

Supplementary material (ESI):

Synthesis, Crystal structure and Optical properties of [Ag(UO₂)₃(OAc)₉][Zn(H₂O)₄(CH₃CH₂OH)₂]: A Novel Compound Containing Closed-shell 3d¹⁰, 4d¹⁰ and 5d¹⁰ Metal Ions

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1. Experimental Procedures

IR spectra were recorded on a Nicolet AVATAT FT-IR360 spectrometer as KBr pellets in the frequency range 4000-400 cm⁻¹. ¹H NMR and ¹³C NMR were acquired in CDCl₃ on Varian Unity⁺ 400MHz NMR spectrometer using TMS as an internal standard. Powder X-ray diffraction (PXRD) data were collected on a Philips X'Pert Pro MPD X-ray diffractometer with Cu K_α radiation equipped with a X'Celerator detector. The elemental analyses (C, H, N contents) were determined on a CE instruments EA 1110 analyzer. The emission of dark-green light from **1** is easily observed when the sample is irradiated with 254 or 356 nm UV light using a handheld UV lamp. Photoluminescence measurements were performed on a Hitachi F-4500 fluorescence spectrophotometer with solid powder on a 1 cm quartz round plate (for emission spectra, excitation wavelength: 365, 424 and 447 nm; emission wavelengths: 400-800 nm; slit width: 2.5 nm; sensitivity: high. for excitation spectra, emission wavelength: 481, 501, 523 and 546 nm; excitation wavelengths: 300-500 nm; slit width: 2.5 nm; sensitivity: high).

2. Tables of anisotropic thermal parameters, selected bond distances and angles and hydrogen bond geometry for **1**.

Table S1. atomic coordinates (x 10⁴) and equivalent isotropic displacement parameters (Å² x 10³) for **1**. U(eq) is defined as one third of the trace of the orthogonalized Uij tensor.

| | x | y | z | U(eq) |
|-----|---------|---------|----------|-------|
| Zn1 | 2500 | 7500 | 0000 | 29(1) |
| Ag1 | 5000 | 2508(1) | 2500 | 43(1) |
| U1 | 5000 | 5406(1) | 2500 | 36(1) |
| U2 | 6470(2) | 846(1) | 2541(3) | 33(1) |
| C1 | 7490(6) | -313(8) | 2587(9) | 36(3) |
| C2 | 8052(8) | -886(1) | 2635(15) | 80(6) |
| C3 | 6122(6) | 746(9) | 3986(9) | 40(3) |
| C4 | 5910(8) | 685(12) | 4704(12) | 70(5) |

| | | | | |
|-----|---------|----------|----------|-------|
| C5 | 5770(6) | 2185(9) | 1048(9) | 46(3) |
| C6 | 5430(8) | 2904(12) | 312(13) | 79(5) |
| C7 | 6134(6) | 4351(10) | 3853(10) | 50(3) |
| C8 | 6706(8) | 3817(11) | 4569(12) | 76(5) |
| C9 | 5000 | 7429(12) | 2500 | 43(5) |
| C10 | 5000 | 8509(13) | 2500 | 67(6) |
| C11 | 3104(7) | 7071(10) | 2173(10) | 56(4) |
| C12 | 3584(8) | 7387(13) | 3135(10) | 69(5) |
| O1 | 5856(4) | -3(6) | 1904(7) | 50(2) |
| O2 | 7078(4) | 1713(6) | 3176(6) | 48(2) |
| O3 | 5881(5) | 1348(6) | 3343(8) | 59(3) |
| O4 | 6552(4) | 185(6) | 4007(6) | 45(2) |
| O5 | 7368(4) | -341(6) | 3245(6) | 45(2) |
| O6 | 7120(4) | 204(6) | 1862(7) | 49(2) |
| O7 | 6226(4) | 1654(6) | 1069(6) | 48(2) |
| O8 | 5645(4) | 2093(6) | 1693(7) | 52(2) |
| O9 | 6149(4) | 5247(6) | 3869(7) | 55(2) |
| O10 | 5619(4) | 3951(6) | 3240(8) | 58(3) |
| O11 | 4691(4) | 5411(6) | 3280(7) | 58(3) |
| O12 | 5517(4) | 6986(6) | 3060(6) | 44(2) |
| O13 | 3197(4) | 7476(6) | 1464(6) | 39(2) |
| O1W | 3134(4) | 8508(6) | -39(6) | 48(2) |
| O2W | 1975(4) | 8531(5) | 260(6) | 39(2) |

