

*Electronic Supplementary Information*

**Hexanuclear Fe<sup>III</sup><sub>2</sub>Co<sup>III</sup><sub>2</sub>M<sup>II</sup><sub>2</sub> (M = Cu, Ni, Mn) Clusters Based on Kläui's Tripodal Ligand and Tricyanometalates: Syntheses, Structures and Magnetic Properties**

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**Table S1.** Selected bond lengths (Å) and angles (deg) for **1**

<i>Bond lengths</i>			
Fe1–C1	1.924(4)	Fe1–C2	1.910(4)
Fe1–N8	1.962(3)	Fe1–C3	1.925(5)
Fe1–N6	1.987(3)	Fe1–N4	1.966(3)
Co1–P3	2.1507(14)	Co1–P1	2.1641(14)
Co1–P2	2.1684(17)	Cu1–N2	1.962(4)
Cu1–N1	1.972(4)	Cu1–O1	1.950(3)
Cu1–O7	1.978(3)	Cu1–O4	2.178(3)
C1–N1	1.132(6)	C2–N2#1	1.145(6)
C3–N3	1.131(6)		
<i>Bond angles</i>			
P3–Co1–P1	88.74(6)	P3–Co1–P2	90.32(6)
P1–Co1–P2	93.22(6)	O1–Cu1–N1	88.15(13)
O1–Cu1–N2	165.95(16)	O1–Cu1–O7	89.00(13)
N2–Cu1–N1	92.56(14)	N1–Cu1–O7	170.14(15)
N2–Cu1–O7	87.94(13)	N2–Cu1–O4	99.60(15)
O1–Cu1–O4	94.25(14)	O7–Cu1–O4	92.86(14)
N1–Cu1–O4	96.76(15)	N1–C1–Fe1	176.9(4)
N2#1–C2–Fe1	179.7(4)	N3–C3–Fe1	179.5(5)
C1–N1–Cu1	173.1(4)	C2#1–N2–Cu1	174.3(4)

Symmetry transformations used to generate equivalent atoms: #1  $-x + 1, -y + 1, -z$

**Table S2.** Selected bond lengths (Å) and angles (deg) for **2**

<i>Bond lengths</i>			
Fe1–C1	1.908(4)	Fe1–C2#1	1.906(4)
Fe1–C3	1.942(4)	Fe1–N4	2.008(3)
Fe1–N6	1.997(4)	Fe1–N8	1.992(3)
Co1–P3	2.1125(12)	Co1–P1	2.1572(12)
Co1–P2	2.1914(12)	Cu1–N1	1.983(3)
Cu1–N2	1.960(4)	Cu1–O1	2.116(3)
Cu1–O4	1.970(3)	Cu1–O7	1.944(3)
C1–N1	1.133(5)	C2–N2	1.165(5)
C3–N3	1.072(5)		
<i>Bond angles</i>			
P3–Co1–P1	94.15(5)	P3–Co1–P2	87.95(5)
P1–Co1–P2	87.94(12)	O1–Cu1–O4	95.75(11)
O1–Cu1–O7	94.29(11)	O4–Cu1–O7	87.94(12)
N1–Cu1–O1	96.12(13)	N1–Cu1–O4	89.71(13)
N1–Cu1–O7	169.50(13)	N2–Cu1–O1	99.91(12)
N2–Cu1–O4	163.93(13)	N2–Cu1–O7	87.30(13)
N1–Cu1–N2	92.20(14)	N1–C1–Fe1	177.9(4)

N2–C2–Fe1#1	178.8(3)	N3–C3–Fe1	177.3(4)
C1–N1–Cu1	172.3(3)	C2–N2–Cu1	172.8(3)

Symmetry transformations used to generate equivalent atoms: #1  $-x + 1/2, -y + 1/2, z$

