

Supplemental information

for

Preparation, Structure and Analysis of the Bonding in the Molecular Entity $(\text{OSO})_2\text{Li}\{[\text{AlF}(\text{OR}_F)_3]\text{Li}[\text{Al}(\text{OR}_F)_4]\}$ ($\text{R}_F =$ $\text{C}(\text{CF}_3)_3$).

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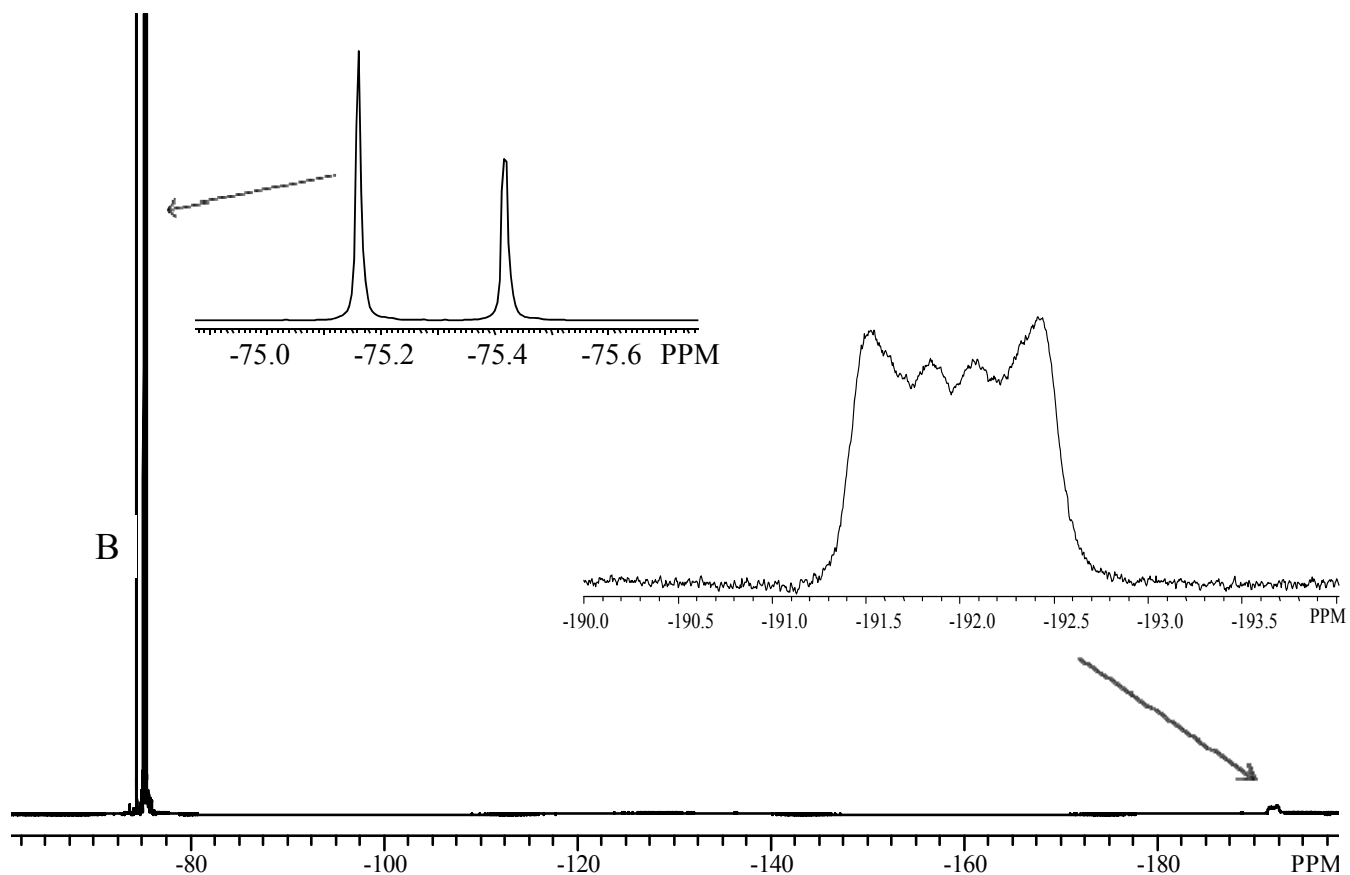


Figure S1. ^{19}F NMR spectrum of **1** in SO_2 at room temperature, **B** = minor impurity of $\text{Li}[\text{Al}(\text{OC}(\text{CF}_3)_3)_4]$. Relative intensities of three resonances of **A** = 36 : 28 : 1. Relative molar intensities of **A** : **B** = 100 : 16. Fine structure of resonance at -192 ppm has four components, separation between them are 122 Hz, 89 Hz, 122 Hz (from left to right).

