Electronic Supplementary Information (ESI)

The first metal-organic framework containing an unprecedented *in situ*-generated C-substituted hexamethylenetetramine ligand

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- **Fig. S1.** 1-D chain formed via connecting Ag1 atoms by the fba ligands with alternate Ag₂O₂ rings and Ag₂C₂O₄ rings.
- **Fig. S2.** 1-D chain formed via connecting Ag2 atoms by the cphmt ligands with protrudent Ag1 atoms.
- Fig. S3 2-D tubelike network, built from the combination of two kinds of chains *via* N1 atom of cphmt ligand. The carbon atoms of hmt section in the new generated ligands denoted as gray ball.
- **Fig. S4.** 3-D framework of **1** formed by aqua ligand supporting the 2-D networks viewed along the *a*-axis. The bridging oxygen atoms of aqua ligand denoted as green ball.
- Fig. S5. TG curves for complex 1 under N₂ atmospheric pressure.
- Scheme S1 Proposed mechanism for the formation of *in-situ* generated ligand.
- Table S1
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Fig. S1 1-D chain formed via connecting Ag1 atoms by the fba ligands with alternate Ag_2O_2 rings and $Ag_2C_2O_4$ rings.

Ag2 protrudent Ag1

Fig. S2 1-D chain formed via connecting Ag2 atoms by the cphmt ligands with protrudent Ag1 atoms.



Fig. S3 2-D tubelike network, built from the combination of two kinds of chains via N1 atom of cphmt ligand. The carbon atoms of hmt section in the new generated ligands denoted as gray ball.



Fig. S4 3-D framework of **1** formed by aqua ligand supporting the 2-D networks viewed along the *a*-axis. The bridging oxygen atoms of aqua ligand denoted as green ball.



Fig. S5 TG curves for complex 1 under N_2 atmospheric pressure.



Scheme S1 Proposed mechanism for the formation of *in-situ* generated ligand.

Proposed mechanism for the formation of *in situ*-generated cphmt ligand was suggested as shown in Scheme S1. The ultimate result can be simply viewed as a substitution, *i.e.* a -CH₂- group in hmt was replaced by a HOOC-Ph-CH- group. By contrast, such C-substituted hmt ligand was not be found in the similar system with Cd(II), Cu(II), Zn(II) metal salts. Thus, the silver(I) ion may play an crucial role in the entire process. Furthermore, silver mirror phenomenon was observed in the reaction system which may suggest the existence of the byproduct HCHO.

Empirical formula	$C_{21}H_{22}N_4O_6Ag_2$	
Formula weight	642.17	
Crystal system	Triclinic	
Space group	ΡĪ	
a/Å	6.0834(5)	
<i>b</i> /Å	12.8715(11)	
$c/{ m \AA}$	13.5851(12)	
$\alpha \prime^{o}$	101.212(3)	
β'°	91.969(3)	
$\gamma/^{\circ}$	97.795(2)	
<i>V</i> /Å ³	1031.78(15)	
Ζ	2	
$D_{\rm c}/{ m mg~m}^{-3}$	2.067	
μ /mm ⁻¹	1.948	
θ range	3.06 to 25.00	
eflections collected/unique [R(int)] 7882/3585 [0.0455]		
<i>F</i> (000)	636	
R1, wR2 $[I > 2\sigma(I)]$	0.0697, 0.1870	
R1, wR2 [all data]	0.0978, 0.2196	
GOF	1.094	

Table S1 Crystal data and structure refinement parameters of complex 1

Table S2 Selected bond distances (Å) data of complex 1^a

Ag(1)-O(3)	2.199(7)	Ag(2)-O(1)	2.205(8)
$Ag(1)-O(4)^{i}$	2.272(7)	Ag(2)-O(2)	2.949(11)
Ag(1)-N(1)	2.414(7)	$Ag(2)-N(2)^{ii}$	2.211(8)
$Ag(1)-O(4)^{v}$	2.798(7)	Ag(2)-O(1W)	2.434(11)
$Ag(1)-Ag(1)^{i}$	2.9343(18)	$Ag(2)-O(1W)^{iii}$	2.786(11)

^a Symmetry operation: i, -x+3, -y+1, -z+1; ii, x-1, y-1, z; iii, -x+1, -y, -z+2; v, x-1, y, z.