

Supplementary Information

for

Pt(II) N-confused porphyrin: An expanded pyrrole that affords stable π -anion

Dong-Hoon Won, Motoki Toganoh, Hidemitsu Uno, and Hiroyuki Furuta*

Table of Contents

Figure S1. ^1H NMR spectrum of 1-Pt in CDCl_3 -----	S2
Figure S2. ^1H NMR spectrum of 2-Pt in CDCl_3 -----	S2
Figure S3. ^1H NMR spectrum of 3 in CDCl_3 -----	S3
Figure S4. ^1H NMR spectrum of 3-Pt in CDCl_3 -----	S3
Figure S5. pH titration curves for 1-Pt and 1-Cu -----	S4
Figure S6. Packing diagram of $\text{Bu}_4\text{N}^+\cdot(\text{1-Pt}^-)$ -----	S5
Figure S7. Calculated charge density for 1-Pt and 1-Pt⁻ -----	S6

Figure S1. ^1H NMR spectrum of **1-Pt** in CDCl_3 .

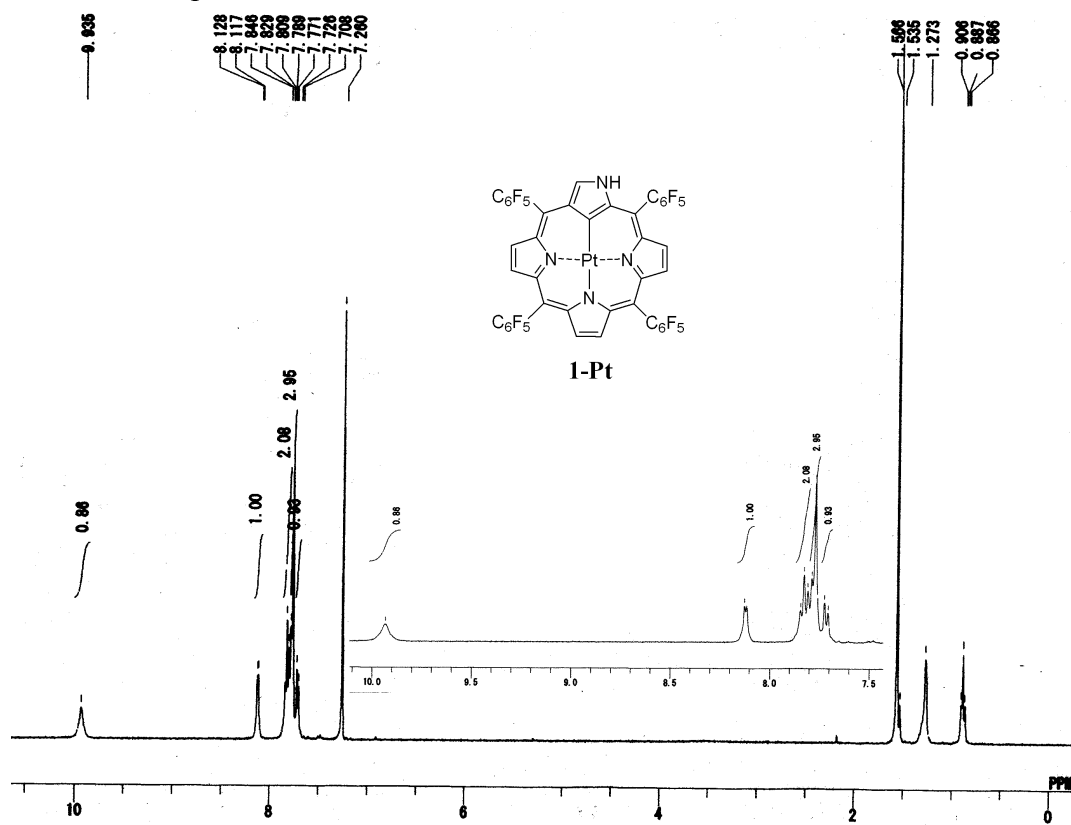


Figure S2. ^1H NMR spectrum of **2-Pt** in CDCl_3 .

