

Synthesis, Characterization, and DFT Studies of Thione and Selone Cu(I) Complexes with Variable Coordination Geometries

Martin M. Kimani, Craig A. Bayse, and Julia L. Brumaghim*

SUPPLEMENTARY INFORMATION

Crystal packing diagrams for $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-I})_2\text{I}_2$ (**1**) and $\text{CuI}(\text{dmit})_2$ (**3a**) (Fig S1-S2), XRD powder patterns for copper complexes **1**, **2**, **3**, **4**, **5**, **6**, **7**, and **8** (Fig S3-S12), cyclic voltammograms for all copper complexes (Fig S13), and DPV voltammograms for tetrameric complexes **1** and **6** (Figs. S14-S15).

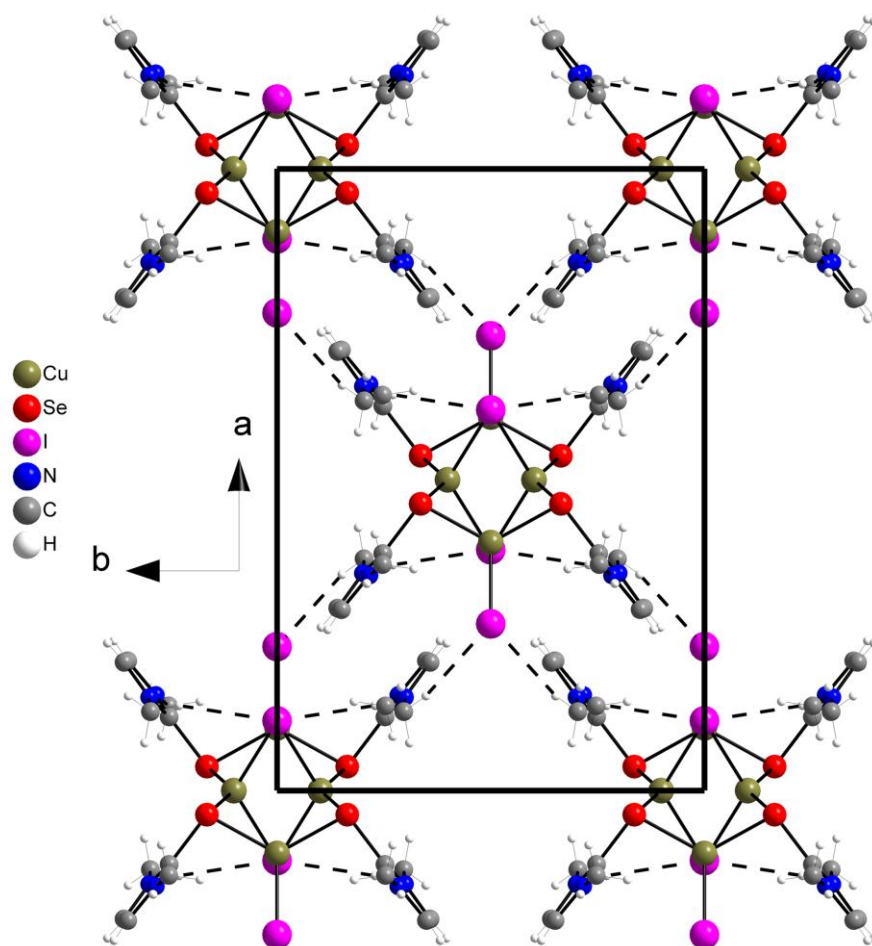


Figure S1. Crystal packing diagram of $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-I})_2\text{I}_2$ (**1**) along the c -axis depicting short contact interactions between Se and H atoms.

