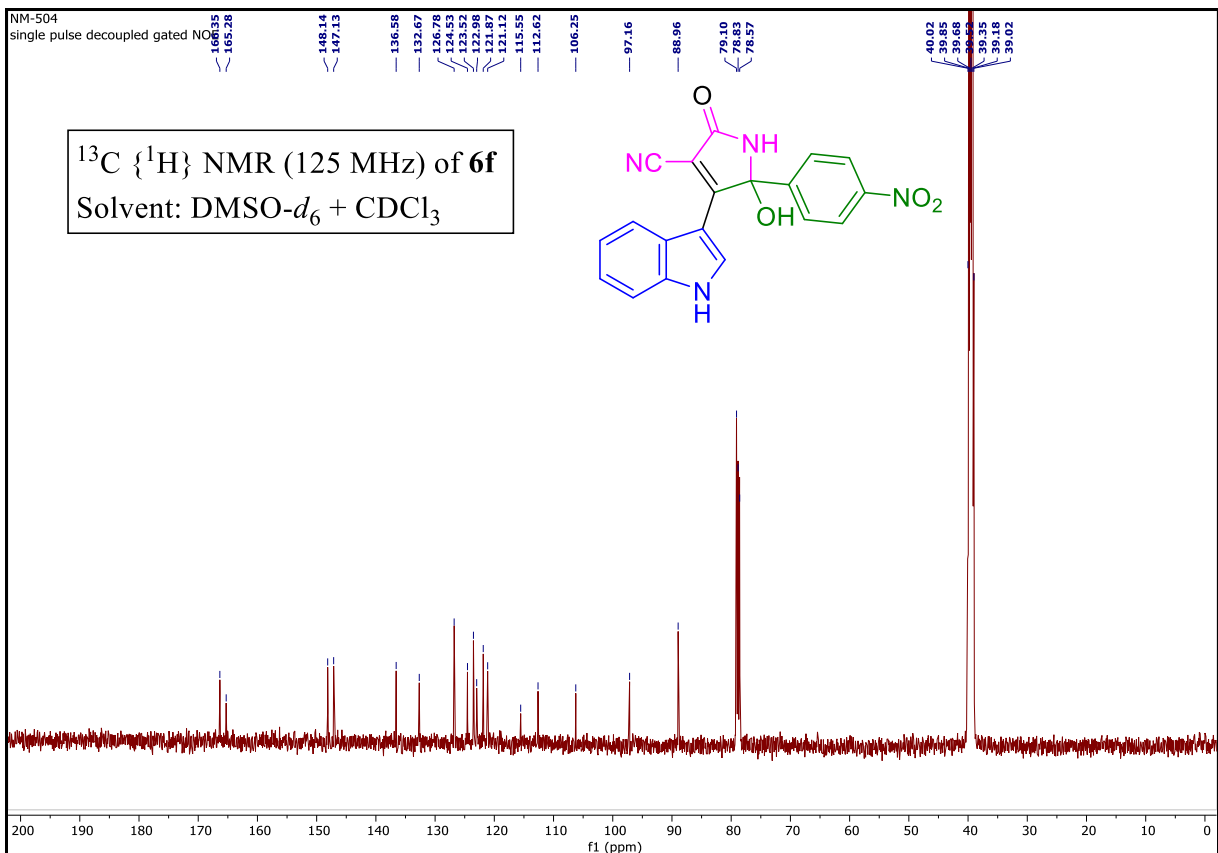
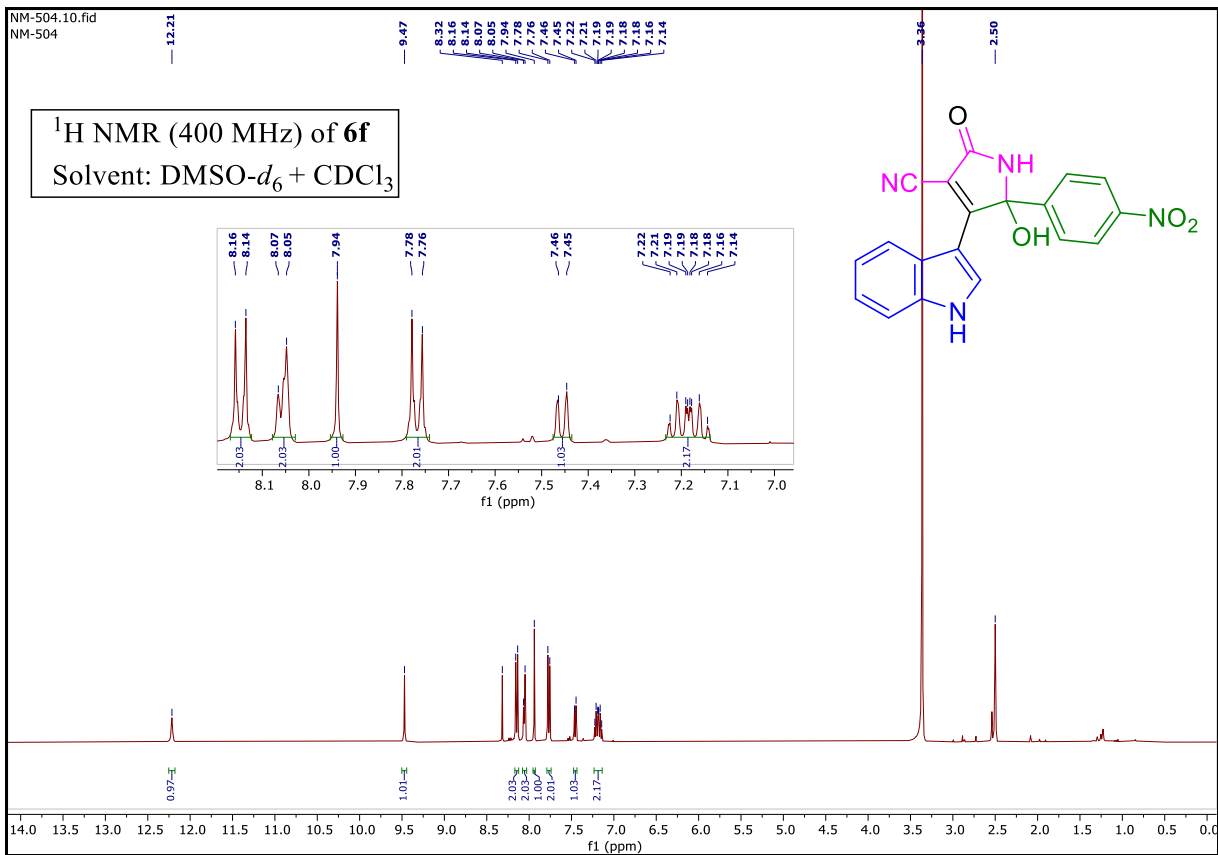


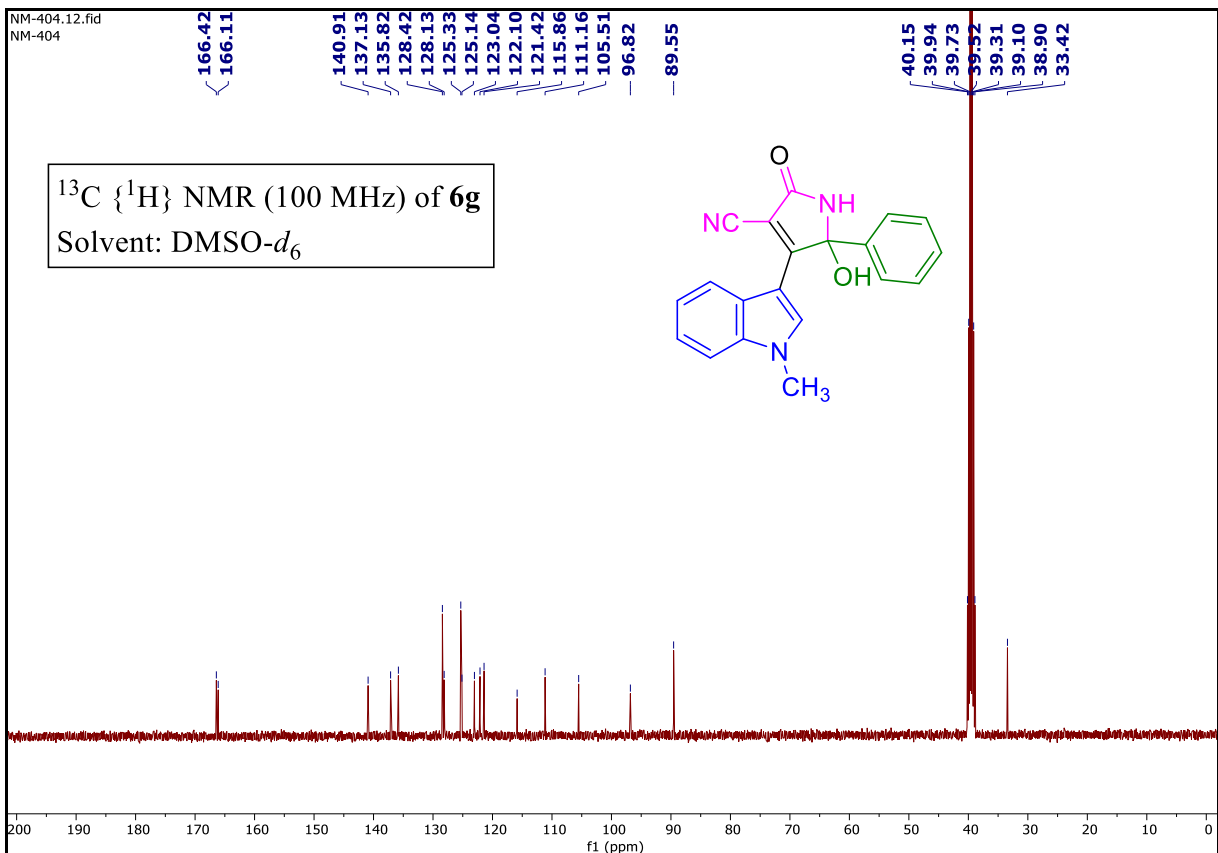
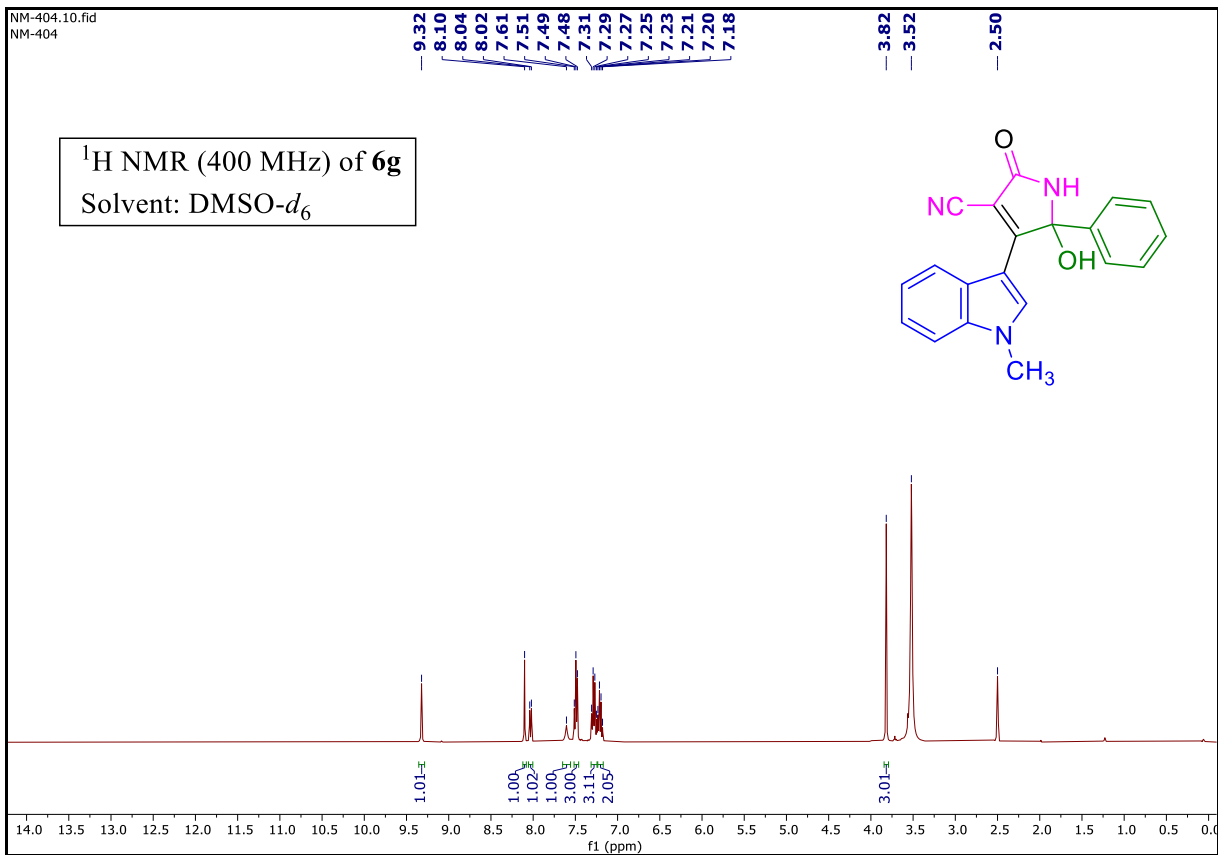


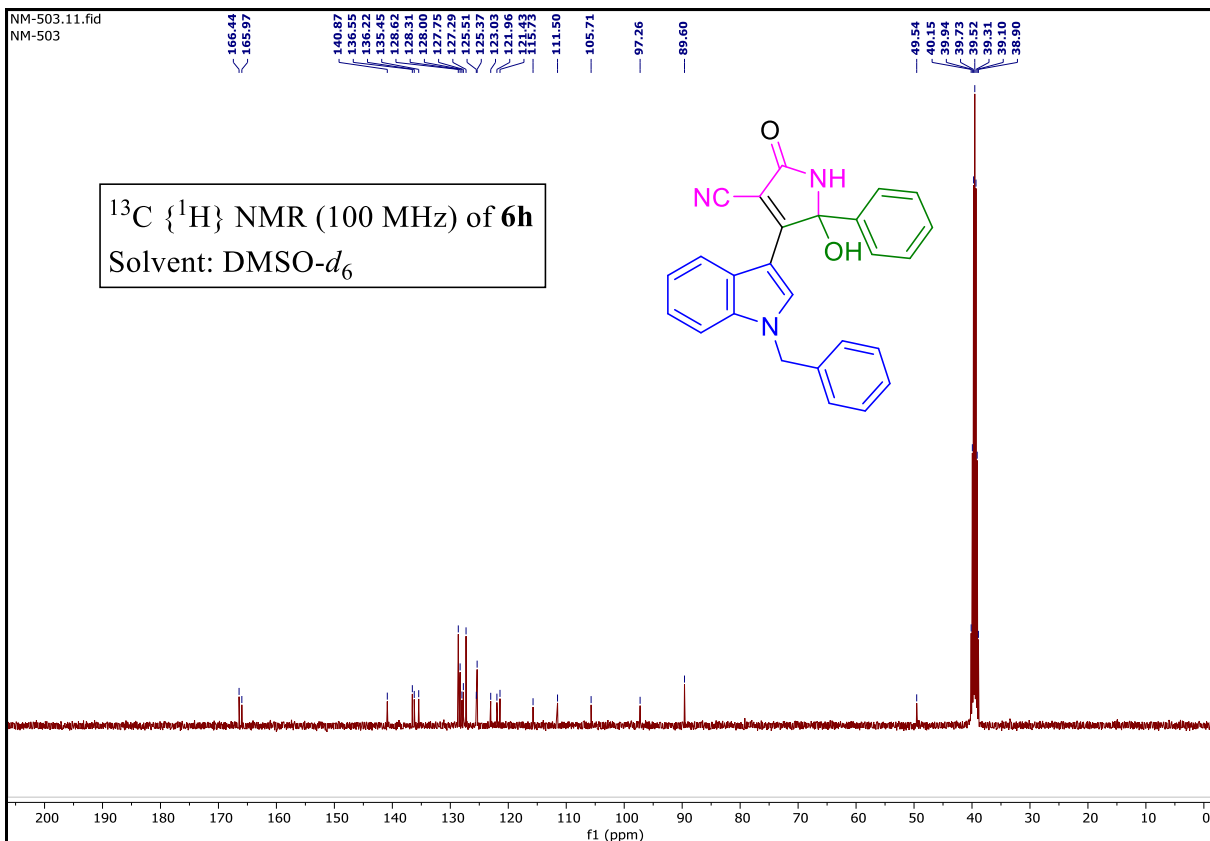
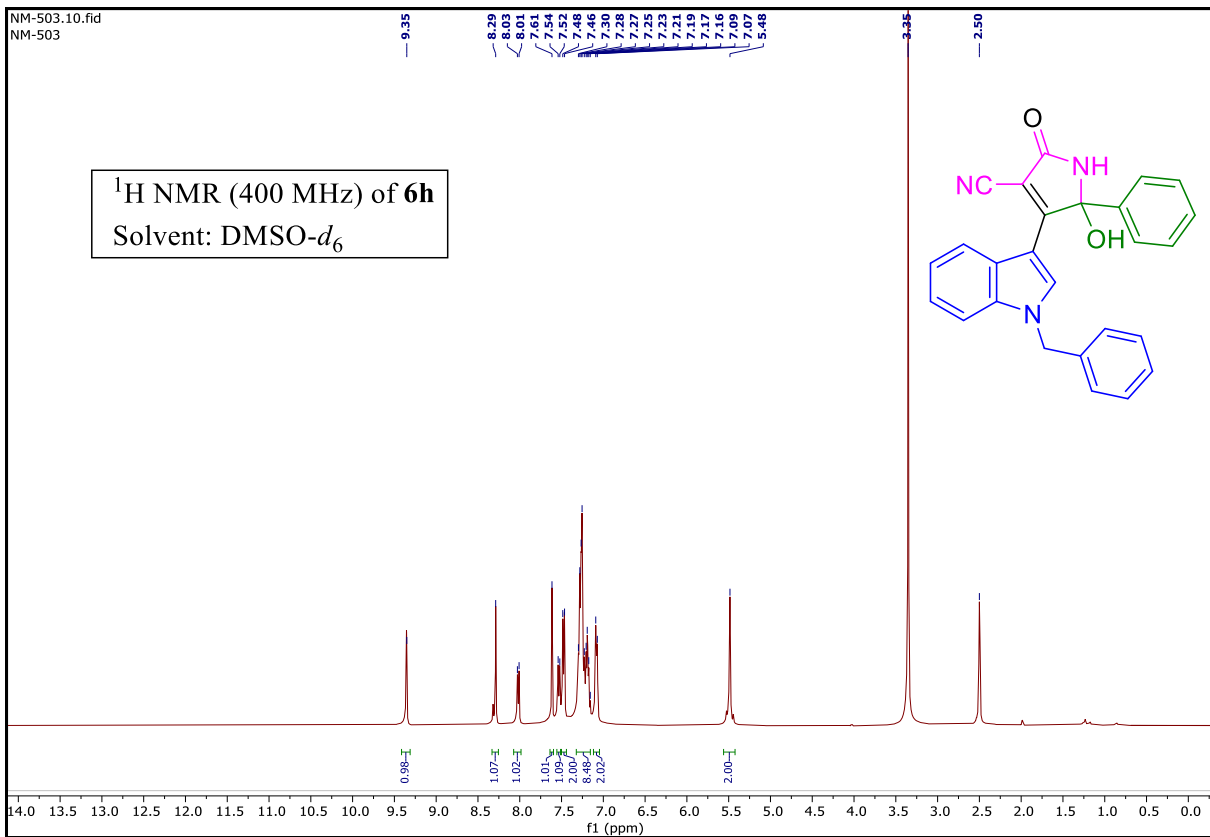