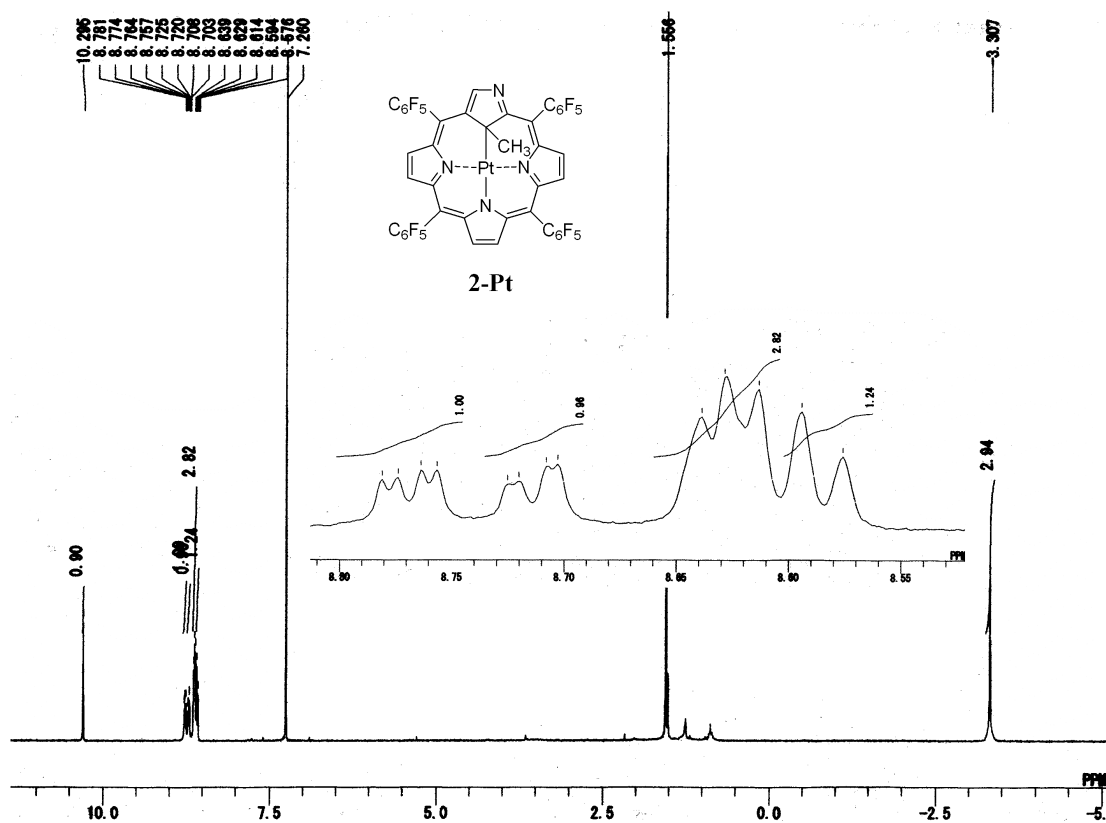


Figure S3. ^1H NMR spectrum of **3** in CDCl_3 .