Table S2 Selected bond lengths (\AA) and angles ($^\circ$) for **1**

| | | | | | |
|---------------------------------|----------|--------------------------|----------|--|----------|
| Ag(1)-O(3) | 2.412(9) | Ag(1)-O(3) ⁱⁱ | 2.412(9) | Ag(1)-O(10) | 2.414(9) |
| Ag(1)-O(10) ⁱⁱ | 2.414(9) | Ag(1)-O(8) | 2.483(8) | Ag(1)-O(8) ⁱⁱ | 2.483(8) |
| U(1)-O(11) | 1.732(9) | U(1)-O(11) ⁱⁱ | 1.732(9) | U(1)-O(10) | 2.427(8) |
| U(1)-O(10) ⁱⁱ | 2.427(8) | U(1)-O(12) | 2.455(8) | U(1)-O(12) ⁱⁱ | 2.455(8) |
| U(1)-O(9) | 2.467(8) | U(1)-O(9) ⁱⁱ | 2.467(8) | U(2)-O(1) | 1.734(9) |
| U(2)-O(2) | 1.743(8) | U(2)-O(3) | 2.402(8) | U(2)-O(6) | 2.423(8) |
| U(2)-O(8) | 2.432(8) | U(2)-O(5) | 2.439(7) | U(2)-O(7) | 2.440(8) |
| U(2)-O(4) | 2.473(8) | Zn(1)-O(13) | 2.101(8) | Zn(1)-O(13) ⁱ | 2.101(8) |
| Zn(1)-O(1w) | 2.047(7) | Zn(1)-O(1w) ⁱ | 2.047(7) | Zn(1)-O(2w) | 2.051(7) |
| Zn(1)-O(2w) ⁱ | 2.051(7) | | | | |
| O(3)-Ag(1)-O(3) ⁱⁱ | 95.6(5) | | | O(10)-Ag(1)-O(3) ⁱⁱ | 164.1(3) |
| O(3)-Ag(1)-O(10) | 99.3(3) | | | O(3)-Ag(1)-O(10) ⁱⁱ | 164.1(3) |
| O(10)-Ag(1)-O(10) ⁱⁱ | 66.6(4) | | | O(3) ⁱⁱ -Ag(1)-O(8) | 95.4(3) |
| O(3)-Ag(1)-O(8) | 66.0(3) | | | O(10)-Ag(1)-O(8) | 95.8(3) |
| O(10) ⁱⁱ -Ag(1)-O(8) | 106.8(3) | | | O(3) ⁱⁱ -Ag(1)-O(8) ⁱⁱ | 66.0(3) |