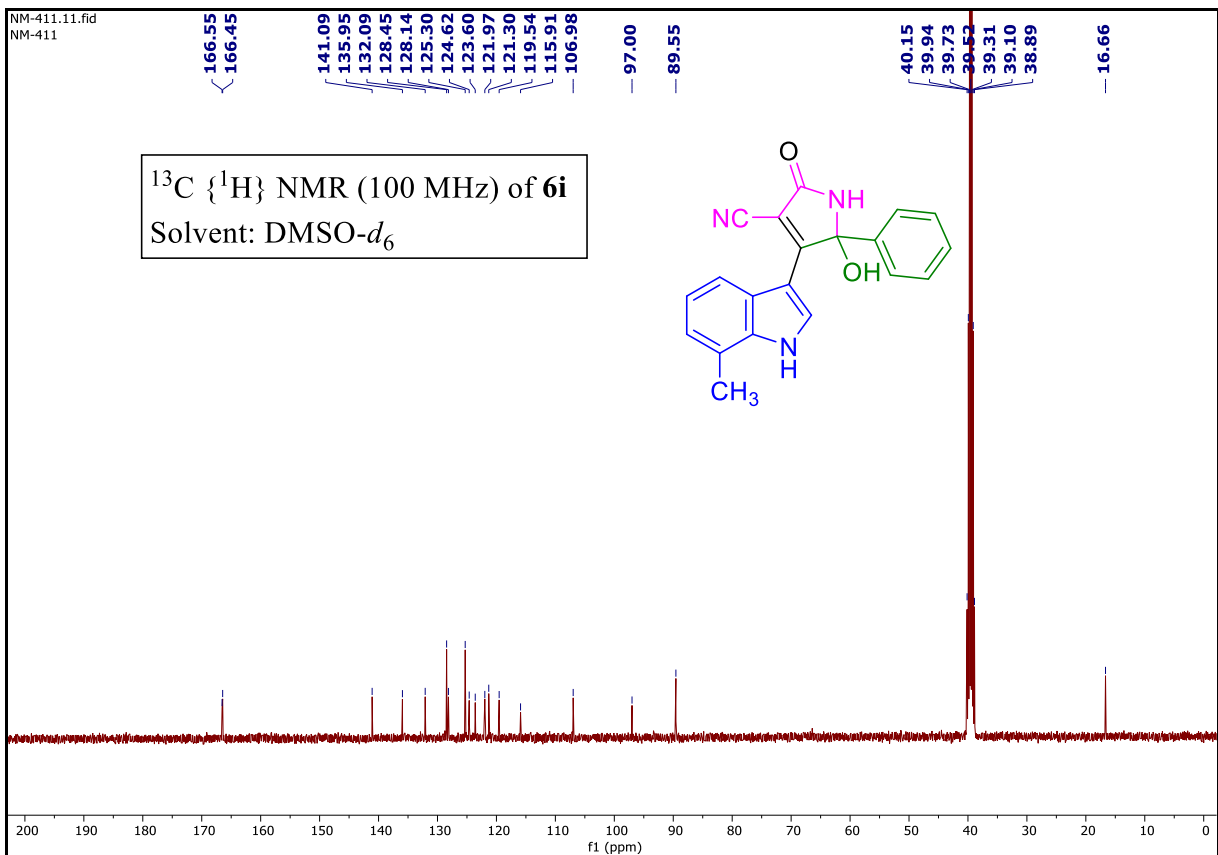
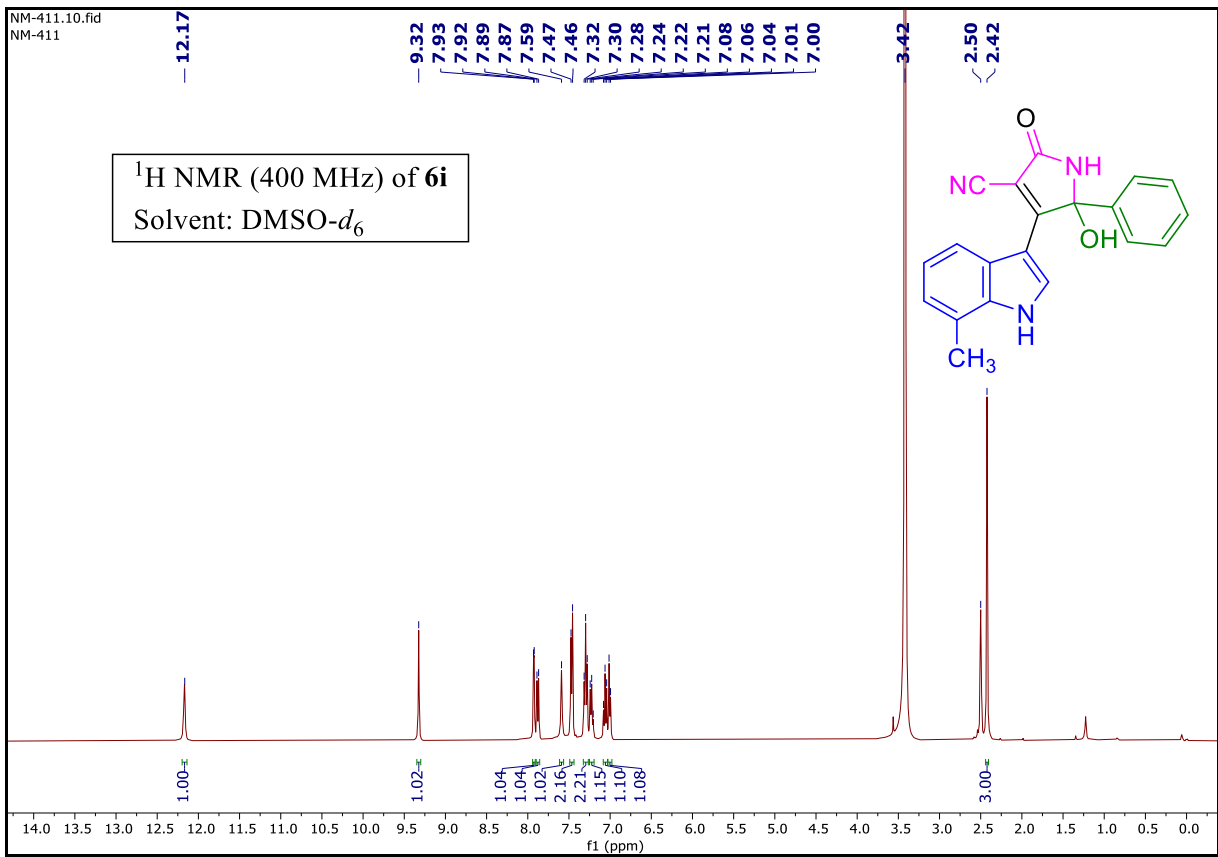


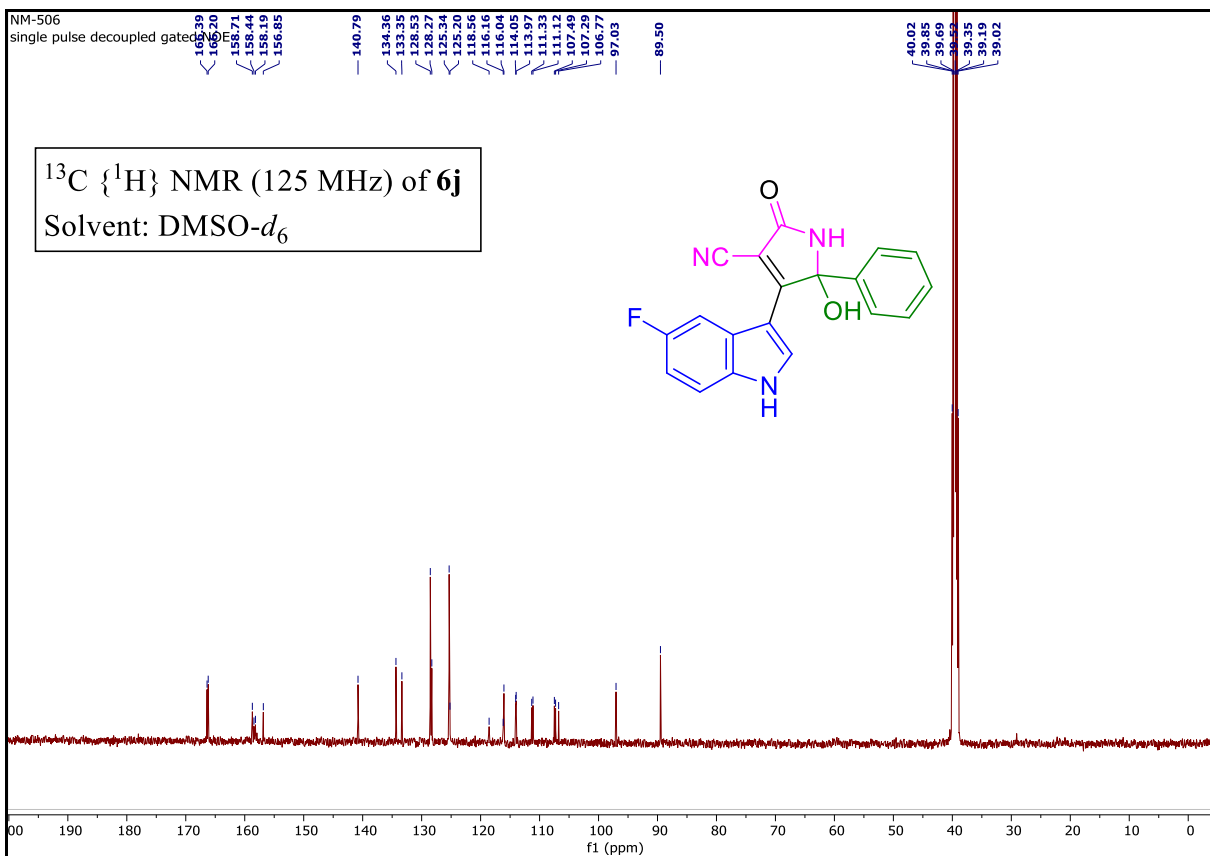
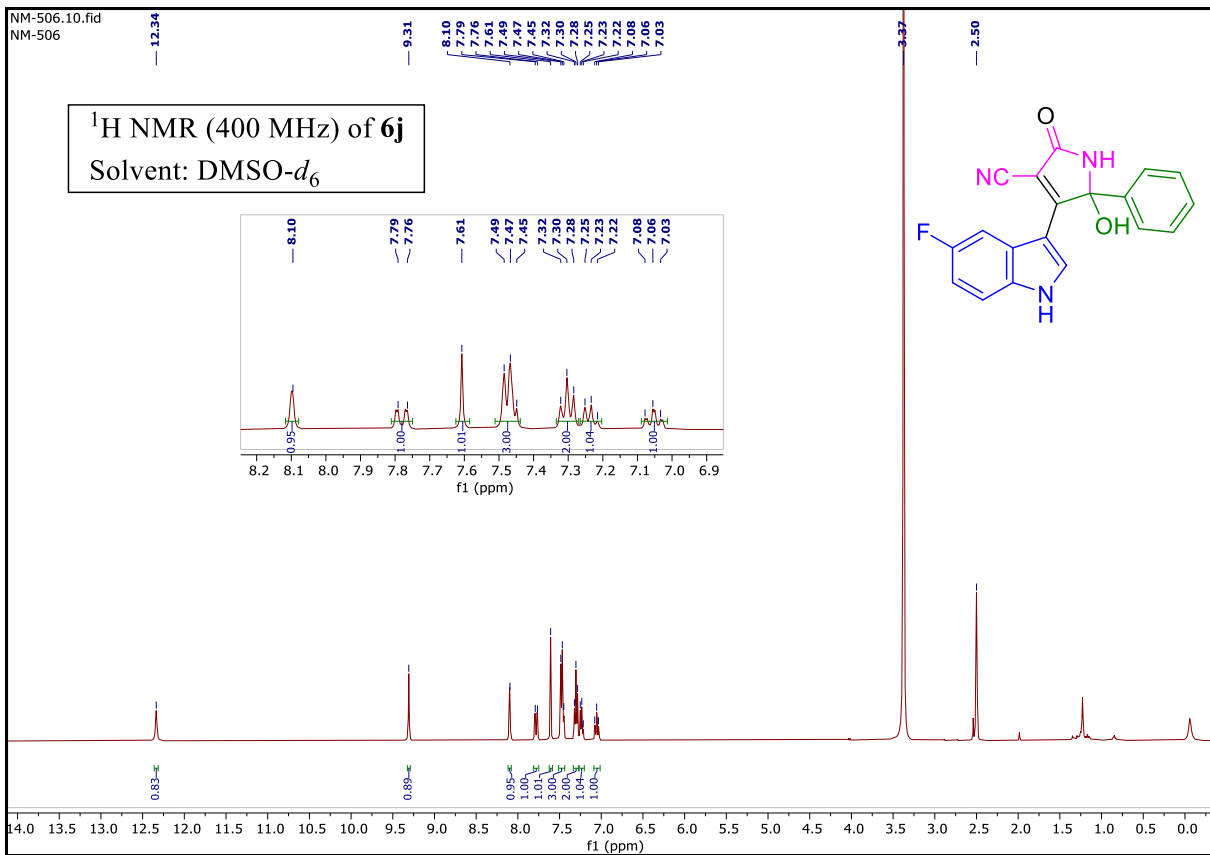


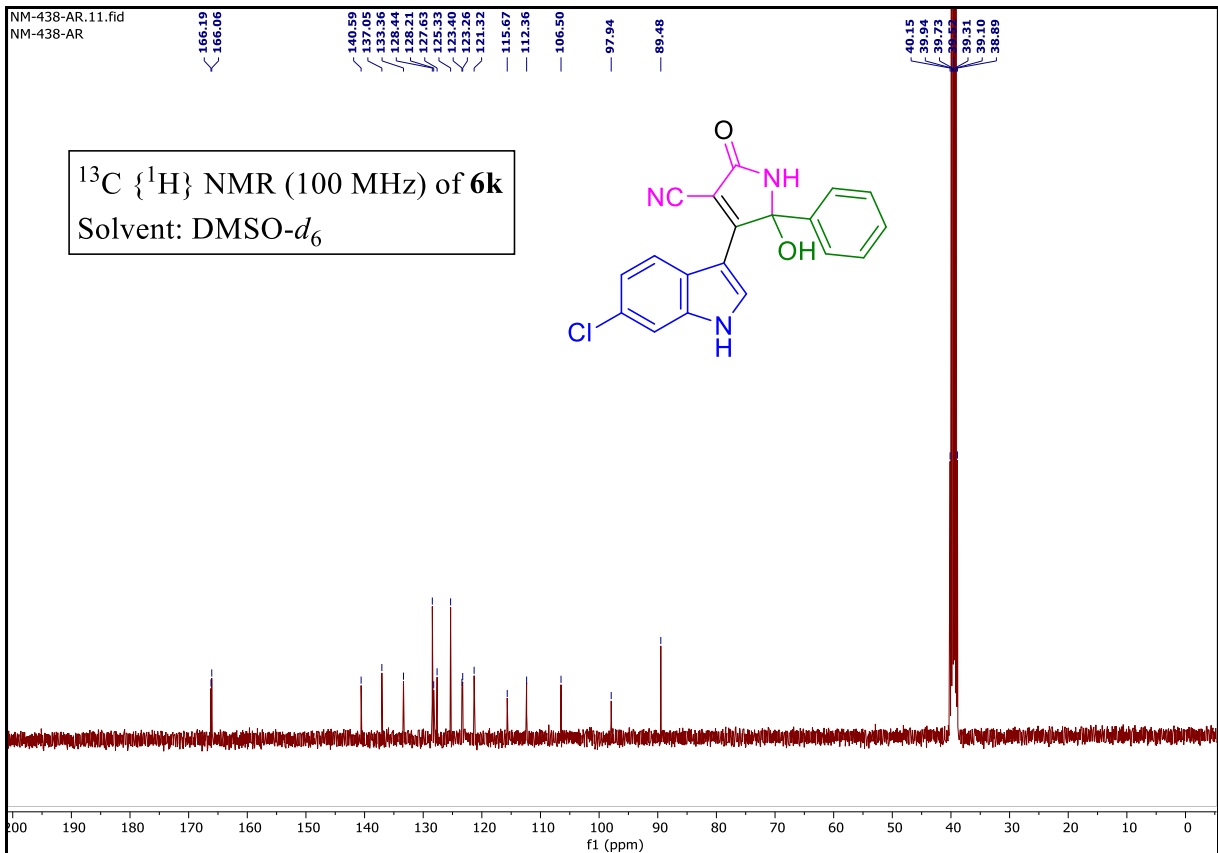
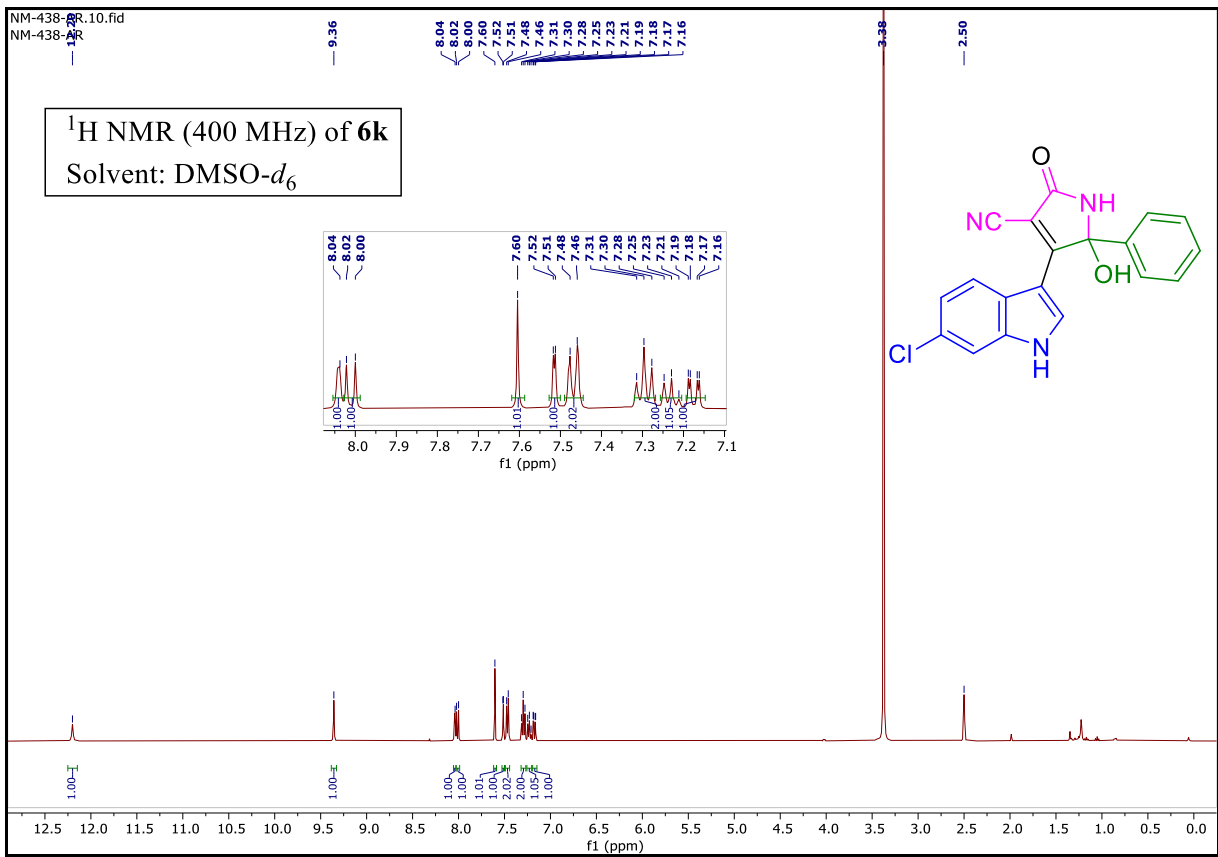


