## Supplementary Information

Bond (Å)/angle (°)	<b>[NiRu]</b> <sup>−</sup> (initial and final steps)	[NiHRu]	[NiHRu]⁻	[NiRu(H <sub>2</sub> )]	[NiRu]
Ni-Ru	2.733	2.934	2.823	2.982	2.819
$Ru-H_1$ $Ru-H_2$		1.614	1.629	1.813 1.700	
Ni-H <sub>1</sub>		2.342	1.951	1.850	
Ni-H <sub>1</sub> -Ru		93.9	103.7	109.0	
Hinge Ni-2S-Ru	97.9	106.7	101.0	110.0	100.9
Hinge 2S-Ni-2S	171.5	176.4	173.8	174.2	174.2

**Table S1**. Structural data for the intermediate species in the H2-evolving catalytic cycle of $[Ni(xbsms)Ru(CO)_2Cl_2]$  (BP86 in DMF)

 Table S2. Mulliken spin densities for the intermediate species in the H2-evolving catalytic cycle of

 [Nixbsms)Ru(CO)2Cl2] (BP86 in DMF)

Atom	[NiRu] <sup>-</sup>	[NiHRu]	[NiHRu] <sup>-</sup>	[NiRu(H <sub>2</sub> )]	[NiRu]
Ru	0.601	0.0	0.015	0.027	0.649
Ni	-0.425	0.0	-0.674	-0.674	0.124
$Cl_2$	0.116	0.0	0.005	0.012	0.143
$\mathbf{S}_1$	-0.080	0.0	-0.104	-0.111	0.026
$S_2$	-0.079	0.0	-0.072	-0.080	-0.002
$S_3$	-0.079	0.0	-0.072	-0.080	-0.002
$S_4$	-0.079	0.0	-0.104	-0.111	0.025

Electronic Supplementary Information for Dalton Transactions This journal is © The Royal Society of Chemistry 2010 **Table S3.** Atomic (ESP) charges for the intermediate species in the H<sub>2</sub>-evolving catalytic cycle of

Atom	[Ni(xbsms)Ru(CO) <sub>2</sub> Cl <sub>2</sub> ]	[NiRu] <sup>-</sup>	[NiHRu]	[NiHRu] <sup>-</sup>	[NiRu(H <sub>2</sub> )]	[NiRu]
Ru	0.108	0.720	0.462	0.680	0.031	0.385
Ni	0.235	0.322	0.277	0.275	0.260	0.320
$Cl_1$	-0.467					
$Cl_2$	-0.406	-0.806	-0.717	-0.816	-0.530	-0.707
$\mathrm{H}_{1}$			-0.209	-0.274	-0.323	
$\mathrm{H}_{2}$					0.845	
$\mathbf{S}_1$	-0.162	-0.503	-0.272	-0.446	-0.387	-0.298
$S_2$	-0.112	-0.317	-0.183	-0.272	-0.277	-0.206
$S_3$	-0.113	-0.316	-0.182	-0.274	-0.279	-0.207
$S_4$	-0.163	-0.493	-0.276	-0.452	-0.395	-0288
Total 4S	-0.550	-1.629	-0.913	-1.443	-1.338	-0.998
$C_{21}O_{21}$	0.058	-0.270	-0.092	-0.189	0.063	-0.132
$C_{22}O_{22}$	0.058	-0.270	-0.090	-0.190	0.060	-0.132

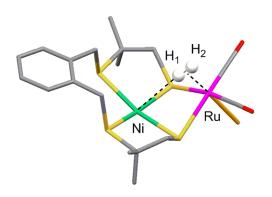
[Ni(xbsms)Ru(CO)<sub>2</sub>Cl<sub>2</sub>] (BP86 in DMF except for X-ray which was gas-phase optimized)

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Table S4. Mayer bond orders for the intermediate species in the H<sub>2</sub>-evolving catalytic cycle of

Atoms	[Ni(xbsms)Ru(CO) <sub>2</sub> Cl <sub>2</sub> ]	[NiRu] <sup>-</sup>	[NiHRu]	[NiHRu] <sup>-</sup>	[NiRu(H <sub>2</sub> )]	[NiRu]
$Ru-H_1$			0.697	0.617	0.464	
Ru-H <sub>2</sub>					0.312	
Ni-H			0.126	0.262		
Ru-Cl <sub>1</sub>	0.083					
$H_1$ - $H_2$					0.374	
Ru-C <sub>21</sub>	1.257	1.428	1.313	1.371	1.192	1.359
Ru-C <sub>22</sub>	1.258	1.430	1.314	1.371	1.190	1.360
Ru-Cl <sub>2</sub>	0.807	0.364	0.462	0.394	0.611	0.457
$Ru-S_1$	0.626	0.593	0.616	0.636	0.681	0.596
Ru-S <sub>4</sub>	0.627	0.594	0.612	0.635	0.683	0.599
$Ni-S_1$	1.050	0.775	1.023	0.735	0.770	1.054
Ni-S <sub>4</sub>	1.052	0.776	1.025	0.729	0.772	1.055
Ni-S <sub>2</sub>	0.916	0.696	0.930	0.752	0.757	0.896
Ni-S <sub>3</sub>	0.915	0.704	0.929	0.755	0.759	0.899
$C_{21}$ - $O_{21}$	2.546	2.347	2.471	2.408	2.558	2.414
C <sub>22</sub> -O <sub>22</sub>	2.546	2.349	2.471	2.407	2.560	2.414

[Ni(xbsms)Ru(CO)<sub>2</sub>Cl<sub>2</sub>] (BP86 in DMF except for X-ray which was gas-phase optimized)



**Figure S1**. Molecular structure of **[NiRu(H<sub>2</sub>)].** The nickel atom is depicted in green, ruthenium in magenta, chloride in orange and sulfurs in yellow.