| | | | |
|---|----------|---|------------|
| O(3)-Ag(1)-O(8) ⁱⁱ | 95.4(3) | O(10) ⁱⁱ -Ag(1)-O(8) | 106.8(3) |
| O(10) ⁱⁱ -Ag(1)-O(8) ⁱⁱ | 95.8(3) | O(10)-Ag(1)-O(8) ⁱⁱ | 106.8(3) |
| O(10) ⁱⁱ -Ag(1)-O(8) ⁱⁱ | 95.7(3) | O(8)-Ag(1)-O(8) ⁱⁱ | 153.0(4) |
| O(11) ⁱⁱ -U(1)-O(11) | 179.5(6) | O(11) ⁱⁱ -U(1)-O(10) ⁱⁱ | 89.6(4) |
| O(11)-U(1)-O(10) ⁱⁱ | 90.8(4) | O(11) ⁱⁱ -U(1)-O(10) | 90.8(4) |
| O(11)-U(1)-O(10) | 89.6(4) | O(10) ⁱⁱ -U(1)-O(10) | 66.2(4) |
| O(11) ⁱⁱ -U(1)-O(12) | 88.2(4) | O(11)-U(1)-O(12) | 91.3(4) |
| O(10) ⁱⁱ -U(1)-O(12) | 172.5(3) | O(10)-U(1)-O(12) | 121.0(3) |
| O(11) ⁱⁱ -U(1)-O(12) ⁱⁱ | 91.3(4) | O(11)-U(1)-O(12) ⁱⁱ | 88.3(4) |
| O(10) ⁱⁱ -U(1)-O(12) ⁱⁱ | 121.0(3) | O(10)-U(1)-O(12) ⁱⁱ | 172.5(3) |
| O(12)-U(1)-O(12) ⁱⁱ | 51.9(4) | O(11) ⁱⁱ -U(1)-O(9) ⁱⁱ | 89.1(4) |
| O(11)-U(1)-O(9) ⁱⁱ | 91.0(4) | O(10) ⁱⁱ -U(1)-O(9) ⁱⁱ | 51.7(3) |
| O(10)-U(1)-O(9) ⁱⁱ | 117.9(3) | O(12)-U(1)-O(9) ⁱⁱ | 121.0(3) |
| O(12) ⁱⁱ -U(1)-O(9) ⁱⁱ | 69.3(3) | O(11) ⁱⁱ -U(1)-O(9) | 91.0(4) |
| O(11)-U(1)-O(9) | 89.1(4) | O(10) ⁱⁱ -U(1)-O(9) | 117.9(3) |
| O(10)-U(1)-O(9) | 51.7(3) | O(12)-U(1)-O(9) | 69.3(3) |
| O(12) ⁱⁱ -U(1)-O(9) | 121.0(3) | O(9) ⁱⁱ -U(1)-O(9) | 169.7(4) |
| O(1)-U(2)-O(2) | 179.1(4) | O(1)-U(2)-O(3) | 90.8(4) |
| O(2)-U(2)-O(3) | 88.6(4) | O(1)-U(2)-O(6) | 88.6(3) |
| O(2)-U(2)-O(6) | 92.0(3) | O(3)-U(2)-O(6) | 173.7(3) |
| O(1)-U(2)-O(8) | 89.0(4) | O(2)-U(2)-O(8) | 90.2(4) |
| O(3)-U(2)-O(8) | 67.0(3) | O(6)-U(2)-O(8) | 119.3(3) |
| O(1)-U(2)-O(5) | 93.4(4) | O(2)-U(2)-O(5) | 87.5(3) |
| O(3)-U(2)-O(5) | 121.0(3) | O(6)-U(2)-O(5) | 52.8(3) |
| O(8)-U(2)-O(5) | 171.6(3) | O(1)-U(2)-O(7) | 90.6(4) |
| O(2)-U(2)-O(7) | 89.0(4) | O(3)-U(2)-O(7) | 118.6(3) |
| O(6)-U(2)-O(7) | 67.6(3) | O(8)-U(2)-O(7) | 51.7(3) |
| O(5)-U(2)-O(7) | 120.1(3) | O(1)-U(2)-O(4) | 88.1(4) |
| O(2)-U(2)-O(4) | 92.0(4) | O(3)-U(2)-O(4) | 51.6(3) |
| O(6)-U(2)-O(4) | 122.1(3) | O(8)-U(2)-O(4) | 118.5(3) |
| O(5)-U(2)-O(4) | 69.7(3) | O(7)-U(2)-O(4) | 170.1(3) |
| U(2)-O(8)-Ag(1) | 111.6(3) | Ag(1)-O(10)-U(1) | 113.6(4) |
| C(11)-O(13)-Zn(1) | 126.7(7) | O(1w)-Zn(1)-O(1w) ⁱ | 180.000(1) |
| O(1w) ⁱ -Zn(1)-O(2w) | 88.8(3) | O(1w) ⁱ -Zn(1)-O(2w) ⁱ | 91.2(3) |
| O(1w)-Zn(1)-O(2w) | 91.2(3) | O(1w)-Zn(1)-O(2w) ⁱ | 88.8(3) |
| O(2w) ⁱ -Zn(1)-O(2w) | 180.0(5) | O(1w)-Zn(1)-O(13) ⁱ | 96.0(3) |
| O(1w) ⁱ -Zn(1)-O(13) ⁱ | 83.6(3) | O(2w) ⁱ -Zn(1)-O(13) ⁱ | 88.4(3) |
| O(2w)-Zn(1)-O(13) ⁱ | 91.6(3) | O(1w)-Zn(1)-O(13) | 84.0(3) |
| O(1w) ⁱ -Zn(1)-O(13) | 96.0(3) | O(2w) ⁱ -Zn(1)-O(13) | 91.6(3) |
| O(2w)-Zn(1)-O(13) | 88.4(3) | O(13) ⁱ -Zn(1)-O(13) | 180.0(4) |