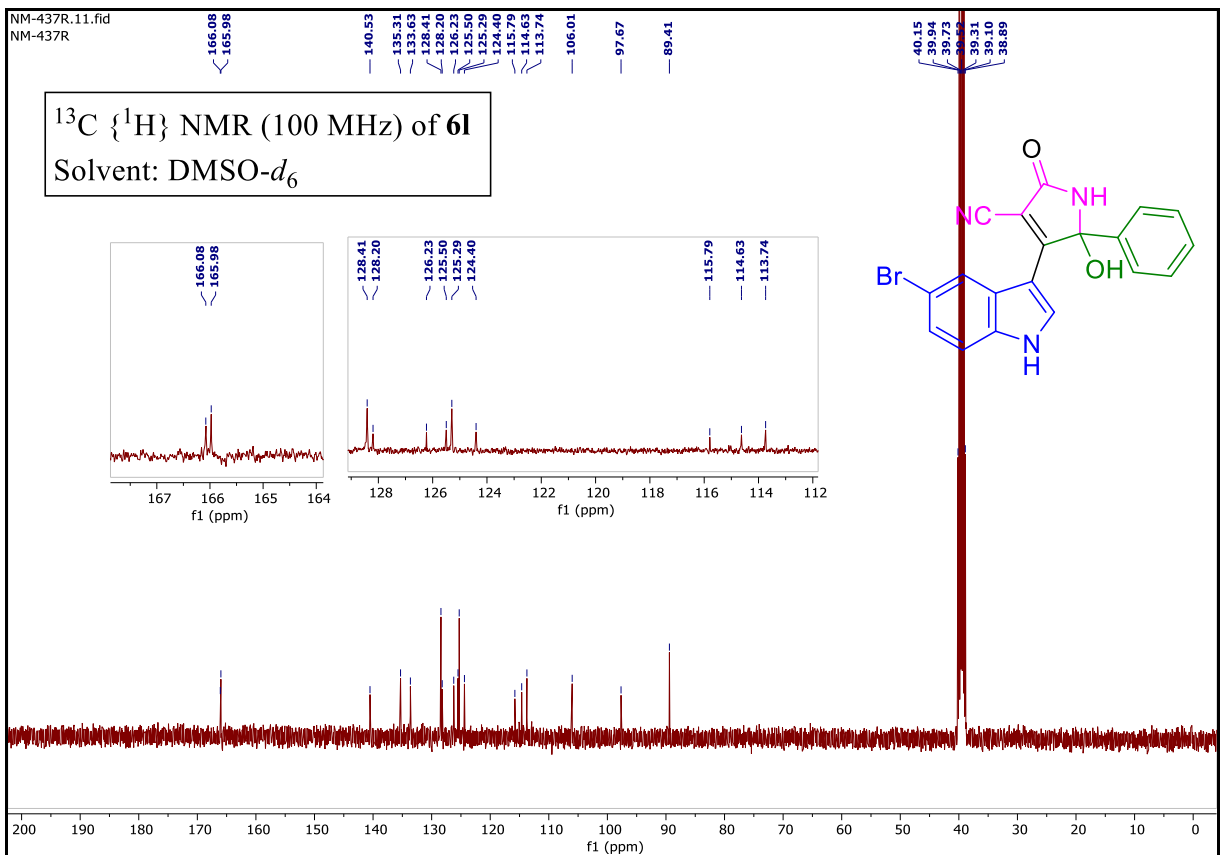
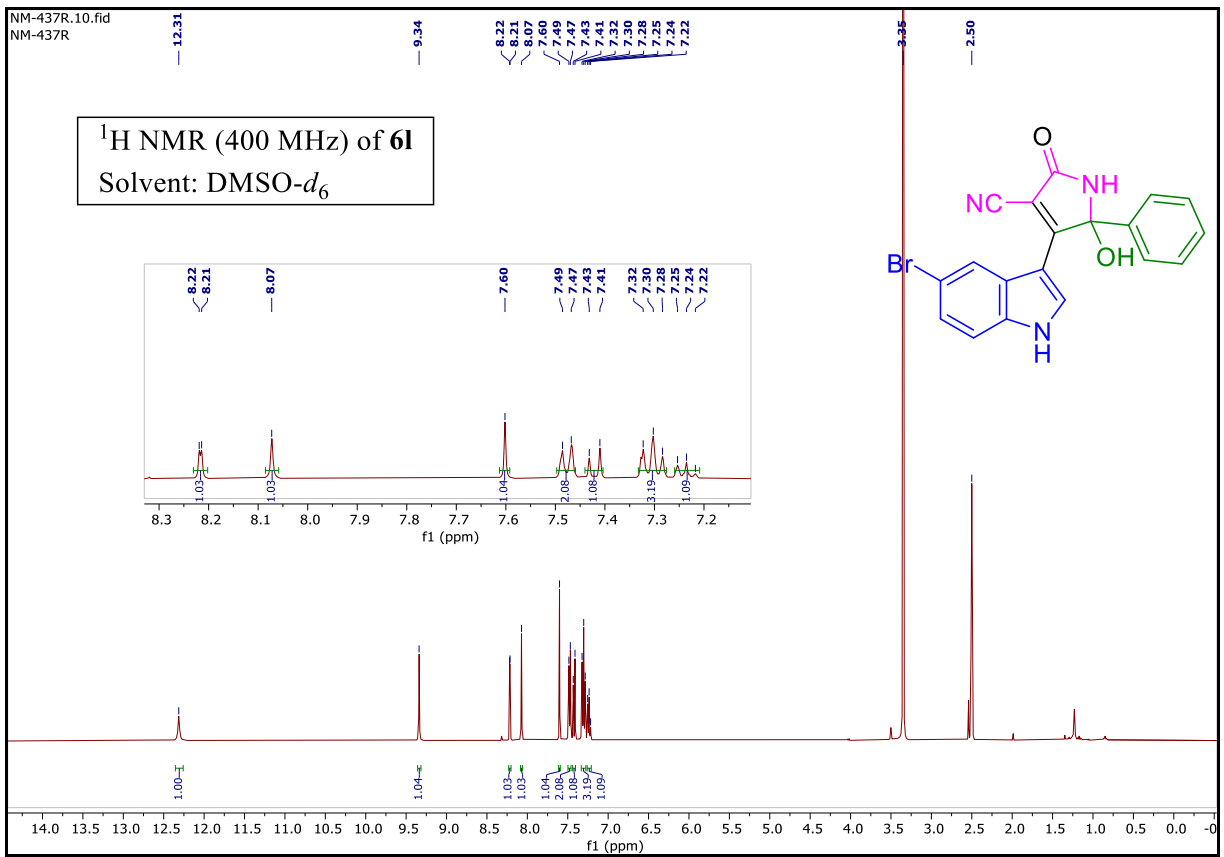


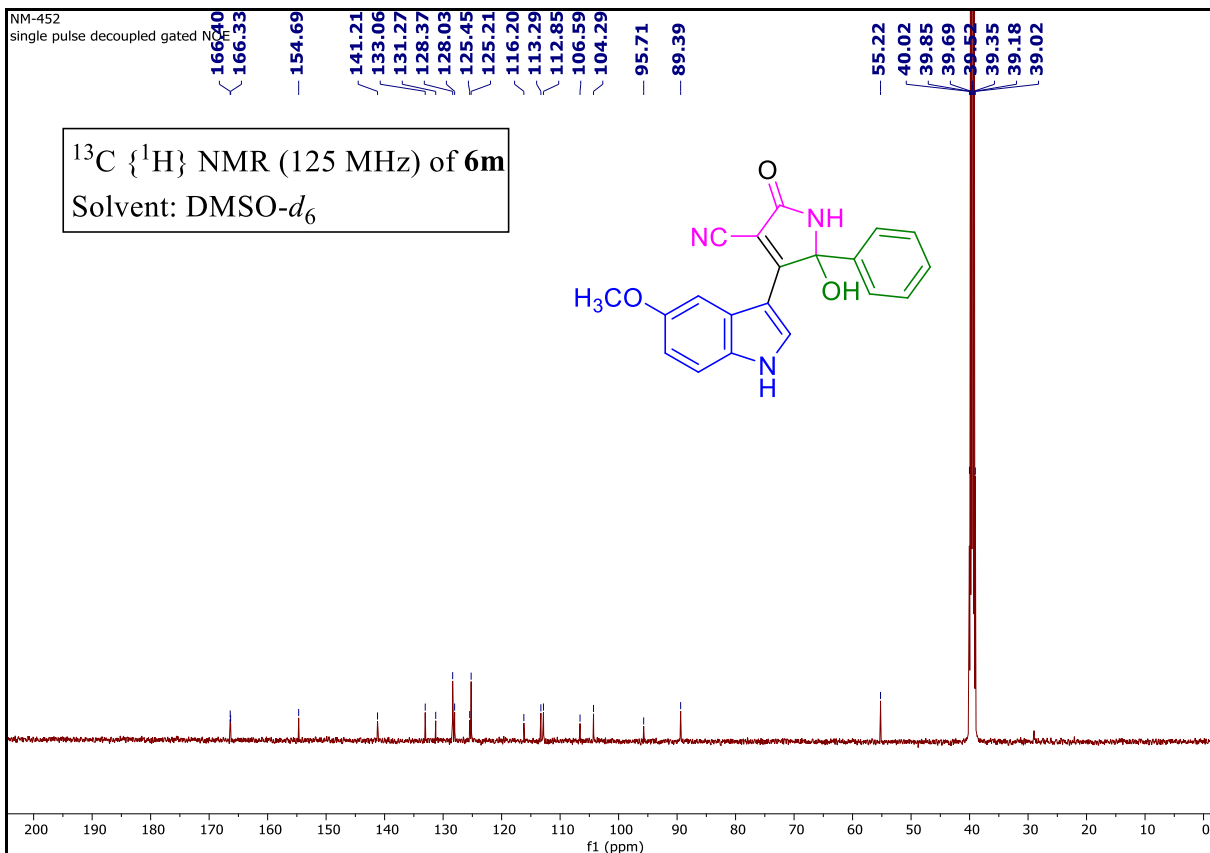
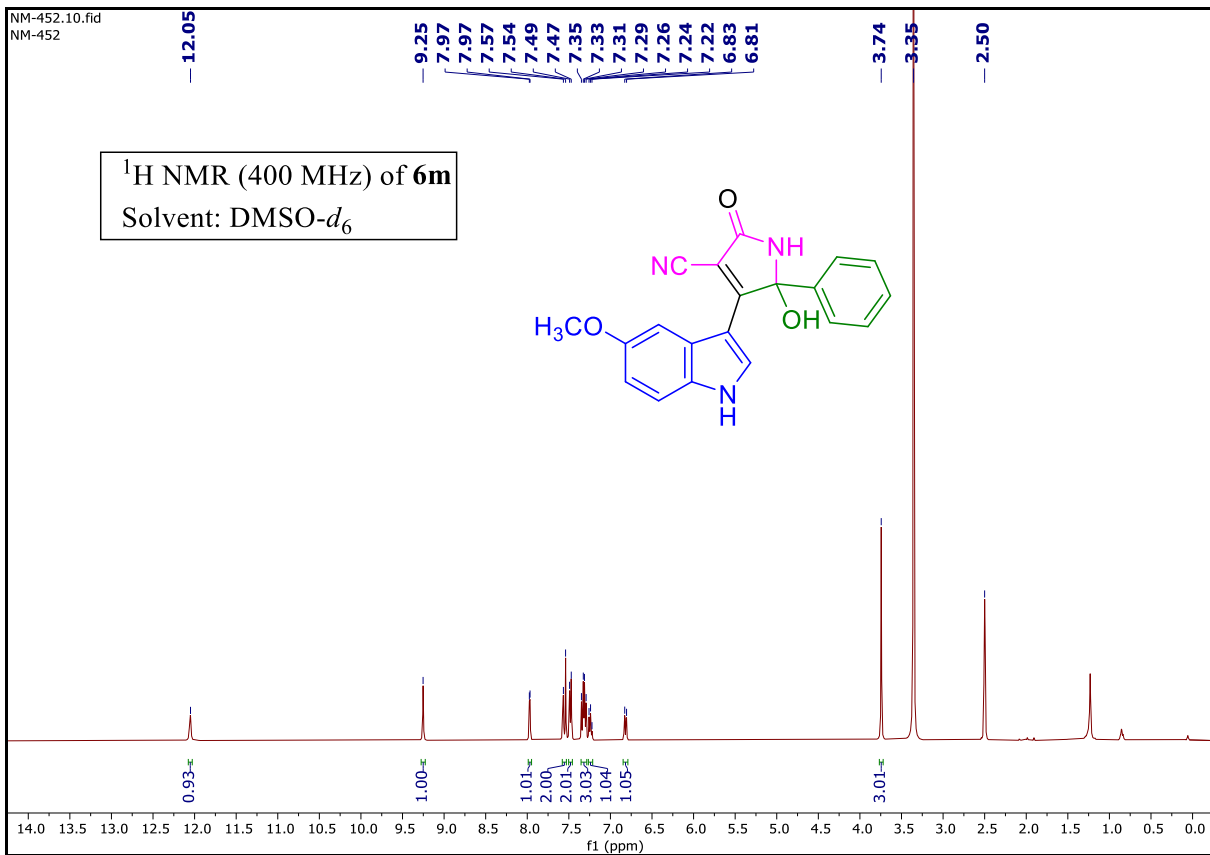


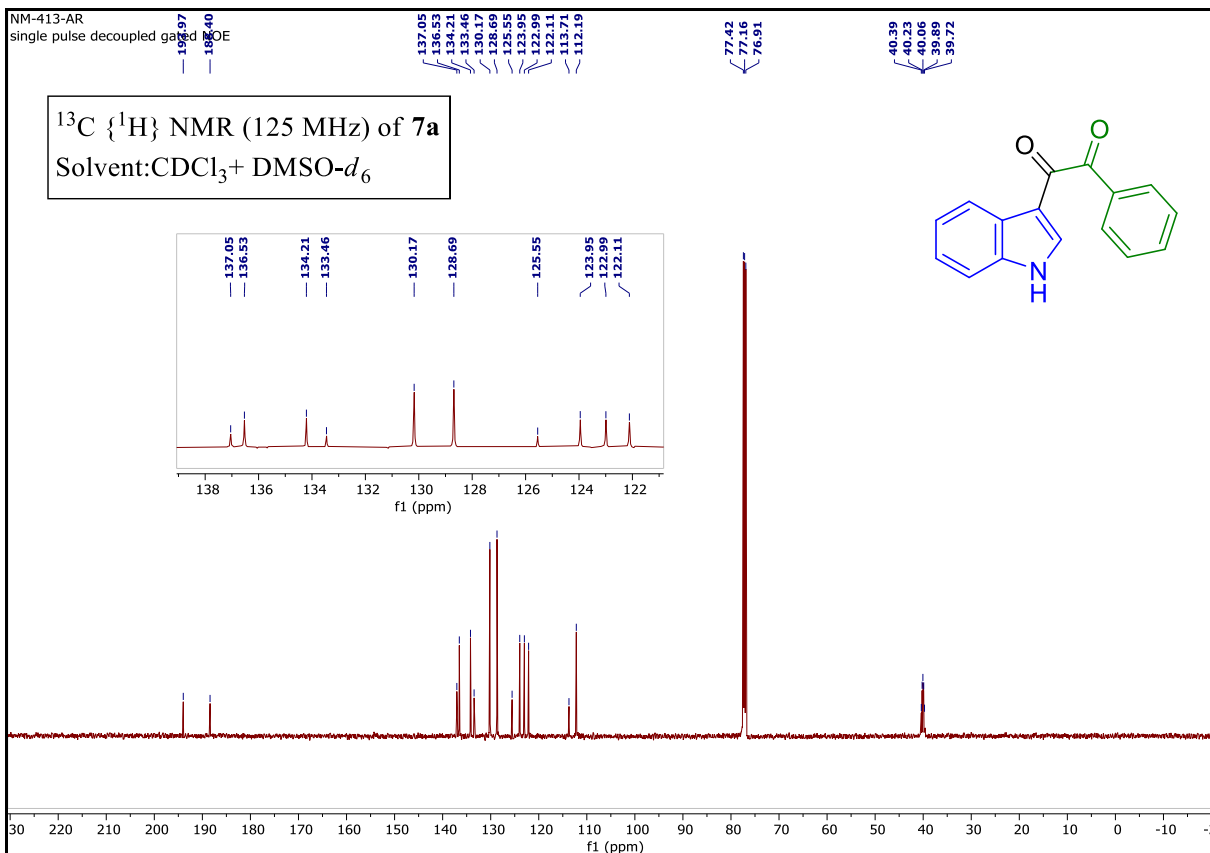
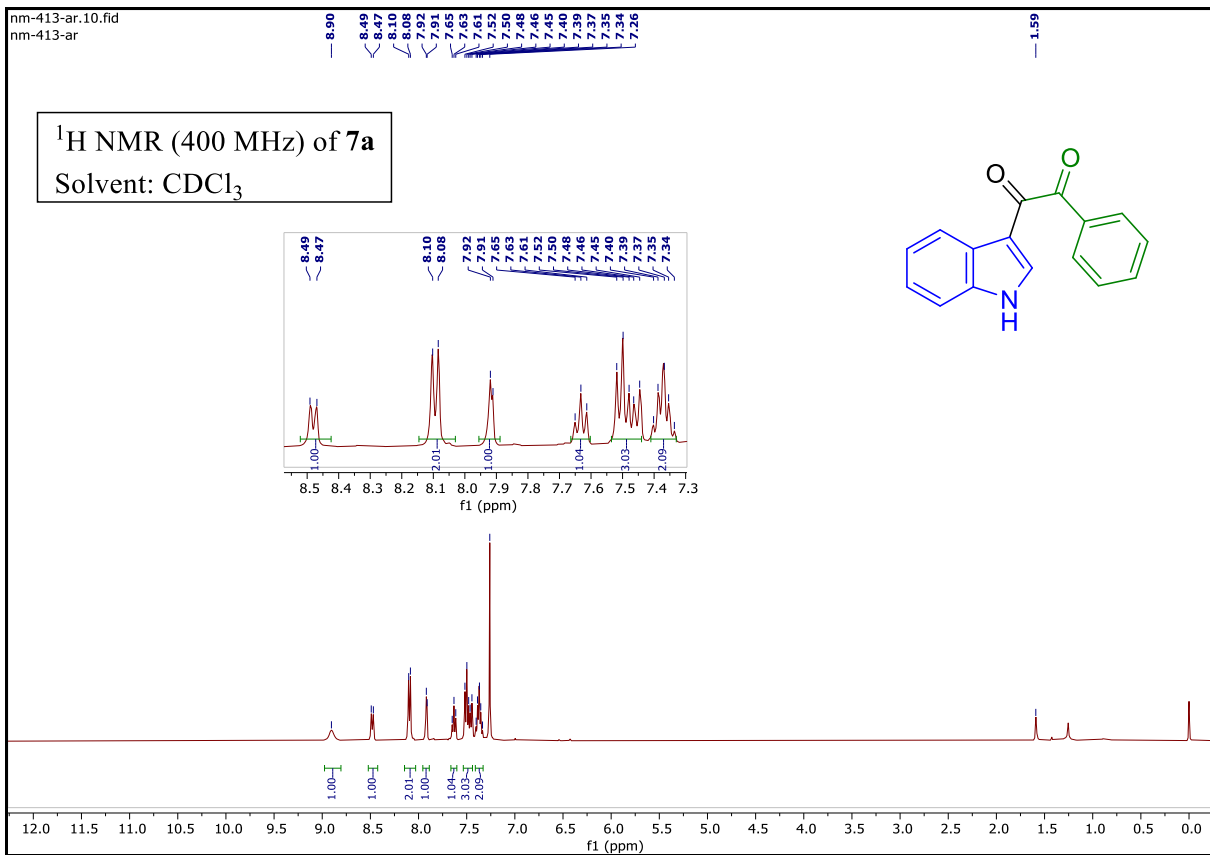