**Table S3.** Selected bond lengths (Å) and angles (deg) for **3**

<i>Bond lengths</i>			
Fe1–C3	1.913(9)	Fe1–C2	1.913(7)
Fe1–N4	1.951(6)	Fe1–C1	1.931(7)
Fe1–N6	1.966(5)	Fe1–N8	1.958(6)
Co1–P3	2.155(2)	Co1–P1	2.156(2)
Co1–P2	2.162(2)	Cu1–N2	1.961(6)
Cu1–N1	1.969(6)	Cu1–O7	1.947(4)
Cu1–O1	1.948(5)	Cu1–O4	2.219(5)
C1–N1	1.131(8)	C2–N2#1	1.168(8)
C3–N3	1.148(10)		
<i>Bond angles</i>			
P3–Co1–P1	92.16(9)	P3–Co1–P2	92.18(9)
P1–Co1–P2	89.53(8)	O1–Cu1–O4	91.6(2)
O1–Cu1–O7	90.7(2)	O4–Cu1–O7	92.9(2)
N1–Cu1–O1	86.9(2)	N1–Cu1–O4	103.2(2)
N1–Cu1–O7	163.8(2)	N2–Cu1–O1	169.4(3)
N2–Cu1–O4	98.9(2)	N2–Cu1–O7	88.1(2)
N1–Cu1–N2	91.3(2)	N1–C1–Fe1	179.3(6)
N2#1–C2–Fe1	178.8(6)	N3–C3–Fe1	177.7(8)
C1–N1–Cu1	172.2(6)	C2#1–N2–Cu1	174.8(6)

Symmetry transformations used to generate equivalent atoms: #1  $-x, -y, -z + 1$

**Table S4.** Selected bond lengths (Å) and angles (deg) for **4**

<i>Bond lengths</i>			
Fe1–C1	1.916(5)	Fe1–C2	1.917(5)
Fe1–C3	1.929(5)	Fe1–N4	1.973(4)
Fe1–N6	1.976(4)	Fe1–N8	1.976(4)
Co(1)–P(3)	2.1566(17)	Co(1)–P(1)	2.1573(17)
Co(1)–P(2)	2.1596(18)	Ni1–N1	2.061(4)
Ni1–N2	2.046(4)	Ni1–N10	2.138(5)
Ni1–O1	2.034(3)	Ni1–O4	2.071(3)
Ni1–O7	2.053(3)	C1–N1	1.141(5)
C2–N2#1	1.142(6)	C3–N3	1.130(6)
<i>Bond angles</i>			
P3–Co1–P1	90.25(6)	P3–Co1–P2	89.78(6)
P1–Co1–P2	90.91(7)	O1–Ni1–O4	89.94(13)