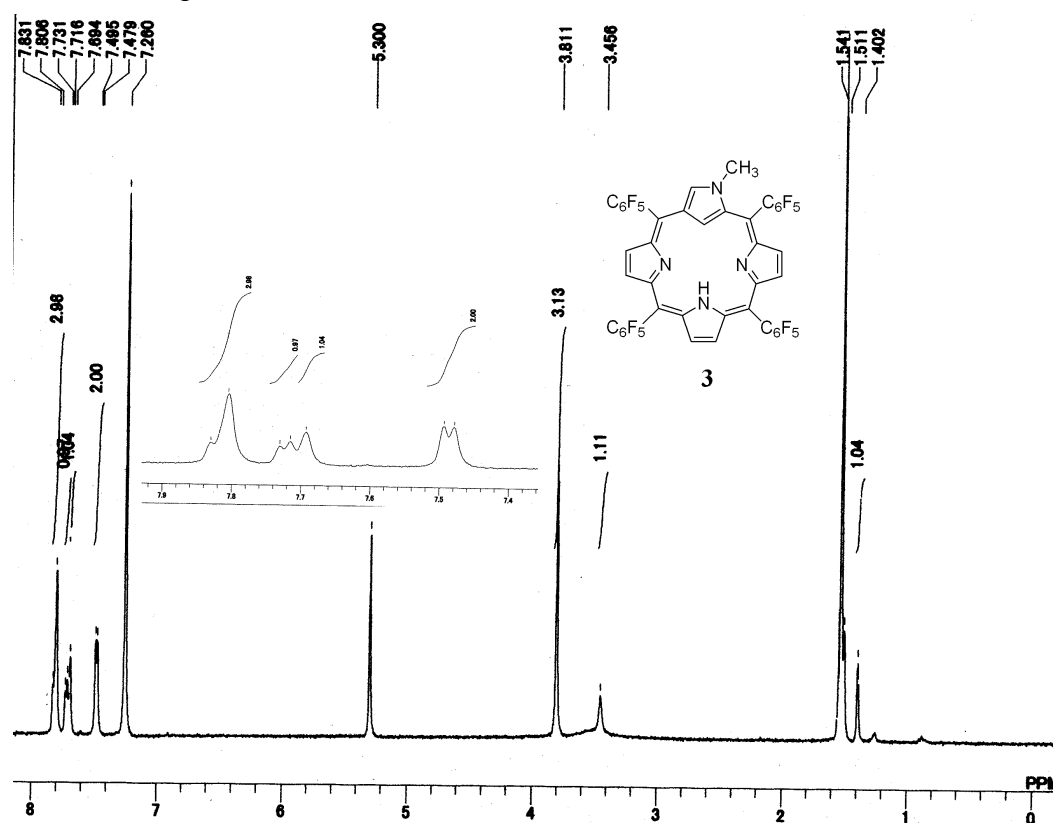


Figure S4. ^1H NMR spectrum of **3-Pt** in CDCl_3 .