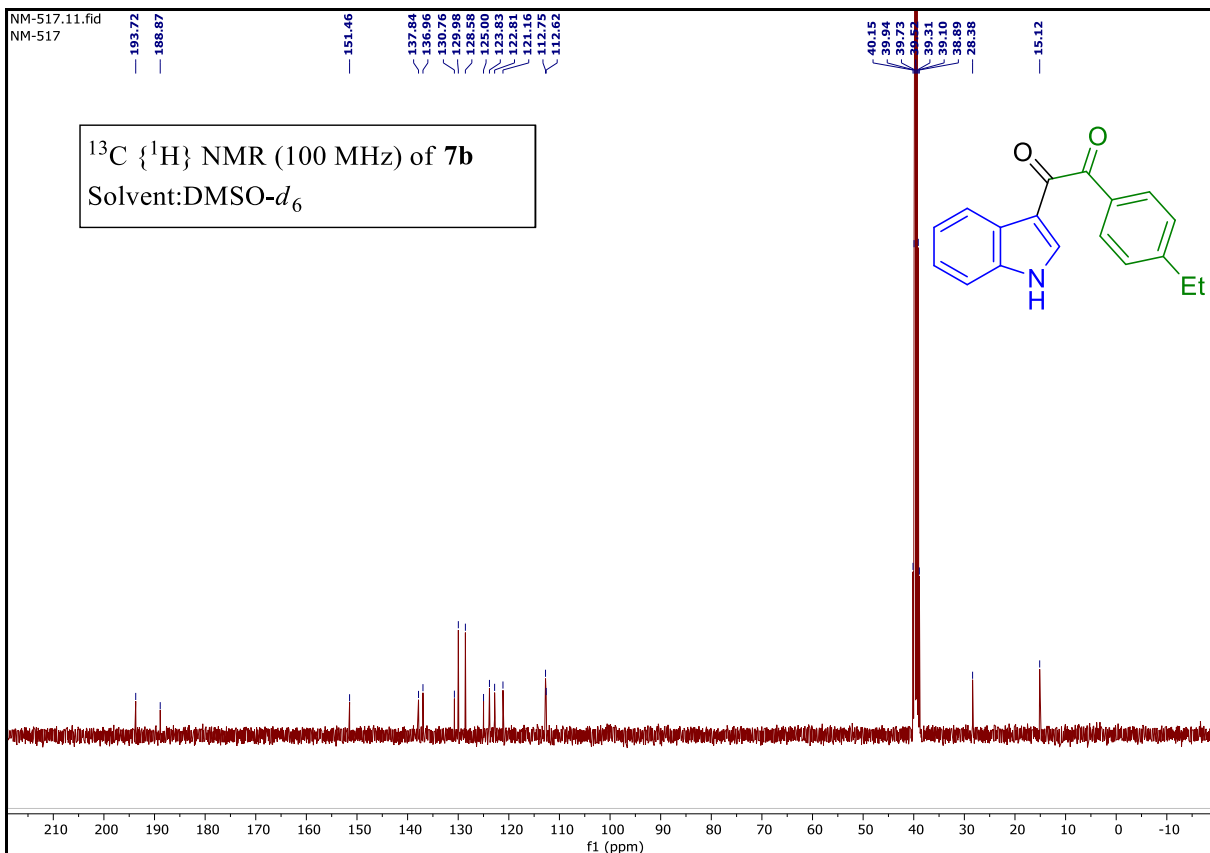
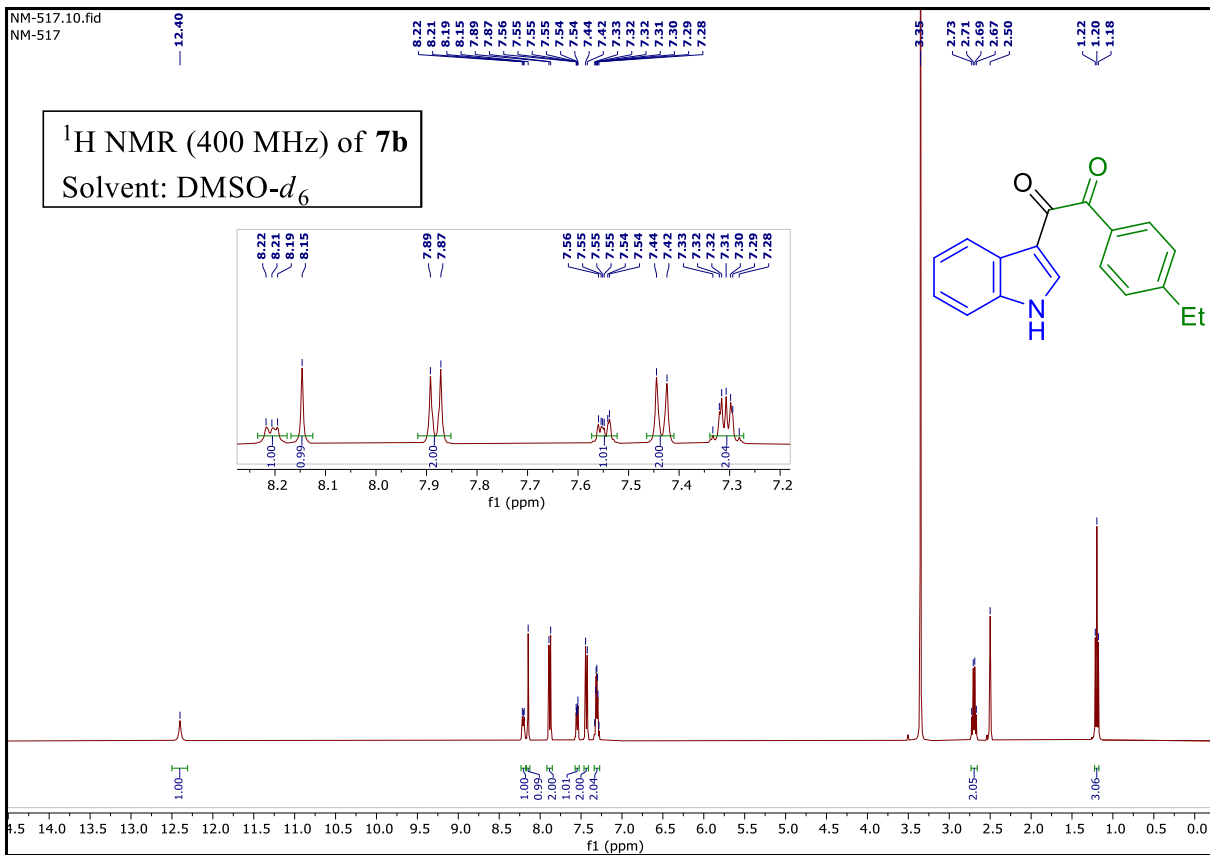


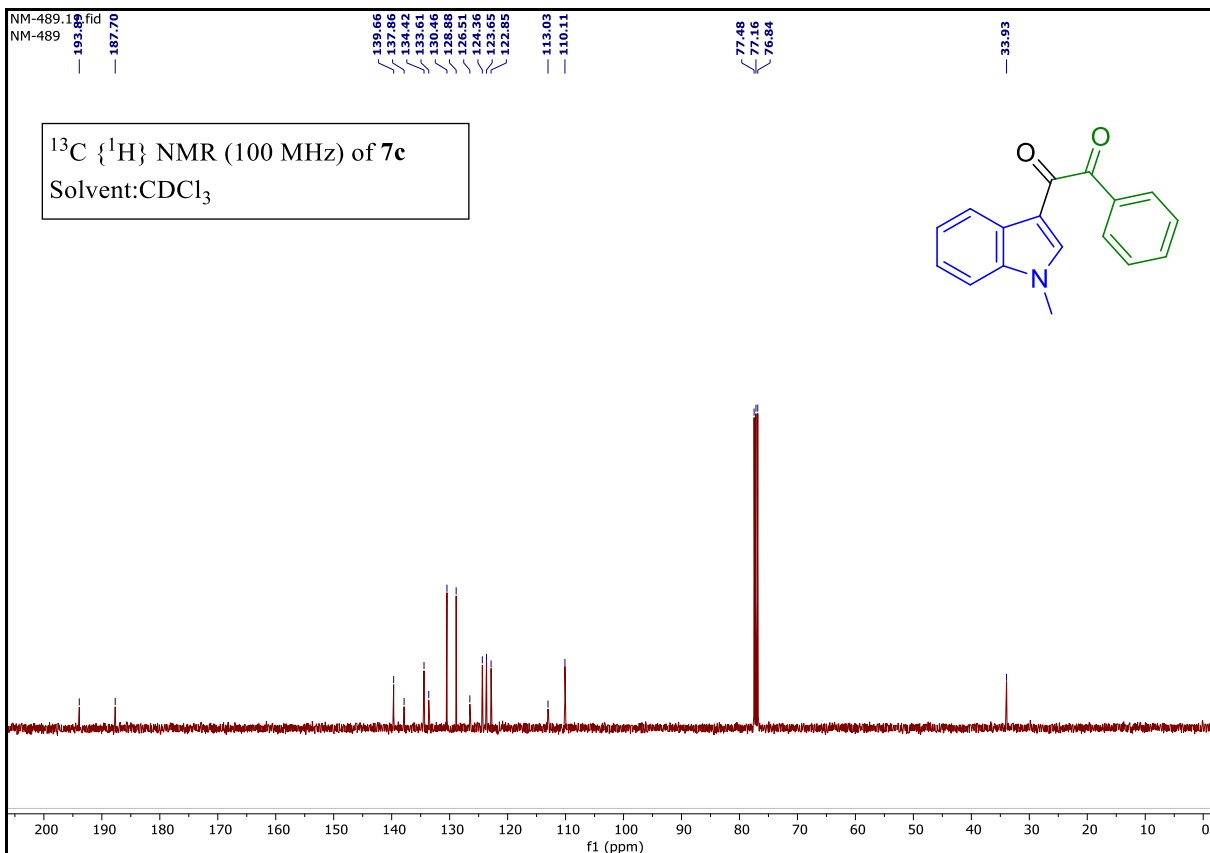
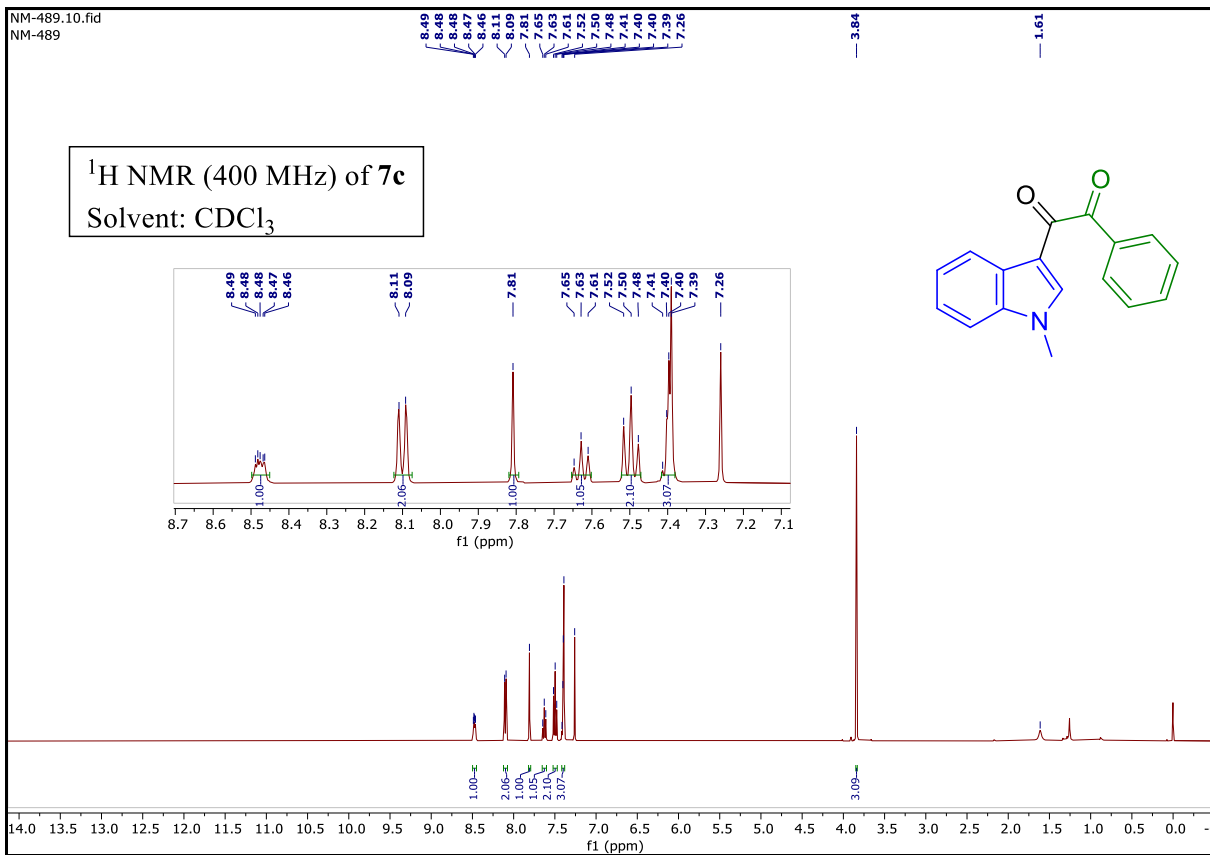


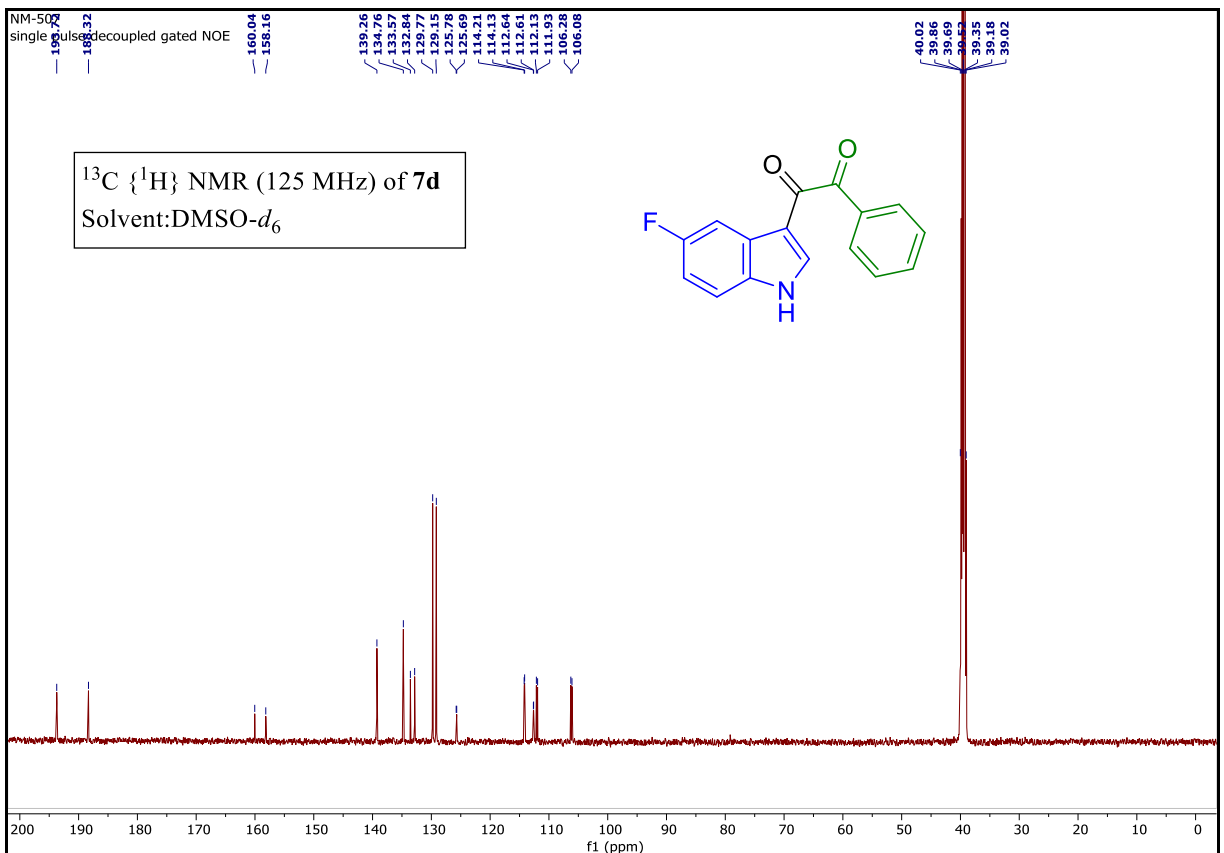
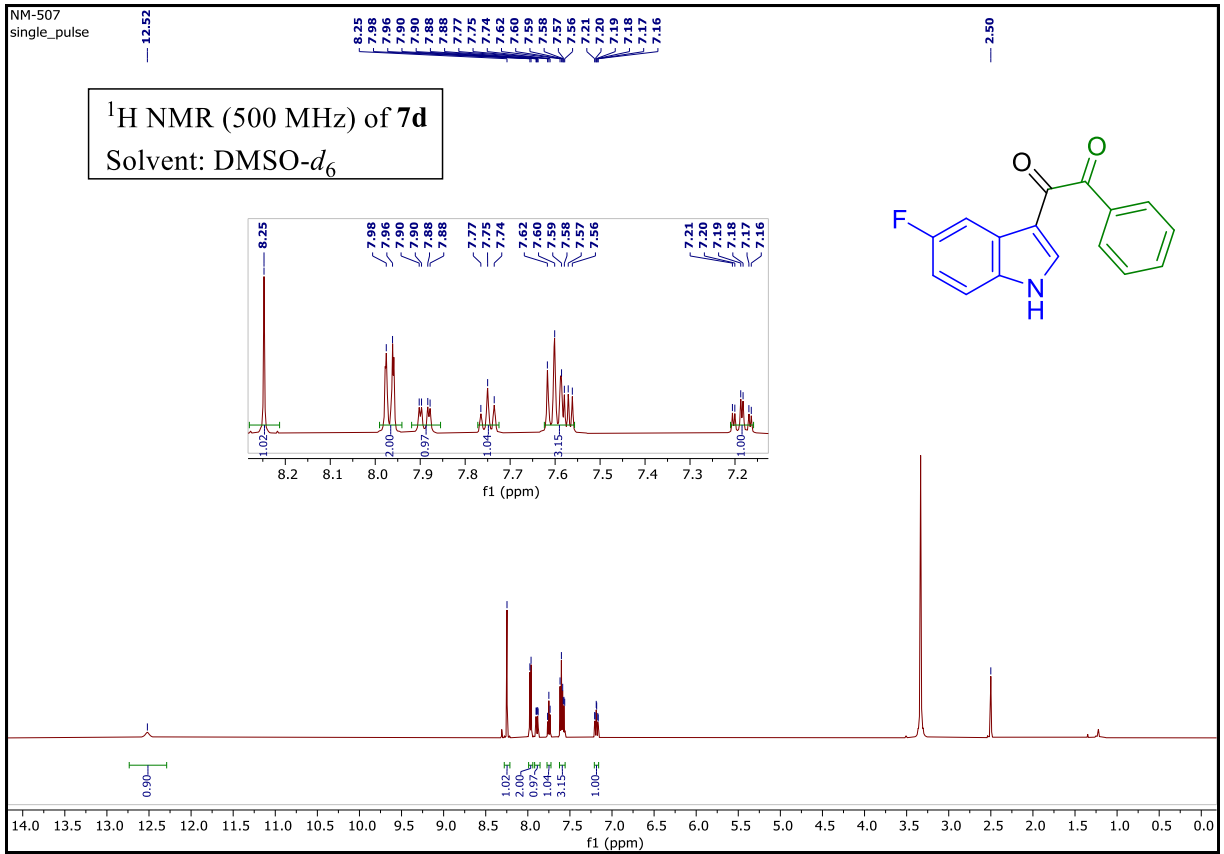