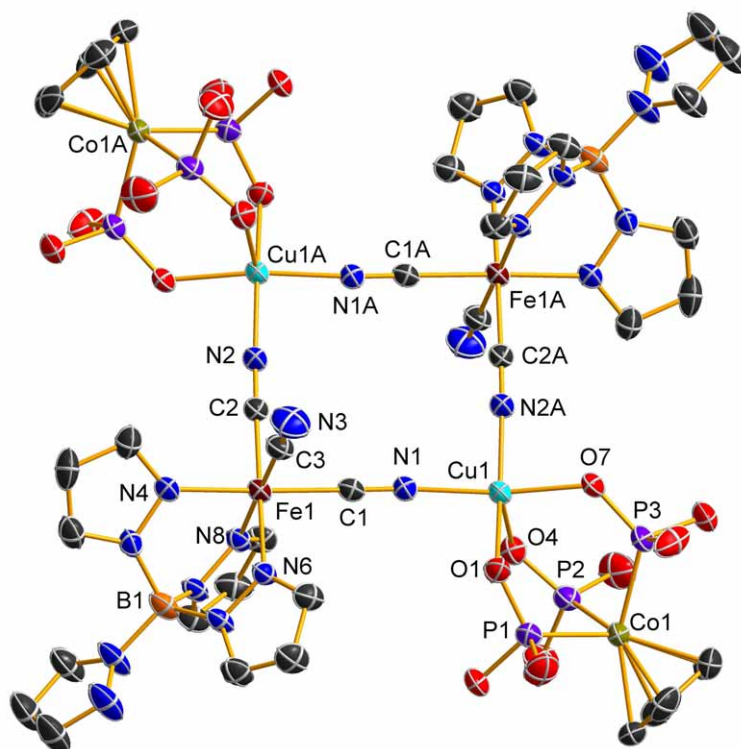
O1–Ni1–O7	88.90(14)	O4–Ni1–O7	89.89(13)
N1–Ni1–O1	89.52(14)	N1–Ni1–O4	89.55(14)
N1–Ni1–O7	178.33(14)	N2–Ni1–O1	176.74(15)
N2–Ni1–O4	92.33(15)	N2–Ni1–O7	88.76(15)
N10–Ni1–O1	91.20(16)	N10–Ni1–O4	177.90(15)
N10–Ni1–O7	91.89(16)	N1–Ni1–N2	92.84(15)
N1–Ni1–N10	88.70(17)	N2–Ni1–N10	86.60(17)
N1–C1–Fe1	178.6(4)	N2#1–C2–Fe1	177.2(4)
N3–C3–Fe1	177.8(5)	C1–N1–Ni1	175.2(4)
C2#1–N2–Ni1	171.8(4)		

Symmetry transformations used to generate equivalent atoms: #1  $-x + 1, -y + 1, -z$

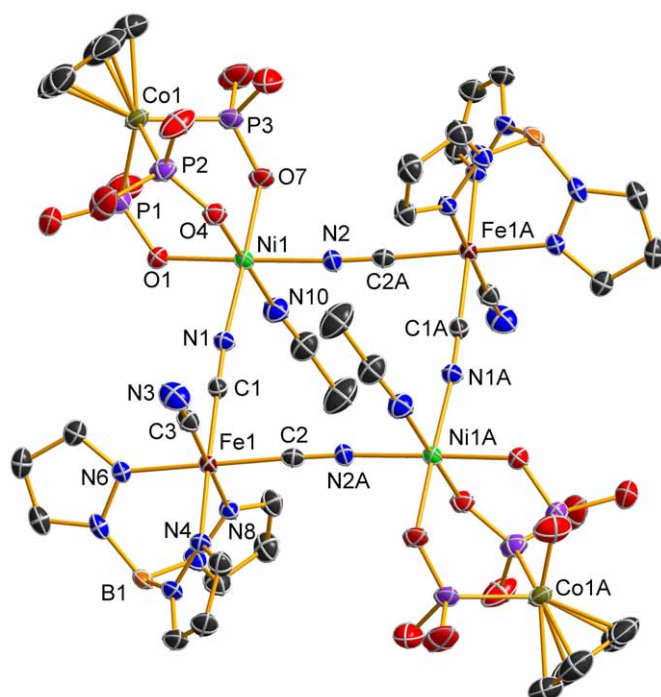
**Table S5.** Selected bond lengths (Å) and angles (deg) for **5**

