

## Supporting Information

# Quantification of Mutual Trans Influence of Ligands in Pd(II) Complexes: A Combined Approach Using Isodesmic Reactions and AIM Analysis

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### (1) Table S1. SCF energies E (in a.u.) of all geometries at MPWB1K- COSMO/BS1 level.

$[\text{PdCl}_3\text{H}_2\text{O}]^-$ , E = -1584.09541137	$[\text{PdCl}_3\text{CO}]^-$ , E = -1620.96688807
$[\text{PdCl}_3\text{Py}]^-$ , E = -1755.88995161	$[\text{PdCl}_3\text{NH}_3]^-$ , E = -1564.25453667
$[\text{PdCl}_3\text{SMe}_2]^-$ , E = -1985.71128760	$[\text{PdCl}_3\text{C}_2\text{H}_4]^-$ , E = -1586.24498739
$[\text{PdCl}_3\text{AsH}_3]^-$ , E = -3743.17116718	$[\text{PdCl}_3\text{PH}_3]^-$ , E = -1850.87524624
$[\text{PdCl}_3\text{AsMe}_3]^-$ , E = -3861.08693813	$[\text{PdCl}_3\text{PMe}_3]^-$ , E = -1968.79627957
$[\text{PdCl}_3\text{PEt}_3]^-$ , E = -2086.66529766	$[\text{PdCl}_3\text{ONO}]^{2-}$ , E = -1712.87492674
$[\text{PdCl}_3\text{F}]^{2-}$ , E = -1607.66446611	$[\text{PdCl}_4]^{2-}$ , E = -1968.12116931
$[\text{PdCl}_3\text{Br}]^{2-}$ , E = -4079.61490270	$[\text{PdCl}_3\text{N}_3]^{2-}$ , E = -1671.96383042
$[\text{PdCl}_3\text{NO}_2]^{2-}$ , E = -1712.88962942	$[\text{PdCl}_3\text{OH}]^{2-}$ , E = -1583.62846621
$[\text{PdCl}_3\text{CN}]^{2-}$ , E = -1600.65643902	$[\text{PdCl}_3\text{C}_6\text{H}_5]^{2-}$ , E = -1739.37234667
$[\text{PdCl}_3\text{H}]^{2-}$ , E = -1508.42417198	$[\text{PdCl}_3\text{CH}_3]^{2-}$ , E = -1547.71228455
$[\text{PdCl}_3\text{SiH}_3]^{2-}$ , E = -1799.15401533	$[\text{PdCl}_2\text{H}_2\text{O}]$ , E = -1123.60320367
$[\text{PdCl}_2\text{CO}]$ , E = -1160.48095317	$[\text{PdCl}_2\text{Py}]$ , E = -1295.41321093
$[\text{PdCl}_2\text{NH}_3]$ , E = -1103.77837583	$[\text{PdCl}_2\text{SMe}_2]$ , E = -1525.24143631