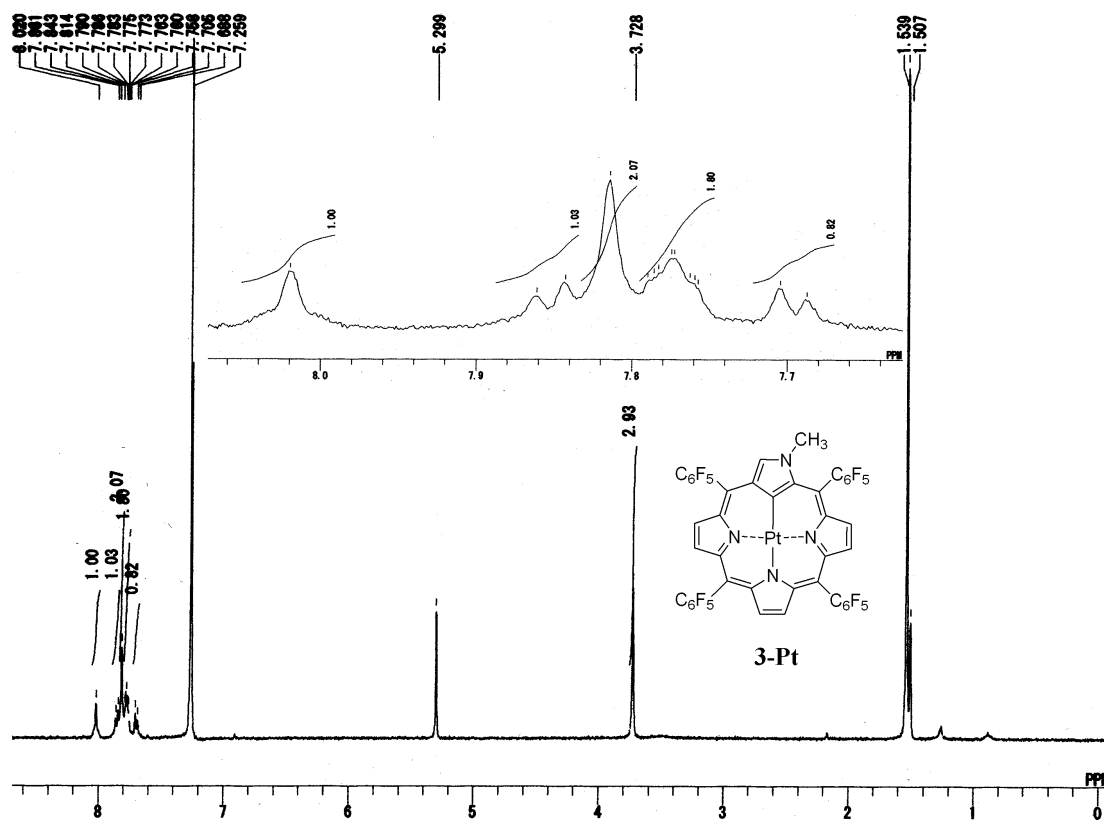


Figure S5. pH titration curves for **1-Pt** and **1-Cu**. Titration was performed in 3.5% sodium dodecyl sulfate by adjusting the pH with HCl-NaOH; [**1-Pt**] = $\sim 2 \times 10^{-5}$ M, [**1-Cu**] = $\sim 1 \times 10^{-5}$ M.

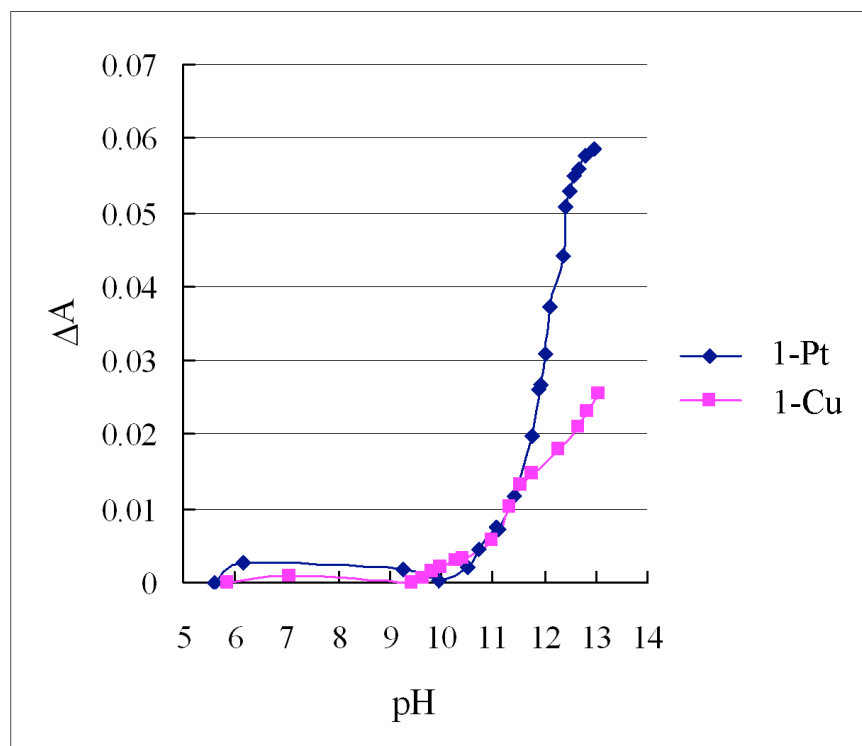


Figure S6. Packing diagram of $\text{Bu}_4\text{N}^+\cdot(1\text{-Pt}^-)$.