^a Symmetry code i, -x+1/2, -y+3/2, -z; ii, -x+1, y, -z+1/2

Table S3 Hydrogen bond geometry (\AA , $^\circ$) for **1**

| D-H...A | D-H | H...A | D...A | D-H...A |
|------------------------------------|------|-------|-----------|---------|
| O(13)-H(13B)...O(12) ⁱ | 0.82 | 1.96 | 2.733(10) | 158.1 |
| O(1w)-H(1wA)...O(4) ⁱⁱ | 0.82 | 2.36 | 2.759(11) | 110.4 |
| O(1w)-H(1wB)...O(7) ⁱⁱⁱ | 0.82 | 1.95 | 2.728(10) | 158.8 |
| O(2w)-H(2wD)...O(9) ^{iv} | 0.82 | 2.08 | 2.708(11) | 132.9 |
| O(2w)-H(2wA)...O(5) ⁱⁱ | 0.82 | 1.94 | 2.645(11) | 143.0 |

Symmetry codes: (i), -x+1, y, -z +1/2; (ii), -x+1, y+1, -z+1/2; (iii), -x+1, -y+1, -z; (iv), x-1/2, -y+3/2, z-1/2

Table S4. Anisotropic displacement parameters ($\text{\AA}^2 \times 10^3$) for **1**. The anisotropic displacement factor exponent takes the form: $-2 \pi^2 [h^2 a^*{}^2 U_{11} + \dots + 2 h k a^* b^* U_{12}]$

| | U11 | U22 | U33 | U23 | U13 | U12 |
|-------|--------|---------|---------|--------|--------|--------|
| Zn(1) | 32(1) | 25(1) | 26(1) | 0(1) | 11(1) | 0(1) |
| Ag(1) | 39(1) | 25(1) | 60(1) | 0 | 23(1) | 0 |
| U(1) | 31(1) | 25(1) | 37(1) | 0 | 6(1) | 0 |
| U(2) | 36(1) | 30(1) | 31(1) | 8(1) | 16(1) | 4(1) |
| C(1) | 41(7) | 35(7) | 35(7) | 6(6) | 22(6) | -2(5) |
| C(2) | 83(12) | 76(11) | 115(17) | 12(11) | 76(14) | 24(9) |
| C(3) | 50(8) | 45(7) | 31(7) | 16(6) | 26(6) | 20(6) |
| C(4) | 82(11) | 82(11) | 62(11) | 36(9) | 51(10) | 34(9) |
| C(5) | 43(8) | 55(8) | 30(7) | 18(7) | 10(6) | 0(6) |
| C(6) | 72(11) | 78(11) | 86(14) | 36(11) | 39(11) | 13(9) |
| C(7) | 32(7) | 48(8) | 42(8) | 6(7) | -1(6) | -3(6) |
| C(8) | 72(11) | 57(9) | 53(10) | 15(8) | -4(8) | 26(8) |
| C(9) | 42(12) | 33(9) | 61(14) | 0 | 29(11) | 0 |
| C(10) | 78(16) | 25(10) | 83(18) | 0 | 26(14) | 0 |
| C(11) | 54(9) | 63(9) | 51(9) | 10(8) | 25(8) | -21(7) |
| C(12) | 63(10) | 115(14) | 18(7) | 7(8) | 8(7) | 6(9) |
| O(1) | 46(5) | 50(5) | 49(6) | 15(5) | 19(5) | -2(4) |
| O(1W) | 58(6) | 41(5) | 44(6) | -15(4) | 28(5) | -19(4) |
| O(2W) | 42(5) | 31(4) | 34(5) | 0(4) | 14(4) | 10(3) |
| O(2) | 45(5) | 44(5) | 39(5) | 7(4) | 9(4) | -2(4) |
| O(3) | 73(7) | 50(6) | 68(7) | 24(5) | 45(6) | 34(5) |
| O(4) | 56(5) | 44(5) | 43(5) | 19(4) | 31(5) | 17(4) |
| O(5) | 53(5) | 40(5) | 39(5) | 20(4) | 20(5) | 26(4) |

| | | | | | | |
|-------|-------|-------|-------|-------|--------|-------|
| O(6) | 57(6) | 50(5) | 39(6) | -6(5) | 23(5) | -8(4) |
| O(7) | 56(6) | 61(6) | 34(5) | 21(5) | 28(5) | 14(5) |
| O(8) | 53(6) | 61(6) | 45(6) | 24(5) | 30(5) | 27(5) |
| O(9) | 43(5) | 31(5) | 46(6) | 0(4) | -14(4) | -5(4) |
| O(10) | 41(5) | 31(5) | 66(7) | 2(4) | 0(5) | 5(4) |
| O(11) | 52(6) | 55(6) | 58(7) | 12(5) | 21(5) | 0(5) |
| O(12) | 26(4) | 39(5) | 50(6) | -5(4) | 6(4) | -4(4) |
| O(13) | 31(5) | 53(5) | 26(5) | 12(4) | 8(4) | 1(4) |