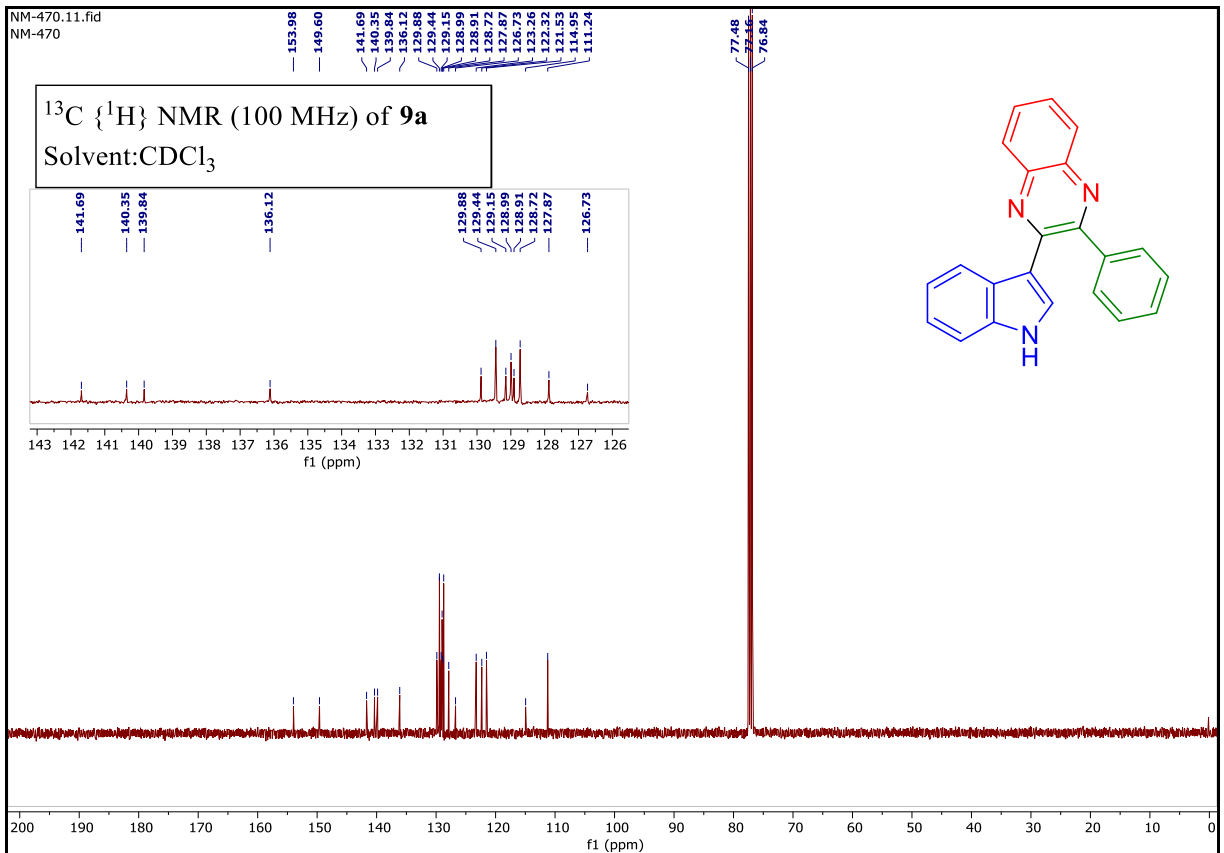
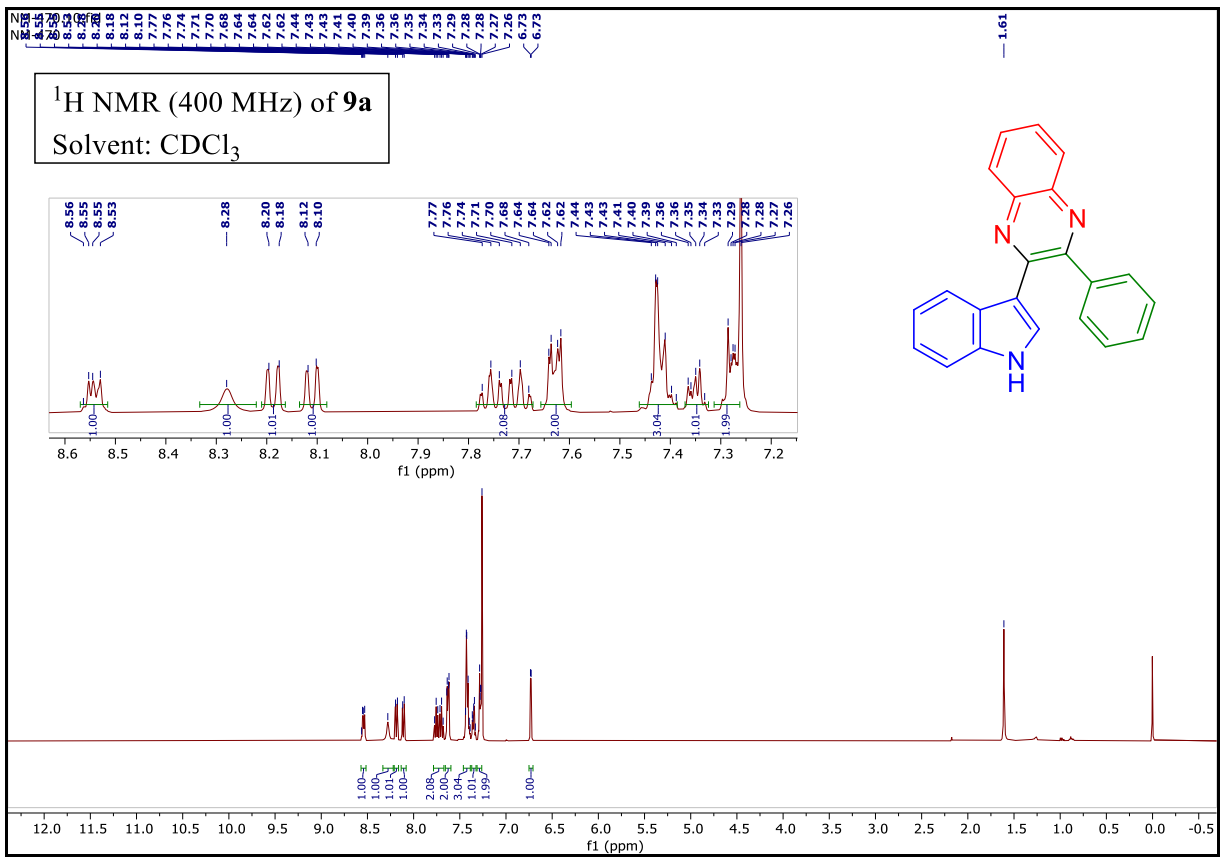


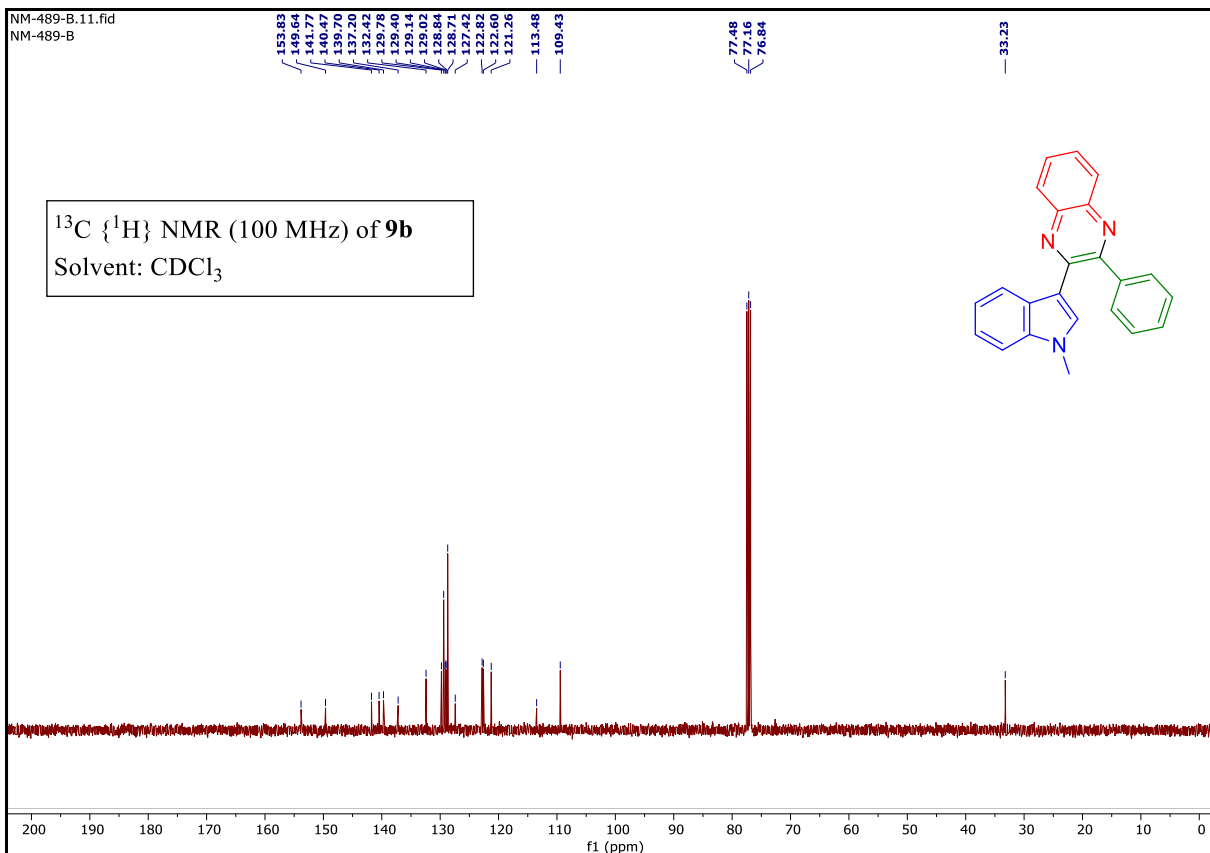
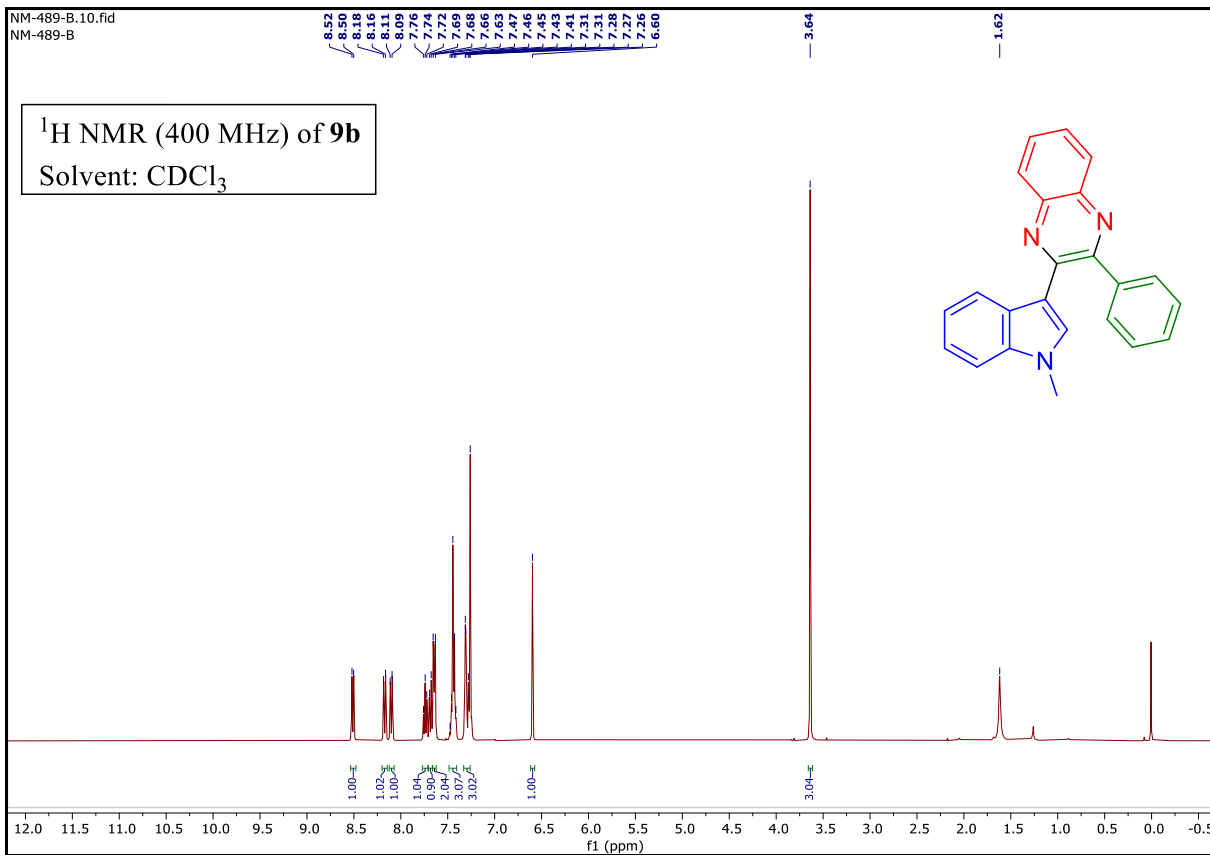




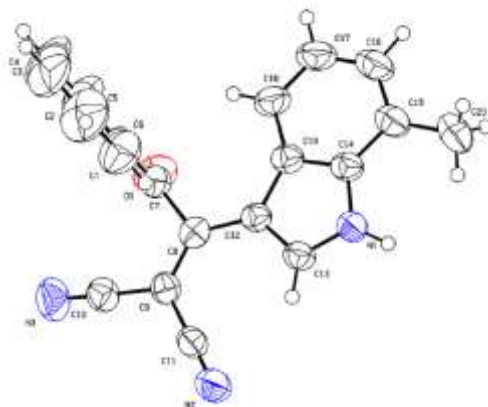








2. Crystallographic description of compound 5n



Ellipsoid plot

Table S1. Crystal data and structure refinement for 5n

CCDC No.	2376014
Empirical formula	C ₂₀ H ₁₃ N ₃ O
Formula weight	311.33
Temperature	296(2) K
Wavelength	0.71073 Å
Crystal system	Monoclinic
Space group	P 21/n
Unit cell dimensions	a = 13.5858(8) Å α = 90°. b = 7.0142(4) Å β = 95.905(2)°. c = 19.7041(11) Å γ = 90°.
Volume	1867.71(19) Å ³
Z	4
Density (calculated)	1.107 Mg/m ³
Absorption coefficient	0.071 mm ⁻¹
F(000)	648
Crystal size	0.184 x 0.128 x 0.075 mm ³
Theta range for data collection	3.272 to 24.998°.
Index ranges	-16 ≤ h ≤ 16, -8 ≤ k ≤ 8, -23 ≤ l ≤ 23
Reflections collected	49056
Independent reflections	3285 [R(int) = 0.0734]
Completeness to theta = 24.998°	99.8 %
Absorption correction	Semi-empirical from equivalents
Max. and min. transmission	0.9602 and 0.6988

Refinement method	Full-matrix least-squares on F ²
Data / restraints / parameters	3285 / 0 / 221
Goodness-of-fit on F ²	1.054
Final R indices [I>2sigma(I)]	R1 = 0.0614, wR2 = 0.1670
R indices (all data)	R1 = 0.0763, wR2 = 0.1784
Extinction coefficient	n/a
Largest diff. peak and hole	0.235 and -0.221 e.Å ⁻³

Table S2. Atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for 5n. U(eq) is defined as one third of the trace of the orthogonalized U^{ij} tensor.

	x	y	z	U(eq)
C(1)	7721(2)	4536(5)	3478(2)	89(1)
C(2)	8550(3)	4495(8)	3112(2)	123(2)
C(3)	8975(3)	2831(10)	2987(2)	136(2)
C(4)	8602(4)	1186(9)	3214(3)	146(2)
C(5)	7779(3)	1180(6)	3576(2)	106(1)
C(6)	7333(2)	2901(4)	3702(1)	62(1)
C(7)	6459(2)	2875(3)	4088(1)	52(1)
C(8)	5972(2)	4769(3)	4220(1)	44(1)
C(9)	5241(2)	5315(3)	3723(1)	46(1)
C(10)	4997(2)	4121(3)	3139(1)	54(1)
C(11)	4693(2)	7038(3)	3727(1)	50(1)
C(12)	6311(2)	5742(3)	4837(1)	43(1)
C(13)	5988(2)	7508(3)	5047(1)	47(1)
C(14)	7127(2)	6554(3)	5869(1)	47(1)
C(15)	7762(2)	6517(4)	6476(1)	58(1)
C(16)	8341(2)	4897(4)	6568(1)	68(1)
C(17)	8283(2)	3429(4)	6096(1)	71(1)
C(18)	7655(2)	3501(4)	5499(1)	60(1)
C(19)	7054(2)	5110(3)	5376(1)	47(1)
C(20)	7803(2)	8130(5)	6977(1)	83(1)
N(1)	6462(1)	7981(3)	5649(1)	52(1)
N(2)	4263(2)	8441(3)	3719(1)	68(1)
N(3)	4810(2)	3173(4)	2672(1)	80(1)
O(1)	6110(1)	1431(2)	4299(1)	76(1)