[PdCl <sub>2</sub> C <sub>2</sub> H <sub>4</sub> ], E = -1125.77407607	[PdCl <sub>2</sub> AsH <sub>3</sub> ], E = -3282.70757895
[PdCl <sub>2</sub> PH <sub>3</sub> ], E = -1390.41189604	[PdCl <sub>2</sub> AsMe <sub>3</sub> ], E = -3400.63133922
[PdCl <sub>2</sub> PMe <sub>3</sub> ], E = -1508.34201363	[PdCl <sub>2</sub> PEt <sub>3</sub> ], E = -1626.21246997
[PdCl <sub>2</sub> ONO], E = -1252.39792104	[PdCl <sub>2</sub> F], E = -1147.18757806
[PdCl <sub>3</sub> ] <sup>-</sup> , E = -1507.65221781	[PdCl <sub>2</sub> Br] <sup>-</sup> , E = -3619.14927801
[PdCl <sub>2</sub> N <sub>3</sub> ] <sup>-</sup> , E = -1211.49760964	[PdCl <sub>2</sub> NO <sub>2</sub> ] <sup>-</sup> , E = -1252.42289625
[PdCl <sub>2</sub> OH] <sup>-</sup> , E = -1123.16457851	[PdCl <sub>2</sub> CN] <sup>-</sup> , E = -1140.19420140
[PdCl <sub>2</sub> C <sub>6</sub> H <sub>5</sub> ] <sup>-</sup> , E = -1278.93243509	[PdCl <sub>2</sub> H] <sup>-</sup> , E = -1047.98275444
[PdCl <sub>2</sub> CH <sub>3</sub> ] <sup>-</sup> , E = -1087.27494392	[PdCl <sub>2</sub> SiH <sub>3</sub> ] <sup>-</sup> , E = -1338.71945016
[PdCl <sub>2</sub> ], E = -1047.15180156	[PdCl <sub>2</sub> COBr] <sup>-</sup> , E = -3732.45986273
[PdCl <sub>2</sub> H <sub>2</sub> OBr] <sup>-</sup> , E = -3695.58911374	[PdCl <sub>2</sub> PH <sub>3</sub> CN] <sup>-</sup> , E = -1483.39990229
[PdCl <sub>2</sub> NH <sub>3</sub> Br] <sup>-</sup> , E = -3675.74712684	[PdCl <sub>2</sub> PH <sub>3</sub> NO <sub>2</sub> ] <sup>-</sup> , E = -1595.63582456
[PdCl <sub>2</sub> AsH <sub>3</sub> CN] <sup>-</sup> , E = -3375.69712860	[PdCl <sub>2</sub> FPH <sub>3</sub> ] <sup>-</sup> , E = -1490.41814350
[PdCl <sub>2</sub> NH <sub>3</sub> NO <sub>2</sub> ] <sup>-</sup> , E = -1309.01998923	[PdCl <sub>2</sub> PyAsH <sub>3</sub> ], E = -3530.93468985
[PdCl <sub>2</sub> COPy], E = -1408.72946711	[PdCl <sub>2</sub> ONON <sub>3</sub> ] <sup>2-</sup> , E = -1416.71742767
[PdCl <sub>2</sub> H <sub>2</sub> ONH <sub>3</sub> ], E = -1180.22411073	[PdCl <sub>2</sub> COCN] <sup>-</sup> , E = -1253.49677671
[PdCl <sub>2</sub> HCH <sub>3</sub> ] <sup>2-</sup> , E = -1087.96363617	[PdCl <sub>2</sub> BrOH] <sup>2-</sup> , E = -3695.12082312
[PdCl <sub>2</sub> OHPH <sub>3</sub> ] <sup>-</sup> , E = -1466.37844263	[PdCl <sub>2</sub> NH <sub>3</sub> SMe <sub>2</sub> ] <sup>-</sup> , E = -1581.84046357
[PdCl <sub>2</sub> OHCN] <sup>2-</sup> , E = -1216.15979452	[PdCl <sub>2</sub> (H <sub>2</sub> O) <sub>2</sub> ], E = -1200.06086964
[PdCl <sub>2</sub> H <sub>2</sub> ON <sub>3</sub> ] <sup>-</sup> , E = -1287.93917013	[PdCl <sub>2</sub> H <sub>2</sub> OCH <sub>3</sub> ] <sup>-</sup> , E = -1163.69600490