3. Figure S1 and S2: Polyhedral representation and Schematic depiction of crystal packing of **1**.

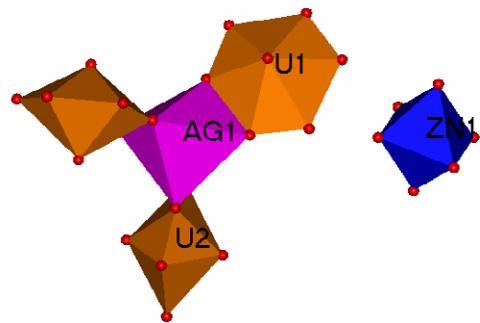


Figure S1 Polyhedral representation showing only the atoms coordinated directly to U, Ag and Zn for clarity.

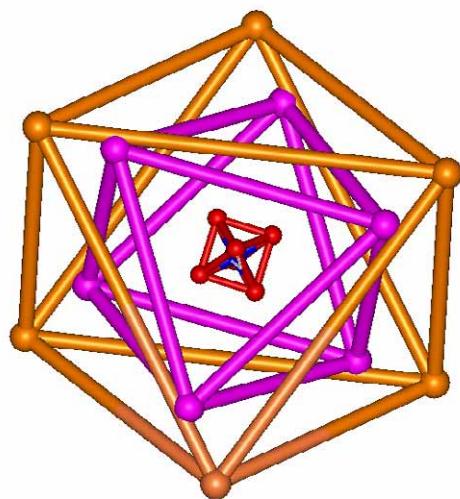


Figure S2 Schematic depiction of crystal packing of **1**.

4. IR spectra of compound **1**(Figure S3)

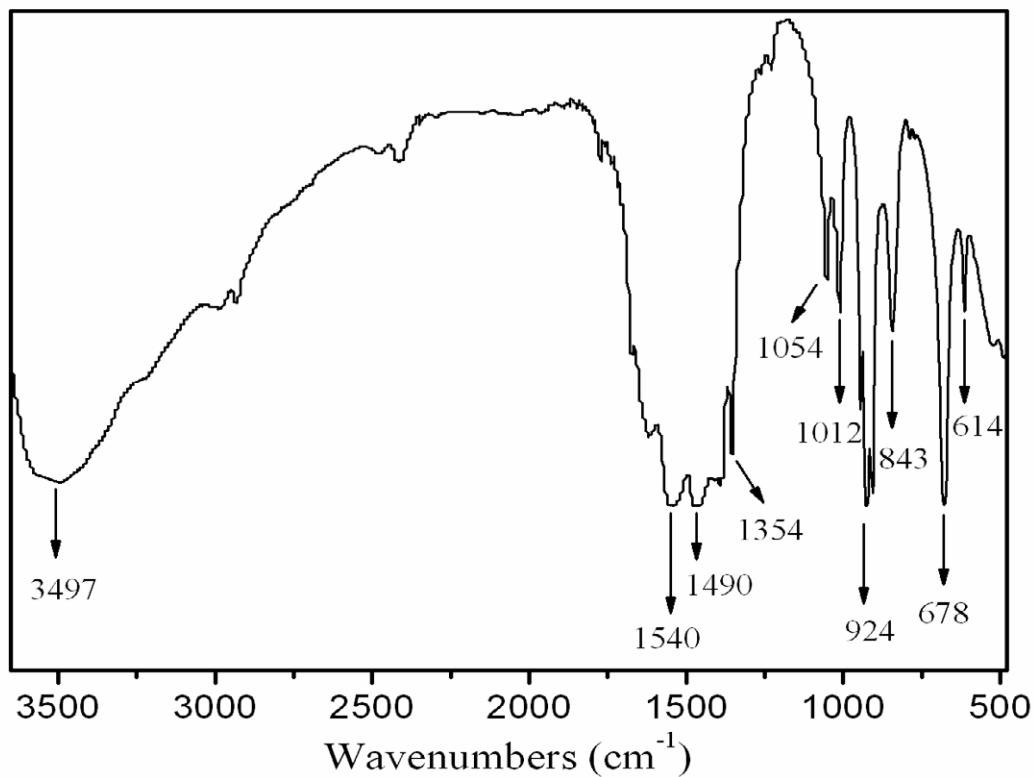


Figure S3: IR spectra of **1**. Note that the peaks at 3497, 1540, 1459 and 924 cm^{-1} , ascribed to $\nu(\text{OH})$, $\nu_{\text{as}}(\text{COO})$, $\nu_{\text{s}}(\text{COO})$ and $\nu(\text{UO}_2^{2+})$ stretching absorptions of **1** respectively.

