## **Supporting Information**

## **Bifunctional Additive-Driven Shape Transitions of Block Copolymer Particles through Synergistic Quaternization and Protonation**

Zhengping Tan,<sup>1,#</sup> Soohyun Ban,<sup>2,#</sup> Younghyeon Ahn,<sup>1</sup> Kang Hee Ku,<sup>2,\*</sup> Bumjoon J. Kim<sup>1,\*</sup>

<sup>1</sup>Department of Chemical and Biomolecular Engineering, Korea Advanced Institute of Science and Technology (KAIST), Daejeon 34141, Republic of Korea <sup>2</sup>School of Energy and Chemical Engineering, Ulsan National Institute of Science and Technology (UNIST), Ulsan 44919, Republic of Korea

\*E-mail: bumjoonkim@kaist.ac.kr (B.J.K), kangheeku@unist.ac.kr (K.H.K)



**Figure S1.** SEM and TEM images of PS-*b*-P2VP(BNA)<sub>*x*</sub> particles at (a, b) x = 0 and (c, d) x = 0.3. Scale bars are 1  $\mu$ m in the SEM images and 100 nm in the TEM images.



**Figure S2.** <sup>1</sup>H NMR spectra of solution containing BNA, PS-*b*-P2VP, PS-*b*-P2VP(BNA)<sub>0.6</sub>. The new proton peaks highlighted by light purple framework between 8.8 ppm and 9.5 ppm in (C) indicate the reacted pyridine group of PS-*b*-P2VP while the light green framework in (B) or (C) indicate the non-reacted pyridine group of PS-*b*-P2VP. Inset above (C) is the magnified NMR between 8.8 ppm and 9.5 ppm for BNA, PS-*b*-P2VP, PS-*b*-P2VP(BNA)<sub>0.6</sub>.

**Table S1.** Interfacial tension ( $\gamma$ , mN m<sup>-1</sup>) between the polymer solution (i.e., PS<sub>50k</sub> or P2VP<sub>50k</sub> or P2VP<sub>50k</sub>(BNA)<sub>x</sub> in chloroform, 10 mg mL<sup>-1</sup>) and the aqueous surfactant solution (i.e., CTAB in water, 10 mg mL<sup>-1</sup>)

PS	P2VP	$P2VP(BNA)_x$							
		x = 0.1	x = 0.3	x = 0.5	x = 0.6	x = 0.8	<i>x</i> = 1.2	<i>x</i> = 1.5	
3.99 ± 0.25	$\begin{array}{c} 4.66 \pm \\ 0.18 \end{array}$	$\begin{array}{c} 4.62 \pm \\ 0.28 \end{array}$	$\begin{array}{c} 4.51 \pm \\ 0.58 \end{array}$	$\begin{array}{c} 4.38 \pm \\ 0.13 \end{array}$	$\begin{array}{c} 4.26 \pm \\ 0.17 \end{array}$	$\begin{array}{c} 3.93 \pm \\ 0.24 \end{array}$	$\begin{array}{c} 3.48 \pm \\ 0.41 \end{array}$	$\begin{array}{c} 3.32 \pm \\ 0.31 \end{array}$	



**Figure S3.** (a) SEM and (b) TEM images of ellipsoidal  $PS_{102k}$ -*b*-P2VP<sub>97k</sub> particles prepared using a mixture of polyvinyl alcohol (PVA) and CTAB in a 4:1 weight ratio, creating a neutral preference of PS-*b*-P2VP to the surrounding aqueous medium. Scale bars are 1  $\mu$ m in the SEM image and 100 nm in the TEM image.



**Figure S4.** (a) Schematic illustration showing the production of PS-*b*-P2VP discs through disassembly of conventional ellipsoidal particles in acidic solution (pH = 3.0). (b) Side-view SEM images of  $PS_{102k}$ -*b*-P2VP<sub>97k</sub> disc-like particles.



**Figure S5.** (a) TEM image of Janus cups. (b) Statistical distribution of L values for Janus cups, determined by analyzing 100 particles from TEM images.

**Table S2.** Degree of reaction of P2VP ( $\alpha$ , %) by BNA at different feed molar ratios (x), which was determined by the molar fraction of reacted 2VP units in the reacted P2VP block [reacted  $n_{2VP}/(\text{reacted } n_{2VP} + n_{2VP})$ ].

x	α (%)
0.1	10.5
0.3	12.5
0.5	14.6
0.6	15.7
0.8	17.8
1.2	19.2
1.5	21.2

Dorticle gueronaion	pH value with				
Particle suspension	BN	NA	BNA		
PS-b-P2VP (pristine)	$6.96\pm0.35$	$6.96\pm0.35$	$6.96\pm0.35$		
PS- <i>b</i> -P2VP(additive) <sub>0.1</sub>	$6.86\pm0.42$	$6.81\pm0.33$	$6.86\pm0.44$		
PS-b-P2VP(additive) <sub>0.3</sub>	$6.82\pm0.15$	$5.72\pm0.27$	$5.57\pm0.19$		
PS-b-P2VP(additive) <sub>0.5</sub>	$6.89 \pm 0.26$	$5.12\pm0.38$	$4.89\pm0.32$		
PS-b-P2VP(additive) <sub>0.6</sub>	$6.91\pm 0.47$	$4.85\pm0.19$	$4.61\pm0.27$		
PS-b-P2VP(additive) <sub>0.8</sub>	$6.85\pm0.29$	$4.73\pm0.22$	$4.52\pm0.35$		
PS- <i>b</i> -P2VP(additive) <sub>1.2</sub>	$6.94\pm0.32$	$3.89\pm0.43$	$3.68\pm0.34$		
PS-b-P2VP(additive) <sub>1.5</sub>	$6.83\pm0.18$	$3.45\pm0.21$	$3.25\pm0.21$		

**Table S3.** The pH values of the PS-*b*-P2VP(BN)<sub>*x*</sub>, PS-*b*-P2VP(NA)<sub>*x*</sub>, PS-*b*-P2VP(BNA)<sub>*x*</sub> particle suspension at different feed molar ratios (*x*). The pH values were measured by pH meter (Ohaus ST3100 pH Bench).



**Figure S6.** FTIR spectra of  $PS_{102k}$ -*b*-P2VP<sub>97k</sub>(BNA)<sub>x</sub> particles obtained at different *x* values. (a) (a) The full FTIR spectra and (b, c) amplified FTIR spectra of  $PS_{102k}$ -*b*-P2VP<sub>97k</sub>(BNA)<sub>x</sub> particles obtained at different *x* values (b) from 950 cm<sup>-1</sup> to 1100 cm<sup>-1</sup> and (c) from 1500 cm<sup>-1</sup> to 1650 cm<sup>-1</sup>. Framework in (b) indicates the broaden spectra around 1000 cm<sup>-1</sup> after quaternization while the arrow in (c) indicates the shift after protonation.



**Figure S7.** SEM and TEM images of PS-*b*-P2VP(BN)<sub>*x*</sub> particles at different *x* values. Scale bars are 1  $\mu$ m in the SEM image and 100 nm in the TEM image.



**Figure S8.** SEM and TEM images of PS-*b*-P2VP(NA)<sub>*x*</sub> particles at different *x* values. Scale bars are 1  $\mu$ m in the SEM image and 100 nm in the TEM image.



**Figure S9.** (a) SEM and (b) TEM images of Janus cups after crosslinked with dibromobutane for 24 h.