Table S3. Bond lengths [Å] for 5n

C(1)-C(6)	1.355(4)	C(1)-C(6)	1.355(4)
C(1)-C(2)	1.398(5)	C(1)-C(2)	1.398(5)
C(1)-H(1)	0.9300	C(1)-H(1)	0.9300
C(2)-C(3)	1.336(7)	C(2)-C(3)	1.336(7)
C(2)-H(2)	0.9300	C(2)-H(2)	0.9300
C(3)-C(4)	1.355(7)	C(3)-C(4)	1.355(7)
C(3)-H(3)	0.9300	C(3)-H(3)	0.9300
C(4)-C(5)	1.386(6)	C(4)-C(5)	1.386(6)
C(4)-H(4)	0.9300	C(4)-H(4)	0.9300
C(5)-C(6)	1.385(4)	C(5)-C(6)	1.385(4)
C(5)-H(5)	0.9300	C(5)-H(5)	0.9300
C(6)-C(7)	1.474(3)	C(6)-C(7)	1.474(3)
C(7)-O(1)	1.210(3)	C(7)-O(1)	1.210(3)
C(7)-C(8)	1.519(3)	C(7)-C(8)	1.519(3)
C(8)-C(9)	1.375(3)	C(8)-C(9)	1.375(3)
C(8)-C(12)	1.428(3)	C(8)-C(12)	1.428(3)
C(9)-C(11)	1.421(3)	C(9)-C(11)	1.421(3)
C(9)-C(10)	1.435(3)	C(9)-C(10)	1.435(3)
C(10)-N(3)	1.142(3)	C(10)-N(3)	1.142(3)
C(11)-N(2)	1.143(3)		

Table S4. Bond angles [°] for 5n

C(6)-C(1)-C(2)	120.8(4)	C(9)-C(8)-C(12)	128.2(2)	C(15)-C(16)-C(17)	122.1(2)
C(6)-C(1)-H(1)	119.6	C(9)-C(8)-C(7)	114.64(18)	C(15)-C(16)-H(16)	119.0
C(2)-C(1)-H(1)	119.6	C(12)-C(8)-C(7)	117.12(18)	C(17)-C(16)-H(16)	119.0
C(3)-C(2)-C(1)	119.9(4)	C(8)-C(9)-C(11)	125.01(19)	C(18)-C(17)-C(16)	122.1(2)
C(3)-C(2)-H(2)	120.0	C(8)-C(9)-C(10)	120.0(2)	C(18)-C(17)-H(17)	118.9
C(1)-C(2)-H(2)	120.0	C(11)-C(9)-C(10)	115.00(19)	C(16)-C(17)-H(17)	118.9
C(2)-C(3)-C(4)	120.0(4)	N(3)-C(10)-C(9)	179.4(3)	C(17)-C(18)-C(19)	118.4(2)
C(2)-C(3)-H(3)	120.0	N(2)-C(11)-C(9)	178.5(3)	C(17)-C(18)-H(18)	120.8
C(4)-C(3)-H(3)	120.0	C(13)-C(12)-C(8)	126.51(18)	C(19)-C(18)-H(18)	120.8
C(3)-C(4)-C(5)	121.4(4)	C(13)-C(12)-C(19)	105.65(17)	C(14)-C(19)-C(18)	117.7(2)
C(3)-C(4)-H(4)	119.3	C(8)-C(12)-C(19)	127.85(19)	C(14)-C(19)-C(12)	106.22(18)
C(5)-C(4)-H(4)	119.3	N(1)-C(13)-C(12)	110.42(18)	C(18)-C(19)-C(12)	136.0(2)
C(6)-C(5)-C(4)	118.7(4)	N(1)-C(13)-H(13)	124.8	C(15)-C(20)-H(20A)	109.5
C(6)-C(5)-H(5)	120.6	C(12)-C(13)-H(13)	124.8	C(15)-C(20)-H(20B)	109.5
C(4)-C(5)-H(5)	120.6	N(1)-C(14)-C(19)	108.00(18)	H(20A)-C(20)-H(20B)	109.5
C(1)-C(6)-C(5)	119.1(3)	N(1)-C(14)-C(15)	127.1(2)	C(15)-C(20)-H(20C)	109.5
C(1)-C(6)-C(7)	122.6(2)	C(19)-C(14)-C(15)	124.9(2)	H(20A)-C(20)-H(20C)	109.5
C(5)-C(6)-C(7)	118.2(3)	C(16)-C(15)-C(14)	114.8(2)	H(20B)-C(20)-H(20C)	109.5
O(1)-C(7)-C(6)	123.5(2)	C(16)-C(15)-C(20)	123.4(2)	C(13)-N(1)-C(14)	109.71(19)
O(1)-C(7)-C(8)	118.7(2)	C(14)-C(15)-C(20)	121.8(2)	C(13)-N(1)-H(1A)	126.5(17)
C(6)-C(7)-C(8)	117.8(2)	C(15)-C(16)-C(17)	122.1(2)	C(14)-N(1)-H(1A)	123.7(17)

Table S5. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for 5n. The anisotropic displacement factor exponent takes the form: $-2 \pi^2 [h^2 a^{*2} U_{11} + \dots + 2 h k a^* b^* U_{12}]$

	U11	U22	U33	U23	U13	U12
C(1)	76(2)	98(2)	98(2)	-6(2)	34(2)	-3(2)
C(2)	92(3)	163(4)	123(3)	-11(3)	54(2)	-18(3)
C(3)	81(3)	216(6)	119(3)	-51(4)	46(2)	6(3)

C(4)	117(4)	166(5)	163(4)	-60(4)	56(3)	38(3)
C(5)	98(2)	98(3)	125(3)	-35(2)	33(2)	21(2)
C(6)	56(1)	71(2)	59(1)	-15(1)	5(1)	5(1)
C(7)	53(1)	46(1)	55(1)	-4(1)	0(1)	-2(1)
C(8)	44(1)	42(1)	49(1)	1(1)	11(1)	-4(1)
C(9)	49(1)	44(1)	45(1)	-1(1)	4(1)	-3(1)
C(10)	53(1)	54(1)	56(1)	-4(1)	3(1)	-6(1)
C(11)	51(1)	51(1)	45(1)	-2(1)	-5(1)	-1(1)
C(12)	42(1)	43(1)	44(1)	4(1)	6(1)	2(1)
C(13)	47(1)	44(1)	47(1)	4(1)	-3(1)	4(1)
C(14)	45(1)	50(1)	47(1)	8(1)	4(1)	4(1)
C(15)	51(1)	72(2)	50(1)	9(1)	-1(1)	2(1)
C(16)	57(1)	89(2)	54(1)	18(1)	-6(1)	10(1)
C(17)	64(2)	76(2)	70(2)	19(1)	2(1)	26(1)
C(18)	59(1)	58(2)	63(1)	8(1)	7(1)	18(1)
C(19)	44(1)	48(1)	48(1)	9(1)	7(1)	4(1)
C(20)	87(2)	96(2)	60(2)	-12(2)	-20(1)	3(2)
N(1)	57(1)	48(1)	48(1)	-4(1)	-4(1)	7(1)
N(2)	72(1)	61(1)	68(1)	-5(1)	-14(1)	11(1)
N(3)	82(2)	83(2)	71(2)	-26(1)	-2(1)	-13(1)
O(1)	79(1)	46(1)	104(2)	7(1)	16(1)	0(1)

Table S6. Hydrogen coordinates ($\times 10^4$) and isotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for 5n

	x	y	z	U(eq)
H(1)	7433	5699	3568	107
H(2)	8805	5625	2956	148
H(3)	9527	2803	2745	163
H(4)	8905	36	3125	175
H(5)	7532	42	3731	127
H(13)	5509	8253	4802	56
H(16)	8783	4791	6959	81
H(17)	8680	2359	6184	85
H(18)	7633	2505	5186	72
H(20A)	7342	9101	6812	125
H(20B)	7635	7668	7409	125

H(20C)	8460	8653	7032	125
H(1A)	6350(20)	9050(40)	5900(14)	75(8)

Table S7. Torsion angles [°] for 5n

C(6)-C(1)-C(2)-C(3)	-0.8(6)		C(8)-C(12)-C(13)-N(1)	179.8(2)
C(1)-C(2)-C(3)-C(4)	0.1(7)		C(19)-C(12)-C(13)-N(1)	0.1(2)
C(2)-C(3)-C(4)-C(5)	0.1(8)		N(1)-C(14)-C(15)-C(16)	179.3(2)
C(3)-C(4)-C(5)-C(6)	0.4(7)		C(19)-C(14)-C(15)-C(16)	-0.1(3)
C(2)-C(1)-C(6)-C(5)	1.3(5)		N(1)-C(14)-C(15)-C(20)	-1.0(4)
C(2)-C(1)-C(6)-C(7)	-179.9(3)		C(19)-C(14)-C(15)-C(20)	179.5(2)
C(4)-C(5)-C(6)-C(1)	-1.1(5)		C(14)-C(15)-C(16)-C(17)	-0.6(4)
C(4)-C(5)-C(6)-C(7)	-179.9(3)		C(20)-C(15)-C(16)-C(17)	179.8(3)
C(1)-C(6)-C(7)-O(1)	-179.1(3)		C(15)-C(16)-C(17)-C(18)	1.0(4)
C(5)-C(6)-C(7)-O(1)	-0.2(4)		C(16)-C(17)-C(18)-C(19)	-0.7(4)
C(1)-C(6)-C(7)-C(8)	1.4(4)		N(1)-C(14)-C(19)-C(18)	-179.1(2)
C(5)-C(6)-C(7)-C(8)	-179.7(2)		C(15)-C(14)-C(19)-C(18)	0.4(3)
O(1)-C(7)-C(8)-C(9)	-89.8(3)		N(1)-C(14)-C(19)-C(12)	0.5(2)
C(6)-C(7)-C(8)-C(9)	89.7(2)		C(15)-C(14)-C(19)-C(12)	-180.0(2)
O(1)-C(7)-C(8)- C(12)	89.5(3)		C(17)-C(18)-C(19)-C(14)	0.0(3)
C(6)-C(7)-C(8)- C(12)	-91.0(2)		C(17)-C(18)-C(19)-C(12)	-179.5(2)
C(12)-C(8)-C(9)- C(11)	2.0(4)		C(13)-C(12)-C(19)-C(14)	-0.4(2)
C(7)-C(8)-C(9)- C(11)	-178.8(2)		C(8)-C(12)-C(19)-C(14)	179.90(19)
C(12)-C(8)-C(9)- C(10)	-179.3(2)		C(13)-C(12)-C(19)-C(18)	179.1(3)
C(7)-C(8)-C(9)- C(10)	-0.1(3)		C(8)-C(12)-C(19)-C(18)	-0.6(4)
C(9)-C(8)-C(12)- C(13)	-2.8(3)		C(12)-C(13)-N(1)-C(14)	0.2(3)
C(7)-C(8)-C(12)- C(13)	178.0(2)		C(19)-C(14)-N(1)-C(13)	-0.5(2)
C(9)-C(8)-C(12)-	176.8(2)		C(15)-C(14)-N(1)-C(13)	-180.0(2)

C(19)			
C(7)-C(8)-C(12)- C(19)	-2.3(3)		

Symmetry transformations used to generate equivalent atoms:

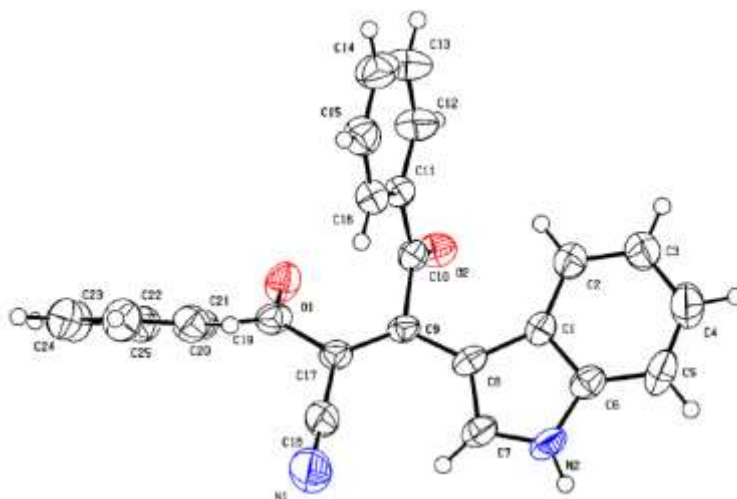
Table S8. Hydrogen bonds for 5n [Å and °].

D-H...A	d(D-H)	d(H...A)	d(D...A)	<(DHA)
C(13)-H(13)...N(2)	0.93	2.59	3.392(3)	145.1
C(13)-H(13)...O(1)#1	0.93	2.60	3.134(3)	116.6
N(1)-H(1A)...N(2)#2	0.92(3)	2.12(3)	3.012(3)	164(2)

Symmetry transformations used to generate equivalent atoms:

#1 x,y+1,z #2 -x+1,-y+2,-z+1

3. Crystallographic description of compound 5v



Ellipsoid plot

Table S9. Crystal data and structure refinement for 5v	
CCDC no.	2354497
Empirical formula	C ₂₅ H ₁₆ N ₂ O ₂
Formula weight	376.40
Temperature/K	298
Crystal system	triclinic
Space group	P-1
a/Å	9.2194(12)
b/Å	10.0506(14)
c/Å	10.4838(14)
α/°	101.017(8)

$\beta/^\circ$	101.391(7)
$\gamma/^\circ$	93.642(7)
Volume/ \AA^3	929.6(2)
Z	2
$\rho_{\text{calc}}/\text{g/cm}^3$	1.345
μ/mm^{-1}	0.086
F(000)	392.0
Crystal size/ mm^3	$0.089 \times 0.056 \times 0.045$
Radiation	MoK α ($\lambda = 0.71073$)
2 Θ range for data collection/ $^\circ$	4.532 to 50.002
Index ranges	$-10 \leq h \leq 10, -11 \leq k \leq 11, -12 \leq l \leq 12$
Reflections collected	12101
Independent reflections	3197 [$R_{\text{int}} = 0.0730, R_{\text{sigma}} = 0.0826$]
Data/restraints/parameters	3197/0/264
Goodness-of-fit on F^2	1.242
Final R indexes [$I \geq 2\sigma(I)$]	$R_1 = 0.1078, wR_2 = 0.1599$
Final R indexes [all data]	$R_1 = 0.1669, wR_2 = 0.1812$
Largest diff. peak/hole / $e \text{\AA}^{-3}$	0.27/-0.23

Table S10. Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 5v. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	x	y	z	U(eq)
O ₂	2656(4)	7362(4)	420(3)	44.7(10)
O ₁	969(4)	5335(4)	1399(4)	52.5(11)
N ₂	7817(5)	5821(5)	1198(4)	45.5(12)
C ₁₀	3147(5)	7205(5)	1531(5)	34.6(12)
C ₉	4075(5)	6031(5)	1734(5)	32.8(12)
C ₈	5564(5)	6252(5)	1541(5)	35.2(12)
C ₁	6307(5)	7484(5)	1325(5)	36.2(12)
C ₁₁	3012(5)	8194(5)	2734(5)	38.3(13)
C ₂₀	1290(6)	4062(5)	3072(5)	37.3(13)
C ₁₇	3444(5)	4952(5)	2122(5)	36.3(13)
C ₁₉	1843(5)	4812(5)	2149(5)	41.0(13)
C ₇	6569(6)	5288(6)	1442(5)	43.4(14)
C ₂	5966(6)	8818(5)	1292(5)	46.6(14)
C ₆	7714(6)	7166(6)	1124(5)	41.4(14)
C ₁₆	3576(6)	8023(5)	4020(5)	43.9(14)
C ₂₅	-115(6)	3361(5)	2688(6)	47.6(15)
C ₃	6994(6)	9731(6)	1055(6)	55.1(16)
C ₁₈	4293(6)	3897(6)	2491(6)	53.1(16)
C ₂₁	2079(6)	4171(6)	4352(5)	49.6(15)
C ₁₂	2356(6)	9370(6)	2576(6)	53.2(16)

Table S10. Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 5v. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{IJ} tensor.

Atom	x	y	z	$U(\text{eq})$
C ₄	8384(7)	9376(7)	868(6)	61.6(18)
C ₁₅	3441(6)	8994(6)	5101(6)	52.7(16)
N ₁	4956(6)	3045(6)	2776(7)	92(2)
C ₂₂	1444(7)	3605(6)	5237(6)	58.1(17)
C ₂₄	-725(7)	2795(6)	3576(6)	57.9(17)
C ₂₃	49(7)	2920(6)	4852(7)	62.4(18)
C ₁₄	2753(7)	10120(6)	4917(6)	62.9(18)
C ₅	8758(6)	8088(7)	881(6)	58.6(17)
C ₁₃	2226(7)	10316(6)	3656(7)	67.5(19)

Table S11. Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 5v. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[\text{h}^2\text{a}^*2U_{11}+2\text{hka}^*\text{b}^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
O ₂	44(2)	51(2)	40(2)	11.9(18)	3.3(18)	15.5(18)
O ₁	35(2)	69(3)	63(3)	33(2)	12.5(19)	17(2)
N ₂	36(3)	58(3)	51(3)	16(2)	19(2)	22(2)
C ₁₀	24(3)	36(3)	42(3)	8(3)	5(2)	3(2)
C ₉	37(3)	35(3)	30(3)	10(2)	10(2)	14(2)
C ₈	29(3)	45(3)	32(3)	5(2)	7(2)	11(2)
C ₁	31(3)	44(3)	35(3)	9(3)	8(2)	4(2)
C ₁₁	34(3)	37(3)	45(3)	10(3)	11(2)	6(2)
C ₂₀	38(3)	38(3)	38(3)	9(3)	9(2)	11(2)
C ₁₇	38(3)	37(3)	39(3)	14(2)	11(2)	14(2)
C ₁₉	30(3)	41(3)	48(3)	3(3)	4(3)	7(2)
C ₇	44(3)	50(4)	41(3)	9(3)	18(3)	15(3)
C ₂	40(3)	49(4)	55(4)	14(3)	15(3)	5(3)
C ₆	37(3)	53(4)	38(3)	13(3)	9(2)	14(3)
C ₁₆	44(3)	40(3)	49(4)	11(3)	13(3)	4(3)
C ₂₅	45(3)	49(4)	49(4)	11(3)	11(3)	8(3)
C ₃	52(4)	51(4)	69(4)	22(3)	19(3)	8(3)
C ₁₈	51(4)	49(4)	70(4)	28(3)	23(3)	14(3)
C ₂₁	50(4)	47(4)	50(4)	6(3)	10(3)	3(3)
C ₁₂	57(4)	50(4)	53(4)	8(3)	11(3)	20(3)
C ₄	51(4)	65(5)	75(5)	28(4)	18(3)	-3(3)
C ₁₅	58(4)	52(4)	49(4)	10(3)	13(3)	3(3)
N ₁	75(4)	83(4)	154(6)	77(4)	48(4)	41(4)
C ₂₂	77(5)	64(4)	36(3)	12(3)	19(3)	6(4)
C ₂₄	51(4)	58(4)	68(5)	14(4)	20(3)	0(3)

Table S11. Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 5v. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
C ₂₃	70(5)	62(4)	66(5)	19(4)	34(4)	9(4)
C ₁₄	66(4)	60(4)	61(4)	-3(3)	22(4)	12(3)
C ₅	40(4)	84(5)	61(4)	27(4)	23(3)	5(3)
C ₁₃	75(5)	53(4)	70(5)	1(4)	7(4)	32(3)

Table S12. Bond Lengths for 5v.

