

Electronic Supplementary Information for Dalton Transactions  
This journal is © The Royal Society of Chemistry 2007

## **Tuning the Self-assembly of Lanthanide Triple Stranded Heterobimetallic Helicates by Ligand Design**

Thomas B. Jensen, Rosario Scopelliti, and Jean-Claude G. Bünzli\*

*Contribution from the Laboratory of Lanthanide Supramolecular Chemistry, École  
Polytechnique Fédérale de Lausanne, BCH 1402, CH-1015 Lausanne (Switzerland)*

**Electronic Supplementary Information**

(11 pages)

**Table S1**  $^1\text{H}$  NMR of  $\text{L}^{\text{AB4}}$  in  $\text{CDCl}_3$ .

	1 or 3	2	23	8 and 10	3 or 1	15 and 17	20	6, 12, 21, 22	7 and 11			
$\delta$ / ppm	8.38	7.93	7.86	7.73	7.71	7.54	7.46	7.45	7.45	7.28 – 7.38	7.25	7.22
	dd 1H	t 1H	dd 1H	s 1H	s 1H	dd 1H	d 1H	d 1H	dd 1H	m 4H	dd 1H	dd 1H
$J$ / Hz	8.1; 1.2	7.9	6.8; 2.2			7.8; 1.1	2.4	2.4	6.5; 1.3		8.3; 1.8	8.3; 1.8
	4, 13, 18	9	24 or 26	28	26 or 24	5	14 and 19	25 or 27	29	27 or 25		
$\delta$ / ppm	4.68 – 4.79	4.30	3.61	3.54	3.36	1.45	1.34	1.32	1.29	1.26	1.08	
	m 6H	s 2H	q 2H	q 4H	q 2H	t 3H	t 3H	t 3H	t 3H	t 6H	t 3H	
$J$ / Hz			7.1	7.1	7.1	7.2	7.1	7.1	7.0	7.2	7.1	

**Table S2**  $^1\text{H}$  NMR of  $\text{L}^{\text{AB5}}$  in  $\text{CDCl}_3$ .

	15 and 17	1 or 3	2	23	8 and 10	3 or 1	20	6, 12, 21, 22	7 or 11			
$\delta$ / ppm	8.39	8.38	8.38	7.93	7.87	7.74	7.70	7.54	7.49	7.33 – 7.41	7.27	
	d 1H	d 1H	dd 1H	t 1H	dd 1H	d 1H	d 1H	dd 1H	dd 1H	m 4H	dd 1H	
$J$ / Hz	1.9	1.9	8.2; 1.1	7.9	7.3; 2.0	1.3	1.3	7.7; 1.1	7.4; 1.8		8.6; 1.5	
	11 or 7	18	13	4	9	24 and 26	5	19	14	25 and 27		
$\delta$ / ppm	7.25	4.81	4.77	4.75	4.31	3.61	3.36	1.45	1.38	1.29	1.29	1.08
	dd 1H	q 2H	q 2H	q 2H	s 2H	q 2H	q 2H	t 3H	t 3H	t 3H	t 3H	t 3H
$J$ / Hz	8.0; 1.5	7.2	7.2	7.1		7.1	7.1	7.1	7.2	7.2	7.1	7.1

**Table S3** Definition of least squares planes.

	$\text{L}^1$	$\text{L}^2$	$\text{L}^3$
<i>bpb moiety</i>			
terminal	1: C1-7;	2: C48-54;	3: C95-101;
benzimidazole	N1-2	N11-12	N21-22
imidazole only	1 <i>i</i> : C1,6-7; N1-2	2 <i>i</i> : C48,53-54; N11-12	3 <i>i</i> : C95,100- 101; N21-22
pyridine	4: C8-12; N3	5: C55-59; N13	6: C102-106; N23
bridging	7: C13-19;	8: C60-66;	9: C107-113;
benzimidazole	N4-5	N14-15	N24-25
imidazole only	7 <i>i</i> : C13-14,19; N4-5	8 <i>i</i> : C60-61,66; N14-15	9 <i>i</i> : C107-108, 113; N24-25
<i>bpa moiety</i>			
bridging	10: C21-27;	11: C68-74;	12: C115-121;
benzimidazole	N6-7	N16-17	N26-27
pyridine	13: C28-32; N8	14: C75-79; N18	15: C122-126; N28