<b>Bond lengths</b>			
Fe1–C1	1.906(8)	Fe1–C2	1.895(8)
Fe1–C3	1.907(8)	Fe1–N8	1.964(5)
Fe1–N6	1.964(6)	Fe1–N4	1.972(6)
Co1–P2	2.163(3)	Co1–P3	2.164(3)
Co1–P1	2.165(2)	Mn1–O4	2.130(5)
Mn1–O7	2.146(5)	Mn1–O1	2.160(5)
Mn1–N2	2.215(7)	Mn1–N1	2.218(7)
Mn1–N10	2.362(7)	C1–N1	1.152(8)
C2–N2#1	1.151(8)	C3–N3	1.150(8)
<b>Bond angles</b>			
P2–Co1–P3	90.79(10)	P2–Co1–P1	92.94(10)
P3–Co1–P1	89.85(10)	O4–Mn1–O7	88.69(19)
O4–Mn1–O1	85.8(2)	O7–Mn1–O1	88.5(2)
O4–Mn1–N2	171.2(2)	O7–Mn1–N2	97.7(2)
O1–Mn1–N2	88.3(2)	O4–Mn1–N1	90.9(2)
O7–Mn1–N1	95.9(2)	O1–Mn1–N1	174.5(2)
N2–Mn1–N1	94.5(2)	O4–Mn1–N10	89.7(2)
O7–Mn1–N10	174.5(2)	O1–Mn1–N10	86.2(2)
N2–Mn1–N10	83.4(2)	N1–Mn1–N10	89.4(2)
N1–C1–Fe1	177.6(7)	N2#1–C2–Fe1	177.0(7)
N3–C3–Fe1	175.9(7)	C1–N1–Mn1	172.3(6)
C2#1–N2–Mn1	168.9(6)		

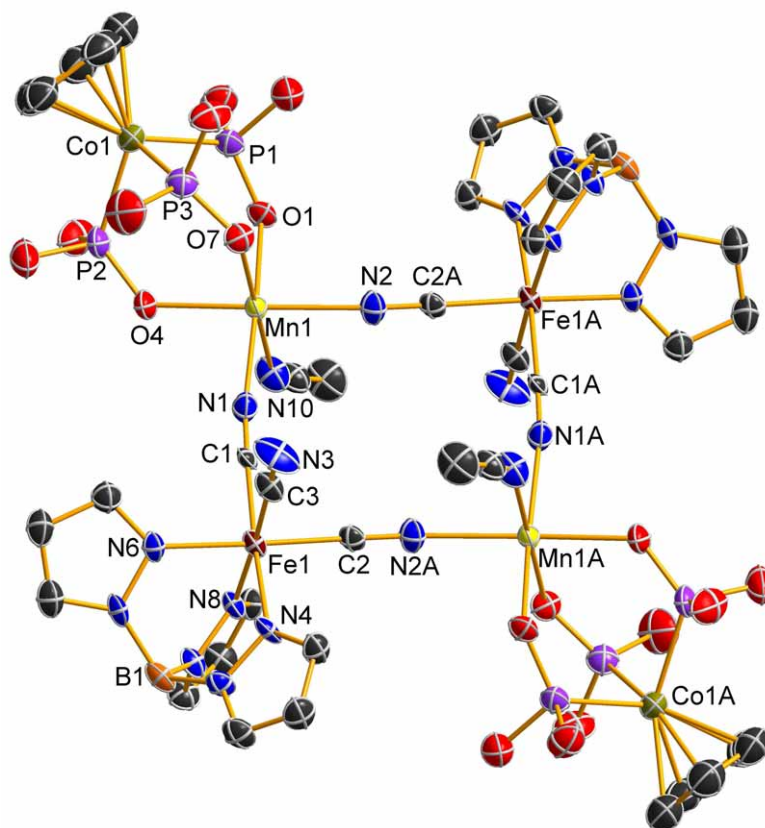
Symmetry transformations used to generate equivalent atoms: #1  $-x + 1, -y + 2, -z + 2$



