Supplementary Materials for

Zinc/Nickel Exchange and Ligand Cannibalism in N₂S₂O_{1,2} Donor Ligand Sets

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Figure S1. Schlenk flasks containing the reaction mixtures from synthetic route (b) on the left and route (b') on the right were photographed against a colored background to highlight the clear and cloudy makeup of each colorless solution.



Figure S2. Reaction flask containing the 2 Zn: 1 Ni zinc displacement reaction solution in CH_3OH . The blue color was observed upon immediate addition of the light green Ni to the colorless Zn solution.



Figure S3. A Schlenk flask containing the zinc displacement reaction of $Ni(BF_4)_2$ (light green) with **Zn-1'-Ac**₂ (colorless) in CH₃OH. The blue color was observed after approx. 30 min of reaction time.

Table S1. Selected Bond Lengths (Å) for [Zn-1']₂, Zn-1'-Ac, and Zn-1'-Ac₂.

	[Zn-1'] ₂	Zn-1'-Ac	Zn-1'-Ac ₂
Zn(1) - S(thiolate)	2.307* 2.496**	2.263	
Zn(1) – S(thioether)		2.587	2.552 2.601
Zn(1) – O(1)		1.983	2.039 2.044

* terminal thiolate, ** bridging thiolate

Table S2. Summary of Crystallographic Data.

	Zn-1'-Ac ₂	Ni-1'-Ac ₂		
empirical formula	$C_{15}H_{30}N_2O_6S_2Zn$	$C_{13}H_{28}N_2O_7S_2N_1$		
formula weight	463.90	447.20		
temperature (K)	110(2)	110(2)		
wavelength (Å)	1.54178	0.71073		
Ζ	2	4		
$D_{\text{calcd}} (\text{g} / \text{cm}^3)$	1.559	1.610		
$\mu (\mathrm{mm}^{-1})$	4.001	1.315		
crystal system	Triclinic	Monoclinic		
space group	P-1	P2(1)		
<i>a</i> (Å)	7.5042(14)	7.731(2)		
<i>b</i> (Å)	8.0905(15)	7.703(2)		
<i>c</i> (Å)	16.774(4)	30.984(9)		
α (°)	78.684(11)	90		
β (°)	81.945(11)	90.282(5)		
γ (°)	86.515(10)	90		
$V(A^{-3})$	988.2(3)	1845.1(9)		
Goodness-of-fit	1.067	1.003		
$R1^{a}, wR2^{b} (\%) [I > 2\sigma(I)]$	0.0368, 0.0982	0.0572, 0.1238		
$R1^{a}$, w $R2^{b}$ (%) (all data)	0.0415, 0.1005	0.0629, 0.1290		
^{<i>a</i>} R1 = $\Sigma F_o - F_c / \Sigma F_o$. ^{<i>b</i>} wR2 = $[\Sigma [w(F_o^2 - F_c^2)^2 / \Sigma w(F_o^2)^2]^{1/2}$.				