## Electronic Supplementary Information for:

Semi-vacant Wells-Dawson anions. Synthesis of tritungsten-vacant derivatives and crystallographic studies of  $[\alpha\beta\beta\alpha-(Cu^{\parallel}OH_{2})_{2}(Cu^{\parallel})_{2}(AsW_{15}(OH_{2})_{3}-(OH)O_{52})_{2}]^{12}$ 

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**Figure S1.** Thermal ellipsoid plot of **2** (50% probability surfaces). The slight disorder (4%) in the  $W_3O_{13}$  "cap" units is shown (each cap is 96% α-isomer and 4% β-isomer).

**Table S1.** W(VI)-based reduction potentials for 1, 1Zn, and 1Cu.

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Figure S1.

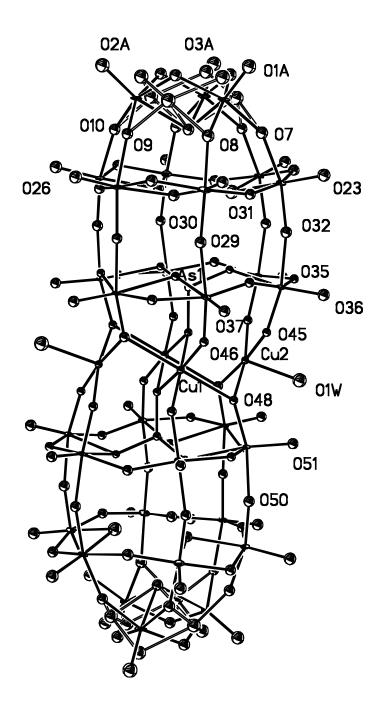


Table S1. W(VI)-based reduction potentials for 1, 1Zn, and 1Cu.<sup>a</sup>

complex	-E <sub>pc1</sub> (V)	-E <sub>pc2</sub> (V)	-E <sub>pc3</sub> (V)
1	0.528	0.644	0.876
1Zn	0.576	0.736	0.924
1Cu	0.588 <sup>b</sup>	0.648	0.820

<sup>a</sup>Conditions: POM concentration:  $2 \times 10^{-4} \text{ M}$  in 0.4 M CH<sub>3</sub>COOLi/CH<sub>3</sub>COOH (pH 5) solution; scan rate:  $10 \text{ mV s}^{-1}$ ; working electrode: glassy carbon; reference electrode: SCE; <sup>b</sup>This wave appears as a shoulder following a copper deposition wave with a cathodic peak potential  $E_{pc} = -0.320 \text{ V}$  and a characteristic desorptive oxidation peak potential located at  $E_{pa} = -0.10 \text{ V}$ .