**Table S4** Angles and distances between least squares planes.

Planes	[Pr <sub>2</sub> (L <sup>AB4</sup> ) <sub>3</sub> ](ClO <sub>4</sub> ) <sub>6</sub>		[Nd <sub>2</sub> (L <sup>AB4</sup> ) <sub>3</sub> ](ClO <sub>4</sub> ) <sub>6</sub>		[Sm <sub>2</sub> (L <sup>AB4</sup> ) <sub>3</sub> ](ClO <sub>4</sub> ) <sub>6</sub>	
	Angle	<i>d</i>	Angle	<i>d</i>	Angle	<i>d</i>
<i>1 - 6</i>	19.4 °	3.474 Å	20.3 °	3.492 Å	20.8 °	3.479 Å
<i>2 - 4</i>	43.4 °	4.247 Å	43.0 °	4.269 Å	44.0 °	4.231 Å
<i>3 - 5</i>	32.6 °	3.969 Å	31.7 °	3.953 Å	32.0 °	3.887 Å
<i>1i - 9i</i>	21.2 °	4.290 Å	22.6 °	4.295 Å	20.3 °	4.291 Å
<i>2i - 7i</i>	10.9 °	3.510 Å	10.9 °	3.505 Å	12.9 °	3.489 Å
<i>3i - 8i</i>	41.1 °	4.141 Å	42.1 °	4.123 Å	43.0 °	4.087 Å
<i>4 - 9</i>	38.5 °	3.909 Å	39.0 °	3.905 Å	36.8 °	3.785 Å
<i>5 - 7</i>	40.4 °	4.169 Å	39.5 °	4.172 Å	40.9 °	4.145 Å
<i>6 - 8</i>	80.1 °	4.962 Å	79.8 °	4.939 Å	79.5 °	4.896 Å
<i>7 - 12</i>	25.7 °	4.191 Å	26.1 °	4.225 Å	26.6 °	4.240 Å
<i>8 - 10</i>	19.3 °	4.204 Å	19.9 °	4.259 Å	20.2 °	4.277 Å
<i>9 - 11</i>	27.9 °	4.416 Å	28.7 °	4.449 Å	27.2 °	4.476 Å
<i>10 - 15</i>	27.9 °	3.945 Å	29.8 °	3.938 Å	28.8 °	3.867 Å
<i>11 - 13</i>	73.3 °	4.836 Å	72.0 °	4.832 Å	74.8 °	4.797 Å
<i>12 - 14</i>	87.9 °	5.182 Å	87.7 °	5.202 Å	88.3 °	5.159 Å

**Table S5** Pitch and Ln<sup>1</sup>-Ln<sup>2</sup> distances in L<sup>AB4</sup> complexes.

	Pr <sub>2</sub>	Nd <sub>2</sub>	Sm <sub>2</sub>
Pitch (Å)	13.4(2)	13.4(2)	13.3(3)
<i>d</i> (Ln <sup>1</sup> -Ln <sup>2</sup> ) (Å)	9.093(2)	9.148(3)	9.156(2)

