Supporting Information for

A Zinc Oxide-Silver Nanocomposites-based SERS Nanoplatform for Ultrasensitive Ofloxacin Determination in Beef and an Ophthalmic Solution: Effects of ZnO and ZnO content on Electron Transfer and SERS enhancements

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Calculation of limit of detection (LOD)

The standard curve of linear detecting range was given as:

$$Y = A + B \times Log(X) \tag{1}$$

where A and B are intercept and slope of regression equation obtained through the plot of the logarithmic SERS intensity (Y) – logarithmic concentration (X).

The LOD is calculated using the following equation [1]:

$$LOD = 10^{\left[(Y_{blank} + 3SD) / Y_{blank} - A \right] / B}$$
(2)

where Y_{blank} and SD are the SERS signal and the standard deviation of blank sample, respectively.

SD is calculated via the well-known formula:

$$SD = \sqrt{\frac{1}{n-1} \times \sum_{i}^{n} (x_i - x_{average})^2}$$
(3)

where x_i if the "i" sample of the series of measurements, $x_{average}$ is the average value of SERS signal obtained from the blank sample repeated n times.

Calculation of relative standard deviation (RSD)

The RSD value of repeatability and reproducibility is calculated via the well-known formula:

$$RSD = \frac{SD \times 100}{x_{average}}$$
(4)

where SD is the standard deviation that calculates using equation 4 and $x_{average}$ is the average value of SERS signal obtained from each measurement.

Calculation of enhancement factor (EF)

The EF value is calculated according to the well-established equation, which was employed in several published studies [2, 3]:

$$EF = \frac{I_{SERS}}{I_{Raman}} \times \frac{N_{bulk}}{N_{surface}}$$
(5)

where I_{SERS} and I_{Raman} are Raman signal intensity of the analyte with and without SERS from the substrate, respectively; and N_{bulk} is the number of analyte molecules that are probed on the Raman spectrum, while $N_{surface}$ is the number of analyte molecules probed using SERS.

N_{bulk} can be calculated following:

$$N_{bulk} = \frac{A_{laser} \times h \times \rho}{M} \times N_A \tag{6}$$

where A_{laser} , h, ρ and m are the laser spot area, the focal length, the density of the solid analyte and its molecular weight, respectively; and N_A is the Avogadro number.

N_{surface} can be expressed as:

$$N_{surface} = \frac{C \times V}{A_{substrate}} \times N_A \times A_{laser}$$
⁽⁷⁾

where C, V, $A_{substrate}$ are the concentration, the volume drop-casted of the analyte, and the area of the substrate, respectively; N_A is the Avogadro number; and A_{laser} is the laser spot area.

Thus, EF can be calculated as:

$$EF = \frac{I_{SERS}}{I_{Raman}} \times \frac{N_{bulk}}{N_{surface}} = \frac{I_{SERS}}{I_{Raman}} \times \frac{h \times \rho \times A_{substrate}}{M \times C \times V}$$
(8)

In our case, I_{SERS} and I_{Raman} is Raman signal intensity with and without SERS substrate of Ofloxacin (1407 cm⁻¹), h = 2 μ m = 2 × 10⁻⁴ cm, ρ = 1.2688 (estimate) g/cm³, M = 361.37 g/mol, A_{substrate} = 4 π mm² = 4 π × 10⁻² cm², C = 10⁻⁶ mol/L, V = 5 μ L = 5 × 10⁻⁶ L.

 I_{SERS} and I_{Raman} values of Ofloxacin based on AgNPs, ZnO/Ag, and Ofloxacin powder were estimated using the spectra in Figure S5.

Calculation of HOMO and LUMO energy levels

The HOMO and LUMO energy levels of Ofloxacin can be estimated based on its onset oxidation and reduction potentials (ϕ_{ox} and ϕ_{red}), respectively, using the equations:[4-6]

$$E_{\text{HOMO}} = -e \left(\phi_{\text{ox}} + 4.8 - \phi_{\text{Fc/Fc+}}\right) \tag{9}$$

$$E_{LUMO} = -e \left(\phi_{red} + 4.8 - \phi_{Fc/Fc+}\right) \tag{10}$$

in which ϕ_{Fc/Fc^+} is the redox potential of ferrocene/ferrocenium couple (Fc/Fc⁺) in the electrochemical system, assuming the energy level of Fc/Fc⁺ to be -4.8 eV below vacuum level. Similar to the previous study, we set up an electrochemical system based on the description of Bin *et al.* with a Pt working electrode and a Ag/AgCl reference electrode.[4] Therefore, ϕ_{Fc/Fc^+} was assumed to be 0.44 V versus Ag/AgCl.[4]



Figure S1: SEM image of the ZnO nanosheets.



Figure S2: Determination of the HOMO–LUMO energy levels of Ofloxacin analyte via cyclic voltammetry measurements.



Figure S3: Structure of Ofloxacin [7].



Figure S4: CV curves (a) recorded on Ag_SPE, ZnO/Ag 16-84_SPE, ZnO/Ag 50-50_SPE in 0.1 M KCl containing 5 mM [Fe(CN)₆] ³⁻/[Fe(CN)₆] ⁴⁻.



Figure S5: The energy level of AgNPs, ZnO nanosheets, and Ofloxacin analyte.



Figure S6: Raman of TCZ; and SERS spectrum of AgNPs and ZnO/Ag for Ofloxacin (10⁻⁶ M).



Figure S 7: Raman spectra of original ophthalmic solution (OFL 0.3%) and several diluted ones.

Real sample	Analyte	Concentration of TCZ (M)	Recovery (%)
Beef	OFL	10-5	91.41
		10-6	89.86
		10-7	89.37
		10-8	86.57
		10-9	87.95

Table S1: The recovery values for five concentrations of OFL in the beef sample.

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