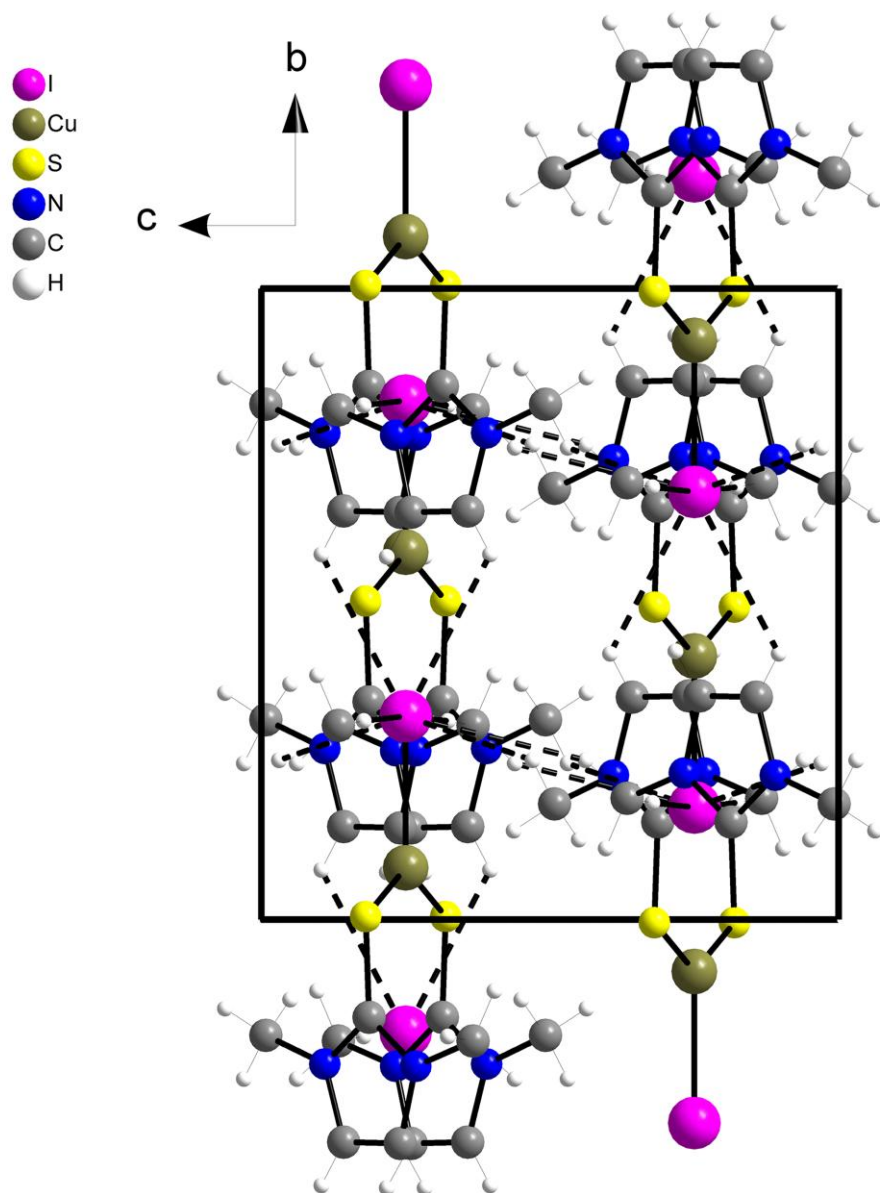


Figure S2. Crystal packing diagram of CuI(dmit)₂ (**3a**) along the *a*-axis depicting short contact interactions between Se and H atoms.

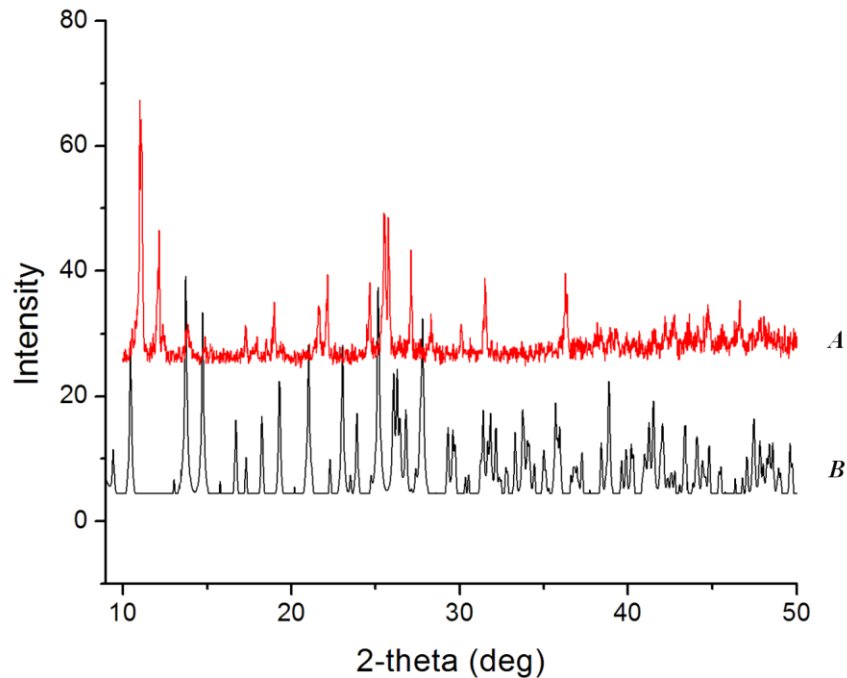


Figure S3. Powder X-ray diffraction pattern of A) $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-I})_2\text{I}_2$ (**1**), vs. simulated powder pattern B) for $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-I})_2\text{I}_2$ (**1**).