Atom	Atom	Length/\AA	Atom	Atom	Length/\AA
O ₂	C ₁₀	1.207(5)	C ₂₀	C ₂₁	1.375(7)
O ₁	C ₁₉	1.227(6)	C ₁₇	C ₁₉	1.480(7)
N ₂	C ₇	1.328(6)	C ₁₇	C ₁₈	1.424(7)
N ₂	C ₆	1.377(6)	C ₂	C ₃	1.368(7)
C ₁₀	C ₉	1.524(6)	C ₆	C ₅	1.386(7)
C ₁₀	C ₁₁	1.481(7)	C ₁₆	C ₁₅	1.379(7)
C ₉	C ₈	1.436(6)	C ₂₅	C ₂₄	1.369(7)
C ₉	C ₁₇	1.365(7)	C ₃	C ₄	1.392(8)
C ₈	C ₁	1.454(7)	C ₁₈	N ₁	1.135(7)
C ₈	C ₇	1.387(6)	C ₂₁	C ₂₂	1.378(7)
C ₁	C ₂	1.402(7)	C ₁₂	C ₁₃	1.363(7)
C ₁	C ₆	1.402(6)	C ₄	C ₅	1.363(8)
C ₁₁	C ₁₆	1.393(7)	C ₁₅	C ₁₄	1.359(8)
C ₁₁	C ₁₂	1.384(7)	C ₂₂	C ₂₃	1.366(8)
C ₂₀	C ₁₉	1.480(7)	C ₂₄	C ₂₃	1.365(8)
C ₂₀	C ₂₅	1.381(7)	C ₁₄	C ₁₃	1.371(8)

Table S13. Bond Angles for 5v.

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C ₇	N ₂	C ₆	110.1(4)	O ₁	C ₁₉	C ₂₀	119.8(5)
O ₂	C ₁₀	C ₉	119.9(4)	O ₁	C ₁₉	C ₁₇	119.3(5)
O ₂	C ₁₀	C ₁₁	122.5(5)	C ₂₀	C ₁₉	C ₁₇	120.9(5)
C ₁₁	C ₁₀	C ₉	117.4(4)	N ₂	C ₇	C ₈	110.5(5)
C ₈	C ₉	C ₁₀	113.8(4)	C ₃	C ₂	C ₁	119.4(5)
C ₁₇	C ₉	C ₁₀	118.0(4)	N ₂	C ₆	C ₁	107.6(5)
C ₁₇	C ₉	C ₈	128.2(4)	N ₂	C ₆	C ₅	128.7(5)
C ₉	C ₈	C ₁	128.2(4)	C ₅	C ₆	C ₁	123.7(5)
C ₇	C ₈	C ₉	126.4(5)	C ₁₅	C ₁₆	C ₁₁	120.1(5)
C ₇	C ₈	C ₁	105.3(4)	C ₂₄	C ₂₅	C ₂₀	120.4(6)
C ₂	C ₁	C ₈	136.4(5)	C ₂	C ₃	C ₄	121.6(6)

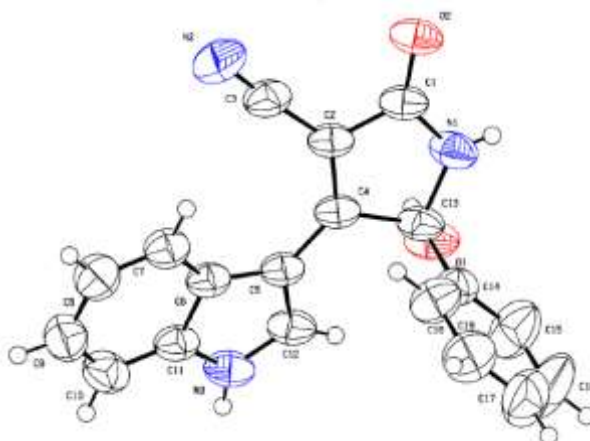
Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C ₆	C ₁	C ₈	106.4(4)	N ₁	C ₁₈	C ₁₇	179.2(8)
C ₆	C ₁	C ₂	117.2(5)	C ₂₀	C ₂₁	C ₂₂	119.6(6)
C ₁₆	C ₁₁	C ₁₀	122.9(5)	C ₁₃	C ₁₂	C ₁₁	120.6(6)
C ₁₂	C ₁₁	C ₁₀	118.6(5)	C ₅	C ₄	C ₃	121.0(6)
C ₁₂	C ₁₁	C ₁₆	118.4(5)	C ₁₄	C ₁₅	C ₁₆	120.2(6)
C ₂₅	C ₂₀	C ₁₉	119.2(5)	C ₂₃	C ₂₂	C ₂₁	120.8(6)
C ₂₁	C ₂₀	C ₁₉	121.2(5)	C ₂₃	C ₂₄	C ₂₅	120.3(6)
C ₂₁	C ₂₀	C ₂₅	119.3(5)	C ₂₄	C ₂₃	C ₂₂	119.6(6)
C ₉	C ₁₇	C ₁₉	121.4(4)	C ₁₅	C ₁₄	C ₁₃	120.2(6)
C ₉	C ₁₇	C ₁₈	121.4(5)	C ₄	C ₅	C ₆	117.2(5)
C ₁₈	C ₁₇	C ₁₉	117.2(5)	C ₁₂	C ₁₃	C ₁₄	120.4(6)

A	B	C	D	Angle/°	A	B	C	D	Angle/°
O ₂	C ₁₀	C ₉	C ₈	75.6(6)	C ₂₀	C ₂₅	C ₂₄	C ₂₃	0.1(9)
O ₂	C ₁₀	C ₉	C ₁₇	-107.2(6)	C ₂₀	C ₂₁	C ₂₂	C ₂₃	1.0(9)
O ₂	C ₁₀	C ₁₁	C ₁₆	-179.9(5)	C ₁₇	C ₉	C ₈	C ₁	-169.8(5)
O ₂	C ₁₀	C ₁₁	C ₁₂	-2.9(8)	C ₁₇	C ₉	C ₈	C ₇	14.5(8)
N ₂	C ₆	C ₅	C ₄	-179.5(6)	C ₁₉	C ₂₀	C ₂₅	C ₂₄	-172.2(5)
C ₁₀	C ₉	C ₈	C ₁	7.1(7)	C ₁₉	C ₂₀	C ₂₁	C ₂₂	171.5(5)
C ₁₀	C ₉	C ₈	C ₇	-168.6(5)	C ₇	N ₂	C ₆	C ₁	-0.5(6)
C ₁₀	C ₉	C ₁₇	C ₁₉	9.2(7)	C ₇	N ₂	C ₆	C ₅	-179.6(5)
C ₁₀	C ₉	C ₁₇	C ₁₈	-172.1(5)	C ₇	C ₈	C ₁	C ₂	-179.7(6)
C ₁₀	C ₁₁	C ₁₆	C ₁₅	178.7(5)	C ₇	C ₈	C ₁	C ₆	-0.6(5)
C ₁₀	C ₁₁	C ₁₂	C ₁₃	-179.1(5)	C ₂	C ₁	C ₆	N ₂	179.9(5)
C ₉	C ₁₀	C ₁₁	C ₁₆	-5.9(7)	C ₂	C ₁	C ₆	C ₅	-0.9(8)
C ₉	C ₁₀	C ₁₁	C ₁₂	171.1(5)	C ₂	C ₃	C ₄	C ₅	1.7(10)
C ₉	C ₈	C ₁	C ₂	3.9(10)	C ₆	N ₂	C ₇	C ₈	0.1(6)
C ₉	C ₈	C ₁	C ₆	-177.0(5)	C ₆	C ₁	C ₂	C ₃	0.7(8)
C ₉	C ₈	C ₇	N ₂	176.8(5)	C ₁₆	C ₁₁	C ₁₂	C ₁₃	-2.0(9)
C ₉	C ₁₇	C ₁₉	O ₁	26.7(8)	C ₁₆	C ₁₅	C ₁₄	C ₁₃	-1.8(9)
C ₉	C ₁₇	C ₁₉	C ₂₀	-152.8(5)	C ₂₅	C ₂₀	C ₁₉	O ₁	32.9(7)
C ₈	C ₉	C ₁₇	C ₁₉	-174.0(5)	C ₂₅	C ₂₀	C ₁₉	C ₁₇	-147.6(5)
C ₈	C ₉	C ₁₇	C ₁₈	4.7(8)	C ₂₅	C ₂₀	C ₂₁	C ₂₂	-1.4(8)
C ₈	C ₁	C ₂	C ₃	179.7(6)	C ₂₅	C ₂₄	C ₂₃	C ₂₂	-0.6(9)
C ₈	C ₁	C ₆	N ₂	0.6(6)	C ₃	C ₄	C ₅	C ₆	-1.8(9)
C ₈	C ₁	C ₆	C ₅	179.8(5)	C ₁₈	C ₁₇	C ₁₉	O ₁	-152.0(5)
C ₁	C ₈	C ₇	N ₂	0.3(6)	C ₁₈	C ₁₇	C ₁₉	C ₂₀	28.5(7)
C ₁	C ₂	C ₃	C ₄	-1.1(9)	C ₂₁	C ₂₀	C ₁₉	O ₁	-140.1(5)
C ₁	C ₆	C ₅	C ₄	1.5(9)	C ₂₁	C ₂₀	C ₁₉	C ₁₇	39.5(7)

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C11	C10	C9	C8	-98.6(5)	C21	C20	C25	C24	0.9(8)
C11	C10	C9	C17	78.7(6)	C21	C22	C23	C24	0.0(9)
C11	C16	C15	C14	0.2(8)	C12	C11	C16	C15	1.7(8)
C11	C12	C13	C14	0.5(10)	C15	C14	C13	C12	1.5(10)

Atom	x	y	z	U(eq)
H ₂	8576.79	5389.85	1101.03	55
H ₇	6395.02	4392.83	1532.46	52
H _{2A}	5050.03	9079.48	1428.77	56
H ₁₆	4045.18	7251.9	4149.85	53
H ₂₅	-650.21	3272.01	1820.11	57
H ₃	6758.89	10611.17	1018.25	66
H ₂₁	3035.01	4622.99	4619.85	59
H ₁₂	2001.06	9515.63	1725.56	64
H ₄	9067.39	10028.11	732.55	74
H ₁₅	3821.54	8878.19	5957.93	63
H ₂₂	1971.76	3690.46	6106.3	70
H ₂₄	-1671.1	2322.18	3307.65	69
H ₂₃	-370.3	2542.22	5456.25	75
H ₁₄	2639.85	10760.42	5648.76	75
H ₅	9673.61	7836.86	732.02	70
H ₁₃	1776.19	11099	3537.05	81

4. Crystallographic description of compound 6a



Ellipsoid plot

CCDC No.	2361275
Empirical formula	C ₁₉ H ₁₃ N ₃ O ₂
Formula weight	315.32
Temperature/K	298
Crystal system	monoclinic
Space group	P2 ₁ /c
a/Å	10.4772(6)
b/Å	22.4464(13)
c/Å	9.0718(4)
α/°	90
β/°	96.262(2)
γ/°	90
Volume/Å ³	2120.7(2)
Z	4
ρ _{calc} /cm ³	0.988
μ/mm ⁻¹	0.066
F(000)	656.0
Crystal size/mm ³	0.214 × 0.136 × 0.123
Radiation	MoKα (λ = 0.71073)
2θ range for data collection/°	4.868 to 52.84
Index ranges	-13 ≤ h ≤ 13, -28 ≤ k ≤ 28, -11 ≤ l ≤ 11
Reflections collected	20924
Independent reflections	4323 [R _{int} = 0.0503, R _{sigma} = 0.0335]
Data/restraints/parameters	4323/0/222
Goodness-of-fit on F ²	1.034
Final R indexes [I >= 2σ (I)]	R ₁ = 0.0735, wR ₂ = 0.2339
Final R indexes [all data]	R ₁ = 0.0826, wR ₂ = 0.2429
Largest diff. peak/hole / e Å ⁻³	0.26/-0.27

Atom	x	y	z	U(eq)
O ₂	6337.9(17)	5241.9(9)	6965.8(16)	64.5(5)
O ₁	5208.8(15)	5670.7(10)	2215.1(18)	67.1(5)
N ₃	7821(2)	4790.9(10)	-375(2)	62.3(6)
N ₁	5933(2)	5749.2(12)	4768(2)	63.8(6)
C ₆	8780(2)	4568.0(10)	1903(2)	50.6(5)
C ₁₄	7007(2)	6290.1(11)	2897(2)	56.1(6)
C ₄	7162(2)	5169.4(10)	3324(2)	49.6(5)
C ₅	7701(2)	4956.9(10)	2036(2)	49.8(5)
C ₂	7254(2)	4959.3(11)	4745(2)	52.1(5)
C ₁	6482(2)	5324.4(11)	5638(2)	54.0(5)

Table S17. Fractional Atomic Coordinates ($\times 10^4$) and Equivalent Isotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 6a. U_{eq} is defined as 1/3 of the trace of the orthogonalised U_{ij} tensor.

Atom	<i>x</i>	<i>y</i>	<i>z</i>	U_{eq}
C ₁₂	7179(2)	5086.1(12)	587(2)	59.0(6)
C ₇	9782(2)	4324.7(11)	2877(3)	59.5(6)
C ₁₁	8817(2)	4472.2(10)	364(2)	54.2(5)
N ₂	8334(3)	4050.0(13)	6041(3)	92.5(8)
C ₁₃	6295(2)	5722.8(11)	3254(2)	54.5(6)
C ₃	7862(3)	4447.7(12)	5413(2)	63.0(6)
C ₁₉	8290(3)	6363.7(13)	3368(3)	72.0(7)
C ₁₀	9755(3)	4130.0(12)	-196(3)	67.0(7)
C ₈	10705(3)	3987.5(13)	2319(3)	70.1(7)
C ₉	10691(3)	3882.7(13)	795(3)	72.4(7)
C ₁₈	8941(4)	6880.1(15)	3071(4)	88.3(9)
C ₁₅	6379(3)	6750.4(16)	2134(5)	96.8(11)
C ₁₇	8297(4)	7326.7(15)	2309(5)	102.3(12)
C ₁₆	7033(5)	7271.1(17)	1825(6)	125.5(16)

Table S18. Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 6a. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U_{11}	U_{22}	U_{33}	U_{23}	U_{13}	U_{12}
O ₂	66.5(10)	93.7(13)	32.7(8)	6.9(7)	2.8(7)	-9.5(9)
O ₁	49.9(9)	101.8(14)	47.4(9)	20.0(8)	-4.6(7)	-4.1(9)
N ₃	72.8(13)	82.1(14)	30.9(9)	5.7(8)	0.7(8)	-6.5(11)
N ₁	56.6(12)	92.9(16)	42.7(10)	11.4(10)	9.3(9)	11.0(11)
C ₆	58.7(12)	54.5(12)	37.6(10)	4.1(8)	1.2(9)	-10.4(10)
C ₁₄	60.2(13)	64.8(14)	43.9(11)	5.5(9)	8.4(9)	7.0(10)
C ₄	52.0(11)	61.2(12)	34.3(10)	8.0(8)	-1.4(8)	-9.7(9)
C ₅	57.6(12)	58.4(12)	32.3(9)	7.0(8)	-0.2(8)	-5.1(10)
C ₂	58.7(12)	64.0(13)	32.7(10)	6.4(9)	0.5(8)	-6.8(10)
C ₁	51.2(12)	73.6(14)	36.2(10)	7.8(9)	-0.5(8)	-10.6(10)
C ₁₂	62.7(14)	78.8(16)	34.6(10)	9.7(10)	0.9(9)	-2.8(11)
C ₇	65.5(14)	66.5(14)	45.0(12)	5.4(10)	-1.4(10)	-1.1(11)
C ₁₁	66.5(14)	57.3(12)	38.3(10)	2.6(9)	2.9(9)	-10.9(10)
N ₂	119(2)	84.2(17)	73.8(16)	30.5(13)	7.9(15)	13.8(16)
C ₁₃	49.4(11)	80.2(15)	33.0(10)	11.7(9)	0.6(8)	2.0(10)
C ₃	78.1(16)	69.2(15)	40.5(11)	11.5(10)	0.8(11)	-2.3(13)
C ₁₉	66.9(16)	74.3(17)	72.1(17)	13.2(13)	-4.8(12)	-5.9(13)
C ₁₀	78.8(17)	71.0(15)	52.4(13)	-9.9(11)	12.9(12)	-9.3(13)
C ₈	69.2(16)	73.0(16)	66.9(15)	4.3(12)	1.7(12)	4.1(13)
C ₉	75.8(17)	66.8(16)	76.5(17)	-7.3(13)	17.1(14)	1.8(13)

Table S18. Anisotropic Displacement Parameters ($\text{\AA}^2 \times 10^3$) for 6a. The Anisotropic displacement factor exponent takes the form: $-2\pi^2[h^2a^{*2}U_{11}+2hka^*b^*U_{12}+\dots]$.

Atom	U ₁₁	U ₂₂	U ₃₃	U ₂₃	U ₁₃	U ₁₂
C ₁₈	88(2)	77(2)	100(2)	0.8(17)	9.2(18)	-18.5(17)
C ₁₅	71.1(19)	86(2)	130(3)	27(2)	-5.5(19)	14.0(16)
C ₁₇	108(3)	59.6(18)	143(3)	2.7(19)	31(2)	-4.2(18)
C ₁₆	115(3)	69(2)	191(5)	40(3)	7(3)	20(2)

Table S19. Bond Lengths for 6a.

Atom	Atom	Length/ \AA	Atom	Atom	Length/ \AA
O ₂	C ₁	1.244(2)	C ₄	C ₁₃	1.536(3)
O ₁	C ₁₃	1.402(3)	C ₅	C ₁₂	1.398(3)
N ₃	C ₁₂	1.334(3)	C ₂	C ₁	1.457(3)
N ₃	C ₁₁	1.378(3)	C ₂	C ₃	1.417(3)
N ₁	C ₁	1.328(3)	C ₇	C ₈	1.368(4)
N ₁	C ₁₃	1.465(3)	C ₁₁	C ₁₀	1.386(4)
C ₆	C ₅	1.444(3)	N ₂	C ₃	1.142(3)
C ₆	C ₇	1.407(3)	C ₁₉	C ₁₈	1.386(4)
C ₆	C ₁₁	1.417(3)	C ₁₀	C ₉	1.373(4)
C ₁₄	C ₁₃	1.528(3)	C ₈	C ₉	1.401(4)
C ₁₄	C ₁₉	1.376(4)	C ₁₈	C ₁₇	1.355(5)
C ₁₄	C ₁₅	1.371(4)	C ₁₅	C ₁₆	1.399(5)
C ₄	C ₅	1.434(3)	C ₁₇	C ₁₆	1.355(6)
C ₄	C ₂	1.366(3)			

Table S20. Bond Angles for 6a.