5. ^1H NMR and ^{13}C NMR of **1**(Figure S4 & S5)

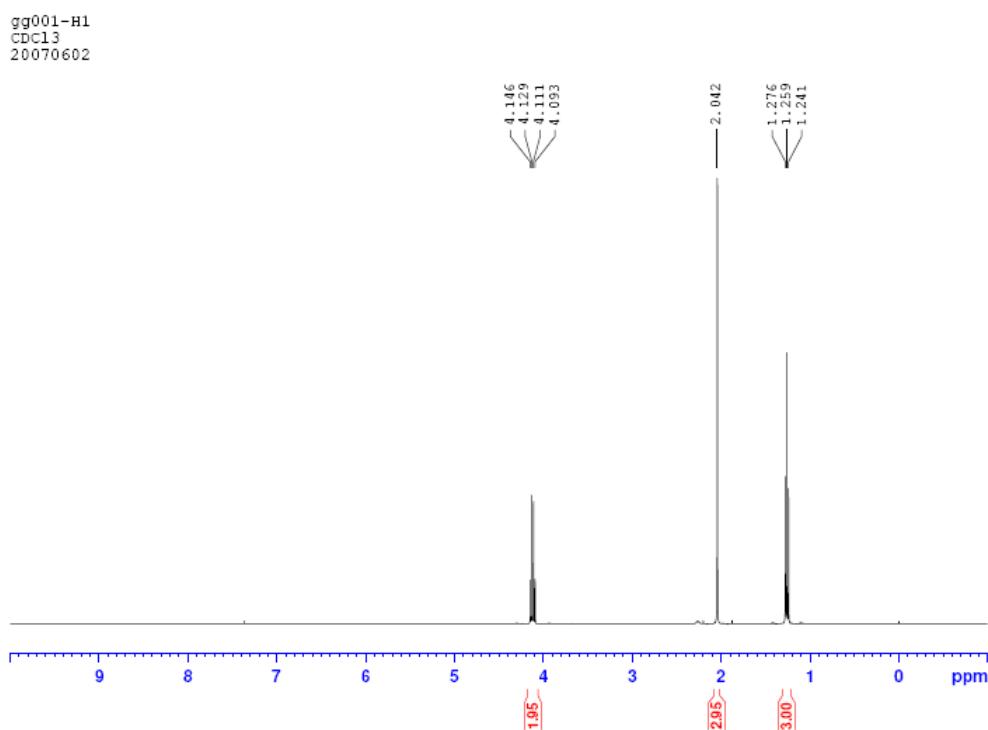


Figure S4: ^1H NMR of **1**. δ (ppm) = 4.13 (2H, m), 2.04(1H, s), 1.26(3H, t), respectively.