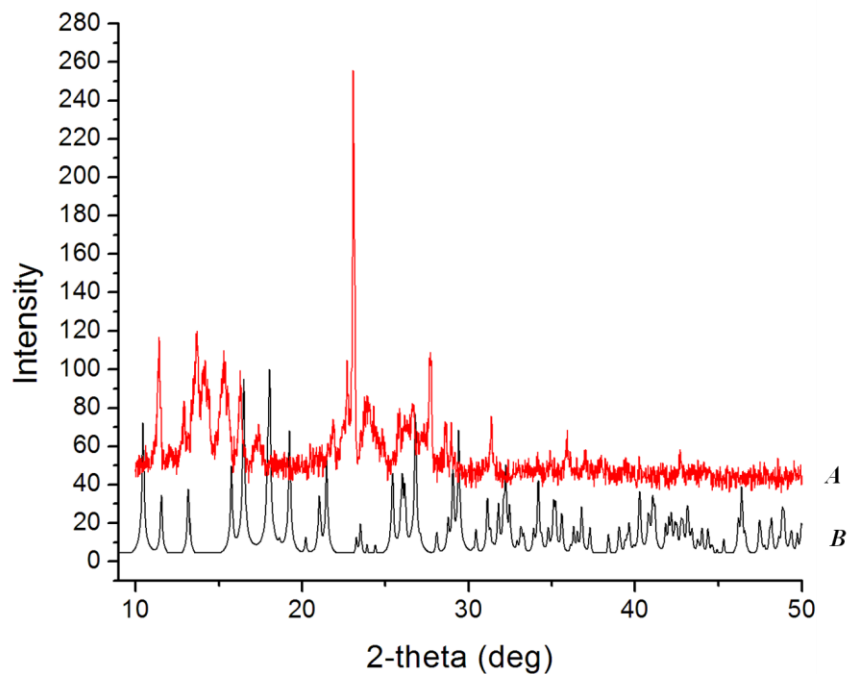


Figure S4. Powder X-ray diffraction pattern for A) $\text{CuI}(\text{dmise})_2$ (**2**), vs. simulated powder pattern for B) $\text{CuI}(\text{dmise})_2$ (**2**).

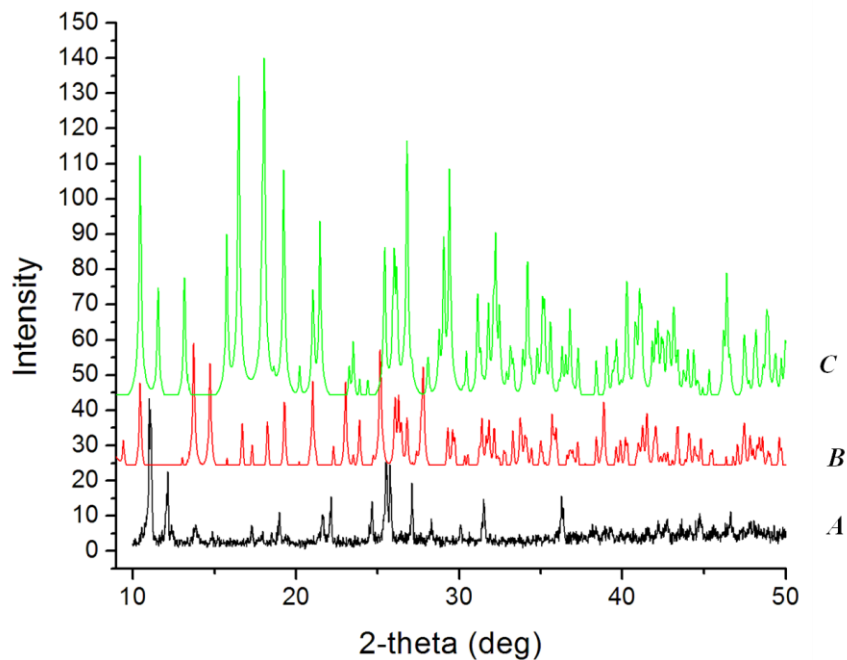


Figure S5. Experimental powder X-ray diffraction pattern of A) $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-I})_2\text{I}_2$ (**1**), vs. simulated powder pattern for B) $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-I})_2\text{I}_2$ (**1**), and C) $\text{CuI}(\text{dmise})_2$ (**2**).

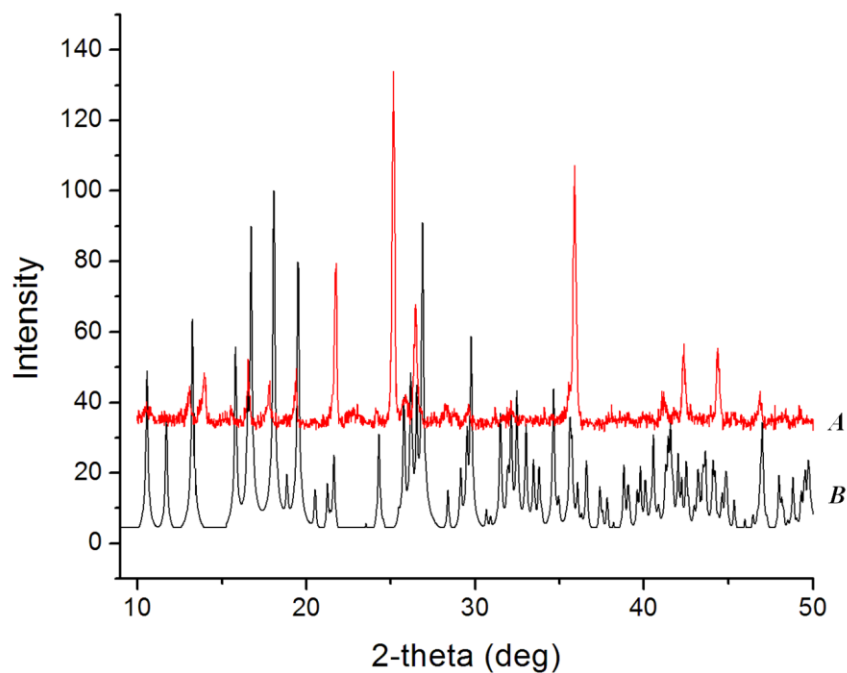


Figure S6. Powder X-ray diffraction pattern for A) $\text{CuI}(\text{dmit})_2$ (**3b**) vs. simulated powder pattern for B) $\text{CuI}(\text{dmit})_2$ (**3b**).