**Table S6** Angles (°) and distances (Å) in the coordination polyhedra of L<sup>AB4</sup> complexes.

benzimidazole- pyridine- carboxamide moiety	$\Phi$	$\theta(\text{P1})$ $\theta(\text{P3})$	$\omega(\text{P1-P3})$ interstrand intrastrand	$\alpha$ P1-P2 P1-P3 P2-P3	$\beta$ P1 P2 P3	Distances	
						P1 - P2 P2 - P3	Ln - P1 Ln - P2 Ln - P3
Pr <sub>2</sub>	175.2	49(2)	15(3)	6.3	12.2	1.402	1.605
		54(4)	56(2)	3.1	6.1	1.783	0.210
				6.6	11.6		1.561
Nd <sub>2</sub>	175.7	48(1)	15(3)	4.7	11.5	1.434	1.622
		53(4)	56(2)	3.9	6.8	1.790	0.194
				5.3	10.6		1.586
Sm <sub>2</sub>	175.4	47(2)	14(3)	5.5	11.6	1.446	1.619
		53(5)	56(3)	3.8	6.4	1.754	0.180
				6.4	11.0		1.560
Regular trigonal tricapped prism	180	$\theta(\text{P1}) =$ $\theta(\text{P3})$	0 60	0 0 0	0 0 0	$d(\text{P1-P2}) =$ $d(\text{P2-P3})$	0
benzimidazole- pyridine- benzimidazole moiety	$\Phi$	$\theta(\text{P4})$ $\theta(\text{P6})$	$\omega(\text{P4-P6})$ interstrand intrastrand	$\alpha$ P4-P5 P4-P6 P5-P6	$\beta$ P4 P5 P6	Distances	
						P4 - P5 P5 - P6	Ln - P4 Ln - P5 Ln - P6
Pr <sub>2</sub>	178.2	51(3)	14(2)	4.4	4.6	1.655	1.681
		50(2)	53(2)	7.5	8.9	1.738	0.036
				3.5	11.6		1.695
Nd <sub>2</sub>	178.4	51(3)	13(2)	4.1	4.9	1.651	1.674
		51(3)	53(2)	7.1	8.9	1.722	0.030
				3.0	11.7		1.686
Sm <sub>2</sub>	178.1	50(3)	12(1)	4.7	3.5	1.646	1.658
		50(3)	55(2)	8.2	7.9	1.693	0.022
				3.5	11.4		1.663
Regular trigonal tricapped prism	180	$\theta(\text{P4}) =$ $\theta(\text{P6})$	0 60	0 0 0	0 0 0	$d(\text{P4-P5}) =$ $d(\text{P5-P6})$	0

**Table S7 Assignment of  $^1\text{H}$  NMR spectra of homobimetallic  $\text{HHH}[\text{Ln}_2(\text{L}^{\text{ABX}})_3]^{6+}$  complexes in  $\text{CD}_3\text{CN}$ .**

