

Reactivity of the bridged-sulfide complex $\text{Pd}_2\text{Cl}_2(\mu\text{-S})(\mu\text{-dmpm})_2$ toward electrophiles ; dmpm = bis(dimethylphosphino)methane

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Supplementary Material

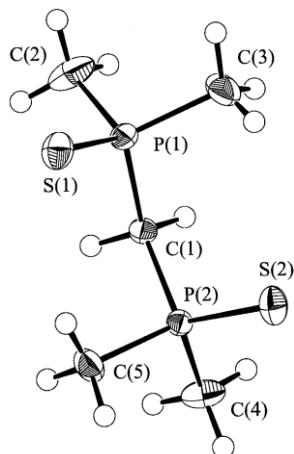


Figure S1. ORTEP diagram of dmpm(S)₂ with 50% probability ellipsoids. Bond lengths (Å) and angles (°): S(1)–P(1)=1.9504(9), S(2)–P(2)=1.9523(9), P(1)–C(1)=1.822(2), P(1)–C(2)=1.798(3), P(1)–C(3)=1.789(3), P(2)–C(1)=1.817(2), P(2)–C(4)=1.796(3), P(2)–C(5)=1.800(3); S(1)–P(1)–C(1)=115.46(8), S(1)–P(1)–C(2)=111.84(10), S(1)–P(1)–C(3)=114.41(10), S(2)–P(2)–C(1)=115.75(9), S(2)–P(2)–C(4)=112.94(10), S(2)–P(2)–C(5)=113.64(9), C(2)–P(1)–C(3)=106.45(13), C(4)–P(2)–C(5)=105.56(12), P(1)–C(1)–P(2)=118.75(12).

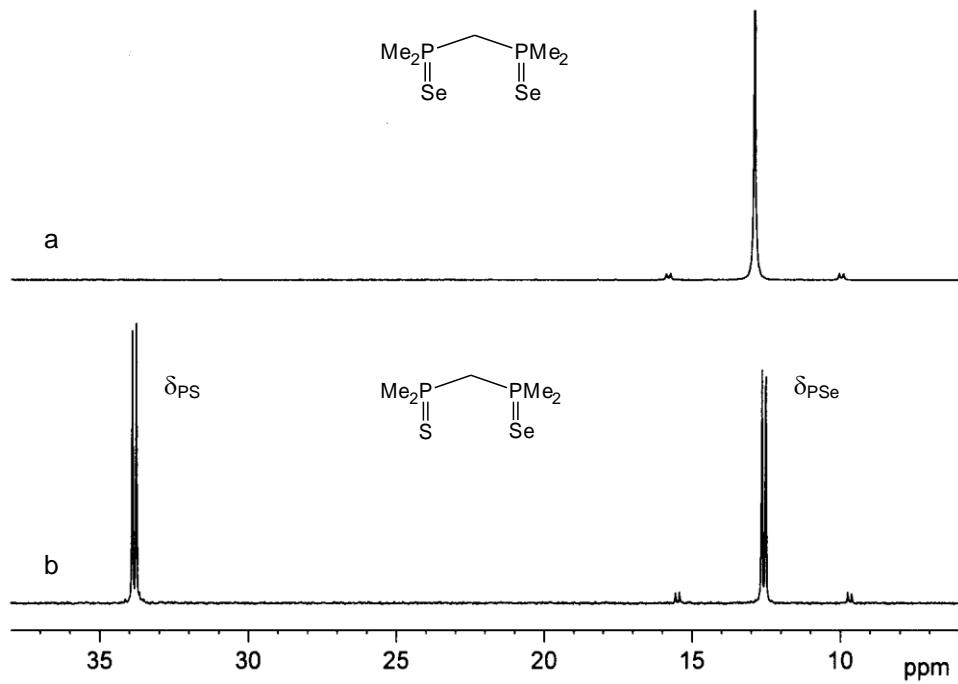


Figure S2. $^{31}\text{P}\{\text{H}\}$ NMR spectra (CDCl_3) of (a) $\text{dmpm}(\text{Se})_2$ and (b) $\text{dmpm}(\text{S})(\text{Se})$, both showing the $^2J_{\text{PP}}$ and $^1J_{\text{PSe}}$ coupling.

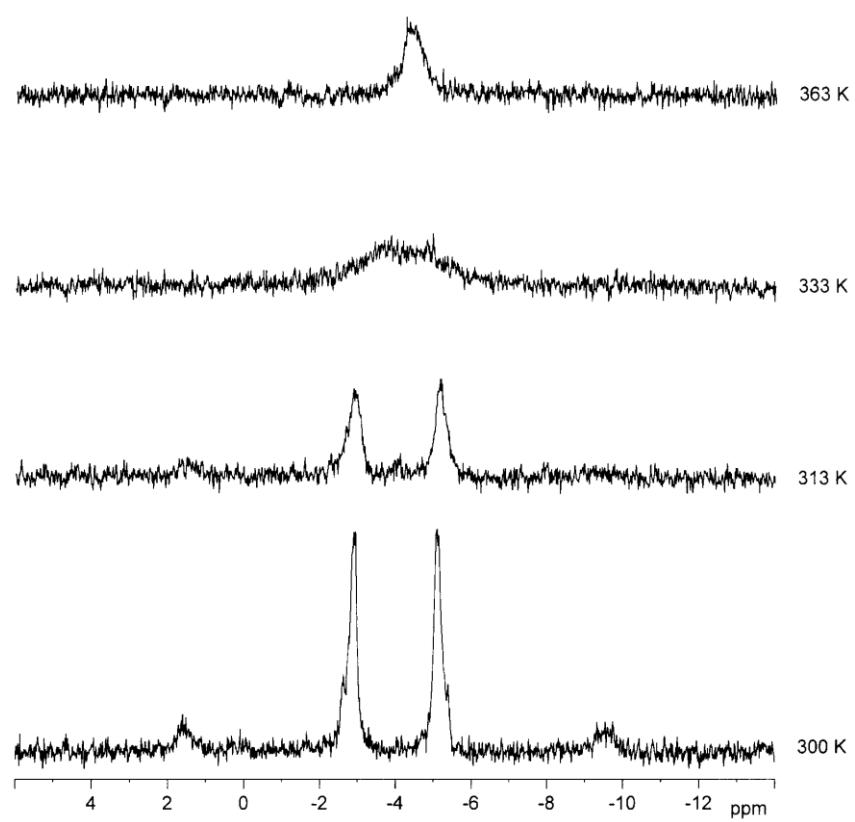


Figure S3. VT $^{31}\text{P}\{\text{H}\}$ NMR spectra (MeNO_2-d_3) of $[\text{Pd}_2\text{Cl}_2(\mu\text{-SMe})(\text{dmppm})_2]\text{OTf}$ (**5**). The AA'BB' signal collapses at $\sim 60^\circ\text{C}$ to give a broad singlet at higher temperature.