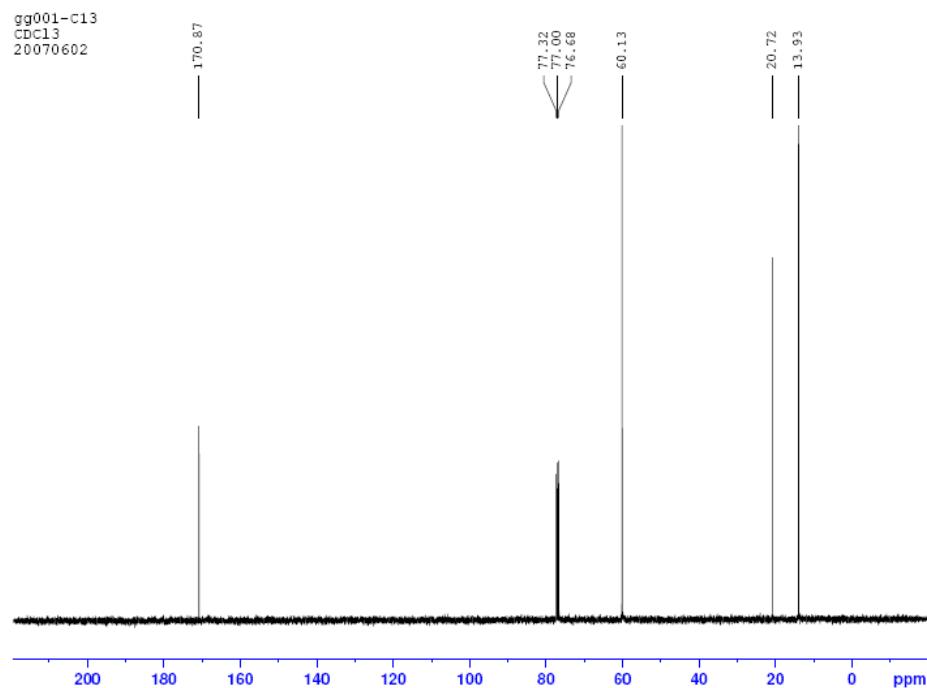


Figure S5: ^{13}C NMR of **1**. δ (ppm) = 13.93, 20.72, 60.13, 76.68, 77.00, 77.32, 170.87.

5. Powder XRD patterns of compound **1**(Figure S6)

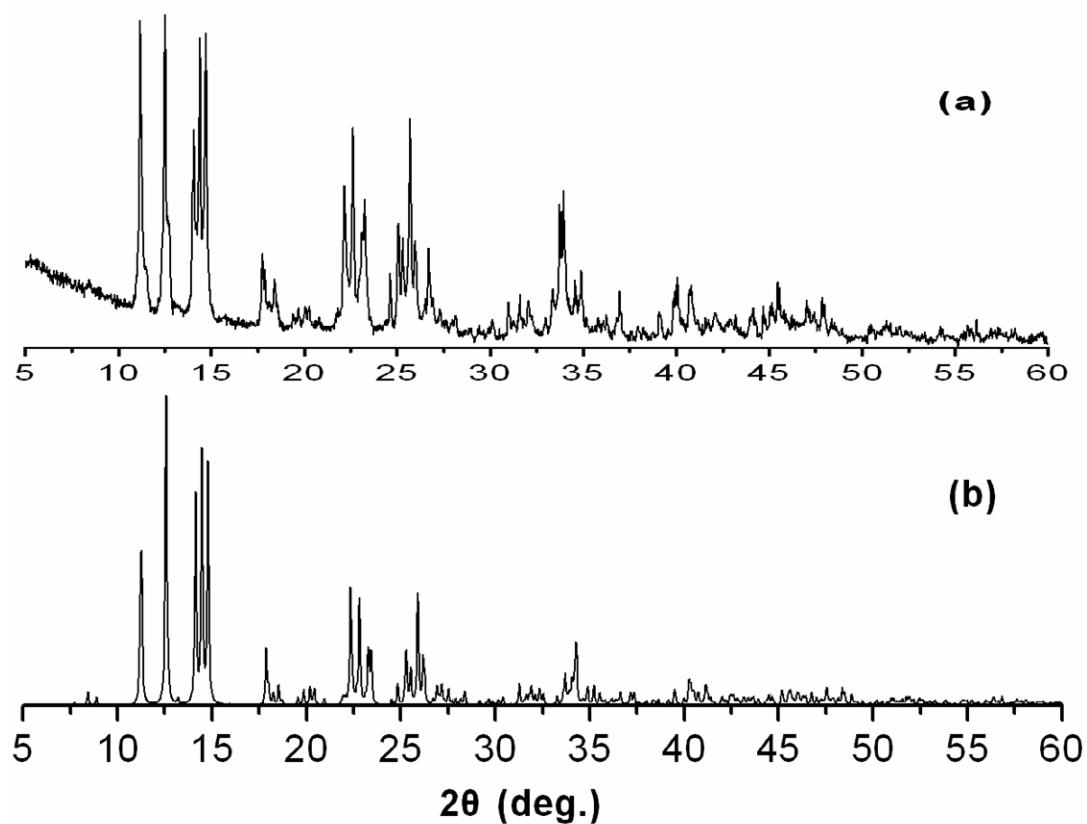


Figure S6: Powder XRD patterns of **1**: (a) Experimental powder X-ray diffraction pattern; (b) simulated from single crystal X-ray data.

