

ESI to accompany

When is a metallopolymer not a metallopolymer? When it is a metallomacrocycle.

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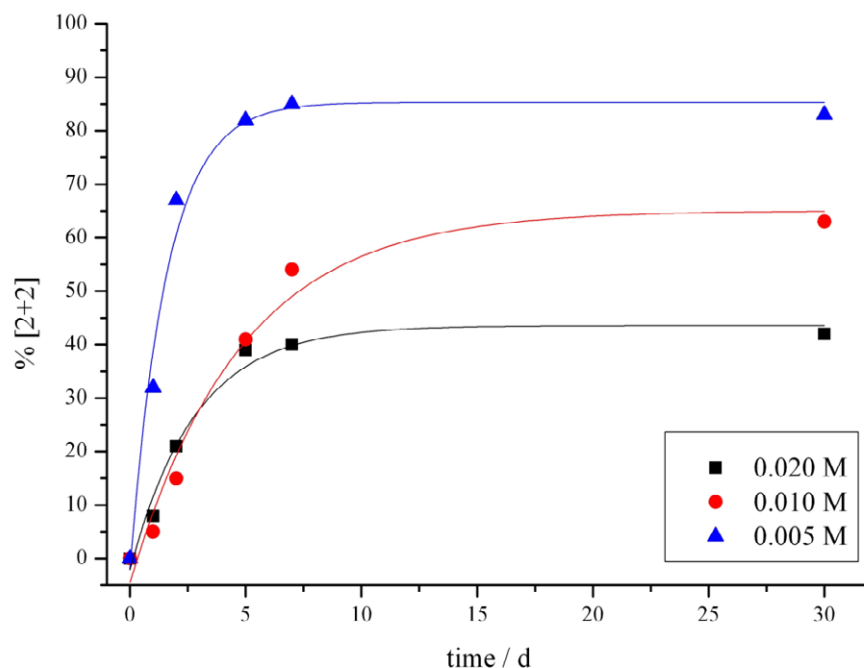


Fig. S1 Graph showing the relative proportions (determined by 250 MHz ^1H NMR spectroscopy of CD_3CN solutions) of the proposed $[\text{Co}_2(\mathbf{3})_2][\text{PF}_6]_4$ species present in mixtures formed in the reaction of $\mathbf{3}$ with $\text{Co}(\text{OAc})_2 \cdot 4\text{H}_2\text{O}$ in $\text{CHCl}_3/\text{MeOH}$ (9:1) followed by precipitation with NH_4PF_6 .

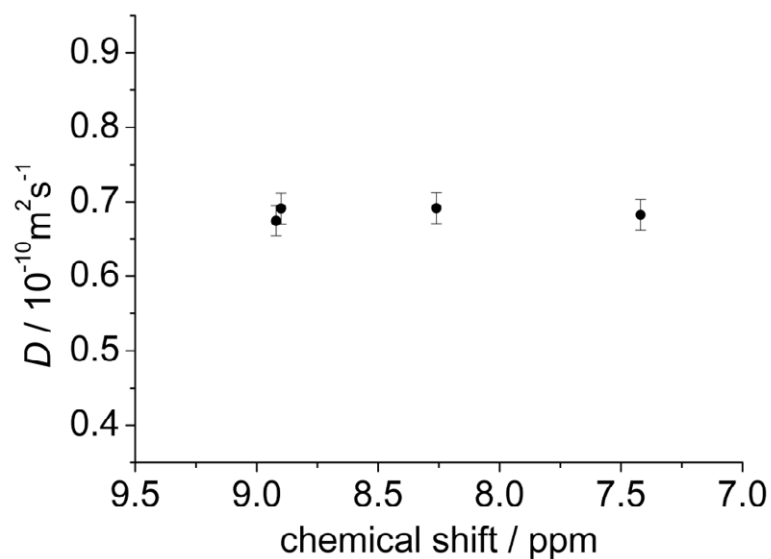


Fig. S2 ^1H DOSY-type plot (500 MHz, $\text{DMSO-}d_6$, 295 K) of an oxidized equilibrium mixture of $[\text{Co}_n(\mathbf{2})_n][\text{PF}_6]_3$ ($\Delta = 48.8$ ms, $\delta = 11.0$ ms).