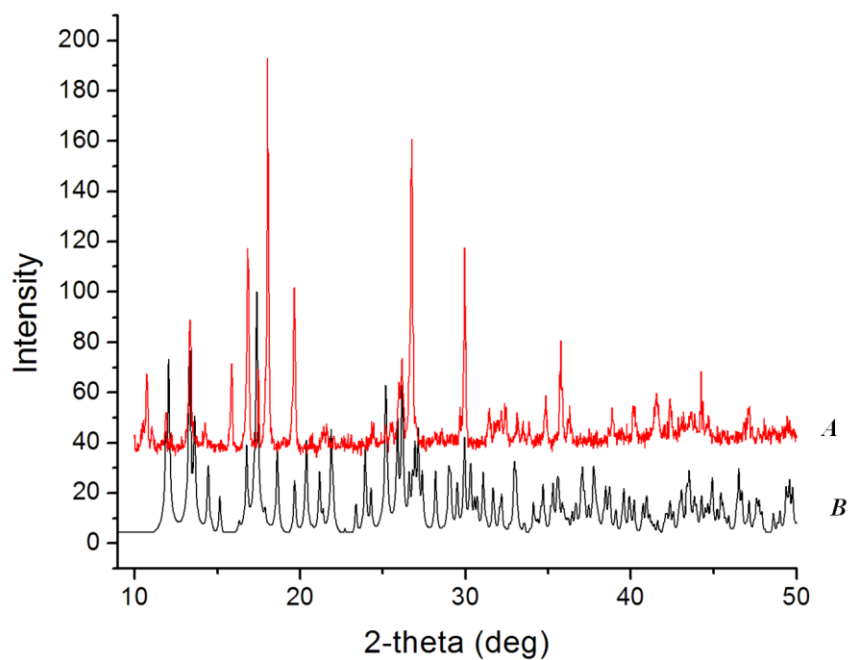


Figure S7. Powder X-ray diffraction pattern for **A**) CuI(dmit)₂ (**3a**) vs. simulated powder pattern for **B**) CuI(dmit)₂ (**3a**).

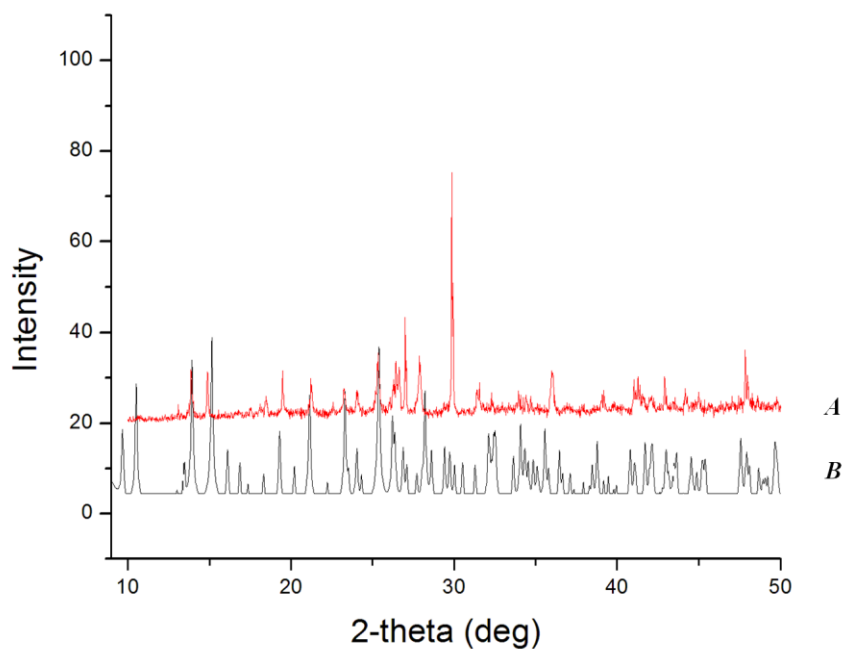


Figure S8. Experimental powder x-ray diffraction pattern of **A**) Cu₄(μ-dmise)₄(μ-Br)₂Br₂ (**6**) vs. simulated powder pattern **B**) for Cu₄(μ-dmise)₄(μ-Br)₂Br₂ (**6**).