[PdCl <sub>2</sub> PyCH <sub>3</sub> ] <sup>-</sup> , E = -1335.47639721	[PdCl <sub>2</sub> PH <sub>3</sub> SiH <sub>3</sub> ] <sup>-</sup> , E = -1681.89359033
[PdCl <sub>2</sub> NH <sub>3</sub> CH <sub>3</sub> ] <sup>-</sup> , E = -1143.83807649	[PdCl <sub>2</sub> CNCH <sub>3</sub> ] <sup>2-</sup> , E = -1180.22770329
[PdCl <sub>2</sub> AsH <sub>3</sub> SiH <sub>3</sub> ] <sup>-</sup> , E = -3574.19212857	[PdCl <sub>2</sub> COH] <sup>-</sup> , E = -1161.26928203
[PdCl <sub>2</sub> OHH] <sup>2-</sup> , E = -1123.91446353	[PdCl <sub>2</sub> AsH <sub>3</sub> C <sub>6</sub> H <sub>5</sub> ] <sup>-</sup> , E = -3514.41240989
[PdCl <sub>2</sub> NO <sub>2</sub> H] <sup>2-</sup> , E = -1253.18159795	[PdCl <sub>2</sub> H <sub>2</sub> OSiH <sub>3</sub> ] <sup>-</sup> , E = -1415.13630745
[PdCl <sub>2</sub> PH <sub>3</sub> C <sub>6</sub> H <sub>5</sub> ] <sup>-</sup> , E = -1622.11389869	[PdCl <sub>2</sub> SMe <sub>2</sub> CH <sub>3</sub> ] <sup>-</sup> , E = -1565.29894798
[PdCl <sub>2</sub> FSiH <sub>3</sub> ] <sup>2-</sup> , E = -1438.69168467	[PdCl <sub>2</sub> SMe <sub>2</sub> SiH <sub>3</sub> ] <sup>-</sup> , E = -1816.73890905
[PdCl <sub>2</sub> COSiH <sub>3</sub> ] <sup>-</sup> , E = -1451.99794713	[PdCl <sub>2</sub> (AsMe <sub>3</sub> ) <sub>2</sub> ], E = -5754.03840377
[PdCl <sub>2</sub> FH] <sup>2-</sup> , E = -1147.96429472	[PdCl <sub>2</sub> (PMe <sub>3</sub> ) <sub>2</sub> ], E = -1969.45515665
[PdCl <sub>2</sub> (AsMe <sub>3</sub> )(PMe <sub>3</sub> )], E = -3861.74682423	[PdCl <sub>2</sub> CN(PMe <sub>3</sub> )] <sup>-</sup> , E = -1601.31826678
[PdCl <sub>2</sub> (AsMe <sub>3</sub> )(PEt <sub>3</sub> )], E = -3979.61623989	[PdCl <sub>2</sub> H <sub>2</sub> O(PMe <sub>3</sub> )], E = -1584.77304790
[PdCl <sub>2</sub> H <sub>2</sub> O(AsMe <sub>3</sub> )], E = -3477.06264116	[PdCl <sub>2</sub> (AsMe <sub>3</sub> )CH <sub>3</sub> ] <sup>-</sup> , E = -3440.65752551
[PdCl <sub>2</sub> (PMe <sub>3</sub> )SiH <sub>3</sub> ] <sup>-</sup> , E = -1799.80233764	[PdCl <sub>2</sub> C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub> ] <sup>2-</sup> , E = -1318.91188476
[PdCl <sub>2</sub> OH(PMe <sub>3</sub> )] <sup>-</sup> , E = -1584.29663414	[PdCl <sub>2</sub> CH <sub>3</sub> SiH <sub>3</sub> ] <sup>2-</sup> , E = -1378.68824353
[PdCl <sub>2</sub> C <sub>6</sub> H <sub>5</sub> SiH <sub>3</sub> ] <sup>2-</sup> , E = -1570.35138503	[PdCl <sub>2</sub> (SiH <sub>3</sub> ) <sub>2</sub> ] <sup>2-</sup> , E = -1630.12830064
[PdCl <sub>2</sub> (CH <sub>3</sub> ) <sub>2</sub> ] <sup>2-</sup> , E = -1127.24685742	[PdCl <sub>2</sub> HSiH <sub>3</sub> ] <sup>2-</sup> , E = -1339.40208335