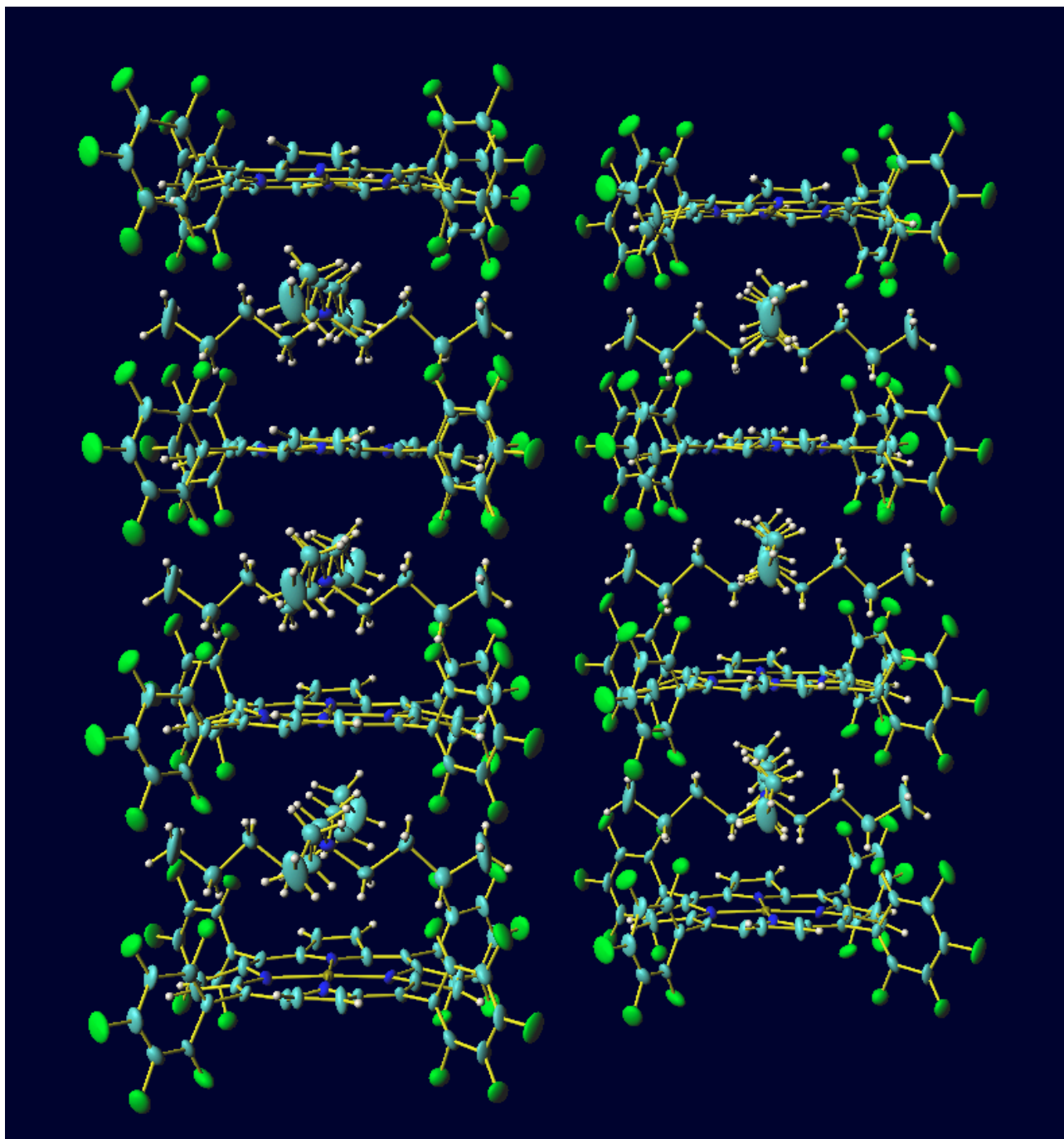


Figure S7. Calculated charge density for a) 1-Pt, b) 1-Pt⁻, and c) their difference.