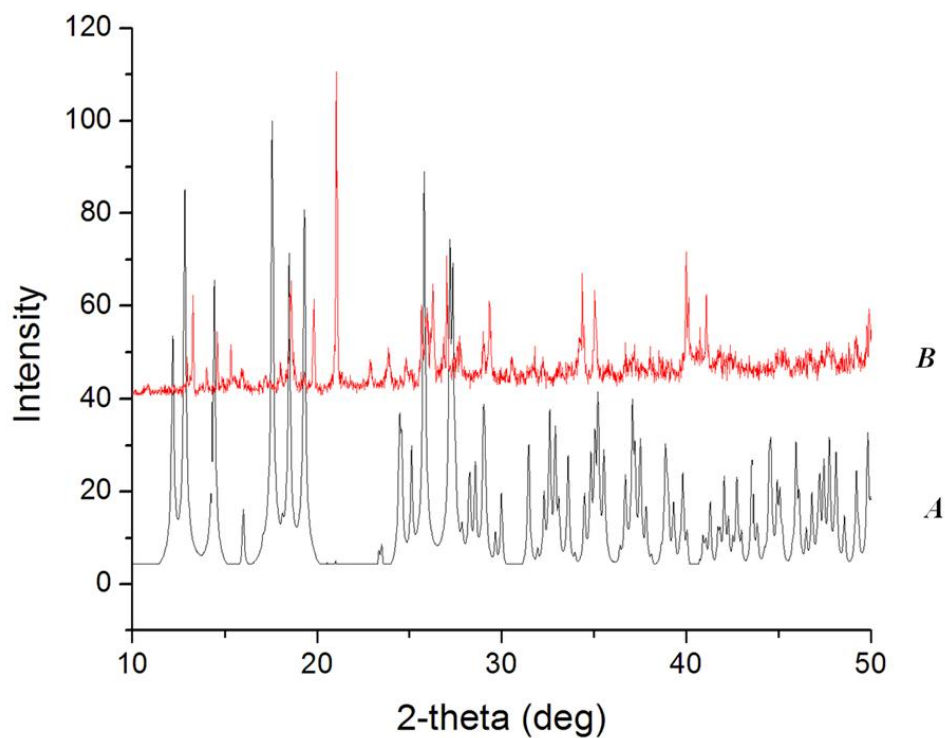


Figure S9. Experimental powder x-ray diffraction pattern of B) $\text{CuBr}(\text{dmise})_2$ (**8**), vs. simulated powder pattern A) for $\text{CuBr}(\text{dmise})_2$ (**8**).

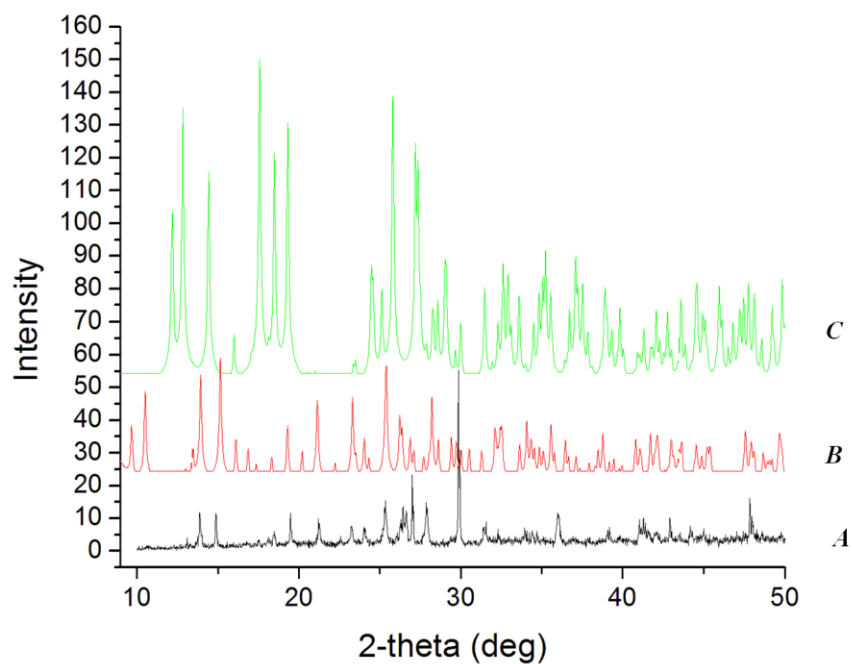


Figure S10. Experimental powder x-ray diffraction pattern of A) $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-Br})_2\text{Br}_2$ (**6**), vs. simulated powder pattern B) for $\text{Cu}_4(\mu\text{-dmise})_4(\mu\text{-Br})_2\text{Br}_2$ (**6**), and C) $\text{CuBr}(\text{dmise})_2$ (**8**).

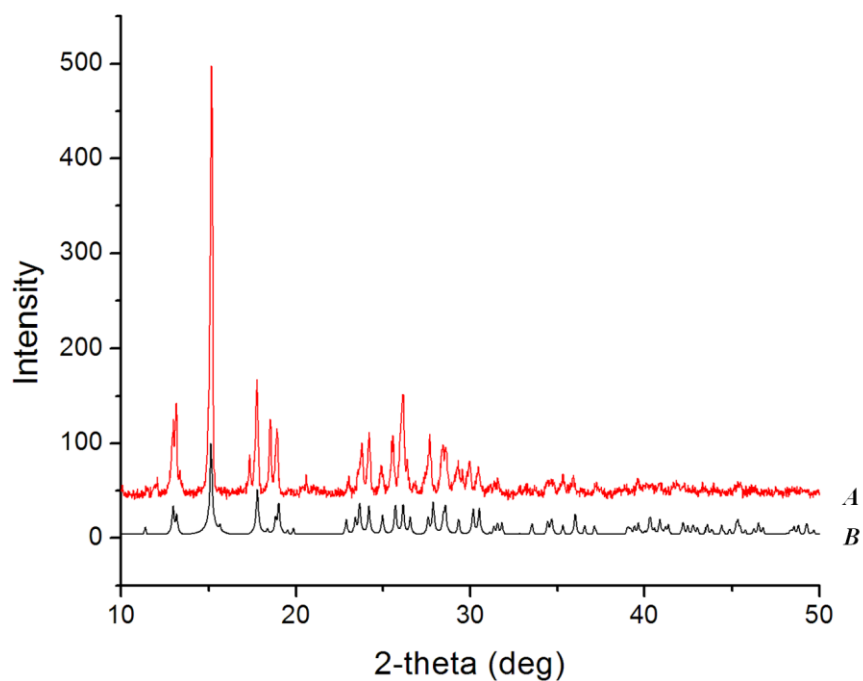


Figure S11. Experimental powder X-ray diffraction pattern of A) CuCl(dmit)_2 (**4**), vs. simulated powder pattern for B) CuCl(dmit)_2 (**4**).