(2) Table S2. Comparison of E<sub>1</sub> values calculated at two different levels of theory.

Trans Ligand	B3LYP-COSMO/ BS1	MPWB1K-COSMO/ BS1
H <sub>2</sub> O	7.84	5.15
CO	12.08	9.09
NH <sub>3</sub>	18.17	15.22
C <sub>2</sub> H <sub>4</sub>	21.68	18.51
PH <sub>3</sub>	25.92	23.26
F <sup>-</sup>	17.79	14.76
Cl <sup>-</sup>	23.03	19.74
Br <sup>-</sup>	25.48	21.83

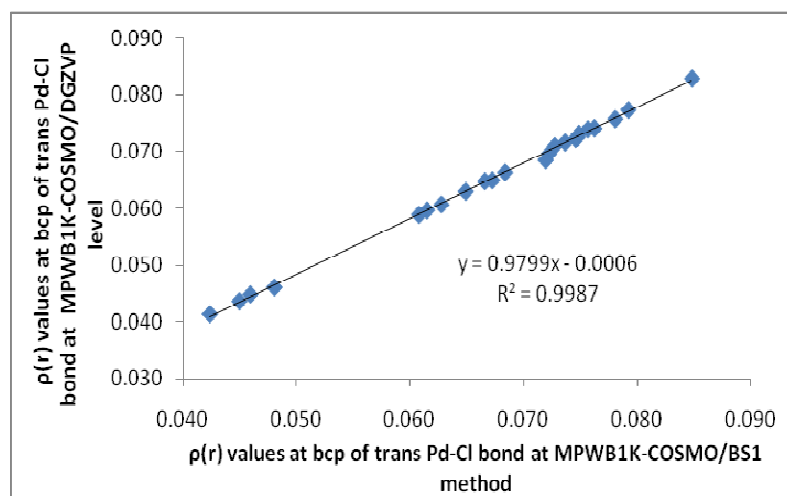
CH <sub>3</sub> <sup>-</sup>	42.29	39.58
SiH <sub>3</sub> <sup>-</sup>	43.75	41.32

**(3) Table S3. Comparison of E<sub>1</sub> values with water and chloroform as solvent obtained at MPWB1K-COSMO/BS1 method.**

Trans Ligand	Water as solvent	Chloroform as solvent
H <sub>2</sub> O	5.15	6.73
CO	9.09	10.22
NH <sub>3</sub>	15.22	17.32
C <sub>2</sub> H <sub>4</sub>	18.51	20.35
PH <sub>3</sub>	23.26	25.66
F <sup>-</sup>	14.76	33.57
Cl <sup>-</sup>	19.74	37.83
Br <sup>-</sup>	21.83	39.71
CH <sub>3</sub> <sup>-</sup>	39.58	58.91
SiH <sub>3</sub> <sup>-</sup>	41.32	59.87

**(4) Table S4. Comparison of ρ(r) values at bcp of trans Pd-X bond in [PdCl<sub>3</sub>X]<sup>n-</sup> obtained at two different levels of theory.**

Trans Ligand	MPWB1K-COSMO/BS1	MPWB1K-COSMO/DGDZVP
H <sub>2</sub> O	0.08481	0.08086
CO	0.07921	0.07609
NH <sub>3</sub>	0.07456	0.07063
C <sub>2</sub> H <sub>4</sub>	0.07473	0.07002
PH <sub>3</sub>	0.06657	0.06316
F <sup>-</sup>	0.07564	0.07177
Cl <sup>-</sup>	0.07484	0.07099
Br <sup>-</sup>	0.07272	0.06848
CH <sub>3</sub> <sup>-</sup>	0.04497	0.04096
SiH <sub>3</sub> <sup>-</sup>	0.04237	0.03908



**Figure S1.** Correlation between  $\rho(r)$  values calculated at MPWB1K-COSMO/BS1 and single point energy calculations at MPWB1K-COSMO/DGDZVP methods.