Atom	Atom	Atom	Angle/ $^\circ$	Atom	Atom	Atom	Angle/ $^\circ$
C ₁₂	N ₃	C ₁₁	110.40(18)	N ₃	C ₁₂	C ₅	109.8(2)
C ₁	N ₁	C ₁₃	112.9(2)	C ₈	C ₇	C ₆	119.4(2)
C ₇	C ₆	C ₅	136.0(2)	N ₃	C ₁₁	C ₆	107.4(2)
C ₇	C ₆	C ₁₁	117.4(2)	N ₃	C ₁₁	C ₁₀	129.7(2)
C ₁₁	C ₆	C ₅	106.28(19)	C ₁₀	C ₁₁	C ₆	122.8(2)
C ₁₉	C ₁₄	C ₁₃	121.2(2)	O ₁	C ₁₃	N ₁	111.07(18)
C ₁₅	C ₁₄	C ₁₃	121.1(2)	O ₁	C ₁₃	C ₁₄	107.83(18)
C ₁₅	C ₁₄	C ₁₉	117.7(3)	O ₁	C ₁₃	C ₄	113.2(2)
C ₅	C ₄	C ₁₃	121.26(18)	N ₁	C ₁₃	C ₁₄	110.3(2)
C ₂	C ₄	C ₅	131.4(2)	N ₁	C ₁₃	C ₄	101.90(17)
C ₂	C ₄	C ₁₃	107.38(19)	C ₁₄	C ₁₃	C ₄	112.52(18)
C ₄	C ₅	C ₆	130.61(19)	N ₂	C ₃	C ₂	175.4(3)
C ₁₂	C ₅	C ₆	106.0(2)	C ₁₄	C ₁₉	C ₁₈	121.7(3)

Atom	Atom	Atom	Angle/°	Atom	Atom	Atom	Angle/°
C ₁₂	C ₅	C ₄	123.3(2)	C ₉	C ₁₀	C ₁₁	117.9(2)
C ₄	C ₂	C ₁	110.1(2)	C ₇	C ₈	C ₉	121.8(3)
C ₄	C ₂	C ₃	131.7(2)	C ₁₀	C ₉	C ₈	120.6(3)
C ₃	C ₂	C ₁	117.97(19)	C ₁₇	C ₁₈	C ₁₉	119.3(3)
O ₂	C ₁	N ₁	126.3(2)	C ₁₄	C ₁₅	C ₁₆	120.8(3)
O ₂	C ₁	C ₂	126.0(2)	C ₁₆	C ₁₇	C ₁₈	120.8(3)
N ₁	C ₁	C ₂	107.69(19)	C ₁₇	C ₁₆	C ₁₅	119.7(3)

A	B	C	D	Angle/°	A	B	C	D	Angle/°
N ₃	C ₁₁	C ₁₀	C ₉	176.4(2)	C ₇	C ₆	C ₁₁	N ₃	-174.7(2)
C ₆	C ₅	C ₁₂	N ₃	2.4(3)	C ₇	C ₆	C ₁₁	C ₁₀	3.1(3)
C ₆	C ₇	C ₈	C ₉	0.8(4)	C ₇	C ₈	C ₉	C ₁₀	1.5(4)
C ₆	C ₁₁	C ₁₀	C ₉	-0.9(4)	C ₁₁	N ₃	C ₁₂	C ₅	-2.2(3)
C ₁₄	C ₁₉	C ₁₈	C ₁₇	0.6(5)	C ₁₁	C ₆	C ₅	C ₄	175.6(2)
C ₁₄	C ₁₅	C ₁₆	C ₁₇	-1.3(7)	C ₁₁	C ₆	C ₅	C ₁₂	-1.7(2)
C ₄	C ₅	C ₁₂	N ₃	-175.1(2)	C ₁₁	C ₆	C ₇	C ₈	-3.0(3)
C ₄	C ₂	C ₁	O ₂	178.2(2)	C ₁₁	C ₁₀	C ₉	C ₈	-1.5(4)
C ₄	C ₂	C ₁	N ₁	-0.1(3)	C ₁₃	N ₁	C ₁	O ₂	-179.6(2)
C ₅	C ₆	C ₇	C ₈	-176.3(3)	C ₁₃	N ₁	C ₁	C ₂	-1.4(3)
C ₅	C ₆	C ₁₁	N ₃	0.5(2)	C ₁₃	C ₁₄	C ₁₉	C ₁₈	-179.0(3)
C ₅	C ₆	C ₁₁	C ₁₀	178.3(2)	C ₁₃	C ₁₄	C ₁₅	C ₁₆	179.3(4)
C ₅	C ₄	C ₂	C ₁	-177.9(2)	C ₁₃	C ₄	C ₅	C ₆	160.7(2)
C ₅	C ₄	C ₂	C ₃	-3.2(4)	C ₁₃	C ₄	C ₅	C ₁₂	-22.4(3)
C ₅	C ₄	C ₁₃	O ₁	58.0(3)	C ₁₃	C ₄	C ₂	C ₁	1.4(2)
C ₅	C ₄	C ₁₃	N ₁	177.28(19)	C ₁₃	C ₄	C ₂	C ₃	176.0(3)
C ₅	C ₄	C ₁₃	C ₁₄	-64.6(3)	C ₃	C ₂	C ₁	O ₂	2.7(4)
C ₂	C ₄	C ₅	C ₆	-20.1(4)	C ₃	C ₂	C ₁	N ₁	-175.5(2)
C ₂	C ₄	C ₅	C ₁₂	156.7(2)	C ₁₉	C ₁₄	C ₁₃	O ₁	-157.9(2)
C ₂	C ₄	C ₁₃	O ₁	-121.38(19)	C ₁₉	C ₁₄	C ₁₃	N ₁	80.6(3)
C ₂	C ₄	C ₁₃	N ₁	-2.0(2)	C ₁₉	C ₁₄	C ₁₃	C ₄	-32.4(3)
C ₂	C ₄	C ₁₃	C ₁₄	116.1(2)	C ₁₉	C ₁₄	C ₁₅	C ₁₆	1.1(6)
C ₁	N ₁	C ₁₃	O ₁	122.9(2)	C ₁₉	C ₁₈	C ₁₇	C ₁₆	-0.7(6)

A	B	C	D	Angle/°	A	B	C	D	Angle/°
C ₁	N ₁	C ₁₃	C ₁₄	-117.6(2)	C ₁₈	C ₇	C ₁₆	C ₁₅	1.0(7)
C ₁	N ₁	C ₁₃	C ₄	2.1(3)	C ₁₅	C ₄	C ₁₃	O ₁	24.0(3)
C ₁₂	N ₃	C ₁₁	C ₆	1.0(3)	C ₁₅	C ₄	C ₁₃	N ₁	-97.5(3)
C ₁₂	N ₃	C ₁₁	C ₁₀	-176.6(3)	C ₁₅	C ₄	C ₁₃	C ₄	149.5(3)
C ₇	C ₆	C ₅	C ₄	-10.7(4)	C ₁₅	C ₄	C ₁₉	C ₁₈	-0.8(5)
C ₇	C ₆	C ₅	C ₁₂	172.1(3)					

Atom	x	y	z	U(eq)
H ₁	4818.34	5365.24	2377.65	101
H ₃	7639.26	4798.66	-1322.35	75
H ₁₂	6488.59	5339.65	331.35	71
H ₇	9817.11	4392.61	3892.24	71
H ₁₉	8732.43	6059.18	3898.26	86
H ₁₀	9750.64	4069.94	-1210.64	80
H ₈	11359.34	3823.4	2968.8	84
H ₉	11320.16	3643.64	451.6	87
H ₁₈	9811.23	6919.53	3390.64	106
H ₁₅	5507.58	6715.97	1817.93	116
H ₁₇	8728.15	7675.7	2116.67	123
H ₁₆	6602.79	7577.51	1289.03	151
H _{1A}	5480(40)	5947(16)	5100(40)	86(11)

Number	X	Y	Z	Volume	Electron count	Content
1	-0.340	0.250	-0.157	369.2	90.5	2 C ₃ H ₆ O ₂ ,2
2	0.340	0.750	-0.959	369.2	90.5	2 C ₃ H ₆ O ₂ ,2

5. Detection of H₂ using gas chromatography (GC)

Gas Chromatography (GC) was performed using a GC-5800 gas chromatograph, Ar was used as carrier gas, thermal conductivity detector (TCD temperature was 50 °C) equipped with carbo-sieve column.

Reaction setup for the H₂ evolution of 4a



UV and HPLC spectroscopy-graded DMF was degassed by using a vacuum pump and sonicator for an hour to remove dissolved oxygen. Then, to a 10 mL cleaned and dried glass-sealed tube **II**, 76 mg (0.5 mmol) of **4a** and a magnetic bead were added. Then 4 mL of degassed DMF was injected and purged with N₂ for an hour. In a similar fashion sealed tube **I** was prepared without compound **4a**. Both tube **I** and tube **II** were heated with continuous stirring at 100 °C for 8 hours. After 8 hours of reaction, tube **I** and tube **II** were sent to the GC lab for analysis. 1 mL of gas from each of the tubes was injected in GC for the analysis.

The standard H₂ shows a retention time of 0.25 min (Fig. S1). The gas from the tube **I** show only one peak of N₂ gas with the retention time of 0.61 min. (Fig. S2). Tube **II** shows two peaks with a retention time of 0.19 min and 0.61 min for H₂ and N₂ respectively (Fig. S3).

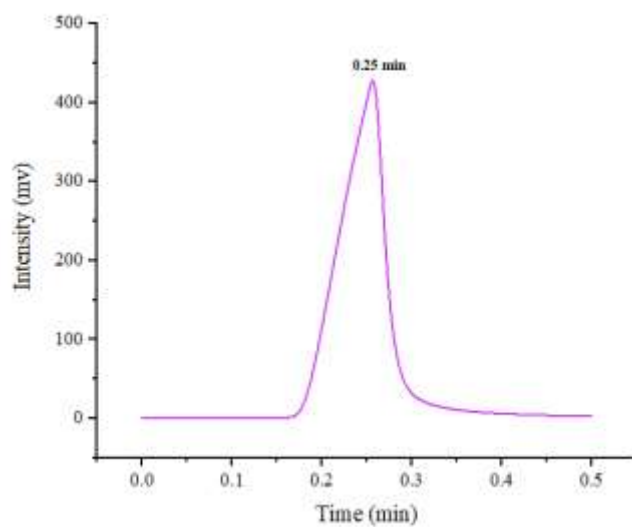


Fig. S1 Chromatogram of reference substance H_2

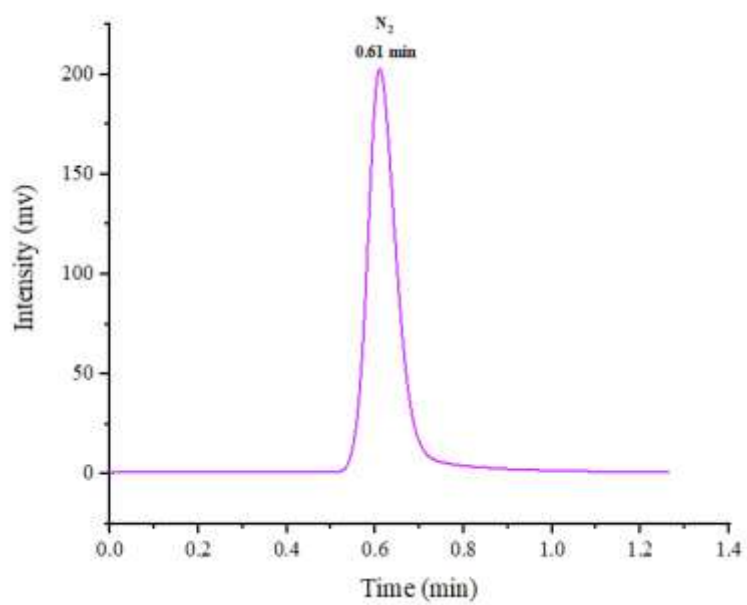


Fig. S2. Chromatogram of gas in the tube I

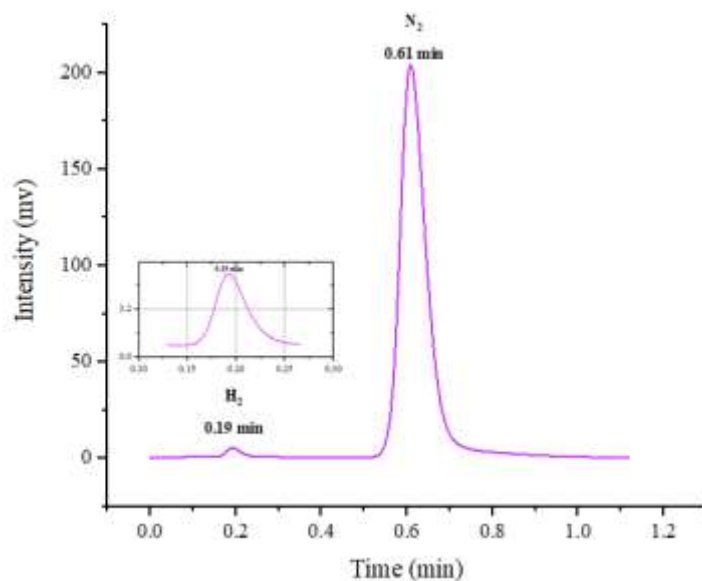
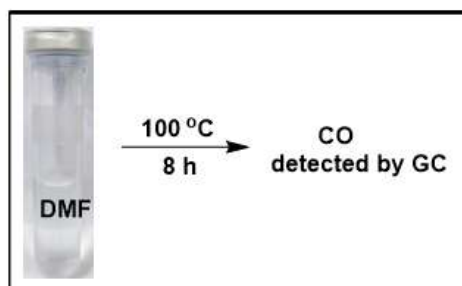


Fig. S3 Chromatogram of gas in the tube II

6. Detection of CO using gas chromatography

The presence of CO during the heating of DMF could indeed indicate the formation of dimethylamine. This transformation typically occurs via decarbonylation, where the carbonyl group (-C=O) in DMF is eliminated as carbon monoxide (CO).



A 10 mL cleaned and dried glass-sealed tube was charged with 4 mL of UV and HPLC grade DMF, which was then degassed using a vacuum pump and sonicated for one hour. The sealed tube containing the DMF was subsequently heated at 100°C for 8 hours. Afterward, 1 mL of gas from the tube was collected and injected into a gas chromatograph

(GC) for CO analysis. The output is presented in Fig. S5. A GC analysis of atmospheric air was also conducted to ensure that CO was only detected from the sealed tube.

The standard CO shows the retention time of 0.9 min (Fig. S4). The CO from DMF was shown at 0.87 min (Fig. S5). The Figure S6 is the chromatogram of atmospheric air.

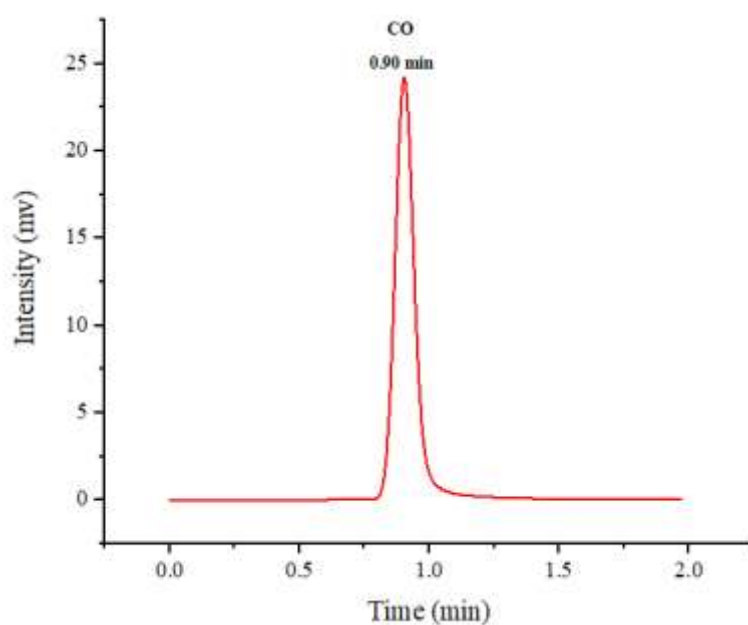


Fig. S4 Chromatogram of reference substance CO

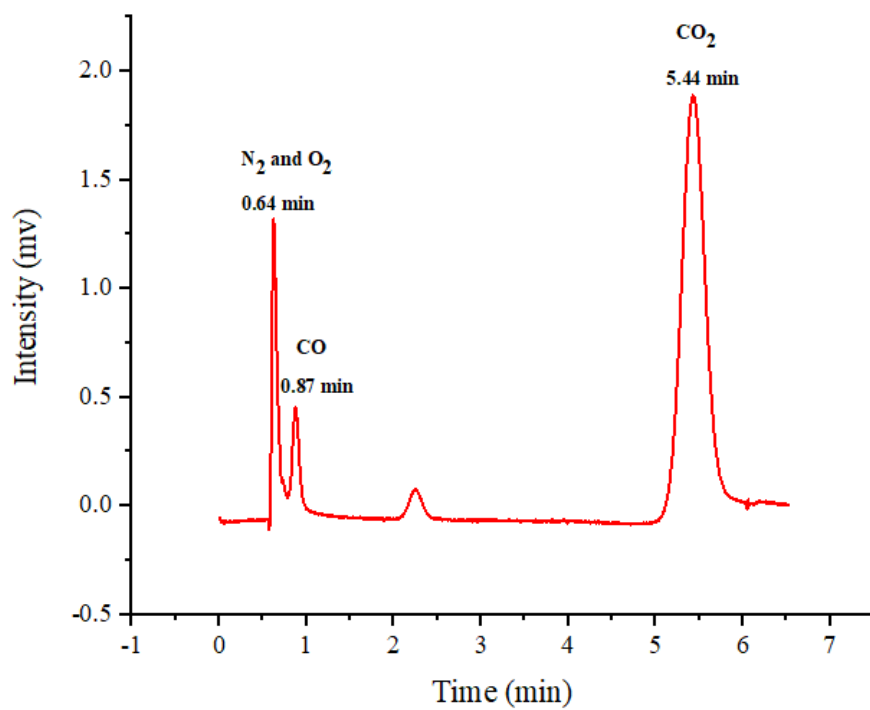


Fig. S5 Chromatogram of gas from heated DMF

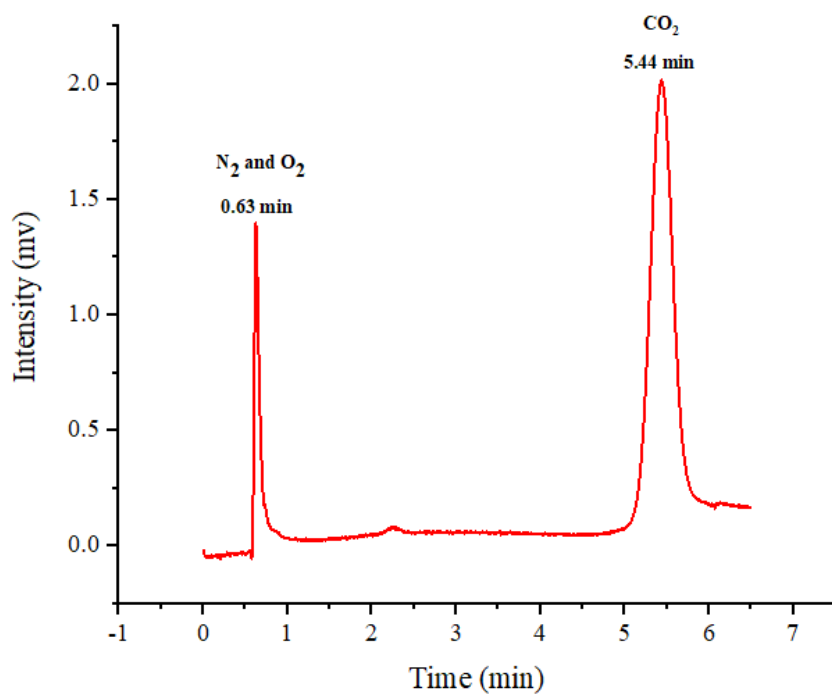


Fig. S6 Chromatogram of atmospheric air