7. Fluorescence excitation spectra and emission spectra of **1** in the solid state.

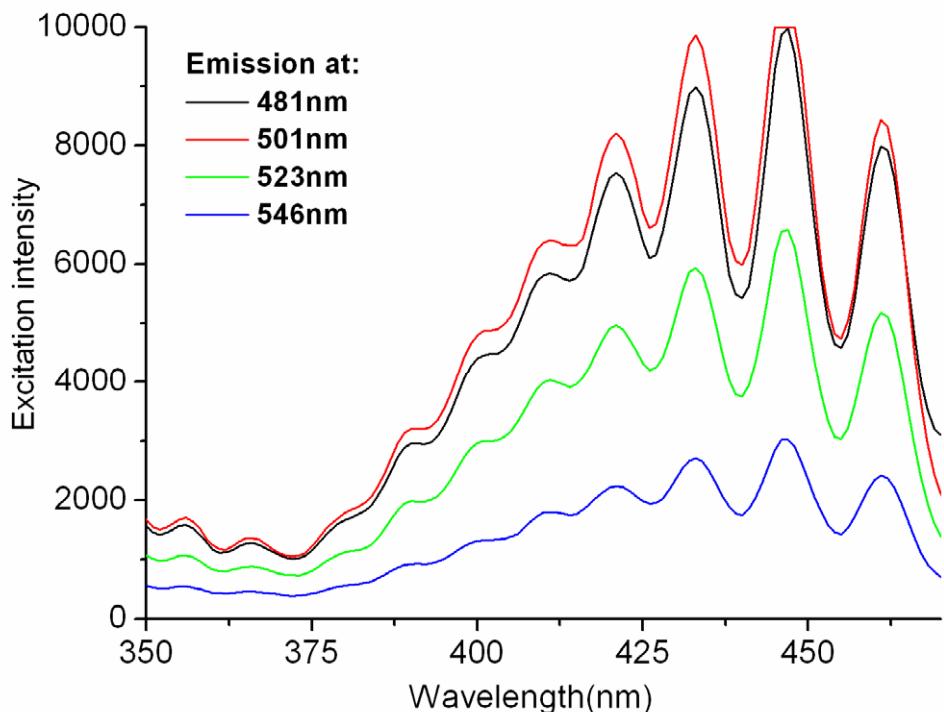


Figure S7-1 Solid-state excitation spectra for **1** at different emission wavelengths at RT.

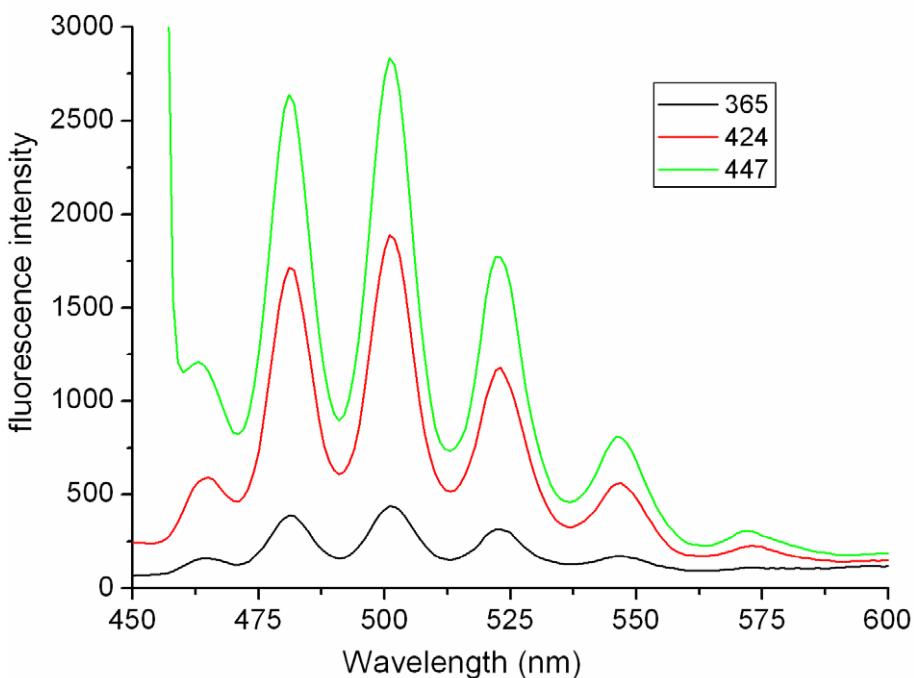


Figure S7-2 Solid-state emission spectra for **1** at different excitation wavelengths at RT.