**(5) Table S5.  $E_{2\text{pred}}$ , actual and predicted dissociation energy (in kcal/mol) of the bonds Pd-X and Pd-Y bonds in  $[\text{Pd}^{\text{II}}\text{Cl}_2\text{XY}]^{n-}$  of 53 complexes.**

The equations used for predictions are,

$$E_{2\text{pred}} = -2709.906\rho_{1X} - 2217.285\rho_{1Y} + 19763.654\rho_{1X}\rho_{1Y} + 274.334$$

$$D_{2X\text{pred}} = D_{1X} + 1924.836\rho_{1X} + 2217.285\rho_{1Y} - 19763.654\rho_{1X}\rho_{1Y} - 198.584$$

$$D_{2Y\text{pred}} = D_{1Y} + 2709.906\rho_{1X} + 1431.93\rho_{1Y} - 19763.654\rho_{1X}\rho_{1Y} - 198.584$$

Sl. No.	Ligands in the		$E_{2\text{pred}}$	Calculated	Predicted	Calculated	Predicted
	trans X	positions Y		$D_{2X}$	$D_{2X}$	$D_{2Y}$	$D_{2Y}$
1	H <sub>2</sub> O	Br <sup>-</sup>	5.15	21.21	27.33	42.81	41.72
2	NH <sub>3</sub>	Br <sup>-</sup>	18.20	39.07	40.90	32.04	28.67
3	PH <sub>3</sub>	CN <sup>-</sup>	35.44	30.76	29.68	43.30	42.84
4	NO <sub>2</sub> <sup>-</sup>	PH <sub>3</sub>	26.45	30.02	31.31	35.29	38.65
5	NH <sub>3</sub>	NO <sub>2</sub> <sup>-</sup>	18.79	38.60	40.30	41.12	38.95
6	Py	AsH <sub>3</sub>	19.20	31.64	35.02	40.57	41.85
7	ONO <sup>-</sup>	N <sub>3</sub> <sup>-</sup>	14.10	27.44	29.63	40.22	40.20
8	CO	CN <sup>-</sup>	17.39	22.94	26.71	60.76	60.89
9	OH <sup>-</sup>	PH <sub>3</sub>	33.07	42.37	38.39	35.88	32.03
10	Br <sup>-</sup>	OH <sup>-</sup>	24.86	24.19	22.03	45.50	46.58
11	OH <sup>-</sup>	CN <sup>-</sup>	34.57	41.77	36.89	47.83	43.70
12	H <sub>2</sub> O	H <sub>2</sub> O	-1.39	32.40	33.87	32.40	33.85
13	OH <sup>-</sup>	PMe <sub>3</sub>	37.60	34.88	33.87	47.84	45.24
14	CN <sup>-</sup>	PMe <sub>3</sub>	41.02	35.93	37.27	42.82	41.81
15	Py	CH <sub>3</sub> <sup>-</sup>	35.82	15.54	18.40	93.07	95.72
16	PH <sub>3</sub>	SiH <sub>3</sub> <sup>-</sup>	55.73	10.95	9.39	68.84	68.19
17	AsH <sub>3</sub>	SiH <sub>3</sub> <sup>-</sup>	52.39	9.95	8.68	70.63	71.53