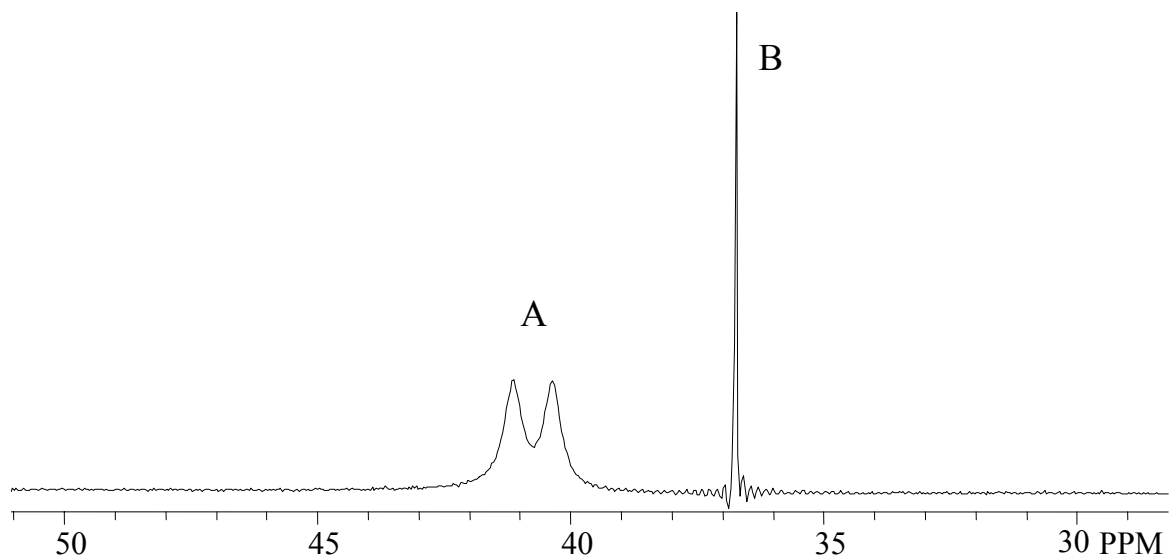


Figure S2. ^{27}Al NMR spectrum of **1** in SO_2 at room temperature, A = 1, B = impurity of $\text{Li}[\text{Al}(\text{OC}(\text{CF}_3)_3)_4]$. Relative intensities A : B = 100 : 19. Separation between two resonances of A is 80.0 Hz. The linewidth for two resonances of **1** is $\Delta\omega_{1/2} = 51$ Hz and 50 Hz. Linewidth of B is $\Delta\omega_{1/2} = 10$ Hz.

Notes and interpretation. The Al-F spin-spin coupling constant is reported only for one aluminum fluoride complex $[\text{AlF}_4]^-$. ^{19}F NMR resonance of $[\text{AlF}_4]^-$ is located at -194.2 ppm, $J_{\text{Al-F}} = 37.8$ Hz; ^{27}Al NMR resonance at 49.2 ppm, $J_{\text{Al-F}} = 37.8$ Hz. The spin-spin coupling constants are equal in ^{19}F and ^{27}Al NMR spectra. Symmetry of environment of both aluminums in **1** is lower than tetrahedral, however separation of components of multiplet (-192.0 ppm) in ^{19}F NMR spectrum and separation between two resonances of A in ^{27}Al NMR spectra are not comparable. Therefore ^{27}Al NMR resonances at 40.4 and 41.3 ppm are due to two magnetically non equivalent aluminums in **1** and not due to coupling Al-F. Fine structure of ^{19}F NMR resonance may be due to spin-spin coupling of fluorine nuclei with the ^{27}Al nuclei and ^6Li and ^7Li nuclei.



Figure S3. Sublimation vessel (Ace Glass Inc.) used for subliming Li[Al(OR_F)₄].