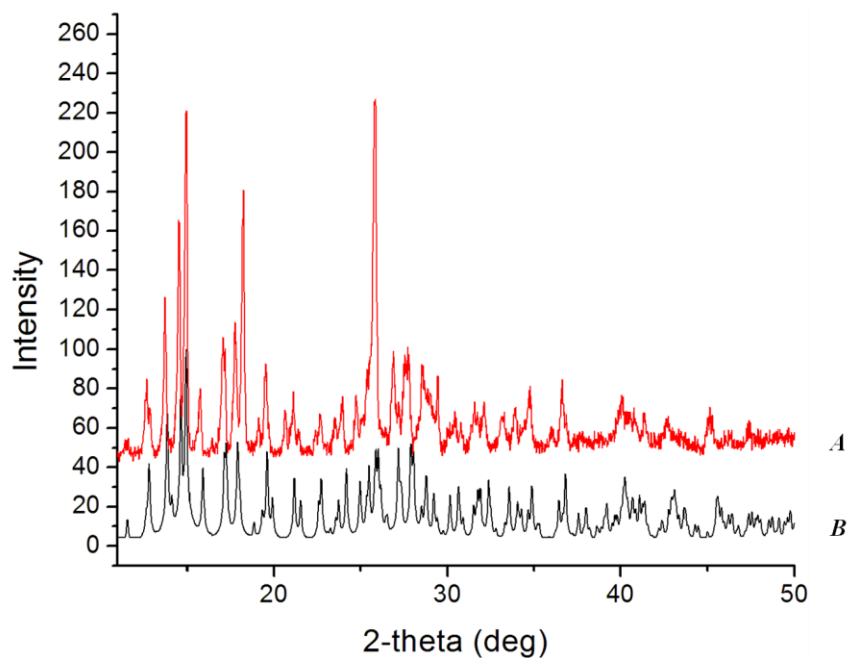


Figure S12. Experimental powder X-ray diffraction pattern of A) CuCl(dmise)_2 (**5**), vs. simulated powder pattern for B) CuCl(dmise)_2 (**5**).