	$\text{L}^{\text{AB4}}$							$\text{L}^{\text{AB5}}$							
	$\text{La}_2$	$\text{Ce}_2$	$\text{Pr}_2$	$\text{Nd}_2$	$\text{Sm}_2$	$\text{Eu}_2$	$\text{Lu}_2$	$\text{La}_2$	$\text{Ce}_2$	$\text{Pr}_2$	$\text{Nd}_2$	$\text{Sm}_2$	$\text{Eu}_2$	$\text{Y}_2$	$\text{Lu}_2$
H1	7.67	8.77	9.53	9.05	7.81	5.57	7.73	7.71	8.97	9.79	9.19	7.88	5.39	7.75	7.75
H2	8.24	9.76	10.56	9.58	8.42	6.57	8.23	8.28	9.99	10.88	9.75	8.50	6.36	8.27	8.26
H3	8.24	10.20	11.65	10.48	8.52	4.93	8.33	8.28	10.61	12.25	10.82	8.65	4.53	8.36	8.37
H4	4.46	4.82	5.07	4.71	4.56	4.17	4.60	4.47	4.99	5.32	4.88	4.70	4.08	4.61	4.67
H4'	4.64	5.34	5.91	5.17	4.81	4.25	4.73	4.68	5.63	6.28	5.38	4.93	3.96	4.76	4.78
H5	1.49	2.46	3.05	2.18	1.67	0.71	1.51	1.51	2.66	3.34	2.33	1.72	0.47	1.52	1.51
H6	7.14	7.01	7.04	7.39	7.03	6.42	7.21	7.16	7.20	7.30	7.57	7.12	6.24	7.22	7.25
H7	6.80	6.21	5.75	6.21	6.66	7.52	6.81	6.93	6.38	5.93	6.39	6.81	7.57	6.91	6.92
H8	6.02	-3.16	-9.95	-1.91	3.94	14.17	5.52	5.97	-1.54	-7.75	-0.74	4.13	12.34	5.51	5.41
H9	3.41	2.75	2.10	2.74	3.29	4.26	3.39	3.47	2.85	2.33	2.83	3.39	4.36	3.49	3.49
H9'	3.69	2.86	2.27	2.78	3.49	4.85	3.63	3.72	3.17	2.57	3.01	3.60	4.51	3.67	3.66
H10	6.12	-5.26	-14.68	-5.39	3.38	18.65	5.60	5.91	-1.85	-8.81	-2.38	4.17	13.57	5.52	5.41
H11	7.26	6.97	6.68	6.95	7.19	7.65	7.21	7.38	7.13	6.88	7.14	7.35	7.73	7.38	7.35
H12	7.54	8.01	8.64	8.61	7.63	5.93	7.55	7.67	7.88	8.39	8.58	7.72	6.33	7.71	7.70
H13	4.26	5.93	7.56	6.48	4.26	1.43	4.36	4.26	4.94	6.18	5.73	4.70	3.05	4.49	4.51
H13'	3.66	5.59	7.56	6.04	4.68	1.02	3.96	3.94	4.68	5.81	5.33	4.48	2.95	4.24	4.29
H14	1.64	2.61	3.51	2.75	1.86	0.34	1.58	1.31	2.02	2.86	2.38	1.55	0.54	1.44	1.44
H15	6.22	8.57	11.25	9.97	6.65	0.84	6.21	7.56	8.21	9.81	9.66	7.62	4.61	7.52	7.51
H17	6.44	8.63	11.12	9.97	6.80	1.14	6.33	7.54	8.32	9.95	9.76	7.61	4.32	7.48	7.47
H18	3.88	6.02	7.79	6.39	4.52	1.12	4.08	4.28	4.94	6.12	5.35	4.31	2.95	4.35	4.37
H18'	4.38	6.05	8.08	6.68	4.81	1.51	4.47	4.52	5.12	6.24	5.87	4.58	3.24	4.57	4.58
H19	1.53	2.27	2.93	2.37	1.70	0.64	1.52	1.49	1.83	2.30	2.02	1.57	1.02	1.48	1.47
H20	7.65	8.40	9.23	8.95	7.74	5.70	7.61	7.69	8.02	8.61	8.70	7.72	6.15	7.72	7.70
H21	7.29	7.25	7.23	7.25	7.28	7.42	7.26	7.38	7.17	7.03	7.20	7.35	7.63	7.34	7.35
H22	6.82	5.93	5.07	5.87	6.58	7.97	6.76	6.88	6.15	5.41	6.02	6.68	7.66	6.82	6.80
H23	6.98	0.12	-6.03	-0.52	5.11	15.44	6.48	6.61	2.60	-2.24	1.32	5.47	11.46	6.31	6.19
H24	3.35	3.53	3.95	3.66	3.46	2.83	3.33	3.37	3.77	4.28	3.82	3.45	2.60	3.33	3.35
H24'	3.25	2.86	2.81	3.17	3.22	3.24	3.33	3.27	3.08	3.11	3.33	3.28	3.04	3.33	3.35
H25	0.92	1.04	1.26	1.11	0.99	0.77	1.05	0.87	1.20	1.51	1.23	1.01	0.57	1.02	1.06
H26	2.81	1.33	0.47	1.72	2.49	4.03	2.59	2.77	1.58	0.87	1.93	2.53	3.72	2.63	2.60
H26'	2.88	2.33	2.05	2.44	2.64	3.26	2.72	2.84	2.60	2.45	2.65	2.69	2.95	2.70	2.72
H27	0.71	-2.33	-4.57	-1.91	0.06	3.64	0.62	0.72	-2.15	-4.31	-1.75	0.10	3.49	0.64	0.63
H28	3.51	4.12	4.79	4.15	3.56	2.75	3.40								
H28'	3.42	4.12	4.79	4.15	3.63	2.53	3.40								
H29	1.18	1.73	2.34	1.87	1.28	0.27	1.10								

**Table S8 Percentages of HHH isomer in  $[\text{Ln}_2(\text{L}^{\text{ABX}})_3]^{6+}$  complexes in  $\text{CD}_3\text{CN}$  solution at room temperature. Table S3 Percentages of HHH isomer in  $[\text{Ln}_2(\text{L}^{\text{ABX}})_3]^{6+}$  complexes in  $\text{CD}_3\text{CN}$  solution at room temperature.**