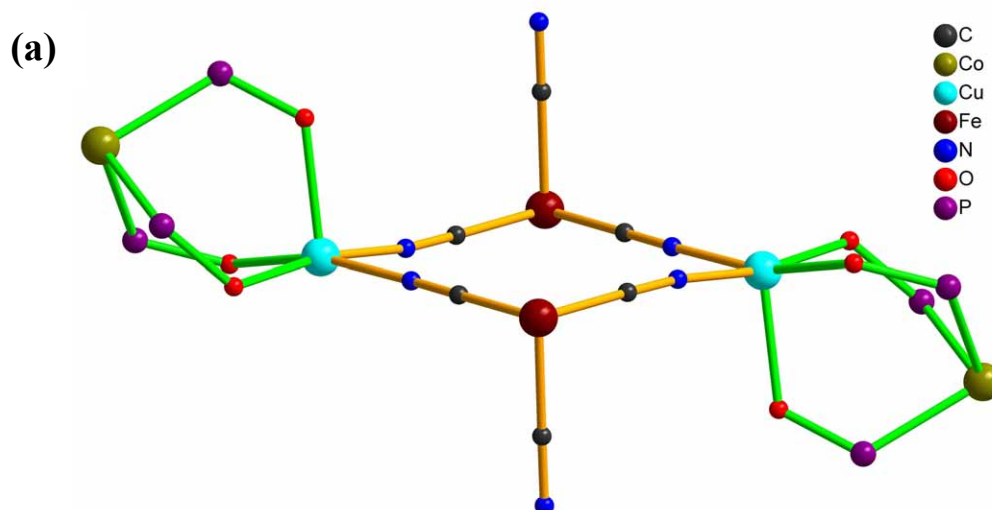
**Fig. S1** ORTEP view of the neutral hexanuclear cluster of **3** with an atom-numbering scheme at the 30% probability level. Ethyl groups of the phosphates and H atoms are omitted for clarity.

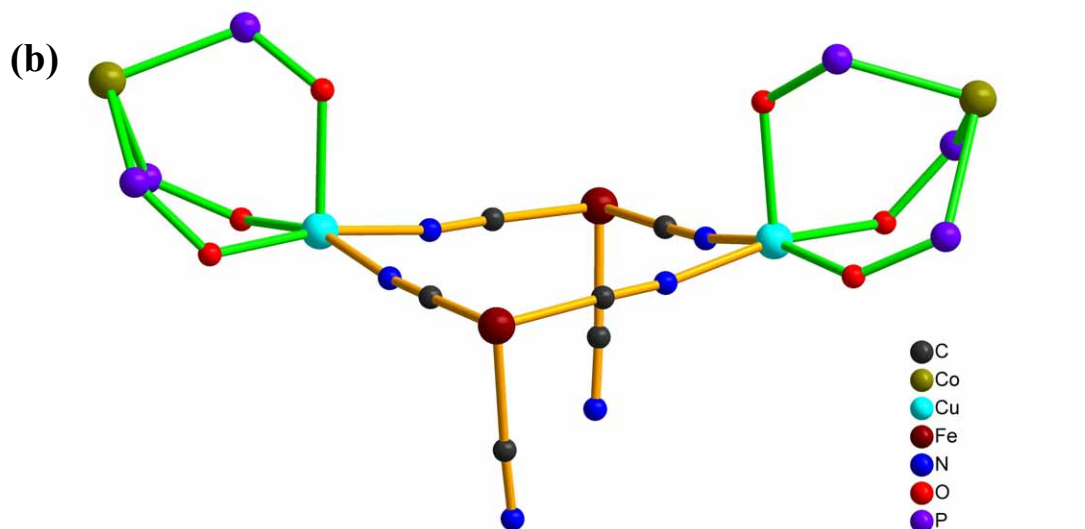


**Fig. S2** ORTEP view of the neutral hexanuclear clusters of **4** with an atom-numbering scheme at the 30% probability level. Ethyl groups of the phosphates and H atoms are omitted for clarity.

