Supporting Information

Solving ZIB Challenges: The Dynamic Role of Water in Deep Eutectic Solvents electrolyte

E. Emanuele^{1*}, G. Batignani², G. Cerullo³ G. Leita¹, N. Madathiparambil Mohanan², E. Mai^{2,4}, M. Martinati² C. Mele⁵, T. Scopigno², B. Bozzini¹

¹Department of Energy, Politecnico di Milano, via Lambuschini 4, 20156 Milano, Italy ²Dipartimento di Fisica, Universitá di Roma "La Sapienza", Roma I-00185, Italy ³IFN-CNR, Dipartimento di Fisica, Politecnico di Milano, Piazza Leonardo da Vinci 32, I-20133 Milano, Italy ⁴Center for Life Nano Science Center for life Nano Science @Sapienza, Istituto Italiano di Tecnologia, Viale Regina Elena 291, I-00161 Roma, Italy ⁵Department of Innovation Engineering, University of Salento, Via Monteroni, 73100 Lecce, Italy



Fig. S1 – Schematic of coin-cell structure and assembly.



Fig. S2 - Spontaneous Raman spectra of pure anhydrous ethaline, 0.1 M and 0.3 M ZnSO₄ DESEG electrolytes.



Fig. S3 - Spontaneous Raman spectra (dot lines) and SRS (solid lines) of DESEG with different hydration levels. A logarithmic y-scale has been used for improved clarity.



Fig. S4 - Spontaneous Raman spectra (black line) and SRS (red line) of anhydrous DESEG.



Fig. S5 –Sketch of the Impulsive Stimulated Raman spectroscopy (IVS) experimental concept: two pulses are exploited to stimulate and read out vibrational excitations directly in the time-domain, as a function of the pulse delay (Δt). Fast Fourier Transforming (FFT) over Δt allows for retrieving the vibrational response in the frequency-domain.



Fig. S6 ISRS spectra normalized to the intensity of the Zn-Cl peak



Fig. S7 - Galvanostatic charge-discharge cycling experiments replicates carried out with Zn symmetric CR2032 coin cells DESEG with different hydration levels.