Ln	$\text{L}^{\text{AB1}}$	$\text{L}^{\text{AB2}}$	$\text{L}^{\text{AB3}}$	$\text{L}^{\text{AB4}}$	$\text{L}^{\text{AB5}}$
La	73	20	79	96	60
Ce	69	13	79	95	53
Pr	67	12	87	93	53
Nd	69	15	87	95	55
Sm	65	11	85	95	56
Eu	69	8	85	96	54
Y	63	8	82		61
Lu	68	6	86	94	60

**Table S9** Assignment of  $^1\text{H}$  NMR spectra of heterobimetallic  $\text{HHH}[\text{Ln}^1\text{Ln}^2(\text{L}^{\text{AB4}})_3]^{6+}$  complexes in  $\text{CD}_3\text{CN}$  solution.

	LaCe	LaPr	LaNd	LaEu	LaLu	CeLu	PrLu	NdLu	EuLu
H1	9.20	10.33	9.51	5.01	7.73	7.31	6.94	7.28	8.26
H2	10.13	11.25	9.99	6.07	8.24	7.87	7.55	7.84	8.71
H3	10.80	12.74	11.12	4.16	8.36	7.78	7.27	7.73	9.08
H4	5.59	6.06	5.28	3.49	4.52	3.98	3.50	3.96	5.27
H4'	5.80	6.90	5.77	3.49	4.72	4.16	3.66	4.12	5.45
H5	2.60	3.75	2.58	0.15	1.46	1.09	0.76	1.07	1.96
H6	7.46	7.94	7.88	5.68	7.05	6.54	6.08	6.53	7.83
H7	6.47	6.31	6.52	7.05	6.72	6.39	6.08	6.36	7.23
H8	-0.15	-3.95	1.45	10.61	5.64	2.72	0.16	2.47	9.22
H9	2.87	2.54	2.99	3.91	3.38	3.19	3.01	3.11	3.79
H9'	3.53	3.44	3.53	3.91	3.67	2.96	2.34	2.88	4.58
H10	3.19	1.25	3.80	8.69	6.03	-2.56	-10.41	-3.17	16.17
H11	7.04	6.81	7.06	7.53	7.27	7.18	7.12	7.15	7.41
H12	7.19	6.92	7.26	7.97	7.60	8.39	9.27	8.90	5.58
H13	3.85	3.60	3.95	4.68	4.31	6.21	8.18	6.76	1.07
H13'	3.19	2.88	3.28	4.08	3.67	6.04	8.18	6.36	0.62
H14	1.39	1.23	1.46	1.92	1.69	2.84	3.86	2.94	0.12
H15	5.71	5.39	5.82	6.67	6.21	9.01	11.96	10.29	0.39
H17	6.06	5.83	6.15	6.78	6.44	8.95	11.62	10.17	0.84
H18	3.50	3.30	3.59	4.25	3.89	6.36	8.26	6.63	0.74
H18'	4.08	3.87	4.15	4.73	4.30	6.36	8.61	6.90	1.18
H19	1.30	1.14	1.35	1.77	1.54	2.52	3.31	2.55	0.41
H20	7.40	7.24	7.46	7.89	7.66	8.65	9.65	9.14	5.45
H21	7.07	6.93	7.12	7.49	7.29	7.49	7.62	7.44	7.21
H22	6.54	6.37	6.61	7.06	6.82	6.25	5.67	6.12	7.72
H23	6.53	6.24	6.61	7.37	6.99	0.76	-4.91	0.02	15.03
H24	3.93	4.70	4.08	2.34	3.33	2.96	2.62	2.94	3.82
H24'	3.24	3.53	3.52	2.77	3.33	2.96	2.64	2.94	3.82
H25	1.35	1.87	1.45	0.34	1.06	0.76	0.49	0.73	1.44
H26	1.70	1.20	2.16	3.57	2.57	2.11	1.71	2.08	3.18
H26'	2.77	2.90	2.94	2.77	2.71	2.33	2.00	2.30	3.18
H27	-1.98	-3.87	-1.53	3.25	0.61	0.24	-0.08	0.22	1.10
H28	3.25	3.30	3.32	3.67	3.49	4.27	5.05	4.27	2.58
H28'	3.25	3.15	3.23	3.58	3.42	4.27	5.05	4.27	2.37
H29	1.02	0.92	1.05	1.32	1.18	1.89	2.57	1.98	0.12

**Table S10** Assignment of  $^1\text{H}$  NMR spectra of heterobimetallic  $\text{HHH}[\text{Ln}^1\text{Ln}^2(\text{L}^{\text{AB5}})_3]^{6+}$  complexes in  $\text{CD}_3\text{CN}$  solution.