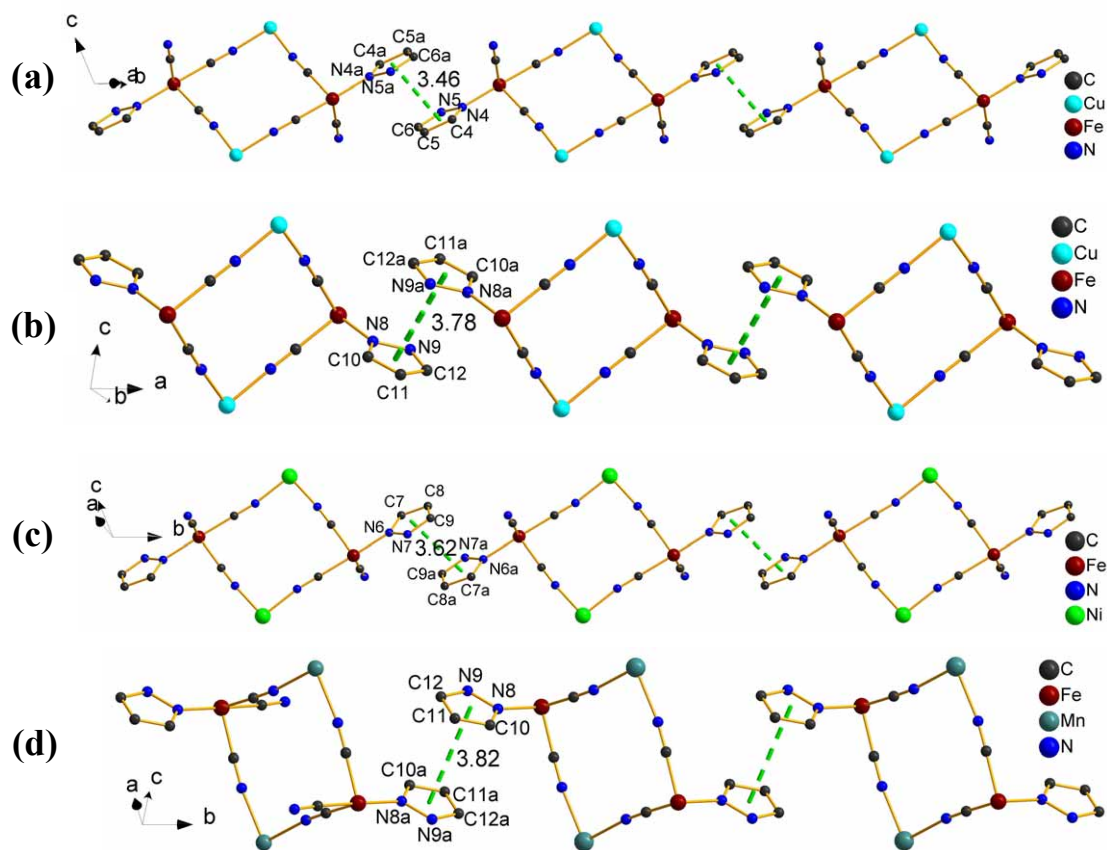


**Fig. S3** ORTEP view of the neutral hexanuclear clusters of **5** with an atom-numbering scheme at the 30% probability level. Ethyl groups of the phosphates and H atoms are omitted for clarity.