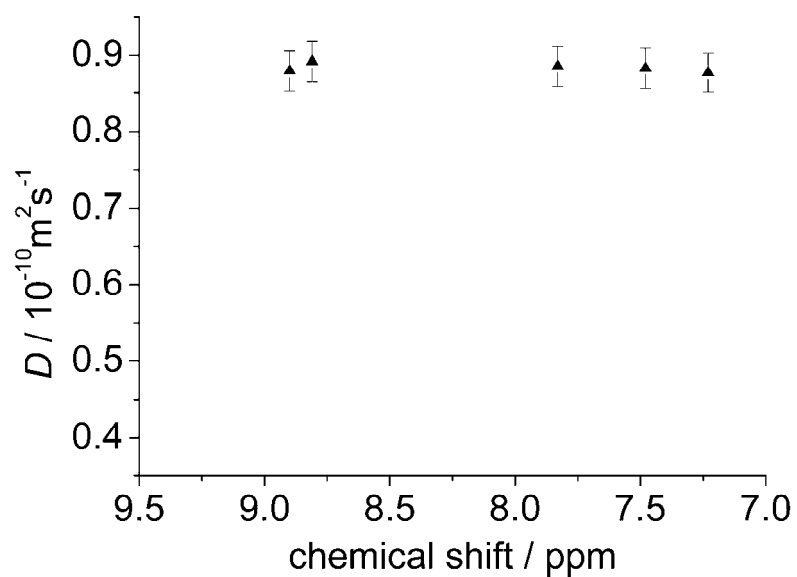


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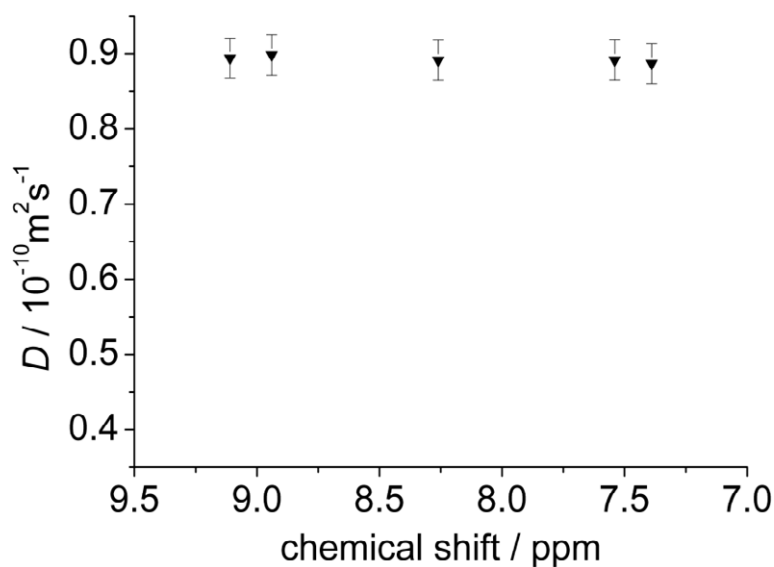


Fig. S4 ¹H DOSY-type plot (500 MHz, DMSO-*d*₆, 295 K) of an oxidized equilibrium mixture of [Co_{*n*}(**4**)_{*n*}][PF₆]₃ ($\Delta = 52.8$ ms, $\delta = 9.0$ ms).

Table S1 Calculated radii for models of the metallomacrocyclic complexes.^a

Complex	<i>r</i> / Å sphere	2 α / Å	2 β / Å	<i>r</i> _e ^d / Å ellipsoid
[Co ₃ (2) ₃] ⁹⁺	12.45	24.90	8.63	8.74 ^b
[Co ₂ (3) ₂] ⁶⁺	10.91	21.77	19.10	9.97 ^c
[Co ₂ (4) ₂] ⁶⁺	13.06	26.12	19.29	10.64 ^c

^aThe length, 2 α , of the major axis of both ellipsoids was taken as the longest distance across the metallomacrocyclic in the crystal structure. The minor axis (length 2 β) of the prolate ellipsoid was measured as the longest distance perpendicular to the major axis, and of the oblate ellipsoid was the longest distance between the H^{A4} protons of two tpy units coordinated to the same Co³⁺ ion, perpendicular to the major axis.

^bOblate ellipsoid model

^cProlate ellipsoid model

^d*r*_e is calculated from 2 α and 2 β according to: K. E. v. Holde, W. C. Johnson and P. S. Ho, *Principles of Physical Biochemistry*, 2nd edn., Pearson Prentice Hall, Upper Saddle River, 2006.

Table S2 Correction factors^a to the Stokes-Einstein equation for non-spherical model metallomacrocycles of molecular size f/f_0 , for the shape of the molecule and c , taking into account the size of the molecule.

The relationship^b used for the diffusion coefficient was:

$$D = \frac{kT}{6\pi\eta rc(f/f_0)} \text{ where } c = \left[1 + 0.695 \left(\frac{r_{\text{solvent}}}{r} \right)^{2.234} \right]^{-1}$$

Complex	f/f_0 , ellipsoid ^c	$c_{\text{ellipsoid}}$	$c_{\text{spherical}}$
$[\text{Co}_3(\mathbf{2})_3]^{9+}$	1.11 ^d	0.959 ^c	0.981
$[\text{Co}_2(\mathbf{3})_2]^{6+}$	1.00 ^e	0.969 ^c	0.975
$[\text{Co}_2(\mathbf{4})_2]^{6+}$	1.01 ^e	0.973 ^d	0.983

^a A. Macchioni, G. Ciancaleoni, C. Zuccaccia and D. Zuccaccia, *Chem. Soc. Rev.*, 2008, **37**, 479.

^b E. Yumet, H. C. Chen and S. H. Chen, *AIChE J.*, 1985, **31**, 76; H. C. Chen and S. H. Chen, *J. Phys. Chem.*, 1984, **88**, 5118.

^c K. E. v. Holde, W. C. Johnson and P. S. Ho, *Principles of Physical Biochemistry*, 2nd edn., Pearson Prentice Hall, Upper Saddle River, 2006.

^d Oblate ellipsoid model

^e Prolate ellipsoid model