	LaCe	LaPr	LaNd	LaEu	LaLu	CeLu	PrLu	NdLu	SmLu	EuLu
H1	9.19	10.26	9.49	5.13	7.76	7.54	7.27	7.46	7.70	8.03
H2	10.21	11.35	10.05	6.09	8.27	8.10	7.85	8.00	8.22	8.50
H3	11.00	13.04	11.29	4.08	8.38	8.10	7.74	8.01	8.31	8.75
H4	5.25	5.87	5.21	3.58	4.51	4.23	3.89	4.17	4.51	4.98
H4'	5.94	6.89	5.76	3.58	4.76	4.47	4.12	4.38	4.70	5.16
H5	2.90	3.80	2.61	0.17	1.47	1.27	1.04	1.22	1.44	1.75
H6	7.42	7.83	7.85	5.74	7.06	6.79	6.45	6.74	7.09	7.60
H7	6.53	6.31	6.60	7.24	6.84	6.62	6.35	6.57	6.83	7.21
H8	-0.33	-4.31	1.30	10.44	5.55	4.02	2.25	3.58	5.15	7.34
H9	2.92	2.57	3.03	3.97	3.45	3.31	3.14	3.21	3.46	3.87
H9'	3.56	3.42	3.54	3.97	3.70	3.31	2.84	3.14	3.59	4.20
H10	2.70	1.01	3.57	8.62	5.91	1.36	-4.05	-0.12	4.64	10.91
H11	7.15	6.93	7.18	7.65	7.40	7.36	7.33	7.34	7.39	7.47
H12	7.31	7.03	7.38	8.07	7.70	8.25	9.02	8.87	7.77	5.97
H13	3.85	3.55	3.94	4.67	4.29	5.32	6.78	6.02	4.64	2.63
H13'	3.49	3.17	3.54	4.30	3.88	5.06	6.50	5.61	4.37	2.68
H14	1.03	0.86	1.13	1.65	1.43	2.31	3.25	2.57	1.64	0.36
H15	7.00	6.66	7.11	7.99	7.50	8.66	10.55	9.97	7.68	4.16
H17	7.14	6.90	7.23	7.88	7.53	8.65	10.46	9.97	7.67	4.06
H18	3.96	3.72	3.99	4.60	4.25	5.26	6.66	5.71	4.52	2.65
H18'	4.23	4.03	4.30	4.82	4.53	5.44	6.71	6.08	4.64	2.90
H19	1.24	1.09	1.30	1.73	1.51	2.09	2.71	2.21	1.60	0.80
H20	7.47	7.31	7.53	7.94	7.72	8.30	9.06	8.90	7.77	5.92
H21	7.17	7.02	7.21	7.58	7.39	7.42	7.45	7.39	7.39	7.42
H22	6.61	6.42	6.66	7.13	6.88	6.48	5.95	6.28	6.74	7.42
H23	6.16	5.87	6.27	7.04	6.68	3.22	-1.21	1.87	5.60	11.00
H24	4.02	4.82	4.15	2.31	3.33	3.11	2.89	3.06	3.29	3.58
H24'	3.28	3.59	3.60	2.77	3.38	3.16	2.92	3.09	3.32	3.62
H25	1.38	1.96	1.47	0.26	1.05	0.90	0.70	0.84	1.02	1.26
H26	1.83	1.40	2.20	3.42	2.55	2.30	2.04	2.26	2.52	2.85
H26'	2.87	3.06	2.95	2.63	2.70	2.51	2.28	2.44	2.67	2.95
H27	-1.95	-3.86	-1.52	3.24	0.62	0.43	0.20	0.36	0.58	0.86



**Table S11** Composition in % of L<sup>AB4</sup> complexes in solution. For hetero pairs, the first mentioned Ln is in the bpb cavity.