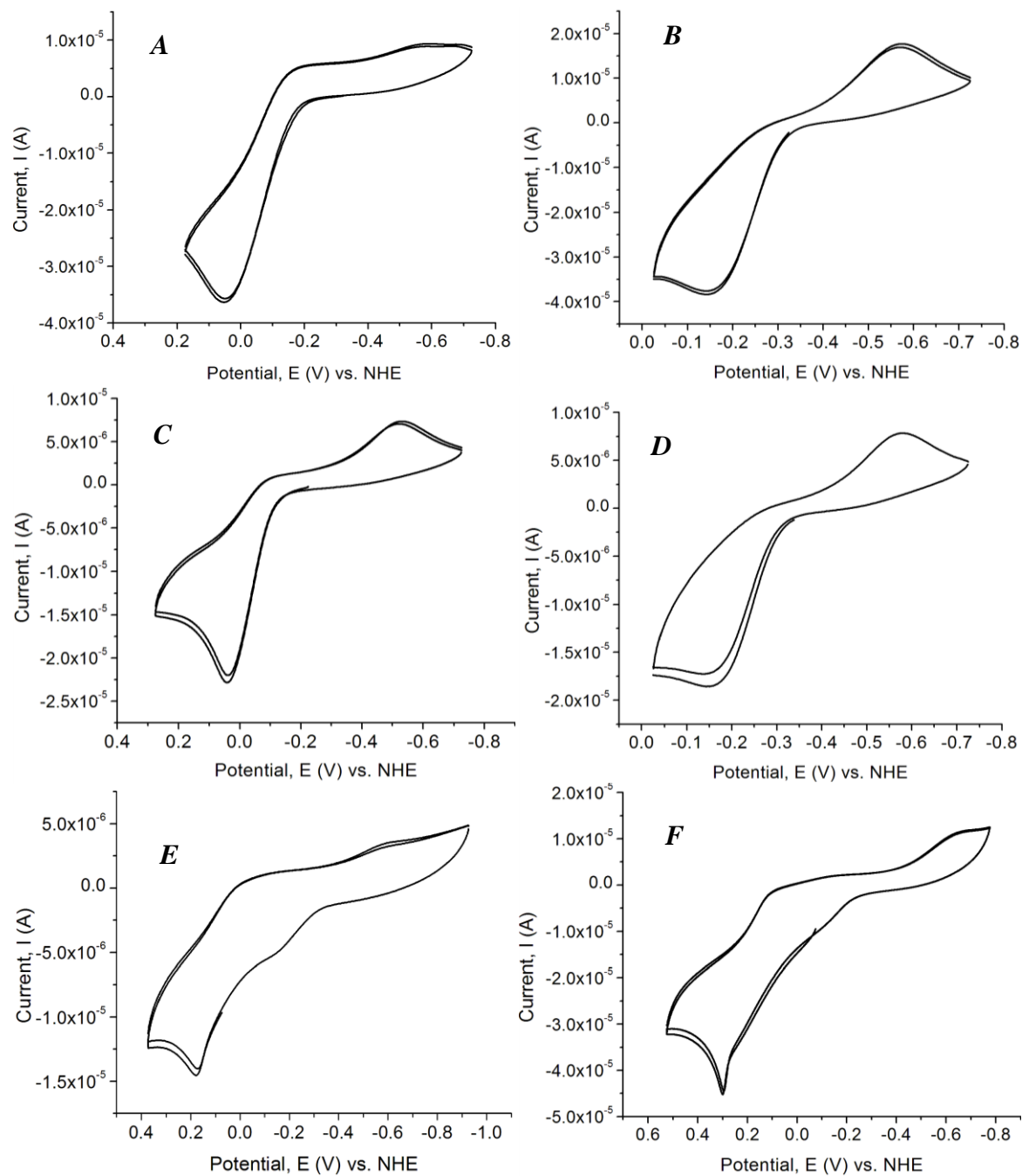


Figure S13. Cyclic voltammety scans for A) $\text{CuCl}(\text{dmit})_2$, B) $\text{CuCl}(\text{dmise})_2$, C) $\text{CuBr}(\text{dmit})_2$, D) $\text{CuBr}(\text{dmise})_2$, E) $\text{CuI}(\text{dmit})_2$ **3a**, F) $\text{CuI}(\text{dmit})_2$ **3b**. All data collected with 10^{-3} M complex in acetonitrile.

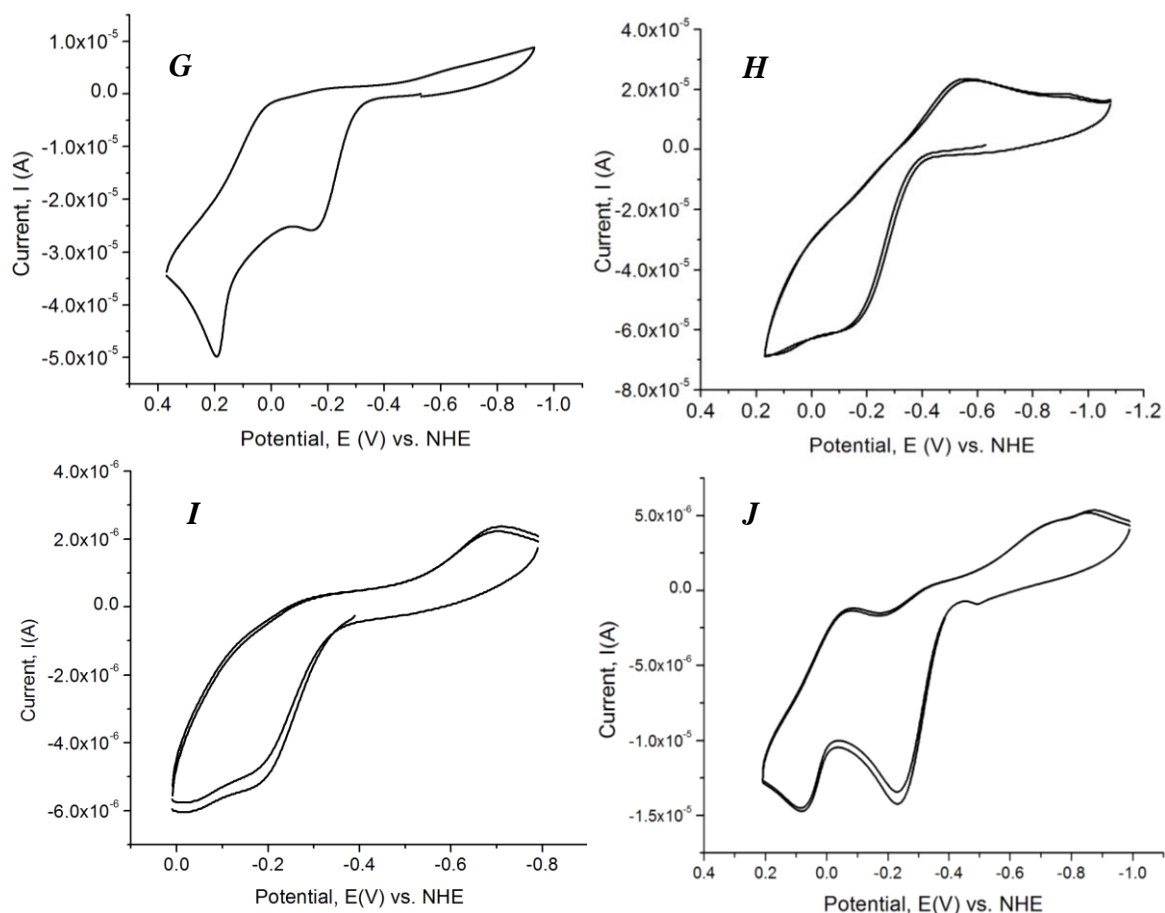


Figure S13 (continued). Cyclic voltammograms for G) mixed *trans*- and *cis*-CuI(dmit)₂ (**3a** and **3b**), H) CuI(dmise)₂, I) Cu₄(μ-dmise)₄(μ-Br)₂Br₂, J) Cu₄(μ-dmise)₄(μ-I)₂I₂. All data collected with 10⁻³ M complex in acetonitrile.