18	CO	H <sup>-</sup>	29.74	12.87	14.35	88.15	87.82
19	NO <sub>2</sub> <sup>-</sup>	H <sup>-</sup>	42.87	14.28	14.89	69.56	74.68
20	PH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub> <sup>-</sup>	50.61	15.55	14.51	76.27	75.05
21	H <sub>2</sub> O	SiH <sub>3</sub> <sup>-</sup>	21.57	6.79	10.91	101.11	102.36
22	F <sup>-</sup>	SiH <sub>3</sub> <sup>-</sup>	38.74	8.31	11.04	82.90	85.18
23	SMe <sub>2</sub>	CH <sub>3</sub> <sup>-</sup>	40.52	13.40	12.82	89.51	91.02
24	CO	SiH <sub>3</sub> <sup>-</sup>	32.06	7.83	12.03	90.99	91.87
25	F <sup>-</sup>	H <sup>-</sup>	36.18	14.15	13.61	80.87	81.38
26	PMe <sub>3</sub>	PMe <sub>3</sub>	46.10	35.97	36.75	35.97	36.73
27	AsMe <sub>3</sub>	PEt <sub>3</sub>	44.90	31.88	32.89	38.65	39.45
28	AsMe <sub>3</sub>	CH <sub>3</sub>	60.34	18.58	17.44	69.85	71.20
29	C <sub>6</sub> H <sub>5</sub> <sup>-</sup>	CH <sub>3</sub> <sup>-</sup>	87.11	35.45	38.56	40.53	44.44
30	CH <sub>3</sub> <sup>-</sup>	CH <sub>3</sub> <sup>-</sup>	92.72	35.80	38.84	35.80	38.82
31	SiH <sub>3</sub> <sup>-</sup>	SiH <sub>3</sub> <sup>-</sup>	101.04	23.13	22.90	23.13	22.89
32	H <sup>-</sup>	SiH <sub>3</sub> <sup>-</sup>	94.38	21.83	23.19	29.70	29.55
33	H <sup>-</sup>	CH <sub>3</sub> <sup>-</sup>	90.98	25.63	26.59	41.42	40.57
34	CO	Br <sup>-</sup>	12.28	27.97	31.81	38.42	34.59
35	AsH <sub>3</sub>	CN <sup>-</sup>	32.89	28.93	28.18	44.27	45.38
36	F <sup>-</sup>	PH <sub>3</sub>	21.27	29.66	28.52	46.36	43.83
37	CO	Py	10.00	31.52	34.09	45.07	44.20
38	H <sub>2</sub> O	NH <sub>3</sub>	4.15	24.91	28.32	53.54	54.92
39	NH <sub>3</sub>	SMe <sub>2</sub>	17.52	39.81	41.57	37.30	35.80
40	H <sub>2</sub> O	N <sub>3</sub> <sup>-</sup>	5.42	22.29	27.05	50.55	48.88
41	H <sub>2</sub> O	AsMe <sub>3</sub>	10.54	15.88	21.93	66.82	67.22
42	H <sub>2</sub> O	PMe <sub>3</sub>	11.23	15.68	21.25	71.56	71.60
43	H <sub>2</sub> O	CH <sub>3</sub> <sup>-</sup>	20.16	9.42	12.31	111.66	111.38
44	NH <sub>3</sub>	CH <sub>3</sub> <sup>-</sup>	38.84	17.29	20.26	90.88	92.71
45	CN	CH <sub>3</sub> <sup>-</sup>	56.46	21.19	21.83	74.44	75.09
46	OH <sup>-</sup>	H <sup>-</sup>	51.43	20.51	20.03	64.03	66.12
47	AsH <sub>3</sub>	C <sub>6</sub> H <sub>5</sub> <sup>-</sup>	47.47	14.52	13.60	78.05	78.19
48	SMe <sub>2</sub>	SiH <sub>3</sub> <sup>-</sup>	42.50	10.55	10.84	78.74	81.42
49	AsMe <sub>3</sub>	AsMe <sub>3</sub>	42.97	33.95	34.82	33.95	34.80
50	AsMe <sub>3</sub>	PMe <sub>3</sub>	44.21	32.53	33.58	37.44	38.62
51	PMe <sub>3</sub>	SiH <sub>3</sub> <sup>-</sup>	65.25	16.98	17.60	55.43	58.67
52	C <sub>6</sub> H <sub>5</sub> <sup>-</sup>	SiH <sub>3</sub> <sup>-</sup>	90.40	32.31	35.27	29.47	33.52
53	CH <sub>3</sub> <sup>-</sup>	SiH <sub>3</sub> <sup>-</sup>	96.17	33.84	35.39	25.92	27.75