Ln <sup>1</sup>	Ln <sup>2</sup>	$\Delta r(\text{Ln}^1-\text{Ln}^2)$ / pm	Ln <sup>1</sup> Ln <sup>2</sup>		Ln <sup>2</sup> Ln <sup>1</sup>		Total hetero	Ln <sup>1</sup> <sub>2</sub>		Ln <sup>2</sup> <sub>2</sub>	
			HHH	HHT	HHT	HHH		HHH	HHT	HHH	HHT
La	Ce	2	34			14	48	10			41
La	Pr	3.7	41			13	54	24			23
La	Nd	5.3	54			7	61	16			23
Eu	Lu	8.8	62				62	14			24
La	Eu	9.6	75				75	14			12
Nd	Lu	13.1	68				68	16			16
Pr	Lu	14.7	82				82	13			5
Ce	Lu	16.4	76				76	14			11
La	Lu	18.4	79				79	11			9

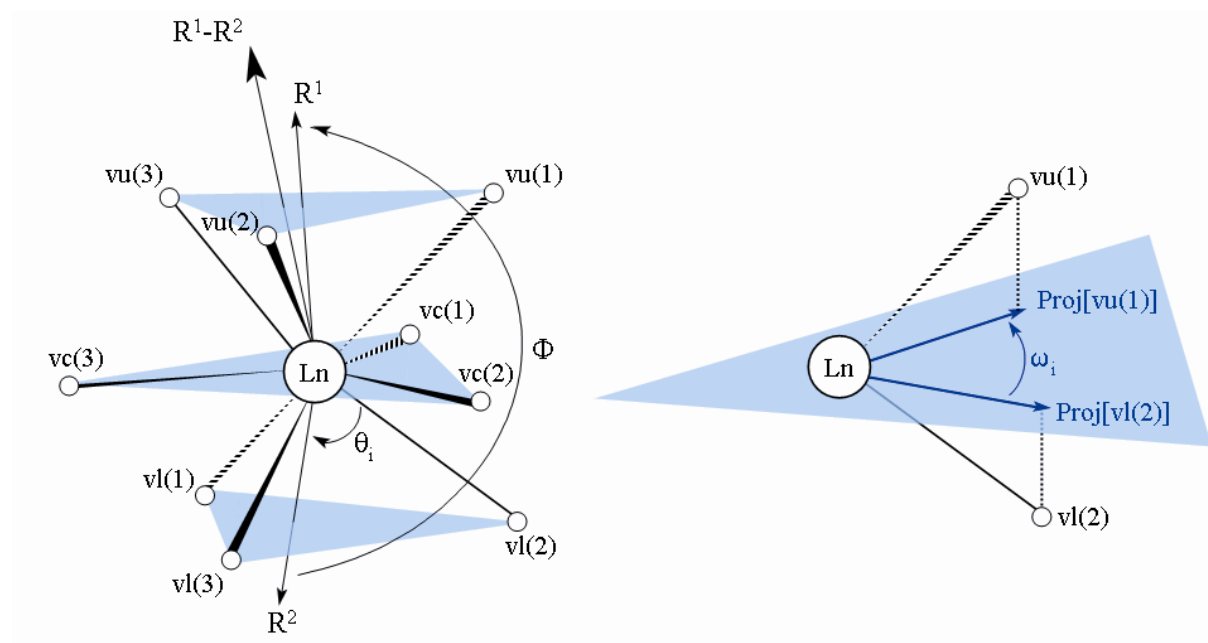
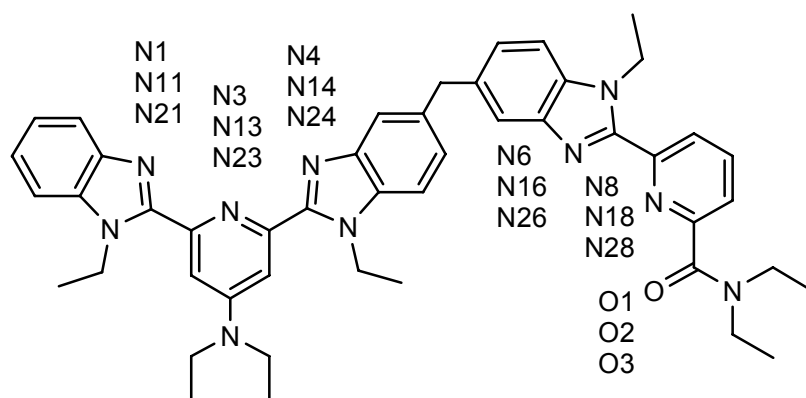
**Table S12** Composition in % of L<sup>ABS</sup> complexes in solution. For hetero pairs, the first mentioned Ln is in the bpb cavity.

