Supporting Information

Photo induced NO release of [Fe₂(µ-SL)₂(NO)₄] complexes and their protein adducts: insights from structure, cytotoxicity and photodynamic studies

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Fig. S1. ¹H NMR spectrum (600 MHz) of the four complexes: (a) complex **1**, (b) complex **2**, (c) complex **3**, (d) complex **4**, (e) crystallized complexes **4**.







Fig. S2. ESI-MS spectra of the four complexes in CH₃OH: (a) complex **1**, (b) complex **2**, (c) complex **3**, (d) complex **4**, (e) crystallized complexes **4**.



Fig. S3. UV-vis spectra of complexes 1–4 in PBS buffer (5% DMSO) for 7 days: (a) complex 1, (b) complex 2, (c) complex 3, and (d) complex 4.



Fig. S4. Cell viability of HeLa cells incubated with complexes **1**–**4** at various concentrations for 24 h under dark and light irradiation: (a) complex **1**, (b) complex **2**, (c) complex **3**, and (d) complex **4**.



Fig. S5. Cell viability of HL7702 cells incubated with complexes 1–4 at various concentrations for 24 h under dark and light irradiation: (a) complex 1, (b) complex 2, (c) complex 3, and (d) complex 4.



Fig. S6. Cell viability of HeLa cells incubated with ferritin complexes 1–4 adducts at various concentrations for 24 h under dark and light irradiation: (a) adduct 1, (b) adduct 2, (c) adduct 3, and (d) adduct 4.



Fig. S7. Red confocal microscopy images of HeLa cells treated with DAX-J2 for 0, 5,10 and 15 min under an LED light (420 nm). Excitation was performed with a 561 nm laser, and the emission wavelength was recorded in the range of 579–701 nm.

	1	2	3	4'
Chemical formula	$C_2H_6Fe_2N_4O_4S_2$	$C_{14}H_{14}Fe_2N_4O_4S_2$	$C_{20}H_{22}Fe_{2}N_{4}O_{8}S_{2}$	$C_{16}H_{20}B_2Fe_2N_4O_8S_2$
Formula weight	325.92	478.11	622.23	593.80
Crystal system	Triclinic	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P</i> -1	$P2_{1}/c$	C2/c	$P2_1/c$
a (Å)	7.0639(4)	7.142(4)	35.2604 (13)	5.0645(2)
b (Å)	8.7633(8)	16.668(11)	5.0790 (2)	23.6415(8)
c (Å)	9.5289(8)	7.986(5)	16.1563 (6)	10.1545(4)
a (°)	89.048(7)	90	90	90
β (°)	68.641(7)	93.36(2)	115.6664 (11)	95.0247(10)
γ (°)	89.088(6)	90	90	90
V (Å ³)	549.25 (8)	949.1 (10)	2607.90 (17)	1211.15(8)
Z	4	2	4	2
Temperature (K)	293 K	298 K	273 K	298 K
DX (g cm ⁻³)	1.971	1.673	1.585	1.628
F (000)	324.0	484.0	1272.0	604.0
Theta range (°)	4.6-28.9°	2.8-27.6°	2.9–28.3°	2.7-28.5°
$R [F^2 > 2 \sigma(F_2)]$	0.040	0.052	0.028	0.029
w $R(F^2)$	0.119	0.093	0.076	0.074
S	1.07	1.01	1.02	1.05
Δ > max, Δ > min (e Å ⁻³)	0.66, -0.61	0.42, -0.40	0.31, -0.28	0.27, -0.22

Table S1. Crystallographic data and refinement statistics for the four nitrosyl iron-sulfur complexes.

Fe1—N1	1.669 (3)	O1—N1	1.162 (4)
Fe1—N2	1.666 (4)	O2—N2	1.155 (5)
Fe1—S1	2.2513 (12)	S1—C1	1.825 (4)
O1—N1—Fe1	169.7 (4)	O2—N2—Fe1	171.0 (3)
Fe1—S1—Fe1 ⁱ	73.54 (4)	S1—Fe1—S1 ⁱ	106.46 (4)

Table S2. Selected bond distances (Å) and bond angles (°) of complex 1.

Table S3. Selected bond distances (Å) and bond angles (°) of complex 2.

Fe1—N1	1.665 (3)	01—N1	1.169 (3)
Fe1—N2	1.666 (3)	O2—N2	1.161 (4)
Fe1—S1	2.2567 (16)	S1—C1	1.839 (3)
O1—N1—Fe1	169.5 (3)	O2—N2—Fe1	167.6 (3)
Fe1—S1—Fe1 ⁱ	73.95 (4)	S1—Fe1—S1 ⁱ	106.05 (4)

 Table S4. Selected bond distances (Å) and bond angles (°) of complex 3.

Fe1—N1	1.6640 (19)	01—N1	1.161 (2)
Fe1—N2	1.6649 (18)	O2—N2	1.157 (2)
Fe1—S1	2.2485 (6)	S1—C1	1.845 (2)
O1—N1—Fe1	172.40 (18)	O2—N2—Fe1	171.9 (2)
Fe1—S1—Fe1 ⁱ	73.018 (19)	S1—Fe1—S1 ⁱ	106.984 (19)

Fe1—N1	1.6653 (16)	O1—N1	1.161 (2)
Fe1—N2	1.6703 (17)	O2—N2	1.155 (2)
Fe1—S1	2.2546 (5)	S1—C1	1.8444 (18)
O1—N1—Fe1	170.52 (18)	O2—N2—Fe1	170.3 (2)
Fe1—S1—Fe1 ⁱ	73.188 (15)	S1—Fe1—S1 ⁱ	106.810 (15)

Table S5. Selected bond distances (Å) and bond angles (°) of complex 4'.

Table S6. In vitro cytotoxic IC_{50} values (μM) for complexes 1–4 and their Ferritincomplex adducts against HeLa cells under dark and light irradiation.

Ac	lducts	1	2	3	4
HeLa	Light (-)	5.03 ± 0.25	4.87 ± 0.41	6.13 ± 0.19	4.46 ± 0.65
	Light (+)	4.02 ± 0.32	3.25 ± 0.26	4.75 ± 0.39	3.36 ± 0.72