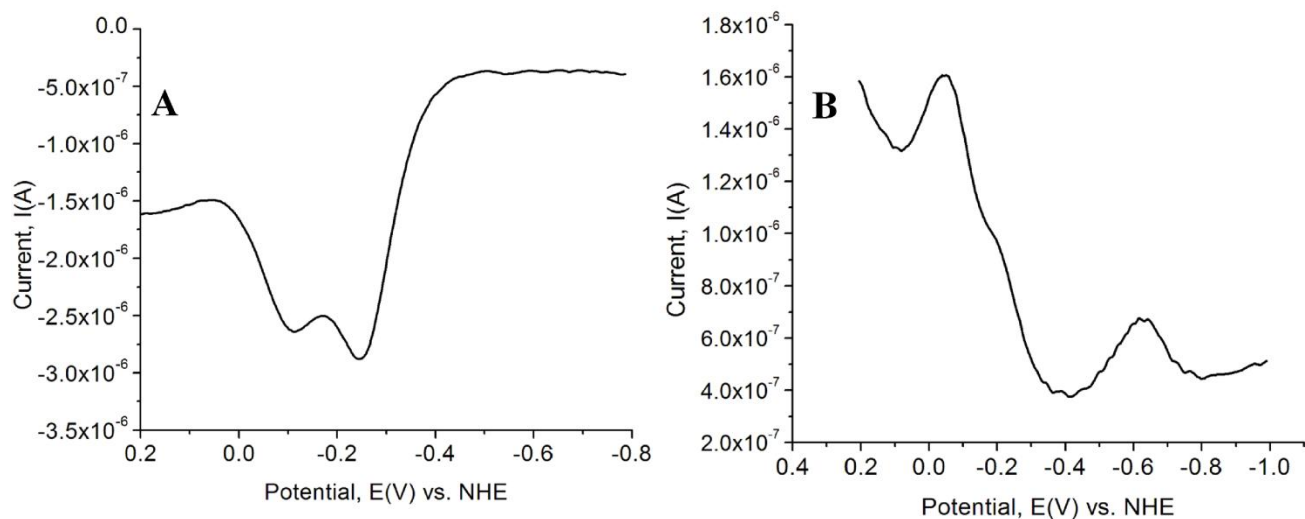


Figure S14. Differential pulse voltammograms: A) positive scan of Cu₄(μ₄-dmise)(μ-Br)₂Br₂ (**6**); B) negative scan of Cu₄(μ₄-dmise)(μ-Br)₂Br₂ (**6**). DPV data were collected at a concentration of 1 mM in acetonitrile.

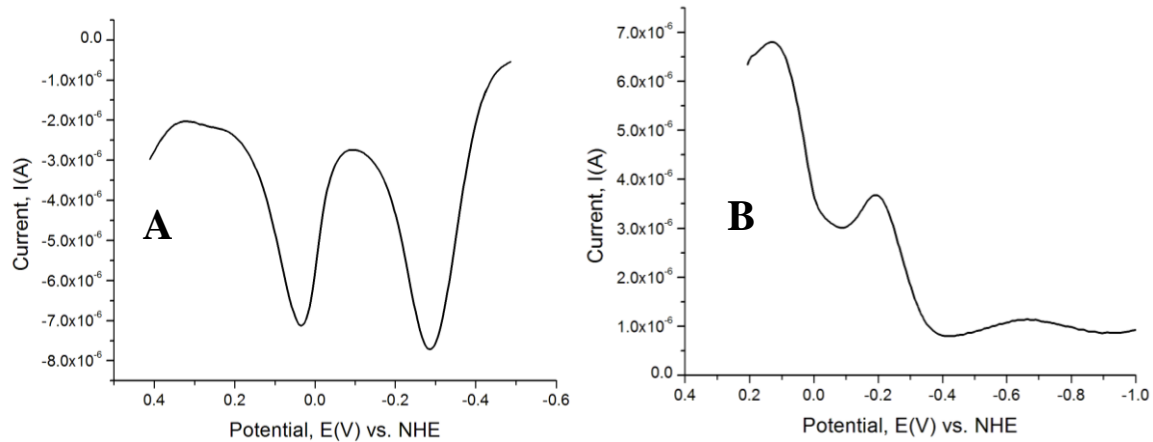


Figure S15. Differential pulse voltammograms: A) positive scan of $\text{Cu}_4(\mu_4\text{-dmise})(\mu\text{-I})_2\text{I}_2$ (**1**); B) negative scan of $\text{Cu}_4(\mu_4\text{-dmise})(\mu\text{-I})_2\text{I}_2$ (**1**). DPV data were collected at a concentration of 1 mM in acetonitrile.