Ln <sup>1</sup>	Ln <sup>2</sup>	$\Delta r(\text{Ln}^1-\text{Ln}^2)$ / pm	Ln <sup>1</sup> Ln <sup>2</sup>		Ln <sup>2</sup> Ln <sup>1</sup>		Total hetero	Ln <sup>1</sup> <sub>2</sub>		Ln <sup>2</sup> <sub>2</sub>	
			HHH	HHT	HHT	HHH		HHH	HHT	HHH	HHT
La	Ce	2	24	12		9	45	15	12	16	12
La	Pr	3.7	29	9		7	45	17	13	13	12
La	Nd	5.3	39	9		4	52	14	10	13	12
La	Sm	8.4	47	9			56	14	11	9	10
Eu	Lu	8.8	39	13			52	15	11	11	11
La	Eu	9.6	48	9			57	10	9	10	15
Sm	Lu	10	57	9			66	10	8	7	9
Nd	Lu	13.1	71	9			80	6	6	5	4
Pr	Lu	14.7	63	7			70	9	9	8	5
Ce	Lu	16.4	79	13			92	4	1	2	1
La	Lu	18.4	77	15			92	3	1	1	1

**Table S13** Total percentages of heterobimetallic  $[\text{Ln}^1\text{Ln}^2(\text{L}^{\text{ABX}})_3]^{6+}$  helicates in  $\text{CD}_3\text{CN}$ .

$\text{Ln}^1\text{Ln}^2$	$\Delta r(\text{Ln}^1-\text{Ln}^2)$ / pm	$\text{L}^{\text{AB1}}$	$\text{L}^{\text{AB2}}$	$\text{L}^{\text{AB3}}$	$\text{L}^{\text{AB4}}$	$\text{L}^{\text{AB5}}$
LaCe	2	42		49	48	45
LaPr	3.7	41		48	54	45
LaNd	5.3	51		57	61	52
LaSm	8.4					56
EuLu	8.8	73	21	75	62	52
LaEu	9.6	64		63	75	57
SmLu	10	84		65		66
NdLu	13.1	90	27	70	68	80
PrLu	14.7	95	51	88	82	70
CeLu	16.4	92	60	80	76	92
LaLu	18.4	96	65	87	79	92

Chart S1 Numbering of ligating atoms in  $L^{AB4}$ .



**Figure S1 Definition of angles in the coordination polyhedron.** The vectors  $vu(n)$ ,  $vc(n)$  and  $vl(n)$  are defined by the Ln atom and the coordinating atoms of ligand  $n$  ( $=1, 2$  or  $3$ ) in the upper, capping and lower positions of the tricapped trigonal prism. The angles between them are denoted  $\theta_i$ . The vectors  $R^1$  and  $R^2$  go through the midpoints of the triangles defined by the upper and lower coordinating atoms, respectively.  $\Phi$  is the angle between them. The projection vectors  $\text{Proj}[vm(n)]$  are the projections of vectors  $vm(n)$  onto a plane perpendicular to  $R^1-R^2$ . The angles between these vectors are denoted  $\omega_i$ . Planes in the coordination polyhedra are defined as follows: P1: O1, O2, O3; P2: N8, N18, N28; P3: N6, N16, N26; P4: N4, N14, N24; P5: N3, N13, N23; P6: N1, N11, N21. The angles  $\alpha$  and  $\beta$  are the angles between planes and the angles between a plane and a plane normal to the molecular (Ln-Ln) axis, respectively.