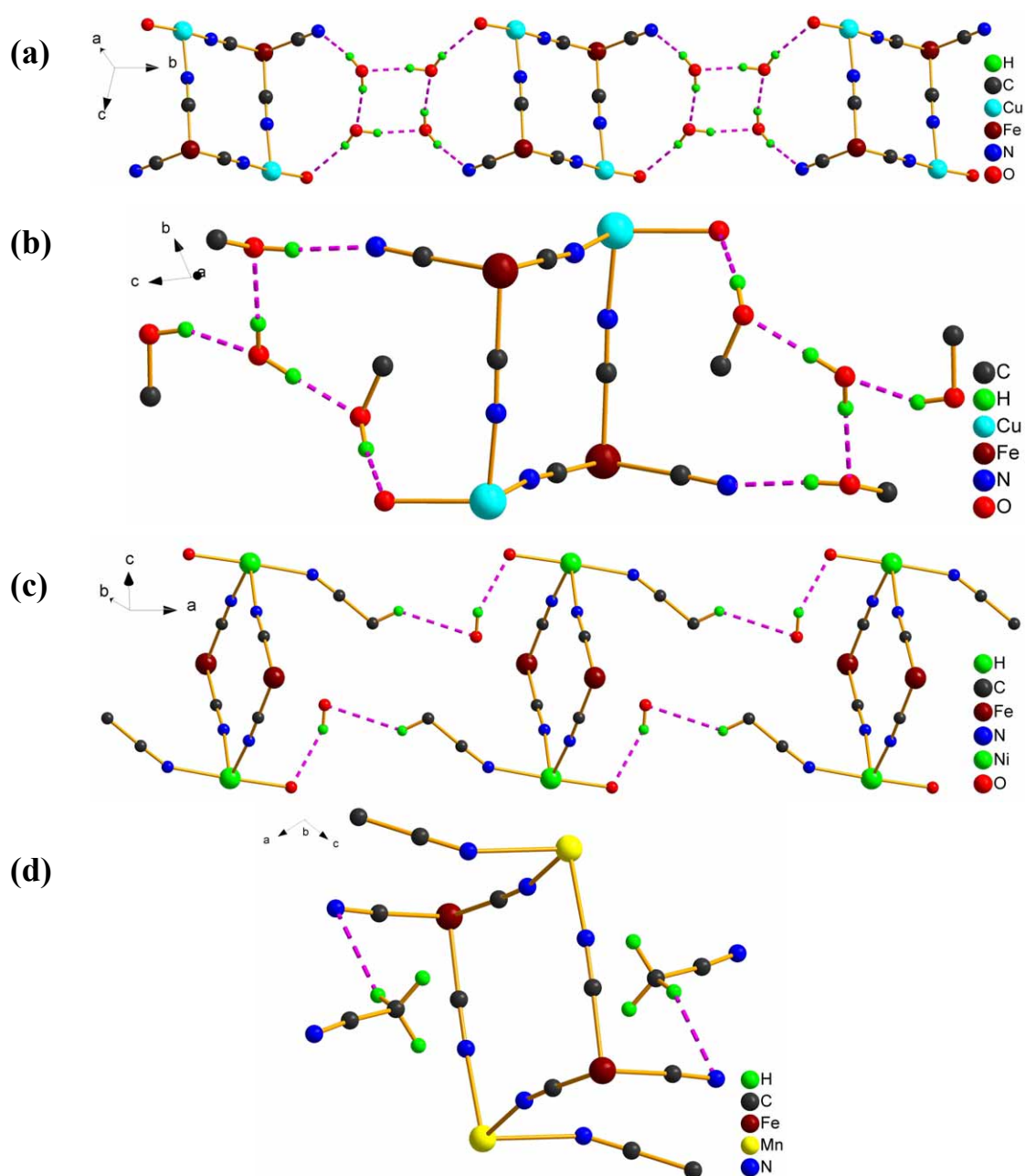
**Fig. S4** (a) A view to show the *trans* conformation of the  $L_{CoEt}$  units in complex **1**.  
(b) A view to show the *cis* conformation of the  $L_{CoEt}$  units in complex **2**.



**Fig. S5** A view to show the chains formed by  $\pi$ - $\pi$  interactions (green dash lines) of the nearby pyrazole rings in complexes **1** (a), **3** (b), **4** (c) and **5** (d).

**Table S6.** Centroid distances of  $\pi$ - $\pi$  stacking interactions between the pyrazole rings of complexes **1**, **3**, **4** and **5**.

Complexes	<b>1</b>	<b>3</b>	<b>4</b>	<b>5</b>
Centroid distances (Å)	3.4604(7)	3.7772(7)	3.6229(10)	3.8232(14)



**Fig. S6** A view to show the hydrogen-bond interactions (pink dash lines) within complexes **1** (a), **3** (b), **4** (c) and **5** (d).