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## Supporting information

## Magnesium Borohydride Pyridine Derivatives as Electrolytes for All-Solid-State Batteries

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Figure S1. <sup>1</sup>H NMR spectra of s1-s3.



**Figure S2.** Rietveld refinement of Mg(BH<sub>4</sub>)<sub>2</sub>·3N(CH)<sub>5</sub> (**s1**) from SR PXD data measured at T = 20 °C,  $\lambda = 0.70870 \text{ Å}$ , showing experimental (red circles) and calculated (black line) PXD patterns, and a difference plot below (blue line). Final discrepancy factors:  $R_p = 2.51 \%$ ,  $R_{wp} = 2.19 \%$  (not corrected for background),  $R_p = 11.2 \%$ ,  $R_{wp} = 18.2 \%$  (conventional Rietveld R-factors),  $R_{Bragg}(Mg(BH_4)_2 \cdot 3N(CH)_5) = 9.80 \%$  and global  $\chi^2 = 10.1$ .



**Figure S3.** Rietveld refinement of Mg(BH<sub>4</sub>)<sub>2</sub>·2N(CH)<sub>5</sub> (**s2**) from SR PXD data measured at T = 20 °C,  $\lambda = 0.826927$  Å, showing experimental (red circles) and calculated (black line) PXD patterns, and a difference plot below (blue line). Final discrepancy factors:  $R_p = 1.60$  %,  $R_{wp} = 2.74$  % (not corrected for background),  $R_p = 20.4$  %,  $R_{wp} = 14.2$  % (conventional Rietveld R-factors),  $R_{Bragg}(Mg(BH_4)_2 \cdot 2N(CH)_5) = 10.2$  % and global  $\chi^2 = 13.8$ .



**Figure S4.** Thermal analysis (TG-MS) data of **s3** during heating with a heating rate of 0.5 °C/min. The apparent increasing weight% is an artifact from the instrument.



**Figure S5.** Fitting of the Nyquist data of  $Mg(BH_4)_2 \cdot 3N(CH)_5$  (s1) at room temperature after stabilization.



**Figure S6.** Nyquist plots of  $Mg(BH_4)_2 \cdot 3N(CH)_5$  (s1) from electrochemical impedance spectroscopy measurements during stabilization at room temperature in intervals of 1 h.



**Figure S7.** Time-evolution of the ionic conductivity of **s1-s3** measured by electrochemical impedance spectroscopy at room temperature in intervals of 1 h.



**Figure S8.** Arrhenius plot of  $\ln(\sigma T)$  vs 1000/T to evaluate the activation energy.